# Orientational ordering of water molecules confined in beryl: A theoretical study

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We present an improved model for studying the interactions between dipole moments of water molecules confined in beryl crystals inspired by recent NMR experiments. Our model is based on a local crystal potential with dihexagonal symmetry for the rotations of water dipole moments, leading to deflection from the ab hexagonal crystallographic plane. This potential shape has significant implications for dipole ordering, which is linked to the nonzero projection of the dipole moment on the hexagonal c axis. To reveal the tendency toward equilibrium-ordered states, we used a variational mean-field approximation, Monte Carlo simulations, and quantum tunneling. Our analysis reveals three types of equilibrium-ordered states: a purely planar dipole order with an antiparallel arrangement in the adjacent ab planes, a configuration with deflected dipole moments ordered in antiparallel directions, and a helical structure of the dipoles twisting along the c axis.

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## I. INTRODUCTION

Water molecules confined in various types of environments exhibit properties substantially different from those of common water phases [1-5]. In most of the confined geometries, these properties are a result of competition between the energy minimization of the hydrogen-bonded network in water and the constraints due to the confinement, such as interactions with the cavity surface, possible quantum coherence, or the necessity to fit within the limited volume. The properties of the usual condensed phases of water are determined, to a large extent, by the ubiquitous network of hydrogen bonds. Among the known confined geometries, the opposite extreme case in terms of hydrogen bonding is represented by crystals containing single molecules of water, where hydrogen bonding among the water molecules cannot occur because the crystal lattice mutually separates them. Examples of such crystals include gypsum [6-9], bassanite [7-9], and cordierite [8,10–14], minerals in which low-temperature ordering has been detected. Furthermore, hydrated beryl (Be<sub>3</sub>Al<sub>2</sub>Si<sub>6</sub>O<sub>18</sub>) is a particular system of this kind that came into the focus of researchers more than 50 years ago [15–17] and it has been studied until now [12,18–30]. In particular, the interest in water confined in beryl intensified after revelations of tendencies of the water dipoles to order anti/ferroelectrically; this ordering develops at low temperatures with the polarization vectors in the *ab* plane, i.e., perpendicular to he crystallographic *c* axis [19,22].

The crystal structure of beryl exhibits hexagonal symmetry (space group P6/mcc), and it contains linear channels of cavities parallel to its hexagonal axis. Each cavity, slightly larger than 5 Å in diameter [22], may host one water molecule. Following the infrared spectroscopic study of Wood and Nassau [16], the water molecules have been assumed to stay in one of

two orientations with respect to the crystal lattice. According to their hypothesis, the molecules occur with the H-H lines oriented either parallel to the hexagonal c axis (designated as "type-I water") or perpendicular to it ("type-II water"). The occurrence of the latter type was assigned to crystals with additional doping, as the oxygen atoms of the water molecules may bind to impurity atoms located within the channels. In the last decade, the incipient ferroelectricity of water molecules in beryl was demonstrated via spectroscopic measurements of their collective vibrations in a partially hydrated crystal, revealing a ferroelectric soft phonon mode. This soft mode was detected in the THz-range spectra of dielectric permittivity, and its parameters were shown to obey the usual Curie-Weiss, and Cochran temperature dependence [22]. In that work, a negative Curie temperature of  $T_{\rm C} \approx -20 \,\rm K$  was determined from the temperature dependence of permittivity, implying that the ferroelectric state could not be reached. Notably, at temperatures lower than about 20 K, the soft phonon vibrational frequency remained almost constant, which was explained by quantum tunneling of the water molecules between equivalent potential minima [20,21].

Most of the theoretical studies on beryl focused on the interactions between the dipole moments of the water molecules, and the models of their collective dynamics assumed that the type-I molecules can rotate around the hexagonal axis of beryl so that the H-H lines remain parallel to the hexagonal axis. At the same time, it was also supposed [18,20,21,23-25] that the molecules are subjected, in their angular orientations, to a local potential exhibiting six equivalent minima separated by an angle of 60°. A recent NMR study of hydrated and deuterated beryl [31] provided new data on the orientations and dynamics of the dipoles of water molecules in beryl. These results show that the traditional view of the molecules' orientations have to be revisited. The numbers and positions of the NMR lines observed are not compatible with the earlier view assuming that the H-H lines of the molecules are parallel to the hexagonal axis. Instead, their mutual

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FIG. 1. (a) Water molecule confined in a void formed by the crystal structure of beryl [17]. Dark blue: SiO<sub>4</sub> tetrahedra, pale blue: AlO<sub>6</sub> octahedra. The thin black line marks a libration axis of the water molecule owing to a transiently formed hydrogen bond between one of the hydrogens and one of the nearest 12 oxygen atoms of the void. The structural data were visualized using the VESTA software [33]. (b) Scheme of the void with 12 possible equilibrium orientations of the molecule. From the beryl structure, only the 12 O atoms nearest to the void center are drawn (red). When the water molecule is constrained to one of the planes shown in light blue, an H atom binds to one of the O atoms in the lower hexagon (dotted line); then, the H-H line (dashed) intersects the c crystal axis above the central O atom and the dipole moment has an upward component along the c axis. The opposite occurs if the H atoms are constrained to the planes shown in light green. The same color coding is used in Fig. 2.

angle deduced from these experiments amounts to about  $\pm 18^{\circ}$  (see Fig. 1). This suggests that hydrogen bonds are formed between one of the hydrogen atoms of water [labeled H<sub>2</sub> in Fig. 1(a)] and any of the 12 nearest-neighboring oxygen atoms forming the cavity, of which 6 are in the upper crystal plane and 6 are in the lower one. The hydrogen bonding mechanism appears plausible, as the distance between any of the 12 O atoms and the center of the void is about 3.43 Å. This is slightly more than the known donor-acceptor distances in other hydrogen-bonded compounds [32]; we note, however, the bond length is generally not a constant, and it increases with decreasing bond energy. Also, in the beryl voids, the water molecules may be slightly off centered owing to the attractive force of the bonds, which would imply shorter donor-acceptor distances than the above value.

The local atomic arrangements of the voids lack the mirror symmetry with respect to the plane perpendicular to the hexagonal axis; consequently, the potential minima from the oxygens atoms in the upper and lower levels surrounding the water molecule are not aligned along the c axis, but rather alternately angularly tilted [see Fig. 1(b)]. The observed temperature dependence of the NMR lines strongly supports a model in which the hydrogen bonds are present, with the bonded proton of the confined water molecule forming a rotation axis around which the other free hydrogen librates near its equilibrium position. It is known from the structural data [17] that the cavity centers are surrounded by 12 oxygen atoms located at equal distances from the centers. These atoms are distributed in two layers parallel to the *ab* planes, namely, at the vertices of two hexagons [see Fig. 1(a)]. Consequently, each water molecule's possible equilibrium dipole positions within its beryl cavity must equal 12.

The aim of this paper is to propose a model of the water molecules in the cavities of the beryl crystal with a confining potential showing dihexagonal symmetry of 12 local minima for the equilibrium positions for their dipole moments. We show that the presence of 12 energetically equivalent orientations of the water molecules has unexpected consequences. Various ordering tendencies along the crystallographic c axis emerge, leading to a wider variety of dipole states than assumed up to now. In particular, we find that the ground state at low temperatures is linked to a chiral ordering along the c axis. The experimental findings indicate that the existing models of water molecules confined in the cavities of the beryl crystal with a sixfold potential degeneracy still need to be completed, revisited, and extended to be compatible with the latest observed behavior. Our model is the first step in this direction.

