Exploring High-Performance *p*-Type Transparent Conducting Oxides Based on Electron Correlation in V₂O₃ Thin Films

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Transparent conducting oxides (TCOs) with the functionality of transparency plus conductivity lie at the center of many technological applications. Commonly used TCOs are *n*-type. However, the performances of *p*-type TCOs lag far behind their *n*-type counterparts and the development of high-performance *p*-type TCOs remains an arduous challenge. The electron correlation due to on-site Coulomb interaction in the correlated metal oxides can not only promote modification of the valence band, which forms the basis for *p*-type conduction, but also shift the screened plasma energy to below the visible region for enhancing optical transparency. In this paper, the electron correlation is utilized as a key material design parameter to improve the performance of *p*-type TCOs. This approach is experimentally verified by decreasing electron correlation from Cr_2O_3 to V_2O_3 , which, despite its high hole concentration (> 10^{22} cm⁻³), has low screened plasma energy (approximately 0.97 eV) and significantly boosts the performance through good balance between the electrical conductivity and optical transparency. These results indicate that the strategy of electron correlation engineering paves an effective way for exploring high-performance *p*-type TCOs in strongly correlated oxides.

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I. INTRODUCTION

Transparent conducting oxides (TCOs), combining the contraindicated properties of high electrical conductivity and high optical transparency, are of great technological interest for a myriad of applications, notably in flat panel displays, photovoltaic cells, touchscreen sensors, and other optoelectronic devices [1-5]. To date, the commonly used TCOs are *n*-type, such as Sn-doped In_2O_3 (ITO), Al-doped ZnO (AZO), and F-doped SnO₂ (FTO), which are obtained by degenerately doping wide band-gap oxides to realize optimal balance between the electrical conductivity and optical transparency. However, the performances of p-type TCOs lag far behind their *n*-type counterparts, which has impeded many of the potential transparent electronic applications from organic and thin-film solar cell designs to the field of transparent electronics [6-10]. This discrepancy stems from the intrinsic electronic structure of metal oxides that usually possess an oxygen 2p-derived valence band (VB) and a conduction band (CB) composed of metal s orbitals. Consequently, these s orbitals are spatially spread and highly dispersed, facilitating a low (high) electron effective mass (mobility) and excellent electrical conductivity when donor doped. In contrast, the localized nature of oxygen 2p orbitals in the VB results in a large (low) hole effective mass (mobility) and poor p-type dopability, which becomes the fundamental obstacle to achieving p-type TCOs [6,11].

The strategy for reducing the localization behavior is modifying the VB through hybridization of oxygen 2p orbitals with the metal d or s orbitals. Kawazoe *et al.* proposed the concept of "chemical modulation of the valence band" (CMVB) to mitigate this conundrum, which spurs the development of a series of p-type TCOs such as Cu-based delafossite oxides and oxychalcogenides [12–23]. Following this design principle, highthroughput material screening has also selected several p-type TCOs candidates in view of low hole effective mass and wide band gap to realize high hole mobility and optical

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transparency, respectively [24–26]. However, these p-type TCOs exhibit imbalance between electrical conductivity and optical transparency, i.e., either low electrical conductivity or high electrical conductivity at the expense of optical transparency, leading to much lower performance as compared to the *n*-type TCOs [6,11]. For example, the Cu-based delafossite oxides are transparent and exhibit *p*-type conductivity; however, their electrical conductivities are usually several orders of magnitude lower than those of *n*-type TCOs arising from the deep level of acceptor defects [27]. Mg-doped CuCrO₂ exhibits a high *p*-type conductivity of 220 S cm⁻¹ concurrently with a low optical transparency of approximately 30% [15]. The other approach includes utilizing electron correlation produced by strong electron-electron Coulomb interaction U as starting points for exploring *p*-type TCOs. Several transition metal oxides (TMOs) such as NiO, Cr₂O₃, LaCrO₃, LaVO₃, Ca₃Co₄O₉, Na_xCoO₂, Bi₂Sr₂Co₂O_{ν}, and $Bi_2Sr_2CaCu_2O_v$ have also been reported as *p*-type TCOs [28-36]. Based on the relative magnitudes of U, electronic bandwidth (W), and charge-transfer energy between O 2pand d orbitals (Δ), these TMOs can be eithor Mott insulators (NiO, Cr₂O₃, LaCrO₃, and LaVO₃) or *p*-type correlated metals ($Ca_3Co_4O_9$, Na_xCoO_2 , $Bi_2Sr_2Co_2O_\nu$, and $Bi_2Sr_2CaCu_2O_{\nu}$ [37]. The common feature of their band structures to support *p*-type conduction is that the VB consists of a hybridizing band between the oxygen 2p orbital and the transition metal d orbital or transition metal dorbital driven by the electron correlation [38–41]. From the viewpoint of band structure illustrated by the Hubbard model, the electron correlation splits the d band into the upper Hubbard band (UHB) and lower Hubbard band (LHB), separated by U [37]. As a result, the electronic structure of these TMOs could be modified by the electron correlation. In this regard, the electron correlation plays the same role as CMVB for modification of the VB, which forms the basis for *p*-type conduction.

