Electronic transport and thermoelectricity in the selenospinel $Cu_{6-x}Fe_{4+x}Sn_{12}Se_{32}$

Yu Liu (刘育)◉[,](https://orcid.org/0000-0002-5567-9677)^{1,∗} Zhixiang Hu (胡之翔),^{1,2} Xiao Tong (佟晓)◑,³ David Gra[f](https://orcid.org/0000-0001-6195-0462)◑,⁴ and C. Petrovi[c](https://orcid.org/0000-0001-6063-1881)◎^{1,2}

¹*Condensed Matter Physics and Materials Science Department, Brookhaven National Laboratory, Upton, New York 11973, USA*

²*Materials Science and Chemical Engineering Department, Stony Brook University, Stony Brook, New York 11790, USA*

³*Center of Functional Nanomaterials, Brookhaven National Laboratory, Upton, New York 11973, USA*

⁴*National High Magnetic Field Laboratory, Florida State University, Tallahassee, Florida 32306-4005, USA*

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We report a study of selenospinel Cu_{6−*x*}Fe_{4+*x*}Sn₁₂Se₃₂ ($x = 0, 1, 2$) single crystals, which crystallize in a cubic structure with the $F d\overline{3}m$ space group, and show typical semiconducting behavior. The large discrepancy between the activation energy for electrical conductivity E_p (32.3–69.8 meV), and for thermopower E_s (3.2–11.5 meV), indicates a polaronic transport mechanism between 350 and 50 K. With decreasing temperature, it evolves into variable-range hopping conduction. Furthermore, the heat capacity shows a hump around 25(5) K and diverges from the Debye $T³$ law at low temperatures, indicating the observation of structural glassy features in these crystalline solids.

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I. INTRODUCTION

New transition metal chalcogenides may enable transformative changes in thermoelectric energy creation and conversion $[1-3]$. The Kondo-insulator-like semiconductor $FeSb₂$ hosts not only strong electronic correlation but also the highest thermoelectric power factor in nature along with thermopower up to \sim 45 mV K⁻¹ [\[4–9\]](#page-4-0). The ternary CoSbS also features a high value of thermopower \sim 2.5 mV K⁻¹ around 40 K but also a relatively large value of thermal conductivity \sim 100 W K⁻¹ m⁻¹ near the thermopower peak temperature [\[10\]](#page-4-0). This inhibits the thermoelectric figure of merit $ZT = (S^2/\rho \kappa)T$, where *S* is thermopower, ρ is electrical resistivity, κ is thermal conductivity, and T is temperature, respectively. All these parameters are interdependent and related to the electronic structure and charge-carrier scattering of the material. To improve a material for thermoelectricity it is important to achieve high thermoelectric power factor (S^2/ρ) and low thermal conductivity κ .

Quaternary materials, especially transition metal chalcogenides, are more tunable than binary and ternary compounds [\[11\]](#page-4-0). Quaternary spinel $Cu₆Fe₄Sn₁₂Se₃₂ features a large slope$ of density of states near the Fermi energy, favorable for high value of thermopower [\[12\]](#page-4-0). This compound crystallizes in a cubic yet complex crystal structure of high symmetry with many atoms where high degeneracy of band extrema are commonly found [\[13–16\]](#page-4-0). Numerous atoms in the high symmetric unit cell along with the Fe/Sn site disorder contribute to relatively low thermal conductivity κ in Cu_yFe₄Sn₁₂Se₃₂ polycrystals [\[17,18\]](#page-4-0). In addition, $Cu_vFe_4Sn_{12}Se_{32}$ is stable at *y* = 6 without impurity, where a large *S* of ~200 μ V K⁻¹

and a rather low ρ coexist at 340 K as well as low κ of \sim 1.5 W K⁻¹ m⁻¹ at the room temperature.

Here we fabricated a series of Cu_{6−*x*}Fe_{4+*x*}Sn₁₂Se₃₂ $(x = 0, 1, 2)$ single crystals to shed light on the electronic transport mechanism. X-ray photoemission spectroscopy (XPS) measurements were performed to investigate the valencies of interpolated cations. We observe a relatively high value of thermopower \sim 0.18 mV/K at 300 K for $x = 0$, which increases to ∼0.36 mV/K for *x* = 2. The large discrepancy between the activation energy for electrical conductivity and for thermopower above 50 K points to the polaronic transport mechanism. With decreasing temperature, polaronic transport evolves into the dominant variable-range hopping (VRH) mechanism.

