

High-pressure structure and electronic properties of YbD₂ to 34 GPa

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Ytterbium dihydride (YbH₂) shows a well-known transition at ≈ 16 GPa from a CaH₂-type structure to a high-pressure (high-*P*) phase with Yb at hcp sites and unknown H positions. Here, we report its complete structure determination by neutron diffraction at 34 GPa. Hydrogen (deuterium) is located at $2a$ and $2d$ positions of space group $P6_3/mmc$, thus forming a high-symmetry “collapsed” close-packed lattice. The transition is sluggish and can be seen as a transfer of $1/2$ of the hydrogen atoms from strongly corrugated H layers to interstitial sites of the Yb lattice. We demonstrate by first-principles calculations that the transition is related to a change from a completely filled *f*-electron configuration to a fractional *f*-hole (≈ 0.25 h) occupation in the high-*P* phase. The $f \rightarrow d$ charge transfer closes the gap at the transition and leads to a metallic ground state with a sizable electron-phonon interaction involving out-of-plane vibrational modes of interstitial hydrogen.

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There is considerable recent interest in hydrides under high pressure following the discovery of high- T_c superconductivity in H₂S compressed to 150 GPa [1]. Indeed, theory [2] and recent experiments [3] give strong indications that superconductivity at elevated temperatures might occur in numerous hydrides, in particular, binary rare-earth hydrides [4] with a high hydrogen (H) content, which are unstable at ambient pressure. The case of ytterbium (Yb) hydrides appears to be interesting for the following reasons: Elemental Yb was recently found to be superconducting beyond 80 GPa [5]. This is highly unexpected, since Yb is diamagnetic at ambient conditions [6], and pressure is believed to turn it magnetic through a $f^{14} \rightarrow f^{13}$ (Yb²⁺ \rightarrow Yb³⁺) valence change [7] supported by extended x-ray absorption fine structure (EXAFS) [8], x-ray absorption near edge structure (XANES) [5], and resonant inelastic x-ray scattering (RIXS) [9] data, thus unlikely to be superconducting.

This surprising finding draws attention to its most common hydride, YbH₂. This material is insulating and crystallizes at 0 GPa in the α phase, whose structure is of CaH₂ orthorhombic type (space group $Pnma$) [10]. Under pressure, it transforms at ≈ 16 – 20 GPa into a phase with Yb sites at a hexagonal-close-packed (hcp) lattice [11]. The H atom positions are presently unknown. Again, this first-order transition ($\Delta V/V = 3\%$) might be driven by a valence change of Yb from $2+$ to $3+$, as indicated by EXAFS [12]. The YbH₂ high-pressure (high-*P*) phase is stable to at least 60 GPa [12], and its electronic properties are unknown.

Given this context, it appears timely to determine the full structure of YbH₂, i.e., the H position as well as its electronic properties beyond 16 GPa. Generally speaking, experimental data on the H location in high-*P* metal hydride phases are

extremely sparse. Almost all structural information above a few GPa has been obtained by synchrotron x-ray diffraction which is blind to H in the presence of heavy atoms. So far, the H positions are simply assumed to be on “favorable” interstitial sites, or deduced from first-principles calculations [13–15].

Here, we present high-*P* neutron diffraction data to 34 GPa which determine the structure of YbH₂, in particular, the H positions. We use the structural data to determine its electronic properties through *ab initio* methods, which indicate a semiconducting-to-metal transition concomitant with the structural transformation. The metallization is driven by an $f \rightarrow d$ charge transfer with partial *f*-hole unbinding.

We used deuterated samples (YbD₂) for the well-known fact that H is a strong incoherent scatterer. From the x-ray diffraction data it is clear that deuteration has no significant structural effect, even at high pressure [12]. Ytterbium has 70 electrons and hence scatters ≈ 5000 times stronger than deuterium (D), i.e., an x-ray diffraction pattern is completely dominated by Yb. This is not the case for neutrons: The coherent neutron cross sections of Yb and D are 19.4 and 5.6 b, i.e., are of the same order of magnitude. Neutron diffraction is most likely the only technique to solve the problem.