### **II. MICROSCOPIC MODEL**

The water molecules are confined in cavities in the beryl crystal with an overall hexagonal symmetry along the c axis (also called "vertical direction" in the following). The center of mass of the water molecules is positioned at the symmetry axis, and their fluctuations can be neglected. The protons of the hydrogen atoms in the water molecules are attracted to the oxygens of the lattice ions, forming a nanocavity for the water molecule. Recent NMR experiments [31] show that in contrast with the widespread assumption of previous theoretical works, the H-H lines of the water molecules are tilted with respect to the c axis. To set up a model potential for the water molecules' reorientations, see Fig. 2; we took advantage of this finding together with the data on the local crystal arrangement [17] showing that six oxygen atoms are positioned slightly above the center of the cavity and six below. The two hexagons are not mirrored by the *ab* plane but angularly turned by  $30^{\circ}$ , forming a perfect dihexagonal spatial structure confining the water molecule in its center of origin. The positions of the oxygen hexagons, probably forming transient hydrogen bonds to the molecules, are at the origin of the angular deflection of the dipole moment of water I from the *ab* plane. The dipole moment can be directed towards one of the oxygens either above or below the center of mass of the water molecule. The deflection angle does not change much with temperature and its experimental value  $\theta \approx 18^{\circ}$  [31] was taken as fixed in the model description. We hence assume dynamics only in the orbital angle  $\phi$  within the *ab* plane.

The dipole moment of the water molecule in the cavity with only rotational degrees of freedom can be described by a quantum-mechanical Hamiltonian

$$H^{\rm loc} = \frac{1}{2I}L^2 + V^{\rm loc},\tag{1}$$

where *L* is its angular momentum, *I* its moment of inertia, and  $V^{\text{loc}}$  is a local potential due to the surrounding walls. We assume that angle  $\phi$  in the *ab* plane orthogonal to the symmetry axis fully describes the rotational degrees of freedom. We neglect the dipole dynamics out of the plane since the angle of deflection from the plane is small, and the energy connected with the vertical movements is negligible.

The local potential of the cavity has six minima for each oxygen hexagon. Since the single water molecule is attracted alternately to two hexagons, we can represent the local



FIG. 2. Scheme of angular reorientations of the water molecule in the beryl cavity. Red circles denote oxygen atoms of both the beryl structure (upper and lower rows) and the water molecule (central area), whereas small blue and green circles stand for hydrogen nuclei. These are attracted by the potential of the cavity (wavy lines; note the opposite signs of the energy scales) which attracts either of the hydrogen nuclei such that a hydrogen bond is formed (dashed lines). The water molecule represented by filled circles corresponds to one orientational configuration of the hydrogen nuclei among the 12 possible ones. The unbound nucleus is partially free, exhibiting librations (angle  $\alpha$ ) around its equilibrium position. Reorientations of the molecule (dotted arrows) may occur in two ways, either via tunneling of the nucleus through the potential barrier, or when an oxygen atom from the opposite side of the cavity traps the librating nucleus, breaking the previous hydrogen bond and freeing the bound nucleus from its previous position. The horizontal axis does not relate to the water molecule's oxygen atom fixed in the cavity center.

potential as a  $2 \times 2$  matrix

$$V^{\text{loc}}(\phi) = \frac{V_b}{2} \begin{pmatrix} 1 + \cos 6\phi & \omega/V_b \\ \omega/V_b & 1 - \cos 6\phi \end{pmatrix}, \quad (2)$$

where  $V_b$  is the potential barrier as in the standard sixfold model and  $\omega$  is the amplitude of the transition of the hydrogen bond from the upper to the lower nearest oxygen. As the lower and upper limits of the potential barrier, we take the values of Ref. [21],  $V_{b1} = 56$  meV and  $V_{b2} = 176$  meV. We will set the value of  $\omega$  self-consistently (see Sec. III B).

The water dipole moments interact via a binary dipoledipole interaction, and the full lattice Hamiltonian reads as

$$H = \sum_{i} H_{i}^{\text{loc}} + \frac{1}{4\pi\epsilon_{0}\epsilon_{r}r_{0}^{3}} \frac{1}{2} \sum_{i\neq j} \sum_{\alpha,\beta} p_{i}^{\alpha} D_{ij}^{\alpha\beta} p_{j}^{\beta} - \sum_{i,\alpha} E_{i}^{\alpha} p_{i}^{\alpha},$$
(3a)

where  $D_{ii}^{\alpha\beta}$  is the dipolar kernel

$$D_{ij}^{\alpha\beta} = \frac{\delta^{\alpha\beta}}{r_{ij}^3} - 3\frac{r_{ij}^{\alpha}r_{ij}^{\beta}}{r_{ij}^5}$$
(3b)

with dimensionless intermolecular distances  $r_{ij}$  (dependent on the lattice structure),  $p_i^{\alpha}$  is a dipole vector of molecule *i*, and **E**<sub>*i*</sub> is the electric field at site *i*. For further calculations, we define an interaction constant  $\bar{J} = p_0^2/(4\pi\epsilon_0\epsilon_r r_0^3)$ , where  $\epsilon_r = 7$  is the background permittivity [22],  $p_0 = 1.85$  D is the magnitude of dipole moment of the water molecule, and  $r_0 = 9.2$  Å is the distance between the nearest dipoles in the *ab* plane (that is  $r_0/2$  in the *c* direction).

## **III. SINGLE WATER MOLECULE**

Before approaching the full lattice problem, we resolve the local problem of a single water molecule in the beryl cavity.

## A. Ground-state energies

The cavity potential (see Fig. 2) has two sixfold-degenerate minima. The eigenvalue problem for the local Hamiltonian  $H^{\text{loc}}$  from Eq. (1) leads to 12 ground-state energies. They are obtained from the equation

$$\left(\frac{-\hbar^2}{2I} + V^{\rm loc}(\phi)\right) \begin{pmatrix} \psi_+(\phi)\\ \psi_-(\phi) \end{pmatrix} = E \begin{pmatrix} \psi_+(\phi)\\ \psi_-(\phi) \end{pmatrix}$$
(4)

with periodic boundary conditions for the planar angle  $\phi$ . For the rotational constant  $\hbar^2/2I$  of water molecule we choose value 3.04 meV [21]. We numerically solve the eigenvalue equations with a small value  $\omega/V_b \ll 1$  to obtain the eigenenergies' explicit  $\omega$  dependence. We end up with seven different lowest-energy states, five doubly degenerate. Their explicit values in the interval between  $V_{b1} = 56$  meV and  $V_{b2} = 176$  meV, with  $\omega$  dependence, are

$$E_0 = 24.5 - 26.3(\omega/V_b),$$
  

$$E_{1,2} = 27.16 - 25.7(\omega/V_b),$$
  

$$E_3 = 24.5 + 26.3(\omega/V_b),$$
  

$$E_{4,5} = 27.16 + 25.7(\omega/V_b),$$
  

$$E_{6,7} = 34.4 - 22.0(\omega/V_b),$$
  

$$E_{8,9} = 34.4 + 22.0(\omega/V_b),$$
  

$$E_{10,11} = 40.5.$$

The eigenenergies depend on  $V_b$  and are given in meV units. The boundary values correspond to the extreme values  $V_{b1}$  and  $V_{b2}$ . These energies determine the phase space for a single dipole moment. The seven quantum energies for  $\omega = \omega_1$  are plotted in Fig. 3 together with the local potential  $V^{\text{loc}}(\phi)$  with six minima from which we can assess the impact of quantum dynamics.

