Electronic structure of ThPd₂Al₃: Impact of the U 5*f* states on the electronic structure of UPd₂Al₃

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The electronic structure of ThPd₂Al₃, which is isostructural to the heavy fermion superconductor UPd₂Al₃, was investigated by photoelectron spectroscopy. The band structure and Fermi surfaces of ThPd₂Al₃ were obtained by angle-resolved photoelectron spectroscopy (ARPES), and the results were well explained by the band-structure calculation based on the local density approximation. The comparison between the ARPES spectra and the band-structure calculation suggests that the Fermi surface of ThPd₂Al₃ mainly consists of the Al 3*p* and Th 6*d* states with a minor contribution from the Pd 4*d* states. The comparison of the band structures between ThPd₂Al₃ and UPd₂Al₃ argues that the U 5*f* states form Fermi surfaces in UPd₂Al₃ through hybridization with the Al 3*p* state in the Al layer, suggesting that the Fermi surface of UPd₂Al₃ has a strong three-dimensional nature.

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I. INTRODUCTION

The coexistence of unconventional superconductivity and antiferromagnetic ordering with a relatively large magnetic moment (0.85 $\mu_{\rm B}$ /U) is the most characteristic feature of the heavy fermion compound UPd₂Al₃ [1]. There are many electronic structure studies for UPd₂Al₃; however, the nature of the U 5f state in UPd₂Al₃ remains contradictory. The de Haas-van Alphen (dHvA) study reported that the observed dHvA branches are mostly explained by the band-structure calculation treating the U 5f states as being itinerant [2]. Its electronic structure has also been studied by photoelectron spectroscopy [3-7], and the itinerant nature of the U 5 f states has been reported. A recent Compton scattering study has also reported that the U 5f states have an itinerant character at temperatures lower than T = 20 K [8], which is consistent with the angle-resolved photoelectron spectroscopy (ARPES) study [3]. Meanwhile, a resonant photoemission study on UPd_2Al_3 has also suggested the itinerant nature of the U 5f state, but there exists a correlated satellite structure due to the strong electron correlation effect [9]. In contrast, the transport properties of UPd₂Al₃ are very similar to those of heavy fermion Ce-based compounds, and they have been essentially understood based on the very localized U 5f picture. For example, the temperature dependence of the magnetic susceptibility has been interpreted based on the crystalline electric field scheme [10]. To understand these physical properties consistently, the dual model of the U 5f states has been proposed for this compound [11]. In this scenario, the U 5fstates are formally divided into two subsystems: an itinerant f^1 component and a localized f^2 component. The result of the recent point contact spectroscopy study on UPd₂Al₃ was also interpreted along with this scenario [12], but the microscopic

information of the U 5f states is still lacking. To understand the electronic structure of UPd₂Al₃, it is essential to identify the contribution of the U 5f state to gain a greater insight regarding its nature in this compound.

In the present paper, we have studied the electronic structure of ThPd₂Al₃, which is the f^0 -reference compound of UPd₂Al₃, by photoelectron spectroscopy. We compared the ARPES spectra between ThPd₂Al₃ and UPd₂Al₃, and the contribution from the U 5*f* states to the band structure of UPd₂Al₃ was clarified. Furthermore, ThPd₂Al₃ is a superconductor with $T_C = 0.2$ K [13]. It is the only case where a uranium heavy fermion superconductor has an isostructural counterpart of the thorium compound, which is also a superconductor. Thus, ThPd₂Al₃ is in itself an important target material to study its electronic structure. We found that the band structure and the topology of the Fermi surface of ThPd₂Al₃ are essentially explained by the band-structure calculation based on the local density approximation.

II. EXPERIMENTAL PROCEDURE AND BAND-STRUCTURE CALCULATION

The ThPd₂Al₃ single crystal was grown by the Czochralski method at a pulling speed of 12 mm/h in a tetra-arc furnace under an Ar protective atmosphere. The final shape of the single crystal was a cylinder with a diameter of 2–3 mm and a length of 15 mm. The high quality of the single crystal was confirmed by the Laue method showing sharp spots and a residual resistance ratio RRR = 37 (the extrapolated value $\rho_0 = 0.7 \ \mu\Omega/cm$) after 2 weeks annealing at 900 °C in an evacuated quartz ampule. Photoemission experiments were performed at the soft x-ray beamline BL23SU of SPring-8 [14]. The overall energy resolution in angle-integrated



FIG. 1. Valence-band spectrum of ThPd₂Al₃ measured at hv = 800 eV. Also, the bottom part of this figure provides a comparison of the valence-band spectra of ThPd₂Al₃ and UPd₂Al₃ measured at hv = 800 eV.

photoemission (AIPES) experiments at hv = 800 eV was approximately 150 meV, and that in the ARPES experiments at hv = 660 eV was approximately 90 meV. Clean sample surfaces were obtained by cleaving the samples *in situ* perpendicular to the *c* axis under ultrahigh vacuum conditions. The vacuum during the course of measurements was typically $<2 \times 10^{-8}$ Pa, and the sample surfaces were stable for the entire duration of the measurements (about 2 days) since no significant changes had been observed in the ARPES spectra during these periods. The positions of the ARPES cuts in the momentum space were determined by assuming a freeelectron final state with an inner potential of $V_0 = 12 \text{ eV}$.

