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# Active Peltier Coolers Based on Correlated and Magnon-Drag Metals

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Active cooling systems use electrical work to bring a hot device (such as an integrated circuit) down to near ambient temperature by draining heat from the hot area of the device to a passive heat sink. Commercial thermoelectric modules are optimized for refrigeration, and are not ideal for active cooling. Refrigeration maintains a temperature that is below the ambient temperature in a device (such as a kitchen refrigerator) by pumping heat from the cold area of the device to a heat sink. The thermoelectric figure of merit zT is used traditionally to evaluate the performance of thermoelectric modules, including refrigeration modules. But it is not a good indicator of the performance of active cooling materials. Here, we describe an efficient, all-solid-state active cooler based on the Peltier effect in metals with high thermoelectric power factor due to electron correlation effects (CePd<sub>3</sub>) or magnon drag (Co) and high passive thermal conductivity. We show theoretically and experimentally that the effective thermal conductance under applied current can exceed the limits imposed by Fourier's heat conduction law. The designed device measures an effective thermal conductance that is an order of magnitude larger than the passive thermal conductance at  $\Delta T = 1$  K with the dynamic response of 4 s.

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### I. INTRODUCTION

As batteries, lasers, high-power electronics, computer chips, and spacecraft evolve toward more compact structures with higher power densities, they require more sophisticated thermal management technology. For example, lithium-ion batteries require careful temperature control, since variations in temperature affect their performance and life cycle [1]. Similarly, heat management is the limiting step in the further miniaturization of integrated circuits, many of which have time-dependent heat loads [2–4].

Solid-state compact coolers without moving parts are of high interest in thermal management [5–7]. They are particularly suitable for cooling problems where the thermal load is transient. In such problems, it is useful to have a passive heat sink to handle the steady-state operation, supplemented by an active cooling system that handles peak loads (e.g., during battery charging). Thus, passive and active components typically are combined and both must be optimized simultaneously. An active cooling system [see Fig. 1(a)] uses electrical work to dissipate heat from the hot area in a device that must be cooled to ambient temperature (e.g., a battery or an electronic circuit) to a reservoir at or near ambient temperature that acts as a passive heat sink. The purpose of active cooling is to maximize the heat drained from the hot reservoir while minimizing the temperature drop,  $\Delta T$ , across the cooler. In an all solid-state device, this amounts to maximizing the effective thermal conductance of the active cooling system, defined as the ratio of the heat flow out of the hot reservoir to  $\Delta T$ . Normalizing this conductance for geometry, as one does in classical heat-transfer problems through passive thermal conductors, one can define an effective thermal conductivity  $\kappa_{\text{eff}}$  for the materials in such an active cooler:  $\kappa_{\rm eff}$  is the ratio of the heat flux they can carry in active mode to the temperature gradient. For clarity, we note that  $\kappa_{\rm eff}$  is not an intrinsic material property. Indeed, we will show that  $\kappa_{\rm eff}$  depends on  $\Delta T$  as well as on the intrinsic material properties  $\kappa$  and the thermoelectric power factor  $PF = \sigma \alpha^2$ , where  $\sigma$  is the electrical conductivity and  $\alpha$ the thermoelectric power of the material under consideration. Here, we reserve the words thermal conductivity  $\kappa$ 

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or passive thermal conductivity for the passive heat flux in Fourier's law.

This article describes an all-solid-state approach to achieve this that is based on the Peltier effect in metals with high thermopower due to electron correlation effects (CePd<sub>3</sub>) or magnon drag (Co). We show the effective thermal conductance can exceed the limits imposed by Fourier's heat conduction law. We also show how this approach is fundamentally different from Peltier cooling based on conventional thermoelectric (TE) materials such as the (Bi, Sb)<sub>2</sub>(Te, Se)<sub>3</sub> alloys used in commercial Peltier modules.

