Perfect kagome-lattice antiferromagnets with $J_{\text{eff}} = \frac{1}{2}$: The Co²⁺ analogs of the copper minerals volborthite and vesignieite

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We report the synthesis, crystal structure, and magnetic properties of Co^{2+} kagome magnets $Co_3V_2O_7(OH)_2 \cdot 2H_2O$ and $BaCo_3(VO_4)_2(OH)_2$, which can be recognized as Co analogs of the intensively researched quantum kagome magnet volborthite $Cu_3V_2O_7(OH)_2 \cdot 2H_2O$ and vesignieite $BaCu_3(VO_4)_2(OH)_2$. For each compound, the ground state is seemingly *A*-type antiferromagnetic order. At low temperatures, applying a magnetic field causes a metamagneticlike transition described by the transition in which antiferromagnetically aligned canted moments change to ferromagnetically aligned ones. These ground and field-induced states include a canted ferromagnetic component perpendicular to the kagome planes favored by Dzyaloshinskii-Moriya interactions. These magnetic properties are well characterized by the $J_{eff} = \frac{1}{2}$ physics. Our findings will be a step toward clarifying the $J_{eff} = \frac{1}{2}$ kagome physics, which has not been thoroughly studied experimentally or theoretically.

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I. INTRODUCTION

Spin- $\frac{1}{2}$ kagome-lattice antiferromagnets (KAFMs) are one of the most promising spin models to realize a highly entangled quantum spin-liquid (QSL) ground state [1–4]. However, the realistic KAFM materials deviate from the ideal model due to various intertwined perturbations. For example, in herbertsmithite ZnCu₃(OH)₆Cl₂, the "gold standard" in the KAFM exhibiting QSL behavior [5,6], the local spin correlations are greatly perturbed by chemical disorder, the nature of which has not yet been revealed [7–10]. It is also predicted that chemical perturbations will mimic QSL-like behavior [11,12]. Thus, it is crucial to deeply understand the individual roles of perturbations that will always be included in realistic materials. Under this circumstance, the development of new kagome-lattice materials is desirable.

The number of ions that can realize the $S = \frac{1}{2}$ state, which can form the kagome lattice, is minimal. Among them, physicists and materials scientists have successively found promising compounds in the mineral database that constitute the kagome lattice composed of Cu²⁺ ions. Thus, synthetic copper minerals have been the mainstream in quantum frustrated magnetism research. Examples include herbertsmithite, volborthite Cu₃V₂O₇(OH)₂·2H₂O [13–15], vesignieite BaCu₃(VO₄)₂(OH)₂ [16], and Cd kapellasite CdCu₃(OH)₆(NO₃)₂ · H₂O [17,18]. However, Cu²⁺ ions are highly susceptible to symmetry breaking because of their intense Jahn-Teller activity. Therefore, they tend to deviate from the ideal kagome model. On the other hand, Ti³⁺ ions can also realize $S = \frac{1}{2}$, and their Jahn-Teller activity is not as significant as that of Cu²⁺ ions, so there is a possibility that an ideal kagome model with minor perturbation can be realized. However, all the Ti^{3+} KAFMs reported so far are strongly distorted, resulting in a significant deviation from the kagome model [19,20].

We focus on Co^{2+} as a candidate ion for achieving a spin- $\frac{1}{2}$ state. The single-ion magnetic property of Co^{2+} in an octahedral environment is determined by the 12-fold degenerated (dodecet) ground state manifold ${}^{4}T_{1g}$ [21–23]. Due to spin-orbit coupling, this dodecet splits into three manifolds—doublet, quartet, and sextet, as shown in Fig. 1(a). The lowest-energy state is given by $J_{\text{eff}} = \frac{1}{2}$ Kramers doublet. As shown in Fig. 1(b), $J_{\text{eff}} = \frac{1}{2}$ is always the lowest energy regardless of the trigonal distortion, and the energy gap from the excited state is approximately $|\xi|$ to $2|\xi|$. Therefore, if the temperature is a much smaller energy scale than the spin-orbit interaction $\xi = -178 \text{ cm}^{-1}$ ($\sim -250 \text{ K}$), the magnetic properties are governed by the lowest-energy $J_{\text{eff}} = \frac{1}{2}$ Kramers doublet. Therefore, treating Co^{2+} compounds as a $J_{\text{eff}} = \frac{1}{2}$ doublet

Therefore, treating Co²⁺ compounds as a $J_{\text{eff}} = \frac{1}{2}$ doublet physics is possible in the low-temperature region. However, physicists must pay careful attention to the temperature dependence in the physical properties. The contribution of the excited states, $J_{\text{eff}} = \frac{3}{2}$ quartet and $J_{\text{eff}} = \frac{5}{2}$ sextet, by the Boltzmann distribution must cause the effective magnetic moment to change with temperature. Figure 1(c) shows the theoretically calculated μ_{eff} for noninteracting Co²⁺ in an octahedral crystal field with spin-orbit couplings $H = \xi LS$, and L is the total orbital quantum number [24]. It is written as

$$\mu_{\text{eff}}(\text{Co}^{2+}) = \sqrt{\frac{3\left\{\frac{63x}{20} + \frac{98}{25} + \left(\frac{640x}{225} + \frac{4312}{2025}\right)e^{\frac{-15x}{4}} + \left(\frac{169x}{36} - \frac{490}{81}\right)e^{-6x}\right\}}{x\left\{3 + 2e^{\frac{-15x}{4}} + e^{-6x}\right\}}},$$
(1)

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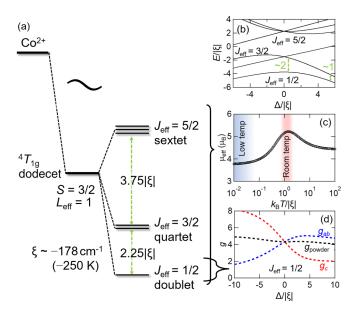


FIG. 1. (a) Energy levels scheme of high-spin d^7 ion under octhahedral crystal field and spin-orbit coupling and (b) their trigonal crystal field Δ derivative. (c) The theoretical effective magnetic moment μ_{eff} as a function of $k_{\text{B}}T/|\xi|$ in Co²⁺ ion with octahedral crystal fields, where k_{B} is the Boltzmann constant. (d) The anisotropic and powder-averaged g factors in Co²⁺ ion as a function of $\Delta/|\xi|$.

where $x = \xi/k_{\rm B}T$. This theoretical calculation yields $\mu_{\rm eff} \sim 5.2\mu_{\rm B}$ near room temperature and $\sim 4 \mu_{\rm B}$ at low temperatures. Conversely, detecting the temperature variation of $\mu_{\rm eff}$ makes it possible to demonstrate a realization of $J_{\rm eff} = \frac{1}{2}$ physics in realistic Co²⁺ materials at low temperatures.