The strength of electron correlation can be quantified by a renormalization factor $Z_k = m_{\text{band}}/m^*$, where m_{band} and m^* are free noninteracting electron effective mass and renormalized electron effective mass incorporating the electron-electron interaction, respectively [42]. In the case of a Mott insulator, all itinerant carriers are localized owing to the electron correlation, which is equivalent to the divergence of m^* ($Z_k = 0$). The effect of electron correlation is determined by the U/W ratio. Decreasing the electron correlation, either by increasing W or decreasing U, would increase Z_k and decrease m^* , which would result in the increase of the carrier itinerancy and ultimately induce an insulator-to-metal transition (Mott transition) [37]. Recently, Zhang et al. have reported that correlated metals of SrVO₃ ($Z_k = 0.33$) thin films with high carrier concentrations (*n*) of approximately 10^{22} cm⁻³ can be developed as n-type TCOs [43]. The design strategy relies on the enhancement of m^* due to electron correlation to reduce the reflection edge below 1.75 eV, keeping simultaneously a high electrical conductivity (σ) of approximately 10⁴ S cm⁻¹ [43]. Stoner *et al.* used the Z_k as a key material design parameter to optimize the optoelectronic performance of SrMoO₃ ($Z_k = 0.48$) thin films as *n*-type TCOs [44].

Besides *n*-type TCOs, the electron correlation could also be used for exploring high-performance *p*-type TCOs. Cr_2O_3 ($Z_k = 0$), a binary oxide of a Mott insulator belonging to an intermediate case between a Mott-Hubbard insulator and a charge transfer insulator, has been considered as a candidate *p*-type TCO, in which the electron correlation promotes strong hybridization between Cr t_{2g} and O 2p orbitals in the VB [30,39,45,46]. However, the mixed character in the VB of Cr₂O₃ tends to exhibit the polaronic conduction of holes, resulting in low hole mobility (approximately 10^{-4} cm² V⁻¹ s⁻¹) and significantly suppress the electrical conductivity [47]. Arca et al. demonstrated that VB modification of Cr₂O₃ by acceptor doping is an effective method to improve its electrical conductivity and performance [46]. In M_2O_3 (M = 3d transition metal ions), the values of U gradually decrease with the decrease of atomic number of transition metal ions, which affects their electronic structures as well as the resultant electrical properties [48–50]. Therefore, starting from Cr_2O_3 , it is expected that the electrical conductivity can be improved by decreasing U while maintaining the consequence of VB modification due to electron correlation for *p*-type conduction to realize good balance between them. Theoretical calculations based on local density approximation with dynamical mean-field theory (LDA + DMFT) and experimental results have revealed that the U and Z_k values of V_2O_3 are about 5 eV and 0.12, which satisfy the above requirements [49–51]. As schematically shown in Fig. 1, the decrease of U from Cr_2O_3 to V_2O_3 can not only retain the VB modification for p-type conduction, but also diminish the hybridization between V t_{2g} and O 2p orbitals to increase the carrier itinerancy (mobility). V_2O_3 ($Z_k \approx$



FIG. 1. Schematic diagram of VB modification due to electron correlation for Cr_2O_3 and V_2O_3 .

0.12), neighboring to Cr₂O₃ with smaller *U*, is a prototypical Mott-Hubbard system showing a metal-to-insulator transition (MIT) at around 155 K from a high-temperature metallic phase to a low-temperature insulating phase [51]. The V 3*d* bands in V₂O₃ are split into twofold e_g and threefold t_{2g} bands due to the crystal field. The trigonal distortion of the rhombohedral lattice further degenerates the t_{2g} bands into low-lying doubly degenerate e_g^{π} and nondegenerate a_{1g} bands [52]. In the high-temperature metallic phase of V₂O₃, the VB that consists of e_g^{π} and a_{1g} bands is partially occupied by two t_{2g} electrons and the Fermi level is located within the upper VB, which accounts for the *p*-type conduction with high σ and *n* to easily evade the hole doping trouble for most *p*-type TCOs [53–55].

The reflection edge resulting from the fundamental excitation of free carriers occurs at the screened plasma energy $\hbar\omega_p = \hbar (e^2/\epsilon_0 \epsilon_r)^{1/2} (n/m^*)^{1/2}$, in which \hbar is the reduced Planck constant, e is the elemental charge, ε_0 is the vacuum permittivity, and ε_r is the relative permittivity. To open an optical transparency window in the visible spectrum (1.75–3.25 eV), the reflection edge should be below 1.75 eV and the strong interband optical transition should be above 3.25 eV. Therefore, the enhancement of m^* due to electron correlation in the correlated metals of V₂O₃ can shift the $\hbar \omega_n$ below 1.75 eV through tuning the n/m^* ratio, which satisfies one of the basic requirements for TCOs [43,44,51]. The optical absorption of V₂O₃ in the visible region is attributed to the d-d transition from occupied t_{2g} to empty t_{2g} bands, which are moderate in order to not reduce the transparency of the V_2O_3 thin films [56]. Accordingly, the strong absorption due to charge transfer excitations from O 2p to V t_{2g} bands in the ultraviolet region determines the upper limit of the transparency window (E_H) [57]. In general, the high-temperature metallic phase of V₂O₃ fulfills all the requirements for *p*-type TCOs based on the criteria of *p*-type conductivity and optical transparency. In this paper, we explore high-performance *p*-type TCOs based on electron correlation in the V_2O_3 thin films. The V_2O_3 thin films exhibit a high electrical conductivity of 400–2122 S cm^{-1} concurrently with a visible transmittance of 40%-78%. The screened plasma energy stemming from a fundamental excitation of the free carriers is approximately 0.97 eV extracted from the ellipsometry measurements. The E_H is approximately 3.6 eV determined by the strong interband transition from the occupied O 2p to the empty V t_{2g} bands. Therefore, the high-performance p-type TCOs based on the V₂O₃ thin films are obtained by the good balance between the electrical and optical properties.

