Phase transitions of LaMnO₃ and SrRuO₃ from DFT + U based machine learning force fields simulations

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Perovskite oxides are known to exhibit many magnetic, electronic, and structural phases as function of doping and temperature. These materials are theoretically frequently investigated by the DFT + U method, typically in their ground state structure at T = 0. We show that by combining machine learning force fields (MLFFs) and DFT + U based molecular dynamics, it becomes possible to investigate the crystal structure of complex oxides as function of temperature and U. Here, we apply this method to the magnetic transition metal compounds LaMnO₃ and SrRuO₃. We show that the structural phase transition from orthorhombic to cubic in LaMnO₃, which is accompanied by the suppression of a Jahn-Teller distortion, can be simulated with an appropriate choice of U. For SrRuO₃, we show that the sequence of orthorhombic to tetragonal to cubic crystal phase transitions can be described with great accuracy. We propose that the U values that correctly capture the temperature-dependent structures of these complex oxides can be identified by comparison of the MLFF simulated and experimentally determined structures.

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I. INTRODUCTION

Complex transition metal oxides, in particular those with a perovskite structure (ABO_3), possess rich phase diagrams, in which many phases with remarkable physical properties can be identified, such as high T_c superconductivity [1], ferromagnetism [2], ferroelectricity, or multiferroic properties, and metal to insulator transitions under external fields, pressure, doping, or temperature change [3]. Many such properties are not only interesting because of fundamental science, but they may also lead to new types of applications in oxide electronics [4,5]. The properties arise due to the intricate interplay between the structure and the electronic interactions, leading to charge order, magnetic order, and/or orbital order [5,6]. The emergence of different structural phases is therefore strongly tied to the nature of the electronic structure.

Many of the interesting phases occur at a finite temperature (T > 0 K) and over a limited temperature range. This calls for simulations that can describe the physics at elevated temperatures. Ordinary molecular dynamics (MD) is of limited use, as it is based upon fixed force fields, which do not consider changes in the electronic structure. In the materials discussed above, the latter is intimately coupled to, if not driving, the phase transitions. *Ab initio* MD would be an option, but unfortunately it is very time consuming, as in every MD step a density functional theory (DFT) calculation has to be performed to obtain the total energy and the forces on the atoms. In practice, only processes occurring over very small time scales can then be simulated. In the past three years this type of on-the-fly MLFFs has been successfully employed in various applications, such as phase transitions in hybrid lead halide perovskites [12,13], superionic transport in AgI [14], melting of solids [15], surface reconstruction of palladium adsorbed on silver [16], NMR ${}^{1}\text{H}-{}^{1}\text{H}$ dipolar coupling in mixed hybrid lead halide perovskites [17], the structure-band gap relationship in SrZrS₃ [18], and the catalysis dynamics of H/Pt [19].

The MLFFs in those studies are based upon results obtained with standard generalized gradient approximation (GGA) functionals. Such functionals, however, tend not to describe interactions between localized electrons very well, such as they occur on transition metal ions in oxides, for instance. The description can be improved significantly by adding a parametrized model electron-electron interaction on the metal ions, leading to the DFT + U method [20]. The latter is designed to capture some of the essential physics of electrons localized in atomic shells [20–24]. Although, in principle, more accurate methods for describing local correlations exist [23,25], the DFT + U method is cost efficient and

Recent developments in machine-learning (ML) interatomic potentials open up new computational routes [7–10]. Here we apply an approach where a ML model of the potential energy surface is trained on-the-fly during *ab initio* MD runs [11] using limited supercell sizes and time scales. Subsequently, the machine learned force field (MLFF) is then used to perform simulations on much larger supercells and time scales. On the one hand, this approach takes the advantages of *ab initio* MD, i.e., it incorporates the information about the electronic structure. On the other hand, a MLFF greatly reduces the computational resources required to run the MD simulations, thereby enabling the ensemble sizes and time scales required to draw statistically valid conclusions.

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currently the practical electronic structure method of choice for calculations on systems with sizable unit cells involving localized electrons, as they occur in many transition metal compounds [23,25,26].

