


Impact of oxidation state on the valence-bond glass physics in the lithium-intercalated Mo_3O_8 cluster Mott insulators

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We have successfully synthesized four Mo_3O_8 -type cluster Mott insulators (CMIs) by intercalating lithium into nonmagnetic precursors to regulate the Mo_3 cluster valence. The resulting materials are $\text{Li}_{1+x}\text{RMO}_3\text{O}_8$ ($R = \text{Sc}, \text{Y}, \text{Lu}$) and $\text{Li}_x\text{Zn}_2\text{Mo}_3\text{O}_8$. Our magnetic susceptibility measurements revealed that these materials display characteristics akin to a valence-bond glass state and suggest the presence of short-range ordering when the Mo_3 cluster valence approximates its ideal value. These findings challenge the prevailing belief that the plaquette charge ordering state is an inherent feature of Mo_3O_8 -type CMIs. Instead, they underscore the importance of Mo_3 cluster valence in determining the physical properties of these systems. These insights furnish a fresh understanding of the Mo_3O_8 -type CMI and open research opportunities in highly frustrated magnetism.

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I. INTRODUCTION

Quantum spin-liquids (QSLs) have garnered significant attention in condensed matter physics due to their potential to host exotic quantum phenomena such as fractionalized excitations and long-range entanglement [1–4]. These QSL materials, characterized by highly frustrated spin configurations and the absence of conventional long-range magnetic order, have posed a challenge for both experimentalists and theoreticians, demanding the development of innovative synthetic methods and advanced observation tools for QSLs. Moreover, the breakthrough of these material discoveries offers distinct perspectives on quantum physics [5–9].

Mo_3O_8 -type cluster Mott insulators (CMIs) are promising candidates for realizing QSLs. These compounds are comprised of Mo atoms arranged in kagome layers trimerized to form a triangular lattice of Mo_3 clusters, as shown in Fig. 1. Studies of $\text{LiZn}_2\text{Mo}_3\text{O}_8$ have revealed an intriguing phenomenon: the Curie constant at low temperatures is observed to be one-third of its value at high temperatures, implying a diminish of two-thirds of the total spins [10]. An initial explanation for this partial spin disappearance mechanism was proffered by forming a spin singlet on the triangular lattice of a honeycomb superlattice, with the remaining one-third of the spin constituting a valence-bond solid (VBS) [11–13]. Subsequently, the Mo_3 CMIs were reinterpreted as the extended Hubbard model on the $1/6$ filled breathing kagome lattice, leading to the development of a theoretical framework to describe the electronic state of the system, postulating the plaquette charge order (PCO) state as the ground state [14–16]. In the PCO state, three unpaired electrons are arranged in a hexagonal configuration, with each electron placed at a vertex and not adjacent to its neighbors. However, the effect of the

PCO state on the magnetic order is not yet understood. While it accounts for the reduced spin in $\text{LiZn}_2\text{Mo}_3\text{O}_8$, its role in the emergence of a quantum spin liquid state is less clear, given the frustrated geometry of the lattice. The potential for non-nearest neighbor interactions to suppress magnetic order invites further scrutiny. The connection between the PCO state and the suppression of magnetic order in Mo_3O_8 -type CMIs is still under investigation. The presumption that spins on a triangular lattice naturally form a spin liquid state is questionable without factoring in extended interactions. Thus, future research explorings, exemplified by Flint and Lee's emergent honeycomb lattice model [11], are essential to discern if diminished superexchange interactions are solely responsible for inducing a spin liquid state or if other dynamics contribute.

Alternatively, recent structural investigations grounded in solid-state chemistry have revealed significant off-stoichiometry in $\text{LiZn}_2\text{Mo}_3\text{O}_8$ [17,18]. The magnetic susceptibility of $\text{LiZn}_2\text{Mo}_3\text{O}_8$ with nearly perfect stoichiometric ratios is not reproduced at all by the PCO model. Instead, it displays magnetic susceptibility characteristics consistent with strongly correlated two-dimensional antiferromagnetic systems [18]. The diminution of the PCO-like behavior as the stoichiometry improves raises questions regarding the validity of the previous hypothesis. While other Mo_3O_8 -type CMIs like $\text{Li}_2\text{InMo}_3\text{O}_8$ [19] demonstrate magnetic ordering, the characterization of $\text{Li}_2\text{ScMo}_3\text{O}_8$ [19] is more complex. Although it was initially thought to exhibit QSL-like behavior probed in the NMR study, subsequent muon spin rotation experiments have provided clear evidence of spin freezing [15,20]. Similarly, for $\text{Na}_3\text{A}_2(\text{MoO}_4)_2\text{Mo}_3\text{O}_8$ ($A = \text{In}, \text{Sc}$) [21,22], the observed magnetic properties cannot be adequately explained by the PCO model alone, indicating the need for a revised understanding or additional theoretical models to account for these behaviors.

As mentioned, despite a spin defect associated with off-stoichiometry in Mo_3O_8 -type CMIs, the experimental results

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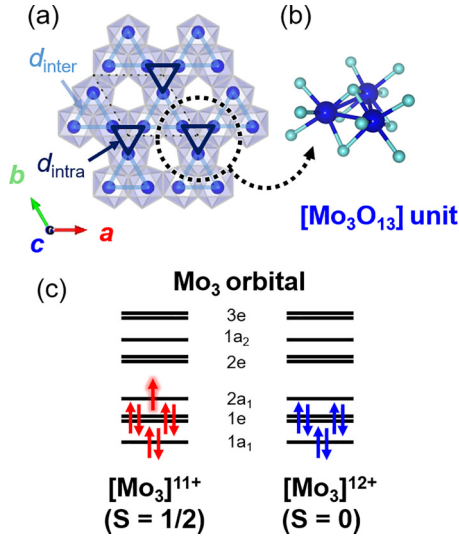


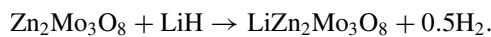
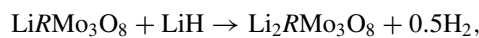
FIG. 1. Triangular lattice constituted by Mo₃ clusters, accompanied by their respective molecular orbitals and electron configurations corresponding to their valence states.

are consistent with theoretical predictions and currently represent the physical properties of Mo₃O₈-type CMI. As shown in Fig. 1(c), magnetic/nonmagnetic characteristics in a Mo₃ cluster are highly susceptible to the valence of molybdenum ion. However, the effect of spin defects caused by deviations from the ideal valence of Mo₃ clusters has not been thoroughly examined. Consequently, we aimed to shed light on the fundamental properties of Mo₃O₈-type CMI by synthesizing different materials through precise control of the valence of Mo₃ clusters, which would be a crucial determinant of the physical properties of these systems.