II. EXPERIMENT

 V_2O_3 thin films are grown on *c*-plane Al₂O₃ substrates by pulsed laser deposition. A KrF excimer laser ($\lambda =$ 248 nm) with a repetition rate of 5 Hz and a laser fluency of 2 J cm⁻² is focused on a V₂O₅ ceramic target. The substrate temperature is kept at 600 °C during the deposition. The films are grown and cooled to room temperature in an Ar atmosphere of 2 Pa. The distance between the target and the substrate is 5 cm. In this growth condition, the growth rate of the V_2O_3 thin film is about 6 nm/min. The structural properties of the V₂O₃ films are characterized by x-ray diffraction (XRD) using K_{α} radiation at room temperature. X-ray photoelectron spectroscopy (XPS) with monochromatic Al K_{α} ($hv = 1486.6 \,\text{eV}$) is used to evaluate the valence states of the elements in an ultrahigh vacuum chamber. Binding energy data are referenced to the C 1s peak at 284.8 eV. Electrical transport properties are measured on a physics properties measurement system (PPMS) using the standard four-probe method. Hall measurements are also carried out on PPMS using the van der Pauw geometry. Optical transmission is measured by a UV/Vis/NIR Lambda 900 spectrometer in the wavelength range from 200 to 2000 nm. Room-temperature ellipsometric spectra are collected at an incidence angle of 65° using single rotating-compensator spectroscopic ellipsometers over the infrared to ultraviolet 0.93-4.77 eV.

III. RESULTS AND DISCUSSION

Figure 2(a) shows the typical XRD pattern of the V₂O₃ thin films grown on *c*-plane Al₂O₃ substrates. Only (000*l*) peaks for the V₂O₃ thin film and Al₂O₃ are observed, confirming a single phase and *c*-axis oriented growth of V₂O₃. Results for the XRD ϕ scan of V₂O₃ (1014) and Al₂O₃ (1014) reflections are shown in Fig. 1(b). Threefold symmetry is clearly observed for both the Al₂O₃ substrate and V₂O₃ thin film due to the same hexagonal structures. This indicates the epitaxial growth of the V₂O₃ thin films on the Al₂O₃ substrates and the epitaxial relation can be written as (0006) V₂O₃ || (0006) Al₂O₃ and (1000) V₂O₃ || (1000) Al₂O₃. The high quality of the V₂O₃ thin film is also confirmed by the rock curve scan with the full width at half maximum (FWHM) of 0.09° as shown in the inset of Fig. 1(a).

XPS measurement has been carried out to probe the oxidation state of the V₂O₃ thin films. Figure 3(a) displays the spectra of the V 2p and O 1s core level, in which the dotted and solid lines represent the experimental data and fitting curve, respectively. The V 2p level is split into V $2p_{3/2}$ and V $2p_{1/2}$ due to spin-orbital coupling with a separated binding energy of approximately 7.6 eV. The fitting binding energy of V $2p_{3/2}$ is approximately 515.7 eV and the line width of FWHM is approximately 3.4 eV. These values are in good agreement with the previous studies of V³⁺ in V₂O₃ [58–60]. The O 1s level can be divided into two peaks centered at 530 and 531.1 eV, corresponding to the lattice oxygen ions and the physically adsorbed oxygen at the film surface, respectively [58]. Figure 3(b) presents the XPS spectra of the V₂O₃ thin film in the VB region. It



FIG. 2. (a) XRD θ -2 θ scan of the 56-nm-thick V₂O₃ thin film on the Al₂O₃ substrate. The inset shows the rocking curve scan of the (0006) diffraction peak of a V₂O₃ thin film. (b) XRD ϕ scans of V₂O₃ (1014) and Al₂O₃ (1014) reflections.

is found that the spectrum can be assigned to the V 3*d* band ranging from 0 to 2.5 eV and the O 2*p* band in the range of 2.5–9 eV, respectively [53,54,59]. Further analysis indicates that the V 3*d* band consists of A (a_{1g}) and B (e_g^{π}) bands and the O 2*p* band can be fitted with the C, D, and E peaks, which are consistent with the previous theoretical calculation and experimental results of V₂O₃ [50,54,61]. As shown in the inset of Fig. 3(b), the Fermi level *E_F* lies within the VB formed by the overlapping V (3*d*) a_{1g} and e_g^{π} bands. Accordingly, the V₂O₃ clearly exhibits a finite spectral weight at *E_F*, which coincides with the metallic behavior at room temperature.

Figure 4(a) shows the temperature dependence of sheet resistance ($R_S = 1/\sigma t$, t film thickness) for the V₂O₃ thin films with different thicknesses. The V₂O₃ thin films with thicknesses of 56 and 28 nm exhibit metallic behaviors, in which the decrease of the R_S with decreasing temperature down to approximately 190 K is followed by a weakly insulating behavior. The obvious thermal hysteresis between the cooling and warming processes indicates the first-order nature of the MIT. Nevertheless, the MIT and its accompanied resistance change are broadened and much smaller compared to that of the single crystal, while the resistivity in the high-temperature metallic state is similar with the single crystal [62]. In contrast, the V₂O₃ thin films with thicknesses of 14 and 7 nm show insulator-insulator transitions, in which the R_S increases



FIG. 3. (a) XPS spectrum of the V 2p and O 1s core levels of the V₂O₃ thin film. (b) XPS spectra of the V₂O₃ thin film in the valence band region. The inset shows the energy level diagram of the V₂O₃ thin film in the high-temperature metallic phase.

slowly with decreasing temperature, followed by a rapid increase of approximately 3 orders of magnitude. Previous studies have revealed the coexistence of metallic and insulating phases in the V_2O_3 thin films associated with the thermodynamic instability near the transition [63,64].



FIG. 4. (a) Temperature-dependent sheet resistance of the V_2O_3 thin films. (b) Evolution of room-temperature Hall resistance with magnetic field. (c) Thickness dependence of room-temperature hole concentration and hole mobility for the V_2O_3 thin films.