A TE module has three modes of operations: (1) powergeneration mode where heat is supplied and electric work is generated; (2) refrigeration mode [Fig. 1(a)] where electric work is used to pump heat from the area of the device to be maintained below ambient temperatures to a heat sink at or near ambient temperature; and (3) active cooling mode [Fig. 1(a)] where electric work is used to drain heat from the area of the device to be maintained at or near ambient temperature to a passive heat sink at or near ambient temperature. Commercial Peltier modules are designed for refrigeration applications, including climate-controlled automotive seats and small-scale refrigeration of various sensors, or to reduce noise and improve performances of electronic and optoelectronic devices [6,8,9]. For the first two modes of operations, materials with a large TE figure of merit, zT, are needed, in particular, materials with low thermal conductivity. In the power-generation mode, low thermal conductivity is needed to maintain the temperature difference. In the refrigeration mode, the idea is to pump heat from cold to hot using electrons. A low lattice thermal conductivity is needed to block the back flux of heat via phonons from hot to cold.

However, for active cooling, material design is different. Traditional high-zT materials are not useful, and instead materials with a simultaneously large TE power factor and passive thermal conductivity are needed to remove heat from a hot area of the device quickly and effectively. Previous studies have already shown that while conventional Peltier TE modules are able to drive a large heat flux [10], their effective thermal conductivity in active mode is too small to compete even with the thermal conductivity of metals used in passive heat exchangers [11]. Thus, classical Peltier modules are most useful only in refrigeration mode where the device must be cooled below the ambient temperature, and thus, the temperature of the passive heat sink. In this study, we first point out the differences between the equations that govern active cooling and those that govern refrigeration. Then, we show how these equations lead to different material selection criteria. Finally, we study TE metals experimentally to investigate how to exceed the thermal conductivity of metals in passive heat sinks.



FIG. 1. Schematic drawings show (a) the difference between active cooling and refrigeration, and (b)–(d) the use of Peltier couples in various modes of heat conduction: (b) passive mode, (c) refrigeration mode, (d) active cooling mode.

## **II. THEORY**

Consider a thermocouple made out of a *p* and an *n* leg, placed between a hot source and a cold sink. Assume temperature-independent materials properties,  $\alpha_{p(n)}$ ,  $\sigma_{p(n)}$ , and  $\kappa_{p(n)}$  to be the Seebeck coefficient, the electrical conductivity, and the thermal conductivity of each leg. Ignoring the contact resistances, the thermal conductance of the TE couple is  $K = (\kappa_p A_p / L_p) + (\kappa_n A_n / L_n)$ , the electrical resistance is  $R = (L_p / \sigma_p A_p) + (L_n / \sigma_n A_n)$ , and the Seebeck coefficient of the TE couple is  $\alpha = \alpha_p - \alpha_n$ .

In the passive state, shown in Fig. 1(b), the rate of the heat conduction at the hot side as well as at the cold side of the thermocouple is given by the Fourier's law as

$$Q_{\rm off} = K(T_H - T_C), \tag{1}$$

where  $T_H$  is the hot source temperature and  $T_C$  is the cold sink temperature.

The device is switched on by driving an electric current, I, through the legs, as shown in Figs. 1(c) and 1(d). Here, there are two possibilities. Figure 1(c) is the refrigeration cycle where heat is pumped from the cold side to the hot side. In this case, the heat rate extracted from the cold side is

$$Q_{\rm on - refrigration} = \alpha T_C I - K(T_H - T_C) - RI^2/2.$$
(2)

Because of the coefficient of performance (COP) of the refrigerator (a measure of its efficiency) and the maximum temperature drop,  $\Delta T_m$ , the amount of heat it is able to deliver is both limited by the quality of its materials, described by the dimensionless TE figure of merit, and the zT of each leg,

$$zT = \frac{\sigma \alpha^2}{\kappa} T.$$
 (3)

Equation (3) demonstrates that zT benefits from minimizing  $\kappa$ ; thus, a small  $\kappa$  is beneficial to a refrigerator. Commercially available Peltier modules are made from tetradymite materials [12], which have a thermal conductivity  $\kappa \sim 1 \text{ Wm}^{-1} \text{ K}^{-1}$ , a power factor PF =  $\sigma \alpha^2 \sim 33 \,\mu\text{W cm}^{-1} \text{ K}^{-2}$ . and material  $zT \sim 1$ .

