# Ab initio study of the energy competition between $\Gamma$ and K valleys in bilayer transition metal dichalcogenides

Sam Olin, <sup>1,\*</sup> Erekle Jmukhadze<sup>®</sup>, <sup>1</sup> Allan H. MacDonald, <sup>2</sup> and Wei-Cheng Lee<sup>®</sup>, <sup>1</sup> Department of Physics, Applied Physics, and Astronomy, Binghamton University, Binghamton, New York 13902, USA <sup>2</sup> Department of Physics, the University of Texas at Austin, Austin, Texas 78712, USA

(Received 30 October 2023; revised 5 February 2024; accepted 11 March 2024; published 1 April 2024)

Moiré engineering in two-dimensional van der Waals bilayer crystals has emerged as a flexible platform for controlling strongly correlated electron systems. The competition between valleys for the band extremum energy position in the parent layers is crucial in deciding the qualitative nature of the moiré Hamiltonian since it controls the physics of the moiré minibands. Here we use density functional theory to examine the competition between K and  $\Gamma$  for the valence-band maximum in homo- and hetero-bilayers formed from the transition metal dichalcogenides (TMD),  $MX_2$  where M=Mo, W and X=S, Se, Te. We shed light on how the competition is influenced by interlayer separation, which can be modified by applying pressure, by external gate-defined electric fields, and by transition metal atom d-orbital correlations. Our findings are related to several recent experiments, and contribute to the development of design rules for moiré materials.

DOI: 10.1103/PhysRevB.109.165101

## I. INTRODUCTION

Because their miniband widths can be narrow, twodimensional van der Waals material multilayer moirés provide an attractive platform for designer quantum materials. One example of moiré engineering is provided by twisted bilayer graphene, in which novel strongly correlated electronic states, including Mott insulators and superconductors, emerge only at certain magic angles between layers [1–6]. The basic mechanism of moiré engineering is periodicity at a length scale that is controllable, and in a range that allows the number of electrons per effective atom to be tuned over ranges larger then one using electrical gates. In twisted bilayer graphene flat bands emerge when the interference between intralayer and interlayer hopping is destructive [7,8].

Recently group VI transition metal dichalcogenides (TMDs) have stimulated enormous research interest because of their potential to host even richer physics under moiré engineering [6,9–13]. In monolayer form, these materials are semiconductors with direct band gaps in which both conduction-band minimum (CBM) and valence-band maximum (VBM) are at the K point [14,15] in the Brillouin zone. Nontrivial Berry curvature near the K points gives rise to a number of unique electronic and optical properties [16,17]. For bilayer TMDs, the VBM can be at  $\Gamma$  or K points depending on a variety of factors [18-24]. Since the location of the VBM controls optical properties, the moiré potential landscape, and the topological properties of the moiré band structures, a systematic investigation of the key factors that determine the location of the VBM in bilayer TMDs is timely [18,25-30].

In this paper, we employ first-principles methods to examine the trends in K and  $\Gamma$  competition within the valence bands

of twist-free homo- and hetero-bilayer TMDs. The band structures of TMDs have been studied in detail before, and several factors such as the interlayer separation [31–35] and applied gate voltage [33,36–38] have been shown to strongly influence the structure and gap. Here we seek to quantitatively expand upon the underlying physical mechanisms of these factors, including correlation and spin-orbit coupling, with particular attention on their competition at the valence band edges of K and  $\Gamma$ . In particular, stronger interlayer tunneling favors the  $\Gamma$  point while stronger spin-orbit coupling favors the K point. On top of these two opposing factors, the influence of electron correlations on the d orbitals tends to increase energy in states with more localized wave functions, and therefore to raise the energy of states near the K point. For this, we test the efficacy of the local modified Becke-Johnson potential [39,40]. In this paper we first explore these competitions at greater depth and then discuss experimentally feasible strategies to control the VBM location for moiré engineering in light of our findings.

# II. METHODS

## A. Structural relaxation

Herein we consider twist-free, homo- and heterobilayer systems with chemical formula  $MX_2$  where M=Mo, W and X=S, Se, Te. For heterobilayers, we consider only combinations containing common chalcogens in the parent layers to avoid possible inaccuracies due to large lattice constant mismatches. While many bilayer stacking arrangements exist, in this work we consider only the two high-symmetry stacking orders denoted as 2H and AA and displayed in Fig. 1. The bilayer AA stacking occurs when the  $\hat{x}$ ,  $\hat{y}$  coordinates of the metal (chalcogen) atom(s) in the top layer are the same as the metal (chalcogen) atom(s) in the bottom layer [41]. The 2H stacking is obtained by a  $180^{\circ}$  rotation about the  $\hat{z}$  axis of the top layer relative to the AA stacking case, and is predicted to be the most stable of the bilayer configurations [42,43].

<sup>\*</sup>solin1@binghamton.edu

<sup>†</sup>wlee@binghamton.edu

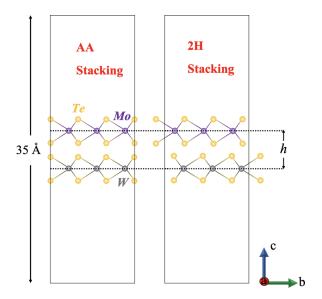


FIG. 1. Example of AA and 2H stacking in the case of the free-standing  $MoTe_2/WTe_2$  TMD heterobilayer. A vacuum of magnitude 20 Å is added to the bulk c axis lattice constant. h is the interlayer separation, characterized by the vertical distance between metal atoms.

In this case, the metal (chalcogen) atom(s) are directly above the chalcogen (metal) atom(s). These arrangements cannot be made equal by relative translations of the layers and therefore do not occur at different positions in the same moiré pattern.