II. EXPERIMENTAL DETAILS

Single crystals of $Cu_{6-x}Fe_{4+x}Sn_{12}Se_{32}$ ($x = 0, 1, 2$) were grown by melting a stoichiometric mixture of Cu and Fe (4N; Alfa Aesar) powder, Sn and Se (5N; Alfa Aesar) shot [\[12\]](#page-4-0). The starting materials were mixed and loaded in 1-cm-diameter Al_2O_3 crucibles, vacuum-sealed in quartz tubes, heated to 650° C over 10 h and dwelled for 12 h, then cooled to 550 °C at a rate of $1 \degree C/h$, and finally quenched in ice water. Crucible-size-limited single crystals were obtained with shining surface and confirmed by back-reflection Laue x-ray photographs.

The powder x-ray diffraction (XRD) data were taken on crushed crystals with Cu $K\alpha$ ($\lambda = 0.15418$ nm) radiation of a Rigaku Miniflex powder diffractometer. The element analysis was performed using energy-dispersive x-ray spectroscopy (EDX) in a JEOL LSM-6500 scanning electron microscope. Multiple points on several samples were examined and the average actual chemical compositions of samples are listed in Table [I.](#page-1-0) XPS measurements were carried out in an ultrahighvacuum (UHV) system with base pressure of \sim 2 × 10⁻¹⁰ Torr

^{*}Present address: Los Alamos National Laboratory, Los Alamos, NM 87545, USA.

TABLE I. The actual chemical composition and lattice parameter a for Cu_{6−*x*}Fe_{4+*x*}Sn₁₂Se₃₂ (*x* = 0, 1, 2).

| Nominal | Actual | a(A) | |
|--------------------------|--|--------|--|
| $Cu6Fe4Sn12Se32$ | $Cu_{5.9(2)}Fe_{4.0(1)}Sn_{13(1)}Se_{32(1)}$ | 10.774 | |
| $Cu5Fe5Sn12Se32$ | $Cu_{5,3(2)}Fe_{3,6(2)}Sn_{14(1)}Se_{31(1)}$ | 10.786 | |
| $Cu_4Fe_6Sn_{12}Se_{32}$ | $Cu_{5.1(2)}Fe_{4.2(2)}Sn_{13(1)}Se_{32(1)}$ | 10.796 | |

and equipped with a hemispherical electron energy analyzer (SPECS; Phoibos 100) and twin anode x-ray source (SPECS; XR50). Al- $K\alpha$ (1486.6 eV) radiation was used at 15 kV and 20 mA. The angle between the analyzer and the x-ray source is 45[°] and photoelectrons were collected along the sample surface normal. In order to remove potential surface contaminations and oxygen layers, each sample was cleaned in UHV by Ar^+ sputtering for 60 min under conditions of Ar gas pressure of 2×10^{-5} Torr and Ar⁺ kinetic energy of 2 keV. XPS data was analyzed using Casa XPS and peak positions were calibrated using residual adventitious carbon C 1*s* at 284.8 eV. The electrical resistivity, thermopower, and thermal conductivity were measured in the Quantum Design thermal transport option (TTO) in a PPMS-9 with the standard four-probe method and in continuous measuring mode. The maximum heater power and period were set as 50 mW and 1430 s along with the maximum temperature rise of 3%. The crystals were cut and polished into rectangular bars with typical dimensions of $4 \times 1 \times 0.8$ mm³. Epoxy and copper leads were used for TTO contacts. The relative error in our measurement for thermopower was below 5% based on the Ni standard measured under identical conditions. Sample dimensions were measured by a Nikon SMZ-800 optical microscope with resolution of 10 μ m.