Furthermore, this instability can be affected by many factors such as strain, substrate disorder, and substrate orientation, which may account for the diverse transport properties observed in the V₂O₃ thin films [65–68]. The $R_S(\sigma)$ at room temperature is 3569 (400), 1028 (695), 209 (1704), and 84 (2122) Ω/sq (S cm⁻¹) for the 7-, 14-, 28-, and 56-nm V_2O_3 thin films, respectively. Figure 3(b) depicts the typical room-temperature Hall resistance versus magnetic field plots for the V₂O₃ thin films. The positive Hall coefficients confirm the p-type carriers in the V₂O₃ thin films. Figure 3(c) shows the derived hole concentration (n_h) and mobility (μ) for the V₂O₃ thin films. The values of n_h (μ) are in the range of 1.77–5.28 × 10²² cm⁻³ $(0.14-0.24 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1})$, which is consistent with that in single crystal and thin films [62,69]. The values of n_h in the V₂O₃ thin films are the highest among *p*-type TCOs [6,11]. For the V₂O₃ ($Z_k \approx 0.12$), the localized V 3d orbital around the VB maximum results in an effective hole mass of $m^* = m_{\text{band}}/Z_k \approx 8.3 m_{\text{band}}$, which would be larger than that of conventional TCOs [24]. The mobility (μ) is defined as $\mu = e\tau/m^*$, with τ the scattering time. As a result, the values of μ would be suppressed by the enhancement of m^* due to strong electron correlation. However, the low μ can be compensated by the high n_h , resulting in high electrical conductivity.

Figure 5 shows optical transmittance of the V₂O₃ thin films on Al₂O₃ substrates from infrared to ultraviolet regions. The optical transmittance in the visible region (*T*) is usually quantified by averaging values at wavelengths of 420, 490, 560, 630, and 700 nm including the substrate contribution. The values of *T* are 78.0%, 72.2%, 54.8%, and 40.0% for the 7, 14, 28, and 56 nm of V₂O₃ thin films,



FIG. 5. (a) Optical transmittance spectra for the Al_2O_3 substrate and V_2O_3 thin films with different thicknesses. The range of the visible spectrum is illustrated. The inset shows the photographs of the Al_2O_3 substrate and V_2O_3 thin films with different thicknesses on the V_2O_3 substrates.

respectively. The inset of Fig. 5 exhibits the photograph of the V_2O_3 thin films on a labeled paper. The 7-nm V_2O_3 thin film is almost indistinguishable from the Al_2O_3 substrate, reflecting the high optical transparency. The V_2O_3 thin films become less transparent with increasing film thickness due to the unavoidable optical absorption in the visible region.

To gain a deeper insight into the optical properties for the V_2O_3 thin film, the ellipsometry measurements are performed on the V_2O_3 thin film (t = 130 nm) on a single-side polished Al₂O₃ substrate at room temperature. Figure 6 shows the complex dielectric constant ($\epsilon =$ $\epsilon_1 + i\epsilon_2$) spectra of the V₂O₃ thin film extracted from the room-temperature ellipsometry measurements. The dielectric response of the V2O3 thin film is modeled using combinations of Drude, Lorentz, and Tauc-Lorentz oscillators (see the Supplemental Material [70]). The real part of the dielectric function (ϵ_1) equals zero at the screened plasma energy ($\hbar\omega_p$). Therefore, the $\hbar\omega_p$ for the V₂O₃ thin film can be determined to be approximately 0.97 eV by setting $\epsilon_1 = 0$, which is consistent with the plasma edge of approximately 1 eV in the V_2O_3 single crystal [71]. This value is well below 1.75 eV required for realizing visible optical transparency. The remarkably reduced $\hbar \omega_p$ of the V_2O_3 thin film in the metallic phase with a high carrier concentration is a direct consequence of the enhancement of m^* due to strong electron correlation, which is similar to that in the correlated metals of SrVO₃ and SrMoO₃ [43,44]. Furthermore, the values of ϵ_2 intimately tied to the optical absorption in the visible region are relatively moderate, which is also important for the realization of high optical transparency in thin-film materials to reduce absorption. The ϵ_2 increases in the infrared region due to the free-carrier excitation.



FIG. 6. Complex dielectric function ($\epsilon = \epsilon_1 + i\epsilon_2$) spectra of the V₂O₃ thin film (t = 130 nm) extracted from room-temperature spectroscopic ellipsometry measurements.



FIG. 7. Optical absorption coefficients for the V₂O₃ thin film derived from transmission (t = 56 nm) and ellipsometry measurements (t = 130 nm), respectively. The inset shows the Tauc plot to estimate E_H .

Another important parameter for the optical transparency window in the visible spectrum is the upper limit of the transparency window (E_H) , above which the material is obscure for the incident photon [57]. Usually, the E_H is determined by the strong optical absorption due to the direct interband transition. Figure 7 shows the experimental absorption spectra for the V_2O_3 thin film derived from transmission and ellipsometry measurements, respectively. The absorption coefficient (α) spectra can be calculated using the equation of $\alpha = 4\pi\kappa/\lambda$, where λ is the wavelength and κ is the extinction coefficient. The relationship between κ and the dielectric constant is $\sqrt{\epsilon_1^2 + \epsilon_2^2 - \epsilon_1}$ /2. It is clear that the optical $\kappa =$ absorption features of the V₂O₃ thin film can be well extracted from both the transmission and ellipsometry spectra. The low-intensity spectral features observed in the 1.0–3.5 eV range can be assigned as the t_{2g} - t_{2g} interband transitions [56,72]. A strong absorption edge is observed at the photon energy of approximately 4.0 eV due to the inter-

band transition from the O 2*p* to the empty V t_{2g} bands. E_H can be determined by fitting the absorption edge through the Tauc relation, $(\alpha hv)^2 = C(hv - E_H)$, where hv is the photon energy and C is a constant. As shown in the inset of Fig. 7, the linear fitting yields an E_H of approximately 3.6 eV, which is well above 3.25 eV for visible transparency.