This paper reports the successful synthesis of four Mo₃O₈-type CMI using Li intercalation to control Mo₃ cluster valence. Our magnetic susceptibility results showed valence-bond glasslike behaviors and short-range ordering as the Mo₃ cluster valence approached the ideal value, contradicting previous understanding of the PCO state as an indispensable attribute of Mo₃O₈-type CMI and highlighting Mo₃ cluster valence as a crucial determinant of these systems' properties. Our findings offer insight into Mo₃O₈-type CMI and the potential for further advancement in frustrated physics.

II. EXPERIMENTAL METHODS

The precursors LiRMO₃O₈ ($R = \text{Sc, Y, Lu}$) and Zn₂Mo₃O₈ were synthesized by conventional solid-state reaction. Li₂RMO₃O₈ and LiZn₂Mo₃O₈ were synthesized by Li intercalation of the obtained precursors according to the following chemical reaction:



The precursors and four-molar excess of LiH were finely ground in an Ar-filled glove box, sealed in an evacuated Pyrex tube, and heated at 300 °C. The optimal molar ratio for the LiH mixture was determined through experimentation. When

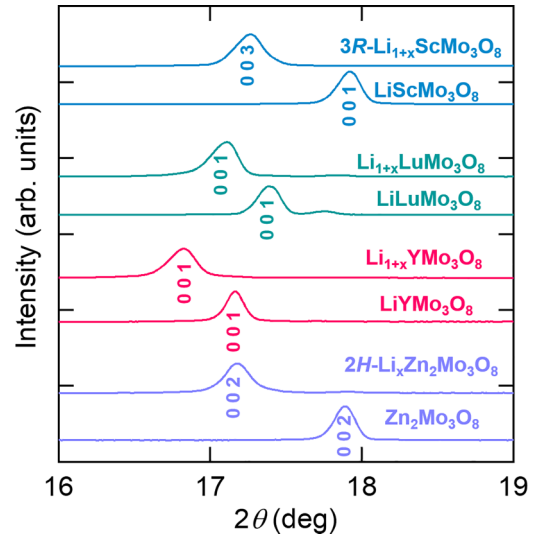


FIG. 2. X-ray diffraction patterns of LiRMO₃O₈ ($R = \text{Sc, Lu, Y}$) and Zn₂Mo₃O₈ and their lithiated form focused on the angular range $16 < 2\theta/\text{deg} < 19$. Intensities have been normalized. Peaks with Miller indices indicate the interlayer spacings d .

the molar amount of LiH was approximately one to three times the stoichiometric ratio, the reaction was incomplete with a remaining unreacted precursor. On the other hand, when the LiH amount was a large excess (about tenfold the stoichiometric ratio), the reduction reaction proceeded further, resulting in the deposition of the molybdenum element. After the reaction, the pellets were washed with methanol to remove residual LiH. The thus-obtained sample was characterized by powder x-ray diffraction (XRD) experiments in a diffractometer with Cu- $K\alpha$ radiation. The cell parameters and crystal structure were refined by the Rietveld method using the Z-RIETVELD v1.1.3 software [24,25] (see the Supplemental Material [26]). The temperature dependence of magnetization was measured under several magnetic fields in a magnetic property measurement system.

III. RESULTS

A. Structural analysis

Using x-ray diffraction analysis, as illustrated in Fig. 2, we observed shifts in the lowest 00 l peaks associated with the Mo₃O₈ interlayer spacing after Li intercalation. In addition, the shifts moved toward lower angles, as shown by comparing the lattice constants of these materials. This change suggests that the intercalation of lithium ions into the samples resulted in a noticeable elongation of interlayer spacing. The observed structural modifications from the precursor materials to lithium-intercalated ones are summarized in Table I. A common change across all samples was the noticeable increase in the interlayer space d after the lithium insertion process. This expansion in interlayer space provides strong evidence that lithium ions have indeed been successfully inserted into the interlayer regions of these materials.

To comprehensively investigate the structural changes caused by lithium insertion, we performed Rietveld analysis on all samples, as depicted in Fig. 3. The detailed

TABLE I. Comparison of structural parameters for precursor and Li-intercalated samples.

	Space group	a (Å)	d (Å)
LiScMo ₃ O ₈	$P3m1$	5.724(1)	4.943(1)
3R-Li _{1+x} ScMo ₃ O ₈	$R3m$	5.771 93(7)	5.137 59(3)
LiLuMo ₃ O ₈	$P3m1$	5.766(1)	5.096(1)
Li _{1+x} LuMo ₃ O ₈	$P3m1$	5.767 3(5)	5.207 43(5)
LiYMo ₃ O ₈	$P3m1$	5.781(1)	5.153(2)
Li _{1+x} YMo ₃ O ₈	$P3m1$	5.767 43(9)	5.260 56(10)
Zn ₂ Mo ₃ O ₈	$P6_3mc$	5.783 5(2)	4.949 8(3)
2H-Li _x Zn ₂ Mo ₃ O ₈	$P6_3mc$	5.731 81(2)	5.168 28(5)

structural parameters are listed in the Supplemental Material [26]. The results revealed that Li_{1+x}LuMo₃O₈, Li_{1+x}YMo₃O₈ and 2H-Li_xZn₂Mo₃O₈ retained the same space group as their precursors ($x = 0$), indicating no change in their crystal structure. On the contrary, for 3R-Li_{1+x}ScMo₃O₈, there was a shift in the space group from $P3m1$ to $R3m$, as detected in the x-ray diffraction patterns. It is important to note that the space groups of 3R-Li₂ScMo₃O₈ and 2H-LiZn₂Mo₃O₈ differed from those synthesized via the high-temperature ceramics method.