In addition, physicists should also pay attention to the effects of low-symmetry crystal fields when considering only the $J_{\rm eff} = \frac{1}{2}$ doublet at low temperatures. In a perfect octahedral crystal field, there is no anisotropy in the Zeeman splitting of the doublet in a magnetic field. On the other hand, in an axis-symmetric crystal field such as a trigonal or tetragonal crystal field, there is a change in energy between the case parallel to the c axis and the case perpendicular to it. Therefore, anisotropy appears in the g value. Figure 1(d)shows the g values expressed as a function of the parameter $\Delta/|\xi|$, where Δ is the trigonal distortion energy [22,25]. Here, $\Delta/|\xi| < 0$ means the Co₆ octahedron is trigonally elongated along the c axis and $\Delta/|\xi| > 0$ is trigonally compressed. The powder-averaged g_{powder} (= $[(2g_{ab}^2 + g_c^2)/3]^{0.5}$) of the powder sample is almost unchanged and is approximately $g_{powder} \sim 4$, although there is a slight Δ dependence. Therefore, the experimental observation of a powder-averaged g value close to 4 at low temperatures in the powder sample is a criterion for satisfying the $J_{\text{eff}} = \frac{1}{2}$ condition. Unlike Cu²⁺ or Ti³⁺ KAFMs, there is a limited number

Unlike Cu²⁺ or Ti³⁺ KAFMs, there is a limited number of candidates for the Co²⁺ KAFMs, so experimental studies have been limited. Co₃Mg(OH)₆Cl₂ contains ~7% site mixture between the Mg²⁺ and Co²⁺ sites, resulting in a spinglass character due to randomness [26]. Moreover, synthesis of BaCo₃(VO₄)₂(OH)₂ [27] and Co₃V₂O₇(OH)₂·2H₂O [28], Co²⁺ analogs of vesignieite and volborthite, has also been reported. Their crystal structures are shown in Figs. 2(a) and

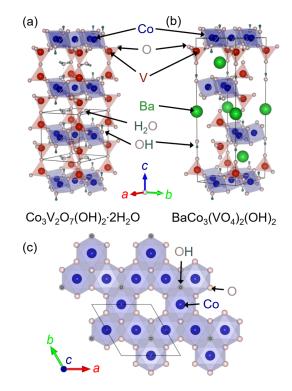


FIG. 2. Crystal structures of (a) $Co_3V_2O_7(OH)_2 \cdot 2H_2O$ and (b) $BaCo_3(VO_4)_2(OH)_2$ viewed along the *c* axis. (c) The $Co_3O_6(OH)_2$ kagome layer viewed along the *ab* plane commonly observed in $Co_3V_2O_7(OH)_2 \cdot 2H_2O$ and $BaCo_3(VO_4)_2(OH)_2$. The VESTA program is used for visualization [31].

2(b). Both compounds commonly comprised the $Co_3O_6(OH)_2$ kagome layer [Fig. 2(c)]. Both compounds exhibit superparamagnetism without magnetic orderings, probably due to finite-size effects [29,30].

This work reports the successful synthesis of high-quality $Co_3V_2O_7(OH)_2 \cdot 2H_2O$ and $BaCo_3(VO_4)_2(OH)_2$ polycrystalline samples, as well as their magnetic properties. The crystal structure analysis demonstrates the formation of a perfect kagome lattice without distortion and site mixture. Moreover, the magnetic and thermodynamic properties commonly observed in both materials are well characterized in the $J_{eff} = \frac{1}{2}$ state. Our findings demonstrate that $Co_3V_2O_7(OH)_2 \cdot 2H_2O$ and $BaCo_3(VO_4)_2(OH)_2$ are promising model compound $J_{eff} = \frac{1}{2}$ KAFMs.

II. EXPERIMENTAL METHODS

Polycrystalline samples were prepared using the hydrothermal method. All starting materials were purchased from FUJIFILM Wako Pure Chemical Corporation. For $Co_3V_2O_7(OH)_2 \cdot 2H_2O$, a Teflon beaker containing 0.37 g of $Co(NO_3)_2 \cdot 6(H_2O)$, 0.1 g of NH_4VO_3 , 0.07 g of NaOH, and 10 ml of pure H_2O was heated at 120 °C for 60 h. For $BaCo_3(VO_4)_2(OH)_2$, a Teflon beaker of 30 ml volume containing 0.35 g of $Ba(OH)_2 \cdot 8H_2O$, 0.78 g of $CoCl_2 \cdot 2H_2O$, 0.14 g of NH_4VO_3 , and 5 ml of pure H_2O was heated at 200 °C for 12 h. In both syntheses, orange-colored powders were obtained after rinsing with distilled water several times and drying at 110 °C. The thus-obtained samples were

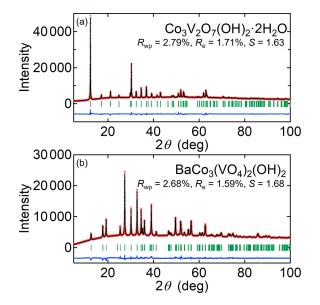


FIG. 3. (a) Powder x-ray diffraction patterns of (a) $Co_3V_2O_7(OH)_2 \cdot 2H_2O$ and (b) $BaCo_3(VO_4)_2(OH)_2$. The observed intensities (red), calculated intensities (black), and their differences (blue) are shown. Green vertical bars indicate the positions of Bragg reflections.

characterized by powder x-ray diffraction (XRD) using Cu $K\alpha$ radiation. The cell parameters and crystal structure were refined by the Rietveld method using the RIETAN-FP version 2.16 software [32]. The temperature dependence of magnetization was measured under magnetic fields up to 7 T in a magnetic property measurement system (MPMS, Quantum Design). The temperature dependence of the heat capacity was measured by a conventional relaxation method in a physical property measurement system (PPMS, Quantum Design). Magnetization curves up to approximately 50 T were measured by the induction method in a multilayer pulsed magnet at the International Mega Gauss Science Laboratory in the Institute for Solid State Physics.