Figure 8 illustrates T and σ for the V₂O₃ thin films (red semisolid stars) and other typical *p*-type TCOs. The σ of *p*-type TCOs are distributed over a wide range of several orders of magnitude. It is well known that the practical applications of the *p*-type TCOs rely on simultaneously high T and σ . Clearly, the V₂O₃ thin film becomes a front



FIG. 8. Graphical representation of T and σ for the V₂O₃ thin films (red semisolid stars) and other typical *p*-type TCOs. The numbers in parentheses show the corresponding values of F^H and F^G in the units of $M\Omega^{-1}$. The *n*-type TCO of ITO thin film is also added for comparison. The blue arrow indicates the enhancement of performance through electron correlation engineering.

runner in the pursuit of high-performance *p*-type TCOs due to the good balance between electrical and optical properties. It is convenient to quantitatively evaluate the performance of *p*-type TCOs by a figure of merit (FOM). Two types of FOM are often calculated by $F^H = T^{10}/R_S$ and $F^G = -1/(R_S \ln T)$, which were proposed by Haacke and Gordon, respectively [73,74]. The larger values of FOM indicate higher performances of the TCOs. However, F^H overestimates the contribution of R_S , while F^G overestimates the importance of T. Therefore, both F^H and F^G should be considered simultaneously to yield a fair evaluation of the performance of *p*-type TCOs. The numbers in parentheses show the corresponding values of F^H and F^G in the units of M Ω^{-1} for the V₂O₃ thin films and other reported *p*-type TCOs (also in Table S2 in the Supplemental Material [70]). In particular, it should be noted that the performance of *p*-type TCOs can be remarkably enhanced in the M₂O₃ compounds from Cr₂O₃ to V_2O_3 . The decrease of U would be the main driving force. This result indicates that the strategy of electron correlation engineering paves an effective way for exploring high-performance *p*-type TCOs in strongly correlated oxides.

IV. CONCLUSION

In summary, we demonstrate that the correlated metallic phase of V_2O_3 thin films can be explored as *p*-type TCOs with high performance. The electron correlated-driven VB engineering in V_2O_3 forms the basis for *p*-type conduction. The intrinsic metallic phase of V_2O_3 with high carrier concentration (> 10^{22} cm⁻³) guarantees its high electrical conductivity. Combining low screened plasma energy (approximately 0.97 eV) and moderate absorption in the visible region, the V₂O₃ thin films exhibit the visible transmittances of 40%–78%. These results indicate that the V₂O₃ thin films can be developed as a high-performance *p*-type TCO and provide an alternative route to explore high-performance *p*-type TCOs in strongly correlated oxides. Furthermore, the V₂O₃ thin films as highperformance *p*-type TCOs can be potentially used in several transparent device applications such as rectifiers, photodetectors, and solar cells [6,75].

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- K. Ellmer, Past achievements and future challenges in the development of optically transparent electrodes, Nat. Photonics 6, 809 (2012).
- [2] C. G. Granqvist, Transparent conductors as solar energy materials: A panoramic review, Sol. Energy Mater. Sol. Cells 91, 1529 (2007).
- [3] E. Fortunato, D. Ginley, H. Hosono, and D. C. Paine, Transparent conducting oxides for photovoltaics, MRS Bull. 32, 242 (2007).
- [4] E. Fortunato, P. Barquinha, and R. Martins, Oxide semiconductor thin-film transistors: A review of recent advances, Adv. Mater. 24, 2945 (2012).
- [5] T. Minami, Transparent conducting oxide semiconductors for transparent electrodes, Semicond. Sci. Technol. 20, S35 (2005).
- [6] Z. Wang, P. K. Nayak, J. A. Caraveo-Frescas, and H. N. Alshareef, Recent developments in *p*-type oxide semiconductor materials and devices, Adv. Mater. 28, 3831 (2016).
- [7] W. Beyer, J. Hupkes, and H. Stiebig, Transparent conducting oxide films for thin film silicon photovoltaics, Thin Solid Films 516, 147 (2007).
- [8] H. Ohta and H. Hosono, Transparent oxide optoelectronics, Mater. Today 3, 42 (2004).
- [9] H. Kawazoe, H. Yanagi, and K. Ueda, Transparent *p*-type conducting oxides: Design and fabrication of *p*-*n* heterojunctions, MRS Bull. 25, 28 (2000).
- [10] A. Banerjee and K. Chattopadhyay, Recent developments in the emerging field of crystalline *p*-type transparent conducting oxide thin films, Prog. Cryst. Growth Charact. Mater. 50, 52 (2005).