Figure 4 depicts the crystal structure transformations through lithium intercalation. LiYMo₃O₈, LiLuMo₃O₈, and

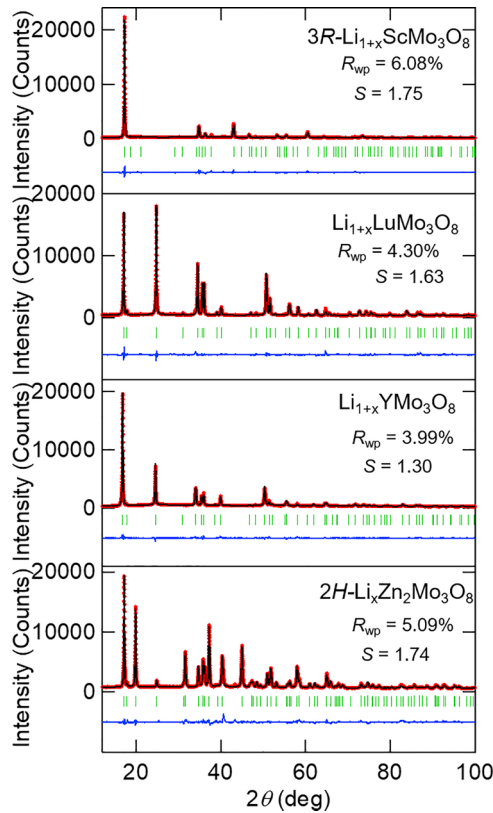


FIG. 3. X-ray diffraction patterns of lithiated LiRMo₃O₈ ($R = \text{Sc, Lu, Y}$) and Zn₂Mo₃O₈. The green vertical lines indicate the Bragg reflection positions of the intercalated samples. The observed intensities (red circles), calculated intensities (black line), and their differences (blue curve at the bottom) are shown.

Zn₂Mo₃O₈ retained their crystal structures, showing identical space groups before and after lithiation. On the other hand, in the case of LiScMo₃O₈, we noticed structural alterations postlithiation. The precursor LiScMo₃O₈ presents a $P3m1$ space group structure akin to LiYMo₃O₈ and LiLuMo₃O₈. However, the layer stacking pattern changed to an $R3m$ structure upon lithium insertion. This observation suggests that the relatively lighter Sc³⁺ ions likely migrate within the crystal structure due to the lithium insertion process, resulting in this remarkable structural transformation.

B. Magnetization measurement

The temperature dependencies of the magnetic susceptibility M/H and the inverse magnetic susceptibility H/M of the intercalated samples are depicted in Fig. 5, with the measurement results of the precursors included for comparison. The measurements were conducted under an applied external magnetic field of 1 T. The linearity observed in the high-temperature region of the inverse susceptibility suggests the presence of localized spins.

We performed Curie-Weiss fitting to χ data at high temperatures (200–300 K), yielding the effective magnetic moments ($\mu_{\text{eff}}^{\text{HT}}$) and Weiss temperatures ($\theta_{\text{W}}^{\text{HT}}$), as summarized in Table II. All intercalated samples exhibited an enhancement in the $\mu_{\text{eff}}^{\text{HT}}$ -value compared to their precursors. This enhancement must be linked to the Li intercalation, which is theorized to cause a reduction in the valence of the Mo₃ cluster from 12+ ($S = 0$) to 11+ ($S = 1/2$). The effective magnetic moment for the precursors is significantly smaller than Li-intercalated samples but not the expected value of zero. This unexpected finding suggests that there is inherent off-stoichiometry also in the precursor and that the Mo clusters are not in their ideal valence states, with evidence of slight valence change from ideal [Mo₃]¹²⁺. Surprisingly, there appears to be a substantial magnetic correlation among the diluted spins in the Mo₃ clusters of the precursor, as probed by the significantly large negative Weiss temperature. This observation is distinct from the Curie-tail behavior characterized by an almost zero Weiss temperature at low temperatures, as seen in the Li-intercalated compounds discussed subsequently. Such a large Weiss temperature has also been observed in Li₂Sc_{1-x}Sn_xMo₃O₈, which exhibits significant spin dilution [27].

Table II presents the inferred Mo₃ cluster valence and the projected chemical formula based on the effective magnetic moment values. The Mo₃ cluster valence was calculated by substituting the μ_{eff} value into the equation as follows:

$$n = 12 - \left(\frac{\mu_{\text{eff}}^{\text{HT}}}{2\sqrt{S(S+1)}} \right)^2 = 12 - \left(\frac{\mu_{\text{eff}}^{\text{HT}}}{\sqrt{3}} \right)^2. \quad (1)$$

Here, the methodology employed was to iteratively compute the μ_{eff} value from the assumed chemical formula and subsequently determine the chemical formula corresponding to the calculated μ_{eff} value until convergence (details are provided in the Supplemental Material [26]). As a result, the calculated μ_{eff} value of all samples is less than the expected $\sim 1.73\mu_{\text{B}}$ for $S = 1/2$, indicating an incomplete occupancy of lithium ions at the vacancy sites of precursors.

As shown in Fig. 5, there is an evident deviation from the Curie-Weiss (CW) law at low temperatures, which is