## B. Dynamics: Discrete clock model

We simplify the dynamics of the dipole within the cavity by explicitly considering only discrete amplitudes of transitions between nearest-neighbor local minima, instead of a continuous movement. We introduce three energy parameters,  $e_0$ , a, and d, which correspond to the ground-state energy and hopping to the locally nearest and next-nearest minima of the sixfold-degenerate local potential  $V^{\text{loc}}$ . Note that the parameter a represents the amplitude for reversing the deflection of the dipole moment from the ab plane in the direction of the caxis. The parameter d describes the hopping between two adjacent minima without changing the orientation of the dipole



FIG. 3. The lowest 12 energy levels of the clock model with the potential barrier  $V_{b1} = 56 \text{ meV}$  and the transition amplitude between the layers  $\omega = 8.2 \text{ meV}$ .

moment along the c axis. The Hamiltonian of the local dipole dynamics reduces the position representation to a matrix

	0	а	d	0	0	0	0	0	0	0	d	a	
$\hat{h}^{loc} = e_0 \hat{I} -$	а	0	а	d	0	0	0	0	0	0	0	d	
	d	а	0	а	d	0	0	0	0	0	0	0	
	0	d	а	0	а	d	0	0	0	0	0	0	
	0	0	d	а	0	а	d	0	0	0	0	0	
	0	0	0	d	а	0	а	d	0	0	0	0	
	0	0	0	0	d	а	0	а	d	0	0	0	,
	0	0	0	0	0	d	а	0	а	d	0	0	
	0	0	0	0	0	0	d	а	0	а	d	0	
	0	0	0	0	0	0	0	d	а	0	а	d	
	d	0	0	0	0	0	0	0	d	а	0	а	
	a	d	0	0	0	0	0	0	0	d	а	0)	
	`											(6	5)

where  $\hat{I}$  is the diagonal matrix. The three model parameters lead to the following eigenvalues of the above matrix:  $\bar{E}_0 =$  $-2a - 2d + e_0$ ,  $\bar{E}_3 = 2a - 2d + e_0$ ,  $\bar{E}_{1,2} = \sqrt{3}a - d + e_0$ ,  $\bar{E}_{4,5} = \sqrt{3}a - d + e_0$ ,  $\bar{E}_{6,7} = -a + d + e_0$ ,  $\bar{E}_{8,9} = a + d + e_0$ ,  $\bar{E}_{10,11} = 2d + e_0$ . These eigenenergies should equal the ground-state energies obtained in the preceding subsection. There are generally seven independent parameters in matrix (6). However, including the full set of eigenenergies would make the solution cumbersome and difficult to control. Therefore, in our simplified matrix, we consider only three eigenenergies: the lowest energy  $E_0$ , the highest energy  $E_{10,11}$ , and the energy  $E_3$  of the singlet state. It is important to note that this reduction does not change the result qualitatively. The equations determining  $e_0$ , a, d then read as

$$-2a - 2d + e_0 = E_0, \tag{7a}$$

$$2d + e_0 = E_{10,11},\tag{7b}$$

$$2a - 2d + e_0 = E_3. \tag{7c}$$

The solution of these equations is  $e_0 \simeq 32.2 \text{ meV}$ ,  $d = 4 \text{ meV} - 0.002 \,\omega^2$ , and  $a = 0.23 \,\omega$ . We assume that the probability of jumps between the upper and lower sites is the same as between the left and right sites, which requires choosing d = 2a. From this condition, we obtain the value of  $\omega$ . As a result, we get these parameter values for the lower bound  $V_{b1} = 56 \text{ meV}$ ,  $\omega_1 = 8.2 \text{ meV}$ ,  $d_1 = 3.9 \text{ meV}$ ,  $a_1 = 1.9 \text{ meV}$  and

for the upper bound  $V_{b2} = 176 \text{ meV}$ ,  $\omega_2 = 3.9 \text{ meV}$ ,  $d_2 = 1.26 \text{ meV}$ ,  $a_2 = 0.63 \text{ meV}$ .

We use these values to characterize the local parameters of individual dipoles in the description of the water molecules in the extended beryl crystal with nonlocal dipole-dipole interaction from Hamiltonian in Eq. (3).

## IV. INTERACTING WATER DIPOLES IN BERYL CRYSTAL

## A. Ground state

Having resolved the behavior of a single water molecule in the beryl cavity, we extend the description to the behavior of water molecules in the extended beryl crystal. We assume a perfect homogeneity and neglect the dilution of water molecules. We describe the model within a tight-binding approximation with each unit cell containing one water molecule, the local behavior of which will be approximated by the clock model with 12-fold degenerate ground-state energy. We first neglect quantum tunneling between the minima. The rotation angle  $\phi$  will describe the positions of the potential minima in the *ab* plane. The  $n_S = 12$  values  $\phi_i \in \{\frac{\pi}{6}n_i\}_{n_i=0}^{n_S-1}$  are the coordinates of the minima. The three-dimensional vector of the dipole moment **p** in the minima deflects from the *ab* plane by an angle  $\theta$ :

$$\mathbf{p}_i = p_0 \cos(\theta) \boldsymbol{\sigma}_i, \tag{8}$$

where  $p_0$  is the magnitude of the dipole moment of the water molecule. Here we introduced a new dimensionless vector variable

$$\boldsymbol{\sigma}_{i} = \left(\cos\frac{2\pi n_{i}}{n_{S}}, \sin\frac{2\pi n_{i}}{n_{S}}, (-1)^{z+n_{i}} \tan\theta\right), \qquad (9)$$

where z denotes the coordinate along the crystallographic c axis. Index i should be considered a three-dimensional integer variable,  $i = \{x, y, z\}$  along axes a, b, c, respectively. The interacting part of the full Hamiltonian can now be represented in the form of the classical Heisenberg spin model with spin vector variables  $\sigma_i$ :

$$H_{I} = \frac{J}{2} \sum_{i \neq j} \boldsymbol{\sigma}_{i} \cdot \mathbf{D}_{ij} \cdot \boldsymbol{\sigma}_{j} - p_{\perp} \sum_{i} \mathbf{E}_{i} \cdot \boldsymbol{\sigma}_{i}, \qquad (10)$$

where  $J \equiv \overline{J} \cos^2 \theta = p_0^2 \cos^2 \theta / (4\pi \epsilon_0 \epsilon_r r_0^3)$  and  $p_{\perp} = p_0 \cos \theta$ . The deflection angle  $\theta = 18.1^{\circ}$  is the same for all dipoles.