The sample was synthesized by heating Yb powder (99.9% metal purity, fresh filings with ca. 0.1 mm grain size from a rod purchased from Goodfellow) in 2000 hPa deuterium atmosphere to 600 °C, hence similar to the procedure described in previous work [10]. Neutron diffraction data of ground powder taken at ambient conditions in a vanadium can reveal single-phase material with the expected CaH₂ structure, space group $Pnma$, $a = 5.8823(2)$ Å, $b = 3.5676(1)$ Å, $c = 6.7588(2)$ Å, and fractional atomic positions $x(\text{Yb}) = 0.2396(3)$, $z(\text{Yb}) = 0.1119(1)$, $x(\text{D1}) = 0.3556(3)$, $z(\text{D1}) = 0.4291(3)$, $x(\text{D2}) = -0.0305(5)$, $z(\text{D2}) = 0.6784(3)$. As expected, the compound is slightly nonstoichiometric with a

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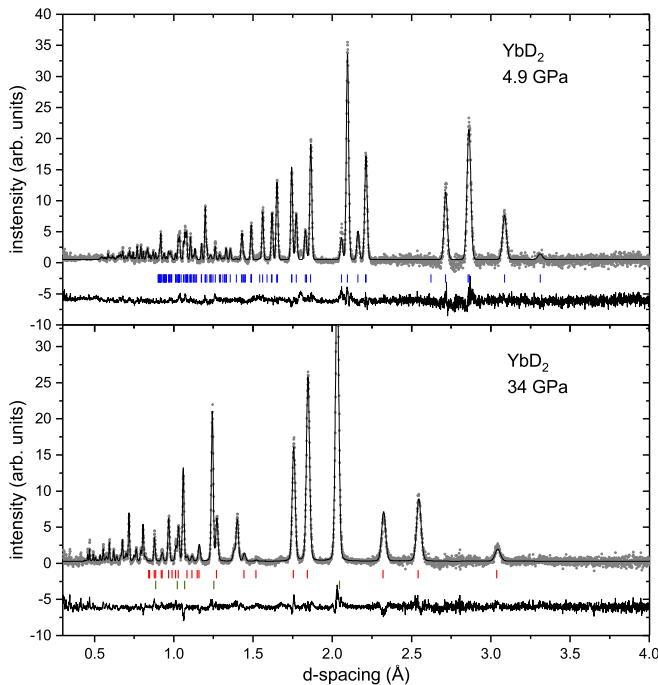


FIG. 1. Neutron diffraction patterns of YbD_2 at 298 K, in the low- P $Pnma$ phase (upper panel) and the high- P $P6_3/mmc$ phase (lower panel). The lines are Rietveld fits to the data (dots). Upper tick marks indicate Bragg reflections of the sample, and lower tick marks of diamond from the anvils. $\chi^2 = 2.05$, $R_{wp} = 7.95\%$ (top); $\chi^2 = 2.27$, $R_{wp} = 8.05\%$ (bottom). Accumulation time is 1 and 2 h, for 4.9 and 34 GPa pressure, respectively.

refined D composition of 1.912(4) instead of 2. This is a well-known phenomenon and was observed in all previous studies on YbD_2 [10]. For the sake of simplicity, we will call the sample YbD_2 throughout the text.

Neutron diffraction measurements were carried out at the high- P beamline PLANET [16] at MLF, the Japan Proton Accelerator Research Complex (J-PARC), Tokai, Ibaraki, Japan. The three high- P runs used three types of double-toroidal sintered diamond anvils [17,18] with maximum sample volumes of 12, 7, and 3 mm³, encapsulating TiZr gaskets and a 4:1 methanol-ethanol mixture as the pressure transmitting fluid. All runs applied a VX4-type Paris-Edinburgh load frame [17] with the position of the sample maintained to within ± 0.1 mm relative to the laboratory frame. The pressure values cited here were determined from the equation of state (EOS) of YbD_2 reported by the x-ray work [12], using the measured ambient-pressure unit cell volume ($V_0 = 141.84 \text{ \AA}^3$) for the low-pressure (low- P) phase. In the first run to 22.6 GPa the sample was temporarily heated to 363 K in each pressure ramp-up to keep the sample hydrostatic up to 16 GPa. In the runs to 26 and 34 GPa compressions were made at 299 K.

