

Emitter-site-selective photoelectron circular dichroism of trifluoromethyloxirane

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The angle-resolved inner-shell photoionization of R-trifluoromethyloxirane, C₃H₃F₃O, is studied experimentally and theoretically. Thereby, we investigate the photoelectron circular dichroism (PECD) for nearly symmetric O 1s and F 1s electronic orbitals, which are localized on different molecular sites. The respective dichroic β_1 and angular distribution β_2 parameters are measured at the photoelectron kinetic energies from 1 to 16 eV by using variably polarized synchrotron radiation and velocity map imaging spectroscopy. The present experimental results are in good agreement with the outcome of *ab initio* electronic structure calculations. We report a sizable chiral asymmetry β_1 of up to about 9% for the K-shell photoionization of oxygen atom. For the individual fluorine atoms, the present calculations predict asymmetries of similar size. However, being averaged over all fluorine atoms, it drops down to about 2%, as also observed in the present experiment. Our study demonstrates a strong emitter and site sensitivity of PECD in the one-photon inner-shell ionization of this chiral molecule.

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I. INTRODUCTION

In 1976, Ritchie predicted theoretically that angle-resolved photoelectron spectra of chiral molecules exhibit a sizable circular dichroism (CD) effect, which in contrast to the normal CD in total absorption spectra is governed by the electric dipole interaction [1]. It took about 25 years to verify these predictions experimentally [2]. Since then, the photoelectron circular dichroism (PECD) in the one-photon ionization of chiral molecules in the gas phase has been extensively studied experimentally and theoretically. At present, numerous independent works have confirmed that for randomly oriented molecules, PECD can be seen as a forward-backward asymmetry in the emission of photoelectrons which is typically on the order of a few percent. Most of those studies are reviewed in Refs. [3–5].

The first experiments performed with circularly polarized synchrotron radiation on randomly oriented bromocamphor [2] and camphor [6] illustrated a sizable PECD of about 3%–4%. It was also noticed that PECD changes with the binding energy of a system, i.e., it is different for the ionization of different molecular orbitals [2]. Furthermore, a strong dependence of PECD on the photoelectron kinetic energy (exciting-photon energy) was found experimentally and theoretically for outer-shell photoionization of methyloxirane [7–9], chiral derivatives of oxirane [10], as well as camphor and fenchone [11–13].

Nowadays, PECD in the one-photon ionization of outer electrons is a well-established research area, which includes studies of molecular dimers [14], clusters [15], metal-organic complexes [16], and even small biological molecules [17–20]. Experimentally, this effect has been studied by tunable circularly polarized synchrotron radiation utilizing different methods of angle-resolved photoelectron spectroscopy. Theoretically, the continuum multiple scattering method with the local $X\alpha$ exchange correlation (CMS- $X\alpha$) [21] and the time-dependent density-functional theory (TDDFT) B -spline LCAO formalism [22] were used in those studies.

Molecular orbitals of the outer valence electrons (HOMO- n) are typically delocalized over a large part of a molecule and are strongly asymmetric. This asymmetry of the initial electronic state is naturally imprinted in the observed PECD through the photoionization amplitudes. Notwithstanding, there is another contribution to PECD, which is related to the final electronic state through the same amplitudes. Indeed, a contrastive theoretical study of chiral derivatives of oxirane [10] has suggested that the magnitude of PECD is also governed by the ability of the outgoing photoelectron continuum wave to probe the asymmetry of the molecular ion potential.

The latter effect of the final electron continuum state plays a decisive role in the photoionization of almost-symmetric inner-shell electrons. The very first study of inner-shell PECD performed for the O=C(1s) ionization of camphor [23] reported a large chiral asymmetry, which was shown to be dependent on the photoelectron kinetic energy scanned up to 65 eV above the ionization threshold. These experimental results were supported by numerical simulations, which suggested that PECD is caused here by the final-state scattering effects on the chiral potential of a molecule [23].

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Recently [24,25], a sizable PECD effect in the photoionization of camphor and fenchone by intense short circularly polarized laser pulses was observed in the 2+1 resonance-enhanced multiphoton ionization regime. Since then, many experimental and theoretical investigations of the multiphoton PECD by femtosecond laser pulses have been reported in literature. In contrast to the one-photon ionization regime, multiphoton PECD provides important complementary information on the effect of intermediate electronic states involved in different multiphoton ionization schemes (see Refs. [26–28] for very recent results).