III. RESULTS

A. Crystal structures

Figures 3(a) and 3(b) show the powder XRD pattern of $Co_3V_2O_7(OH)_2 \cdot 2H_2O$ and $BaCo_3(VO_4)_2(OH)_2$. For both samples, the observed XRD patterns are perfectly reproduced using previously reported crystal structure parameters without changing any parameters, confirming that our samples are single phase and match the trigonal structures. The crystal structure commonly in Co₃V₂O₇(OH)₂·2H₂O and BaCo₃(VO₄)₂(OH)₂ contains an edge-sharing network of CoO_6 octahedra as shown in Fig. 2(c), where the Co atoms form a perfect kagome lattice. There is only one crystallographic Co site and Co-Co distance in the kagome lattice. Each Co ion is surrounded by six anions: two OH ions and four O ions. The bonds are short for the OH ions at the trans position, and long for the four lateral O ions. The twoshort four-long-type crystal field must affect the single-ion anisotropy in the $J_{\rm eff} = \frac{1}{2}$ pseudospin of Co²⁺.

B. Magnetic property

Figures 4(a) and 4(b) show the temperature dependencies of magnetic susceptibility χ and their inverse $1/\chi$ for $Co_3V_2O_7(OH)_2 \cdot 2H_2O$ and $BaCo_3(VO_4)_2(OH)_2$. In the $1/\chi$ curves, a slope change is observed around 50 K. The effective magnetic moments μ_{eff} and Weiss temperature θ_{W} obtained by the Curie-Weiss (CW) fits of the $1/\chi$ curves in high temperatures (150-300 K) and low temperatures (10-25 K) are listed in Table I. In common with both materials, the $\mu_{\rm eff}$ values are $\sim 5.2 \ \mu_{\rm B}$ for high-temperature regions and ~ 4.2 $\mu_{\rm B}$ for low-temperature regions, which is roughly consistent with the theoretical calculation shown in Fig. 1(c). Thus, the $\mu_{\rm eff}$ shrinkage is explained in terms of level split owing to spin-orbit couplings. These negative Weiss temperatures indicate that the interaction between Co^{2+} spins is predominantly antiferromagnetic. Furthermore, the change in slope of $1/\chi$ as a function of temperature is observed. For Co^{2+} ions in an octahedral environment, the spacings of the multiplet levels $(J = \frac{1}{2}, \frac{3}{2}, \text{ and } \frac{5}{2})$ are not so large compared to $k_{\rm B}T$. Thus, the observation of μ_{eff} decreasing demonstrates a crossover of the multiplet levels.

In both materials, the χ data increase and then tend to saturate at low temperatures, a characteristic of ferromagnets. On the contrary, as shown in Fig. 4(c), the χ data of $Co_3V_2O_7(OH)_2 \cdot 2H_2O$ and $BaCo_3(VO_4)_2(OH)_2$ under lower applied fields (10 mT) exhibit antiferromagneticlike magnetic anomalies at $T_{\rm N} = 2.7$ and 4.9 K, respectively. A clear magnetic order appeared in these materials, whereas no magnetic order was found in previous reports [29,30], probably due to improved crystallinity and sample quality. In addition, the slight separation of the zero-field cooling (ZFC) and field cooling (FC) curves below T_N suggests an emergent magnetic ordering. Figure 4(d) shows the isothermal magnetizations M and their derivative dM/dH at 2 K for Co₃V₂O₇(OH)₂·2H₂O and BaCo₃(VO₄)₂(OH)₂. The peaks in the dM/dH data at 75 mT for $Co_3V_2O_7(OH)_2 \cdot 2H_2O$ and $\mu_0H_M = 16 \text{ mT}$ for $BaCo_3(VO_4)_2(OH)_2$ indicate a magnetic field induced phase transition. The M^2 vs H/M plot [the inset of Fig. 4(e)] shows an S-like behavior because of first-order metamagnetic transition according to Landau theory [33]. Also, metamagnetism is universally observed in A-type antiferromagnets with intralayer ferromagnetically coupled layers [34-38]. The origin of the A-type antiferromagnetic-ferromagnetic phase transition at low magnetic fields appears to be due to weak interlayer coupling. Therefore, we conclude that the title materials are A-type antiferromagnets.

The good linearity of the M^2 vs H/M plot with a positive value intercept in the 1–5 T field region indicates that the spontaneous magnetization M_0 in the category of Landau theory is an order parameter in the field-induced phase [39]. The M_0 values obtained as the intercept of the linear fit are $M_0 = 1.21 \,\mu_B$ for Co₃V₂O₇(OH)₂·2H₂O and $M_0 = 0.924 \,\mu_B$ for BaCo₃(VO₄)₂(OH)₂. In general, the ground state of Co²⁺ is well described by the $J_{eff} = \frac{1}{2}$ pseudospin as shown in Fig. 1(a), and the powderaveraged g factor is ~4 [see Fig. 1(d)]; thus, $M_{sat} \sim$ $2 \,\mu_B$. Compared to M_{sat} , the observed M_0 is relatively small. Figure 4(f) shows the isothermal magnetizations of Co₃V₂O₇(OH)₂·2H₂O and BaCo₃(VO₄)₂(OH)₂ up to 50 T. Above 40 T, the M data increase almost linearly due to the

