



## Theoretical investigations on elastic properties, phase stability, and magnetism in Ni<sub>2</sub>Mn-based all-*d*-metal Heusler compounds

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The elastic properties, phase stability, and magnetism and correlations among them in all-*d*-metal Heusler compounds, i.e., Ni<sub>2</sub>Mn*T* (*T* = Sc, Ti, V, Cr, Y, Zr, Nb, Mo, Hf, Ta, and W) were systematically investigated by first-principles calculations. The results indicated that Ni<sub>2</sub>Mn*T* compounds were not fully consistent with the conventional atomic preferential occupation rule in the Heusler family. Within the scope of Heusler structures, Ni<sub>2</sub>Mn*T* compounds containing early transition metal atoms preferred *L*<sub>2</sub>-type structure, while those with late transition metal atoms were relatively stable in Hg<sub>2</sub>CuTi-type structure. In *d-d* interatomic hybridization-controlled all-*d*-metal Ni<sub>2</sub>Mn*T* Heusler compounds, the atomic radius determined the lattice sizes. Owing to the strong couplings among elastic parameters, phonon modes, and electronic structure, the most likely martensitic phase transition could be expected in weakly magnetic Ni<sub>2</sub>Mn*T* compounds with late transition metal atoms. By applying hydrostatic pressure or imposing chemical pressure via adjusting the composition of Ti in an off-stoichiometric Ni-Mn-Ti system, magnetism was weakened, and the suppressed martensitic phase transition could be re-evoked. In this paper, we also revealed that experimentally observed antiferromagnetism in Ni<sub>2</sub>MnTi originated from the arrest of the atomic diffusion process during the transition from the high-temperature chemically disordered paramagnetic state to the low-temperature chemically and magnetically ordered ferromagnetic state, which resulted in the formation of an intermediate metastable and partially disordered antiferromagnetic phase. Comparatively, Ni<sub>2</sub>Mn*T* compounds with early transition metals showed better ductility. In representative Ni<sub>2</sub>MnY and Ni<sub>2</sub>MnTa, it was found that nondirectional *d-d* interatomic hybridization became prevailing and helped establish the metal bonding character, which consequently enhanced the ductility. This paper can provide more insight into understanding the mechanism of martensitic phase transition and the origin of experimental anomalous magnetic states as well as the scheme to design multiple functional magnetic materials with outstanding ductility in the all-*d*-metal Heusler family. Experimentally observed exceptional multicaloric effects in Ni-Mn-based compounds with outstanding mechanical properties make all-*d*-metal Heusler compounds attractive for potential solid-state refrigeration application.

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### I. INTRODUCTION

Energy-efficient and environment-friendly solid-state refrigeration is the most promising alternative to the conventional vapor-compression cooling technology, and the essential prerequisites to successfully achieve solid-state refrigeration include high-performance phase transition materials with high cooling capacity. These materials possess outstanding phase transition characteristics so that they can produce large isothermal entropy change and adiabatic temperature change, which are ideal for high-efficiency cooling devices [1–8]. Among this type of materials, the Heusler family constitutes the most promising candidate materials. Their rich candidate material system, tunable phase transition temperature and hysteresis, quick response to multiple fields, and higher multiple-caloric effects make Heusler compounds

more attractive from the perspective of not only industrial application but also fundamental research [7,8]. The other side of conventional Heusler compounds includes their inherent brittleness due to the strong *p-d* covalent hybridization between the first-nearest-neighbor (1nn) atomic pairs of transition metal atoms and main group elements [9–14]. The poor ductility degrades the fatigue life and service ability, posing the biggest challenges for their utilization as solid-state refrigerant materials. To the greatest possibility of maintaining their functionalities, concrete and strenuous efforts have been devoted to improve their ductility by introducing the second phase [9,10], strengthening the grain boundary [11,12], and microalloying the rare earth elements [13,14]. Owing to the lack of fundamentally changing *p-d* covalent hybridization-dominated atomic bonding, the improvement of ductility is limited. Thus, systematic exploration of the phase transition materials with outstanding ductile properties is imperative.

Apart from the *p-d* covalent hybridization between different *p*- and *d*-group elements in the Heusler family, the *d-d* hybridization between different *d*-group elements also

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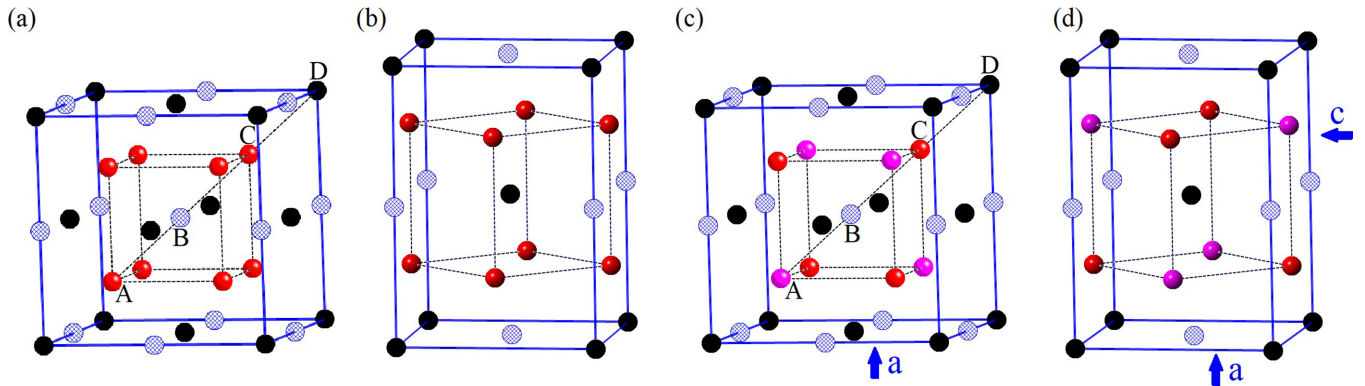


FIG. 1. (a) Regular cubic Heusler structure, i.e.,  $L2_1$ -type or  $\text{Cu}_2\text{MnAl}$ -type structure (space group  $Fm\bar{3}m$ , No. 225). (b) Regular tetragonal Heusler structure (space group  $I4/mmm$ , No. 139). (c) Inverse cubic Heusler structure, i.e.,  $\text{Hg}_2\text{CuTi}$ -type structure (space group  $F\bar{4}3m$ , No. 216). (d) Inverse tetragonal Heusler structure (space group  $I\bar{4}m2$ , No. 119). In cubic structure, A, B, C, and D represent four interpenetrating face-centered cubic (fcc) sublattices, whose Wyckoff coordinates are  $(0\ 0\ 0)$ ,  $(\frac{1}{4}\ \frac{1}{4}\ \frac{1}{4})$ ,  $(\frac{1}{2}\ \frac{1}{2}\ \frac{1}{2})$ , and  $(\frac{3}{4}\ \frac{3}{4}\ \frac{3}{4})$ , respectively.

to some extent contributed to the formation and stabilization of Heusler phases [15]. Based on this, Wei *et al.* [15,16] experimentally realized that the Heusler phase could be formed without the  $p$ -group elements. They classified this type of material as all- $d$ -metal Heusler compounds, which enriched the category of the Heusler family and expanded its research field. Recently, the concept of all- $d$ -metal Heusler compounds has been well established [17]. Experimentally, all- $d$ -metal Heusler compounds with typical phase transition exhibit improved mechanical properties. Owing to the good mechanical properties and large volume change observed across phase transition, the colossal elastocaloric effect under an applied stress of 700 MPa [18] and the giant barocaloric effect at 4 kbar [19] were experimentally achieved in polycrystalline all- $d$ -metal Ni-Mn-Ti shape memory compounds. Moreover, in directionally solidified all- $d$ -metal Heusler metamagnetic shape memory compounds, the martensitic transformation was found to be sensitive to applied pressure and stress. Low pressure-induced giant barocaloric effect [20] and large elastocaloric effect under a low critical stress [21] were observed in directionally solidified Ni-Co-Mn-Ti compounds. Moreover, mechanically and chemically stable Ni-Co-Mn-Ti rapidly solidified ribbons and quenched samples exhibited giant enhancement of the magnetocaloric response for a moderate field change of 2 T [22,23]. It is comparable with or larger than those of the best-known materials for near-room-temperature applications. Furthermore, due to the strong magnetostructural coupling in Ni-Mn-Ti-based compounds, their martensitic phase transition temperature and entropy changes can be jointly tailored by applying a magnetic field and hydrostatic pressure. When the magnetic field and hydrostatic pressure are applied in a proper sequence, the reversible entropy change gets effectively enhanced [20,24]. All the abovementioned advantages make Ni-Mn-based all- $d$ -metal Heusler compounds popular for the design and exploration of multicaloric materials for solid-state refrigeration application. Detailed comparisons are shown in the Supplemental Material [25] (see also Refs. [5,8,11,12,18–22,26–59] therein). However, current experiments mainly concentrate on the Ni-Mn-Ti system and its derivative [15,16,18–24,60,61]. Thus, identification of more

outstanding all- $d$ -metal Heusler compounds that can serve as potential solid-state refrigeration materials is important. Currently, despite the potential importance of all- $d$ -metal Heusler compounds for solid-state refrigeration application, fully fundamental understanding about the mechanism of martensitic phase transition and anomalous magnetic behavior as well as improved ductility is still lacking.

In this paper, elastic properties, phase stability, and magnetism and correlations among them in all- $d$ -metal  $\text{Ni}_2\text{MnT}$  ( $T = \text{Sc, Ti, V, Cr, Y, Zr, Nb, Mo, Hf, Ta, and W}$ ) Heusler compounds were systematically investigated by first-principles calculations. Furthermore, the mechanism of martensitic phase transition, origin of experimental anomalous magnetic states, and improved ductility are comprehensively discussed. It is believed that this paper can help explore more excellent phase transition materials for solid-state refrigeration, with outstanding mechanical properties from both experimental and theoretical perspectives.

## II. CALCULATION METHODS

Like those in conventional Ni-Mn-based full Heusler compounds, two possible high-ordered cubic structures, i.e.,  $L2_1$ - and  $\text{Hg}_2\text{CuTi}$ -type structures, were considered herein for  $\text{Ni}_2\text{MnT}$  compounds. They are shown in Figs. 1(a) and 1(c), respectively. First, the specific structure was determined for  $\text{Ni}_2\text{MnT}$  compounds. Then based on the obtained structures, the first-principles calculations were performed.

In this paper, structure information, elastic parameters, electronic structure, phonon dispersions, and magnetism of  $\text{Ni}_2\text{MnT}$  compounds were investigated by first-principles calculations. When computing the total energy, the adoption of the full charge density via the exact muffin-tin orbitals (EMTO) method not only improves the calculation efficiency but also leads to high accuracy [62]. Thus, the theoretical calculations were performed mainly by the EMTO method. Moreover, random occupation of atoms in off-stoichiometric Heusler compounds was investigated by using the coherent potential approximation (CPA) [63–65].

Previously, numerous studies have verified its feasibility on alloys and compounds [18,66–70]. The EMTO basis sets included  $s$ ,  $p$ ,  $d$ , and  $f$  orbitals. The muffin-tin potential parameters, i.e., the radius of the muffin-tin potential sphere of the Ni site ( $R_{\text{mt}}^{\text{Ni}}$ ) and the atomic radius of the Ni site ( $R_{\text{ws}}^{\text{Ni}}$ ), were selected as  $R_{\text{mt}}^{\text{Ni}} = 0.95R_{\text{ws}}$  and  $R_{\text{ws}}^{\text{Ni}} = 1.10R_{\text{ws}}$ , where  $R_{\text{ws}}$  is the average Wigner-Seitz radius [71–73]. For the other sublattices, a default setup  $R_{\text{mt}} = R_{\text{ws}}$  was adopted. The soft-core approximation was employed. The Ni  $3d^8 4s^2$ , Mn  $3d^5 4s^2$ , Sc  $3d^1 4s^2$ , Ti  $3d^2 4s^2$ , V  $3d^3 4s^2$ , Cr  $3d^4 4s^2$ , Y  $4d^1 5s^2$ , Zr  $4d^2 5s^2$ , Nb  $4d^3 5s^2$ , Mo  $4d^4 5s^2$ , Hf  $5d^2 6s^2$ , Ta  $5d^3 6s^2$ , and W  $5d^4 6s^2$  states were treated as the valence electron states. The paramagnetic (PM) state was described by using the disordered local moment approach [74,75]. The density of states (DOS) was calculated by adopting the integration method of interpolation. The electron localization function (ELF) and phonon properties (with the method of finite displacement) were calculated directly by the pseudopotential method implemented in the CASTEP code. The size of the supercell was 27 times that of the primitive cell used in the calculation of the phonon properties. The energy convergence criterion and density plane-wave cutoff were set to  $1 \times 10^{-6}$  eV atom $^{-1}$  and 500 eV, respectively. The ultrasoft pseudopotential was selected to describe the interactions between the atomic core and the valence electrons [76]. A Monkhorst-Pack phonon  $q$ -vector grid of  $8 \times 8 \times 8$  was adopted for the calculation of phonon dispersions and phonon DOSs. During all calculations, the exchange-correlation energy of electrons was solved by using GGA-PBE [77]. A Monkhorst-Pack  $k$ -point grid of  $15 \times 15 \times 15$  was used for the electronic properties.