In the band-structure calculations, relativistic linear augmented plane-wave (RLAPW) calculations [15] within the local density approximation (LDA) [16] were performed, treating all U 5*f* electrons as itinerant. In this approach, the Dirac-type Kohn-Sham equation has been formulated, and the spin-orbit interaction is exactly taken into account [17]. To compare the results of the calculation with the ARPES spectra, we have simulated the ARPES spectral functions on the basis of the band-structure calculation. In the simulation, the following effects were taken into account: (i) the broadening along the k_{\perp} direction due to the finite escape depth of the photoelectrons, (ii) the lifetime broadening of the photohole, (iii) the photoemission cross sections of orbitals, and (iv) the energy resolution and the angular resolution of the electron analyzer. The details are outlined in Ref. [18].

III. RESULTS AND DISCUSSION

Figure 1 shows the AIPES valence-band spectrum of ThPd₂Al₃ measured at hv = 800 eV and its comparison with

that of UPd₂Al₃. According to the photoionization cross sections, the contributions from the Pd 4d and U 5f states are dominant in this photon energy [19], and thus three prominent peaks located between $E_{\rm B} = 2.5$ and 6 eV are ascribed to the contributions from the Pd 4d states. The bottom part of this figure demonstrates the comparison between the valence-band spectra of ThPd₂Al₃ and UPd₂Al₃ measured at $h\nu = 800$ eV. The comparison shows that the sharp peak at the Fermi energy exists only in the spectrum of UPd₂Al₃, suggesting that the peak found at the Fermi energy represents the contribution of the U 5f states in UPd₂Al₃. This is consistent with the result of the resonant photoemission experiment for UPd₂Al₃ in which a similar similar sharp peak was observed at the Fermi energy of UPd_2Al_3 [9]. A further important point to note is that all three Pd 4d peaks in the valence-band spectrum of UPd₂Al₃ are located in higher binding energies than those in the valence-band spectrum of ThPd₂Al₃. A very similar rigid shift of the transition metal d bands has been observed in the valence-band spectra of URu₂Si₂ and ThRu₂Si₂ [20]. The amount of energy shift is approximately 200 meV, which is also very similar to the case between ThRu₂Si₂ and URu₂Si₂. This energy shift of the *d* band is in contrast with the case of the U 5 f localized compound UPd₃ where no shift was observed in the Pd 4d bands between ThPd₃ and UPd₃ [21]. The rigid shift in the Pd 4d states between UPd₂Al₃ and ThPd₂Al₃ suggests that the U 5f states are involved in the valence-band structure of UPd₂Al₃, and they are not impuritylike contributions as expected when the U 5f states are almost localized. Next, we discuss the band structure and Fermi surface of ThPd₂Al₃ measured by ARPES. Figure 2 summarizes the result of the ARPES study of ThPd₂Al₃. Figure 2(a) shows the ARPES spectra of ThPd₂Al₃ along the $K-M-K-\Gamma-M-\Gamma$ high-symmetry line measured at hv = 660 eV. The spectra consist of two different types of energy dispersions, namely very dispersive bands at $E_{\rm F} \lesssim E_{\rm B} \lesssim 2.5~{\rm eV}$ and less dispersive bands with enhanced intensities distributed at $E_{\rm B} \gtrsim 2.5 \text{ eV}$. Since the photoionization cross section of the Pd 4d states is more than 30 times enhanced than those of Al 3s, 3p and Th 6*d* states [19], the prominent dispersions at $E_{\rm B} \gtrsim 2.5 \text{ eV}$ are the contributions from the Pd 4d states. Furthermore, the dispersive bands at $E_{\rm F} \lesssim E_{\rm B} \lesssim 2.5$ eV reflect the contributions mainly from the Al and Th states.