Figure 1 shows diffraction patterns along with Rietveld refinements [19] of YbD_2 in the low- P α phase at 4.9 GPa and in the high- P phase at 34 GPa. The transition is found to be sluggish in all runs (see Fig. 2), and heating to 363 K in the first loading had no significant effect on its kinetics. It starts at 20 GPa (approximately consistent with previous data) and

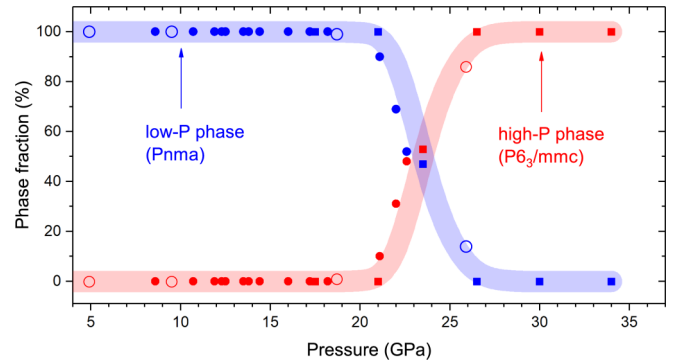


FIG. 2. Phase fraction of YbD_2 as pressure is increased (up-stroke). Solid circles, open circles, and squares indicate three different runs to 22.6, 26, and 34 GPa, respectively. Broad lines are guides to the eye.

ends slightly above 26 GPa, thus higher than what has been reported from x-ray studies [11].

The determination of the high- P structure is facilitated by the fact that the heavy-element (Yb) sublattice is known to be hcp [11] and the number of possibilities of incorporating H(D) therein with the required stoichiometry is limited. With Yb placed at the $2c$ position of space group $P6_3/mmc$, nine configurations were tested with H(D) on the remaining $2a$, $2b$, $2d$, and $4f$ positions. These include hence the tetrahedral ($2a$) and octahedral ($4f$) sites which are preferentially occupied by H(D) in various other hcp metals. Wyckoff sites with multiplicity 6, 12, and 24 would place H(D) on very low-symmetry sites with partial occupancy which hardly can be justified given the high-symmetry environment of the Yb sublattice. Pattern simulations [20] show that out of these nine configurations, *only one* is compatible with the measured diffraction data, and all others give strongly different intensities and hence can be safely excluded. The structure consists of one H(D) on the $2a$ position (on the octahedral hcp site) and the other on the $2d$ site (see Fig. 3). The refinement of the pattern at 34 GPa gives $a = 3.5088(2) \text{ \AA}$, $c = 4.6413(3) \text{ \AA}$. Apart from lattice constants, the refinements include only a minimal set of parameters, i.e., phase fraction, profile, and preferred orientation. Interestingly, this H(D) configuration was guessed 36 years ago [10] from purely geometrical arguments applied to the low- P α phase.

Inspection of the α phase along its b axis reveals an interesting relationship with the high- P phase, and details of the transition mechanism: In the α phase, Yb is on a distorted hcp lattice, the c/b ratio at 15 GPa is 1.90 compared to 1.633 for an ideal hcp structure, and there are further small displacements along the orthorhombic a axis. H(D) in this structure is stored in strongly corrugated layers separated from the Yb layers. The shortest H-H distance (d_{HH}) inside a layer is 2.64 \AA . The high- P transition renders a relatively irregular arrangement into a highly symmetric crystal, by transferring half of all H(D) into planar interstitials of the Yb layers. By this mechanism, the in-plane d_{HH} becomes larger and equal to 3.56 \AA , whereas the shortest d_{HH} between neighboring planes reduces only slightly to 2.37 \AA . This picture is confirmed by phonon calculations in the high- P phase, where its low-pressure instability is driven by a phonon softening at $\mathbf{q} = M$. The related distortion implies the doubling of the hexagonal