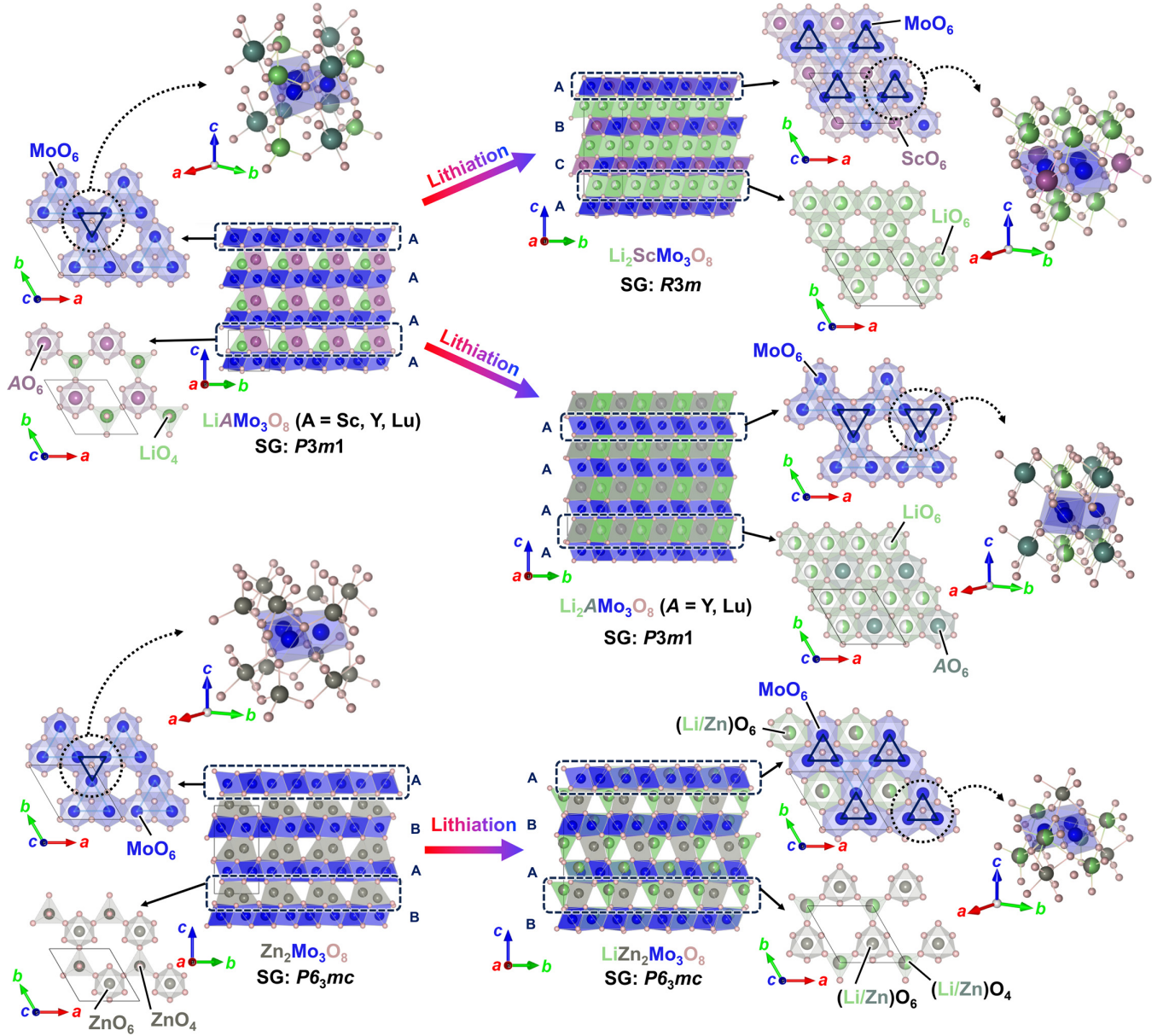


FIG. 4. The crystal structure of nonmagnetic precursors LiRMO_3O_8 ($R = \text{Lu, Y, Sc}$) and $\text{A}_2\text{MO}_3\text{O}_8$ ($A = \text{Mg, Zn}$) and their lithiated products $\text{Li}_2\text{RMO}_3\text{O}_8$ ($R = \text{Lu, Y, Sc}$) and $\text{Li}_2\text{A}_2\text{MO}_3\text{O}_8$ ($A = \text{Mg, Zn}$) along with the transformation. The layer-by-layer crystal structure and the local environment around the Mo_3O_{13} cluster of each compound are also shown. The VESTA program is used for visualization [23].

particularly noticeable in the behavior of the inverse susceptibility as the temperature decreases. However, as the temperature drops, a trend of linear inverse susceptibility with temperature change reemerges. These behaviors are ob-

served across all materials under study. The CW fitting in the low-temperature region was carried out separately from the fitting in the high-temperature range. The resultant values for the effective magnetic moments ($\mu_{\text{eff}}^{\text{LT}}$) and the Weiss

TABLE II. The Mo_3 cluster valences, high-temperature and low-temperature μ_{eff} , and θ_{W} values for lithiated products.

Sample	$x =$	$[\text{Mo}_3]^{n+}$	$\mu_{\text{eff}}^{\text{HT}}/\text{Mo}_3(\mu_{\text{B}})$	$\theta_{\text{W}}^{\text{HT}}$ (K)	$\mu_{\text{eff}}^{\text{LT}}/\text{Mo}_3(\mu_{\text{B}})$	$\theta_{\text{W}}^{\text{LT}}$ (K)	Reference
ideal		11	$\sqrt{3}$		1		
$6R\text{-Li}_{0.95}\text{Zn}_{1.92}\text{Mo}_3\text{O}_8$		11.11	1.63(12)	-440(110)	0.607(2)	-5.2	[18]
$3R\text{-Li}_{1+x}\text{ScMo}_3\text{O}_8$	0.86	11.14	1.609(5)	-215(5)	0.552(8)	-5.6(5)	this work
$2H\text{-Li}_x\text{Zn}_2\text{Mo}_3\text{O}_8$	0.77	11.23	1.521(2)	-202(2)	0.338(7)	-3.9(3)	this work
$\text{Li}_{1+x}\text{LuMo}_3\text{O}_8$	0.77	11.23	1.516(2)	-103.7(9)	0.732(8)	-3.8(2)	this work
$6R\text{-Li}_{1.2}\text{Zn}_{1.8}\text{Mo}_3\text{O}_8$		11.36	1.39	-220	0.8	-14	[10]
$\text{Li}_{1+x}\text{YMo}_3\text{O}_8$	0.62	11.38	1.362(7)	-211(5)	0.603(2)	-0.78(7)	this work

