Multiorbital effects in strong-field ionization and dissociation of aligned polar molecules CH3I and CH3Br

Sizuo Luo,¹ Shushan Zhou,¹ Wenhui Hu,¹ Xiaokai Li,¹ Pan Ma,¹ Jiaqi Yu,¹ Ruihan Zhu,² Chuncheng Wang,¹

Fuchun Liu,¹ Bing Yan,¹ Aihua Liu,¹ Yujun Yang,¹ Fuming Guo,^{1,*} and Dajun Ding^{1,†}

¹*Institute of Atomic and Molecular Physics, Jilin Provincial Key Laboratory of Applied Atomic and Molecular Spectroscopy,*

Jilin University, Changchun 130012, China

²*School of Science, Changchun University of Science and Technology, Changchun 130022, China* (Received 18 December 2016; published 19 December 2017)

Controlling the molecular axis offers additional ways to study molecular ionization and dissociation in strong laser fields. We measure the ionization and dissociation yields of aligned polar CH₃ X ($X = I$, Br) molecules in a linearly polarized femtosecond laser field. The current data show that maximum ionization occurs when the laser polarization is perpendicular to the molecular C-*X* axis, and dissociation prefers to occur at the laser polarization parallel to the C-*X* axis. The observed angular distributions suggest that the parent ions are generated by ionization from the HOMO. The angular distribution of fragment ions indicates that dissociation occurs mainly from an ionic excited state produced by ionization from the HOMO-1.

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I. INTRODUCTION

Ionization is considered the first step when a strong laser field interacts with molecules and a trigger for subsequent dynamic processes, such as laser-induced electron diffraction [\[1,2\]](#page-5-0), high-harmonic generation [\[3,4\]](#page-5-0), and Coulomb explosion [\[5,6\]](#page-5-0), which are important for understanding strong-field molecular physics and developing technologies for ultrafast imaging of molecular orbital and structure. Recently, various studies have shown that the single-active-electron (SAE) approximation cannot be translated well to describe the strongfield ionization and dissociation of multielectron polyatomic molecules $[7-10]$; the multiple orbital contribution is required to illustrate many phenomena [\[9–11\]](#page-5-0). Experimentally, observing the contribution of tunneling ionization from lowlying orbitals and disentangling the different symmetry and dynamics of multiple orbitals upon ionization and dissociation require advanced techniques based on molecular alignment [\[12–14\]](#page-5-0) and coincidence measurement [\[15–18\]](#page-5-0).

The measurement of angular-dependent strong-field ionization and dissociation yields reveals the contribution of multiple orbitals in the molecular frame [\[10,11\]](#page-5-0). Previous studies demonstrate that the angular-dependent yields of parent ions for N_2 , O_2 , and CO_2 molecules provide a possibility to image highest occupied molecular orbitals (HOMOs) [\[12,13\]](#page-5-0), but the results give an angular distribution of the ionization from $CO₂$ very different from the theoretical prediction using the conventional Molecular Ammosov-Delone-Krainov method based on the single-active-electron approximation. This discrepancy has been assigned to the contribution of dynamics exchange [\[19\]](#page-5-0). Similarly, ionization of the polar OCS molecule has attracted intensive attention. Holmegaard *et al.* measured the photoelectron angular distributions of oriented OCS molecules in a circularly polarized laser; they showed that the discrepancy between the calculated and the observed

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ionization asymmetries may be attributed to the neglect of the Stark effect in the theoretical treatment [\[20\]](#page-5-0). Hansen *et al.* also observed that the ionization yields cannot reflect the shape of the single active orbital for OCS molecules and the obtained angular distribution of the parent ion cannot be reproduced by theory even with a Stark-shift correction [\[21\]](#page-5-0). Different methods have been used to reproduce the observations, such as the weak-field asymptotic theory [\[22\]](#page-5-0) and adiabatic strong-field approximation (SFA) with orbital distortion [\[23\]](#page-5-0). Johansen *et al.* measured the angular distribution of OCS using a midinfrared laser; they also found the discrepancy between theory and experiment [\[24\]](#page-5-0), and the calculation could be improved by including the alignment-dependent saturation factor but still could not reproduce the results under a high laser intensity. These attempts are far from satisfactory to reproduce the experimental observation of OCS molecules. The participation of excited states has been put forward to explain the deviation, which is difficult to include in the calculations. Moreover, it is found that the Stark effect and dynamic core polarization have a large influence in the case of polar molecules [\[25–27\]](#page-5-0). The orbital distortion contribution induced by the Stark effect has also been observed for the polar molecules CH3F and CH3Br [\[28\]](#page-5-0) recently, and Walt *et al*. demonstrated the Stark effect and multielectron effects to be important in strong-field ionization and fragmentation of polar molecules CH_3X ($X = F$, Cl, Br, I) [\[29\]](#page-5-0). Therefore, these studies indicate that the angular distribution of strong-field ionization of polar molecules with degenerate asymmetric *π* orbitals is an interesting test for theoretical calculations.

