

Observation of orbital selective charge transfer in a Fe/BaTiO₃ interfacial two-dimensional electron gas

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Two-dimensional electron gas (2DEG) states at oxide interfaces between two ferroic materials have been fertile ground to realize controllable multiferroicity. Here, we investigate the 2DEG states at the interface of ferroelectric BaTiO₃ and a magnetic layer of iron using angle-resolved photoemission spectroscopy. Orbital-selective charge transfer occurs on the surprisingly robust 2DEG. Based on first-principles calculations, we show how the interfacial hybridization can give rise to the unexpected charge transfer in the magnetic 2DEG. Our study reveals a close interplay on a 2DEG between magnetic and ferroelectric interfaces, which sheds light on future design principles of multiferroic 2DEG states.

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Realizing spin electronics (spintronics) by controlling spin and charge degrees of freedom has been intensively studied for decades [1,2]. Multiferroicity, where two or more ferroic orders such as ferroelectricity, ferromagnetism, and ferroelasticity are exhibited simultaneously, has been of keen interest as one of the candidates for promising spintronic devices [3,4]. In particular, magnetoelectric multiferroic materials where ferroelectricity and ferromagnetism coexist have been of great interest for future device applications due to the possibility of reduced power for switching the device state [3–5]. However, due to the contradictory requirement of multiferroicity in *d*-band transition metal oxides—that is, ferroelectricity is favored by *d*⁰ transition metal ions, whereas ferromagnetism is preferred in partially filled *d* (or *f*) shells [6,7]—relatively few single-phase multiferroic materials have been reported [8–10]. Moreover, the empty *d* orbital state makes multiferroic materials prefer insulating ground states, which makes it challenging to utilize multiferroicity for current-controlled electronic devices.

One solution is to design heterogeneous interfaces consisting of both ferromagnetic and ferroelectric materials to realize interfacial multiferroicity [11–15]. In addition, several ferroelectric/metallic-ferromagnetic oxide interfaces such as BaTiO₃/Fe [16–18] and (La,Sr)MnO₃/BaTiO₃ [19,20] have been proposed to address the lack of electrical conductivity in multiferroic systems. These heterostructures are often

susceptible to forming two-dimensional electron gas (2DEG) states, which can have high carrier mobility [21–23], can be tuned under electric/magnetic fields [5,24], and can operate sustainably in atmospheric conditions [25]. As a result, many spintronic devices have been conceived based on oxide heterostructures with ferroic 2DEG states [5,26,27], including a recent report on Ca-SrTiO₃/EuTiO₃/LaAlO₃ heterostructures [28].

However, in magnetic metal/ferroelectric oxide heterostructures, interfacial charge transfer is an essential variable that can substantially alter the original magnetic and ferroelectric orders. Of particular importance, interfacial oxygen diffusion effects have not been well studied in this context, where the metallic layer tends to extract oxygen atoms from the ferroelectric side. It remains unclear how the charge transfer and bonding across such interfaces will occur across the interface. Therefore, investigating directly the electronic and magnetic structures of such interfaces is crucial [29–31] to understand exotic interface-driven phenomena such as multiferroic 2DEG [28]. Especially for tunnel devices, the detailed electronic structure plays a determinant role in the tunnel probability and the macroscopic behavior of the tunnel process through the Landauer-Buttiker formula [19]. Here, we measure these electronic structure details by fabricating a monolayer-thick Fe magnetic layer on oxide ferroelectric thin films of BaTiO₃ (BTO) [Fig. 1(a)] and perform *in situ* investigations of the electronic structure of the 2DEG with angle-resolved photoemission spectroscopy (ARPES). Combining experimental and theoretical investigations, we report a nontrivial charge transfer in the 2DEG due to atomic bonding on the interface. Our study clarifies how the electronic structures are influenced by the interfacial interactions at ferroic oxide interfaces.