In this paper we address the question of whether the on-the-fly MLFF method can learn a representation of the potential energy surface of materials studied by DFT + Uand correctly capture the physics of the interplay between the electronic and crystal structure at finite temperatures. Several first-principles techniques exist that allow for a calculation of the on-site Coulomb or exchange parameters (U, J), based upon linear response [27], unrestricted Hartree-Fock [28,29], constrained random-phase approximation [30–32], or machine learning [33], but in practice these parameters are often treated as empirical. Furthermore, there may be a spread in the parameter values obtained with different theoretical approaches and not all values are satisfactory [34-36]. Finite temperature simulations using MLFFs might then allow for testing those values, while producing data that can be compared to experiment [37].

We study two archetypal complex transition metal oxide perovskites, $LaMnO_3$ and $SrRuO_3$, for which one expects to see a varying degree of electron localization on the transition metal ion. Both these materials show structural phase transitions as a function of temperature and, as we will see, the proper simulation of those requires an adequate description of that localization.

The transition metal Mn in LaMnO₃ has a very localized 3*d* shell with a d^4 configuration in a high spin state (three electrons in t_{2g} states and one in a e_g state, all parallel, leading to a maximum magnetic moment $M = 4 \mu_B$). Bulk LaMnO₃ is an insulator with antiferromagnetic order at low temperature and a Neél temperature of 140 K [38]. Because the e_g states are filled with a single electron, the MnO₆ octahedra in LaMnO₃ are Jahn-Teller (JT) distorted [39].

It results in orbital ordering, which controls the orthorhombic structure observed for LaMnO₃ at temperatures T < 750 K. LaMnO₃ shows a somewhat unusual structural phase transition at T = 750 K, where the JT distortion is suppressed and the lattice parameters become equal. However, in this "metric cubic" phase octahedral tilting remains present and the symmetry of the crystal remains orthorhombic [39]. Using results from our MLFF MD simulations, we present a direct comparison to crystal structure data obtained from x-ray diffraction experiments at finite temperatures.

The 4*d* transition metal Ru in the perovskite SrRuO₃ also has a d^4 electron configuration. The Ru 4*d* shell is much less localized than the Mn 3*d* shell and the properties of SrRuO₃ are markedly different from those of LaMnO₃. The Ru 4*d* electrons adopt a low spin configuration with all electrons in t_{2g} states. This can lead to a maximum atomic magnetic moment $M = 2 \mu_B$, which in SrRuO₃ decreases somewhat, because of hybridization of atomic states and delocalization of electrons [40].

SrRuO₃ is a half-metallic ferromagnet [41–46]. It exhibits an orthorhombic phase at lower temperatures [47], becomes tetragonal at 820 K, and cubic at 950 K [40]. In contrast to the phase transition in LaMnO₃, the sequence of structures characterizing the phase transitions in SrRuO₃ is more standard for perovskite structures. The low temperature, orthorhombic phase is dictated by structural parameters (Goldschmidt tolerance factors) that trigger octahedral tilting. Each phase transition removes one or more of these tilts, which increases the symmetry of the crystal.

In this work we show that, for these two, electronically very different, complex oxides, MLFFs can capture the physics of the structural phase transitions. The simulated phase transition temperatures depend on the U values chosen, demonstrating on the one hand the strong connection between structure and electron localization and, on the other hand, providing the possibility of assessing the value of U by means of the transition temperatures.

II. METHODS

Our MD simulations are divided into two stages, i.e., a first, training stage and a second, production stage. The first stage aims at the on-the-fly training of the MLFF, using the techniques described in Refs. [12,15] with parameter settings motivated by those studies, such as the use of a $2 \times 2 \times 2$ LaMnO₃ or SrRuO₃ supercell. The MLFF potential is constructed based on a variant of the GAP-SOAP method [8,10] with a Gaussian width of 0.5 Å. The local atomic configurations are described within a cutoff radius of 6 and 5 Å for the two- and three-body terms, respectively. To study the influence of the U value, a MLFF is generated for each of the values mentioned in the Results section. The regression results of all the learned models can be found in the Supplemental Material, Sec. S1 [48].

Collinear spin-polarized DFT calculations are carried out using VASP 6.3 [49,50], which applies the projector augmented-wave method [51] to calculate the electronic states. A 500 eV cutoff energy for the plane-wave basis and a Gaussian smearing of $\sigma = 0.05$ eV are used. A $2 \times 2 \times 2$ and a $4 \times 4 \times 4 \Gamma$ -centered *k*-point grid are used for LaMnO₃ and SrRuO₃, respectively. The density of states (DOS) of both compounds are calculated with a fully relaxed primitive orthorhombic unit cell and a $4 \times 4 \times 4 \Gamma$ -centered *k*-point grid.