Though single-active-electron theories of molecular ionization can contribute to the identification of new angularresolved phenomena in tunneling ionization [\[10–12,19–21\]](#page-5-0), a fundamental issue that has received extensive attention is related to the role of the multiorbital effect in strong-field molecular ionization and dissociation. For many molecules, strong-field dissociation occurs via the excited states of molecular ions generated either by tunneling ionization from low-lying orbital or by postionization excitation from ionic states of molecules [\[11,14–18\]](#page-5-0). The distinction of these two

^{*}guofm@jlu.edu.cn

[†] dajund@jlu.edu.cn

excitation pathways also relates to the ionization mechanisms of molecules, such as tunneling ionization at high laser intensities and multiphoton ionization at lower laser intensities. In the tunneling ionization regime, Akagi *et al*. found that both the HOMO and the HOMO-1 contribute to producing fragment ions Cl^+ and H^+ from HCl molecules, and the HOMO has an important influence on dissociation when laser polarization is perpendicular to the direction of the molecular axis [\[11\]](#page-5-0). Xie *et al*. performed measurements of alignment-dependent evolutions of C_2H_2 parent and fragment ions [\[14\]](#page-5-0). The observed evolutions show that the parent ion is generated by tunneling ionization from the HOMO but the fragment ions are produced from low-lying orbitals. Wu *et al*. studied the contribution of the HOMO to the next two lower-lying molecular orbitals in dissociation and Coulomb explosion of CO molecules from measurement of the orientation-dependent yields of fragment ions [\[15\]](#page-5-0). On the other hand, several other experiments at lower laser intensities show that postionization excitation makes an important contribution to the angular distribution of fragment ions [\[16–18\]](#page-5-0), in which the molecule first is ionized from its HOMO and then absorbs extra photons for the transition to ionic excited states. Sándor *et al*. [\[17\]](#page-5-0) and Zhao *et al*. [\[18\]](#page-5-0) measured multiorbital contributions to the strong-field ionization and dissociation of several molecules by using coincident electron spectroscopy. Their results indicated that fragment ions can be generated by both ionization from a lower-lying orbital and the HOMO. The contribution of these two pathways is also determined by the energy gap between nondissociative and dissociative states involved in molecular dissociation; for example, when the energy gap is smaller than the energy of one photon, these two pathways are comparably important, and if the energy gap is larger than the energy of one photon, postionization excitation is dominant. All of these findings suggest that the multiorbital contribution plays an important role in complex ionization and dissociation dynamics of molecules in strong laser fields.

In this work, we measure the angular-dependent yields of ionization and dissociation from CH_3X ($X = I$, Br) by controlling the alignment of molecules using a nonresonant femtosecond laser to investigate the contribution of asymmetric *π*-orbital and multiorbital effects on ionization and dissociation of polar molecules in strong laser fields. The results demonstrate that the ionization of different orbitals correspond to the formation of parent and fragment ions. To identify the contributions of different molecular orbitals, discussion is also presented based on these observations together with some theoretical simulations.

II. EXPERIMENTAL AND THEORETICAL METHODS

The experimental setup is schematically described in Fig. $1(a)$ and is similar to that used in our previous studies [\[30–32\]](#page-6-0). The experiments utilize a Ti:sapphire laser to produce 50-fs, 800-nm linearly polarized laser pulses with an energy of up to 4 mJ at a 1-kHz repetition rate. A Mach-Zehnder interferometer is constructed to split the laser beam into two parts for pump-probe measurement, and the intensities of both parts are variable with the use of a half-wave plate and a Glan polarizer. In order to achieve a high degree of alignment without significant ionization, a pump pulse passes through a