We employ the Vienna *ab initio* simulation package (VASP) [44–46] to perform structural relaxation for each 2H system under the following procedure. First, full relaxation in which both the volume and shape of the unit cell may vary

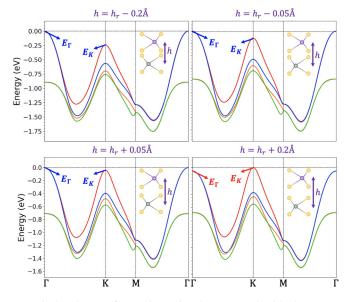


FIG. 2. The top four valence bands computed with the LmBJ functional and spin-orbit coupling, as a function of h for 2H-stacked MoS<sub>2</sub>/WS<sub>2</sub> TMD bilayers. A sketch of each crystal structure is set into each plot. The energies of the  $\Gamma$  and K points are labeled  $E_{\Gamma}$  and  $E_K$ , respectively. These results demonstrate that  $E_K$  increases relative to  $E_{\Gamma}$  as h increases, as observed in all cases considered.

to minimize the total energy and force is performed on the bulk structures using three functionals; the Perdew-Burke-Ernzerhof exchange-correlation functional with spin-orbit coupling, both with (PBE-SO-D3) and without (PBE-SO) the van der Waals correction, and the local density approximation LDA + SO. Additionally, we include the relaxation via PBE alone (as performed in Ref. [47]) for comparison. Second, we construct each "free-standing" bilayer system from the bulk lattice constants found from the LDA + SO relaxation. The total length of the c axis is set to 35 Å (Fig. 1), to isolate the bilayer and limit the unphysical interaction between periodic unit cells. This structure is then relaxed with fixed volume using LDA-SO in VASP, allowing atomic positions to change. From these structures, the bands are computed as a function of a variety of tuning parameters. For AA systems, we only employ the LDA-SO functional, for reasons explained in Sec. IV B.

#### **B.** Electronic structure

To comprehensively explore the energy competition between  $\Gamma$  and K valleys, we perform density functional theory (DFT) calculations using the full-potential linearized augmented plane wave (LAPW) method as provided by the WIEN2K package [48,49]. In addition to the standard local density approximation (LDA) functional, we employ the local modified Becke-Johnson (mBJ) functional to investigate the effect of electronic correlation [50,51]. This relatively new functional is ideally suited to low-dimensional, heterogeneous, and interface systems due to the fact that its mixing parameter takes local changes into affect via the smearing parameter in Eq. (6) of Ref. [40]. It has been shown that the mBJ functional gives very accurate band gaps in many transition metal oxides [52,53] and semiconductors [54], including VO<sub>2</sub> [55,56] and monolayer TMDs [15], with much less computational time than hybrid functional, or GW methods. Furthermore, the local mBJ (LmBJ) functional has been developed [39] for accurate prediction of band gaps in systems with vacuum space. As such, with the growing interest in first-principles-based moiré modeling, a comprehensive study on the predictive power of this potential for valence-band competition is needed. The smearing parameter, used to tune the degree of localization for computing averages was chosen to be 3.78 bohr, while the Wigner-Seitz radius was chosen to be 5 bohr, as recommended by the original authors. An unfortunate drawback, is that LmBJ cannot be used for structural relaxation. The DFT energy convergence was chosen to be 0.1 mRy while the charge convergence was set to  $0.001e^{-}$ . In all cases, we include spin-orbit interactions and employ an additional  $p_{1/2}$  radial basis function, called the relativistic local orbital (RLO) provided by WIEN2K, for the metal atoms to improve the basis functions, aiding in convergence. For the band structure, we use an in-plane momentum  $\vec{k}$  mesh of  $12 \times 12 \times 1$ , and trace a  $\Gamma \to K \to M \to \Gamma$  path in the Brillouin-zone.

The interlayer separation h, defined henceforth as the distance measured purely along the  $\hat{z}$  axis between metal atoms in each layer (shown in Fig. 1), has been shown to strongly alter the bilayer tunneling and as such is a key physical parameter [57] that influences the energy competition between

TABLE I. Relaxed bulk and bilayer structural and electronic data summarized in Å and meV. Footnotes a indicate that materials are  $\Gamma$  valleys while footnotes b indicate that materials are K valleys according the LmBJ.

2H [Å, meV]	Bulk PBE	Bulk PBE-SO	Bulk PBE-SO-D3	Bulk LDA-SO	LDA-SO Rel. bulk $h(\frac{c}{2})$	LDA-SO Rel. bilayer $h_r$	LDA+SO K-Γ	Local mBJ K-Γ
$WS_{2}^{1}$ (a) $WS_{2}^{1}$ (c)	3.184223 13.37829	3.182996 13.81513	3.146435 12.08426	3.123324 12.15246	6.076231	6.078068	-245.48	-151.53 <sup>1</sup>
$MoS_{2}^{1}$ (a) $MoS_{2}^{1}$ (c)	3.192238 13.37829	3.182996 13.81513	3.147325 12.09522	3.123039 12.08398	6.041991	6.044029	-446.68	$-309.88^{1}$
$WSe_2^2$ (a) $WSe_2^2$ (c)	3.319933 13.73719	3.314109 14.38835	3.272225 12.75565	3.247144 12.79507	6.383032	6.400151	19.99	85.02 <sup>2</sup>
$MoSe_2^1$ (a) $MoSe_2^1$ (c)	3.32226 13.54296	3.316714 14.39224	3.274179 12.70686	3.247787 12.72364	6.361821	6.364676	-162.43	$-69.83^{1}$
$WTe_2^2$ (a) $WTe_2^2$ (c)	3.561547 14.84723	3.551663 14.95786	3.492066 13.68179	3.471253 13.77909	6.889546	6.893730	129.51	$209.08^2$
$MoTe_2^2$ (a) $MoTe_2^2$ (c)	3.565654 14.64777	3.550858 14.91474	3.490808 13.65810	3.470279 13.73352	6.866760	6.870946	-47.52	24.39 <sup>2</sup>
$MoS_2/WS_2^{-1}$				3.123039		6.049050	-199.32	$-73.35^{1}$
MoSe <sub>2</sub> /WSe <sub>2</sub> <sup>2</sup>				3.247787		6.379695	44.96	142.17 <sup>2</sup>
MoTe <sub>2</sub> /WTe <sub>2</sub> <sup>2</sup>				3.470279		6.879600	130.65	231.21 <sup>2</sup>

 $<sup>^{1}\</sup>Gamma$  valley according to LmBJ.

the band structure and gap [31,32,36,43,58]. While TMD band structures under changing layer separation have been studied before, we find that an explicit and comprehensive study of the K- $\Gamma$  competition in particular is still needed. We therefore vary this parameter for values in the neighborhood of the relaxed interlayer separation,  $h_r$ , and examine the valley competition for each material in the 2H configuration. The total length along  $\hat{z}$  (vacuum + c lattice constant) is kept fixed while the interlayer separation h is varied. The magnitude of the a and b lattice constants are fixed throughout all calculations. This approach keeps the total system + vacuum unit cell volume fixed throughout the calculations, ensuring a consistent total energy.

In addition to interlayer separation, which can be adjusted by applying pressure, applied gate voltages are an experimentally accessible tuning parameter that influences the K- $\Gamma$  competition [33,36–38,59–66]. To finalize our discussion on key factors contributing to K- $\Gamma$  band engineering, we perform electronic structure calculations for these materials under various electric fields.