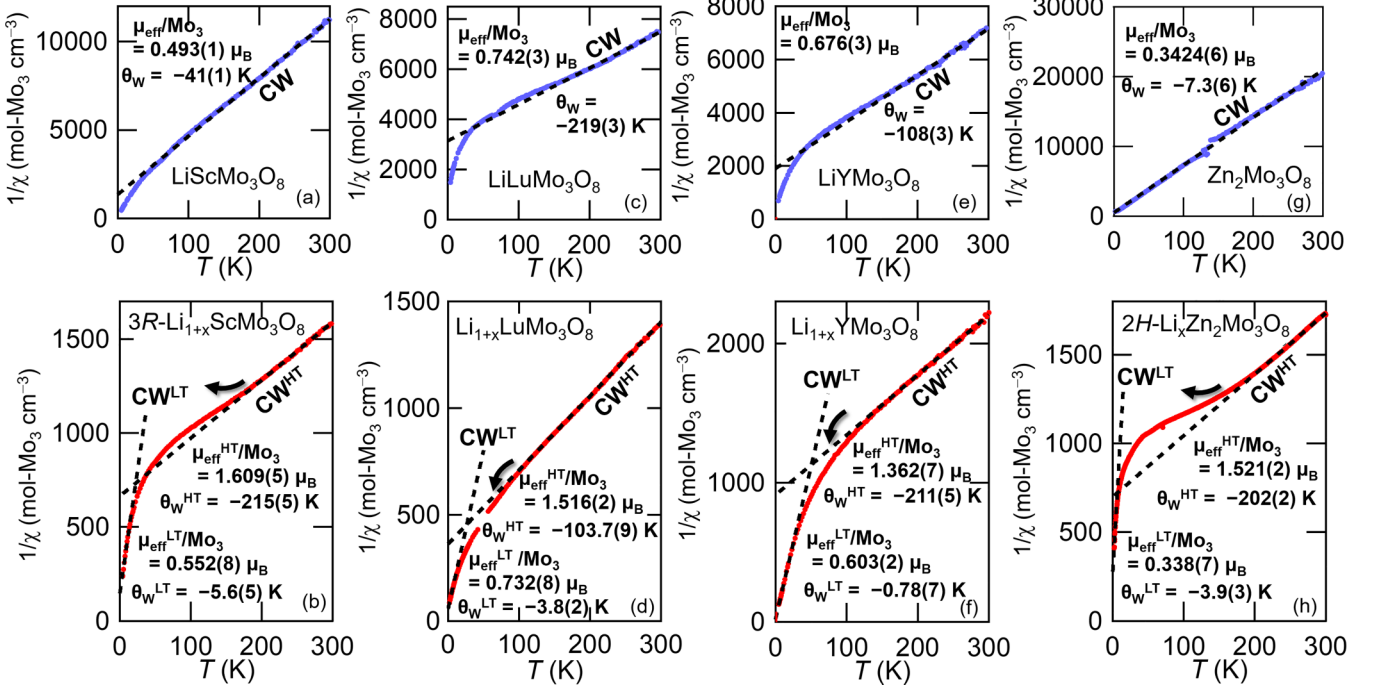


FIG. 5. Temperature dependence of inversed magnetic susceptibility M/H for the following compounds: (a) $\text{LiScMo}_3\text{O}_8$, (b) $3R\text{-Li}_{1+x}\text{ScMo}_3\text{O}_8$, (c) $\text{LiLuMo}_3\text{O}_8$, (d) $\text{Li}_{1+x}\text{LuMo}_3\text{O}_8$, (e) LiYMo_3O_8 , (f) $\text{Li}_{1+x}\text{YMo}_3\text{O}_8$, (g) $\text{Zn}_2\text{Mo}_3\text{O}_8$, and (h) $2H\text{-Li}_x\text{Zn}_2\text{Mo}_3\text{O}_8$. The dotted lines in all panels indicate the results of Curie-Weiss fits. The sinuous arrows delineate the temperature regions wherein the observed M/H curves exhibit deviation from the Curie-Weiss fit, distinguishing the manner of deviation by convex up and down arrows.

temperatures (θ_W^{LT}) obtained from this low-temperature CW fitting are detailed in Table II. The absolute values of θ_W^{LT} and $\mu_{\text{eff}}^{\text{LT}}$ are smaller than θ_W^{HT} and $\mu_{\text{eff}}^{\text{HT}}$. This trend is consistent across all intercalated samples. This phenomenon can be attributed to the partial suppression of magnetic moments commonly observed in $\text{LiZn}_2\text{Mo}_3\text{O}_8$ [18], $\text{Li}_2\text{Sc}_{1-x}\text{In}_x\text{Mo}_3\text{O}_8$ [15], and $\text{Li}_2\text{Sc}_{1-x}\text{Sn}_x\text{Mo}_3\text{O}_8$ [27].

The behavior of the inverse susceptibility in the intermediate temperature range displays deviations from the CW line in all Li-intercalated Mo_3O_8 materials. These deviations, however, are not uniform across the different materials and can be grouped into two distinct classes. The first group, $\text{Li}_{1+x}\text{LuMo}_3\text{O}_8$ and $\text{Li}_{1+x}\text{YMo}_3\text{O}_8$, displays a downward deviation from the CW line as temperature decreases. Conversely, the second group, comprising $3R\text{-Li}_{1+x}\text{ScMo}_3\text{O}_8$ and $2H\text{-Li}_x\text{Zn}_2\text{Mo}_3\text{O}_8$, shows an upward deviation from the CW line. Differences in behavior across various materials can be linked to their respective μ_{eff} values—the first group materials exhibit relatively small μ_{eff} values, whereas the second group demonstrates larger μ_{eff} values. The intricate relationship between these values and the observed susceptibility behaviors will be discussed in subsequent sections.

Figure 6 shows the isothermal magnetization measured at $T = 4.2$ K. All samples exhibit magnetization curves characterized by upward convexity, indicative of free spin presence. To quantify the free spin amount, we employed a magnetization model for analysis comprising two components: a quasi-free-spin component modeled by the Brillouin function of $S = 1/2$ and a strongly correlated spin component, which is

linearly dependent on the applied magnetic field,

$$M = \chi H + f N_A \mu_B \times B_s [H \mu_B / k_B (T - \theta_w)], \quad (2)$$

where χ corresponds to the magnetic susceptibility attributable to correlated spins, and θ_w represents the Weiss temperature associated with the nearly free spins. The symbols N_A , k_B , and B_s denote the Avogadro number, Boltzmann's constant, and Brillouin function, respectively. The parameter f quantifies the proportion of free spins. The optimal fits are

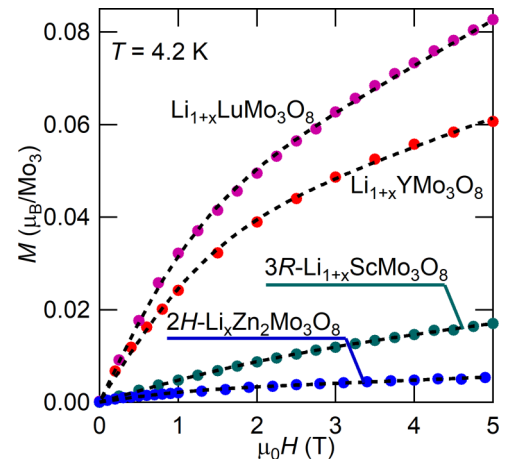


FIG. 6. The isothermal magnetization of at 4.2 K (circles) and those fitting curves (see the text).