FIG. 1. (a) Pump-probe scheme for measuring the alignmentdependent ionization of CH_3X ($X = I$, Br) molecules. (b) HOMO and HOMO-1 of the $CH₃I$ molecule. (c) Schematic of ionization and fragmentation pathways and relevant energy levels of CH_3X ($X = I$, Br) molecules; colored arrows indicate directly tunneling ionization from different orbitals.

piece of 70-mm BK7 glass to stretch it to approximately 300 fs. The spot size of the pump beam is reduced by using an aperture to ensure that its focus is larger than that of the probe beam. The probe beam is sent through a computer-controlled delay stage and a half-wave plate rotates the laser polarization. The pump and probe pulses are collinear and focused with a 250-mmfocal-length lens onto the molecular beam. The spectrometer contains three sections: a source chamber, a hexapole chamber, and a velocity-map-imaging chamber. In the source chamber, the supersonic molecular beam is produced by a pulsed valve (General Valve series 9) with 2.5% CH₃I or CH₃Br seeded in neon at a 4-b stagnation pressure through a 0.5-mm orifice. In the hexapole chamber, a 1-m hexapole rotational-state selector, composed of six hexagonally placed rods of 4-mm diameter, is used to prepare a specific rotational-state-selected molecular beam for the purpose of achieving a high degree of alignment. The $CH₃X$ molecules after the selector are mainly populated at $|J, K, M\rangle = |1, \pm 1, \pm 1\rangle$ rotational states. The alignment of these rotational-state-selected molecules has been studied perviously $[31,32]$; the nonresonant laser (the pump beam) interacts with these state-selected molecules and the obtained maximum degree of alignment is $\langle \cos^2 \theta \rangle = 0.7$. These wellaligned molecules are used as the targets for measuring the angular-dependent ionization yields in the strong laser field (the probe beam). The ions generated by the laser field are extracted and accelerated by an electrostatic lens system and are projected onto a detector at the end of a time-of-flight tube. The mass spectra can be obtained with a photomultiplier tube, accumulated and averaged by a digital storage oscilloscope, and, finally, transferred into a computer. The intensity of the femtosecond laser is calibrated by the saturated intensity of Xe ionization [\[33\]](#page-6-0).

The SFA [\[20,23](#page-5-0)[,34,35\]](#page-6-0) and time-dependent density functional theory (TDDFT) [\[36,37\]](#page-6-0) methods have been widely applied to describe molecular ionization in strong laser fields. In order to explain the experimental results, we performed calculation of the angular-dependent ionization rates using the molecular strong-field approximation (MO-SFA) method [\[34,35\]](#page-6-0). The SFA method treats the Coulomb potential in a perturbative way and the external laser field in a nonperturbative way after ionization and has the advantage of calculating the ionization rate from multiphoton to tunneling ionization. The transition amplitudes of ionization for molecules at the length gauge can be written as

$$
f_{\alpha}(\vec{p}) = -i \int_{-\infty}^{+\infty} \langle \vec{p} + \vec{A}_{\alpha}(t) | \vec{E}(t) \cdot \vec{r} | \varphi_0 \rangle e^{-iS_{\alpha}(\vec{p}, t)} dt, \quad (1)
$$

where

$$
S_{\alpha}(\vec{p},t) = \int_{t}^{+\infty} [\vec{p} + \vec{A}_{\alpha}(t')]^{2}/2 + I_{p}, \qquad (2)
$$

with

$$
|\vec{p} + \vec{A}_{\alpha}(t)\rangle = 1/(2\pi)^{3/2} e^{i[\vec{p} + \vec{A}_{\alpha}(t)] \cdot \vec{r}}, \tag{3}
$$

and φ_0 is the molecular orbital function, $E(t)$ is the laser electric field, $A_\alpha(t)$ is the vector potential of the laser field, \vec{p} is the momentum of the electron, and I_p is the ionization potential of the molecule. The ionization yield for molecules at different angles α can be given by

$$
P(\alpha) = \int |f_{\alpha}(\vec{p})|^2 dp^3.
$$
 (4)

The wave function φ_0 of the molecular orbital in the MO-SFA calculation is obtained using GAMESS software with the basis set SPK-QZP [\[38\]](#page-6-0). Schematics of the HOMO and HOMO-1 of $CH₃I$ molecules are shown in Fig. [1\(b\).](#page-1-0) Different orbital wave functions are used in calculating the ionization rates from the HOMO and HOMO-1. A schematic of the ionization and dissociation pathways is shown in Fig. $1(c)$. The fragment ions are generated by dissociation of excited molecular ions. These ions further are dissociated either into $CH₃⁺$ and *X* (I, Br) when populated in the A^2A_1 state or into CH₃ and X^+ $(X = I, Br)$ when populated in the $E_{3/2,1/2}$ state [\[39,40\]](#page-6-0).