#### III. RESULTS

We focus on exploring the key factors that determine the location of the valence-band maximum for group VI TMD systems. As shown in Fig. 2, we denote the valence-band energies at  $\Gamma$  and K as  $E_{\Gamma}$  and  $E_{K}$ , respectively, and we employ DFT to calculate the energy difference  $\Delta E_{K-\Gamma} = E_{K} - E_{\Gamma}$  as a function of interlayer distance h. To shed light on the role of electronic correlations in the d orbitals of the transition metal atoms, we compare results obtained using the local mBJ functional with those obtained using the standard LDA functional.

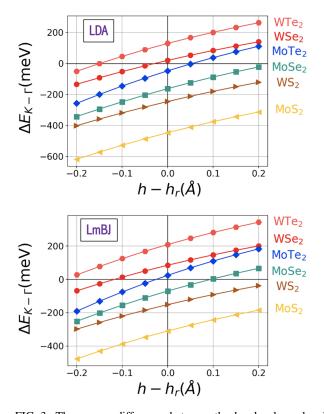


FIG. 3. The energy difference between the local valence-band maxima at  $\Gamma$  and K points,  $\Delta E_{\mathrm{K}-\Gamma} \equiv E_K - E_\Gamma$ , as a function of the interlayer distance h calculated by LDA and local mBJ functionals for 2H stacked homobilayers. The value of h at which  $\Delta E_{\mathrm{K}-\Gamma}$  crosses from negative ( $\Gamma$  valley) to positive (K valley) decreases when correlations are included by using the local mBJ potential. All calculations include spin-orbit coupling.

<sup>&</sup>lt;sup>2</sup>K valley according to LmBJ.

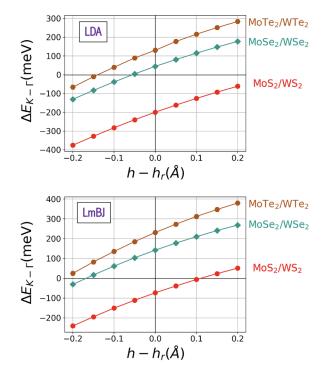


FIG. 4. The energy difference between the local band maxima  $\Gamma$  and K points  $\Delta E \equiv E_K - E_\Gamma$  as a function of the interlayer distance h calculated by LDA and local mBJ functionals for heterobilayers. Analysis of the band characteristics shows that the metal atom at the K valley of the highest valence band is that of tungsten for all heterobilayers.

Figure 3 summarizes our results for 2H stacked  $MX_2$  homobilayers with  $M=\mathrm{Mo}$ , W and  $X=\mathrm{S}$ , Se, Te. We plot the energy difference  $\Delta E_{\mathrm{K}-\Gamma}=E_K-E_\Gamma$  as a function of  $h-h_r$  calculated by LDA (top) and local mBJ (bottom) functionals. Each system's equilibrium interlayer separation  $h_r$  and its  $\Delta E_{\mathrm{K}-\Gamma}=E_K-E_\Gamma$  may be found in Table I. Even though pressure cannot increase h, we include positive values of  $h-h_r$  due to the range of possible separations reported in bulk TMDs. Here it can be observed that the use of the local mBJ potential tends to increase  $\Delta E_{\mathrm{K}-\Gamma}$ . The same behavior is clearly observed in the case of heterobilayers as shown in Fig. 4. Selenide and telluride heterostructures remain K

valley while shifting from equilibrium in nearly all cases. The sulfide heterostructure may achieve a K VBM at the highest separations, while neither sulfide-based homostructure can achieve a K VBM using a physically plausible increase in spacing alone.

After structural relaxation subsequent DFT calculations reveal that the total energy of each AA stacked system is consistently greater ( $\approx$ 70–110 meV per unit cell) than that of its corresponding 2H form, as may be seen in Table II. In this regard, the same overall trend as in the 2H case in the K- $\Gamma$  competition is expected as a function of  $h-h_r$ , but modulating the interlayer separation via pressure is not an experimentally feasible approach due to the predicted instability of the AA structure. The AA structures are only relaxed with LDA-SO, and the  $\Delta E_{K-\Gamma} = E_K - E_\Gamma$  is computed for only the relaxed bilayer separation, as displayed in Table II.

#### IV. DISCUSSION

#### A. Effects of element type

There are several element-related trends that can be seen in our results. To make the analysis concrete, we investigate the role of the element type in the 2H homobilayers only (Fig. 3). First, we consider the trend of varying the metal atom in the  $MX_2$  with the chalcogenide atom X fixed. It can be seen clearly that in this case,  $WX_2$  always has a larger value of  $\Delta E_{K-\Gamma}$  than MoX<sub>2</sub> does. This can be attributed to the fact that the W atom has a larger atomic number than the Mo atom, and therefore larger spin-orbit coupling. To demonstrate this point, we define  $\Delta E_{SO}$  as the energy difference between the top two valence states at the K point as shown in Fig. 2, which can serve as a good estimation of the spin-orbit coupling. Figure 5 (top) plots  $\Delta E_{SO}$  as a function of the absolute interlayer separation h for  $MoX_2$  and  $WX_2$  with X = S, Se and Te. Clearly,  $\Delta E_{SO}$  depends strongly on the metal atom but weakly on the height h and the chalcogenide atom X, which is expected since the spin-orbit coupling is an intrinsic property of the metal

Now we study the trend of varying the chalcogen atom in  $MX_2$  with the metal atom M fixed. We see a monotonic trend where  $\Delta E_{K-\Gamma}$  is generally increased when going from  $S \rightarrow Se \rightarrow Te$ . We note that because chalcogenide atoms X

TABLE II. The K-Γ competition for relaxed AA structures, obtained using just LDA + SO and the local mBJ functionals, in Å, meV for the K-Γ, and Rydberg for  $\Delta E_{AA-2H}$ . The quantity  $\Delta E_{AA-2H} = E_{AA}^{LDA+SO} - E_{2H}^{LDA+SO}$ , where  $E_{AA}^{LDA+SO}$  and  $E_{2H}^{LDA+SO}$  are the total energy in Rydberg for LDA + SO of the AA structure and 2H structures, respectively. The difference is always positive due to the sharp raise in total energy of the AA structures, observed in relaxation.