Finding the zero-temperature energy of any periodic configuration of the normalized polarization vectors  $\sigma_i$  is straightforward. Angle  $\theta$  is a free parameter. The ground-state energy is obtained from minimizing a binary form

$$\mathcal{E}(\{\boldsymbol{\sigma}_i\}) = J(\theta) \sum_j \boldsymbol{\sigma}_0 \cdot \mathbf{D}_{0j} \cdot \boldsymbol{\sigma}_j \tag{11}$$

with respect to the dipole configurations, depending on  $\theta$ . An ordered configuration gives the lowest energy. It is, however, anisotropic and depends on the deflection angle  $\theta$ . The dipole-dipole interaction drives the system to the saturated planar ferroelectric order within the *ab* planes. The order along the *c* axis is antiferroelectric for small deflection angles  $\theta < 8^{\circ}$  (see Fig. 4). The ground state displays a chiral structure for larger angles. In the chiral structure, the dipoles are ferroelectrically



FIG. 4. Comparison of energies of the two expected ground-state structures. Antiferroelectrically ordered ferroelectric planes vs a chiral structure with 12 positions along the crystallographic *c* axis as a function of the dipole deflection angle  $\theta$  from the *ab* plane are plotted. Energy is calculated in units where  $\bar{J} = 1$ .

ordered within each *ab* plane, but their angular coordinate  $\phi$  switches to the nearest minimum in one direction (either clockwise or anticlockwise) along the *z* coordinate until it goes through the whole circle. It means that 12 coupled states are periodically repeated along the symmetry axis. The unit cell of this helical structure is graphically presented in Fig. 5.

## **B.** Variational mean-field approximation

The ground-state analysis shows the correct way of selecting the unit cell of the periodic structure used in the



FIG. 5. Visualized chiral structure of the dipoles ordered along the *c* axis in the ground state. On varying the *z* coordinate, the dipoles go through all the 12 minima of the local potential either clockwise or anticlockwise. Left: four unit cells with the planar (*ab*) hexagonal structure. Right: a single unit cell with the dipole projections onto the *ab* plane (blue arrows) and *c* axis (green arrows).

thermodynamic description of the water molecules in the beryl crystal. We start with a mean-field description neglecting low-temperature quantum tunneling between the minima of the degenerate ground state. The unit cell will have an internal structure with 12 water molecules in the vertical direction. The vector order parameter, the thermally averaged value of the dipole moment  $\mathbf{p}(z) \sim \langle \boldsymbol{\sigma}_z \rangle$ , contains index  $z = 0, \ldots 11$  labeling the vertical coordinate within the unit cell conformal with the helical symmetry of the ground state.

We use the variational mean field and introduce a variational vector parameter  $\lambda_z$  with periodicity  $n_S$  ( $\lambda_z = \lambda_{z+n_S}$ ) that enters into the unperturbed Hamiltonian via

$$H_0(\boldsymbol{\lambda}) = -\sum_{\iota,z} \boldsymbol{\lambda}_z \cdot \boldsymbol{\sigma}_{\iota,z}, \qquad (12)$$

where  $\iota = \{x, y\}$  is the lattice index in the *ab* plane.

We add the interacting part and treat it as a correction to the unperturbed one,

$$H_{1} \equiv H_{I} - H_{0} = \frac{J}{2} \sum_{i,j} \boldsymbol{\sigma}_{i} \cdot \mathbf{D}_{ij} \cdot \boldsymbol{\sigma}_{j} - \sum_{\iota} (\mathbf{E} - \boldsymbol{\lambda}_{z}) \cdot \boldsymbol{\sigma}_{\iota,z}.$$
(13)

The variational free energy

$$F_{\text{var}}(\boldsymbol{\lambda}) \equiv F_0(\boldsymbol{\lambda}) + \langle H_1(\boldsymbol{\lambda}) \rangle_0 \tag{14}$$

is minimized to determine the variational parameters  $\lambda_z$ . The unperturbed free energy is denoted  $F_0$ , and the unperturbed average is

$$\langle X \rangle_0 \equiv \frac{1}{Z_0} \text{Tr} X e^{-\beta H_0}, \qquad (15)$$

with the unperturbed partition sum

$$Z_0 \equiv \text{Tr}e^{-\beta H_0} = \left(\prod_{z=0}^{n_s-1} \zeta_z(\lambda_z)\right)^{N/n_s}.$$
 (16)

The vector with the independent self-consistently determined parameters is  $\lambda_z = \{\lambda_z^x, \lambda_z^y, \lambda^z\}$ . The partition sum of the unit cell with the fixed vertical coordinate *z* reads as

$$\zeta_{z}(\boldsymbol{\lambda}_{z}) = \sum_{l=0}^{n_{S}-1} e^{\beta \boldsymbol{\lambda}_{z} \cdot \boldsymbol{\sigma}_{z}(l)}, \qquad (17)$$

where  $\sigma_z(l)$  is defined in the same way as in Eq. (9),  $\sigma_z(l) \equiv \{\cos(2\pi l/n_S), \sin(2\pi l/n_S), (-1)^{l+z} \tan \theta\}.$ 

The thermal average of the nonlocal interaction term in  $H_1$ in the canonical ensemble with respect to  $H_0$  is standardly decoupled,  $\langle \sigma_i \sigma_j \rangle_0 = \langle \sigma_i \rangle_0 \langle \sigma_j \rangle_0$ . Moreover,  $\langle \sigma_i \rangle_0$  does not on the index  $\iota$  (*x*, *y* components of index *i*) but only on the *z* component. We denote it by  $\Pi_z(\lambda_z)$  and it can be expressed as

$$\langle \boldsymbol{\sigma}_{z} \rangle_{0}(\boldsymbol{\lambda}_{z}) = \frac{1}{\beta} \frac{\partial}{\partial \boldsymbol{\lambda}_{z}} \log \zeta_{z}(\boldsymbol{\lambda}_{z})$$
$$= \frac{1}{\zeta_{z}(\boldsymbol{\lambda}_{z})} \sum_{l=0}^{n_{s}-1} \boldsymbol{\sigma}_{z}(l) e^{\beta \boldsymbol{\lambda}_{z} \cdot \boldsymbol{\sigma}_{z}(l)} \equiv \boldsymbol{\Pi}_{z}(\boldsymbol{\lambda}_{z}).$$
(18)

The variational free-energy density of the unit cell with  $n_S$  water molecules along the *c* axis  $f_{\text{var}} \equiv F_{\text{var}} n_S / N$  can be

written as

$$f_{\text{var}} = -\frac{1}{\beta} \sum_{z=0}^{n_{s}-1} \log \zeta_{z}(\boldsymbol{\lambda}_{z})$$
  
+ 
$$\frac{J}{2} \sum_{z,z'=0}^{n_{s}-1} \boldsymbol{\Pi}_{z} \cdot \boldsymbol{D}_{zz'} \cdot \boldsymbol{\Pi}_{z'} - \sum_{z=0}^{n_{s}-1} (\mathbf{E}_{z} - \boldsymbol{\lambda}_{z}) \cdot \boldsymbol{\Pi}_{z}, \quad (19)$$

where

$$\mathbf{D}_{zz'} \equiv \frac{N_z}{N} \sum_{\iota, \iota'} \mathbf{D}_{\iota z, \iota' z'} = \frac{1}{N_z} \sum_{k_z} \tilde{\mathbf{D}}(k_x = 0, k_y = 0, k_z) e^{ik_z(z - z')}.$$
(20)

The *z* component of the wave vector acquires integer multiples of  $2\pi/n_s$  in the dimensionless representation. We denote by  $\tilde{\mathbf{D}}(\mathbf{k})$  the dipole interaction in the momentum (wave vector) space

$$\tilde{\mathbf{D}}(\mathbf{k}) \equiv \frac{1}{N} \sum_{i,j} \mathbf{D}_{ij} e^{-i\mathbf{k} \cdot (\mathbf{r}_i - \mathbf{r}_j)},$$
(21)

that is evaluated for a given structure of an infinite or periodic system by using the Ewald method (see, e.g., [34]).