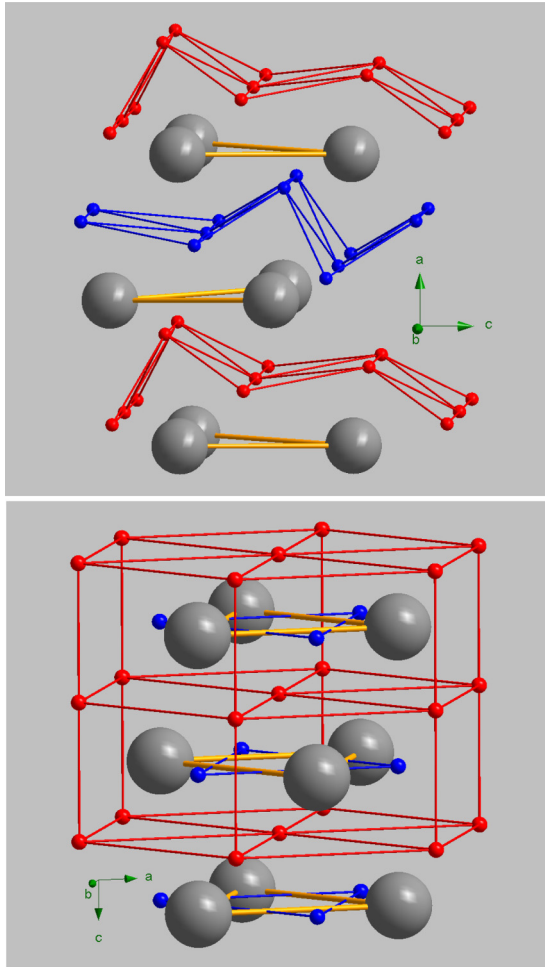


FIG. 3. Structure of YbD_2 in its low- P $Pnma$ (top) and the high- P $P6_3/mmc$ phases (bottom). Large symbols are Yb atoms, and small symbols H(D). Different colors for H(D) highlight different layers. Lines are guides to the eye, and arrows refer to the orthorhombic (top) and hexagonal (bottom) axes.

unit cell, with a phonon pattern that brings this structure back to the known α phase [20].

Such an H arrangement seems to be unique among all hexagonal transition and rare-earth (R) hydrides where the hydrogen positions are known from neutron diffraction. In the well-studied monohydrides FeH and CrH, hydrogen is located exclusively at the octahedral sites between the hcp layers [21,22]. In the hexagonal rare-earth hydrides (all with compositions RH_3) it is located at both octahedral and tetrahedral sites, or close to them, i.e., again *between* the metal planes [23]. High- P YbH_2 seems therefore to be the only hydride known up to now adopting a structure of this type.

TABLE I. Atomic orbital occupations per YbH_2 unit for the α (ambient pressure) and high- P (26 GPa) phases.

Orbital symmetry	α	High- P
$4f(\text{Yb})$	14.00	13.75
$5d(\text{Yb})$	1.30	1.80
$6s6p(\text{Yb}) + 1s(\text{H}_2)$	2.70	2.45

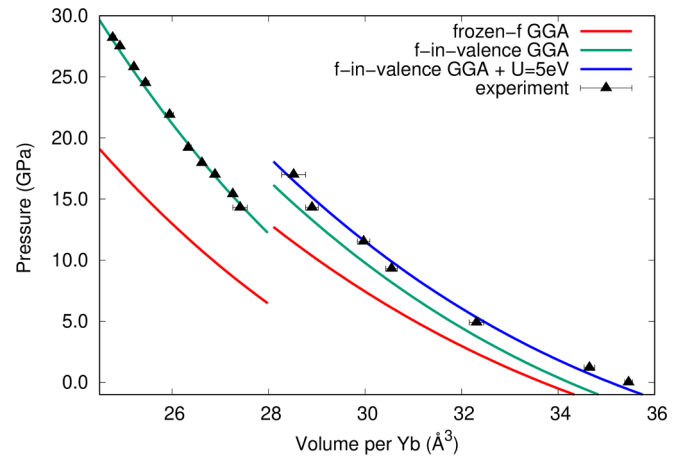


FIG. 4. EOS for α and high- P phases by *ab initio* calculations with different flavors, and compared with experiment for YbH_2 [11]. DFT-GGA calculations with the PBE functional are shown in red and green, for the frozen- f and f -in-valence PAW pseudopotentials, respectively. DFT-GGA calculations with f -in-valence pseudopotential plus Hubbard repulsion (GGA + U) are shown in blue for the α phase.