- [11] K. H. L. Zhang, K. Xi, M. Blamire, and R. G. Egdell, *P*-type transparent conducting oxides, J. Phys.: Condens. Matter 28, 383002 (2016).
- [12] H. Kawazoe, M. Yasukawa, H. Hyodo, M. Kurita, H. Yanagi, and H. Hosono, *P*-type electrical conduction in transparent thin films of CuAlO₂, Nature **389**, 939 (1997).
- [13] H. Yanagi, S. Inoue, K. Ueda, H. Kawazoe, H. Hosono, and N. Hamada, Electronic structure and optoelectronic properties of transparent *p*-type conducting CuAlO₂, J. Appl. Phys. 88, 4159 (2000).
- [14] K. Ueda, T. Hase, H. Yanagi, H. Kawazoe, H. Hosono, H. Ohta, M. Orita, and M. Hirano, Epitaxial growth of transparent *p*-type conducting CuGaO₂ thin films on sapphire (001) substrates by pulsed laser deposition, J. Appl. Phys. **89**, 1790 (2001).
- [15] R. Nagarajan, A. D. Draeseke, A. W. Sleight, and J. Tate, *p*-type conductivity in CuCr_{1-x}Mg_xO₂ films and powders, J. Appl. Phys. **89**, 8022 (2001).
- [16] N. Duan, A. W. Sleight, M. K. Jayaraj, and J. Tate, Transparent *p*-type conducting $CuScO_{2+x}$ films, Appl. Phys. Lett. **77**, 1325 (2000).
- [17] M. Marquardt, N. Ashmore, and D. P. Cann, Crystal chemistry and electrical properties of the delafossite structure, Thin Solid Films 496, 146 (2006).
- [18] A. Kudo, H. Yanagi, H. Hosono, and H. Kawazoe, SrCu₂O₂: A *p*-type conductive oxide with wide band gap, Appl. Phys. Lett. **73**, 220 (1998).
- [19] E. Fortunato, R. Barros, P. Barquinha, V. Figueiredo, S. K. Park, C. Hwang, and R. Martins, Transparent p-type SnO_x thin film transistors produced by reactive rf magnetron sputtering followed by low temperature annealing, Appl. Phys. Lett. **97**, 052105 (2010).
- [20] M. Dekkers, G. Rijnders, and D. H. A. Blank, ZnIr₂O₄, a *p*-type transparent oxide semiconductor in the class of spinel zinc-d⁶-transition metal oxide, Appl. Phys. Lett. **90**, 021903 (2007).
- [21] R. H. Wei, X. W. Tang, L. Hu, Z. Z. Hui, J. Yang, H. M. Luo, X. Luo, J. M. Dai, W. H. Song, Z. R. Yang, X. B. Zhu, and Y. P. Sun, Facile chemical solution synthesis of *p*-type delafossite Ag-based transparent conducting AgCrO₂ films in an open condition, J. Mater. Chem. C 5, 1885 (2017).
- [22] H. Hiramatsu, K. Ueda, H. Ohta, M. Hirano, M. Kikuchi, H. Yanagi, T. Kamiya, and H. Hosono, Heavy hole doping of epitaxial thin films of a wide gap *p*-type semiconductor, LaCuOSe, and analysis of the effective mass, Appl. Phys. Lett. **91**, 012104 (2007).
- [23] H. Hiramatsu, K. Ueda, H. Ohta, M. Orita, M. Hirano, and H. Hosono, Preparation of transparent *p*-type $(La_{1-x}Sr_xO)CuS$ thin films by r.f. sputtering technique, Thin Solid Films **411**, 125 (2002).
- [24] G. Hautier, A. Miglio, G. Ceder, G. Rignanese, and X. Gonze, Identification and design principles of low hole effective mass *p*-type transparent conducting oxides, Nat. Commun. 4, 2292 (2013).
- [25] N. Sarmadian, R. Saniz, B. Partoens, and D. Lamoen, Easily doped *p*-type, low hole effective mass, transparent oxides, Sci. Rep. 6, 20446 (2016).
- [26] A. Bhatia, G. Hautier, T. Nilgianskul, A. Miglio, J. Sun, H. J. Kim, K. H. Kim, S. Chen, G. Rignanese, X. Gonze, and J. Suntivich, High-mobility bismuth-based transparent

p-type oxide from high-throughput material screening, Chem. Mater. **28**, 30 (2016).

- [27] J. Tate, H. L. Ju, J. C. Moon, A. Zakutayev, A. P. Richard, J. Russell, and D. H. McIntyre, Origin of *p*-type conduction in single-crystal CuAlO₂, Phys. Rev. B 80, 165206 (2009).
- [28] H. Sato, T. Minami, S. Takata, and T. Yamada, Transparent conducting *p*-type NiO thin films prepared by magnetron sputtering, Thin Solid Films 236, 27 (1993).
- [29] P. Zhai, Q. Yi, J. Jian, H. Wang, P. Song, C. Dong, X. Lu, Y. Sun, J. Zhao, X. Dai, Y. Lou, H. Yang, and G. Zou, Transparent *p*-type epitaxial thin films of nickel oxide, Chem. Commun. **50**, 1854 (2014).
- [30] E. Arca, K. Fleischer, and I. V. Shvets, Magnesium, nitrogen codoped Cr₂O₃: A *p*-type transparent conducting oxide, Appl. Phys. Lett. **99**, 111910 (2011).
- [31] K. H. L. Zhang, Y. Du, A. Papadogianni, O. Bierwagen, S. Sallis, L. F. J. Piper, M. E. Bowden, V. Shutthanandan, P. V. Sushko, and S. A. Chambers, Perovskite Sr-doped LaCrO₃ as a new *p*-type transparent conducting oxide, Adv. Mater. 27, 5191 (2015).
- [32] L. Hu, R. H. Wei, J. Yan, D. Wang, X. W. Tang, X. Luo, W. H. Song, J. M. Dai, X. B. Zhu, C. J. Zhang, and Y. P. Sun, La_{2/3}Sr_{1/3}VO₃ thin films: A new *p*-type transparent conducting oxide with very high figure of merit, Adv. Electron. Mater. 4, 1700476 (2018).
- [33] M. Aksit, S. K. Kolli, I. M. Slauch, and R. D. Robinson, Misfit layered Ca₃Co₄O₉ as a high figure of merit *p*-type transparent conducting oxide film through solution processing, Appl. Phys. Lett. **104**, 161901 (2014).
- [34] D. C. Yuan, J. L. Wang, N. Fu, X. L. Wu, Y. J. Ma, and S. F. Wang, Transparent conducting properties of *c*-axisoriented Na_xCoO₂ epitaxial thin films, Sci. China-Phys. Mech. Astron. **61**, 107321 (2018).
- [35] R. H. Wei, X. W. Tang, L. Hu, Z. Z Hui, J. Yang, H. M. Luo, X. Luo, J. M. Dai, w. H. Song, Z. R. Yang, X. B. Zhu, and Y. P. Sun, Transparent conducting *p*-type thin films of *c*-axis self-oriented Bi₂Sr₂Co₂O_y with high figure of merit, Chem. Commun. **50**, 9697 (2014).
- [36] R. H. Wei, L. Zhang, L. Hu, X. W. Tang, J. Yang, J. M. Dai, W. H. Song, X. B. Zhu, and Y. P. Sun, *p*-type transparent conductivity in high temperature superconducting Bi-2212 thin films, Appl. Phys. Lett. **112**, 251109 (2018).
- [37] M. Imada, A. Fujimori, and Y. Tokura, Metal-insulator transitions, Rev. Mod. Phys. 70, 1039 (1998).
- [38] J. Y. Zhang, W. W. Li, R. L. Z. Hoye, J. L. MacManus-Driscoll, M. Budde, O. Bierwagen, L. Wang, Y. Du, M. J. Wahila, L. F. J. Piper, T.-L. Lee, H. J. Edwards, V. R. Dhanak, and K. H. L. Zhang, Electronic and transport properties of Li-doped NiO epitaxial thin films, J. Mater. Chem. C 6, 2275 (2018).
- [39] A. B. Kehoe, E. Arca, D. O. Scanlon, I. V. Shvets, and G. W. Watson, Assessing the potential of Mg-doped Cr₂O₃ as a novel *p*-type transparent conducting oxide, J. Phys.: Condens. Matter 28, 125501 (2016).
- [40] Z. Q. Yang, Z. Huang, L. Ye, and X. D. Xie, Influence of parameters U and J in the LSDA + U method on electronic structure of the perovskites LaMO₃ (*M* = Cr, Mn, Fe, Co, Ni), Phys. Rev. B 60, 15674 (1999).
- [41] T. Takeuchi, T. Kondo, T. Takami, H. Takahashi, H. Ikuta, U. Mizutani, K. Soda, R. Funahashi, M. Shikano, M.