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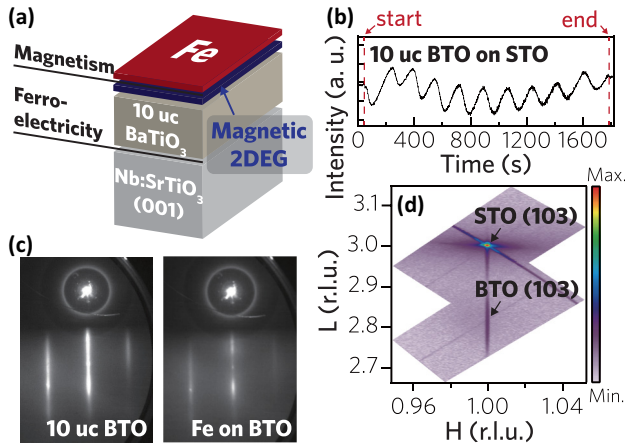


FIG. 1. Structure of interface-induced magnetic two-dimensional electron gas (2DEG). (a) A schematic of the interface between a magnetic layer of Fe and a ferroelectric layer of 10-unit-cell (uc) BaTiO₃ (BTO). (b) Reflection high-energy electron diffraction (RHEED) intensity plot during the growth of a 10-uc BTO film. (c) RHEED patterns along the [11]_{pc} direction before and after depositing Fe atoms on a 10-uc BTO film. (d) Reciprocal space mapping of a 10-uc BTO film.

To study the 2DEG states between magnetic and ferroelectric layers, two thin films have been prepared: 10-unit-cell (uc) BTO thin films on Nb-doped SrTiO₃ (STO) (001) are prepared, and capped with 1.74 monolayer (ML) of Fe deposited at 60 °C in vacuum (1 ML = 6.578×10^{14} atoms/cm² = 1 atom/(3.905 \AA)², the area of a SrTiO₃ surface mesh). Hereinafter, we denote the two types of samples as BTO and Fe/BTO. Figure 1(b) shows a reflection high-energy electron diffraction (RHEED) intensity plot for the BTO film. RHEED patterns are observed during the growth and remain unchanged after Fe atoms are deposited [Fig. 1(c)]. The observed diffraction is consistent with the epitaxial growth of Fe₂O₂ on the BTO with an orientation and structure shown in Fig. 4(a), used for the first principles modeling, and observed in scanning transmission electron microscopy for thicker Fe/BTO interfaces prepared in a manner similar to that used here [32]. Figure 1(d) shows reciprocal space mapping measured on the BTO thin film. The BTO film is fully strained to the STO substrate below it, and the difference in the out-of-plane lattice constants is 6.3% ($c = 4.151 \text{ \AA}$) and consistent with elastically strained bulk tetragonal BTO, which gives 4.13 Å (See Sec. V of the Supplemental Material for discussion [33]). The c -axis lattice constant of our film is almost the same as that of the reported 7-nm BTO film on an STO substrate, $c_{\text{BTO}} = 4.16 \text{ \AA}$ [52], larger than for BTO/STO superlattices grown on STO $c_{\text{BTO}} = 4.04 \text{ \AA}$ [53], and smaller than a SrRuO₃/BaTiO₃/SrRuO₃ structure where $c_{\text{BTO}} = 4.21 \text{ \AA}$ [54].

To measure the electronic structure of the 2DEG, we perform ARPES on the BTO and Fe/BTO films and compare the resulting spectra [Figs. 2(a) and 2(b)]. Figure 2(c) shows a schematic of the Fermi surface of the ultraviolet light (UV) induced 2DEG states. Note that the schematic of the Fermi surface has been examined by conducting ARPES with different photon polarizations (see Fig. S1 for the discussion about the schematic [33]). An electron pocket marked with