Knowing the structure of YbD_2 allows us to derive electronic properties through first-principles calculations at fixed experimental geometries. We carried out density functional theory (DFT) calculations within the generalized gradient approximation (GGA) built in the Perdew-Burke-Ernzerhof (PBE) functional [24,25]. We used the plane-wave implementation as coded in the QUANTUM ESPRESSO package [26,27]. The key question we address here is about the role of f electrons in driving the structural transition. Hypotheses have been made about a possible valence change in YbH_2 between the α and high- P phase, where a tight competition between the atomic $4f^{14}(5d6s6p)^2$ ($2+$) and $4f^{13}(5d6s6p)^3$ ($3+$) configurations could be at play [11]. Yb and Eu are the rare-earth elements where divalent and trivalent states are the closest in energy [28], and they are the only ones where the $2+$ valence is the most stable in the solid state at ambient pressure [29]. To investigate the role of the f electrons in YbH_2 , we ran two types of calculations, one with the f manifold frozen in the Yb pseudopotential, the other with in-valence f electrons. Both pseudopotentials [projector augmented-wave (PAW) type] are fully relativistic and keep the $5s5p$ semicore states in valence [30].

The results for the EOS are shown in Fig. 4 and compared with experimental data for YbH_2 [11]. The geometries used in our DFT calculations have both Yb and H positions determined from experiment [31]. The frozen- f calculations poorly reproduce the experimental EOS, with a significant pressure discrepancy at the phase transition. The f -in-valence calculations substantially improve the agreement with the experiment, signaling the importance of the structural feedback on the f -band shape. In the α phase, the agreement is further improved by including an Hubbard repulsion term within the DFT + U scheme [32], while in the high- P phase the agreement between theory and experiment is very good already at the DFT level. This points towards an f manifold more correlated in the α than in the high- P phase. The experimental

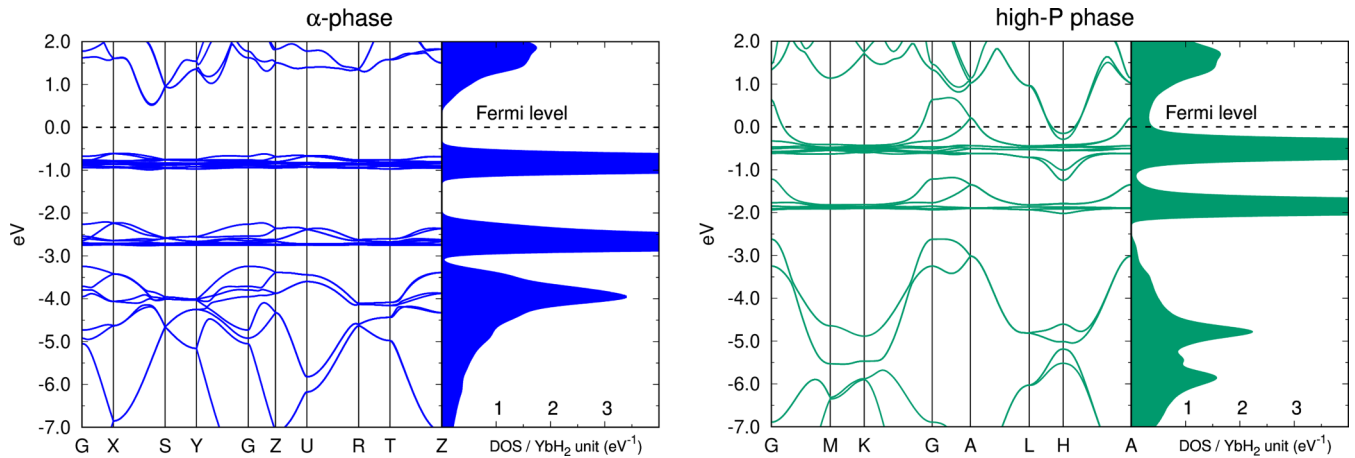


FIG. 5. Band structure and DOS for the α phase at ambient pressure (left panel), and for the high- P phase at 26 GPa (right panel). In both cases, calculations are done with the fully relativistic PBE functional and f -in-valence PAW pseudopotentials.

knowledge of both Yb and H positions gives us a unique chance to assess the validity of the GGA approximation with f -in-valence electrons for this kind of system. Other widely used DFT approximations, such as the local density one (LDA), turn out to perform more poorly, as shown by additional calculations we report in the Supplemental Material (SM) [20].