In the case of active cooling [Fig. 1(d)], the aim is to increase the rate of heat dissipation from a hot source, adding to Fourier heat conduction. The rate of heat conduction at the hot side is then equal to the sum of Fourier heat conduction and Peltier heat, minus a contribution from Joule heating. Assuming constant temperature boundary conditions, half of the generated Joule heat returns to each end of the leg [13]:

$$Q_{\text{on - active cooling}} = K(T_H - T_C) + \alpha T_H I - R I^2 / 2. \quad (4)$$

Taking the derivative, an optimal switching current and maximum rate of heat dissipation are determined:

$$I_{S,\text{optimal}} = \alpha T_H / R, \tag{5}$$

$$Q_{\text{on-max}} = K(T_H - T_C) + (\alpha T_H)^2 / 2R$$
$$= \left(K + \frac{(\alpha T_H)^2}{2R\Delta T}\right) \Delta T.$$
(6)

In order to achieve the maximum rate of heat dissipation, TE legs should be chosen with high thermal conductance, large Seebeck coefficient, and low resistance. Optimizing these three parameters is in opposition to the idea that zT must be maximized.

By analogy with Fourier's law for passive conduction, one can define  $K + (\alpha[T_H])^2/2R\Delta T$  as the maximum effective thermal conductance. Equation (6) applies to a TE thermocouple, but it could be extended to a single TE leg by writing the equation in the form of heat flux  $(q'' \text{ in units of W/m}^2)$  so that geometrical dependences are removed:

$$q_{\text{on-max}}'' = \left(\kappa + \frac{\text{PF}T_H^2}{2\ \Delta T}\right)\nabla T,\tag{7}$$

$$\kappa_{\rm eff} = \left(\kappa + \frac{{\rm PF}T_H^2}{2\,\Delta T}\right).\tag{8}$$

 $\kappa_{\text{eff}}$  is the sum of the passive thermal conductivity  $\kappa$  and active thermal conductivity,  $\kappa_{\text{ac}} = \text{PF}T_H^2/2\Delta T$ . It is clear that the active term is large when the hot-side temperature is high and the temperature difference is small.

Equation (8) shows that the materials that maximize  $\kappa_{\rm eff}$ have a large  $\kappa$  and a large PF. Choosing a material with large  $\kappa$  is most important for the passive cooling operating mode. The material property that maximizes  $\kappa_{ac}$ , which determines the performance of the material during active cooling, is the PF. Materials with a simultaneously large  $\kappa$ and  $\sigma$  are metals, and their PF is large when their thermopower is large. Such metals come essentially in two classes: metals with Kondo effects [14] or correlated electron effects (e.g., CePd<sub>3</sub>) [15], and metals in which the thermopower is dominated by magnon drag (e.g., Co) [16]. In both systems, the Kondo or magnon-drag thermopower can be 10 times larger than the diffusion thermopower in metals, and result in metals with power factors typically two to three times larger than those of high-zTsemiconductors [17,18].

In active mode, the thermoelectric module serves only as a pump and the amount of heat  $Q_C$  that the heat sink at  $T_C$  must reject is comprised of both the heat drained at  $T_H$ and the Joule heat dissipated by the Peltier current. Therefore, the active Peltier cooler described requires a larger

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heat sink at the cold end than a purely passive cooling system would unless use can be made of a thermal storage system to handle transient active cooling. One can derive that the ratio of heat that must be rejected at the heat sink in active mode to that in passive mode at a Peltier current of  $I_{S,optimal}$ :

$$\frac{Q_{\rm on}}{Q_{\rm off}} = \left[1 + \frac{zT_H}{\Delta T} \left(T_C + \frac{T_H}{2}\right)\right].$$
(9)