Stationary equations  $\partial f_{var}/\partial \lambda_z = 0$  determine the equilibrium values of the variational parameters  $\bar{\lambda}_z$ , mean fields,

$$\bar{\boldsymbol{\lambda}}_{z} = \mathbf{E}_{z} - J \sum_{z'=0}^{n_{s}-1} \mathbf{D}_{zz'} \cdot \boldsymbol{\Pi}_{z'}(\bar{\boldsymbol{\lambda}}_{z'}).$$
(22)

The corresponding conjugate equations for the order parameters, thermally averaged local polarizations, read as

$$\langle \boldsymbol{\sigma}_{z} \rangle_{0} = \boldsymbol{\Pi}_{z} \left( \mathbf{E}_{z} - J \sum_{z'=0}^{n_{s}-1} \mathbf{D}_{zz'} \cdot \langle \boldsymbol{\sigma}_{z'} \rangle_{0} \right).$$
 (23)

## C. Spatial fluctuations: Monte Carlo simulations

The mean-field approximation averages spatial fluctuations and assumes their effect as a mean homogeneous environment, a mean field. The dipole-dipole interaction has a long-range and anisotropic character, and replacing it with an effective constant is a crude approximation. Such an approximation generically overestimates the tendency toward a long-range order. Hopping and tunneling in the clock model only partly suppress this tendency. The spatial fluctuations significantly affect ordering, particularly in low spatial dimensions. We used the local clock model, and instead of using the mean-field solution for the crystal equilibrium state, we used the classical Metropolis Monte Carlo simulations [35,36] and at low temperatures, also the Wang-Landau (WD) flat histogram method [37,38].

## D. Low-temperature quantum tunneling

The classical mean-field solution reflects well the hightemperature behavior of statistical models. Low-temperature asymptotics may, however, be strongly affected by quantum fluctuations. In extreme situations, they can destroy the classical behavior by the existence of quantum criticality with quantum critical points. Very recently, a quantum critical point was predicted in a linear chain of the dipole moments of the water molecules [30]. We assess the impact of quantum fluctuations within the clock model in a simplified form by allowing the hopping of the dipole moments only between the nearest and next-to-nearest potential minima. The hopping amplitudes a and d describe the transitions between the minima from the opposite and the same oxygen planes with and without changing the vertical component of the dipole moment, respectively, as shown in Fig. 2 by the oblique and horizontal dotted arrows. The new variational unperturbed Hamiltonian of the mean-field approximation with quantum tunneling remains local, but it becomes nondiagonal

$$\hat{H}_0(\boldsymbol{\lambda}) = \hat{h}^{\text{loc}} - \sum_{\iota, z} \boldsymbol{\lambda}_z \cdot \hat{\boldsymbol{\sigma}}_{\iota, z}, \qquad (24)$$

where  $\hat{h}^{\text{loc}}$  is a 12 × 12 matrix defined in Eq. (6). The local mean-field partition function (17) is then

$$\zeta(\boldsymbol{\lambda}_z) = \mathrm{Tr} e^{-\beta \hat{h}_{mf}^{\mathrm{loc}}} = \sum_{n=0}^{n_S} e^{-\beta v_n(z)}, \qquad (25)$$

where the  $v_n(z)$  are eigenvalues of the matrix

$$\hat{h}_{mf}^{\text{loc}} = \hat{h}^{\text{loc}} - \{\delta_{ij}\boldsymbol{\lambda}_z \cdot \boldsymbol{\sigma}_z(j)\}_{i,j=1}^{n_s}.$$
(26)

The mean-field polarization is then defined from the local partition function in Eq. (18). The nonlocal interacting part is added in the same way as in the classical case to reach the corresponding mean-field solution with quantum tunneling.

## V. THERMODYNAMIC PROPERTIES

## A. Mean-field approximation

We learned from the ground-state calculations that we can expect two ordered phases in the low-temperature limit. Approaching this limit from high temperatures, we start with the variational mean-field approximation with the unit cell containing  $n_S = 12$  water molecules along the *c* axis to assess the trends in the thermodynamic behavior when decreasing temperature. The first ordering emerges at  $T_{c1} \sim 31.6$  K below which the thermally averaged moments are planar and ferroelectrically ordered in the individual ab planes. Due to the anisotropic character of the dipole-dipole interaction, the ferroelectric planes order in antiparallel directions in the adjacent planes. The response to the electric field along the z axis increases with decreasing temperature and diverges at another critical point  $T_{c2} \sim 16.8$  K (Fig. 6). Below  $T_{c2}$ , the thermal average of the c projection of the dipole moment becomes nonzero and parallelly ordered along the c axis.

The local minima of the single dipole moment are spread between two adjacent *ab* planes. The dipole moments can be perfectly antiparallel in the adjacent planes only if they stay planar,  $T_{c1} > T > T_{c2}$ . Once the *c* projection of the dipole moment becomes macroscopically nonzero, the perfect antiparallel order is no longer possible (Fig. 7). The planar component of the dipole moment starts to drift in one direction, either clockwise or anticlockwise. It means that the *a* and *b* projections of the dipole moment are not the same when the *c* projection is nonzero (Fig. 8). Notice that the orientation of the water dipoles in this phase is quite similar to the configuration reported recently in cordierite [11].



FIG. 6. Staggered (along the *c* axis) static susceptibility  $\chi^A$  (electric field in *a* direction), and static susceptibility  $\chi^Z$  (electric field in *c* direction) indicating the two critical points below which the planar  $T_{c1}$  and vertical  $T_{c2}$  orders emerge, respectively. The homogeneous susceptibility  $\chi^S$  with the right vertical scale displays cusps at both transition points.

The projection of the dipole moment in the ab plane changes its direction when moving along the c axis. We take the unit cell with 12 water molecules in the c direction to characterize these changes. We define an order parameter for this unit cell, denoted by **P**, which reflects the antiferroelectric order in the ab plane along the c axis and the ferroelectric order in the c axis. The parameter is given by

$$\mathbf{P} \equiv \frac{1}{n_S} \sum_{z=0}^{n_S-1} \{(-1)^z p^x(z), (-1)^z p^y(z), p^z(z)\},$$
(27)

where  $n_S = 12$  represents the number of molecules in the unit cell. This parameter is referred to as the antiferroelectric/ferroelectric (AF/F) order parameter.

The modulus of this thermally averaged vector parameter is nonzero below  $T_{c1}$ , and the vector remains planar for  $T > T_{c2}$ . To understand what happens below  $T_{c2}$ , we need to introduce another vector parameter that can distinguish a drift of the planar projection of the dipole moment in one direction from random fluctuations when moving along the *c* axis. If one direction of the drift is preferred, we end up with a



FIG. 7. Left panel: Ferroelectric order emerges in the *ab* planes below  $T_{c1}$ , with the planes alternating the direction of the dipole moment, forming an antiferroelectric structure along the *c* axis. The instantaneous directions of the dipole moments (red arrows) deflect from the *ab* planes, but the thermal average of the *c* components (green arrows) is zero for  $T > T_{c2}$ . Right panel: Configuration of the water molecules for  $T < T_{c2}$  with long-range ferroelectric order of the *c* projections of the dipole moments. The vertical components of the dipole moments are aligned parallel, whereas the *ab*-plane components (blue arrows) in the neighboring planes are one angular step away from the prohibited antiparallel orientations.