AA	LDA + SO K-Γ (meV)	Local mBJ K-Γ (meV)	$\Delta E_{AA-2H}$ via LDA + SO (Ry)	h (Å)
$\overline{\mathrm{WS}_2}$	87.4	126.8	5.69	6.776537
$MoS_2$	-86.1	3.0	5.36	6.769552
WSe <sub>2</sub>	368.3	399.1	5.72	7.128487
MoSe <sub>2</sub>	218.2	271.9	6.34	7.100237
WTe <sub>2</sub>	559.3	610.3	8.09	7.769766
MoTe <sub>2</sub>	443.3	483.0	8.19	7.761531
$MoS_2/WS_2$	147.0	215.0	5.63	6.769726
MoSe <sub>2</sub> /WSe <sub>2</sub>	406.2	453.2	6.12	7.111891
$MoTe_2/WTe_2$	588.8	649.9	8.10	7.767940

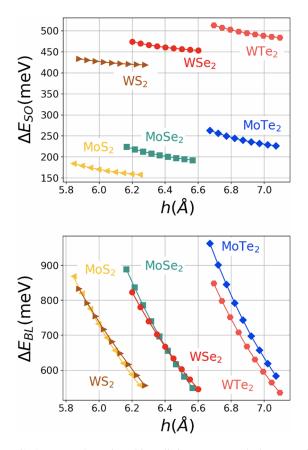


FIG. 5. (top) The spin-orbit splitting  $\Delta E_{SO}$  and (bottom) the bilayer splitting  $\Delta E_{BL}$  as a function of the absolute interlayer separation h for MoX<sub>2</sub> and WX<sub>2</sub> with X = S, Se, and Te. The 2H stacked results using the local mBJ functional are presented.

exist between metal atoms, as shown in Fig. 1 the channels going through the p orbitals of X make the major contribution to the interlayer coupling. We have concluded this by calculating the projected density of states using WIEN2K (not shown), confirming the result in Ref. [34]. Generally speaking, as the atomic number of X increases, orbitals on the outermost shell becomes more delocalized, which leads to a stronger wave function overlap with d orbitals of the metal atom and consequently a larger bilayer splitting at the  $\Gamma$ point. We define  $\Delta E_{BL}$  as the energy difference between the top two valence states at  $\Gamma$  point as shown in Fig. 2, which can be a good estimation of the interlayer coupling.  $\Delta E_{BL}$  as a function of the absolute interlayer separation h for Mo $X_2$  and  $WX_2$  with X = S, Se, and Te is presented in Fig. 5 (bottom). It is clearly observed that bilayer splitting depends strongly on interlayer separation, and that the equilibrium separation of each material monotonically increases as  $S \rightarrow Se \rightarrow Te$ , indicating that stronger bilayer splitting heavily influences the relaxation.

#### **B.** Discussion of structural relaxation

The structural relaxation is performed in VASP, following the procedure outlined in Sec. II A. The resulting structural data are summarized in Table I. For clarity and discussion, we additionally include a plot of the bulk a and c lattice constants

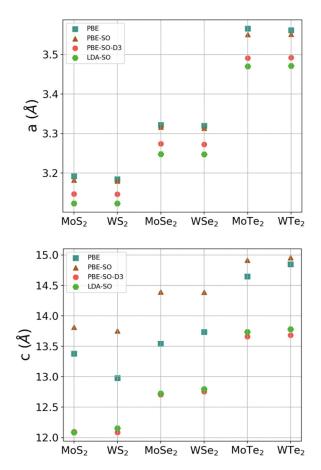


FIG. 6. The (top) a and (bottom) c bulk lattice constants predicted by various potentials employed in VASP.

in Fig. 6. Relaxation has been performed frequently for TMD bilayers, trilayers, and bulk—our results here mostly confirm those previously accepted findings [34,35,43,67]. There are a number conclusions that may be drawn from the relaxation procedure. Inclusion of the vacuum after LDA-SO bulk relaxation can be seen to increase the metal-to-metal atom distance in bilayers, which will reduce interlayer tunneling. The reduction of h in bulk systems is expected [32] since the van der Waals attraction between layers is expected to be stronger when every layer has two neighboring layers. If this consideration is correct, then the encapsulating layers present in most experimental systems may also play a role in determining the equilibrium layer separation. Our DFT calculations confirm this assertion as we see a boost in the K valley while comparing bulk to bilayer results. As we employ both VASP and WIEN2K, we may also comment that the LDA + SO band structures of VASP's pseudopotential method and the all-electron method of WIEN2K are nearly identical if the same structure is provided.1

<sup>&</sup>lt;sup>1</sup>Furthermore, we find that the total energy of the bilayers predicted by LDA + SO using WIEN2K is *in general* at the global minimum when the equilibrium interlayer separation is used. The energy generally grows as we consider the shifting of  $\pm 0.2$  Å, with a sharper growth with positive shifting in interlayer separation.

It is clear from the step pattern in Fig. 6 that the magnitude of the lattice constants is determined mainly by the type of chalcogen in the structure, as predicted in previous works [34,35,43]. This is particularly true for the methods of PBE-SO-D3 and LDA-SO in the case of the c lattice constant. With comparison to the range of literature reported bulk values [67–72], we may confirm that PBE and PBE-SO both tend to overestimate the lattice constants alone, and the inclusion of van der Waals is necessary. As expected [34,35,43,67], even in the absence of the van der Waals force LDA-SO agrees with PBE-SO-D3 quite well for layered systems such as the group VI TMDs, and is in some cases more accurate with respect to experiments. Hence, we employ only LDA-SO for the AA stacking. Our bilayer electronic structure calculations are carried out using constants from the LDA-SO approach, as explained in Sec. II A.

## C. General discussion

The energy competition between  $\Gamma$  and K valleys is influenced by the strengths of interlayer tunneling, spin-orbit coupling, and electronic correlations. Generally speaking, interlayer tunneling produces a more sizable bilayer splitting at the  $\Gamma$  point pushing the top valence state up, with negligible effect at the K point. As a result, a smaller distance between the bilayers favors the valence-band maxima to be at  $\Gamma$  point due to the larger bilayer splitting, as expected [32]. For heterobilayers, the symmetry between layers is broken and the bilayer splitting at the K is not negligible, which explains why the heterobilayers tend to favor the K valley when compared with the homobilayers of their parent atoms. On the other hand, the energy splitting due to spin-orbit coupling is very small near the  $\Gamma$  point but much more pronounced near the K point. Consequently, stronger spin-orbit coupling favors the top valence-band maxima to be at K point. Since varying the distance between bilayers can modulate the interlayer tunneling strength, without influencing the spin-orbit coupling strongly (see Fig. 5), it is an ideal experimental parameter to engineer the  $\Gamma$ -K competition.