The elastic properties of cubic all- $d$ -metal Heusler compounds can be fully described by using three independent elastic constants, i.e.,  $C_{11}$ ,  $C_{12}$ , and  $C_{44}$ . The  $C_{11}$  and  $C_{12}$  were obtained from the tetragonal shear modulus  $C' = (C_{11} - C_{12})/2$  and the bulk modulus  $B = (C_{11} + 2C_{12})/3$  [62]. In this paper, the equilibrium volume and bulk modulus were determined by fitting the calculated total energy vs volume to a Morse function [78]. The tetragonal shear modulus  $C'$  and the cubic elastic modulus  $C_{44}$  were derived from the volume-conserving orthorhombic deformation and monoclinic deformation, namely,  $\begin{pmatrix} 1+\delta_o & 0 & 0 \\ 0 & 1-\delta_o & 0 \\ 0 & 0 & \frac{1}{1-\delta_o^2} \end{pmatrix}$  and  $\begin{pmatrix} 1 & \delta_m & 0 \\ \delta_m & 1 & 0 \\ 0 & 0 & \frac{1}{1-\delta_m^2} \end{pmatrix}$ , respectively. Six distortions  $\delta = 0.00, 0.01, \dots, 0.05$  were adopted to calculate the total energies  $E(\delta_o)$  and  $E(\delta_m)$ . By fitting the total energies with respect to  $\delta_o$  and  $\delta_m$  as  $E(\delta_o) = E(0) + 2VC'\delta_o^2$  and  $E(\delta_m) = E(0) + 2VC_{44}\delta_m^2$ ,  $C'$  and  $C_{44}$  were obtained. By these methods, the three elastic constants  $C_{11}$ ,  $C_{12}$  and  $C_{44}$  can be determined.

The calculated structure information in Sec. III A, elastic related parameters in Sec. III B, electronic structures in Sec. III C, and magnetic moment in Sec. III E by the EMTO method can be compared with available values calculated by CASTEP and other methods. This consistency provides a prerequisite for the reasonable combination of the CASTEP method to discuss trends in chemical bonding by ELF in Sec. III B and describe phase stability by phonon properties in Secs. III D and III E. The effectively combination of these two methods can help us deeply

understand the mechanism of martensitic phase transition and enhanced ductility in Ni<sub>2</sub>Mn-based all- $d$ -metal Heusler compounds.

### III. RESULTS AND DISCUSSION

#### A. Structure information at the ground states

The conventional full Heusler compounds with the general formula  $X_2YZ$  can crystallize into  $L2_1$ - or Hg<sub>2</sub>CuTi-type structures. Based on available knowledge and findings about all- $d$ -metal Heusler compounds, these two possible crystal structures were considered herein to investigate the ground state properties. The calculated structural information of all- $d$ -metal Ni<sub>2</sub>Mn $T$  Heusler compounds is presented in Fig. 2. In this paper, the relative stability of Ni<sub>2</sub>Mn $T$  compounds within the scope of Heusler structures was determined by the energy difference  $\Delta E$  between the  $L2_1$ - and Hg<sub>2</sub>CuTi-type structures. The positive  $\Delta E$  indicates that the Hg<sub>2</sub>CuTi-type structure is relatively stable compared with the  $L2_1$ -type structure. Otherwise, the  $L2_1$ -type structure is relatively stable. First, Fig. 2(a) illustrates that Ni<sub>2</sub>Mn $T$  compounds containing late transition metals V, Cr, Mo, and W show positive  $\Delta E$ . However, the  $\Delta E$  values for Ni<sub>2</sub>Mn $T$  compounds with early transition metals on the left of periodic table are all negative. It indicates that, like conventional Ni<sub>2</sub>MnGa, Ni<sub>2</sub>Mn $T$  ( $T = \text{Sc, Ti, Y, Zr, Nb, Hf, and Ta}$ ) compounds preferentially remain in  $L2_1$ -type structures. Although Ni<sub>2</sub>Mn $T$  (V, Cr, Mo, and W) compounds consist of similar constituent atoms, they are relatively stable in the Hg<sub>2</sub>CuTi-type structure. The two Ni atoms occupy the A and B sites of the Hg<sub>2</sub>CuTi-type structure. Moreover, Mn and the late transition metal atoms (V, Cr, Mo, and W) reside in C and D sites, respectively. This atomic configuration behavior disobeys the generalized adopted preferential occupation rule for atoms in conventional Heusler compounds [83]. Similar atomic occupation behavior was also observed in other V-based all- $d$ -metal Heusler compounds [80,82]. It is thus speculated that this anomalous atomic configuration in all- $d$ -metal Heusler compounds can bring out unique properties.

The cohesive energy is an important magnitude to predict the viability of experimental synthesis of Ni<sub>2</sub>Mn $T$  compounds [84,85]. Theoretically, the cohesive energy of Ni<sub>2</sub>Mn $T$  ( $T = \text{Sc, Ti, V, Cr, Y, Zr, Nb, Mo, Hf, Ta, and W}$ ) compounds in this paper can be obtained by using  $E_{\text{coh}} = (E_{\text{Ni}}^{\text{iso}} + E_{\text{Mn}}^{\text{iso}} + E_T^{\text{iso}}) - E_{\text{Ni}_2\text{Mn}T}^{\text{Total}}$ , where  $E_{\text{Ni}}^{\text{iso}}$ ,  $E_{\text{Mn}}^{\text{iso}}$ , and  $E_T^{\text{iso}}$  are isolated atomic energy of Ni, Mn, and  $T$  atoms, respectively, and  $E_{\text{Ni}_2\text{Mn}T}^{\text{Total}}$  denotes the total energy of Ni<sub>2</sub>Mn $T$ . To characterize the possibility of experimental synthesis of Ni<sub>2</sub>Mn $T$  compounds within their specific Heusler structures, the cohesive energies were calculated in this paper and are shown in Fig. 2(b). The cohesive energy of Ni<sub>2</sub>MnTi is 21.14 eV/f.u. It has been successfully prepared experimentally in the Heusler-type structure [15]. For well-known Ni<sub>2</sub>MnGa, the calculated cohesive energy is 17.64 eV/f.u. It was also experimentally prepared in the Heusler-type structure [86]. For other Ni<sub>2</sub>Mn $T$  compounds, their cohesive energy ranges from 17.06 to 21.82 eV/f.u. Theoretically, they can be compared with values of both Ni<sub>2</sub>MnTi and Ni<sub>2</sub>MnGa. Thus, Ni<sub>2</sub>Mn $T$  compounds in this paper may be experimentally synthesized at their specific Heusler

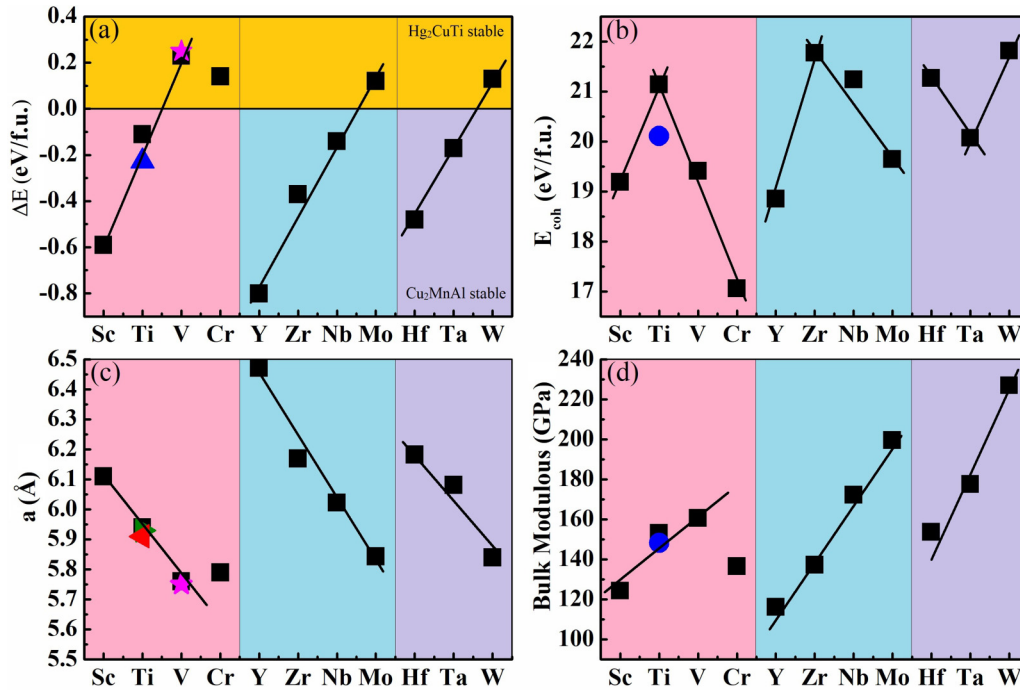


FIG. 2. Ground state structure parameters of  $\text{Ni}_2\text{MnT}$  ( $T = \text{Sc, Ti, V, Cr, Y, Zr, Nb, Mo, Hf, Ta, and W}$ ) at theoretical equilibrium states: (a) The energy difference  $\Delta E$  (in eV/f.u.) between the  $\text{Cu}_2\text{MnAl}$ -type (i.e.,  $L2_1$ -type) and  $\text{Hg}_2\text{CuTi}$ -type structures of  $\text{Ni}_2\text{MnT}$ . The line in figure is to guide the eye to separate the  $\text{Hg}_2\text{CuTi}$ - or  $\text{Cu}_2\text{MnAl}$ -type structure, which corresponds to the positive and negative  $\Delta E$  range, respectively. The other theoretical  $\Delta E$  values of  $\text{Ni}_2\text{MnTi}$  (triangle) from Ref. [79] and  $\text{Ni}_2\text{MnV}$  (star) from Ref. [80] also are listed. (b) Cohesive energies  $E_{\text{coh}}$  (in eV/f.u.) of  $\text{Ni}_2\text{MnT}$ . The other theoretical  $E_{\text{coh}}$  (circle) for  $\text{Ni}_2\text{MnTi}$  is from Ref. [81]. (c) Lattice parameter  $a$  (in Å) of  $\text{Ni}_2\text{MnT}$ . The experimental and/or other theoretical lattice sizes of  $\text{Ni}_2\text{MnT}$  are from Refs. [15,81] ( $\text{Ni}_2\text{MnTi}$ , triangle) and [80,82] ( $\text{Ni}_2\text{MnV}$ , star). (d) Bulk modulus  $B$  (in GPa) of  $\text{Ni}_2\text{MnT}$ . The other theoretical bulk modulus (circle) of  $\text{Ni}_2\text{MnTi}$  is from Ref. [81]. Herein, the calculations in Refs. [80,82] and [79,81] were done by CASTEP and VASP, respectively.

structures. Notably, the crystal structures closely depend on preparation methods and subsequent heat-treatment technology; the further detailed experiments are encouraged to identify the feasible structures for currently studied  $\text{Ni}_2\text{MnT}$  compounds.