#### **III. EXPERIMENT**

To experimentally verify the model and the statements made above, an active cooler is constructed from a Co-CePd<sub>3</sub> TE couple and its performance is measured in the configuration shown in Fig. 1(d). Cobalt foil is selected for the *n*-type leg because it has a peak PF of  $160 \,\mu\text{W}\,\text{cm}^{-1}\,\text{K}^{-2}$  in the temperature range of 300–400 K, which is about 3–4 times larger than conventional bismuth telluride [17]. Fewer options are available for the choice of high-PF high- $\kappa p$ -type metals. CePd<sub>3</sub> is selected for its peak PF ~90  $\mu$ W cm<sup>-1</sup> K<sup>-2</sup>, though it has a lower thermal conductivity [15].

The materials are cut into proportions that have the same length of 4.3 mm. The two materials have dissimilar thermopower and resistivity. Therefore, the cross section of the CePd<sub>3</sub> leg is about 3.7 times larger than the Co leg to match their electrical and thermal conductance to each other, as is done with commercial Peltier modules [18]. Each leg is soldered to an alumina base plate, which is attached to a copper heat sink with thermally conductive paste. A resistive heating element is mounted on top with a copper heat spreader. Type-T thermocouples of 25- $\mu$ m diameter and a few cm in length measure temperature difference along the length of the thermocouple, they drain negligible heat out of the Co-CePd<sub>3</sub> thermocouple. Experiments are performed at room temperature in a vacuum chamber with a radiation shield over the sample.

### **IV. RESULTS**

Combining the data from Watzman *et al.* [16] with the formula for effective thermal conductivity in Eq. (8), Co is modeled to reach  $\kappa_{\text{eff}} \sim 1500 \text{ W m}^{-1} \text{ K}^{-1}$  at  $\Delta T = 1 \text{ K}$ , as shown in Fig. 2(a). Similarly, using data from Boona and Morelli [15], CePd<sub>3</sub> is expected to have a  $\kappa_{\text{eff}} \sim 600 \text{ W m}^{-1} \text{ K}^{-1}$  at a  $\Delta T = 1 \text{ K}$  [see Fig. 2(b)].

Both metals are combined into one module as explained in the experimental section. First, a refrigeration experiment is performed in the configuration of Fig. 1(c) to determine optimal electric current: the value of  $T_C - T_H$  in refrigeration mode is measured as a function of  $I_S$ . This follows a parabolic relation  $(T_C - T_H)_{\text{refrigeration}} = -0.14 - 3.04$  $I_S + 0.28 I_S$  [2], a function that has a minimum for  $I_S = 5.4 \pm 0.6$ ; the value of  $(T_C - T_H)$  varies by less than 1% over that interval, which is the measurement accuracy of the thermometry. We choose an optimal switching current at the low end of this range to limit the heat load on the heat exchanger:  $I_{S,optimal} = 5$  A, which gives a  $\Delta T_{\text{refrigeration}} = 8.5 \text{ K}$  in the absence of a heat load. Second, thermal conductance is measured by the static heater-andsink method [configurations Figs. 1(b) and 1(d)]. A heat load is generated at the top of the thermocouple so that the device sits between a hot source and cold sink. The resulting temperature difference is measured as a function of heater power in both passive [zero I<sub>S</sub>, configuration Fig. 1(b)] and active [current  $I_{S,optimal}$ , configuration Fig. 1(d)] cooling states. Figure 3(a) shows the steady-state measurement of the heat flow Q for a given temperature drop  $\Delta T$ both in passive mode and active mode under optimum current. Transient measurements of the  $\Delta T$  vs time are shown in Fig. 3(b): an experimental time constant of  $\tau \sim$ 4 s is determined by normalizing the data and fitting an exponential function.