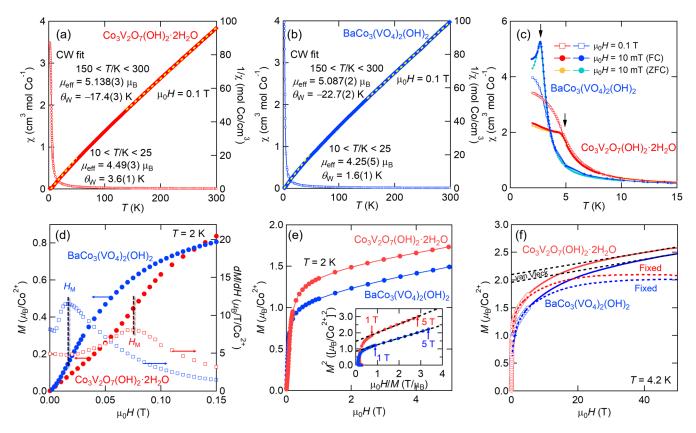


FIG. 4. Temperature dependence of the magnetic susceptibility χ (open symbols) and inverse magnetic susceptibility $1/\chi$ (closed symbols) for (a) Co₃V₂O₇(OH)₂·2H₂O and (b) BaCo₃(VO₄)₂(OH)₂. The dotted lines on the $1/\chi$ data represent fits to the Curie-Weiss (CW) model at high- (150 < *T/*K) and low-temperature (10 < *T/K* < 25) regions. (c) Temperature dependences of χ measured for magnetic fields of 0.1 T and 10 mT. (d) Low-field (<0.15 T) magnetization curves *M* and their derivative *dM/dH* measured at 2 K for Co₃V₂O₇(OH)₂·2H₂O and BaCo₃(VO₄)₂(OH)₂. The metamagneticlike transition is observed at $\mu_0 H_M = 75$ mT for Co₃V₂O₇(OH)₂·2H₂O and $\mu_0 H_M = 16$ mT for BaCo₃(VO₄)₂(OH)₂. (e) The *M* data at 2 K up to 5 T for Co₃V₂O₇(OH)₂·2H₂O and BaCo₃(VO₄)₂(OH)₂. The inset shows the *M*² vs $\mu_0 H/M$ plot. (f) Magnetization curves *M* measured at 4.2 K under pulsed magnetic fields up to 5 T for Co₃V₂O₇(OH)₂·2H₂O and BaCo₃(VO₄)₂(OH)₂. The inset shows the *M*² vs $\mu_0 H/M$ plot. (f) Magnetization curves *M* measured at 4.2 K under pulsed magnetic fields up to 50 T for Co₃V₂O₇(OH)₂·2H₂O and BaCo₃(VO₄)₂(OH)₂. The magnitudes are calibrated to the data measured under static fields up to 5 T (open circles).

large temperature-independent van Vleck paramagnetism of octahedrally coordinated Co²⁺ ions. Linear fits of the *M* data above 40 T yield the van Vleck paramagnetic susceptibility χ_{vv} is 5.57 × 10⁻³ cm³/mol Co²⁺ for Co₃V₂O₇(OH)₂·2H₂O and 5.59 × 10⁻³ cm³/mol Co²⁺ for BaCo₃(VO₄)₂(OH)₂. Then, the saturation magnetizations M_{sat} were obtained as $M_{sat} = 2.10 \,\mu_{B}/\text{Co}^{2+}$ for Co₃V₂O₇(OH)₂·2H₂O and $M_{sat} = 1.98 \,\mu_{B}/\text{Co}^{2+}$ for BaCo₃(VO₄)₂(OH)₂. Thus, the observed M_{sat} value provides strong evidence that the $J_{eff} = \frac{1}{2}$ ground state is realized in these materials, yielding the respective powder-averaged *g* values as follows: g = 4.20 in Co₃V₂O₇(OH)₂·2H₂O and g = 3.96 in BaCo₃(VO₄)₂(OH)₂. The observation of $M_{sat} \sim 2$ demonstrates that the reduced

TABLE I. Experimentally obtained Curie-Weiss parameters at high (150–300 K) and low temperatures (10–25 K) in $Co_3V_2O_7(OH)_2$ ·2H₂O and BaCo₃(VO₄)₂(OH)₂.

Formula	150–300 K		10–25 K	
	$\mu_{\mathrm{eff}} \left(\mu_{\mathrm{B}} \right)$	$\theta_{\mathrm{W}}\left(\mathrm{K}\right)$	$\mu_{\mathrm{eff}}~(\mu_{\mathrm{B}})$	$\theta_{W}(K)$
	5.138(3) 5.087(2)	-17.4(3) -22.7(2)	4.49(3) 4.25(5)	3.6(1) 1.6(1)

 M_0 is not due to temperature effects but rather due to the presence of a field-induced phase with a small spontaneous magnetization.

C. Thermodynamic property

Figure 5(a) shows the temperature dependence of the heat capacity divided by temperature C/T of $Co_3V_2O_7(OH)_2 \cdot 2H_2O$ and BaCo₃(VO₄)₂(OH)₂. Clear λ -shaped peaks are observed at 4.6 and 2.7 K for Co₃V₂O₇(OH)₂·2H₂O and $BaCo_3(VO_4)_2(OH)_2$, respectively. Therefore, both in $Co_3V_2O_7(OH)_2 \cdot 2H_2O$ and $BaCo_3(VO_4)_2(OH)_2$, second-order magnetic phase transitions with bulk nature occur at $T_{\rm N}$. As shown in the inset, even at 0.1 T, where the magnetic anomaly seems to disappear in the magnetic susceptibility, a λ -type peak is almost unchanged. This fact indicates that the magnetic field region above H_M is a robust magnetically ordered state different from the ground state, contrasting with a polarized phase.