FIG. 8. Three thermodynamic phases are observed in the meanfield solution of the water molecules in the beryl cavities. At high temperatures  $(T > T_{c1})$ , the system is in the paraelectric (P) phase, which lacks long-range order. At intermediate temperatures  $(T_{c1} > T > T_{c2})$ , the system exhibits planar ferroelectric order with antiparallel alignment along the *c* axis (AF). At low temperatures  $(T < T_{c2})$ , the system exhibits long-range order with all three projections nonzero and ferroelectric order (F) along the *c* direction. In this phase, the splitting of the *a* and *b* projections is clearly evident.

chiral order. To assess the chiral order quantitatively, we introduce a parameter  $O^{ch}$  using polarizations in three consecutive layers [39]:

$$\mathbf{O}^{\mathrm{ch}} = \frac{1}{n_S} \sum_{z=0}^{n_S-1} \mathbf{p}_z \cdot (\mathbf{p}_{z+2} \times \mathbf{p}_{z+4}).$$
(28)

This order parameter becomes nonzero whenever  $P^c \neq 0$  as demonstrated in Fig. 9. The chiral phase is a local minimum of the free energy for  $T < T_{c2}$ . The thermally averaged parameter **P** has nonzero only the *c* component below  $T_{c2}$  in the chiral state.

We see that below the critical point  $T_{c2}$ , at which the vertical component of the thermally averaged dipole moment becomes nonzero, both the unit-cell antiferroelectric order parameter **P** and chiral order parameter **O**<sup>ch</sup> can be nonzero in equilibrium. These parameters describe different thermodynamic states that generally differ in free energy.



FIG. 9. Three thermodynamic order parameters connected with antiferroelectrically ordered planes P, c projection of the dipole moment  $P^c$ , and the chiral parameter  $O^{ch}$ .



FIG. 10. Temperature dependence of free energy of the antiferroelectric and chiral ordering of the dipole moments along the *c* axis. They cross at a first-order phase transition point  $T_0 \simeq 4$  K. The black line is the equilibrium free energy.

However, only one of the solutions represents the true equilibrium state. The free energies in the mean-field approximation are plotted in Fig. 10. The chiral state has a lower free energy below  $T_0 \sim 4$  K, at which a first-order transition to equilibrium takes place. This conclusion is supported by the internal energy and heat capacity (Fig. 11). The corresponding behavior of the thermodynamic order parameters is plotted in Fig. 12.



FIG. 11. Temperature dependence of the mean-field internal energy (upper figure) and heat capacity (lower figure) indicating the three transition points. The upper two cusps of the heat capacity correspond to continuous transitions at temperatures  $T_{c1}$  and  $T_{c2}$ , and the lowest one to a discontinuous transition at  $T_0$ .



FIG. 12. Temperature dependence of the mean-field order parameters as discussed in the text calculated in the true equilibrium state. Parameter  $\mathbf{P}_{A/F}^{ab} = 0$  in the chiral state (ch) when  $O^{ch}$  saturates.

#### **B.** Monte Carlo simulations

The variational mean-field approximation suppresses spatial fluctuations and replaces the crystal with a mean environment. To improve this simplification, we used Monte Carlo simulations to take the spatial fluctuations into account more sophisticatedly.

We employed the standard Metropolis dynamics with single-site updates and improved them with nonlocal updates at low temperatures to explore our system's phase space more effectively. We investigated the system with linear size in aand b directions of  $L_a = L_b \in \{3, 4, 5, 6\}$  and up to 8 cells of 12 atoms in the c direction, assuming periodic boundary conditions. Averages were calculated using at least  $2 \times 10^6$ Monte Carlo steps (MCS) per dipole after discarding other  $1 \times 10^6$  MCS for equilibration. At low temperatures, up to  $5 \times 10^{6}$  MCS were used. The heat capacity calculated with the Monte Carlo simulations is plotted in Fig. 13. The broad peak in the plot indicates the second-order phase transition to the planar order at  $T_{c1} \approx 9.5$  K. We observe a cusp at  $\approx 5.5$  K indicating a nonanalytic behavior connected with the emergence of ferroelectrically ordered chains of the dipoles thermodynamically averaged only along the c axis. The vertical chains are, however, not ordered when changing their ab coordinates. The convergence significantly slows down when decreasing temperature. Two metastable states of the c-ordered chains coexist on long timescales. One is the bulk ferroelectric state, and the other is a state with disordered ferroelectric chains. The whole system gets stuck in one of these metastable states during the Monte Carlo simulations. The time needed to overcome the barrier between these states grows exponentially at very low temperatures. We, therefore, alternatively used the Wang-Landau (WD) flat histogram method [37,38] to handle this problem. It allowed us to calculate the density of states, from which we could directly deduce the temperature dependence of the free energy, internal energy, and heat capacity. Based on this, we identified a transition at  $T_{c2} \approx 2$  K to a bulk ferroelectric order where the ferroelectric c chains are aligned parallelly. A transition to the helical state was observed at  $T_0 \approx 1$  K. Sharp heat-capacity peaks indicate a discontinuity in the internal energy at both transitions. This conclusion was confirmed by discontinuities in the polarization shown in



FIG. 13. Temperature dependence of the Monte Carlo heat capacity of the water dipole moments with three transitions to ordered phases. The heat capacity exhibits a cusp at around 5.5 K. Different ways for calculating the heat capacity were employed, including descending (desc) and ascending (asc) temperatures as well as the Wang-Landau method. The first two direct approaches show hysteresis with two transition temperatures. In contrast to the mean-field solution, the Monte Carlo simulations suggest a first-order transition to the phase with ferroelectrically ordered c projections of the polarization.

Fig. 14. The Monte Carlo simulations indicate that the phase transitions to the bulk ferroelectric ordering of the *c* chains at  $T_{c2}$  and to the helical state at  $T_0$  are of first-order type. The transition to the ferroelectric order along the *c* axis is second order in the mean-field approximation.

The precise determination of the transition temperature  $T_{c2}$  from Monte Carlo simulations becomes difficult due to the coexistence of two metastable states. We observed a hysteresis depending on whether we approached the transition from above or from below. When starting from the ordered low-temperature state, the transition temperature with a discontinuity in the internal energy is higher than that approached from the high-temperature disordered phase. This hystere-



FIG. 14. Temperature dependence of the Monte Carlo polarizations when the system is heated from the ground state. Planar order antiferroelectrically aligned along the *c* axis (blue dots), ferroelectric order of the *c* projection of the dipole moment (green), and helical order parameter (red). We can observe that the effective dipole moment's planar projection in the unit cell with  $n_s = 12$  water molecules becomes zero in the helical state.

sis is, however, suppressed by the Wang-Landau method (Fig. 13). It is worth mentioning that the cusp near 5.5 K is invisible in systems with only a few cells along the c axis. Long-range spatial fluctuations generally pull down the transition temperatures. The Monte Carlo simulations with spatial fluctuations support, nevertheless, the overall picture of the mean-field solution. They additionally predict the existence of inhomogeneous metastable states, possibly leading to a first-order transition to the state with a nonzero projection of the polarization to the c axis. A more detailed analysis of the model with Monte Carlo simulations will be published separately.