Mikami, S. Tsuda, T. Yokoya, S. Shin, and T. Muro, Contribution of electronic structure to the large thermoelectric power in layered cobalt oxides, Phys. Rev. B **69**, 125410 (2004).

- [42] D. N. Basov, R. D. Averitt, D. van der Marel, M. Dressel, and K. Haule, Electrodynamics of correlated electron materials, Rev. Mod. Phys. 83, 471 (2011).
- [43] L. Zhang, Y. Zhou, L. Guo, W. Zhao, A. Barnes, H.-T. Zhang, C. Eaton, Y. Zheng, M. Brahlek, H. F. Haneef, N. J. Podraza, M. H. W. Chan, V. Gopalan, K. M. Rabe, and R. Engel-Herbert, Correlated metals as transparent conductors, Nat. Mater. 15, 204 (2016).
- [44] J. L. Stoner, P. A. E. Murgatroyd, M. O'Sullivan, M. S. Dyer, T. D. Manning, J. B. Claridge, M. J. Rosseinsky, and J. Alaria, Chemical control of correlated metals as transparent conductors, Adv. Funct. Mater. 29, 1808609 (2019).
- [45] F. Lebreau, M. M. Islam, B. Diawara, and P. Marcus, Structural, magnetic, electronic, defect, and diffusion properties of Cr₂O₃: A DFT + U study, J. Phys. Chem. C 118, 18133 (2014).
- [46] E. Arca, A. B. Kehoe, T. D. Veal, A. Shmeliov, D. O. Scanlon, C. Downing, D. Daly, D. Mullarkey, I. V. Shvets, V. Nicolosi, and G. W. Watson, Valence band modification of Cr₂O₃ by Ni-doping: Creating a high figure of merit *p*-type TCO, J. Mater. Chem. C 5, 12610 (2017).
- [47] L. Farrell, K. Fleischer, D. Caffrey, D. Mullarkey, E. Norton, and I. V. Shvets, Conducting mechanism in the epitaxial *p*-type transparent conducting oxide Cr_2O_3 : Mg, Phys. Rev. B **91**, 125202 (2015).
- [48] T. Uozumi, K. Okada, A. Kotani, R. Zimmermann, P. Steiner, S. Hufner, Y. Tezuka, and S. Shin, Theoretical and experimental studies on the electronic structure of M_2O_3 (M = Ti, V, Cr, Mn, Fe) compounds by systematic analysis of high-energy spectroscopy, J. Electron. Spectrosc. Relat. Phenom. **83**, 9 (1997).
- [49] K. Held, G. Keller, V. Eyert, D. Vollhardt, and V. I. Anisimov, Mott-Hubbard Metal-insulator Transition in Paramagnetic V_2O_3 : An LDA + DMFT(QMC) Study, Phys. Rev. Lett. **86**, 5345 (2001).
- [50] G. Keller, K. Held, V. Eyert, D. Vollhardt, and V. I. Anisimov, Electronic structure of paramagnetic V₂O₃: Strongly correlated metallic and Mott insulating phase, Phys. Rev. B 70, 205116 (2004).
- [51] M. K. Stewart, D. Brownstead, S. Wang, K. G. West, J. G. Ramirez, M. M. Qazilbash, N. B. Perkins, I. K. Schuller, and D. N. Basov, Insulator-to-metal transition and correlated metallic state of V₂O₃ investigated by optical spectroscopy, Phys. Rev. B 85, 205113 (2012).
- [52] F. Rodolakis, J.-P. Rueff, M. Sikora, I. Alliot, J.-P. Itie, F. Baudelet, S. Ravy, P. Wzietek, P. Hansmann, A. Toschi, M. W. Haverkort, G. Sangiovanni, K. Held, P. Metcalf, and M. Marsi, Evolution of the electronic structure of a Mott system across its phase diagram: X-ray absorption spectroscopy study of (V_{1-x}Cr_x)₂O₃, Phys. Rev. B 84, 245113 (2011).
- [53] E. Papalazarou, M. Gatti, M. Marsi, V. Brouet, F. Iori, L. Reining, E. Annese, I. Vobornik, F. Offi, A. Fondacaro, S. Huotari, P. Lacovig, O. Tjernberg, N. B. Brookes, M. Sacchi, P. Metcalf, and G. Panaccione, Valence-band electronic

structure of V_2O_3 : Identification of V and O bands, Phys. Rev. B **80**, 155115 (2009).