TABLE III. Results of fits to the isothermal magnetization data using the formula described in the text. The fitting parameters g and S are fixed to 2 and 0.5, respectively.

Sample	f (%)	θ_W (K)	χ (cm ³ /mol-Mo ₃)
3R-Li _{1+x} ScMo ₃ O ₈	0.61(6)	2.68(1)	$1.23(8) \times 10^{-3}$
Li _{1+x} LuMo ₃ O ₈	3.4(1)	3.31(4)	$5.4(2) \times 10^{-3}$
Li _{1+x} YMo ₃ O ₈	3.1(2)	3.22(7)	$3.4(5) \times 10^{-3}$
2H-Li _x Zn ₂ Mo ₃ O ₈	0.21(2)	3.45(8)	$3.8(5) \times 10^{-4}$

illustrated by the dashed lines in Fig. 6, with the corresponding fitting parameters presented in Table III. The ability of this model to accurately characterize the magnetization process suggests the presence of free spins at low temperatures, which is also probed by the nearly zero Weiss temperature. However, the data do not provide enough information to ascertain whether these spins are extrinsic, potentially due to defects or impurities, or if they are an intrinsic feature of the system. Thus, the observed Curie-tail-like divergence in the temperature dependence of magnetic susceptibility can be attributed to the presence of nearly free spins.

IV. DISCUSSION

The estimated $\mu_{\text{eff}}^{\text{HT}}$ value in all lithiated Mo₃O₈-type CMI was less than anticipated for the $S = 1/2$ spin state. As displayed in Table II, this discrepancy suggests that some Mo₃ clusters possess their nonmagnetic [Mo₃]¹²⁺ state rather than magnetic [Mo₃]¹¹⁺. This situation is visually represented using an $S = 1/2$ dilute triangular lattice model, shown in Fig. 7(a). The magnetism of CMI in the high-temperature region is consistent with the CW law with strong antiferromagnetic interactions. Interestingly, CMIs stray from the Curie-Weiss law in the intermediate temperature range as the temperature decreases, only to return to compliance in the low-temperature region. The difference between the two CW

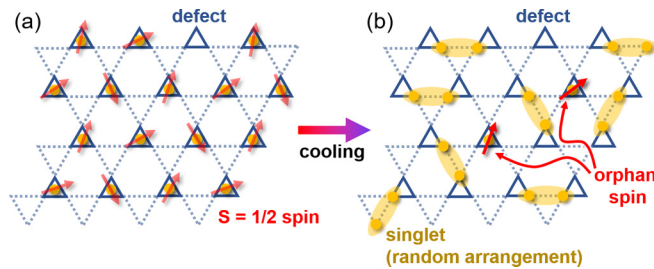


FIG. 7. (a) The partial spin defect model on a triangular lattice constituted by Mo₃ clusters. The Mo₃ clusters are depicted as equilateral triangles. On these triangles, yellow circles with red arrows are placed, symbolizing unpaired electrons corresponding to the $S = 1/2$ spin derived from the [Mo₃]¹¹⁺ cluster. The unmarked triangle represents the [Mo₃]¹²⁺ cluster referred to as a nonmagnetic defect due to its lack of unpaired electrons. (b) The formation diagram of VBG. Yellow ellipses signify spin-singlet pairs, indicating paired electrons with opposite spins that result in no net magnetic moment. The isolated spins, on the other hand, are still represented by red arrows.

regimes is marked by a decline in the μ_{eff} and θ_W values from high to low temperatures.

The valence-bond glass (VBG) model reasonably explains the observed behaviors [28,29]. This model suggests that influences such as spin defects can instigate bond randomness. This randomness creates spin singlets in a disordered state, leaving isolated spins at overlooked sites during singlet formation, as visualized in Fig. 7(b). This occurrence helps interpret the noticed reduction in spin behaviors as temperature decreases. At high temperatures, the system shows clear paramagnetism. However, as temperatures decrease, a deviation from the CW law is noted, correlating with the gradual formation of spin singlets due to strong spin interactions. At low temperatures, isolated spins manifest in a VBG state. These states align with the two CW regimes and offer a plausible explanation for the decrease in both μ_{eff} and $|\theta_W|$ values observed during our experiments. Remarkably, as shown in Fig. 5, the H/M curves of the precursors also display a two CW regime, which persists even with a low concentration of spins. This phenomenon could potentially be accounted for by the VBG mechanism.

LiZn₂Mo₃O₈ also exhibits a spin reduction behavior, as previously documented [10]. The PCO mechanism has been proposed to elucidate the magnetism of LiZn₂Mo₃O₈ [14]. In the PCO state, three unpaired electrons, referred to as “plaquettes,” are distributed evenly across the lattice, ensuring that no two plaquettes share a vertex of the hexagon formed by three Mo₃ clusters. This model successfully explains the magnetism that appears to be a 2/3 spin reduction. The magnetic properties of LiZn₂Mo₃O₈, particularly the high-temperature μ_{eff} and θ_W values, are similar to those of Li_{1+x}YMo₃O₈ and Li_{1+x}LuMo₃O₈. The present study has successfully applied the VBG model to describe the magnetic properties of Li_{1+x}YMo₃O₈ and Li_{1+x}LuMo₃O₈, suggesting the presence of partial spin defects in these compounds. This scenario is further supported by the temperature dependence of inverse susceptibility in α -LiZn₂Mo₃O₈ [10], which aligns qualitatively and quantitatively with the magnetism observed in Li_{1+x}YMo₃O₈ and Li_{1+x}LuMo₃O₈. Notably, subsequent investigations have revealed a significant Li/Zn off-stoichiometry in LiZn₂Mo₃O₈ [18].