Alternatively, the TDDFT method is also used to calculate molecular ionization rates because it naturally includes multielectron dynamics [\[36,37\]](#page-6-0). We calculate the angulardependent multiorbital ionization rates of $CH₃X$ by employing a set of time-dependent Kohn-Sham (TDKS) equations to construct the evolutions of electron density in laser fields. The TDKS equations are

$$
i\frac{\partial}{\partial t}\Psi_i(r,t) = \left[-\frac{\nabla^2}{2} + V_{\text{eff}}(r,t)\right]\Psi_i(r,t) \quad (i = 1,2,\ldots,N),\tag{5}
$$

where $\Psi_i(r,t)$ is the Kohn-Sham orbital, *N* is the number of orbitals, and the time-dependent Kohn-Sham potential $V_{\text{eff}}(r,t)$ is

$$
V_{\text{eff}}(r,t) = V_{\text{ne}}(r) + V_H(r,t) + V_{\text{xc}}(r,t) + r \cdot E(t), \qquad (6)
$$

where V_{ne} is the electron-nuclear potential, V_H is the Hartree potential, and V_{xc} is the exchange-correlation potential, which all include the multielectron effect, and the local density approximation has been used to construct the exchangecorrelation potential, $E(t)$ is the electric field of the laser, which is a trapezoidal field in our calculation. Different pseudopotentials have been used to represent the electron-ion interaction of atoms, the electron-ion interaction for the C atom is represented by norm-conserving Troullier-Martins pseudopotentials [\[41\]](#page-6-0), and those for the H, Br, and I atoms

FIG. 2. Measured parent ion yields of $CH₃I⁺$ and $CH₃Br⁺$ as a function of the pump-probe delay. Black circles represent the alignment-dependent yield; blue lines, the time-dependent alignment evolution $\langle \cos^2 \theta \rangle$ calculated by the time-dependent Schrödinger equation.

are represented by Hartwigsen-Goedecker-Hutter pseudopotentials [\[42\]](#page-6-0). The TDKS equations are calculated by using OCTOPUS [\[43,44\]](#page-6-0); the time-dependent survival probability of a particular Kohn-Sham orbital is

$$
N_i(t) = \int |\Psi_i(r,t)|^2 dr,\tag{7}
$$

and the ionization probability of a particular Kohn-Sham orbital is

$$
P_i(t) = 1 - N_i(t). \tag{8}
$$

The angular distribution of multiorbital ionization is the ionization probability obtained from different angles between the laser field $E(t)$ and the molecular axis.

III. RESULTS AND DISCUSSION

We measured the alignment dependence and angular distributions of parent and fragment ions of polar molecules CH_3X ($X = I$, Br). The alignment-dependent ionization yields are shown in Fig. 2, in which the measured time evolution of the parent ion CH_3I^+ and CH_3Br^+ signal intensity is shown by the black circles and the calculated time evolution of alignment $\langle \cos^2 \theta \rangle$ by the blue lines. The degree of alignment is calculated using the time-dependent Schrödinger equation [\[45\]](#page-6-0). It is clearly shown that the time evolutions of parent ion yields are out of phase by π with respect to the time evolutions of alignment, indicating that $CH₃I$ and $CH₃Br$ prefer to be ionized when the probe laser polarization is perpendicular to the molecular axis. In Fig. 2, we also see that, in order to measure the angular-dependent ionization yield of CH3*X*, the probe laser interacts with the molecules at a delay time

FIG. 3. Time-of-flight mass spectra of (a) $CH₃$ I and (b) $CH₃Br$ irradiated with 800-nm, 50-fs laser pulses at intensities of 5.4×10^{13} and 6.0×10^{13} W/cm².

of 68.1 or 53.1 ps for CH₃I or CH₃Br, respectively, where the molecules are maximum aligned along the pump-laser polarization. Typical time-of-flight mass spectra of CH3I and CH3Br molecules under an 800-nm laser with intensities of 5.4×10^{13} and 6.0×10^{13} W/cm² are shown in Fig. 3. The Keldysh parameter $\gamma \approx 1.2$ for these two molecules at the present laser intensities. The dominant peaks in the mass spectra are the $\text{CH}_3 X^+$ parent ions and the fragments CH_3^+ and X^+ .