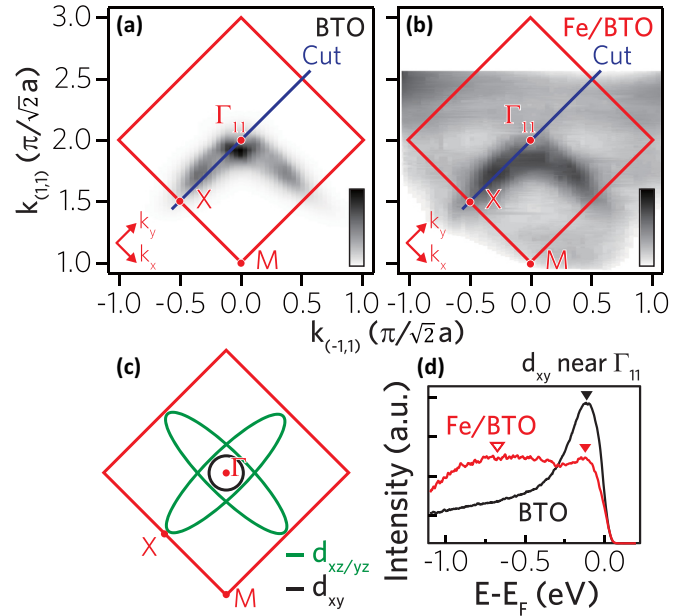


FIG. 2. Robust 2DEG prior to and after Fe deposition. (a), (b) Fermi surfaces of 2DEG in BTO and Fe/BTO films. Red squares represent the Brillouin zone boundary. (c) Schematic of the Fermi surface of the 2DEG. A black (green) solid line indicates a d_{xy} ($d_{xz/yz}$) band. (d) Energy distribution curves of 2DEG in BTO (black) and Fe/BTO (red) films at $(k_{(-1,1)}, k_{(1,1)}) = (0, 1.9)$. Filled inverted triangles indicate coherent peaks near E_F . After Fe deposition, incoherent spectral weight appears around $E = E_F - 0.7 \text{ eV}$ (open inverted triangle).

a black solid line near the Γ point consists of Ti $3d_{xy}$ orbitals, whereas two ellipsoidal electron pockets elongated along the Γ - X direction (marked with a green solid line) consist mainly of Ti $3d_{xz}$ and $3d_{yz}$ orbitals [51]. For compactness, we denote Ti $3d_{xy}$, $3d_{yz}$, and $3d_{xz}$ orbitals as d_{xy} , d_{yz} , and d_{xz} , respectively. Figures 2(a) and 2(b) show the measured Fermi surfaces of the 2DEG states for BTO and Fe/BTO. Both bare BTO and Fe/BTO surfaces are exposed to UV irradiation of 150-eV photons for more than an hour until the pocket sizes stabilize, which is due to the formation of a sufficient number of oxygen vacancies under UV irradiation [55].

Figure 2(d) shows energy distribution curves (EDCs) of the d_{xy} band near the Γ_{11} point. We observe coherent quasiparticle peaks in both samples indicated by filled inverted triangles. In Fe/BTO, an incoherent spectral hump appears in the high-binding energy region ($-1 \text{ eV} < E - E_F < -0.3 \text{ eV}$, where E_F is the Fermi level), highlighted by an open inverted triangle. Since the coherent quasiparticle peaks remain sharply defined in momentum space, the incoherent spectrum is unlikely from simple elastic impurity scattering of the 2DEG electrons [56]. We have found that Fe is oxidized in Fe/BTO by performing core-level photoemission spectroscopy (PES) [see Fig. S2(a) [33]].