To gain insight into the role played by the f electrons as valence states, we plot in Fig. 5 the band structure and the density of states (DOS) at ambient and high pressure for the approximations reproducing best the experimental EOS, i.e., the DFT and DFT + U for the high- P and α phase, respectively. The ambient-pressure phase is a semiconductor, in agreement with experiment, with a gap of about 1 eV between the empty states (mainly of d character) and the narrow $f_{j=7/2}$ bands. In contrast, the f manifold in the high- P phase has a much wider bandwidth, because of strong hybridization with the d states. This is a typical signature of the f -electron delocalization. The f states are pushed up in energy, and the $f_{7/2}$ multiplet crosses the Fermi level. The high- P phase is hence clearly a metal. The $f_{j=7/2}$ - $f_{j=5/2}$ splitting of more than 1 eV is a straightforward manifestation of the strong spin-orbit coupling (SOC) in YbH_2 . While this splitting significantly reduces the band gap in the insulating α phase, the fermiology of the metallic phase seems to be only marginally affected by SOC (see SM [20]).

The f -character change across the transition is also reflected by the atomic orbital occupation analysis, reported in Table I. One can see that the $P6_3/mmc$ phase is in a mixed valence configuration, through the formation of conducting $f_{7/2}$ holes with a fractional occupation (0.25 per YbH_2 unit at 26 GPa). There is also a simultaneous increase of the $5d$ occupation and a slight depletion of the $6s6p$ manifold. In other words, the structural transition is accompanied by a charge transfer towards the d orbitals, and by the partial delocalization of f holes. Both phenomena are responsible for

a tighter chemical bond between two neighboring Yb atoms, whose distance is indeed much shorter in the high- P phase [33].

The hybridization between the $4f$ and $5d$ orbitals increases the f bandwidth and thereby reduces correlations. This rationalizes the fact that the EOS of the high- P phase is well described already at the DFT level, while in the α phase, showing very flat f electrons and band-insulating character, the addition of an explicit Hubbard term seems necessary for a quantitative agreement with the experiment. It is interesting to note that, qualitatively, one can get the same physical picture for the transition even without U . The only difference is a smaller band gap in the α phase (0.2 eV) and, consequently, a lower transition pressure obtained by standard GGA (see Fig. 4 and SM [20]). We verified that the metallicity of this phase is robust against the possible occurrence of magnetic order arising from the f -hole moments [34]. Neither a ferromagnetic nor an interlayer antiferromagnetic order is a stable ground state at the DFT level.

Given its metallic character, superconductivity in high- P YbH_2 appears to be a possibility to explore. Calculations reveal a non-negligible electron-phonon coupling involving the out-of-plane modes of interstitial hydrogen at 26 GPa. However, the total integrated coupling is not large enough to yield a sizable T_c at these pressures [20]. Superconductivity could nevertheless emerge at higher pressures, possibly driven by enhanced charge fluctuations [35] in the mixed-valence regime.

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[1] A. Drozdov, I. Erements, I. Troyan, V. Ksenofontov, and S. Shylin, *Nature (London)* **525**, 73 (2015).

[2] D. Duan, Y. Liu, Y. Ma, Z. Shao, B. Liu, and T. Ciu, *Nat. Sci. Rev.* **4**, 121 (2016).