III. RESULTS AND DISCUSSIONS

Figure $1(a)$ shows the crystal structure of selenospinel $Cu_{6-x}Fe_{4+x}Sn_{12}Se_{32}$ ($x = 0, 1, 2$). The spinel-type compounds have a general formula of *AB*2*X*4, where *A* and *B* are metals encapsulated by tetrahedra and octahedra, and *X* is oxygen or chalcogen elements. Herein, the unit cell contains numerous 54 atoms [Fig. 1(a)] with *A*-site Cu deficiencies; Cu atoms are coordinated tetrahedrally by $X = \text{Se}$; *B*-site Fe/Sn atoms are randomly distributed in the same crystallographic site and are coordinated octahedrally by Se [\[12\]](#page-4-0). We note that $Cu_yFe_4Sn_{12}Se_{32}$ is stable at $y = 6$ $[17,18]$. Figure $1(b)$ shows the structural refinement of powder XRD for $Cu_{6-x}Fe_{4+x}Sn_{12}Se_{32}$ ($x = 0, 1, 2$), indicating that all reflections can be well indexed in the $Fd\overline{3}m$ space group. No impurity peaks are observed, confirming higher tolerance of Cu deficiencies in Cu_{6−*x*}Fe_{4+*x*}Sn₁₂Se₃₂. The determined lattice parameter *a* increases slightly from 10.774(2) to 10.796(2) Å with increasing *x* [Fig. 1(c)]. The value of *a* is close to the previous result for $Cu₆Fe₄Sn₁₂Se₃₂ [17]$ $Cu₆Fe₄Sn₁₂Se₃₂ [17]$, while the rise in *x* suggests larger ionic size of Fe than Cu ions.

To investigate the valencies of interpolated cations, we measured the XPS where all three samples show similar valence states for each element. As shown in Fig. [2\(a\),](#page-2-0) the Cu 2*p*³/² binding energy of 932.9 eV suggests the Cu is in

FIG. 1. (a) Crystal structure of selenospinel Cu6−*^x*Fe4+*^x*Sn12Se32 $(x = 0, 1, 2)$. (b) Powder x-ray diffraction patterns and (c) the evolution of lattice parameter *a*.

metallic state Cu⁰ or Cu¹⁺. The Auger spectra show Cu LMM at kinetic energy of 916.8 eV [Fig. $2(b)$], which exclude the possibility of Cu^{0} and directly confirm the Cu^{1+} state. The highest intensity of Fe 2*p* is coincidentally overlapping with Sn $3p_{3/2}$ [Fig. [2\(c\)\]](#page-2-0). By deconvoluting each component, the binding energy of Fe 2*p*3/² is located at 708.2 eV and shifted toward high binding energy about 1.5 eV when compared to the binding energy of pure metal Fe $2p_{3/2}$ at 706.7 eV [Fig. $2(d)$]. Compared to the binding energy of Sn $3d_{5/2}$ at 485.2 eV for pure metal, the binding energy of Sn $3d_{5/2}$ is located at 486.5 eV [Fig. $2(e)$] and shifted toward high binding energy about 1.3 eV. These observations suggest electropositive Fe (mostly Fe^{2+}) and Sn (mostly Sn^{4+}) ions in these compounds [\[19,20\]](#page-4-0). Moreover, the binding energy of Se $3d_{5/2}$ is located at 54.5 eV and shifted toward lower binding energy about −0.9 eV when compared to the binding energy of pure metal Se $3d_{5/2}$ at 55.4 eV [Fig. [2\(f\)\]](#page-2-0), suggesting electronegative Se (mostly Se^{2-}) ions in the compounds. The subtle difference in binding energy of Sn and Se indicates a slight local structural change induced by Fe substitution, calling for further synchrotron x-ray diffraction study on the site occupancies.

The temperature dependence of electrical resistivity $\rho(T)$ for $Cu_{6-x}Fe_{4+x}Sn_{12}Se_{32}$ ($x = 0, 1, 2$) is depicted in Fig. [3\(a\),](#page-2-0) showing an obvious semiconducting behavior. The value of room temperature resistivity ($\rho_{300 \text{ K}}$) is about $2.2 \times 10^{-4} \Omega$ m for Cu₆Fe₄Sn₁₂Se₃₂, which monotonically increases to 8.0 \times 10⁻⁴ and 5.1 × 10⁻³ Ω m for $x = 1$ and 2, respectively. In such a complex disordered system, the electronic conduction mechanism is of great interest. In general, three typical

FIG. 2. Typical XPS spectra for the indicated elements of $Cu_6Fe_4Sn_{12}Se_{32}$ (bottom), $Cu_5Fe_5Sn_{12}Se_{32}$ (middle), and $Cu₄Fe₆Sn₁₂Se₃₂$ (top) samples.

models are considered to describe the semiconducting behavior: (i) thermally activated model $\rho(T) = \rho_0 \exp(E_\rho / k_B T)$, where E_{ρ} is activation energy and k_{B} is the Boltzmann constant; (ii) adiabatic small polaron hopping model $\rho(T)$ = $AT \exp(E_p/k_B T)$; and (iii) VRH model $\rho(T) = \rho_0 \exp(T_0/T)$ ^v $[21–28]$, where T_0 is a characteristic temperature and is related to the density of states available at the Fermi level and carrier localization length, and ν depends on the dimensionality. To understand the transport mechanism, it is necessary to fit the temperature-dependent resistivity curves based on these three formulas.