To compare the band structures of the two systems, we plot in Figs. 3(a) and 3(b) energy-momentum (E - k) slices of the spectra along the Γ_{11} - X direction [solid blue lines in Figs. 2(a) and 2(b)]. Two electronlike pockets (d_{xy} and d_{xz}) are observed in both E - k cuts. Momentum distribution

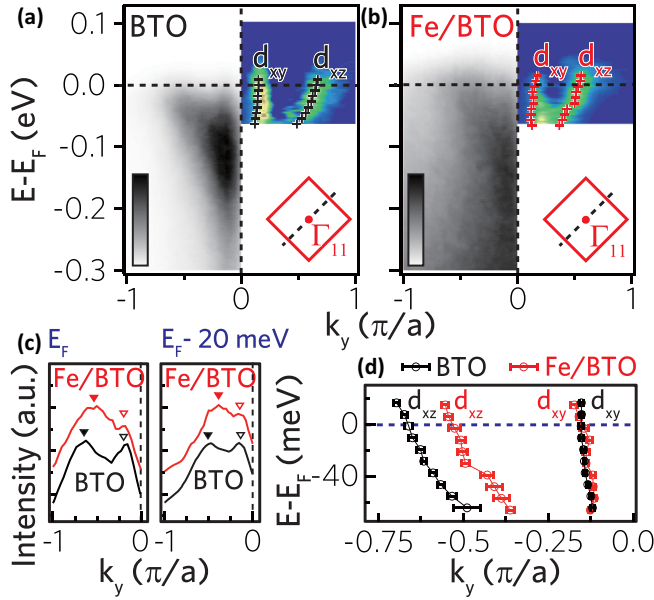


FIG. 3. Fe-induced, orbital-selective charge transfer at the interface. (a), (b) E - k dispersion of BTO and Fe/BTO films along the k_y direction. Here, we define k_y at Γ_{11} as zero. 2D curvature band dispersions [60] and fit results with a Lorentzian function are plotted on right. (c) Momentum distribution curves (MDCs) at the Fermi level (E_F) and $E_F - 20$ meV. Open (closed) inverted triangles represent peaks for the d_{xy} (d_{xz}) bands. (d) Peak positions of d_{xz} and d_{xy} bands of BTO and Fe/BTO films. Black (red) open circles represent data from BTO (Fe/BTO) film.

curves (MDCs) at E_F [Fig. 3(c)] show two peaks in each cut, where the peaks from the d_{xz} (d_{xy}) orbital are marked with closed (open) inverted triangles. Fitted 2DEG band dispersions before and after Fe deposition are shown in the insets of Figs. 3(a) and 3(b). To compare the difference in detail, Fig. 3(d) shows the peak positions of the d_{xz} and d_{xy} bands versus momentum. The peak positions of the d_{xz} band change with Fe deposition, and its Fermi momentum k_F changes from -0.53 to -0.42 \AA^{-1} , corresponding to an estimated change in carrier density from 8.3×10^{13} to $6.6 \times 10^{13} \text{ cm}^{-2}$. On the other hand, the peak positions of the d_{xy} orbital remain at the same position with $k_F \pm 0.02 \text{ \AA}^{-1}$ corresponding to a carrier density of $7.7 \times 10^{12} \text{ cm}^{-2}$. Similar analysis along the Γ_{11} - M direction also shows little change in the d_{xy} pocket size (Fig. S3 [33]). This defies the natural expectation of Fe $4s$ electrons being completely ionized and filling the low-energy Ti $3d$ bands near E_F . As a result, the Fe on the ferroelectric surface induces an unconventional orbital-selective charge transfer in the 2DEG state.

Direct structural characterization of such a monolayer metal-oxide interface is challenging given the necessity to maintain an *in situ* environment to prevent extrinsic oxidation, especially when possible interfacial oxygen diffusion is part of the investigation. To decipher the property of the interface, we combine *in situ* surface electronic and structural characterizations with *ab initio* calculations of a comprehensive set of possible surface configurations. First, the RHEED patterns of the Fe/BTO film suggest that Fe grows as a flat single-crystalline monolayer on the BTO surface [Fig. 1(c)].