- [54] S. S. Majid, D. K. Shukla, F. Rahman, K. Gautam, R. J. Choudhary, V. G. Sathe, and D. M. Phase, Stabilization of metallic phase in V_2O_3 thin film, Appl. Phys. Lett. **110**, 173101 (2017).
- [55] I. Lo Vecchio, L. Baldassarre, F. Dapuzzo, O. Limaj, D. Nicoletti, A. Perucchi, L. Fan, P. Metcalf, M. Marsi, and S. Lupi, Optical properties of V₂O₃ in its whole phase diagram, Phys. Rev. B **91**, 155133 (2015).
- [56] S. E. Chamberlin, I. H. Nayyar, T. C. Kaspar, P. V. Sushko, and S. A. Chambers, Electronic structure and optical properties of α -(Fe_{1-x}V_x)₂O₃ solid-solution thin films, Appl. Phys. Lett. **106**, 041905 (2015).
- [57] K. R. Eppelmeier and J. M. Rondinelli, Correlated oxides: Metals amassing transparency, Nat. Mater. 15, 132 (2016).
- [58] S. Autier-Laurent, B. Mercey, D. Chippaux, P. Limelette, and Ch. Simon, Strain-induced pressure effect in pulsed laser deposited thin films of the strongly correlated oxide V_2O_3 , Phys. Rev. B 74, 195109 (2006).
- [59] E. Hryha, E. Rutqvist, and L. Nyborg, Stoichiometric vanadium oxides studied by XPS, Surf. Interface Anal. 44, 1022 (2012).
- [60] X. B. Liu, H. B. Lu, M. He, K. J. Jin, and G. Z. Yang, Room-temperature epitaxial growth of V₂O₃ films, Sci. China-Phys. Mech. Astron. 57, 1866 (2014).
- [61] R. L. Kurtz and V. E. Henrich, Surface electronic structure and chemisorption on corundum transition-metal oxides: V_2O_3 , Phys. Rev. B **28**, 6699 (1983).
- [62] D. B. McWhan and J. P. Remeika, Metal-insulator transition in $(V_{1-x}Cr_x)_2O_3$, Phys. Rev. B **2**, 3734 (1970).
- [63] S. Lupi *et al.*, A microscopic view on the Mott transition in chromium-doped V₂O₃, Nat. Commun. 1, 105 (2010).
- [64] A. S. McLeod, E. van Heumen, J. G. Ramirez, S. Wang, T. Saerbeck, S. Guenon, M. Goldflam, L. Anderegg, P. Kelly, A. Mueller, M. K. Liu, K. Schuller Ivan, and D. N. Basov, Nanotextured phase coexistence in the correlated insulator V₂O₃, Nat. Phys. **13**, 80 (2016).

- [65] Q. Luo, Q. Guo, and E. G. Wang, Thickness-dependent metal-insulator transition in V₂O₃ ultrathin films, Appl. Phys. Lett. 84, 2337 (2004).
- [66] B. S. Allimi, M. Aindow, and S. P. Alpay, Thickness dependence of electronic phase transitions in epitaxial V₂O₃ films on (0001) LiTaO₃, Appl. Phys. Lett. **93**, 112109 (2008).
- [67] J. Brockman, M. G. Samant, K. P. Roche, and S. S. P. Parkin, Substrate-induced disorder in V₂O₃ thin films grown on annealed *c*-plane sapphire substrates, Appl. Phys. Lett. **101**, 051606 (2012).
- [68] J. Sakai, P. Limelette, and H. Funakubo, Transport properties and c/a ratio of V₂O₃ thin films grown on C- and R-plane sapphire substrates by pulsed laser deposition, Appl. Phys. Lett. **107**, 241901 (2015).
- [69] C. Grygiel, Ch. Simon, B. Mercey, W. Prellier, R. Fresard, and P. Limelette, Thickness dependence of the electronic properties in V₂O₃ thin films, Appl. Phys. Lett. **91**, 262103 (2007).
- [70] See the Supplemental Material at http://link.aps.org/supple mental/10.1103/PhysRevApplied.12.044035 for the spectroscopic ellipsometry measurement and modeling of the V₂O₃ thin film. Also, the performances of reported *p*-type TCOs in the literature and the V₂O₃ thin films are given.
- [71] A. S. Barker and J. P. Remeika, Optical properties of V_2O_3 doped with chromium, Solid State Commun. **8**, 1521 (1970).
- [72] M. M. Qazilbash, A. A. Schafgans, K. S. Burch, S. J. Yun, B. G. Chae, B. J. Kim, H. T. Kim, and D. N. Basov, Electrodynamics of the vanadium oxides VO₂ and V₂O₃, Phys. Rev. B 77, 115121 (2008).
- [73] C. Haacke, New figure of merit for transparent conductors, J. Appl. Phys. 47, 4086 (1976).
- [74] R. G. Gordon, Criteria for choosing transparent conductors, MRS Bull. 25, 52 (2000).
- [75] C. Chen, T. Wei, P. Hsiao, and C. Hung, Vanadium oxide as transparent carrier-selective layer in silicon hybrid solar cells promoting photovoltaic performances, ACS Appl. Energy Mater. 2, 4873 (2019).