The measured angular distributions of the parent and fragment ions from the ionization of $CH₃I$ and $CH₃Br$ molecules are shown in Fig. 4. The angular distribution shows a minimum at 0° and a maximum at 90° for parent ions $\rm CH_{3}I^{+}$

FIG. 4. Angular distribution of parent and fragment ions measured from the well-aligned molecular targets CH₃I and CH₃Br at the pump-probe delay time of 68.1 and 53.1 ps, respectively. Angular distributions for (a) CH_3I^+ , (b) I^+ , and (c) CH_3^+ and for (d) CH_3Br^+ , (e) Br^+ , and (f) CH_3^+ . Error bars are estimated from the different experimental runs and blue lines are the fitted curves using Legendre polynomials $I = 1 + \sum \beta_{2i} P_{2i}$.

FIG. 5. Experimentally deconvolved angular-dependent ionization and dissociation probability $S(\theta)$ of CH_3X molecules, together with the MO-SFA and TDDFT calculation results. (a, b) Angular distributions of parent and fragment ions from $CH₃I$ and $CH₃Br$ (c, d) Angular distributions of HOMO (solid and dashed black lines) and HOMO-1 (solid and dashed red lines) orbitals from MO-SFA and TDDFT calculations for $CH₃I$ and $CH₃Br$. Angular distributions are normalized at the maximum intensities.

and CH_3Br^+ . However, the fragment ions show very different angular distributions, with a maximum at $0°$ and a minimum at 90°. In order to compare them with the theoretical calculation, the measured angular distributions of ions must be converted to the molecular frame from the experimental frame. The angular-dependent ionization and dissociation probabilities in the molecular frame, $S(\theta)$, are obtained from the deconvolution of the measured angular-dependent ion intensities, $M(\alpha)$, using the alignment distribution $A(\theta', \varphi')$ [\[12\]](#page-5-0),

$$
M(\alpha) = \int_{\varphi'=0}^{2\pi} \int_{\theta'=0}^{\pi} S[\theta(\theta', \varphi'; \alpha)] A(\theta', \varphi') \sin(\theta') d\theta' d\varphi', \quad (9)
$$

where θ' and φ' are the polar and azimuthal angles in the experimental frame with respect to the polarization axis of the aligning beam, and the angle θ is given by

$$
\cos(\theta) = \cos(\alpha)\cos(\theta') - \sin(\alpha)\sin(\theta')\sin(\varphi'). \qquad (10)
$$

We expand the angular-dependent ionization and dissociation $\sum_{i=0}^{3} a_{2i} p_{2i} \cos(\theta)$. Figure 5 gives the angular-dependent ionization and dissociation probabilities *S*(*θ*) of these two probabilities in terms of Legendre polynomials as $S(\theta)$ = molecules after deconvolution.

The results for the parent ions $CH₃H⁺$ and $CH₃Br⁺$, shown by the black circles in Figs. $5(a)$ and $5(b)$, are similar to the previous observations for OCS ⁺ [\[21\]](#page-5-0). This is because these molecules have similar HOMO distributions. In the case of $CH₃X$, the outer orbital of CH₃ becomes an orbital with $a₁$ symmetry in the C_{3v} point group and it combines with the n*pσ* orbital of the halogen atom to give rise to a bondingantibonding pair of *A*¹ orbitals and form nonbonding HOMOs [\[46\]](#page-6-0). It is expected that these degenerated HOMOs are close to

FIG. 6. Normalized angular distributions of parent ions for (a) $CH₃I$ and (b) $CH₃Br$ at different laser intensities.