Generally, to evaluate the magnetic contribution, the heat capacity of the nonmagnetic analog is used to determine its lattice contributions. Unfortunately, the nonmagnetic counterpart $BaZn_3(VO_4)_2(OH)_2$ is unavailable, and the C/T data of $Zn_3V_2O_7(OH)_2 \cdot 2H_2O$ (as a natural mineral

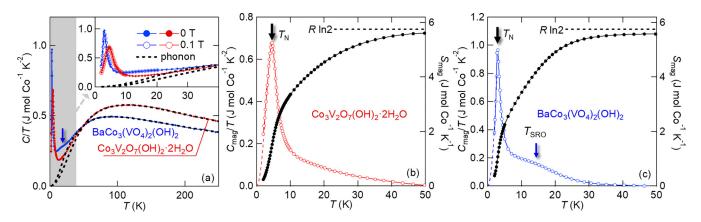


FIG. 5. (a) The total heat capacities divided by temperature C/T of $\text{Co}_3\text{V}_2\text{O}_7(\text{OH})_2 \cdot 2\text{H}_2\text{O}$ and $\text{BaCo}_3(\text{VO}_4)_2(\text{OH})_2$. The inset shows an enlarged view of the shaded area. The C/T data with 0.1 T magnetic field applied are also shown as open circles. The dashed lines on the data represent the lattice contribution estimated by fitting the data above 100 K, as described in the text. The temperature dependences of the magnetic heat capacity divided by temperature C_{mag}/T after the subtraction of lattice contribution and magnetic entropy S_{mag} for (b) $\text{Co}_3\text{V}_2\text{O}_7(\text{OH})_2 \cdot 2\text{H}_2\text{O}$ and (c) $\text{BaCo}_3(\text{VO}_4)_2(\text{OH})_2$. The horizontal dashed lines indicate the value of $S_{\text{mag}} = R \ln 2$, which is the total magnetic entropy derived from $J_{\text{eff}} = \frac{1}{2}$.

martyite [40]) does not match the high-temperature *C/T* data of $Co_3V_2O_7(OH)_2\cdot 2H_2O$ at all (see the Supplemental Material [41]). Thus, to extract the magnetic contribution to the heat capacity, the lattice contribution was estimated by fitting the high-temperature part where the magnetic heat capacity may be negligible. It is well known that a lattice heat capacity C_{latt} consists of contributions from three acoustic phonon branches and 3n-3 optical phonon branches, where *n* is the number of atoms per formula unit [42]; *n* equals 22 for $Co_3V_2O_7(OH)_2\cdot 2H_2O$ and 18 for $BaCo_3(VO_4)_2(OH)_2$. The acoustic and optical contributions are described by the Debyeand Einstein-type heat capacities C_D and C_E , respectively. Provided that C_{latt} is the sum of C_D and C_E , the *C/T* data above 100 K, where the magnetic heat capacity may be negligible, are fitted to the equation

$$C_{\text{latt}} = C_{\text{D}} + C_{\text{E}}$$

= $9R(T/\theta_{\text{D}})^3 \int_0^{\theta_{\text{D}}/T} \frac{x^4 \exp(x)}{[\exp(x) - 1]^2} dx$
+ $R \sum_{i=1}^3 n_i \frac{(\theta_{\text{Ei}}/T)^2 \exp(\frac{\theta_{\text{Ei}}}{T} - 1)}{\exp(\frac{\theta_{\text{Ei}}}{T} - 1)},$ (2)

where *R* is the gas constant, θ_D is the Debye temperature, and $\theta_{\rm Ei}$ is the Einstein temperatures. The best fits are shown by the dashed lines with $\theta_D = 180(2)$ K, $\theta_{\rm E1} = 260(9)$ K, $\theta_{\rm E2} = 552(12)$ K, $\theta_{\rm E3} = 1677(24)$ K, $n_1 = 17$, $n_2 = 23$, and $n_3 = 23$ for Co₃V₂O₇(OH)₂·2H₂O and $\theta_D = 153(4)$ K, $\theta_{\rm E1} = 227(9)$ K, $\theta_{\rm E2} = 633(11)$ K, $\theta_{\rm E3} = 1544(52)$ K, $n_1 = 16$, $n_2 = 17$, and $n_3 = 18$ for BaCo₃(VO₄)₂(OH)₂. Figures 5(b) and 5(c) show the magnetic contribution $C_{\rm mag}/T$ obtained by subtracting this lattice contribution from the experimental data and the magnetic entropy $S_{\rm mag}$ calculated by integrating the magnetic $C_{\rm mag}/T$ with respect to *T* assuming the $C_{\rm mag}/T$ value equals 0 at 0 K following the third law of thermodynamics. The asymptotic value of $S_{\rm mag}$ at high temperatures close to $R \ln 2 = 5.76 \, {\rm mol}^{-1} \, {\rm K}^{-1}$ expected for a doublet demonstrates that the $J_{\rm eff} = \frac{1}{2}$ doublet is realized.

As shown in Fig. 5(c), the C_{mag}/T data of BaCo₃(VO₄)₂(OH)₂ show a broad hump at $T_{SRO} \sim 15 \text{ K}$ higher than T_N , indicating a short-range ordering. This hump can also be seen in the unsubtracted C/T data [indicated by the blue arrow in Fig. 5(a)]. On the other hand, as shown in Fig. 5(b), the C_{mag}/T data for $\text{Co}_3\text{V}_2\text{O}_7(\text{OH})_2\cdot 2\text{H}_2\text{O}$ show no apparent hump. However, up to around 50 K, which is considerably higher than T_N , there is a finite C_{mag}/T that decays with increasing temperature, indicating the development of short-range order. Furthermore, the S_{mag} data at T_{N} reach approximately 20% and 12% for $Co_3V_2O_7(OH)_2 \cdot 2H_2O$ and $BaCo_3(VO_4)_2(OH)_2$, respectively. These observations indicate that the development of short-range magnetic correlations above T_N has released magnetic entropy. The difference in the presence/absence of a distinct hump structure in C_{mag}/T data associated with the short-range ordering probably depends on the difference in magnetic dimensionality derived from their crystal structures—the V_2O_7 polyanions bound to the kagome layer of $Co_3V_2O_7(OH)_2 \cdot 2H_2O$ are tightly chemically bound to the neighboring kagome layer, whereas the VO₄ polyanions bound to the kagome layer of $BaCo_3(VO_4)_2(OH)_2$ are isolated from the neighboring kagome layer. Therefore, the magnetic dimensionality must be derived from the "built-in" character of a crystal structure, being consistent with the higher $T_{\rm N}$ and $H_{\rm M}$ in Co₃V₂O₇(OH)₂·2H₂O.