Because the highest energy valence bands are composed primarily of d orbitals from the transition metal atoms, electronic correlations also influence the  $\Gamma$ -K energy competition. Given that the wave function is more extended for states near the  $\Gamma$  point and more localized for states near the K point, correlations tend to push states near the K point to higher relative energy. This trend is observed in all of our DFT calculations. One may observe that in molybdenum-based structures, the inclusion of exact electronic correlation via LmBJ produces a larger upward push, due to the 4d orbital characteristic as opposed to the 5d of the stable tungsten-based structures.

With respect to the stacking order, we note that the AA stacked materials possess a much larger metal to metal distance when fully relaxed compared with their 2H counterparts. As a result, we expect that the AA bilayer splitting is smaller than in 2H, contributing to a lower  $\Gamma$  valley. In Table II, we see that all of the equilibrium AA stacked materials are K valleys for LmBJ, due primarily to this weaker interlayer tunneling. As in the 2H cases, the correlation effect of the LmBJ potential is to push the K valley up, but the magnitude of the increase is lesser in AA materials than in 2H.

## V. EXTERNAL TUNING PARAMETERS

## A. Methods to change interlayer distance

Pressure engineering has been widely used as one of the ways to tune an interlayer distance in bilayer and multilayer TMDs [22,31,58,73–77]. Decrease in the interlayer distance has been achieved experimentally by applying hydrostatic pressure through a diamond anvil cell (DAC) [73,75–77]. Pressure used to decrease the interlayer separation can be used to modify the band gap, and in the semiconducting regime the change of interlayer distance estimated from optical probes can be up to 10% within 10 GPa. [75–77]. While we do not attempt to make a quantitative study of the pressure-induced effects, we would like to see if our results are reasonably consistent with experiments. For this purpose, we employ the formula used in Ref. [22] that approximates pressure as

$$P \approx \frac{E_{\text{tot}}(h) - E_{\text{tot}}(h_r)}{A(h_r - h)},\tag{1}$$

where  $E_{\text{tot}}(h)$  is the total energy (from LDA + SO) at interlayer separation h,  $h_r$  is the relaxed interlayer separation, and A is the unit-cell area in the ab plane. For the heterobilayer  $\text{MoSe}_2/\text{WSe}_2$  that has been experimentally studied [75], we find that in order to achieve 3% change of the height, the required pressure is predicted to be  $\approx 1$  GPa. Although the order of magnitude seems to be reasonably within the experimental range, more calculations must be done to have a quantitative description of the pressure effects and more experimental data will be needed for comparison. Applying pressure that decreases the layer separation can only enable transitions from K to  $\Gamma$ —and not the reverse transition.

While applying pressure is an experimentally feasible way to decrease the interlayer distance, the strain engineering offers new possibilities to increase the interlayer distance. By applying tensile or compressive strain to the bilayer TMD, interlayer distance can be either increased or decreased. Transition between direct and indirect band gaps has been reported in experimental and theoretical studies [27,78].

# **B.** Electric gating

The most convenient way of engineering bilayer electronic structure is to apply electrical gate fields [33,36-38,59–64,66]. We employ the "zigzag" approach in WIEN2K [49,79] to apply a simulated gate voltage across the layers of  $\approx$ 0.2V, yielding transverse electric fields with magnitudes of  $\approx 0.01$ –0.1 V/A in the samples. Examples of the resulting band structure change for the experimentally relevant [17,80–83] cases of homobilayer MoTe<sub>2</sub>, and heterobilayer of MoSe<sub>2</sub>/WSe<sub>2</sub> are shown in Fig. 7, and Figs. 8 and 9, respectively. We note that the electric field produces the stronger effect near K than  $\Gamma$  valleys. That the K valley is influenced more than the  $\Gamma$  valley by the external field can be explained by the same model presented in the study of the bilayer graphene [84]. In essence, since the external electric field produces an electric potential difference between layers, its effect can be described by a two-band model capturing the layer degrees of freedom in Eq. (5) of Ref. [84]. While this explanation is qualitative, the self-consistency calculations take this model into account quantitatively. First, in the two-band

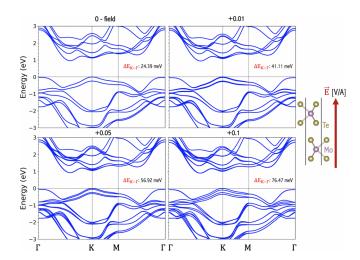


FIG. 7. The electric field effect on the band structure of 2H MoTe<sub>2</sub> bilayers. For homobilayers the direction of the field is not relevant. Starting from zero applied field at the top left, the magnitude of the electric field is increased yielding an enhanced splitting near the K valley, and in between the  $\Gamma$  and K valleys, without strongly affecting the  $\Gamma$  valley.

model, the electric potential difference appears in the diagonal terms, whose effect in the eigenenergies will be reduced in general in the presence of the off-diagonal terms due to the interlayer tunneling. As a result, the shift of the eigenenergies is expected to be larger at momentum where tunneling is small. This explains our observation of the external electrical field having stronger effect near K than  $\Gamma$  valleys. We note that the evolution of the  $\Delta E_{K-\Gamma}$  is further complicated by the screening effect due to the Hartree potential that is shown to be large and well-captured by DFT calculations. In general, the screening effect decreases the electric potential difference actually seen in the bilayer systems in a nonmonotonic way.

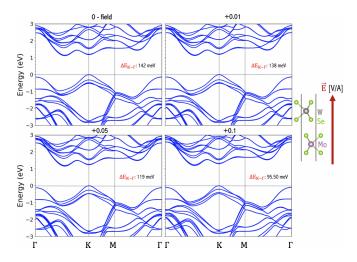


FIG. 8. The electric field effect on the band structure of 2H  $MoSe_2/WSe_2$  bilayers. Starting from zero applied field at the top left, the field pointing from the bottom layer of  $MoSe_2$  to the top layer of  $WSe_2$  is increased in magnitude. For heterobilayers, we find that the direction is important (see text).

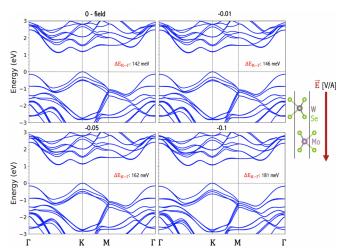


FIG. 9. The electric field effect on the band structure of 2H MoSe<sub>2</sub>/WSe<sub>2</sub> bilayers. Starting from zero applied field at the top left, the field pointing from the top layer of WSe<sub>2</sub> to the bottom layer of MoSe<sub>2</sub> is increased in magnitude. For heterobilayers, we find that the direction is important (see text).