The measured thermal conductance  $(Q/\Delta T)$  is converted into an effective thermal conductivity, shown in Fig. 4 for various values of  $\Delta T = T_H - T_C$ . The values are compared to theoretical values calculated from the predicted effective



FIG. 2. Calculated values of effective thermal conductivity. The red lines represent the  $\kappa_{\text{eff}}$  of Co and black dashes represent that of CePd<sub>3</sub> (a) for  $\Delta T = 1$  K and (b) for  $\Delta T = 10$  K.



thermal conductivity of each material [Eq. (8)] and their relative cross section areas

$$\kappa_{\rm th} = (A_{\rm Co}\kappa_{\rm eff,Co} + A_{\rm CePd_3}\kappa_{\rm eff,CePd_3})/(A_{\rm Co} + A_{\rm CePd_3}).$$
(10)

As predicted by theory, the effective thermal conductivity of the cooler can exceed  $1000 \text{ W} \text{ m}^{-1} \text{ K}^{-1}$  in its active mode vs  $40 \text{ W} \text{m}^{-1} \text{K}^{-1}$  in the passive mode. The upper frame of Fig. 4 shows the relative deviation between the values calculated from the material properties and Eq. (8) and the measured values. The measured values are within 10% of the theoretical values for this temperature range; since thermal contact resistances are ignored in the calculation, the agreement mainly signifies that contact resistances are also much less of a problem with metal TE elements than they are with semiconductor ones. For small values of thermal conductivity, experimental measurements tend to be greater than theoretical values, indicating radiative losses in the experiment. Under the condition of optimal electric current and small temperature difference, thermal conductance is large and radiative losses become negligible. Electrical contact resistances are not quantified, since calculations and experiments consider only currents, but no additional heating is observed that could be ascribed to them.

The time constant measured in Fig. 3(b) is also calculated using the formula  $\tau = C_{\text{th}}/K$ . The device thermal conductance,  $K = 0.008 \ W/K$ , is calculated from the passive thermal conductivity of each element and its dimensions. Combining K with the calculated thermal capacity of each of the components in the cooling module, a time constant  $\tau$  is estimated to be 3.6 s, which is close to the measured value. Thus, the response time is a function only of the passive thermal diffusivity of the materials. Therefore, the response time is expected to be proportional to the square of the length of the TE legs. Furthermore, the high value of  $\kappa$  in metals is beneficial to the transient response of the coolers. Finally, for any particular application, optimum designs can be found that will trade off the response time for current and the  $Q/(T_H - T_C)$  ratio. FIG. 3. Data measured on a Co-CePd<sub>3</sub> TE couple cooler under a Peltier current  $I_{S_{-}}$  (a) The temperature difference as a function of the heat load; the slope of these lines gives the thermal conductance. (b) The time dependence of the temperature drop after  $I_{S}$  is switched on.

In summary, the effective thermal conductivity, and thus the cooling capacity, of an active Peltier cooling module constructed from high PF and high thermal conductivity metals is demonstrated to exceed the thermal conductivity of materials used in passive heat sinks by up to an order of magnitude, depending on the temperature differences. As predicted by theory, effective thermal conductivity diverges at small temperature differences. A Co-CePd<sub>3</sub> Peltier couple produces a measurable conductivity of over



FIG. 4. Effective thermal conductivity of the Co-CePd<sub>3</sub> TE couple cooler for two values of  $I_S$ . In the bottom frame, the symbols represent measured points and lines represent theory. The top frame (the same colors identify  $I_S$  as in the bottom frame) shows the deviation between theory and experiment to be less than 10% for the same currents as are used in the bottom frame.

1000 W m<sup>-1</sup> K<sup>-1</sup> under temperature differences of less than one degree, which is 25 times larger than the passive thermal conductivity of the couple. Materials for the *p*-type leg are harder to find than for the *n*-type leg: new Pd-free *p*-type metals are needed to lower costs. Metal TEs can have an additional advantage in active cooling because, unlike conventional materials, they can be integrated with existing fin heat exchangers. When TE legs are used as fins, much larger heat fluxes can be extracted.

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