#### C. Quantum tunneling

The behavior of the dipole moments at low temperatures is strongly affected by quantum fluctuations. We included quantum fluctuations via tunneling between the local minima of the classical model. The new parameters of the local Hamiltonian are a ranging from  $a_2 \approx 0.63$  meV to  $a_1 \approx 1.9$  meV and d from  $d_2 \approx 1.26 \text{ meV}$  up to  $d_1 \approx 3.9 \text{ meV}$  for the hopping amplitude to the nearest minimum, and the hopping amplitude to the next-nearest minimum, respectively. The local hopping of the unit dipole moment changes the energetic balance and affects the response to the external electric field. We plotted the antiferroelectric  $\chi^A$  and ferroelectric  $\chi^Z$  susceptibilities along the c axis in Fig. 15. In the first setting  $V_{b1} = 56 \text{ meV}$ and  $\omega_1 = 8.2$  meV, the transition to the planar order is shifted to a lower temperature,  $T_{c1} \approx 20.2$  K. The quantum tunneling with these model parameters fully suppresses the emergence of the ferroelectric order along the c axis. In contrast, the second limiting setting with  $V_{b2} = 176 \text{ meV}$  and  $\omega_2 = 3.9 \text{ meV}$ shifts the transition temperature only to  $T_{c1} \approx 30$  K and allows for the emergence of a vertical ferroelectric order at nonzero temperature  $T_{c2} \approx 12.7$  K. The hopping amplitudes between the local minima for the local dipole moment, which affect the low-temperature thermodynamic behavior, were derived from the input physical parameters  $V_b$  and  $\omega$ . The former was taken from Ref. [21], which describes the hexagonal symmetry of the local water dipole moment within a single beryl cavity. The amplitude connecting the two hexagons of the crystallographic cavity for the water molecule  $\omega$  was chosen to keep all the local minima equivalent. The existence of the thermodynamic order along the *c* axis is strongly affected by the value of  $\omega$ , which is reflected in the hopping amplitude a. We plotted the dependence of the planar, vertical, and chiral order parameters on the relative hopping amplitude  $a/a_1$ at zero temperature in Fig. 16. We see that the greater the amplitude with greater quantum fluctuations, the lower the probability of the vertical order. The limiting case  $a = a_1 \approx$ 1.9 meV is beyond the edge of the existence of the vertical order at zero temperature. Therefore, we do not observe a divergence of the ferroelectric susceptibility at nonzero temperatures. The vertical order occurs at a value of  $a = 0.41 a_1$ , and the vertical polarization is reduced to  $\sim 60\%$  of the original classical values at zero temperature for  $a = a_2 \approx 0.63$  meV. Chiral order at zero temperature is suppressed by quantum tunneling above  $a_{ch} \approx 0.052 a_1$ . Recent experimental data supported by Monte Carlo simulations revealed that deuterated water molecules reduce quantum tunneling and enhance



FIG. 15. Antiferroelectric (blue) and ferroelectric (green) susceptibilities calculated with quantum tunneling amplitudes as discussed in Sec. IV D. Dynamical quantum fluctuations decrease the transition temperatures, as do the static spatial fluctuations. The quantum tunneling for the setting labeled "1" (with  $V_{b1}$ ,  $\omega_1$ ) appears to lead to a zero *c* projection of the thermally averaged dipole moment (top figure). The inset shows the ferroelectric susceptibility with a magnified vertical scale. The second setting (bottom figure) labeled "2" (with  $V_{b2}$ ,  $\omega_2$ ) allows for the existence of vertical ferroelectric order at a nonzero temperature  $T_{c2}$ , although with a suppressed chiral order.



FIG. 16. Dependence of the in-plane antiferroelectric polarization  $P^{ab}$  (blue curve), the vertical polarization  $P^c$  (green curve), and the chiral order parameter  $O^{ch}$  (red curve) on the relative strength of quantum tunneling defined by the hopping parameter  $a/a_1$  at zero temperature. The value  $a/a_1 = 1$  corresponds to setting "1," while  $a/a_1 = 0.33$  corresponds to setting "2," representing different strengths of quantum tunneling. Chiral order is suppressed for values above  $a_{ch} \approx 0.052 a_1$ .

the tendency towards long-range antiferroelectric ordering [40]. Our mean-field approximation of the deuterated water decreases quantum tunneling in beryl by a factor of 2.

#### VI. CONCLUSIONS

We have developed a model that considers 12 equivalent orientations for the interacting dipole moments of water molecules confined in beryl crystal cavities. Recent NMR experiments [31] revealed that the dipole moments of type-I water molecules are not exactly orthogonal to the hexagonal caxis but instead deflect by approximately  $\pm 18^{\circ}$  from the crystallographic *ab* plane. Moreover, the potential minima for the hydrogen nuclei pointing to the oxygen atoms of the confining beryl molecules are shifted by 30° in the neighboring layers. This situation contradicts previously used theoretical models of water molecules in beryl crystals, prompting us to introduce a model of the orientational potential for the dipole moments of water molecules with dihexagonal symmetry, involving alternating minima with the c component of the dipole moment pointing up and down.

We used a classical variational mean-field approximation to obtain a qualitative thermodynamic picture of the behavior of the interacting dipole moments and their equilibrium configurations. We used the standard dipole-dipole interactions and included the lattice potential via a clock model where individual dipoles can adopt only one of the 12 minima of the lattice potential. We also used Monte Carlo simulations to more accurately include spatial fluctuations and included quantum tunneling between the minima to assess the lowtemperature macroscopic behavior better.

The classical mean-field solution predicts a nonzero Curie temperature  $T_{c1}$  below which the dipole moments are ferroelectrically ordered within the crystallographic *ab* planes, but the orientations of the planar dipole moments are antialigned, antiferroelectrically ordered along the c axis. The thermodynamic order along this axis remains zero down to another critical point  $T_{c2}$ , below which a different order emerges with a nonzero thermally averaged c projection of the dipole moment. The central and most important feature of our model, hitherto absent in existing models with only planar dipole moments in the crystallographic ab plane, is the possibility of the existence of a nonzero c projection of the thermally averaged dipole moment on the c axis. The order along the c axis leads to unexpected consequences for the symmetry of the equilibrium state. The ferroelectric order, together with the nonisotropic character of the dipole-dipole interaction, produces a helical structure of the dipole moments in which their directions go through all the local minima along the c axis within the unit cell containing  $n_s = 12$  positions symmetrically distributed to cover the whole angle of 360°.

Monte Carlo simulations, treating spatial fluctuations more realistically, essentially supported the thermodynamic picture of the water dipole moments obtained from classical variational mean-field theory. However, the simulations suggested a less straightforward transition to the ferroelectrically ordered c projection of the polarization. An intermediate state with a partial order of ferroelectric vertical chains was predicted, smearing the transition to the bulk ferroelectric state with a nonzero c projection of the polarization and making it of a first-order type. More precise simulations of spatial fluctuations are necessary to clarify this issue.

The classical model was improved by quantum tunneling to better assess the low-temperature asymptotics. The long-range order along the c axis appears to be significantly breached by quantum fluctuations that ultimately destroy the ordering at nonzero temperatures for the limiting values of the parameters in the setting "1." However, accurately assessing the impact of quantum fluctuations is currently difficult due to the lack of experimental data from which we could extract the necessary microscopic model parameters for the hopping amplitudes aand d between the local minima of the dipole moments of the confined water molecules in beryl. Additionally, a direct comparison with the experiment is hindered by the fact that the water molecules fill only some of the cavities of the beryl crystals, which is not accounted for in our model. Hence, the experimental critical temperatures are overestimated in our theory depending on the hydration level. Nevertheless, our study clearly shows that the observed deflection of the dipole moments from the crystallographic *ab* plane may lead to a nontrivial structure with a nonzero polarization along the *c* axis, which is worth further experimental exploration.