Figure $3(b)$ exhibits the fitting result of the adiabatic small polaron hopping model from 50 to 350 K. The extracted activation energy E_{ρ} is about 32.3(4) meV for $x = 0$, in agreement with the previous result [\[17\]](#page-4-0), and gradually increases to 37.3(5) and 69.8(9) meV for $x = 1$ and 2, respectively. However, the $\rho(T)$ curves can also be fitted by the thermally activated model (see discussion below). With decreasing temperature, the resistivity increases abruptly below 50 K, indicating a strong localization regime with a VRH conductivity, $\rho(T) = \rho_0 \exp(T_0/T)^\nu$. We use the logarithmic-derivative method for an accurate estimate of the v exponents, i.e., plot $\ln(-\delta \ln \rho / \delta \ln T)$ versus $\ln(T)$ [Fig. $3(c)$], where the slopes of the fit give the values of expo-

FIG. 3. (a) Temperature dependence of electrical resistivity $\rho(T)$ for the indicated samples. (b) $\ln(\rho/T)$ versus $1000/T$ curves fitted by the adiabatic small polaron hopping model $\rho(T) = AT \exp(E_p / k_B T)$. (c) $ln(-δ ln ρ/δ ln T)$ versus $ln(T)$, where the slopes give the exponents for variable-range hopping model $\rho(T) = \rho_0 \exp(T_0/T)^{\nu}$.

nent ν. The obtained ν (0.29–0.44) situates between 0.25 for the Mott's VRH and 0.5 for the Efros-Shklovskii-type VRH conductivity.

To distinguish the thermally activated model and the polaron hopping model, we further measured temperaturedependent thermopower *S*(*T*). The *S*(*T*) exhibits large positive values in the whole temperature range [Fig. $4(a)$], indicating dominant hole-type carriers. The room temperature value $S_{300 \text{ K}}$ of $x = 0$ is about 180 μ V K⁻¹; the large value of *S* is attributed to the small number of holes. The rise in *x* further increases thermopower, reaching $S_{300 \text{ K}} = 360 \text{ }\mu\text{V K}^{-1}$ for $x = 2$. The $S(1000/T)$ curves of all samples show similar shape [Fig. $4(b)$] and can be fitted with the equation $S(T) = (k_B/e)(\alpha + E_S/k_B T)$ from 50 to 350 K [\[22\]](#page-4-0), where E_S is activation energy and α is constant. The obtained acti-

FIG. 4. (a) Temperature dependence of thermopower $S(T)$ for the indicated samples. (b) $S(T)$ vs $1000/T$ curves fitted using $S(T) = (k_B/e)(\alpha + E_S/k_B T)$ from 50 to 350 K. (c) The evolution of E_ρ and E_S .

FIG. 5. Temperature dependence of (a) thermal conductivity $\kappa(T)$, (b) phonon mean free path l_{κ} , (c) the figure of merit ZT, and (d) specific heat C_p divided by T^3 for the indicated samples.

vation energy for thermopower, E_S (3.2–11.5 meV), are much smaller than those for conductivity, E_{ρ} (32.3–69.8 meV), as shown in Fig. $4(c)$. This large discrepancy between E_S and *E*^ρ indicates a polaron transport mechanism of carriers. According to the polaron model, the E_S is the energy required to activate the hopping of carriers, while E_{ρ} is the sum of the energy needed for the creation of carriers and activating the hopping of carriers [\[22\]](#page-4-0). Therefore, E_S is smaller than E_o . The weak temperature-dependent $S(T)$ at high temperatures also supports the small polaron conduction; however, the *S*(*T*) changes its slope at lower temperatures where it evolves into the VRH conduction.