Under the abovementioned atomic preferential configuration, the lattice parameters of  $\text{Ni}_2\text{MnT}$  ( $T = \text{Sc, Ti, V, Cr, Y, Zr, Nb, Mo, Hf, Ta, and W}$ ) were obtained by geometry optimization, as shown in Fig. 2(c). For  $T$  atoms in the same period, the lattice parameters of  $\text{Ni}_2\text{MnT}$  decrease with the increase in the number of valence electrons. Comparison of lattice parameters indicates that, for  $\text{Ni}_2\text{MnT}$  compounds, the bulk modulus follows the general trend that a larger bulk modulus corresponds to a smaller lattice size. Thus, the bulk modulus dependent on specific atoms shows inverse change behavior to the lattice size, as shown in Fig. 2(d). Currently, the difference in the lattice size among all- $d$ -metal Heusler compounds originates from the atomic radii of transition metals. For atoms in the same period, their atomic radii decrease with the increase of valence electrons from left to right in the periodic table. The variation mode of the lattice parameter is dependent on specific atoms in  $\text{Ni}_2\text{MnT}$  compounds, which is like the distribution of the atomic size, as shown in Fig. 2(c). It indicates that the inherent atomic size is the main contributor of lattice parameters in all- $d$ -metal Heusler compounds. However, in conventional Heusler compounds, the  $p$ - $d$  covalent hybridization between 1nn main group el-

ements and transition metals is also crucially important in determining the lattice size. In the main group element-rich compounds such as cubic  $\text{Mn}_{2-x}\text{CoGa}_{1+x}$  [87] and tetragonal  $\text{Ni}_{50}\text{Mn}_{5+x}\text{Ga}_{35-x}\text{Cu}_{10}$  [88] systems, the enhanced covalent hybridization results in kink behavior in the lattice parameter dependence of the main group element composition curve. Basically, in conventional Heusler compounds, the inherent atomic radius and the  $p$ - $d$  hybridization codetermine the lattice size. Herein, in all- $d$ -metal Heusler compounds, the main group elements are fully replaced with transition metals. As a result, covalent hybridization gets weakened, and it does not dominate the lattice size anymore. Experimentally, this weakened hybridization also lowered the atomic order level from perfect  $L2_1$ -type/ $\text{Hg}_2\text{CuTi}$ -type structure to partially disordered B2 structure or the mixture of  $L2_1$ -type/ $\text{Hg}_2\text{CuTi}$ -type and B2 structures [15]. For simplification, in current research, only fully ordered structures are considered. Theoretically, the calculated lattice constant of  $\text{Ni}_2\text{MnTi}$  was found to be 5.94 Å, which is very close to the experimentally observed value of 5.93 Å [15]. The marginal difference may originate from the disordered arrangement of atoms, which was experimentally observed in  $\text{Ni}_2\text{MnTi}$ . Furthermore, the theoretical lattice size of  $\text{Ni}_2\text{MnV}$  (5.76 Å) based on the EMTO method is consistent with the published values (5.75 Å) obtained by using the CASTEP code [80,82]. These agreements suggest that current calculations are practical. Based on these reliable calculations, the elastic properties, lattice dynamics, electronic structure,

TABLE I. Calculated elastic constants  $C_{ij}$  (in GPa), shear modulus  $G$  (in GPa), and melting temperature  $T_m$  (in K) for all- $d$ -metal Heusler compounds  $\text{Ni}_2\text{Mn}T$  ( $T = \text{Sc, Ti, V, Cr, Y, Zr, Nb, Mo, Hf, Ta, and W}$ ). For comparison, the related parameters of well-known conventional  $\text{Ni}_2\text{MnGa}$  are also included here. Comparison done with experiments or previous calculations, wherever data are available as given.

	$C_{11}$	$C_{12}$	$C_{44}$	$G_V$	$G_R$	$G_H$	$T_m$
$\text{Ni}_2\text{MnSc}$	145.12	113.86	68.77	47.51	29.14	38.33	$1410.81 \pm 300$
$\text{Ni}_2\text{MnTi}$	167.41	146.14	82.71	53.88	22.29	38.08	$1542.59 \pm 300$
	157.3 <sup>a</sup>	143.6 <sup>a</sup>	73.7 <sup>a</sup>				
$\text{Ni}_2\text{MnV}$	166.52	157.74	126.71	77.78	10.43	44.10	$1537.28 \pm 300$
$\text{Ni}_2\text{MnCr}$	128.59	140.45	120.58	69.98	-16.01	26.99	$1313.10 \pm 300$
$\text{Ni}_2\text{MnY}$	142.49	103.01	25.31	23.08	22.74	22.91	$1395.25 \pm 300$
$\text{Ni}_2\text{MnZr}$	156.70	127.56	55.19	38.94	26.09	32.52	$1479.24 \pm 300$
$\text{Ni}_2\text{MnNb}$	189.78	163.52	75.70	50.67	26.05	38.36	$1674.81 \pm 300$
$\text{Ni}_2\text{MnMo}$	190.21	204.17	138.59	80.36	-18.89	30.74	$1677.31 \pm 300$
$\text{Ni}_2\text{MnHf}$	180.20	140.53	58.67	43.14	32.90	38.02	$1618.16 \pm 300$
$\text{Ni}_2\text{MnTa}$	185.37	173.78	68.10	43.18	12.85	28.01	$1648.71 \pm 300$
$\text{Ni}_2\text{MnW}$	216.68	232.20	155.66	90.29	-20.96	34.67	$1833.80 \pm 300$
	160.20	155.48	109.80	66.82	5.71	36.27	$1499.78 \pm 300$
	163.6 <sup>a</sup>	155.4 <sup>a</sup>	106.7 <sup>a</sup>				
	152 <sup>b</sup>	143 <sup>b</sup>	103 <sup>b</sup>				
$\text{Ni}_2\text{MnGa}$	162.2 <sup>c</sup>	157.07 <sup>c</sup>	108.72 <sup>c</sup>				
	166.7 <sup>d</sup>	151.7 <sup>d</sup>	113.4 <sup>d</sup>				
	165.4 <sup>e</sup>	149.4 <sup>e</sup>	127.0 <sup>e</sup>				

<sup>a</sup>Reference [81] by VASP.

<sup>b</sup>Reference [89] by experiments.

<sup>c</sup>Reference [90] by CASTEP.

<sup>d</sup>Reference [91] by EMTO.

<sup>e</sup>Reference [92] by EMTO.

and magnetism were investigated, and the discussion is presented in the following sections.

### B. Elastic parameters determined martensitic phase transition and ductile properties

Based on the atomic configuration at the equilibrium states, elastic parameters of all- $d$ -metal Heusler compounds  $\text{Ni}_2\text{Mn}T$  ( $T = \text{Sc, Ti, V, Cr, Y, Zr, Nb, Mo, Hf, Ta, and W}$ ) were calculated and listed in Table I. For comparative analysis, elastic parameters of well-studied  $\text{Ni}_2\text{MnGa}$  are also included herein. Table I presents the high consistency between our data about  $\text{Ni}_2\text{MnGa}$  and previous results, which again indicates that current calculations are valid and suitable to deal with Ni-Mn-based Heusler compounds.

All- $d$ -metal Heusler compounds  $\text{Ni}_2\text{Mn}T$  ( $T = \text{Sc, Ti, V, Cr, Y, Zr, Nb, Mo, Hf, Ta, and W}$ ) exhibit cubic structures. They have three independent elastic constants, i.e.,  $C_{11}$ ,  $C_{12}$ , and  $C_{44}$ . Their sufficient and necessary mechanical stability conditions are given by (1)  $C_{11} > 0$ , (2)  $C_{44} > 0$ , (3)  $C_{11} + 2C_{12} > 0$ , and (4)  $C' = \frac{C_{11} - C_{12}}{2} > 0$  [93]. The elastic constants summarized in Table I indicate that all the compounds satisfy the first three conditions well. The last condition of the tetragonal shear elastic constant  $C'$  is usually used to characterize the martensitic phase transition temperature ( $T_M$ ), as shown in the equation of  $\frac{dT_M}{dC'} = -\frac{1}{\beta C'}$ , where  $\beta = \frac{dC'}{TdT}$  [94]. Owing to softening, the  $\beta$  factor is always positive, so that the right side of the abovementioned equation is negative. Therefore, higher  $C'$  corresponds to lower  $T_M$ . This equation has been successfully adopted to discuss the possibility and tendency of martensitic phase transition in Ni-Mn-based

Heusler compounds [95,96]. Furthermore, a perfect cubic Heusler compound is isotropic. The two shear moduli  $C'$  and  $C_{44}$  are equal, and the value of the Zener anisotropy factor ( $A$ ) ( $= \frac{2C_{44}}{C_{11} - C_{12}} = \frac{C_{44}}{C'}$ ) is 1. Commonly,  $A$  is considered a parameter to characterize the elastic anisotropy strength of cubic compounds. However, herein, it is used in a generalized manner to characterize the phase stability. Theoretically, any deviation of  $A$  from 1 indicates the instability of the cubic compounds. When the value of  $A$  is  $> 2$ , the cubic compounds get transformed to low-symmetry martensitic phase [97,98].

In this paper, the tetragonal shear elastic constant  $C'$  and the Zener anisotropy factor ( $A$ ) of  $\text{Ni}_2\text{Mn}T$  ( $T = \text{Sc, Ti, V, Cr, Y, Zr, Nb, Mo, Hf, Ta, W, and Ga}$ ) compounds at theoretical equilibrium states were calculated, and the corresponding results are shown in Fig. 3. Their validity as predictors of the phase stability has been verified, and they are still widely adopted in Ni-Mn-based Heusler compounds [100,101]. Obviously, compounds with late transition metals ( $T = \text{Cr, Mo, and W}$ ) show negative  $C'$  and  $A$ , which indicates that martensitic phase transition can occur in them. Even though the values of  $C'$  and  $A$  in  $\text{Ni}_2\text{MnV}$  are both positive, they are comparable with the theoretical values of well-known  $\text{Ni}_2\text{MnGa}$  (see Fig. 3), which are very consistent with the published experimental data and theoretical results [89,90,92,100,102]. Thus, martensitic phase transition can also be expected in  $\text{Ni}_2\text{MnV}$ . In contrast, for compounds with early transition metals ( $T = \text{Sc, Y, Zr, Nb, and Hf}$ ), their higher  $C'$  and lower  $A$  support more stable cubic structures. For  $\text{Ni}_2\text{MnTi}$  and  $\text{Ni}_2\text{MnTa}$  compounds, their medium values of  $C'$  and  $A$  indicate that the two compounds remain on the edge of the martensitic phase transition. Under certain external stimuli

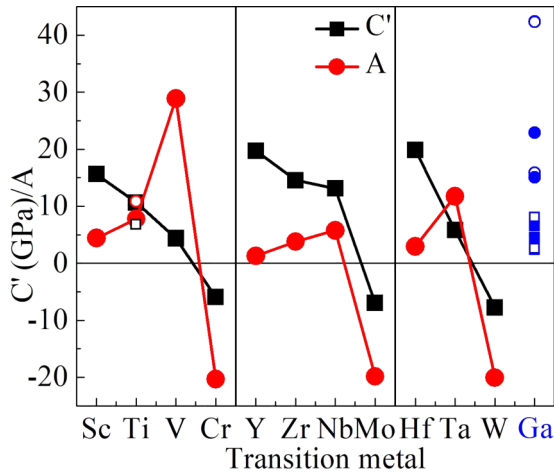


FIG. 3. Tetragonal shear elastic constant  $C'$  (in GPa) and Zener anisotropy factor ( $A$ ) of  $\text{Ni}_2\text{Mn}T$  ( $T = \text{Sc}, \text{Ti}, \text{V}, \text{Cr}, \text{Y}, \text{Zr}, \text{Nb}, \text{Mo}, \text{Hf}, \text{Ta}, \text{W}$ ) compounds at theoretical equilibrium states. These two parameters describe the possibility of martensitic phase transition. Other theoretical  $C'$  and  $A$  of  $\text{Ni}_2\text{MnTi}$  are from Ref. [81] by VASP. For comparative analysis,  $C'$  and  $A$  (in both, open blue circles are for theory, solid ones are for experiments) of  $\text{Ni}_2\text{MnGa}$  are also collected and included herein. These values are from this paper and Refs. [89,90,92,99]. Theoretical values in Refs. [90,92] and [92] were obtained by CASTEP and EMTO methods, respectively. The data in Refs. [89,99] are from experiments.

such as high pressure and magnetic field, martensitic phase transition can be induced.