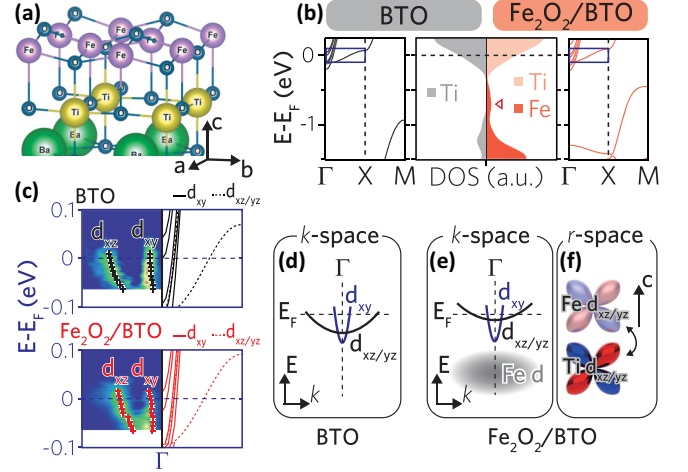


FIG. 4. *Ab initio* results for the $\text{Fe}_2\text{O}_2/\text{BTO}$ interface. (a) Geometry of the monolayer $\text{Fe}_2\text{O}_2/\text{BTO}$ heterostructure. (b) *Ab initio* calculated band structures and density of states (DOS) of (left) BTO and (right) $\text{Fe}_2\text{O}_2/\text{BTO}$ films. Atom-projected DOS of the BTO ($\text{Fe}_2\text{O}_2/\text{BTO}$) film is plotted at the center with black (red) depending on the type of atom. A red open triangle is placed at $E = E_F - 0.7$ eV. (c) (left) Enlarged 2D curvature band dispersions with fit results [adapted from Figs. 3(a) and 3(b)] and (right) calculated E - k dispersions for bands with mainly Ti $3d$ orbital character. The enlarged regions are represented with blue rectangles in (b). (d)–(f) Schematic band structures and atomic orbital interaction in (d) BTO and (e),(f) $\text{Fe}_2\text{O}_2/\text{BTO}$ films. The double-ended arrows represent electron hopping between orbitals.

The iron flux administered during the *in situ* synthesis corresponds to approximately two irons per primitive BTO surface unit cell (see the Methods section in the Supplemental Material [33]). Meanwhile, core-level PES results [Fig. S2(a) [33]] suggest oxidized Fe with a predominately and Fe^{2+} oxidation state, consistent with Fe_2O_2 (see Sec. IV in the Supplemental Material [33]) [57]. To further understand the interfacial composition and structure, it is useful to perform *ab initio* calculations on a large number of surface structures where we consider either one or two Fe atoms per surface unit cell along with zero, one, or two added oxygen atoms. Multiple initial guesses for the initial starting geometry are used for structural relaxations, and a range of Fe $3d$ and Ti $3d$ U values (within DFT + U theory) are used along with ferromagnetic and antiferromagnetic Fe magnetic moment alignments.

After identifying the ground state lattice structure for each composition, we compute the electronic band structures for structure screening. The correct interface structure should obey two qualitative APRES observations: (i) after Fe deposition, there are no additional bands crossing the Fermi energy (e.g., Fe $4s$ or $3d$ bands); (ii) both d_{xz}/d_{yz} and d_{xy} Ti $3d$ bands are present at the Fermi energy before and after Fe deposition. We find that only the Fe_2O_2 composition with Fe $3d$ $U \geq 3$ eV obeys both conditions. The structure for $U = 3$ eV is shown in Fig. 4(a): the Fe_2O_2 layer has a buckled rocksalt structure, both Fe atoms are bonded to the O atoms of the TiO_2 layer below, a lower buckled O is bonded to the Ti below, while the upper buckled O is above a void site of the TiO_2 layer. This particular structure for $U = 3$ eV has an antiferromagnetic

ground state, but a competing ferromagnetic state may still be stabilized by fine-tuning the Fe U value. Regarding the oxygen content, we find that O has a strong energetic preference to oxidize the initially metallic Fe even under the high-vacuum conditions in the experiment (see Sec. III in the Supplemental Material [33]).