- [3] M. Somayazulu, M. Ahart, A. K. Mishra, Z. M. Geballe, M. Badini, Y. Meng, V. V. Struzhkin, and R. J. Hemley, *Phys. Rev. Lett.* **122**, 027001 (2019).
- [4] F. Peng, Y. Sun, C. J. Pickard, R. J. Needs, Q. Wu, and Y. Ma, *Phys. Rev. Lett.* **119**, 107001 (2017).
- [5] J. Song, G. Fabbris, W. Bi, D. Haskel, and J. S. Schilling, *Phys. Rev. Lett.* **121**, 037004 (2018).
- [6] E. Bucher, P. Schmidt, A. Jayaraman, K. Andres, J. Maita, K. Nassau, and P. Dernier, *Phys. Rev. B* **2**, 3911 (1970).
- [7] J. Lock, *Proc. Phys. Soc., London, Sect. B* **70**, 476 (1957).
- [8] K. Syassen, G. Wortmann, J. Feldhaus, K. H. Frank, and G. Kaindl, *Phys. Rev. B* **26**, 4745 (1982).
- [9] C. Dallera, O. Wesseley, M. Colarieti-Tosti, O. Eriksson, R. Ahuja, B. Johansson, M. I. Katsnelson, E. Annese, J.-P. Rueff, G. Vankó, L. Braicovich, and M. Grioni, *Phys. Rev. B* **74**, 081101(R) (2006).
- [10] B. Lebech, N. Hessel Andersen, S. Steenstrup, and A. Schroder Pedersen, *Acta Crystallogr. C* **39**, 1475 (1983).
- [11] J. Olsen, B. Buras, L. Gerward, B. Johansson, B. Lebech, H. Skriver, and S. Steenstrup, *Phys. Scr.* **29**, 503 (1984).
- [12] J. Olsen, S. Steenstrup, and L. Gerward, in *High Pressure Science and Technology: Proceedings of the Joint XV AIRAPT and XXXIII EHPRG International Conference*, edited by W. A. Trzeciakowski (World Scientific, Singapore, 1996), p. 549.
- [13] I. Goncharenko, M. I. Erements, M. Hanfland, J. S. Tse, M. Amboage, Y. Yao, and I. A. Trojan, *Phys. Rev. Lett.* **100**, 045504 (2008).
- [14] B. Li, Y. Ding, D. Y. Kim, R. Ahuja, G. Zou, and H.-K. Mao, *Proc. Natl. Acad. Sci. USA* **108**, 18618 (2001).
- [15] C. M. Pépin, A. Dewaele, G. Geneste, P. Loubeyre, and M. Mezouar, *Phys. Rev. Lett.* **113**, 265504 (2014).
- [16] T. Hattori, A. Sano-Furukawa, H. Arima, K. Komatsu, Y. Yamada, Y. Inamura, T. Nakatani, Y. Seto, T. Nagai, W. Utsumi, T. Itaka, H. Kagi, Y. Katayama, T. Inoue, T. Otomo, K. Suzuya, T. Kamiyama, M. Arai, and T. Yagi, *Nucl. Instrum. Methods Phys. Res., Sect. A* **780**, 55 (2015).
- [17] S. Klotz, *Techniques in High Pressure Neutron Scattering* (Taylor and Francis/CRC, Boca Raton, FL, 2013).
- [18] T. Hattori, A. Sano-Furukawa, S. Machida, J. Abe, K. Funakoshi, H. Arima, and N. Okazaki, *High Press. Res.* **39**, (2019), doi: 1080/08957959.2019.1624745.
- [19] J. Rodríguez-Carvajal, *Physica B* **192**, 55 (1993).
- [20] See Supplemental Material at <http://link.aps.org/supplemental/10.1103/PhysRevB.100.020101> for the determination of the hydrogen/deuterium location through the simulation of the neutron diffraction pattern, the thermal expansion of rare-earth dihydrides [36], the assessment of pseudopotentials, density functionals, Hubbard and SOC interactions by direct comparison with experimental geometries and EOS, the structural instability studied by means of phonon calculations, and the electron-phonon coupling calculations.
- [21] V. Antonov, M. Baier, B. Dorner, V. Fedotov, G. Grosse, A. Kolznikov, E. Ponyatovsky, G. Schneider, and F. Wagner, *J. Phys.: Condens. Matter* **14**, 6427 (2002).
- [22] G. Albrecht, F. Doenitz, K. Kleinstück, and M. Betzl, *Phys. Status Solidi* **3**, K249 (1963).
- [23] K. Mackay, *Hydrogen Compounds of the Metallic Elements* (E. and F. N. Spon, London, 1966).