For conventional Heusler compounds, the directional covalent bonding between main group elements and transition metal atoms leads to inherent brittleness, seriously affecting their service performance. To improve the ductile properties and satisfy more flexible applications, one ideal and effective way is to design Heusler compounds with less covalent bonding. Thus, all- $d$ -metal Heusler compounds have attracted more attention and have become popular. Other than the outstanding functional properties, great improvements in ductility can also be expected. For all- $d$ -metal Heusler compounds  $\text{Ni}_2\text{Mn}T$  ( $T = \text{Sc}, \text{Ti}, \text{V}, \text{Cr}, \text{Y}, \text{Zr}, \text{Nb}, \text{Mo}, \text{Hf}, \text{Ta}, \text{W}$ ), the ductile properties are evaluated from Poisson's ratio ( $\nu$ ) and the  $B/G$  ratio, followed by the discussion on interatomic orbital hybridization style from calculated ELF. Herein, these adopted methods and theories are still widely used to reasonably evaluate the ductility and reveal its physical origin in materials, including but not limited to cubic Heusler compounds [18,103–106].

By using the Voigt-Reuss-Hill (VRH) approximation, elastic moduli were obtained herein from the elastic constants of a single crystal [107–109]. With the help of Voigt and Reuss approximations, the bulk and shear moduli of cubic Heusler compounds were derived by  $B_V = B_R = \frac{C_{11} + 2C_{12}}{3}$ ,  $G_V = \frac{(C_{11} - C_{12} + 3C_{44})}{5}$ , and  $G_R = \frac{5(C_{11} - C_{12})C_{44}}{4C_{44} + 3(C_{11} - C_{12})}$ . Based on VRH approximation, average bulk and shear moduli ( $B_H = \frac{B_V + B_R}{2}$  and  $G_H = \frac{G_V + G_R}{2}$ ) can be deduced. The detailed shear moduli are listed in Table I. Furthermore, by  $\nu = \frac{3B - 2G}{2(3B + G)}$ , Poisson's ratio (details in Fig. 4) was obtained. First, the ductility of  $\text{Ni}_2\text{Mn}T$  compounds was directly evaluated from Poisson's

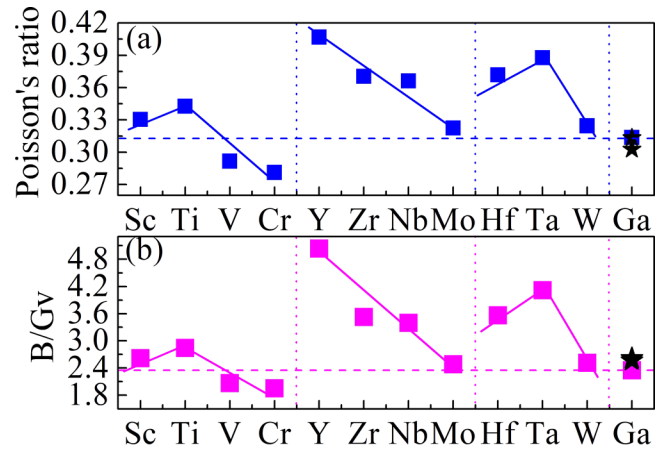


FIG. 4. Parameters describing the ductility of  $\text{Ni}_2\text{Mn}T$  ( $T = \text{Sc}, \text{Ti}, \text{V}, \text{Cr}, \text{Y}, \text{Zr}, \text{Nb}, \text{Mo}, \text{Hf}, \text{Ta}, \text{W}$ ) compounds. (a) Poisson's ratio, (b)  $B/G_V$  ratio. For comparative analysis, parameters of well-known  $\text{Ni}_2\text{MnGa}$  are also included. The other Poisson's ratio and  $B/G_V$  ratio of  $\text{Ni}_2\text{MnGa}$  are obtained from Refs. [90,110] and [90,111], respectively. Herein,  $\text{Ni}_2\text{Mn}T$  compounds with early transition metals show better ductility than conventional  $\text{Ni}_2\text{MnGa}$  and  $\text{Ni}_2\text{Mn}T$  compounds with late transition metals. Theoretical values in Refs. [90,111] were obtained by CASTEP and VASP methods, respectively. The data in Ref. [110] are from experiment.

ratio. In general, materials with  $\nu > \frac{1}{3}$  are ductile, whereas those with  $\nu < \frac{1}{3}$  are brittle. Figure 4(a) illustrates that the calculated  $\nu$  of well-studied  $\text{Ni}_2\text{MnGa}$  ( $=0.31$ ) is less than the empirical boundary value, indicating its brittle nature. This claim agrees with previous observations [14,112] and further proves the validity of the current calculations. In contrast, all- $d$ -metal Heusler compounds  $\text{Ni}_2\text{Mn}T$  ( $T = \text{Sc}, \text{Ti}, \text{V}, \text{Cr}, \text{Y}, \text{Zr}, \text{Nb}, \text{Mo}, \text{Hf}, \text{Ta}, \text{W}$ ) show different ductility. Compounds with early transition metals, i.e., Sc, Ti, Y, Zr, Nb, Hf, and Ta, possess higher  $\nu$  values, which are all above critical value and consequently exhibit better ductility. Instead, compounds with late transition metals, i.e., V, Cr, Mo, and W, show lower  $\nu$  values, which are below the separating line and drive these compounds to be brittle. Second, based on Pugh's research, the  $B/G$  ratio was proposed as another parameter to separate brittle and ductile materials [113]. The boundary value to separate ductile and brittle materials is  $\sim 1.75$ . Commonly for materials, the higher the  $B/G$  ratio, the more the ductile nature and vice versa [113]. A previous related study about Ni-Mn-based Heusler compounds indicates that the  $B/G_V$  ratio based on  $G_V$  can serve as a better parameter to separate ductile and brittle systems [90,103]. For  $\text{Ni}_2\text{Mn}T$  ( $T = \text{Sc}, \text{Ti}, \text{V}, \text{Cr}, \text{Y}, \text{Zr}, \text{Nb}, \text{Mo}, \text{Hf}, \text{Ta}, \text{W}$ ) compounds investigated herein, all  $B/G_V$  ratios are above the critical value. Thus, these all- $d$ -metal Heusler compounds can be considered ductile materials from the perspective of the  $B/G_V$  ratio. This is different from the judgment by Poisson's ratio. Owing to the fact that the critical value of Poisson's ratio used herein was not defined explicitly in the literature and only based on properties reported for the materials tabulated by Frantsevich [114], it is an empirical separating line practically. Finally, although the classification is imprecise, both the  $B/G_V$  ratio and Poisson's ratio of compounds with early

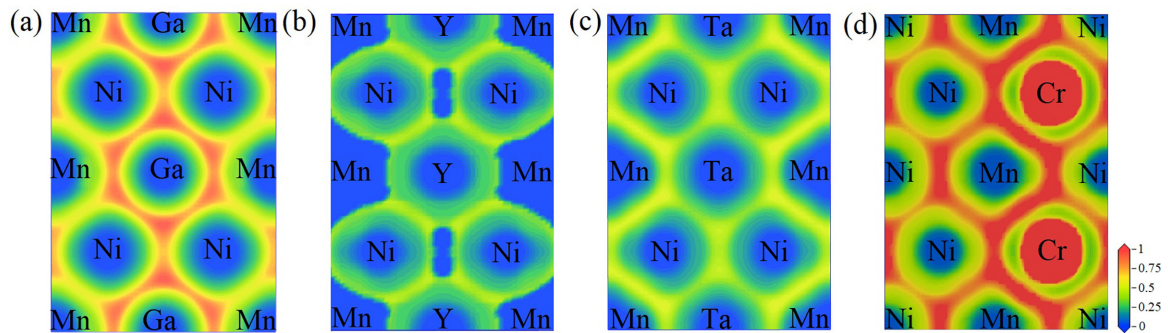


FIG. 5. Electron localization function (ELF) distributions of (a)  $\text{Ni}_2\text{MnGa}$ , (b) isostructural  $\text{Ni}_2\text{MnY}$ , and (c)  $\text{Ni}_2\text{MnTa}$  in  $\text{Cu}_2\text{MnAl}$ -type structure, and (d)  $\text{Ni}_2\text{MnCr}$  in  $\text{Hg}_2\text{CuTi}$ -type structure on (110) plane of unit cell. The scale bar from blue to red corresponds to increasing electron localization (namely, increasing covalent bonding). Compared with Ni-Y and Ni-Ta atom pairs in  $\text{Ni}_2\text{MnY}$  and  $\text{Ni}_2\text{MnTa}$ , there are higher values between Ni-Ga atom pairs in  $\text{Ni}_2\text{MnGa}$ .

transition metals are higher than those of conventional  $\text{Ni}_2\text{MnGa}$  and  $\text{Ni}_2\text{MnT}$  compounds consisting of late transition metals V, Cr, Mo, and W. This is clearly illustrated in Figs. 4(a) and 4(b).

Inherently, ductile performance in solids can be directly reflected from the angular character of bonding between constituent atoms. Physically, covalent bonding is directional with the angular character and responsible for brittleness and lower ductility in solids [115]. To discuss trends in chemical bonding and reveal the mechanism of enhanced ductility in  $\text{Ni}_2\text{MnT}$  compounds with early transition metals compared with  $\text{Ni}_2\text{MnGa}$ , representative ELF [116,117] of  $\text{Ni}_2\text{MnGa}$ ,  $\text{Ni}_2\text{MnY}$ ,  $\text{Ni}_2\text{MnTa}$ , and  $\text{Ni}_2\text{MnCr}$  were calculated and shown in Fig. 5. The ELF is often used to investigate the nature of chemical bonding in different types of materials [88,118]. For instance, in  $\text{Cu}_2\text{MnAl}$ -type  $\text{Ni}_2\text{MnGa}$ ,  $\text{Ni}_2\text{MnY}$ , and  $\text{Ni}_2\text{MnTa}$ , the Ga, Y, and Ta atoms locate in the cubic crystal field with 8 Ni atoms as their 1st neighbors. It provides a convenient platform to investigate the change of chemical bonding by comparing Ni-Ga, Ni-Y, and Ni-Ta bonding. The topological analysis of the ELF indicates the occurrence of strong covalent hybridization between Ga and Ni atoms in  $\text{Ni}_2\text{MnGa}$ , which results from the  $p$ - $d$  orbital hybridization between  $p$  electrons of Ga and  $d$  electrons of Ni atoms. It shows an obvious directional character. The brittleness in the intermetallic compound  $\text{Ni}_2\text{MnGa}$  could intrinsically originate from this interatomic covalent hybridization. However, in all- $d$ -metal Heusler compounds  $\text{Ni}_2\text{MnY}$  and  $\text{Ni}_2\text{MnTa}$ , the interatomic covalent hybridization strength between Y/Ta and Ni atoms decreases significantly. Correspondingly, the metallic bonding becomes dominant and does not remain directional anymore. Thus, the increased metallic bonding could be largely responsible for the enhancement of ductility in  $\text{Ni}_2\text{MnY}$  and  $\text{Ni}_2\text{MnTa}$ . Similar comparisons of ELF between  $\text{Ni}_2\text{MnGa}$  and  $\text{Ni}_2\text{MnTi}$  compounds were also investigated, as presented in Fig. S4 in the Supplemental Material [25] and Refs. [61,81]. However, in  $\text{Hg}_2\text{CuTi}$ -type  $\text{Ni}_2\text{MnCr}$ , it is another case. The interatomic covalent hybridization in  $\text{Hg}_2\text{CuTi}$ -type  $\text{Ni}_2\text{MnCr}$  is obviously stronger compared with  $\text{Ni}_2\text{MnGa}$ , which results in lower ductile parameters, as presented in Fig. 4.

Practically and importantly, due to the relatively low energy of metallic bonding compared with covalent bonding,

the nucleation and propagation of dislocation in real metallic bonding materials become relatively easy since the energy consumed to overcome bond distortion and breaking decreases [30]. It is speculated that this may result in higher density of dislocation and corresponding increase in plastic performance of all- $d$ -metal Heusler compounds  $\text{Ni}_2\text{MnT}$  ( $T = \text{Sc, Ti, Y, Zr, Nb, Hf, and Ta}$ ). Combining the calculated ELF, significant enhancement in ductile properties was experimentally observed in Ni-Mn-Ti compounds [18,60,61], which can be mainly ascribed to improved metallic bonding. Thus, based on chemical bonding engineering, tuning the bonding type of interatomic coupling has been proven to be an effective approach to design high-ductility multiple functional materials.