Figure 2(b) shows the calculated band structure and the simulation of the ARPES spectra based on the band-structure calculation. The solid lines and the density plot represent the calculated energy dispersions and the simulation, respectively. The color coding of each band represents the contribution from the Pd 4d states. The experimental band structure is well explained by the band-structure calculation. For example, the parabolic dispersions centered at the Γ point that forms the Fermi surface are in good agreement with the calculated bands 16-18. There are also parabolic dispersions along the *K*-*M*-*K* high-symmetry line, which are also well explained by the calculated band 15. At higher binding energies, the nearly flat Pd 4d bands in the experimental ARPES spectra agree with the bands 2-13 in the calculation. Although the Pd 4dbands are not clearly resolved one by one in the experimental spectra, the overall features agree very well between experimental ARPES spectra and the calculated band structure. Note that the contribution from the Pd 4d states is distributed in



FIG. 2. Band structure and Fermi surface of ThPd₂Al₃ obtained by ARPES. (a) ARPES spectra of ThPd₂Al₃ measured along the *K-M-K*- Γ -*M*- Γ high-symmetry line at hv = 660 eV. (b) Calculated band structure and simulation of the ARPES spectra based on the band-structure calculation. (c) Fermi-surface map obtained by the integration of the ARPES spectra over 100 meV across $E_{\rm F}$. (d) Calculated Fermi surface and the simulation of the experimental Fermi-surface map. (e) Three-dimensional shape of the calculated Fermi surfaces.

higher binding energies ($E_B \ge 2.5$ eV), and the Fermi surface of ThPd₂Al₃ mainly consists of the Al and Th states. This situation is significantly different from the case of ThRu₂Si₂ where the transition metal *d* band is close to the Fermi energy, and the *d* bands are hybridized with the U 5*f* states in URu₂Si₂ [20,22]. We further discuss the orbital character of the Fermi surface of ThPd₂Al₃ and UPd₂Al₃ in the latter part of this section.

Figure 2(c) shows the experimental Fermi-surface map of ThPd₂Al₃, which was obtained by integrating the photo emission intensity over 100 meV across $E_{\rm F}$. There is a hexagonal-shaped feature with enhanced intensity centered at the Γ point. In addition, the intensity at the K point is also enhanced and these points are connected to each other and form a very complex shape. In Fig. 2(d), we illustrate the calculated Fermi surface using solid curves, and we also demonstrate the simulation of the Fermi-surface map based on the bandstructure calculation as a density plot. The three-dimensional shape of the calculated Fermi surfaces of ThPd₂Al₃ is also shown in Fig. 2(e). In the band-structure calculation, the bands 16-18 form Fermi surfaces. Bands 16 and 18 form a holetype Fermi surface around the M point and an electron-type Fermi surface around the Γ point, respectively. In contrast, band 17 forms a very complicated Fermi surface with a three-dimensional shape, but forms the electron-type Fermi surface with a starlike shape around the Γ point. Experimental and calculated Fermi surfaces agree very well although the features are broader in the experimental map compared to the calculated map. Despite the fact that the details of the experimental Fermi surface are not very clear, the topology of the Fermi surface agrees with the result of the band-structure

calculation. Accordingly, the band structure and Fermi surface of $ThPd_2Al_3$ were well explained by the band-structure calculation.

Next, we performed a comparison between the ARPES spectra of ThPd₂Al₃ and UPd₂Al₃ to reveal the contribution of the U 5 f states in the band structure of UPd₂Al₃, as shown in Fig. 3. Figures 3(a) and 3(b) represent the experimental ARPES spectra of ThPd₂Al₃ and the corresponding calculated band structure, respectively. The color coding of the calculated bands corresponds to the contributions from the Th 6d and the Al 3p states. Note that the contribution from the Al 3s is almost negligible in these binding energies. The experimental ARPES spectra are well explained by the bandstructure calculation, as discussed in Figs. 2, and thus the orbital character of each band should also agree with the result of the band-structure calculation. The calculation suggests that the Fermi surface of ThPd₂Al₃ mainly consists of the Al 3p and Th 6d states, but it has an enhanced Al 3p character. In particular, bands 17 and 18, which form the Fermi surfaces, have a dominant contribution from the Al 3p states, and the overall good agreement between the experimental data and the respective calculations suggests that the Fermi surfaces should have an enhanced contribution from the Al 3pstates.