Another intriguing observation is that for homobilayers (Fig. 7) direction of the field does not matter as both layers maintain the same on-site potential in each layer and have a mirror symmetry between layers, and the field makes a positive contribution to  $\Delta E_{K-\Gamma}$ . For heterobilayers (Figs. 8 and 9), we observe that changing the field direction produces opposite effects. If the field points from MoSe<sub>2</sub> to WSe<sub>2</sub>, the top valence band (which is dominated by d orbitals of W atoms) is pushed away from the Fermi energy and the gap is enlarged resulting in a reduction in  $\Delta E_{K-\Gamma}$ . If the field points from WSe<sub>2</sub> to MoSe<sub>2</sub>, the top valence band is pushed closer to the Fermi energy and the gap is reduced. This can also been explained by the two-band model [84]. Because in the heterobilayer the on-site energy is already different between layers, a diagonal term in the two band model breaking the layer symmetry is already present at zero electric field. Therefore, the external field can drive K closer to or further away from the VBM in heterobilayers, depending on whether the direction of the electric field aligns with the chemical potential difference. Accordingly, the gate voltage is a powerful tuning parameter in TMD systems.

#### VI. CONCLUSION

In this paper, we have explored fundamental ingredients that influence the competition between valleys within the valence bands of the experimentally promising group VI transition metal dichalcogenide systems. Our results show that several physical ingredients, namely, the interlayer tunneling, spin-orbital coupling, gate voltage, and the electron correlation between the d orbitals of the metal atoms, all play a significant role in the calculated energy at the K and  $\Gamma$  points, particularly at the top of the valence band.

To study these physical processes, we first completed comprehensive structural relaxation for 2H and AA stacking configurations using VASP. Our results confirm that for the PBE functional, both spin orbit and van der Waals corrections (SO-D3) are necessary to ensure that lattice constants are not overestimated, which would in turn misrepresent the bilayer splitting at  $\Gamma$ . We find that the local density approximation with spin-orbit correction agrees well with the full PBE-SO-D3, which provides a computationally practical alternative for structural calculations in group VI TMDs. Our findings lead us to the conclusion that the magnitude of lattice constants is primarily determined by the chalcogen in the system. We also report that for the LDA + SO band structure, the pseudopotential and all-electron approaches of VASP and WIEN2K, respectively, strongly agree.

Using density function theory, we performed a systematic study of the interlayer splitting as a function of layer separation, and found that the energy at the K point can be driven higher as a result of increasing separation, while conversely, the energy at  $\Gamma$  is decreased. Furthermore, we applied an external field and observed an increase in the K valley energy in homostructures, and a direction-dependent movement of the same valley in heterostructures. Our DFT calculations employed both the standard LDA function and the Local mBJ functional. As the d orbitals of the transition metals from the parent layers are known to be dominant contributors to the top valence bands, including the electronic correlation effect from the mBJ potential is physically warranted. We find that the inclusion of the electronic correlation further pushes the K point nearer to the Fermi energy, with a negligible effect on the  $\Gamma$ point, consistent with the idea that the wave function is more localized at K. Additionally, we find that the spin degeneracy

at K is lifted with the inclusion of the spin-orbit coupling, further separating the top valence bands. At the same time, this inclusion does not drastically change the energy at the  $\Gamma$  point.

Our findings show that applying gate voltage and tuning of the interlayer separation are powerful tuning knobs for topological and band engineering. For example, K and  $\Gamma$  valleys may be tuned to be nearly degenerate within realistic range for experiments, which can lead to an effective two-dimensional strongly correlated two-orbital material after an induced moiré twist. The inclusion of the electronic correlation causes this K-point energy to raise even higher, making a crossover separation more achievable. Increasing the separation drives the K point higher *without* affecting the degeneracy-lifting spin-orbit interaction, meaning that our findings outline an experimentally feasible approach for TMD engineering.

## **ACKNOWLEDGMENTS**

S.O., E.J., and W.-C.L. were supported by the Air Force Office of Scientific Research Multi-Disciplinary Research Initiative (MURI) entitled, Cross-disciplinary Electronic-ionic Research Enabling Biologically Realistic Autonomous Learning (CEREBRAL) under Award No. FA9550-18-1-0024 administered by Dr. Ali Sayir. A.H.M. was supported by U.S. Department of Energy, Office of Basic Energy Sciences (DESC0021984).

- [1] Y. Cao, V. Fatemi, A. Demir, S. Fang, S. L. Tomarken, J. Y. Luo, J. D. Sanchez-Yamagishi, K. Watanabe, T. Taniguchi, E. Kaxiras, R. C. Ashoori, and P. Jarillo-Herrero, Nature (London) 556, 80 (2018).
- [2] Y. Cao, V. Fatemi, S. Fang, K. Watanabe, T. Taniguchi, E. Kaxiras, and P. Jarillo-Herrero, Nature (London) 556, 43 (2018).
- [3] X. Lu, P. Stepanov, W. Yang, M. Xie, M. A. Aamir, I. Das, C. Urgell, K. Watanabe, T. Taniguchi, G. Zhang, A. Bachtold, A. H. MacDonald, and D. K. Efetov, Nature (London) 574, 653 (2019).
- [4] F. Wu, T. Lovorn, E. Tutuc, and A. H. MacDonald, Phys. Rev. Lett. 121, 026402 (2018).
- [5] F. Wu, A. H. MacDonald, and I. Martin, Phys. Rev. Lett. 121, 257001 (2018).
- [6] H. C. Po, L. Zou, A. Vishwanath, and T. Senthil, Phys. Rev. X 8, 031089 (2018).
- [7] R. Bistritzer and A. H. MacDonald, Proc. Natl. Acad. Sci. USA 108, 12233 (2011).
- [8] E. J. Mele, Phys. Rev. B 81, 161405(R) (2010).
- [9] B. L. Chittari, G. Chen, Y. Zhang, F. Wang, and J. Jung, Phys. Rev. Lett. 122, 016401 (2019).
- [10] M. H. Naik and M. Jain, Phys. Rev. Lett. **121**, 266401 (2018).
- [11] L. Wang, E.-M. Shih, A. Ghiotto, L. Xian, D. A. Rhodes, C. Tan, M. Claassen, D. M. Kennes, Y. Bai, B. Kim, K. Watanabe, T. Taniguchi, X. Zhu, J. Hone, A. Rubio, A. N. Pasupathy, and C. R. Dean, Nat. Mater. 19, 861 (2020).