For engineering applications, the thermodynamic properties are also significantly important. Herein, this issue is discussed from the perspective of the melting point. Empirically, it can be obtained by using an approximate equation of  $T_m = 553 \text{ K} + \frac{5.91 \text{ K}}{\text{GPa}} C_{11}$ , in which  $C_{11}$  is elastic constant, as presented in Table I. The standard error is about  $\pm 300 \text{ K}$  [119]. Notably, good agreements with experiments were achieved by this method to predict melting points in cubic compounds [120]. Herein, the calculated melting point for  $\text{Ni}_2\text{MnGa}$  is  $1499.78 \pm 300 \text{ K}$ . Its experimental value is  $1380 \text{ K}$  [121], which lies well in the confidence interval of the calculated value. Again, this agreement confirms the feasibility of our calculation. Based on this method, melting points of  $\text{Ni}_2\text{MnT}$  ( $T = \text{Sc, Ti, V, Cr, Y, Zr, Nb, Mo, Hf, Ta, and W}$ ) compounds were obtained and are listed in Table I. In general, a higher melting point indicates a stronger resistance to temperature. Thus, the high melting point values of  $\text{Ni}_2\text{MnT}$  ( $T = \text{Sc, Ti, V, Cr, Y, Zr, Nb, Mo, Hf, Ta, and W}$ ) compounds indicate that their ground state properties can be preserved as much as possible when temperature is increased [115,122]. It enables these compounds to serve more flexible situations. The melting points of  $\text{Ni}_2\text{MnT}$  predicted in this paper call for further experimental study for verification. Furthermore, systematic investigations are required to increase the accuracy of the method to predict the melting point of Heusler compounds.

For a reliable evaluation of the martensitic phase transition in cubic compounds, electronic structure and phonon dispersion were also involved and are discussed in the following

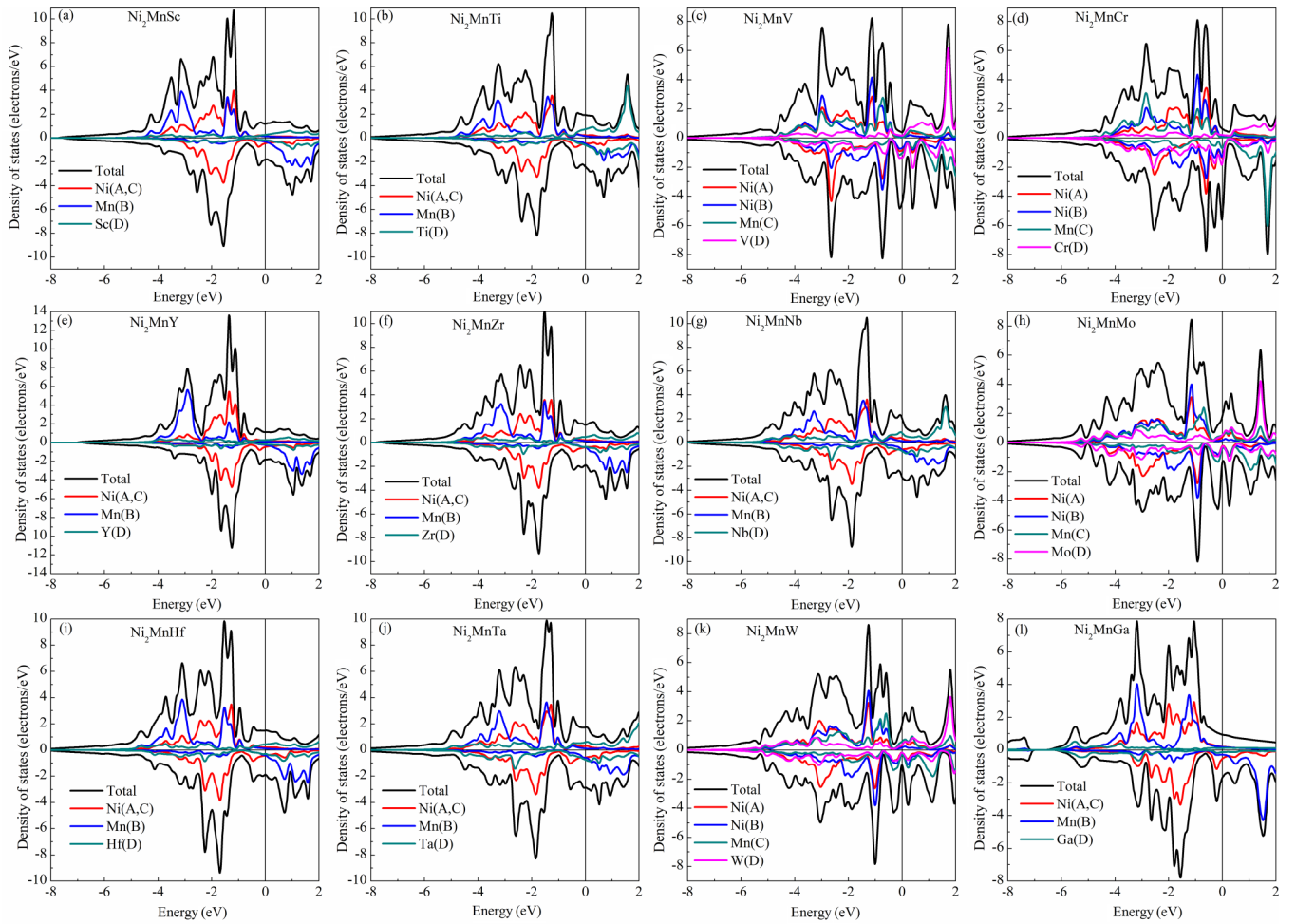


FIG. 6. (a)–(k) Calculated spin-projected density of states (DOS) plots of  $\text{Ni}_2\text{Mn}T$  ( $T = \text{Sc}, \text{Ti}, \text{V}, \text{Cr}, \text{Y}, \text{Zr}, \text{Nb}, \text{Mo}, \text{Hf}, \text{Ta},$  and  $\text{W}$ ) compounds. The total and partial DOSs for  $d$  electrons of atoms on specific crystalline sites are included. The upper halves of each panel display the spin-up states. For comparison, the DOSs of  $\text{Ni}_2\text{MnGa}$  are illustrated in (l). Herein, the DOSs of  $\text{Ni}_2\text{MnGa}$ ,  $\text{Ni}_2\text{MnV}$ , and  $\text{Ni}_2\text{MnTi}$  are like those by CASTEP (for  $\text{Ni}_2\text{MnGa}$  in Ref. [90] and  $\text{Ni}_2\text{MnV}$  in Ref. [82]) and VASP (for  $\text{Ni}_2\text{MnTi}$  in Ref. [81]).

sections. Importantly, comprehensive analysis of these factors can aid in achieving consistent conclusions about the martensitic phase transition of cubic  $\text{Ni}_2\text{Mn}T$  compounds.

### C. Electronic structures

The electronic structure information at the ground state, such as DOS, depends on the specific crystal structure and directly reflects physical properties of materials. In conventional  $\text{Ni}_2\text{Mn}$ -based Heusler compounds, the structural instability of the cubic phase was characterized by the typical peak structure below the Fermi level ( $E_F$ ) in the spin-down band. This peak structure predominantly stemmed from Ni  $3d$   $eg$  orbitals (for details, see Refs. [123–126]). In conventional  $\text{Ni}_2\text{Mn}$ -based Heusler compounds, it can result in the band Jahn-Teller instability of the cubic structures [126–128]. Thus, this peak structure can usually serve as an intuitive criterion in determining the phase stability of Heusler compounds. To further clarify the physical origin of the stability in cubic  $\text{Ni}_2\text{Mn}T$  compounds, the spin-polarized total and partial DOSs were calculated herein, and the corresponding results are shown in Fig. 6 for analysis. Herein, the DOSs of all compounds show

peak structures just below  $E_F$  in the spin-down band (0 to  $-0.6$  eV). Dependent on specific transition metals, the DOSs of  $\text{Ni}_2\text{Mn}T$  ( $T = \text{Sc}, \text{Ti}, \text{V}, \text{Cr}, \text{Y}, \text{Zr}, \text{Nb}, \text{Mo}, \text{Hf}, \text{Ta},$  and  $\text{W}$ ) compounds near  $E_F$  manifest diverse distributions. Among all compounds studied herein,  $\text{Ni}_2\text{MnV}$ ,  $\text{Ni}_2\text{MnCr}$ ,  $\text{Ni}_2\text{MnMo}$ , and  $\text{Ni}_2\text{MnW}$  show typical peak structures below  $E_F$ , as observed in well-known  $\text{Ni}_2\text{MnGa}$ . Moreover, these peaks possess relatively high values compared with the remaining  $\text{Ni}_2\text{Mn}T$  compounds investigated in this paper. Different from the case of  $\text{Ni}_2\text{MnGa}$ , the  $3d$  states of Ni and  $T$  ( $=\text{V}, \text{Cr}, \text{Mo},$  and  $\text{W}$ ) atoms all contribute to these prominent peak structures below  $E_F$ . These prominent peak structures near  $E_F$ , more often than not, cause high energy and consequently lead to poor structural stability in the parent cubic phase [81, 127, 129]. A tetragonal distortion is an effective way to reduce the peak structures near  $E_F$  and avoid the instability [130]. Thus, the most likely tetragonal distortion can be expected in  $\text{Ni}_2\text{MnV}$ ,  $\text{Ni}_2\text{MnCr}$ ,  $\text{Ni}_2\text{MnMo}$ , and  $\text{Ni}_2\text{MnW}$ . In high-symmetry Heusler compounds, the equivalent  $k$  points in the Brillouin zone lead to the same band energies. Owing to tetragonal distortion, some originally equivalent  $k$  points become inequivalent, leading to unequal energy values at



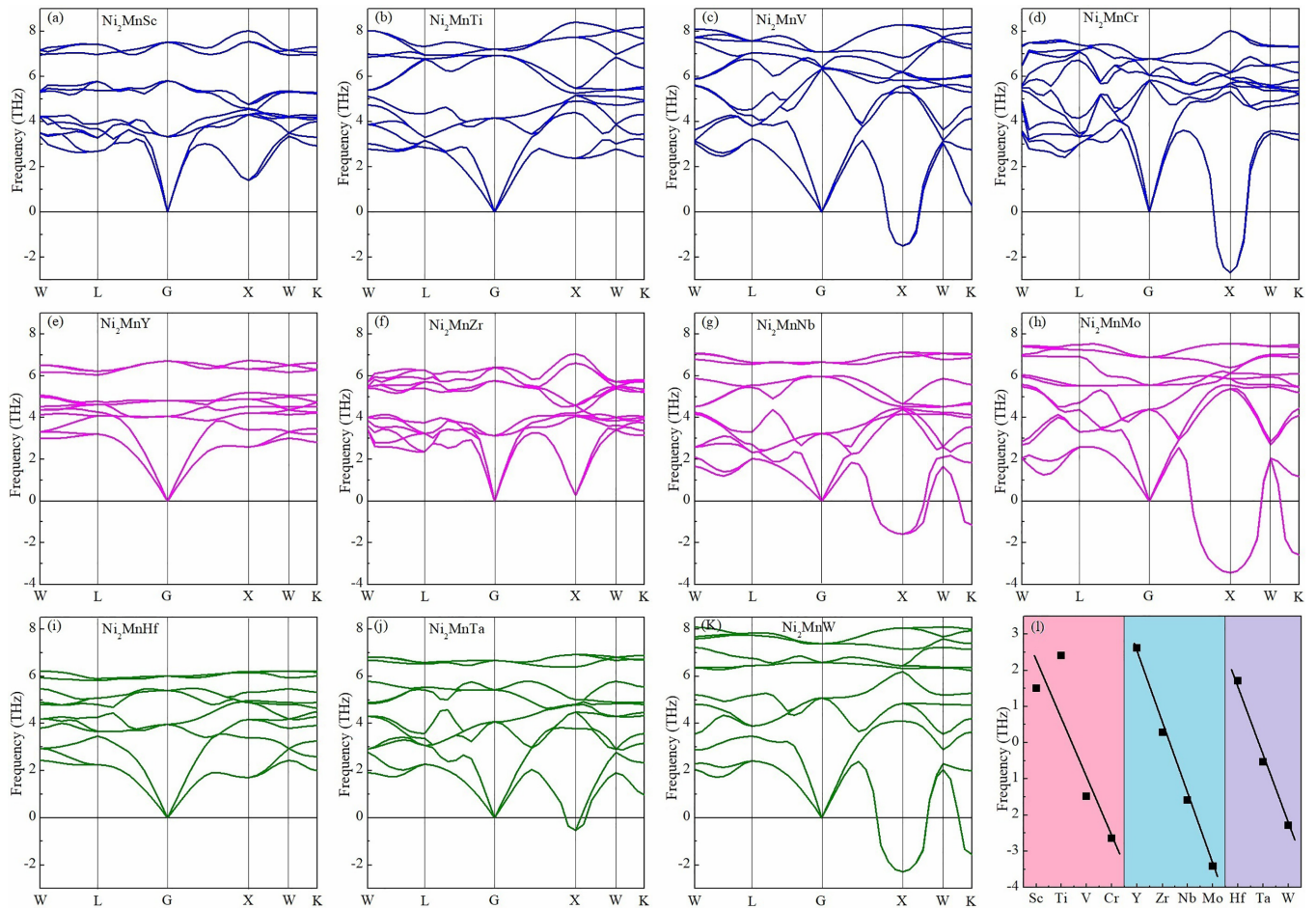


FIG. 7. (a)–(k) The calculated phonon dispersions of  $\text{Ni}_2\text{Mn}T$  ( $T = \text{Sc, Ti, V, Cr, Y, Zr, Nb, Mo, Hf, Ta, and W}$ ) compounds. Imaginary frequencies of the unstable modes are shown in negative frequency range. The high-symmetry  $q$  point path in the Brillouin zone is chosen as  $W(0.5, 0.25, 0.75) \rightarrow L(0.5, 0.5, 0.5) \rightarrow G(0, 0, 0) \rightarrow X(0.5, 0, 0.5) \rightarrow W(0.5, 0.25, 0.75) \rightarrow K(0.375, 0.375, 0.75)$ . Frequency values at the  $X$  point dependent on specific transition metal atoms are presented in (l).