To construct the MLFFs as described in [12] an explicit electronic structure calculation, in this case DFT + U, is done whenever the Bayesian uncertainty in the predicted forces by the MLFF is too large. The Perdew-Burke-Ernzerhof (PBE) generalized gradient approximation (GGA) is used for LaMnO₃ as part of the DFT + U description and PBEsol for SrRuO₃. Generally, PBEsol tends to give somewhat better lattice parameters than PBE, but we mainly based our choice of functionals on previous studies on these materials; see the next paragraph.

For LaMnO₃ a Hubbard term according to the rotationally invariant description of Dudarev *et al.* [52] is used, with $U_{\text{eff}} = U - J = 3.5 \text{ eV}$ for the Mn 3*d* shell, which is a choice guided by previous work [53]. An A-type antiferromagnetic configuration is used for the magnetic ordering in LaMnO₃ at T = 0. For SrRuO₃ a Hubbard term is introduced according to the DFT + U method of Liechtenstein *et al.* [54], with parameters U and J for the Ru 4*d* orbitals set to 2 eV and 0.6 eV, respectively, again a choice that is consistent with previous studies [24,45,46,55]. Here, a ferromagnetic ordering is set for T = 0.

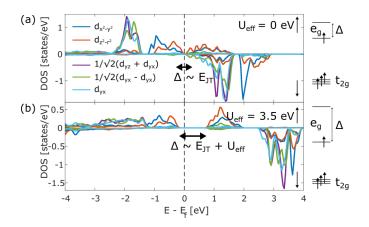


FIG. 1. Density of states (DOS) of LaMnO₃ in its ground state structure, projected on the *d* orbitals of a single Mn atom for (top) $U_{\text{eff}} = U - J = 0$ eV and (bottom) $U_{\text{eff}} = 3.5$ eV. The right-hand side shows schematically the splitting of the *d* orbitals due to the crystal field, including the effect of the Jahn-Teller (JT) distortion.

In the second stage, after the MLFFs have acquired sufficient accuracy, production runs are made using $6 \times 6 \times 6$ supercells, where the forces are calculated solely with the MLFFs, without resorting to DFT + U calculations. The isothermal-isobaric MD simulations with time steps of 0.5 fs are performed with a Langevin thermostat and, in calculating the phase transitions, the temperature is varied from 100 K to 1100 K at a rate of 0.5 K/ps under constant atmospheric pressure.

III. RESULTS

A. LaMnO₃

The basic electronic structure of LaMnO₃ in its ground state structure, resulting from a DFT + U calculation, is shown in Fig. 1. The octahedral coordination of Mn gives the standard $t_{2g}-e_g$ split d orbitals, where for the Mn d^4 configuration the lower t_{2g} states are half filled and the upper e_g states contain one electron, resulting in a high spin state, with a magnetic moment $M = 4 \mu_B$ per Mn atom and antiferromagnetic ordering in the ground state of moments on different Mn atoms.

The singly occupied e_g states give rise to a Jahn-Teller (JT) distortion and concomitant orbital ordering, which has a marked influence on the density of states (DOS), Fig. 1. In a calculation without Hubbard terms, $U_{\text{eff}} = 0$ eV, the JT distortion is small even at T = 0, giving a difference in Mn-O bond lengths of ~0.07 Å only. This results in a small splitting of the e_g orbitals and a corresponding vanishing band gap Δ , whereas the experimental band gap is 1.2 eV [56].

Setting $U_{\text{eff}} = 3.5 \text{ eV}$ increases the JT distortion significantly, enlarging the difference in Mn-O bond lengths to ~ 0.3 Å. Correspondingly, the DFT + U band gap increases to 0.88 eV, which is much closer to the experimental value. At T = 0, these calculations give an orthorhombic structure for LaMnO₃, with optimized lattice constants a = 5.57, b = 5.90, and c = 7.73 Å. The experimental low temperature structure is indeed orthorhombic, with lattice parameters at T = 300 K of a = 5.54, b = 5.74, and c = 7.69 Å [39]. The

overestimation of the lattice parameters by the DFT calculation is a common consequence of the PBE functional.