Taken in totality, we conclude that when Fe is deposited on the BTO surface, the resulting interface is $\text{Fe}_2\text{O}_2/\text{BTO}$ rather than the prevalently discussed Fe/BTO . This is also consistent with the observation of the lack of large charge transfer from Fe $4s$ to Ti $3d$. The *ab initio* calculated band structures of BTO and $\text{Fe}_2\text{O}_2/\text{BTO}$ films are compared in Fig. 4(b). Both show 2DEG bands with Ti $3d$ orbital components at and near E_F , suggesting the Fe_2O_2 layer is not responsible for 2DEG states. On the other hand, the $\text{Fe}_2\text{O}_2/\text{BTO}$ film additionally displays band dispersions from Fe $3d$ orbitals in the range $E_F - 0.5$ eV to $E_F - 1.5$ eV. Our calculated density of states (DOS) shows the Fe-induced states [center panel of Fig. 4(b)], which are marked with a red open triangle at $E = E_F - 0.7$ eV. This result is consistent with our experimental observation of the peak of the incoherent spectral weight near the same energy [Fig. 2(d)]. The *ab initio* results are also supported by Fe-resonant ARPES data (Fig. S2 [33]), which show a strong enhancement of Fe spectral weight at high binding energy and suppression of Ti $3d$ weight.

We now turn to the low-energy electronic structure of this system. As shown in Fig. 3, the out-of-plane d_{xz} band is lifted in energy and subsequently loses charge, whereas the in-plane (d_{xy}) bands are almost unchanged in energy or occupancy. Such an orbital-selective charge transfer in Ti t_{2g} bands is reproduced in our *ab initio* calculation and is compared to experimental results in Fig. 4(c). Analysis of the first principles electronic structure tells us the microscopic mechanism of such an orbital-selective charge transfer, as illustrated in Figs. 4(d) and 4(e), is due to Ti-Fe hybridization [12,32,58,59]. We start with the bare BTO surface with a 2DEG having in-plane d_{xy} and out-of-plane $d_{xz/yz}$ bands. When the Fe_2O_2 layer is placed on the BTO surface, the wave functions of the in-plane Ti d_{xy} bands barely couple to the overlayer and are thus essentially unperturbed, while the wave functions for the out-of-plane $d_{xz/yz}$ bands hybridize with the out-of-plane Fe $3d$ bands. As the Fe $3d$ bands are mainly

below E_F (and thus below the Ti $d_{xz/yz}$ bands), the hybridization leads to upward level repulsion pushing the Ti $d_{xz/yz}$ bands up in energy and selectively depopulating them.

Future experiments that build on this work include in-plane magnetotransport measurements to verify both the conducting nature of the 2DEG and its carrier density, and field effect measurements of the 2DEG through the application of a voltage on the Nb:STO substrate to flip the ferroelectric polarization and modulate the in-plane transport [13]. The devices required for this work will involve architectures that generate and maintain the 2DEG during fabrication and measurement. For example, capping layers are needed to protect the oxygen vacancies responsible for the 2DEG [51]. Ultimately, one may test whether such an interface, especially with the surprisingly robust 2DEG and iron local moments, becomes truly multiferroic.

In conclusion, we investigate interfacial 2DEG states emergent at a magnetic and ferroelectric interface with ARPES measurements. A single magnetic layer on a ferroelectric thin film induces orbital-dependent charge transfer in out-of-plane Ti $3d_{xz/yz}$ bands while the in-plane Ti $3d_{xy}$ bands remain the same. Our *ab initio* calculations show that the anisotropic hybridization between Fe and Ti $3d$ orbitals across the interface leads to orbital-dependent charge transfer. Our results indicate a possibility to perform orbital specific engineering at an interface, which may help increase spin polarization for potential device applications.

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