Here, we leave aside the possibility of VBG and consider whether the PCO model can explain each. The temperature dependence of the magnetic susceptibility in the PCO state is predicted by the following equation:

$$\chi_{\text{PCO}} = \frac{\beta(T)C}{T - \beta(T)\theta_W} \left(\beta(T) = \frac{1 + 5 \exp(-\Delta E/k_B T)}{1 + \exp(-\Delta E/k_B T)} \right), \quad (3)$$

where C is the Curie constant, and ΔE is the energy gap between the ground state with $S_{\text{tot}} = 1/2$ and the excited state with $S_{\text{tot}} = 3/2$, comprised of three spins forming a plaquette. The “phenomenological PCO model” derived by Akbari-Sharbat *et al.* [15] significantly simplifies the theoretical ideas originally proposed by Chen *et al.* [14], yet it is known to reproduce experimental data well. We fit the H/M curves of Li-intercalated Mo₃O₈ compounds to the phenomenological PCO model with Eq. (3), as shown in Fig. 8. The phenomenological PCO model provides a

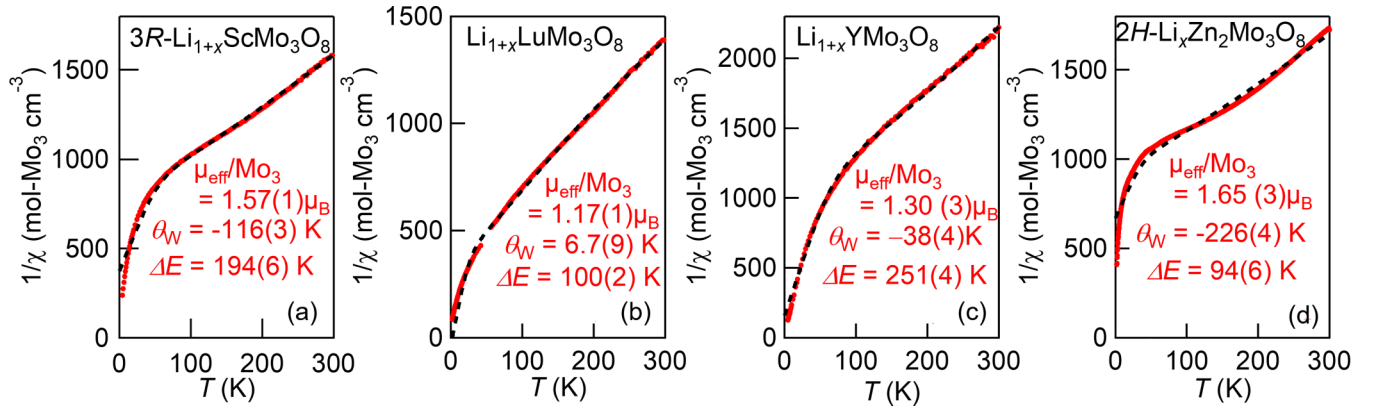


FIG. 8. Results obtained from fitting the phenomenological PCO model to the H/M data for (a) $\text{Li}_{1+x}\text{ScMo}_3\text{O}_8$, (b) $\text{Li}_{1+x}\text{LuMo}_3\text{O}_8$, (c) $\text{Li}_{1+x}\text{YMo}_3\text{O}_8$, (d) $\text{Li}_x\text{Zn}_2\text{Mo}_3\text{O}_8$. Values near the fitting result curves are fitting parameters (see the main text).

satisfactory approximation of the magnetic susceptibility only for $\text{Li}_{1+x}\text{YMo}_3\text{O}_8$ and $\text{Li}_{1+x}\text{LuMo}_3\text{O}_8$. However, it fails to accurately represent the χ data for $3R\text{-Li}_{1+x}\text{ScMo}_3\text{O}_8$ and $2H\text{-Li}_x\text{Zn}_2\text{Mo}_3\text{O}_8$. This discrepancy suggests that these materials manifest magnetic states beyond the predictive capability of the phenomenological PCO model. Furthermore, while the effective magnetic moments estimated by the phenomenological PCO fit for $\text{Li}_{1+x}\text{YMo}_3\text{O}_8$ approximately agree with that by the CW fit, this is not true for $\text{Li}_{1+x}\text{LuMo}_3\text{O}_8$, which is seemingly well fit by the phenomenological PCO model. This variation undermines the reliability of the phenomenological PCO fit.

The reproducibility of the phenomenological PCO model for Mo_3O_8 -type CMIs is limited, as its applicability is restricted to certain materials such as $\text{Li}_{1+x}\text{YMo}_3\text{O}_8$ with a $\mu_{\text{eff}}^{\text{HT}}$ value of approximately $1.4\mu_{\text{B}}$. This constraint also applies to $6R\text{-LiZn}_2\text{Mo}_3\text{O}_8$ [10] and $\text{Li}_2\text{Sc}_{0.6}\text{In}_{0.4}\text{Mo}_3\text{O}_8$ [15], each exhibiting $\mu_{\text{eff}}^{\text{HT}}$ values around $1.4\mu_{\text{B}}$. In addition, The phenomenological PCO model accurately replicates the entire temperature profile of the inverse susceptibility for $\text{Li}_2\text{Sc}_{0.8}\text{Sn}_{0.2}\text{Mo}_3\text{O}_8$ ($\mu_{\text{eff}}^{\text{HT}} = 1.39\mu_{\text{B}}$), in which the valence is deliberately decreased by substituting Sc^{3+} with Sn^{4+} of $2H\text{-Li}_2\text{ScMo}_3\text{O}_8$, as detailed in the Supplemental Material [26]. On the other hand, the phenomenological PCO model cannot accurately represent the magnetization of three particular compounds, $3R\text{-Li}_{1+x}\text{ScMo}_3\text{O}_8$ and $2H\text{-Li}_x\text{Zn}_2\text{Mo}_3\text{O}_8$. This discrepancy is primarily due to the appearance of antiferromagnetic short-range order (SRO), a behavior not accounted for in the phenomenological PCO model. Similar observations have been noted in the optimally synthesized $\alpha\text{-LiZn}_2\text{Mo}_3\text{O}_8$ sample with an almost perfect $\mu_{\text{eff}}^{\text{HT}}$ value of 1.73 , demonstrating SRO without PCO behavior, as reported by Sandvik *et al.* [18]. Consequently, these findings bring into question the validity of the phenomenological PCO model, especially for the $[\text{Mo}_3]^{11+}$ state, considered ideal within this framework. Furthermore, these results suggest that attaining the predictions of the PCO mechanism might be impracticable in situations devoid of spin defects.

Our study reveals several key findings about Mo_3O_8 -type CMIs. First, our findings indicate that magnetic susceptibility curves consistent with the phenomenological PCO model are detectable within a limited Mo_3 valence range characterized by notable spin defects.