an atomic *p* orbital, with the major character of the *p* electron distribution perpendicular to the C-*X* bonding axis. Therefore, it is reasonable to believe that the angular distribution of parent ions is maximum with the laser polarization perpendicular to the C-*X* bonding axis when strong-field ionization occurs and leads to the formation of CH_3X^+ in its electronic ground state X^2E [\[29\]](#page-5-0). We exclude the influence of excitation because the angular distributions of parent ions are similar with varying laser intensities as shown in Fig. 6. The parent ions should have different angular distributions considering that excited states have a laser-intensity-dependent population [\[47\]](#page-6-0). A trend of less anisotropic angular distributions appears with increasing laser intensity, due to the fact that more orbitals participate in the ionization process and the saturation effect under higher laser intensities. The ionization yield at 0*^o* will increase at higher laser intensities because of the contribution from the HOMO-1 [\[48\]](#page-6-0). Meanwhile the saturation effect will reduce the ionization signal at 90◦. These effects will reduce the sharpness of the lobe in the measured angular distribution, as shown in Fig. 6. Furthermore, the distribution near 0° for $CH₃I⁺$ is less than that for $CH₃Br⁺$ [see Figs. [5\(a\)](#page-3-0) and [5\(b\)\]](#page-3-0). These distributions can be considered the contribution from HOMOs of non-*p*-electron character in the ionization. This interpretation is consistent with a recent calculation in which the contribution of the nonbonding p electron has been demonstrated to be different for different methyl halide molecules. The ionization is almost exclusively from the *p* orbital, with almost no contribution from the methyl group for CH3I and with a nonnegligible contribution from the methyl group for CH_3Br [\[49\]](#page-6-0).

For further interpretation of the measured angular distributions, calculated results based on the MO-SFA at length gauge (black lines) and the TDDFT (red lines) are presented in Figs. $5(c)$ and $5(d)$. It is found that in the case of CH₃I the angular distributions of the HOMO obtained from the TDDFT and SFA methods are almost the same as the experimental angular distribution of $CH₃I⁺$, as shown by the black circles in Fig. $5(a)$. This supports that the main contribution is from HOMOs of a major n*p*-orbital character during the formation of the parent ions of CH_3I^+ . But in the case of CH_3Br^+ , the discrepancies between the measured and the calculated angular distributions are obvious [see Figs. $5(b)$ and $5(d)$] when the calculations consider HOMO electron ionization using

the same theoretical approaches. This situation may come from the molecular orbital modification by the strong laser fields, which can affect the angular distribution of ionization significantly, as observed in the orientation-dependent highharmonic generation of $CH₃Br [28]$ $CH₃Br [28]$.

Measurement of the angular-dependent dissociation yields offers an opportunity to reveal the dissociation mechanisms [\[14–16](#page-5-0)[,50\]](#page-6-0). Figures $5(a)$ and $5(b)$ also show the measured angular distributions of the fragment ions CH_3 ⁺ (red squares) and X^+ (blue triangles) for CH₃I and CH₃Br, respectively. With the probe laser intensity at 5.4×10^{13} or 6.0×10^{13} W*/*cm2, these fragment ions can be produced only from dissociation; no evidence has been observed of Coulomb explosion from multicharged parent ions (Fig. [3\)](#page-3-0). Interestingly, these fragment ions show similar angular distributions after deconvolution, with a maximum at $0°$ and a minimum at $90°$, the reverse of that of the parent ions. Also, the calculated results by MO-SFA or TDDFT for ionization of the HOMO-1 give similar angular distributions, as shown by red lines in Figs. $5(c)$ and $5(d)$ for CH₃I and CH₃Br, respectively, and the differences appearing at 90◦ may be caused by the HOMO contribution [\[11\]](#page-5-0). Thus, the present analyses suggest that the fragment ions of CH_3X ($X = I$, Br) are likely generated mainly from tunneling ionization of the HOMO-1. The present work indicates that the multiorbital contribution is important in understanding strong-field ionization of complex molecules. For the molecules CH_3X , this contribution can exert different influences with changing halogen atoms from I to Br. Measurement and reproduction of the angular distributions for either parent or fragment ions can provide sensitive orbital testing, as well as other dynamic effects. For these complex molecules, further theoretical developments and modeling are definitely required to fully establish this aspect.

IV. CONCLUSION

We have performed experimental measurements of angulardependent ionization and dissociation yields of two polar molecules, CH_3X ($X = I$, Br), in strong laser fields. The obtained angular distributions suggest that the HOMO makes an important contribution to the production of parent ions. The dissociation mechanisms have been identified by comparing the measured angular distributions of fragment ions with theoretical calculations based on ionization from different molecular orbitals, and the agreement between experiment and theory indicates that the fragment ions are mainly generated from ionization of the molecules' HOMO-1. Experimental measurements and theoretical calculations on strong-field ionization and dissociation of polyatomic polar molecules will promote understanding of strong-field and attosecond science regarding complex polyatomic molecules.

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