IV. DISCUSSION

 $Co_3V_2O_7(OH)_2 \cdot 2H_2O$ and $BaCo_3(VO_4)_2(OH)_2$ are Co^{2+} analogs of the kagome copper minerals volborthite and vesignieite. The observed effective magnetic moment, saturated magnetization, and magnetic entropy suggest a realization of $J_{eff} = \frac{1}{2}$ doublet at low temperatures.

First, the observed metamagneticlike transition with reduced M_0 is discussed. Both title materials exhibit $\sim \frac{1}{2}M_{\text{sat}}$ after metamagneticlike transition, distinct from conventional metamagnets. Moreover, the maximum magnetization

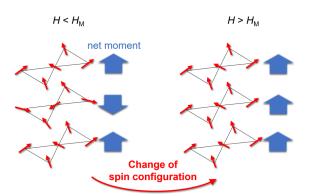


FIG. 6. Schematic view of the change of spin configuration in which the antiferromagnetically aligned canted moments change to ferromagnetically aligned ones in $Co_3V_2O_7(OH)_2$ ·2H₂O and BaCo₃(VO₄)₂(OH)₂.

increases without saturation on increasing the applied magnetic field above $H_{\rm M}$. The observed $M_{\rm sat} \sim 2$ expected for $J_{\rm eff} = \frac{1}{2}$ probed by pulsed magnetization measurements demonstrates that the small M_0 is not due to temperature effects. This behavior is consistent with a noncollinear spin structure such as a canted antiferromagnetic order. In a noncollinear spin structure, spin canting creates an uncompensated net moment. The coplanar 120° spin structure is most stable without Dzyaloshinskii-Moriya (DM) interactions in the classical kagome-lattice antiferromagnets [43].

On the other hand, due to finite DM interactions, all spins tilt from the kagome plane to an umbrellalike structure, resulting in a spontaneous magnetization [44–46]. Essentially, there is no inversion center at the middle point of the coupling between neighboring magnetic sites in the kagome lattice. Such magnetic structures have been universally observed in several jarosite-type materials [47,48] and rare-earth kagome $R_3Sb_3Mg_2O_{14}$ (R = Nd, Dy) [49,50]. Hence, spontaneous magnetization emerges within a single kagome layer. In this situation, a metamagneticlike transition in which antiferromagnetically aligned ones can emerge, as shown in Fig. 6. Such a metamagneticlike transition of the net moment has been observed in jarosite-type $AFe_3(OH)_6(SO_4)_2$ (A = K, Ag) [51,52].

Next, we discuss the magnitude of the single-layer net magnetization. The order of the D vector can be estimated using the following equation [53,54],

$$|\boldsymbol{D}| \sim \frac{g - g_{\rm spin}}{g} 2J, \qquad (3)$$

where $g_{spin}(=2)$ is the spin-only g value, and J is the exchange interaction. The canting angle θ from the plane of coplanar spin structure with the competition and DM interactions is

$$\tan \theta = \frac{|\boldsymbol{D}|}{2J} \sim \frac{g - g_{\rm spin}}{g}.$$
 (4)

The expected in-plane spontaneous magnetization M_0 using the θ value obtained with the experimentally observed $g \sim 4$ is calculated to be $M_0 \sim 0.45M_{\text{sat}}$. Therefore, it is possible

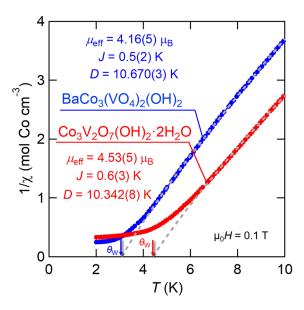


FIG. 7. Results of the modified Curie-Weiss fit incorporating the Dzyaloshinskii-Moriya interaction in the molecular field approximation to the the $1/\chi$ data on the temperature dependence of the inverse susceptibility at low-temperature (7 < T/K < 10) regions.

to explain the observed reduced M_0 from the DM interaction alone in Co²⁺ ions. Note that the spontaneous magnetization of the field-induced phases is powder averaged since the measurements were performed on powder samples. Thus, it is difficult to determine whether the reduced M_0 originates solely from DM interactions in the present stage. Therefore, the future establishment of single-crystal growth will clarify the exact origin of the reduced M_0 . Moreover, the absence of the $\frac{1}{3}$ plateau characteristic of kagome-lattice antiferromagnets would be due to the in-plane net moment of $\sim \frac{1}{2}M_{\text{sat}}$ in canting spin structure transcending that of $\frac{1}{3}M_{\text{sat}}$ in the up-up-down one. Neutron diffraction experiments are a future issue to clarify the details of these magnetic structures, and it is urgent to establish a synthetic method for deuterated title materials Co₃V₂O₇(OD)₂ · 2D₂O and BaCo₃(VO₄)₂(OD)₂.