Figures 3(c) and 3(d) show the experimental ARPES spectra of UPd₂Al₃ and the corresponding calculated band structure. The ARPES spectra of UPd₂Al₃ were taken from Ref. [7]. The U 5*f* difference spectrum obtained by a resonant photoemission measurement [9] and the calculated U 5*f* density of states (DOS) are shown in the right panel of Fig. 3(c). The ARPES spectra were recorded at hv = 600 eV, and the



FIG. 3. Comparison between the ARPES spectra of ThPd₂Al₃ and UPd₂Al₃, and the result of the band-structure calculations. (a) ARPES spectra of ThPd₂Al₃ measured along the Γ -*K*-*M* high-symmetry line at hv = 660 eV. (b) Result of the band-structure calculation for ThPd₂Al₃. The color coding of each band represents the contributions from the Th 6*d* and the Al 3*p* states. (c) ARPES spectra of UPd₂Al₃ measured along the Γ -*K*-*M* high-symmetry line at hv = 600 eV. Data adapted from Ref. [7]. The U 5*f* difference spectrum obtained by resonant photoemission measurement [9] and the calculated U 5*f* DOS are shown in the right panel. (d) Result of the band-structure calculation for UPd₂Al₃. The color coding of each band represents the contributions from the U 5*f*, U 6*d*, and the Al 3*p* states.

sample temperature was 20 K. The experimental energy dispersions of ThPd₂Al₃ and UPd₂Al₃ are very similar to each other, but the intensity of the energy dispersions distributed at $E_{\rm B} = E_{\rm F} - 1.2$ eV is enhanced in the spectra of UPd₂Al₃. Moreover, there exist very flat features in the vicinity of $E_{\rm F}$, which represent the contributions from the U 5*f* states since the photoionization cross section of the U 5*f* states is dominant at the used photon energy. The overall structures of the experimental spectra of UPd₂Al₃ are also essentially explained by the band-structure calculation although the detail of each dispersion was not resolved experimentally. In particular, the very flat features at $E_{\rm F}$ originate from the renormalized U 5*f* bands due to the strong electron correlation effect.

The comparison between the ARPES spectra of ThPd₂Al₃ and UPd_2Al_3 indicates that the U 5f states are strongly hybridized with the non-f dispersive bands in ThPd₂Al₃ which correspond to the calculated bands 15-18 of ThPd₂Al₃. These calculated bands have an enhanced contribution from the Al 3p states, suggesting that the U 5f states are strongly hybridized with the Al 3p states in UPd₂Al₃. The crystal structure of UPd₂Al₃ consists of alternating stacks of U-Pd and Al layers along the c axis, and the presence of the enhanced U 5*f*-Al 3*p* hybridization suggests that the U 5*f* states have a strong three-dimensional nature similar to the case of the heavy fermion compound URu₂Si₂ [22]. As a result, the Fermi surface of UPd₂Al₃ should also have a strong three-dimensional nature due to the enhanced U 5f-Al 3phybridization. Previous ARPES studies of UPd₂Al₃ [4,6] have shown a cylindrical Fermi surface along the Γ - A line. The Fermi surface has an enhanced contribution from the U 5fstates, but the present result suggests that it also has an enhanced contribution from the Al 3p states.

Note that the U 5f states have a strong electron correlation effect, and the U 5f states in UPd₂Al₃ have an incoherent peak distributed at approximately $E_{\rm B} = 0.2-1$ eV [9]. Thus, the experimental bands around this binding energy should have the contribution from the incoherent component of the U 5f states. This is consistent with the dual nature of the U 5f states in UPd_2Al_3 as proposed theoretically, where the heavy quasiparticle band is described by the effective renormalized theory while the high-energy structure is explained by the multiplet sidebands arising from the Hund's rule [23,24]. Experimentally, there are no nondispersive but dispersive bands in this binding energy region, and thus the energy dispersions in $E_{\rm B} = 0.2$ -1 eV are hybridized with the incoherent U 5f states. It is theoretically proposed that the localized multiplet bands are hybridized with dispersive bands with non-f character [24], and the experimental spectra are consistent with the theory.

IV. CONCLUSION

In the present study, we investigated the electronic structure of ThPd₂Al₃ using photoelectron spectroscopy, and the results were compared with the band-structure calculation and the spectra of the isostructural heavy fermion superconductor UPd₂Al₃. The Pd 4*d* states in the valence-band spectrum of ThPd₂Al₃ were found to be shallower than those in the valence-band spectrum of UPd₂Al₃ by approximately 200 meV, suggesting that the U 5*f* states are involved in the valence-band structure of UPd₂Al₃, and are not impuritylike contributions. The Fermi surface and the band structure of ThPd₂Al₃ obtained by ARPES were well explained by the band-structure calculation. The electronic structure in the very vicinity of E_F is dominated by contributions from the Al 3*p* and Th 6*d* states with minor contributions from the Pd 4*d* states. The comparison between the ARPES spectra of ThPd₂Al₃ and UPd₂Al₃ suggests that the electronic structure of UPd₂Al₃ in the very vicinity of E_F is dominated by the enhanced U 5*f*-Al 3*p* hybridization. This indicates that the electronic structure of UPd₂Al₃ has a three-dimensional nature [25–31].

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