these points and consequently smooth distribution of DOSs near  $E_F$  [129]. Moreover, due to tetragonal distortion, the symmetry of the system gets lowered. The degenerate bands in the cubic system do not remain degenerate in the tetragonal system. Bands that overlap in the direction of contraction become broader. A repopulation of electrons was allowed to avoid the electronic instability of the band Jahn-Teller type [131,132]. All these effects can smoothen the peak structures of DOSs near  $E_F$  and lower the energy of the cubic phase.

Owing to similar electronic configurations of Sc, Ti, Y, Zr, Nb, Hf, and Ta, the DOSs of  $\text{Ni}_2\text{Mn}T$  ( $T = \text{Sc, Ti, Y, Zr, Nb, Hf, and Ta}$ ) compounds distribute in a similar manner in the entire energy span. Their peak structures below  $E_F$  are shallow with relatively low values. Thus, comparison with  $\text{Ni}_2\text{MnV}$ ,  $\text{Ni}_2\text{MnCr}$ ,  $\text{Ni}_2\text{MnMo}$ , and  $\text{Ni}_2\text{MnW}$  indicates that their peak structure characteristics are weaker. The cubic structures in these compounds are relatively stable. All judgments about the phase stability of  $\text{Ni}_2\text{Mn}T$  ( $T = \text{Sc, Ti, V, Cr, Y, Zr, Nb, Mo, Hf, Ta, and W}$ ) compounds are consistent with the evaluation based on the elastic moduli, strongly supporting the current assessment.

#### D. Lattice dynamics properties

For an in-depth investigation of the dynamic stability of  $\text{Ni}_2\text{Mn}T$  ( $T = \text{Sc, Ti, V, Cr, Y, Zr, Nb, Mo, Hf, Ta, and W}$ ) compounds, the phonon dispersions along the symmetry line at ground state were calculated, and the corresponding results are shown in Fig. 7. In phonon dispersion curves of  $\text{Ni}_2\text{Mn}T$  ( $T = \text{Sc, Ti, V, Cr, Y, Zr, Nb, Mo, Hf, Ta, and W}$ ) compounds, the most important phenomenon is the softening of two transverse acoustic (TA) phonon modes along  $G$ - $X$ - $W$  directions, all reaching a local minimum at the  $X$  point of the Brillouin zone. Figure 7(l) demonstrates that these minima dependent on specific transition metal atoms gradually become weak with the increase of valence electrons in each period of periodic table. The minima of  $\text{Ni}_2\text{Mn}T$  with late transition metal atoms locate at obvious imaginary frequencies zone. Compared with experimental phonon dispersions in Ni-Mn-Ga compounds [133,134], these anomalies strongly indicate that cubic phases of  $\text{Ni}_2\text{Mn}T$  with late transition metal atoms are unstable to shear deformation corresponding to the eigenvector of TA modes. Possible martensitic phase transition can be expected in them. Thus, these abnormal softening behaviors of TA modes closely relate to

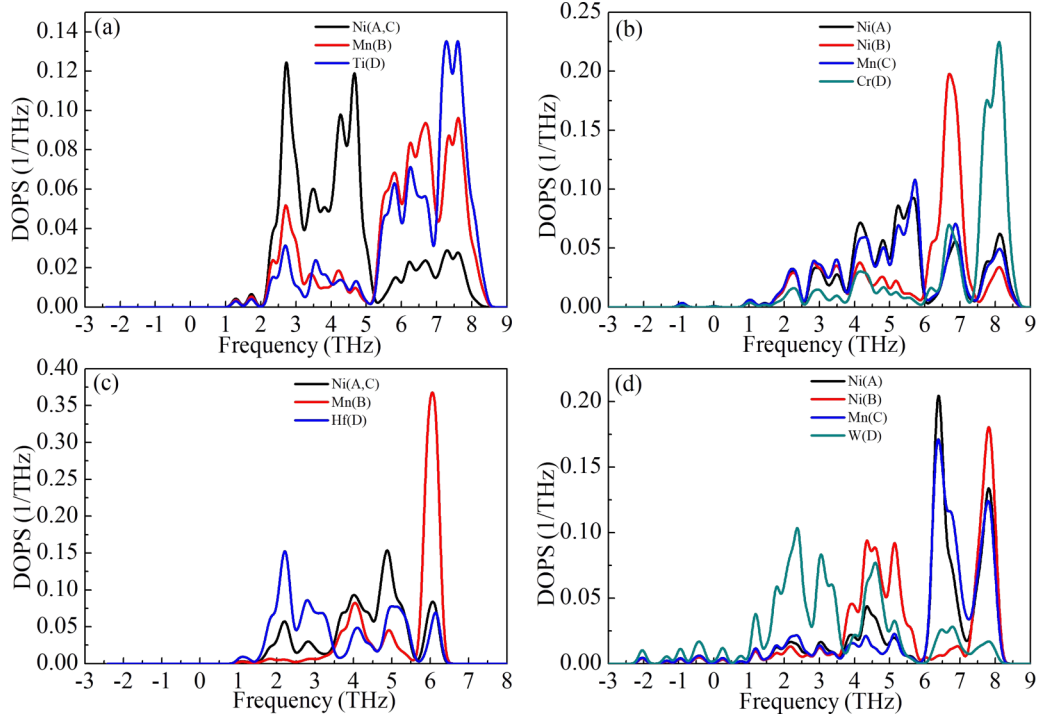


FIG. 8. (a)–(d) The selected site projected phonon density of states of  $\text{Ni}_2\text{MnTi}$ ,  $\text{Ni}_2\text{MnCr}$ ,  $\text{Ni}_2\text{MnHf}$ , and  $\text{Ni}_2\text{MnW}$ , respectively, showing the distribution of amplitudes of atoms in the frequency range. In  $\text{Ni}_2\text{MnCr}$  and  $\text{Ni}_2\text{MnW}$ , the visible anomalous behavior of medium frequencies phonon dispersions of Mn(C), which lie below the optical modes of those heavier Ni(B) atoms, is to act via hybridization as a repulsive force on the transverse acoustic modes.

$C'$ . The lower the value of  $C'$ , the more unstable the cubic phase [115]. The relative phase stability of  $\text{Ni}_2\text{MnT}$  ( $T = \text{Sc}, \text{Ti}, \text{V}, \text{Cr}, \text{Y}, \text{Zr}, \text{Nb}, \text{Mo}, \text{Hf}, \text{Ta}, \text{and W}$ ) compounds is consistent with the judgment made based on elastic parameters.

To further identify the origin of these anomalous softening behaviors, the partial phonon DOSs of  $\text{Ni}_2\text{MnTi}$ ,  $\text{Ni}_2\text{MnCr}$ ,  $\text{Ni}_2\text{MnHf}$ , and  $\text{Ni}_2\text{MnW}$  were analyzed and are presented in Fig. 8. The sequence of atomic contribution to phonon dispersions is mainly separated by mass. Normally, heavier atoms dominate the low-frequency phonon modes and vice versa [128,135]. In  $\text{Ni}_2\text{MnTi}$  and  $\text{Ni}_2\text{MnHf}$ , the atomic contribution to phonon dispersions shows the normal sequence, resulting in stable states without obvious softening. However, in both  $\text{Ni}_2\text{MnCr}$  and  $\text{Ni}_2\text{MnW}$ , Mn(C) atoms dominate the medium frequency phonon dispersions, which lie below the optical modes of heavier Ni(B) atoms. This inversion acts via hybridization as a repulsive force on the acoustic modes and leads to the deep softening of low-lying TA phonon modes [128]. Moreover, consistent judgments on the stability of  $\text{Ni}_2\text{MnTi}$ ,  $\text{Ni}_2\text{MnCr}$ ,  $\text{Ni}_2\text{MnHf}$ , and  $\text{Ni}_2\text{MnW}$  from the perspective of both phonon dispersions and DOSs indicate the occurrence of strong interaction between electronic structure and phonon dispersion [136,137]. Undeniably, a lot more systematic explorations and experiments, such as inelastic neutron scattering, are further demanded to verify the phonon dispersions and to make comparison with available results of conventional Ni-Mn-based shape memory Heusler compounds.

### E. Pressure-induced martensitic phase transition in $\text{Ni}_2\text{MnTi}$

According to the literature, pressure can reduce magnetism and volume, which provides an effective and clear approach to tune the phase stability of cubic Ni-Mn-based Heusler compounds [2,3,5,8]. Herein, considering the  $\text{Ni}_2\text{MnTi}$  compound as a representative example, the effect of external pressure on the cubic phase stability of strongly ferromagnetic all- $d$ -metal Heusler compounds was investigated, as shown in Fig. 9. First, Fig. 9(a) exhibits that the application of pressure suppresses the spin polarization of  $\text{Ni}_2\text{MnTi}$  and decreases its magnetic moment. Herein, the calculated magnetic moment of  $\text{Ni}_2\text{MnTi}$  at ground state is  $4.33 \mu_B/\text{f.u.}$ , which can be comparable with published values ( $4.07 \mu_B/\text{f.u.}$  in Ref. [81] and  $3.90 \mu_B/\text{f.u.}$  in Ref. [79]) by VASP. The loss of the magnetic moment is also accompanied by shrinkage of the lattice size when pressure is applied. Thus, it is speculated that loss of magnetism and contracted lattice size can destabilize the cubic phase [138]. Second, due to bond stretching in the shrunken cubic structure [139], TA modes in phonon dispersions gradually shift to the lower frequency zone with the increase of pressure, as shown in Fig. 9(b). TA branching along  $G$ - $X$ - $W$  directions shows the obvious softening as a function of pressure. As expected, when pressure is 16 GPa and beyond in Fig. 9(c), low-lying TA branches along  $G$ - $X$ - $W$  directions anomalously move to negative frequency ranges. This softening character of TA branches indicates that the cubic structure becomes dynamically unstable and martensitic phase transition can occur when the pressure is applied. Third, the pressure-induced martensitic phase transition in  $\text{Ni}_2\text{MnTi}$  can