- [12] E. C. Regan, D. Wang, C. Jin, M. I. B. Utama, B. Gao, X. Wei, S. Zhao, W. Zhao, Z. Zhang, K. Yumigeta, M. Blei, J. D. Carlstrm, K. Watanabe, T. Taniguchi, S. Tongay, M. Crommie, A. Zettl, and F. Wang, Nature (London) 579, 359 (2020).
- [13] C. Jin, E. C. Regan, A. Yan, M. I. B. Utama, D. Wang, S. Zhao, Y. Qin, S. Yang, Z. Zheng, S. Shi, K. Watanabe, T. Taniguchi, S. Tongay, A. Zettl, and F. Wang, Nature (London) 567, 76 (2019)
- [14] H. Zeng and X. Cui, Chem. Soc. Rev. 44, 2629 (2015).
- [15] W. Li, C. F. J. Walther, A. Kuc, and T. Heine, J. Chem. Theory Comput. 9, 2950 (2013).
- [16] D. Xiao, G.-B. Liu, W. Feng, X. Xu, and W. Yao, Phys. Rev. Lett. 108, 196802 (2012).
- [17] K. F. Mak, D. Xiao, and J. Shan, Nat. Photonics 12, 451 (2018).
- [18] M. Angeli and A. H. MacDonald, Proc. Natl. Acad. Sci. USA 118, e2021826118 (2021).
- [19] Y. Sun, D. Wang, and Z. Shuai, J. Phys. Chem. C 120, 21866 (2016).
- [20] R. Lv, J. A. Robinson, R. E. Schaak, D. Sun, Y. Sun, T. E. Mallouk, and M. Terrones, Acc. Chem. Res. 48, 56 (2015).
- [21] R. Roldán, J. A. Silva-Guillén, M. P. López-Sancho, F. Guinea, E. Cappelluti, and P. Ordejón, Ann. Phys. (Berlin, Ger.) 526, 347 (2014).
- [22] S. Bhattacharyya and A. K. Singh, Phys. Rev. B 86, 075454 (2012).
- [23] T. Li and G. Galli, J. Phys. Chem. C 111, 16192 (2007).
- [24] S. Tongay, J. Zhou, C. Ataca, K. Lo, T. S. Matthews, J. Li, J. C. Grossman, and J. Wu, Nano Lett. 12, 5576 (2012).

- [25] S. K. Pandey, R. Das, and P. Mahadevan, ACS Omega 5, 15169 (2020).
- [26] M. Koperski, M. R. Molas, A. Arora, K. Nogajewski, A. O. Slobodeniuk, C. Faugeras, and M. Potemski, Nanophotonics 6, 1289 (2017).
- [27] A. Kumar and P. K. Ahluwalia, Modell. Simul. Mater. Sci. Eng. 21, 065015 (2013).
- [28] K. P. Dhakal, S. Roy, H. Jang, X. Chen, W. S. Yun, H. Kim, J. Lee, J. Kim, and J.-H. Ahn, Chem. Mater. 29, 5124 (2017).
- [29] D. Pei, B. Wang, Z. Zhou, Z. He, L. An, S. He, C. Chen, Y. Li, L. Wei, A. Liang, J. Avila, P. Dudin, V. Kandyba, A. Giampietri, M. Cattelan, A. Barinov, Z. Liu, J. Liu, H. Weng, N. Wang, Phys. Rev. X 12, 021065 (2022).
- [30] D. Kozawa, R. Kumar, A. Carvalho, K. K. Amara, W. Zhao, S. Wang, M. Toh, R. M. Ribeiro, A. H. C. Neto, K. Matsuda, and G. Eda, Nat. Commun. 5, 4543 (2014).
- [31] P. Ci, Y. Chen, J. Kang, R. Suzuki, H. S. Choe, J. Suh, C. Ko, T. Park, K. Shen, Y. Iwasa, S. Tongay, J. W. Ager, L.-W. Wang, and J. Wu, Nano Lett. 17, 4982 (2017).
- [32] L. Zhang and A. Zunger, Nano Lett. 15, 949 (2015).
- [33] J. Xiao, M. Long, X. Li, Q. Zhang, H. Xu, and K. S. Chan, J. Phys.: Condens. Matter 26, 405302 (2014).
- [34] H.-P. Komsa and A. V. Krasheninnikov, Phys. Rev. B 88, 085318 (2013).
- [35] C. Lechner, B. Pannier, P. Baranek, N. C. Forero-Martinez, and H. Vach, J. Phys. Chem. C 120, 5083 (2016).
- [36] N. A. Lanzillo, T. P. O'Regan, and S. K. Nayak, Comput. Mater. Sci. **112**, 377 (2016).
- [37] N. Lu, H. Guo, L. Li, J. Dai, L. Wang, W.-N. Mei, X. Wu, and X. C. Zeng, Nanoscale **6**, 2879 (2014).
- [38] W. Li, T. Wang, X. Dai, X. Wang, C. Zhai, Y. Ma, S. Chang, and Y. Tang, Solid State Commun. 250, 9 (2017).
- [39] T. Rauch, M. A. L. Marques, and S. Botti, J. Chem. Theory Comput. 16, 2654 (2020).
- [40] T. Rauch, M. A. L. Marques, and S. Botti, Phys. Rev. B 101, 245163 (2020).
- [41] T. Devakul, V. Crépel, Y. Zhang, and L. Fu, Nat. Commun. 12, 6730 (2021).
- [42] X. Li, X. Shi, D. Marian, D. Soriano, T. Cusati, G. Iannaccone, G. Fiori, Q. Guo, W. Zhao, and Y. Wu, Sci. Adv. 9, eade5706 (2023).
- [43] J. He, K. Hummer, and C. Franchini, Phys. Rev. B 89, 075409 (2014).
- [44] G. Kresse and J. Hafner, Phys. Rev. B 47, 558 (1993).
- [45] G. Kresse and J. Furthmüller, Comput. Mater. Sci. 6, 15 (1996).
- [46] G. Kresse and J. Furthmüller, Phys. Rev. B 54, 11169 (1996).
- [47] A. Jain, S. P. Ong, G. Hautier, W. Chen, W. D. Richards, S. Dacek, S. Cholia, D. Gunter, D. Skinner, G. Ceder *et al.*, APL Mater. 1, 011002 (2013).
- [48] P. Blaha, K. Schwarz, G. Madsen, D. Kvasnicka, J. Luitz, R. Laskowsk, F. Tran, L. Marks, and L. Marks, WIEN2k: An Augmented Plane Wave Plus Local Orbitals Program for Calculating Crystal Properties (Techn. Universitat, 2019).
- [49] K. Schwarz and P. Blaha, Comput. Mater. Sci. 28, 259 (2003).
- [50] A. D. Becke and E. R. Johnson, J. Chem. Phys. 124, 221101 (2006).
- [51] F. Tran and P. Blaha, Phys. Rev. Lett. **102**, 226401 (2009).
- [52] Z. Zhu and U. Schwingenschlögl, Phys. Rev. B 86, 075149 (2012).