Finally, we discuss the influence of DM interactions on magnetic susceptibility. By Moriya, an extended Curie-Weiss rule that incorporates the DM interaction by molecular field approximation has been given as follows,

$$\chi = \frac{N\mu_{\rm eff}^2}{3k_{\rm B}(T - \theta_{\rm w})} \frac{(T + T_0)}{(T + \theta_{\rm w})} + \chi_0,$$
(5)

with

$$\theta_{\rm w} = \frac{JzS(S+1)}{3k_{\rm B}} \left[1 + \left(\frac{|\boldsymbol{D}|}{J}\right)^2 \right]^{\frac{1}{2}},\tag{6}$$

$$T_0 = \frac{J_z S(S+1)}{3k_{\rm B}},$$
 (7)

where z indicates the coordination number; z = 4 in the kagome lattice [54]. The results of the fitting using Eq. (5) for the $1/\chi$ data below 10 K are shown in Fig. 7. Here, to reduce fitting errors, χ_0 was fixed to the value of χ_{vv} estimated from the high-field magnetization process [see Fig. 4(f)]. The fits

yield $\mu_{\text{eff}} = 4.53(5) \,\mu_{\text{B}}, J = 0.6(3) \,\text{K}$, and $|D| = 10.342(8) \,\text{K}$ for $\text{Co}_3\text{V}_2\text{O}_7(\text{OH})_2 \cdot 2\text{H}_2\text{O}$, and $\mu_{\text{eff}} = 4.16(5) \,\mu_{\text{B}}, J = 0.5(2)$ K, and |D| = 10.670(3) K for BaCo₃(VO₄)₂(OH)₂. The thusobtained θ_w values, corresponding to the intercept of the T axis, are 4.4 and 3.0 K for $Co_3V_2O_7(OH)_2 \cdot 2H_2O$ and $BaCo_3(VO_4)_2(OH)_2$, respectively, roughly consistent with the magnetic transition temperatures $T_{\rm N}$. These results indicate that relatively large DM interactions dominantly stabilize the magnetic order. Here, given the relationship in Eq. (3), it is mysterious that J is so small relative to |D|, of which the reasonable origin is a cancellation of several competing superexchange interactions with different signs. For example, ferromagnetic Kitaev interactions expected for Co²⁺ magnets possibly compete with conventional Heisenberg antiferromagnetic interactions [55,56]. In the present stage, extracting them independently from the magnetic susceptibility analysis is impossible. Therefore, inelastic neutron scattering experiments are planned to estimate the exchange parameters needed to construct reasonable spin models for Co₃V₂O₇(OH)₂·2H₂O and $BaCo_3(VO_4)_2(OH)_2$.

- [1] L. Balents, Nature (London) 464, 199 (2010).
- [2] L. Savary and L. Balents, Rep. Prog. Phys. 80, 016502 (2017).
- [3] Y. Zhou, K. Kanoda, and T.-K. Ng, Rev. Mod. Phys. 89, 025003 (2017).
- [4] C. Broholm, R. J. Cava, S. A. Kivelson, D. G. Nocera, M. R. Norman, and T. Senthil, Science 367, eaay0668 (2020).
- [5] M. A. de Vries, J. R. Stewart, P. P. Deen, J. O. Piatek, G. J. Nilsen, H. M. Rönnow, and A. Harrison, Phys. Rev. Lett. 103, 237201 (2009).
- [6] T.-H. Han, J. S. Helton, S. Chu, D. G. Nocera, J. A. Rodriguez-Rivera, C. Broholm, and Y. S. Lee, Nature (London) 492, 406 (2012).
- [7] M. R. Norman, Rev. Mod. Phys. 88, 041002 (2016).
- [8] M. A. de Vries, K. V. Kamenev, W. A. Kockelmann, J. Sanchez-Benitez, and A. Harrison, Phys. Rev. Lett. 100, 157205 (2008).
- [9] A. Olariu, P. Mendels, F. Bert, F. Duc, J. C. Trombe, M. A. de Vries, and A. Harrison, Phys. Rev. Lett. 100, 087202 (2008).
- [10] D. E. Freedman, T. H. Han, A. Prodi, P. Müller, Q.-Z. Huang, Y.-S. Chen, S. M. Webb, Y. S. Lee, T. M. McQueen, and D. G. Nocera, J. Am. Chem. Soc. **132**, 16185 (2010).
- [11] Z. Zhu, P. A. Maksimov, S. R. White, and A. L. Chernyshev, Phys. Rev. Lett. **119**, 157201 (2017).
- [12] I. Kimchi, J. P. Sheckelton, T. M. McQueen, and P. A. Lee, Nat. Commun. 9, 4367 (2018).
- [13] Z. Hiroi, M. Hanawa, N. Kobayashi, M. Nohara, H. Takagi, Y. Kato, and M. Takigawa, J. Phys. Soc. Jpn. 70, 3377 (2001).
- [14] H. Yoshida, J. Yamaura, M. Isobe, Y. Okamoto, G. J. Nilsen, and Z. Hiroi, Nat. Commun. 3, 860 (2012).
- [15] H. Ishikawa, M. Yoshida, K. Nawa, M. Jeong, S. Krämer, M. Horvatić, C. Berthier, M. Takigawa, M. Akaki, M. T. A.Miyake, K. Kindo, J. Yamaura, Y. Okamoto, and Z. Hiroi, Phys. Rev. Lett. 114, 227202 (2015).
- [16] Y. Okamoto, H. Yoshida, and Z. Hiroi, J. Phys. Soc. Jpn. 78, 033701 (2009).
- [17] R. Okuma, T. Yajima, D. Nishio-Hamane, T. Okubo, and Z. Hiroi, Phys. Rev. B 95, 094427 (2017).

V. SUMMARY

We have synthesized the high-quality polycrystalline samples of $\text{Co}_3\text{V}_2\text{O}_7(\text{OH})_2\cdot 2\text{H}_2\text{O}$ and $\text{BaCo}_3(\text{VO}_4)_2(\text{OH})_2$ via a hydrothermal route and investigated their magnetism and thermodynamic properties. We have shown that they are promising model compounds for the $J_{\text{eff}} = \frac{1}{2}$ kagome-lattice magnets. The observed canted ferromagnetic state with a halfway-saturated magnetic net moment originates from the DM interaction and the large *g* factor, which would be characterized as an emblematic property of the $J_{\text{eff}} = \frac{1}{2}$ kagome magnets. The $J_{\text{eff}} = \frac{1}{2}$ kagome magnets will provide us with a unique platform to study quantum magnetism.