The size of the band gap is coupled to that of the JT distortion and both are affected by the effective localization of the electrons in the Mn d orbitals, as controlled by the value of U. It should be noted, however, that when directly comparing with experimental data, calculating band gaps with small unit cells may give slightly misleading results and supercells that have a substantial number of local structural and magnetic degrees of freedom may be required [57,58].

A MD simulation using the MLFF, while increasing the temperature linearly, allows us to study the structural transition of LaMnO₃. Figure 2(a) shows the lattice parameters of LaMnO₃ as function of temperature. How the orthorhombic cell parameters are extracted from the $6 \times 6 \times 6$ supercell simulations is explained in the Supplemental Material, Sec. S2 [48]. At low temperature, the MD simulations clearly maintain an orthorhombic structure for LaMnO₃, which is in agreement with experiment. Every tenth MD step is plotted in Fig. 2(a) and a moving average of 500 steps is used to show the trend. In the entire temperature range the lattice vectors remain perpendicular to each other.

The MD calculations preserve the orthorhombic (Ort) phase in the temperature range $T \lesssim 700$ K and give a transition to a metric cubic phase (Cub) above 700 K, where the lattice parameters become equal, but the crystal structure does not adopt a cubic symmetry. Experimentally, a Ort-Cub phase transition is observed at T = 750 K [39]. In the MD run, an extremely sharp transition is not expected, due to the still relatively fast heating rate that is demanded for in the simulations [12,59]. The temperature at which the change of the lattice parameters as function of temperature is at a maximum [indicated by the gray bar at $T \approx 680$ K in Fig. 2(a)] is used to pinpoint the Ort-Cub phase transition temperature.

From the perspective of the lattice parameters, the global structure becomes cubic for temperatures above the transition temperature. However, the microscopic structure reveals that the symmetry of the crystal is still orthorhombic in the Cub phase, which is in agreement with experiment [39]. In the low temperature Ort phase the MnO_6 octahedra are significantly distorted and tilted, with all Mn-O-Mn bonds between neighboring octahedra close to 150° [or $360^{\circ} - 150^{\circ}$; see Fig. 2(e)]. In the high temperature Cub phase, the tilting persists, with angles close to 155°; see Fig. 2(b). For a description of the averaging procedure used to obtain the tilting angles see the Supplemental Material, Sec. S3 [48]. At higher temperatures, the distribution of angles around 155° broadens. In addition, a secondary distribution appears at $(360 - 155)^{\circ}$, which is an indication of octahedra flipping over their tilt; see the Supplemental Material, Sec. S3 [48].

The microscopic structural changes in the Orb-Cub phase transition can also be monitored by inspecting the Mn-O bond lengths during the MD run; see Fig. 2(c). The calculations predict a significant JT distortion for lower temperatures in all MnO₆ octahedra, where Mn-O bond lengths can be divided into three classes, i.e., a long (\sim 2.27 Å), medium (\sim 2.00 Å), and short (\sim 1.94 Å) bond. This JT distortion is almost completely suppressed in each octahedron for

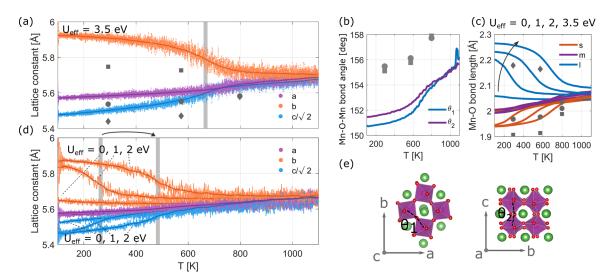


FIG. 2. Results of MD simulations on LaMnO₃, using machine learning force fields. (a) Lattice parameters as function of temperature, sampled at every tenth MD step (5 fs). The solid line is a moving average of 500 steps. The symbols •, \blacksquare , and \blacklozenge indicate the experimental *a*, *b*, and *c* lattice parameters, respectively, taken from Ref. [39]. (b) Moving average of bond angles as function of temperature; the symbols indicate the experimental data. (c) Moving average of bond length as function of temperature for $U_{\text{eff}} = 1, 2, 3, \text{ and } 3.5 \text{ eV}$. The symbols indicate the experimental data. (d) Lattice parameters as function of temperature for $U_{\text{eff}} = 0, 1, \text{ and } 2 \text{ eV}$. (e) Definitions of the bond angles plotted in (b).