Second, our study showed a magnetization divergence in lower temperature regions, likely due to VBG-like states arising from disorder. Furthermore, a previous exact diagonalization-based theoretical investigation reported by Watanabe has revealed a notable coexistence of short-range order and emergent free spin in a spin-1/2 random Heisenberg antiferromagnetic model on a triangular lattice [29], which aligns with our observations.

Third, we observed that an SRO emerges as the Mo_3 -cluster valence nears its ideal $[\text{Mo}_3]^{11+}$ state, suggesting the gradual realization of an $S = 1/2$ Heisenberg triangular lattice antiferromagnetic model. However, defects have significant implications: although the system can support the emergence of SRO with defects, these defects can also foster the development of VBG-like states, triggering random dimer formation.

As mentioned, the magnetic characteristics of Mo_3O_8 -type CMIs are significantly influenced by valence. Figure 9 encapsulates that two-dimensional short-range order is discernible in systems where the effective magnetic moment is close to $\mu_{\text{eff}} = 1.73\mu_{\text{B}}$ expected for $S = 1/2$ spins. Conversely, systems including highly concentrated spin defects with μ_{eff} ranging $1.3 - 1.4\mu_{\text{B}}$ show the χ curves as

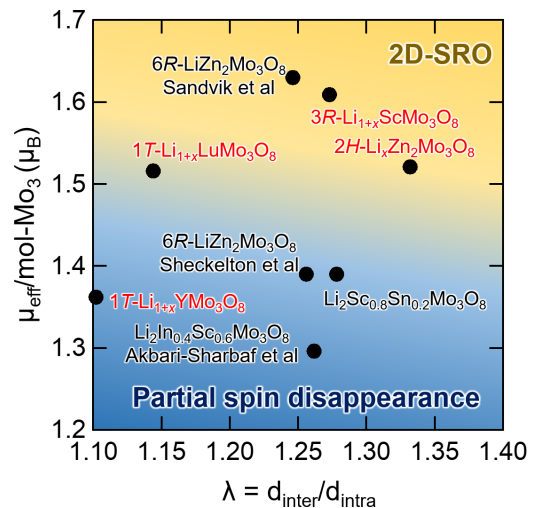


FIG. 9. Relation between effective magnetic moment, breathing parameter λ , and magnetic states in Mo_3O_8 -type CMIs.

seemingly characterized for the PCO state. Hence, depending on the valence within the Mo_3 clusters, the magnetic behavior can be bifurcated into two distinct categories. Contrary to theoretical predictions for Mo_3O_8 -type CMIs, which posit that magnetism is predominantly influenced by the breathing parameter λ (defined as the ratio of intercluster Mo-Mo d_{inter} to intracuster Mo-Mo bond distances d_{intra}), our observations, in conjunction with previous reports, suggest that λ does not systematically correlate with variations of magnetic ground states. However, λ would be associated with the valence of the Mo_3 clusters. As shown in Fig. 1(c), a valence state of $[\text{Mo}_3]^{11+}$ ($S = 1/2$) results in a single unpaired electron in the bonding orbital $2a_1$, whereas for the nonmagnetic $[\text{Mo}_3]^{12+}$ ($S = 0$) state, this orbital is unoccupied. The difference in valence modifies the “bond order” within the Mo_3 clusters; a higher bond order diminishes the Mo-Mo bond length to lower the energy level of the $2a_1$ orbital. Then, an elevation in the bond order would concomitantly increase the λ parameter. Consequently, the λ parameter varies solely with the valence of the Mo_3 cluster and appears not to serve as a direct determinant of magnetism. It is plausible to assert that magnetism is exclusively regulated by the valence, with the λ parameter being determined solely by the interplay between the valence and the lattice constant.

An integrated analysis of our findings supports that the VBG model is a good solution for magnetism in CMIs, although there is no smoking gun to rule out the PCO model. Therefore, the simplicity and directness of the VBG model make it a compelling approach to understanding the mysterious magnetism in CMIs.

V. SUMMARY

We successfully synthesized four Mo_3O_8 -type CMIs: $\text{Li}_{1+x}\text{RMO}_3\text{O}_8$ ($R = \text{Sc}, \text{Y}, \text{or Lu}$) and $\text{Li}_x\text{Zn}_2\text{Mo}_3\text{O}_8$, achieved by lithiation into nonmagnetic precursors, thus controlling the valence of the Mo_3 clusters in the process. Following synthesis, we then performed magnetization measurements on these Mo_3O_8 -type CMIs. The results revealed some intriguing properties: when the Mo_3 cluster valence neared the ideal value, the materials exhibited both VBG behaviors and evidence

of short-range ordering. These findings are notable because they question previous beliefs about the essential properties of Mo_3O_8 -type CMIs, including the assumed ubiquity of the PCO state. We propose that the valence state of the Mo_3 cluster is a critical factor influencing the physical properties of Mo_3O_8 -type CMIs. Despite the significance of this notion, it has not received much emphasis in prior reports. In conclusion, our study’s insights pave the way for further exploration and deepened understanding of frustrated magnetic materials that may contain various chemical defects beyond Mo_3O_8 -type CMIs.

Note added. Recently, we became aware that a similar conceptual study has been reported independently by Wyckoff *et al.* [30]. In this report, Li^+ ions were intercalated into $\text{LiScMo}_3\text{O}_8$ through electrochemical methods, enabling the synthesis of compounds with varying Li contents. Although a detailed interpretation of the temperature dependence of the magnetic susceptibility is not provided, these efforts have been successful in observing both magnetic ordering and changes in the effective magnetic moment that are dependent on the Li composition in compounds near the $\text{Li}_2\text{ScMo}_3\text{O}_8$ composition, with nearly complete control over the Li content. We believe that $\text{Li}_x\text{ScMo}_3\text{O}_8$ systems with such continuously variable defects can also be understood within the framework of the VBG model. Furthermore, the fact that $\text{Li}_2\text{ScMo}_3\text{O}_8$ samples synthesized by high-temperature solid-state reaction methods [19] alongside those obtained through the electrochemical method [18] and the current hydride methods exhibit different space groups suggests a polymorphism derivative of the synthesis method. This polymorphism underlines the diverse potential for the further advancement of research in Mo_3O_8 -type CMIs.

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