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- [18] R. Okuma, D. Nakamura, T. Okubo, A. Miyake, A. Matsuo, K. Kindo, M. Tokunaga, N. Kawashima, S. Takeyama, and Z. Hiroi, Nat. Commun. 10, 1229 (2019).
- [19] M. Goto, H. Ueda, C. Michioka, A. Matsuo, K. Kindo, and K. Yoshimura, Phys. Rev. B 94, 104432 (2016).
- [20] R. Shirakami, H. Ueda, H. O. Jeschke, H. Nakano, S. Kobayashi, A. Matsuo, T. Sakai, N. Katayama, H. Sawa, K. Kindo, C. Michioka, and K. Yoshimura, Phys. Rev. B 100, 174401 (2019).
- [21] A. Abragam and M. H. L. Pryce, Proc. R. Soc. A 206, 173 (1951).
- [22] M. E. Lines, Phys. Rev. 131, 546 (1963).
- [23] H. Shiba, Y. Ueda, K. Okunishi, S. Kimura, and K. Kindo, J. Phys. Soc. Jpn. 72, 2326 (2003).
- [24] O. Kahn, *Molecular Magnetism* (Wiley-VCH, Weinheim, 1993).
- [25] F. Lloret, M. Julve, J. Cano, R. Ruiz-García, and E. Pardo, Inorg. Chem. Acta 361, 3432 (2008).
- [26] M. Fujihala, X. G. Zheng, Y. Oohara, H. Morodomi, T. Kawae, A. Matsuo, and K. Kindo, Phys. Rev. B 85, 012402 (2012).
- [27] T. Đorđević and L. Karanović, Acta Crystallogr., Sect. C: Struct. Chem. 69, 114 (2013).
- [28] S. B. Ni, T. Li, and X. L. Yang, Mater. Lett. 65, 2662 (2011).
- [29] S. Zhang, H. Yun, and H. Yu, J. Alloys Compd. 735, 700 (2018).
- [30] R. Dessapt, L. Lajaunie, J. J. Calvino, P. Deniard, I. Trenquea, and C. Payen, J. Mater. Chem. C 10, 3287 (2022).
- [31] K. Momma and F. Izumi, J. Appl. Crystallogr. 44, 1272 (2011).
- [32] F. Izumi and K. Momma, Solid State Phenom. 130, 15 (2007).
- [33] S. K. Banerjee, Phys. Lett. 12, 16 (1964).
- [34] P. F. Bongers, C. F. van Bruggen, J. Koopstra, W. P. F. A. M. Omloo, G. A. Wiegers, and F. Jellinek, J. Phys. Chem. Solids 29, 977 (1968).
- [35] B. Huang, G. Clark, E. Navarro-Moratalla, D. R. Klein, R. Cheng, K. L. Seyler, D. Zhong, E. Schmidgall, M. A. McGuire, D. H. Cobden *et al.*, Nature (London) 546, 270 (2017).

- [36] J. Wang, J. Deng, X. Liang, G. Gao, T. Ying, S. Tian, H. Lei, Y. Song, X. Chen, J.-g. Guo, and X. Chen, Phys. Rev. Mater. 5, L091401 (2021).
- [37] J. Yang, B. Chen, H. Wang, Q. Mao, M. Imai, K. Yoshimura, and M. Fang, Phys. Rev. B 88, 064406 (2013).
- [38] S. Kobayashi, H. Ueda, C. Michioka, and K. Yoshimura, Inorg. Chem. 55, 7407 (2016).
- [39] A. Arrott, Phys. Rev. 108, 1394 (1957).
- [40] A. R. Kampf and I. M. Steele, Can. Mineral. 46, 687 (2008).
- [42] P. Svoboda, P. Javorský, M. Diviš, V. Sechovský, F. Honda, G. Oomi, and A. A. Menovsky, Phys. Rev. B 63, 212408 (2001).
- [43] A. B. Harris, C. Kallin, and A. J. Berlinsky, Phys. Rev. B 45, 2899 (1992).
- [44] M. Elhajal, B. Canals, and C. Lacroix, Phys. Rev. B 66, 014422 (2002).
- [45] M. Elhajal, B. Canals, and C. Lacroix, Phys. B (Amsterdam) 312-313, 716 (2002).
- [46] R. Ballou, B. Canals, M. Elhajal, C. Lacroix, and A. S. Wills, J. Magn. Magn. Mater. 262, 465 (2003).

- [47] S.-H. Lee, C. Broholm, M. F. Collins, L. Heller, A. P. Ramirez, Ch. Kloc, E. Bucher, R. W. Erwin, and N. Lacevic, Phys. Rev. B 56, 8091 (1997).
- [48] T. Morimoto, M. Nishiyama, S. Maegawa, and Y. Oka, J. Phys. Soc. Jpn. 72, 2085 (2003).
- [49] A. Scheie, M. Sanders, J. Krizan, Y. Qiu, R. J. Cava, and C. Broholm, Phys. Rev. B 93, 180407(R) (2016).
- [50] J. A. M. Paddison, H. S. Ong, J. O. Hamp, P. Mukherjee, X. Bai, M. G. Tucker, N. P. Butch, C. Castelnovo, M. Mourigal, and S. E. Dutton, Nat. Commun. 7, 13842 (2016).
- [51] D. Grohol, K. Matan, J. H. Cho, S. H. Lee, J. W. Lynn, D. G. Nocera, and Y. S. Lee, Nat. Mater. 4, 323 (2005).
- [52] K. Matan, B. M. Bartlett, J. S. Helton, V. Sikolenko, S. Matas, K. Prokes, Y. Chen, J. W. Lynn, D. Grohol, T. J. Sato, M. Tokunaga, D. G. Nocera, and Y. S. Lee, Phys. Rev. B 83, 214406 (2011).
- [53] I. E. Dzyaloshinskii, J. Phys. Chem. Solids 4, 241 (1958).
- [54] T. Moriya, Phys. Rev. **120**, 91 (1960).
- [55] H. Liu, J. Chaloupka, and G. Khaliullin, Phys. Rev. Lett. 125, 047201 (2020).
- [56] R. Sano, Y. Kato, and Y. Motome, Phys. Rev. B 97, 014408 (2018).