- [53] M. J. Wahila, G. Paez, C. N. Singh, A. Regoutz, S. Sallis, M. J. Zuba, J. Rana, M. B. Tellekamp, J. E. Boschker, T. Markurt, J. E. N. Swallow, L. A. H. Jones, T. D. Veal, W. Yang, T.-L. Lee, F. Rodolakis, J. T. Sadowski, D. Prendergast, W.-C. Lee, W. A Doolittle, Phys. Rev. Mater. 3, 074602 (2019).
- [54] P. Borlido, T. Aull, A. W. Huran, F. Tran, M. A. L. Marques, and S. Botti, J. Chem. Theory Comput. 15, 5069 (2019).
- [55] G. J. Paez, C. N. Singh, M. J. Wahila, K. E. Tirpak, N. F. Quackenbush, S. Sallis, H. Paik, Y. Liang, D. G. Schlom, T.-L. Lee, C. Schlueter, W.-C. Lee, and L. F.J. Piper, Phys. Rev. Lett. 124, 196402 (2020).
- [56] C. N. Singh, L. F. J. Piper, H. Paik, D. G. Schlom, and W.-C. Lee, Phys. Rev. B 105, 035150 (2022).
- [57] J. Xu, J. Zhang, W. Zhang, and C.-S. Lee, Adv. Energy Mater. 7, 1700571 (2017).
- [58] X. Fan, C.-H. Chang, W. T. Zheng, J.-L. Kuo, and D. J. Singh, J. Phys. Chem. C 119, 10189 (2015).
- [59] Q. Liu, L. Li, Y. Li, Z. Gao, Z. Chen, and J. Lu, J. Phys. Chem. C 116, 21556 (2012).
- [60] A. Ramasubramaniam, D. Naveh, and E. Towe, Phys. Rev. B 84, 205325 (2011).
- [61] D. Huang and E. Kaxiras, Phys. Rev. B 94, 241303(R) (2016).
- [62] E. J. G. Santos and E. Kaxiras, ACS Nano 7, 10741 (2013).
- [63] M. Sharma, A. Kumar, P. K. Ahluwalia, and R. Pandey, J. Appl. Phys. 116, 063711 (2014).
- [64] Z. Zheng, X. Wang, and W. Mi, J. Phys.: Condens. Matter 28, 505003 (2016).
- [65] F. Xu, Z. Sun, T. Jia, C. Liu, C. Xu, C. Li, Y. Gu, K. Watanabe, T. Taniguchi, B. Tong, J. Jia, Z. Shi, S. Jiang, Y. Zhang, X. Liu, and T. Li, Phys. Rev. X 13, 031037 (2023).
- [66] X. Dai, W. Li, T. Wang, X. Wang, and C. Zhai, J. Appl. Phys. 117, 084310 (2015).
- [67] A. Molina-Sánchez and L. Wirtz, Phys. Rev. B 84, 155413 (2011).
- [68] D. Puotinen and R. E. Newnham, Acta Crystallogr. 14, 691 (1961).
- [69] R. Coehoorn, C. Haas, J. Dijkstra, C. J. F. Flipse, R. A. de Groot, and A. Wold, Phys. Rev. B 35, 6195 (1987).
- [70] N. Wakabayashi, H. G. Smith, and R. M. Nicklow, Phys. Rev. B 12, 659 (1975).
- [71] W. Schutte, J. De Boer, and F. Jellinek, J. Solid State Chem. 70, 207 (1987).
- [72] N. V. Podberezskaya, S. A. Magarill, N. V. Pervukhina, and S. V. Borisov, J. Struct. Chem. 42, 654 (2001).
- [73] X. Dou, K. Ding, D. Jiang, and B. Sun, ACS Nano 8, 7458 (2014).
- [74] X. Su, R. Zhang, C. Guo, J. Zheng, and Z. Ren, Phys. Lett. A 378, 745 (2014).
- [75] J. Xia, J. Yan, Z. Wang, Y. He, Y. Gong, W. Chen, T. C. Sum, Z. Liu, P. M. Ajayan, and Z. Shen, Nat. Phys. 17, 92 (2021).
- [76] A. P. Nayak, S. Bhattacharyya, J. Zhu, J. Liu, X. Wu, T. Pandey, C. Jin, A. K. Singh, D. Akinwande, and J.-F. Lin, Nat. Commun. 5, 3731 (2014).
- [77] A. P. Nayak, Z. Yuan, B. Cao, J. Liu, J. Wu, S. T. Moran, T. Li, D. Akinwande, C. Jin, and J.-F. Lin, ACS Nano 9, 9117 (2015).
- [78] S. Pak, J. Lee, Y.-W. Lee, A.-R. Jang, S. Ahn, K. Y. Ma, Y. Cho, J. Hong, S. Lee, H. Y. Jeong, H. Im, H. S. Shin, S. M. Morris, S. Cha, J. I. Sohn, and J. M. Kim, Nano Lett. 17, 5634 (2017).

- [79] P. Blaha, K. Schwarz, F. Tran, R. Laskowski, G. K. H. Madsen, and L. D. Marks, J. Chem. Phys. 152, 074101 (2020).
- [80] J. Cai, E. Anderson, C. Wang, X. Zhang, X. Liu, W. Holtzmann, Y. Zhang, F. Fan, T. Taniguchi, K. Watanabe *et al.*, Nature (London) **622**, 63 (2023).
- [81] Y. Zeng, Z. Xia, K. Kang, J. Zhu, P. Knüppel, C. Vaswani, K. Watanabe, T. Taniguchi, K. F. Mak, and J. Shan, Nature (London) 622, 69 (2023).
- [82] C. Jin, Z. Tao, T. Li, Y. Xu, Y. Tang, J. Zhu, S. Liu, K. Watanabe, T. Taniguchi, J. C. Hone, L. Fu, J. Shan, and K. F. Mak, Nat. Mater. 20, 940 (2021).
- [83] P. Rivera, K. L. Seyler, H. Yu, J. R. Schaibley, J. Yan, D. G. Mandrus, W. Yao, and X. Xu, Science 351, 688 (2016).
- [84] H. Min, B. Sahu, S. K. Banerjee, and A. H. MacDonald, Phys. Rev. B **75**, 155115 (2007).