 $T \gtrsim 800$ K, in agreement with the experiment. Nevertheless, as described above, the tilting of octahedra persists even at higher temperatures.

The structural phase transition can be linked to the influence of electron localization by evaluating this transition for different values of U_{eff} . Figure 2(d) shows the lattice parameters as function of temperature, characterizing the Ort-Cub phase transition, as obtained from MD runs with MLFFs, trained for DFT calculations with different settings of the Hubbard parameter, $U_{\text{eff}} = 0$, 1, and 2 eV, respectively. Ignoring the on-site Hubbard term, $U_{\text{eff}} = 0$, the orthorhombic distortion is severely underestimated already at a very low temperature and there is no clear transition to a different phase at any temperature.

If one increases the on-site repulsion $U_{\rm eff}$ to 1 and 2 eV, the orthorhombic distortion at low temperature becomes much more pronounced. Moreover, as indicated by the gray bars in Fig. 2(d), the transition temperature for the Ort-Cub phase transition increases with increasing $U_{\rm eff}$, from 270 K, 490 K, to 680 K for $U_{\rm eff}$ to 1, 2, and 3.5 eV, respectively. Comparing this to experiment [39], the latter value thus seems the most reasonable. Figure 2(c) shows that the size of the JT distortion at low temperature increases monotonically with increasing $U_{\rm eff}$ and that the Ort-Cub transition at higher temperature is accompanied by a suppression of the JT distortion for all values of $U_{\rm eff}$.

B. SrRuO₃

Although in both cases the transition metal ion has a d^4 configuration, the basic electronic structure of SrRuO₃ differs markedly from that of LaMnO₃; compare Figs. 3 and 1. Setting the Coulomb and exchange parameters U, J = 2.6, 0.6 eV, as in previous calculations [24,45,46,55], the Ru ions adopt a low spin state, with the four *d* electrons occupying t_{2g} states. The calculated magnetic moments on the

Ru atoms are $M = 1.44 \mu_{\rm B}$. There is no occupation of the e_g orbitals and no JT distortion. SrRuO₃ is a half-metallic ferromagnet in its ground state; see Fig. 3(b).

In contrast, neglecting the on-site terms, U, J = 0, 0 eV, in the calculations, SrRuO₃ becomes a standard metallic ferromagnet. The e_g states remain unoccupied, but in addition the majority spin t_{2g} states are no longer fully occupied; see Fig. 3(a). Correspondingly, the calculated magnetic moments $(M = 1.30 \mu_B)$ are smaller than in the case discussed in the previous paragraph.

Like LaMnO₃, SrRuO₃ shows structural phase transitions as a function of temperature, although quantitatively the structural changes are smaller than those in LaMnO₃. The low temperature structure of SrRuO₃ is characterized by a relative tilting of RuO₆ octahedra, mainly driven by geometric

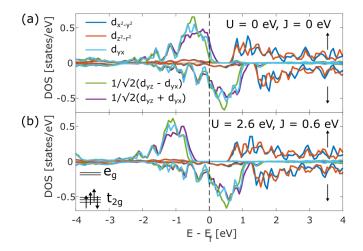


FIG. 3. Density of states (DOS) of SrRuO₃ projected on the individual *d* orbitals of Ru for U = J = 0 eV and U = 3.5 and J = 0.6 eV.

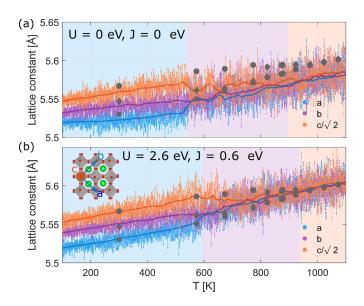


FIG. 4. Results of MD calculations on SrRuO₃, using machine learning force fields, for the cases (a) U = J = 0 eV and (b) U = 2.6, J = 0.6 eV. Lattice constants as function of temperature. The solid lines are a moving average of 500 MD steps. The symbols •, **a**, and \blacklozenge indicate the experimental *a*, *b*, and *c* lattice parameters, respectively, taken from [40,47].

considerations (Goldschmidt tolerance factors). Experimentally, for T < 600 K, SrRuO₃ adopts an orthorhombic phase. In the temperature range 600 K < T < 900 K, the structure is tetragonal and for T > 900 K the structure becomes cubic [40,47].