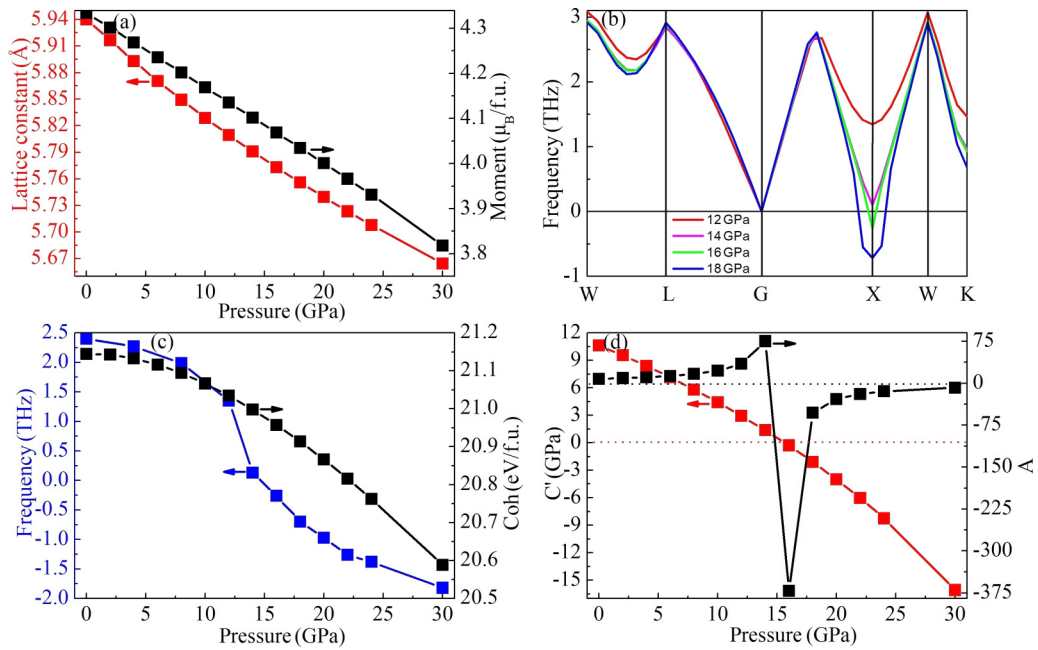


FIG. 9. Pressure dependence of physical parameters, indicating martensitic phase transition in  $\text{Ni}_2\text{MnTi}$ : (a) Pressure dependence of lattice parameter and magnetic moment. (b) Phonon dispersions of low-lying transverse acoustic (TA) branches at 12, 14, 16, and 18 GPa. The occurrence of imaginary frequency at 16 GPa indicates that applied pressure destabilizes cubic structure. (c) TA frequencies at high-symmetry point X and cohesive energy as a function of pressure. (d) Pressure dependence of tetragonal shear elastic constant  $C'$  and Zener anisotropy factor ( $A$ ). Their change behaviors support pressure-induced martensitic phase transition in  $\text{Ni}_2\text{MnTi}$ .

also be supported by the calculations of tetragonal shear elastic constant  $C'$  and Zener anisotropy factor ( $A$ ). Figure 9(d) illustrates that the values of  $C'$  and  $A$  at the pressure of 14 GPa are  $\sim 1.38$  GPa and 75.38, respectively. These values are close to theoretical values ( $C' = 2.36$  GPa,  $A = 42.43$  in this paper derived from data presented in Table I) and experimental results of well-studied  $\text{Ni}_2\text{MnGa}$  [89,90,92,99,100,102]. The values of  $C'$  and  $A$  both become negative when the pressure reaches 16 GPa and beyond. Furthermore, the decreased cohesive energy under pressure also destabilizes cubic  $\text{Ni}_2\text{MnTi}$ .

Importantly, when external pressure is applied on  $\text{Ni}_2\text{MnTi}$ , the inherent nature of unit cell size contraction during martensitic phase transition shortens the interatomic distance and consequently leads to the enhancement in the  $d$ - $d$  interatomic orbital hybridization [138]. The enhanced  $d$ - $d$  interatomic orbital hybridization can result in the improvement of ductility in  $\text{Ni}_2\text{MnTi}$ . Theoretically, ductile parameters of Poisson's ratios and  $B/G_V$  ratios shown in Fig. 10 both get improved when pressure is applied on  $\text{Ni}_2\text{MnTi}$ , indicating better ductile properties. Practically, the enhanced ductility can benefit the future application of the barocaloric and elastocaloric effects in all- $d$ -metal Heusler compounds.

Owing to the difference between atomic radii in the all- $d$ -metal Heusler compound  $\text{Ni}_2\text{MnTi}$ , the so-called *chemical pressure* can be induced during atomic substitution, which can work like the applied external hydrostatic pressure when tuning the phase stability. Herein, the phase stability of off-stoichiometric  $\text{Ni}_2\text{Mn}_{1+x}\text{Ti}_{1-x}$  compounds was theoretically investigated. The dependence of calculated  $C'$  and experimental martensitic phase transition temperature ( $T_M$ ) on Ti content

is shown in Fig. 11. As indicated by the circle in Fig. 11, the linearly fitted  $T_M$  value of  $\text{Ni}_2\text{MnTi}$  is near zero. It indicates fewer possibilities of martensitic phase transition in stoichiometric  $\text{Ni}_2\text{MnTi}$ , which is consistent with the judgment from obtained elastic parameters, electronic structure, and phonon dispersion. Experimentally, martensitic phase transition was not observed in  $\text{Ni}_2\text{MnTi}$  even when temperature was decreased significantly to 50 K [15]. When Ti was doped with Mn in  $\text{Ni}_2\text{Mn}_{1+x}\text{Ti}_{1-x}$  compounds, the  $C'$  could be softened and even made negative, as shown in Fig. 11, which led to the decreased stability of cubic structure and experimentally led

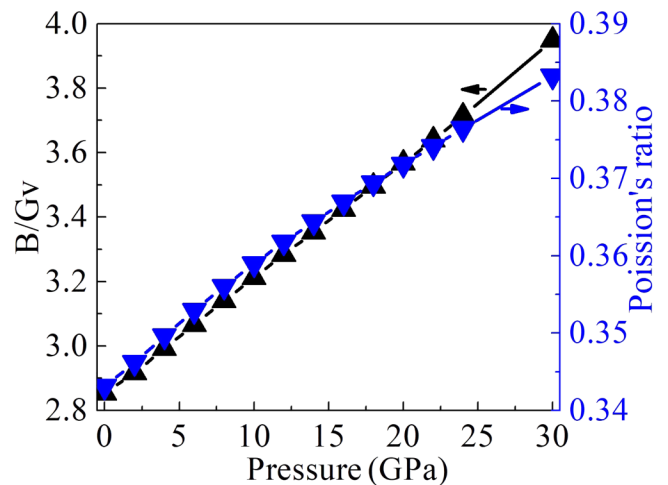


FIG. 10. Pressure dependence of ductile parameters of  $B/G_V$  ratio and Poisson's ratio in  $\text{Ni}_2\text{MnTi}$ . Their synchronous increase under applied pressure shows improved ductility.

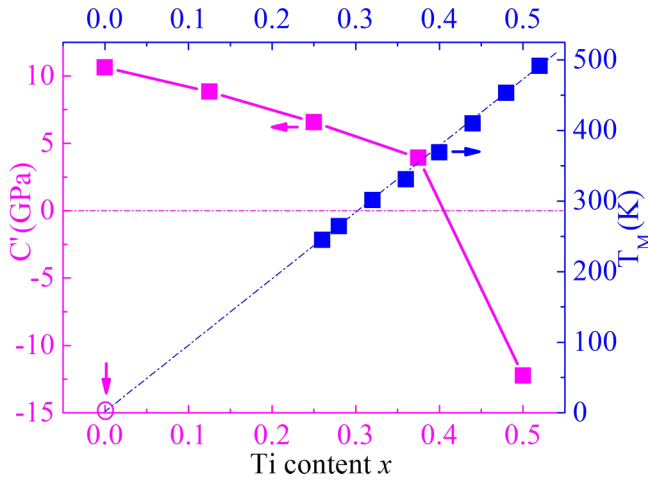


FIG. 11. Ti content  $x$  dependence of tetragonal shear elastic constant  $C'$  and experimental martensitic phase transition temperature ( $T_M$ ) in  $\text{Ni}_2\text{Mn}_{1+x}\text{Ti}_{1-x}$  ( $x = 0-0.5$ ) Heusler compounds.  $T_M$  was selected from Refs. [15,18]. The circle and arrow indicate that the martensitic phase transition temperature of stoichiometric  $\text{Ni}_2\text{MnTi}$  is near zero.

to  $T_M$  of  $\text{Ni}_2\text{Mn}_{1+x}\text{Ti}_{1-x}$  being up to room temperature when  $x$  was  $\sim 0.30$  [15,18,19]. This changing behavior of dependence of phase transition temperature on composition is very similar to that observed in conventional  $\text{Ni}_2\text{Mn}_{1+x}\text{Z}_{1-x}$  ( $Z = \text{Al}, \text{Ga}, \text{In}, \text{Sn}, \text{Sb}$ ) compounds [16].

When the smaller Mn atom dopes larger Ti atoms, the lattice size of  $\text{Ni}_2\text{Mn}_{1+x}\text{Ti}_{1-x}$  compounds decreases gradually with the increase of doping content of Mn [15]. Figures 9(a) and 9(d) present the lattice size and  $C'$  of stoichiometric  $\text{Ni}_2\text{MnTi}$ , revealing that  $C'$  decreases with the contraction of lattice size. From this perspective, the decrease of  $C'$  and consequently the instability of cubic  $\text{Ni}_2\text{Mn}_{1+x}\text{Ti}_{1-x}$  compounds could result from the contribution of volume effect. Comparison of the lattice size dependence of  $C'$  in off-stoichiometric  $\text{Ni}_2\text{Mn}_{1+x}\text{Ti}_{1-x}$  and stoichiometric  $\text{Ni}_2\text{MnTi}$  reveals that  $C'$  in off-stoichiometric  $\text{Ni}_2\text{Mn}_{1+x}\text{Ti}_{1-x}$  decreases far faster than that in stoichiometric  $\text{Ni}_2\text{MnTi}$  (see Fig. S5 in the Supplemental Material [25]). This difference indicates that the change of instability in off-stoichiometric  $\text{Ni}_2\text{Mn}_{1+x}\text{Ti}_{1-x}$  is not merely a volume effect. Moreover, in off-stoichiometric  $\text{Ni}_2\text{Mn}_{1+x}\text{Ti}_{1-x}$ , the valence electron concentration increases

with the increase of Mn content since Mn possesses more valence electrons than Ti. In general, the larger the valence electron concentration, the higher the martensitic phase transition temperature. It suggests that the increased valence electron concentration lowers the stability of cubic structure, which corresponds to the smaller value of  $C'$ . Therefore, the faster decrease of  $C'$  in off-stoichiometric  $\text{Ni}_2\text{Mn}_{1+x}\text{Ti}_{1-x}$  also results from the contribution of the increased valence electron concentration. Thus, in  $\text{Ni}_2\text{Mn}_{1+x}\text{Ti}_{1-x}$  compounds, the coupling of the volume effect and valence electron concentration is responsible for the changes of phase stability and consequently martensitic phase transition temperature.

#### F. Atomic diffusion process arrest-induced antiferromagnetism in $\text{Ni}_2\text{MnTi}$

Mn-containing Heusler compounds have become popular subjects of investigation because of their interesting feature of forcing the Mn atoms to occupy the sites with different neighboring relationships. This allows them to exchange couple in a ferromagnetic, ferrimagnetic, antiferromagnetic order, and even nonmagnetic manner [87,88,115]. Like conventional Ni-Mn-based Heusler compounds,  $\text{Ni}_2\text{MnTi}$  was theoretically predicted to be stable in ferromagnetic  $L2_1$ -type structure at ground state [79,81]. Notably, the Mn atom is the key contributor of magnetism. Rather differently, experimental magnetization measurements revealed that the  $\text{Ni}_2\text{MnTi}$  compound was antiferromagnetic, characterized by a low Néel temperature. Disordered arrangement of Mn and Ti atoms in metastable B2 structure was considered responsible for antiferromagnetism [15]. It indicates that antiferromagnetism can result from the disordered arrangement of Mn and its neighboring atoms.