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- S. Chakraborty, H. Kumar, C. Dasgupta, and P. K. Maiti, Acc. Chem. Res. 50, 2139 (2017).
- [2] M. F. Chaplin, Structuring and Behaviour of Water in Nanochannels and Confined Spaces (Springer, Dordrecht, 2010).
- [3] W. H. Thompson, J. Chem. Phys. 149, 170901 (2018).
- [4] N. E. Levinger, Science 298, 1722 (2002).
- [5] P. Bampoulis, K. Sotthewes, E. Dollekamp, and B. Poelsema, Surf. Sci. Rep. 73, 233 (2018).
- [6] M. Bartl, Analyst 97, 854 (1972).
- [7] C. Yan, J. Nishida, R. Yuan, and M. D. Fayer, J. Am. Chem. Soc. 138, 9694 (2016).
- [8] B. Winkler and B. Hennion, Phys. Chem. Miner. 21, 539 (1994).
- [9] J. L. Bishop, M. D. Lane, M. D. Dyar, S. J. King, A. J. Brown, and G. A. Swayze, Am. Mineral. 99, 2105 (2014).
- [10] B. Winkler, G. Coddens, and B. Hennion, Am. Mineral. 79, 801 (1994).
- [11] M. A. Belyanchikov, M. Savinov, Z. V. Bedran, P. Bednyakov, P. Proschek, J. Prokleska, V. A. Abalmasov, J. Petzelt, E. S. Zhukova, V. G. Thomas *et al.*, Nat. Commun. **11**, 3927 (2020).
- [12] A. I. Kolesnikov, L. M. Anovitz, E. Mamontov, A. Podlesnyak, and G. Ehlers, J. Phys. Chem. B 118, 13414 (2014).
- [13] A. P. Dudka, M. A. Belyanchikov, V. G. Thomas, Z. V. Bedran, and B. P. Gorshunov, J. Surf. Investigation 14, 718 (2020).
- [14] M. A. Belyanchikov, Z. V. Bedran, M. Savinov, P. Bednyakov, P. Proschek, J. Prokleska, V. A. Abalmasov, E. S. Zhukova, V. G. Thomas, A. Dudka *et al.*, Phys. Chem. Chem. Phys. 24, 6890 (2022).
- [15] Y. Sugitani, K. Nagashima, and S. Fujiwara, Bull. Chem. Soc. Jpn. 39, 672 (1966).
- [16] D. L. Wood and K. Nassau, J. Chem. Phys. 47, 2220 (1967).
- [17] G. Gibbs, D. Breck, and E. Meagher, Lithos 1, 275 (1968).
- [18] B. P. Gorshunov, E. S. Zhukova, V. I. Torgashev, V. V. Lebedev, G. S. Shakurov, R. K. Kremer, E. V. Pestrjakov, V. G. Thomas, D. A. Fursenko, and M. Dressel, J. Phys. Chem. Lett. 4, 2015 (2013).
- [19] B. P. Gorshunov, E. S. Zhukova, V. I. Torgashev, E. A. Motovilova, V. V. L. and Anatoly S. Prokhorov, G. S. Shakurov, R. K. Kremer, V. V. Uskov, E. V. Pestrjakov, V. G. Thomas *et al.*, Phase Trans. **87**, 966 (2014).

- [20] E. S. Zhukova, V. I. Torgashev, B. P. Gorshunov, V. V. Lebedev, G. S. Shakurov, R. K. Kremer, E. V. Pestrjakov, V. G. Thomas, D. A. Fursenko, A. S. Prokhorov *et al.*, J. Chem. Phys. **140**, 224317 (2014).
- [21] A. I. Kolesnikov, G. F. Reiter, N. Choudhury, T. R. Prisk, E. Mamontov, A. Podlesnyak, G. Ehlers, A. G. Seel, D. J. Wesolowski, and L. M. Anovitz, Phys. Rev. Lett. **116**, 167802 (2016).
- [22] B. P. Gorshunov, V. I. Torgashev, E. S. Zhukova, V. G. Thomas, M. A. Belyanchikov, C. Kadlec, F. Kadlec, M. Savinov, T. Ostapchuk, J. Petzelt *et al.*, Nat. Commun. 7, 12842 (2016).
- [23] Y. Finkelstein, R. Moreh, S. L. Shang, Y. Wang, and Z. K. Liu, J. Chem. Phys. **146**, 124307 (2017).
- [24] M. A. Belyanchikov, E. S. Zhukova, S. Tretiak, A. Zhugayevych, M. Dressel, F. Uhlig, J. Smiatek, M. Fyta, V. G. Thomas, and B. P. Gorshunov, Phys. Chem. Chem. Phys. 19, 30740 (2017).
- [25] M. Dressel, E. S. Zhukova, V. G. Thomas, and B. P. Gorshunov, J. Infrared Milli Terahz Waves 39, 799 (2018).
- [26] E. S. Zhukova, M. Belyanchikov, M. Savinov, P. Bednyakov, V. Thomas, L. Kadyrov, E. Simchuk, Z. Bedran, V. Torgashev, A. Dudka *et al.*, EPJ Web Conf. **195**, 06018 (2018).
- [27] V. A. Abalmasov, J. Phys.: Condens. Matter 33, 34LT01 (2021).
- [28] M. A. Belyanchikov, M. Savinov, P. Proschek, J. Prokleška, E. S. Zhukova, V. G. Thomas, Z. V. Bedran, F. Kadlec, S. Kamba, M. Dressel *et al.*, Nano Lett. **22**, 3380 (2022).
- [29] Y. Bassie, M. Mahmud, and P. M. Bekele, SSRN Electronic J. (2022).
- [30] T. Serwatka, R. G. Melko, A. Burkov, and P.-N. Roy, Phys. Rev. Lett. 130, 026201 (2023).
- [31] V. Chlan, M. Adamec, H. Štěpánková, V. G. Thomas, and F. Kadlec, J. Chem. Phys. 158, 124308 (2023).
- [32] C. Schran and D. Marx, Phys. Chem. Chem. Phys. 21, 24967 (2019).
- [33] K. Momma and F. Izumi, J. Appl. Cryst. 44, 1272 (2011).
- [34] D. Rapaport, *The Art of Molecular Dynamics Simulation*, 2nd ed. (Cambridge University Press, Cambridge, 2004).
- [35] N. Metropolis, A. W. Rosenbluth, M. N. Rosenbluth, A. H. Teller, and E. Teller, J. Chem. Phys. 21, 1087 (1953).

- [36] D. Frenkel, *Computer Modelling of Fluids Polymers and Solids* (Springer, Dordrecht, 1990), pp. 83–123.
- [37] F. Wang and D. P. Landau, Phys. Rev. Lett. 86, 2050 (2001).
- [38] F. Wang and D. P. Landau, Phys. Rev. E 64, 056101 (2001).
- [39] A. M. Tsvelik and O. M. Yevtushenko, Phys. Rev. Lett. 119, 247203 (2017).
- [40] Y. T. Chan, E. Uykur, M. A. Belyanchikov, M. Dressel, V. A. Abalmasov, V. Thomas, E. S. Zhukova, and B. Gorshunov, Phys. Rev. Res. 4, 023205 (2022).