This sequence of phases, which is not uncommon in perovskites, is determined by a gradual removal of octahedral tilting. The orthorhombic phase is characterized by two tilt angles and is labeled $b^-b^-a^+$ in Glazer notation [40]. The tetragonal phase is a structure with one tilt angle, labeled $a^0a^0c^-$, whereas all tilting disappears in the cubic structure. For the Pb-halide based perovskites, similar subtle structural changes and the related phase transitions have been shown to be well described by MLFFs [13]. The question here is whether the U, J terms play a role in these phase transitions and give the right transition temperatures.

Figure 4 shows the evolution of the lattice constants of SrRuO₃ as function of temperature for the cases U, J = 0, 0 and U, J = 2.6, 0.6 eV, as calculated with MLFFs, using the same procedure as for LaMnO₃. For the case of U, J = 2.6, 0.6 eV the calculations show a clear orthorhombic phase for T < 600 K, a clear cubic phase is present for T > 900 K, and for 600 K < T < 900 K the structure is tetragonal, in agreement with experiment. Moreover, also quantitatively the lattice parameters are in good agreement with experimental results [40,47].

For U, J = 0, 0 eV the agreement with experiment is less gratifying. The calculated lattice parameters are too small over the whole temperature range and the temperatures at which the phase transitions occur are underestimated by ~100 K. The effect of including U, J terms on the finite temperature behavior of SrRuO₃ is less dramatic than in the case of LaMnO₃. Nevertheless, also in SrRuO₃, including on-site electron-electron interactions is important to obtain quantitative agreement with experiment.

IV. DISCUSSION

The MD simulation with the MLFF predicts the sequence of structural phases of LaMnO₃ as a function of temperature in good agreement with experiment, provided the MLFF is trained on-the-fly in an *ab initio* MD simulation based upon a DFT + U description of the electronic structure. The on-site Coulomb interaction U on the Mn ions plays an essential role in improving the description of the finite temperature structures over those given by regular DFT functionals (such as PBE).

The low temperature phase is dominated by a JT distortion of the MnO_6 octahedra, which dictates orbital ordering resulting in an orthorhombic structure. Using only the PBE functional, the ground state of LaMnO₃ is still described correctly as an antiferromagnetic insulator. However, the JT distortion is severely underestimated, as are the band gap and the phase transition temperature.

Upon introducing U, the JT distortion is enlarged to a quantitatively realistic value. One of the consequences of the on-site terms in the DFT + U formalism is to more localize the d electrons on the Mn ions in comparison to a standard DFT GGA functional, thereby amplifying the effects of such a localization, such as the size of the JT distortion of the MnO₆ octahedra and its consequences for the finite temperature behavior of LaMnO₃ as discussed above.

The size of the effects scale with the size of U. For U = 3.5 eV, the structure of the initial orthorhombic phase and the transition to the metric cubic phase around 750 K are reasonably well described. This structural transition is accompanied by a suppression of the JT distortion on each MnO₆ octahedron. We conclude that the JT distortion is the driving force behind the dissimilar lattice parameters at lower temperatures.

The temperature at which the JT is suppressed, and thus the lattice parameters become equal, is proportional to the electronic energy gained by breaking the symmetry of the lattice in the form of a JT distortion. This energy becomes larger with increasing U [60–62]. So, it is reasonable then that, if U is made smaller, the JT distortion, the electronic band gap, and the Ort-Cub transition temperature all become smaller. The value U = 3.5 eV gives a transition temperature that is quite close to the experimental value.

We have used the simplest formulation of the DFT + U method in the calculations on LaMnO₃ [52] and, although the structural phase transition temperature is well predicted, there may be room for improvement, such as orbital dependent Coulomb U parameters, the introduction of exchange J parameters, or by also applying U terms to the ligand orbitals [63–65]. Other possibilities can be found in the use of spatially dependent values of U [66,67] or in the introduction of intersite terms [68].

Also for SrRuO₃ the MD simulation with the MLFF predicts the sequence of structural phases well as function of temperature. The number of d electrons on the Ru ions is formally the same as on the Mn ions in LaMnO₃, but the effects of on-site U and J terms are much smaller. There is no JT distortion of the RuO₆ octahedra and SrRuO₃ is a (ferromagnetic) metal. SrRuO₃ has an orthorhombic structure at low temperature, a cubic structure at high temperature, and a tetragonal structure at intermediate temperatures. As in many perovskites, this sequence of phases is dictated by geometric structural parameters.