To better comprehend and explore the origin of antiferromagnetism in  $\text{Ni}_2\text{MnTi}$ , five possible interatomic occupation styles were considered assuming that  $\text{Ni}_2\text{MnTi}$  remained in  $\text{Cu}_2\text{MnAl}$ -type and  $\text{Hg}_2\text{CuTi}$ -type structures: (i) atomic disorder between Mn(B) and Ti(D), Ni(A, C) in  $\text{Cu}_2\text{MnAl}$ -type structure and (ii) atomic disorder between Mn(C) and Ni(A), Ni(B), and Ti(D) in  $\text{Hg}_2\text{CuTi}$ -type structure. The total magnetic moments of  $\text{Ni}_2\text{MnTi}$  dependent on interatomic disorder level under these five possible disorderly arranged styles are shown in Figs. 12(a) and 12(b). The quantity of interatomic disorder level  $S$  is defined as follows:

$$S = \frac{\text{No. of A atoms on sublattice A when fully ordered} - \text{No. of A atoms on sublattice A when disordered}}{\text{No. of A atoms on sublattice A when fully ordered}}.$$

Other than the experimentally obtained disordered arrangement style between B and D sites in  $\text{Cu}_2\text{MnAl}$ -type structure [15], theoretical interatomic disordered occupation between Ni(A, C) and Mn(B) in the  $\text{Cu}_2\text{MnAl}$ -type structure, and Mn(C) and Ni(B)/Ti(D) in the  $\text{Hg}_2\text{CuTi}$ -type structure can also induce antiferromagnetism in  $\text{Ni}_2\text{MnTi}$ . Compared with the high-ordered ferromagnetic ground state of the  $\text{Ni}_2\text{MnTi}$  compound, the antiferromagnetic state in disordered  $\text{Ni}_2\text{MnTi}$  was found to be metastable, which is like those theoretically

observed in B2-type  $\text{Ni}_8\text{Mn}_5\text{Ti}_3$  [61]. To further explore the most possible occupation style of atoms at finite temperature and their physical origin, the energy dependent on the disorder level at the PM state was calculated, and the results are presented in Figs. 12(c) and 12(d). At PM states, when disordered occupation occurs between Mn and Ti in  $\text{Ni}_2\text{MnTi}$  [i.e., Mn(B) and Ti(D) in the  $\text{Cu}_2\text{MnAl}$ -type structure, and Mn(C) and Ti(D) in the  $\text{Hg}_2\text{CuTi}$ -type structure], the total energy dependent on the disorder level increases at a certain disorder

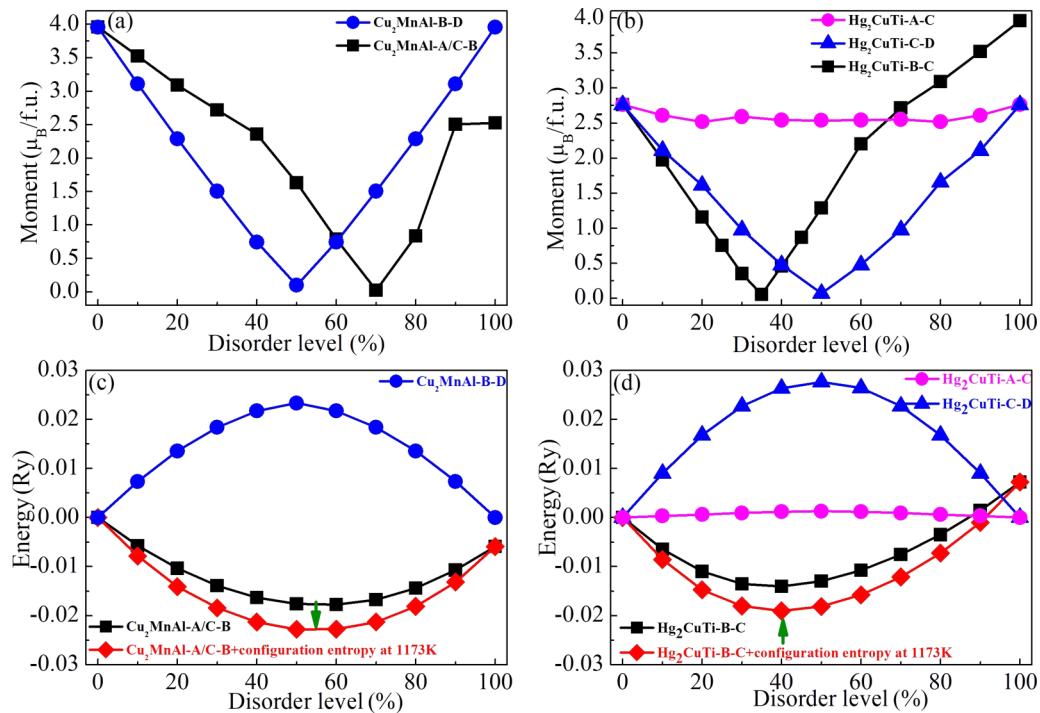


FIG. 12. Disorder-level dependence of magnetic moment in  $\text{Ni}_2\text{MnTi}$  at ground state with (a)  $\text{Cu}_2\text{MnAl}$ -type and (b)  $\text{Hg}_2\text{CuTi}$ -type structures. The local minimum indicates antiferromagnetism. Relative paramagnetic (PM) total energy and PM total energy plus the contribution from the configuration entropy at 1173 K as a function of the disorder level at PM state in  $\text{Ni}_2\text{MnTi}$  with (c)  $\text{Cu}_2\text{MnAl}$ -type and (d)  $\text{Hg}_2\text{CuTi}$ -type structures, considering the ground state energy of the ordered  $\text{Ni}_2\text{MnTi}$  as reference. The green arrows in (c) and (d) indicate the energy minimum at disorder level of  $\sim 55$  and  $40\%$  with  $\text{Cu}_2\text{MnAl}$ - and  $\text{Hg}_2\text{CuTi}$ -type structures, respectively.

range and reaches a local maximum at the disorder level of  $\sim 50\%$ . This is not a favorable condition at the PM state. Inversely, when atomic disorder occurs between Mn and its 1nn Ni in  $\text{Ni}_2\text{MnTi}$  [i.e., Mn(B) and Ni(A/C) in the  $\text{Cu}_2\text{MnAl}$ -type structure, and Mn(C) and Ni(B) in the  $\text{Hg}_2\text{CuTi}$ -type structure], the PM total energy dependence of the disorder level shows a local minimum at the disorder level of  $\sim 56\%$  in the  $\text{Cu}_2\text{MnAl}$ -type structure and  $\sim 38\%$  in the  $\text{Hg}_2\text{CuTi}$ -type structure. It indicates stable atomic configurations at PM states. Furthermore, by considering the configuration entropy at 1173 K (representing the experimental annealing temperature in Ref. [15]), the energy minima become significantly low [see Figs. 12(c) and 12(d)], indicating more stable states. Furthermore, at 1173 K, the preferential disorder levels are  $\sim 55\%$  in the  $\text{Cu}_2\text{MnAl}$ -type structure and  $\sim 40\%$  in the  $\text{Hg}_2\text{CuTi}$ -type structure, showing only slight changes with temperature. Comparatively, the dependence of the energy of a local minimum on disorder level curve of the  $\text{Hg}_2\text{CuTi}$ -type structure is lower than that of the  $\text{Cu}_2\text{MnAl}$ -type structure, which indicates that the  $\text{Hg}_2\text{CuTi}$ -type structure becomes dominant at the high-temperature PM state. It is expected that, when the atomic disordered arrangement occurs either between Mn(B) and Ni(A/C) in the  $\text{Cu}_2\text{MnAl}$ -type structure or between Mn(C) and Ni(B) in the  $\text{Hg}_2\text{CuTi}$ -type structure,  $\text{Ni}_2\text{MnTi}$  can remain in a mixture of  $\text{Hg}_2\text{CuTi}$ - and  $\text{Cu}_2\text{MnAl}$ -type structures. Consequently, current calculations and analysis reveal that the disordered occupation of atoms between Mn and its 1nn Ni atoms is preferential in  $\text{Ni}_2\text{MnTi}$  around the experimental annealing temperature, where it is in the PM state.

Based on the above-stated calculation results, the following scenario was suggested to reveal the physical origin of experimentally observed antiferromagnetism in  $\text{Ni}_2\text{MnTi}$ . When  $\text{Ni}_2\text{MnTi}$  was cooled down from the high-temperature PM state, the disorder-order transition during the cooling processes aided in determination of atomic configuration and consequently the magnetic states. Commonly, the complete atomic diffusion process during cooling can promise a full transformation from the high-temperature atomic disordered PM state to the chemically and magnetically ordered ferromagnetic state. Otherwise, any atomic diffusion process arrest can change the ultimate magnetic state. When quenching  $\text{Ni}_2\text{MnTi}$  from the PM state at 1173 K, the shortened atomic diffusion process blocks the transition from the atomic disordered PM state to the ordered ferromagnetic state. Owing to the atomic diffusion process arrest during cooling process,  $\text{Ni}_2\text{MnTi}$  gets locked in an intermediate state, which consequently maintains the relatively high atomic disorder level inherited from the high-temperature PM state. The atomic disorder level cannot further evolve toward the ordered ferromagnetic state anymore. Thus, the generation of the antiferromagnetic  $\text{Ni}_2\text{MnTi}$  is due to the incomplete transition from the high-temperature PM state to the ferromagnetic state when cooling. Further neutron diffraction experiments are encouraged to confirm the above-discussed atomic occupation styles. Our current theoretical observations indicate that suitable thermodynamics states when annealing and the atomic diffusion process when cooling the samples are crucial for optimizing magnetic states and further realizing their functional applications.

#### IV. CONCLUSIONS

In this paper, the elastic properties, phase stability, and magnetism and correlations among them in all- $d$ -metal  $\text{Ni}_2\text{Mn}T$  ( $T = \text{Sc, Ti, V, Cr, Y, Zr, Nb, Mo, Hf, Ta, and W}$ ) Heusler compounds were systematically investigated by first-principles calculations. The main results are summarized as follows:

(1) Within the scope of Heusler structures,  $\text{Ni}_2\text{Mn}T$  compounds with early transition metal atoms ( $T = \text{Sc, Ti, Y, Zr, Nb, Hf, and Ta}$ ) were relatively stable in the  $\text{Cu}_2\text{MnAl}$ -type structure, which is consistent with the conventional atomic preferential occupation rule in the Heusler family. In contrast,  $\text{Ni}_2\text{Mn}T$  compounds with late transition metal atoms ( $T = \text{V, Cr, Mo, and W}$ ) preferred the  $\text{Hg}_2\text{CuTi}$ -type structure. For transition metal atoms lying in the same period in the periodic table, lattice parameters of  $\text{Ni}_2\text{Mn}T$  decreased with the increase of valence electrons. Different from  $p$ - $d$  covalent hybridization-dominated conventional Heusler compounds, the atomic radius was the determining factor for the lattice size in  $d$ - $d$  metal bonding-controlled all- $d$ -metal Heusler compounds.

(2) Owing to the strong coupling among lower tetragonal shear elastic constant  $C'$ , higher Zener anisotropy factor  $A$ , deeply softened TA phonon modes, and typical peak structures in DOS below the Fermi level, the most likely martensitic phase transition can be expected in weakly magnetic  $\text{Ni}_2\text{Mn}T$  compounds with late transition metal atoms ( $T = \text{V, Cr, Mo, and W}$ ). Practically by applying hydrostatic pressure or imposing chemical pressure via adjusting the composition of Ti in the off-stoichiometric Ni-Mn-Ti system, magnetism got weakened, and the suppressed martensitic phase transition was re-evoked.

(3) Compared with  $p$ - $d$  covalent hybridization-dominant conventional Heusler compounds, in representative all- $d$ -metal Heusler compounds  $\text{Ni}_2\text{MnY}$  and  $\text{Ni}_2\text{MnTa}$ , nondirectional  $d$ - $d$  metal bonding becomes prevailing. This fundamental transition was responsible for the enhancement of

ductility and consequently helped in improving the fatigue life and service ability of all- $d$ -metal Heusler compounds. Experimentally, it is speculated that low-energy  $d$ - $d$  metal bonding can also benefit the nucleation and propagation of dislocations, which can further promote their ductile properties. Thus, tunable martensitic phase transition and improved ductility can enable all- $d$ -metal Heusler compounds to be more promising and attractive as multiple functional magnetic materials.

(4) In this paper, we also reveal that experimentally observed antiferromagnetism in  $\text{Ni}_2\text{MnTi}$  originated from atomic diffusion process arrest during the transition from the high-temperature chemically disordered PM state to the low-temperature chemically and magnetically ordered ferromagnetic state, which resulted in the intermediate metastable and partially disordered antiferromagnetic phase.

It is believed that comprehensive and in-depth understanding of the mechanism of martensitic phase transition, improved ductility, and magnetism in all- $d$ -metal Heusler compounds described in this paper can be helpful to explore martensitic phase transition materials with outstanding mechanical properties from both experimental and theoretical perspectives. Systematic exploration of this type of all- $d$ -metal Heusler compounds is of importance in the context of promising solid-state cooling application based on the caloric effects of phase transition materials.

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