Figure $5(a)$ shows the temperature dependence of thermal conductivity $\kappa(T)$ for Cu_{6−*x*}Fe_{4+*x*}Sn₁₂Se₃₂ (*x* = 0, 1, 2). In general, $\kappa_{\text{total}} = \kappa_{\text{e}} + \kappa_{\text{L}}$, consists of the electronic part κ_e and the phonon term κ_L . Herein, the κ_e estimated from the Wiedemann-Franz law is negligibly small due to large electrical resistivity, indicating a predominantly phonon contribution. At room temperature, the κ shows a relatively low value of 1.32–1.57 W], K^{-1} m⁻¹, arising from its structural complexity, i.e., large Cu deficiencies and Fe/Sn site disorder as phonon scattering centers. Interestingly, the low-temperature $\kappa(T)$ follows a quasilinear *T* dependence [inset in Fig. 5(a)]. This deviates from the common $\kappa \sim T^3$ usually observed in bulk crystals or thin films [\[29,30\]](#page-4-0), implying nanostructural differences that are induced by different vacancies, in particular grains and associated phonon frequency changes [\[31,32\]](#page-4-0). The Fe-substituted samples show slightly larger values of $\kappa(T)$ when compared to $Cu_6Fe_4Sn_{12}Se_{32}$. This also leads to the increase in phonon mean free path l_k [Fig. 5(b)] estimated from the heat capacity and thermal conductivity via $\kappa_L = C_p v_s l_k / 3$, where κ_L is the lattice thermal conductivity, C_p is the heat capacity, v_s is the average sound speed calculated via Debye temperature [\[33,34\]](#page-4-0), and l_k is the phonon mean free path.

The figure of merit *ZT* shows maxima around the room temperature, reaching ∼0.03 for Cu₆Fe₄Sn₁₂Se₃₂ at 300 K [Fig. $5(c)$]. Note that the increase of κ above 200 K can be due to radiation and as a consequence the real *ZT* can be ∼0.044

TABLE II. Variable-range hopping exponents and parameters from specific heat measurements for indicated samples.

| Sample | ν | B $(mJ \text{ mol}^{-1} \text{ K}^{-4})$ | $\Theta_{\rm D}$ (K) | $v_{\rm c}$ $(m s^{-1})$ |
|--------------------------|---------|---|-------------------------|-----------------------------|
| $Cu6Fe4Sn12Se32$ | 0.38(2) | 87(1) | 107(1) | 1030 |
| $Cu5Fe5Sn12Se32$ | 0.29(2) | 53(1) | 125(1) | 1210 |
| $Cu_4Fe_6Sn_{12}Se_{32}$ | 0.44(3) | 81(1) | 109(1) | 1050 |

for $Cu_6Fe_4Sn_{12}Se_{32}$ at 300 K. Although the thermopower is enhanced by Fe substitution, the electrical resistivity and thermal conductivity are also increased in Fe-substituted samples. Further efforts in nanostructuring of single-crystal alloys and/or doping with heavier element would be helpful to obtain better thermoelectricity performance at room temperature.

The temperature dependence of specific heat C_p divided by T^3 of the indicated samples is depicted in Fig. $5(d)$. Two interesting features are observed: a hump around 25(5) K and, it diverges from the Debye T^3 law at low temperatures, similar to those observed in amorphous solids arising from the enhanced density of states of acoustic phonons caused by the disordered structure [\[35](#page-4-0)[–40\]](#page-5-0). The coefficients $\beta = 53(1)$ –87(1) mJ mol⁻¹ K⁻⁴ were estimated from the local minimum values of C_p/T^3 , as shown by dashed lines in Fig. $4(d)$. The derived Debye temperature $\Theta_{\text{D}} = 107(1) - 125(1) \text{ K}$ by using the equation $\Theta_{\text{D}} =$ $[12\pi^4 NR/(5\beta)]^{1/3}$, which implies the average sound velocity of $v_s \approx 1030-1210$ m s⁻¹ (Table II). We should note that the large values of γ can also be from magnetic spin glassy state in this high Fe-content disordered system. Further synchrotron local structural (pair distribution function analysis) and neutron scattering experiments will be helpful to unveil its origin.

IV. CONCLUSIONS

In summary, we synthesized and studied single crystals of $Cu_{6-x}Fe_{4+x}Sn_{12}Se_{32}$ ($x = 0, 1, 2$). An electronic transport mechanism on cooling is dominated by polaron effect down to 50 K. On further cooling the VRH mechanism dominates. The Fe substitution enhances the thermopower possibly due to an increase of the slope and shift of the density of states near the Fermi level. The figure of merit *ZT* reaches ∼0.03–0.044 around the room temperature for $Cu₆Fe₄Sn₁₂Se₃₂$, calling for further carrier optimization.

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