The transitions between these phases are correctly described by the MLFF. Using the parameters U, J = 2.6, 0.6 eV gives excellent agreement with experiment of the lattice parameters as a function of temperature. In contrast, using U, J = 0, 0 eV, the agreement becomes much less impressive and, in particular, the temperatures at which the phase transitions are predicted to occur are too low. We conclude therefore that also in the case of SrRuO₃ it is important to include explicit on-site electron terms to correctly describe the finite temperature behavior of the material.

In addition, a clear difference in the T = 0 K band structure is observed between the DFT and DFT + U method. DFT predicts SrRuO₃ to be a conventional metallic ferromagnet. In contrast, the DFT + U method states that SrRuO₃ is a halfmetallic ferromagnet, as reported earlier by other calculations [45,69] and suggested by experiments [70–72].

In the calculations discussed in this paper, we have used relatively simple DFT + U methods, which of course have their limitations [21,67,73–75]. In particular, one could mention the use of a single U parameter [76–79], the PAW projector functions used to define the on-site terms [80–84], and treating the Coulomb and exchange U and J as adaptable parameters [80,85–87]. In principle, such parameters can be calculated from first principles [27,81,88]. One may then consider finite temperature simulations with MLFFs to be a consistency check on the values of such parameters.

In principle, it is possible to go beyond DFT + U and base a MLFF on first-principles calculations with a more elaborate description of the electronic structure, such as, for instance, hybrid (range-separated) functionals [85,89,90] or the random phase approximation (RPA) [91–94]. However, these methods are computationally much more expensive.

The MD runs in the training phase are not sufficiently accurate to capture the magnetic phase transitions. For instance, the Néel temperature of LaMnO₃ is 140 K, whereas the AFM ordering persists up to \sim 500 K. It should be noted, however, that only in the training phase of the force field, where one has access to the DFT electronic structure, is the full information on the magnetic ordering available. The training phase uses a small supercell and it is quite conceivable that this supercell is too small to properly describe a magnetic phase transition.

Nevertheless, the initial magnetic properties of $LaMnO_3$ and $SrRuO_3$ are important to the structures and the structural phase transitions. At low temperature, $LaMnO_3$ and $SrRuO_3$ have an AFM and FM ordering of magnetic moments on the transition metal ions, respectively, and the MLFFs have been trained starting from these ground states. Starting from an incorrect ground state to train a MLFF, FM ordering in the case of LaMnO₃, for instance, gives a temperature for the phase transition that is significantly too low and starting from an unpolarized ground state does not give a phase transition at all; see the Supplemental Material, Sec. S4 [48]. There is an intimate connection between the electronic structure and the magnetic ordering and a correct electronic structure at low temperature in the training phase is required for capturing the structural phase transitions in the production phase correctly.

V. CONCLUSION

Complex oxides are frequently modeled with DFT + U, where the U should capture the physics of the complex interplay between crystal structure, magnetism, and electron localization. Here we use the new on-the-fly machine learning force field approach to train ML potentials by DFT + Uon two archetypal perovskite oxide materials, LaMnO3 and SrRuO₃, with transition metals in a d^4 configuration. For the antiferromagnetic insulator LaMnO₃, with the correct value for U and antiferromagnetic ordering, the model is able to correctly predict the structural phase transition and the suppression of the JT distortion around 750 K. Likewise, for the ferromagnetic metal SrRuO₃, the structural phase transitions simulated with the MLFF are in good agreement with experiment. We show that the physics of complex oxides can be captured sufficiently well by DFT + U to predict these phase transitions. Furthermore, we demonstrate how the crystal structure at finite temperatures depend on the parameter U. This suggests that, by comparing the experimental crystal structure data with simulations, active learning of ML interatomic potentials could serve as an additional approach to assess values for U.

Research data for this paper has been made available through a data set in the 4TU.ResearchData repository; see Ref. [95]. The following data is stored. (i) The electronic structure databases, including structures and corresponding DFT + U energies, forces, and stress tensors, used to train the MLFFs (ML_AB files). (ii) A high level analysis of the electronic structure databases presented by pdf fact sheets generated with open-source FPDATAVIEWER software [96]. (iii) VASP input files (INCAR, KPOINTS) corresponding to the on-thefly MLFF generation.

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