- [24] J. P. Perdew, K. Burke, and M. Ernzerhof, *Phys. Rev. Lett.* **77**, 3865 (1996).
- [25] J. P. Perdew, K. Burke, and M. Ernzerhof, *Phys. Rev. Lett.* **78**, 1396 (1997).
- [26] P. Giannozzi, S. Baroni, N. Bonini, M. Calandra, R. Car, C. Cavazzoni, D. Ceresoli, G. L. Chiarotti, M. Cococcioni, I. Dabo, A. D. Corso, S. de Gironcoli, S. Fabris, G. Fratesi, R. Gebauer, U. Gerstmann, C. Gougoussis, A. Kokalj, M. Lazzeri, L. Martin-Samos *et al.*, *J. Phys.: Condens. Matter* **21**, 395502 (2009); P. Giannozzi, O. Andreussi, T. Brumme, O. Bunau, M. B. Nardelli, M. Calandra, R. Car, C. Cavazzoni, D. Ceresoli, M. Cococcioni, N. Colonna, I. Carnimeo, A. D. Corso, S. de Gironcoli, P. Delugas, R. A. DiStasio, A. Ferretti, A. Floris, G. Fratesi, G. Fugallo *et al.*, *ibid.* **29**, 465901 (2017).
- [27] The calculations used a plane-wave cutoff of 120 Ry for the wave function and 480 Ry for the charge, an $8 \times 8 \times 8$ electron \mathbf{k} grid, and a Methfessel-Paxton smearing of 0.01 Ry in the integration in a self-consistent loop. Further calculations with a $32 \times 32 \times 32$ grid and tetrahedra interpolation were performed for more accurate computations of the DOS and the Fermi level, starting from a previously converged $8 \times 8 \times 8$ self-consistent electron density.
- [28] B. Johansson, *Phys. Rev. B* **20**, 1315 (1979).
- [29] P. Strange, A. Svane, W. Temmerman, Z. Szotek, and H. Winter, *Nature (London)* **399**, 756 (1999).
- [30] The pseudopotentials have been generated using the ATOMIC code of the QUANTUM ESPRESSO package [26]. The reference atomic calculation is done in the $[\text{Xe}]4f^{14}5d^{0.0}6s^{1.5}6p^{0.5}$ electronic configuration. Nonlinear core corrections have been used.
- [31] The unit cell of the high- P structure has only two variables, i.e., a and c . The volume was varied by changing a , while keeping $c/a = 1.33$ fixed at its 26 GPa value. For the α phase, a , b , and c and internal coordinates were determined by experiment at ambient pressure. We fixed the ambient-pressure values of $b/a = 0.606$, $c/a = 1.15$, and the internal atomic positions, and changed a to vary the volume. The variation of b/a and c/a is marginal in the volume range spanned by our calculations.
- [32] We used the DFT+ U formulation of Refs. [37,38] with $U = 5$ eV, $J = 0.5$ eV, and F_4/F_2 and F_6/F_2 Slater integral ratios taken from their atomic values. The U and J values are those of Ce [39], assuming an electron-hole symmetry in the f manifold. We verified that the same value of U is optimal to reproduce the EOS in Fig. 4.
- [33] At 22 GPa, the Yb-Yb out-of-plane distance in the high- P phase is 3.167 Å against an average of 3.34 Å in the α phase, while for the in-plane Yb-Yb distance we find 3.37 Å in the high- P phase versus 3.435 Å (average) in the α phase.
- [34] W. Iwasieczko, M. Drulis, and H. Drulis, *J. Alloys Compd.* **327**, 11 (2001).
- [35] E. R. Ylvisaker, J. Kuneš, A. K. McMahan, and W. E. Pickett, *Phys. Rev. Lett.* **102**, 246401 (2009).
- [36] J. Bonnet and J. Daou, *J. Appl. Phys.* **48**, 964 (1977).
- [37] V. I. Anisimov, J. Zaanen, and O. K. Andersen, *Phys. Rev. B* **44**, 943 (1991).
- [38] A. I. Liechtenstein, V. I. Anisimov, and J. Zaanen, *Phys. Rev. B* **52**, R5467(R) (1995).
- [39] F. Nilsson, R. Sakuma, and F. Aryasetiawan, *Phys. Rev. B* **88**, 125123 (2013).