At present, there are several works reporting PECD after inner-shell photoionization of chiral molecules [10–12,23,29–31]. As an advantage, it allows one to selectively address electronic orbitals localized on a particular chiral center of a molecule. Importantly, inner-shell ionization, followed by an ultrafast Auger decay and subsequent fragmentation of the resulting dication by Coulomb explosion, offers the possibility to access molecular frame photoelectron angular distributions by multicoincident detection techniques [32]. This principle has recently been utilized to study PECD in the O 1s photoionization of uniaxially oriented methyloxirane [31].

Very recently [5,33], PECD in the one-photon ionization of electrons from the HOMO and HOMO-1 of trifluoromethyloxirane ($C_3H_3F_3O$) has been studied in the vibrationally resolved mode. For this chiral molecule, PECD after inner-shell photoionization has not yet been investigated. In order to study the PECD at different emitter sites of this molecule, we address in this work the O 1s and F 1s photoionization of R-trifluoromethyloxirane. Our experimental and theoretical methods are described in Sec. II. The measured and calculated dichroic and angular distribution parameters are compared and discussed in Sec. III. We conclude in Sec. IV with a brief summary and outlook.

II. EXPERIMENT AND THEORY

The present experiments were performed at the Stanford Synchrotron Radiation Lightsource (SSRL) at the BL13-2 beamline [34,35]. Equipped with the 26-pole elliptically polarizing undulator [36], this synchrotron beamline can produce $94 \pm 6\%$ circularly polarized soft x rays. The spherical grating monochromator at BL13-2 can achieve a spectral resolving power of 10^4 over an energy range from 180 to 1100 eV with a flux of 10^{11} to 10^{12} photons/s.

The sample, (R)-(+)-3,3,3-Trifluoro-1,2-epoxypropane (97%, Sigma Aldrich), was introduced into the interaction chamber by an effusive gas jet via a 0.5 mm skimmer to reach the vacuum conditions necessary for the beamline operation. The interaction region of gas jet and synchrotron radiation coincided with the focus point of the velocity map imaging (VMI) spectrometer of the LAMP chamber of the LCLS AMO beamline [37]. The electric fields were optimized to collect electrons with kinetic energies of around 10 eV with an energy resolution of about 5%. The acquired data sets for the O and F *K* edges cover photoelectron kinetic energies up to 16 eV while using circularly (with both positive “+” and negative “−” helicities) and linearly polarized light.

An inverse Abel transformation was applied to reconstruct slices of the 3D photoelectron angular distributions out of the 2D-VMI images. Several transformation methods, namely the BASEX method [38], direct integration of the Abel integral [38], an iterative approach [39], the lin-BASEX [40], the Hansen-Law algorithms [41], and onion peeling [42–44] were used as an evaluation stability proof and as a means to estimate the uncertainty. The dichroic parameter β_1 and angular distribution parameter β_2 were determined independently. The difference between two normalized photoelectron angular distributions measured with the circularly polarized light was fitted to extract β_1 using the relation $I_+(\theta) - I_-(\theta) = 2\beta_1 P_1(\cos\theta)$. The normalized photoelectron angular distribution measured with the linearly polarized light revealed β_2 via the parametrization $I_0(\theta) = 1 + \beta_2 P_2(\cos\theta)$. The latter procedure was cross-checked by extracting β_2 from the sum of normalized $I_{\pm}(\theta)$ images using the relation $I_+(\theta) + I_-(\theta) = 2 - \beta_2 P_2(\cos\theta)$.

Before carrying out experiments with the enantiomer-pure sample, several tests were performed with the racemic mixture of the target (99%, Sigma Aldrich) in order to improve the data accuracy. Thereby, an intrinsic detector calibration to ensure symmetric angular distributions (the absence of a PECD) for linearly and circularly polarized light with both helicities was accomplished. In addition, the degree of circular polarization was estimated with the help of angular distributions acquired by the linearly polarized light under assumption of almost 100% degree of polarization of the latter. The degree of circular polarization and purity of the enantiomeric target were taken into account in the data analysis for dichroic β_1 and angular distribution β_2 parameters.

Each presently recorded velocity map image exhibits a weak symmetric background due to low-energy electrons induced by secondary processes. Contrary to the main photoelectron signal which has a well-defined energy, this position-dependent noise covers all electron energies. In addition, secondary electrons originated at a copper mesh in front of the MCP detector have introduced an artificial mesh-type signal. The main part of this signal was filtered out before the Abel reconstruction. These are the main sources of experimental uncertainty. Further uncertainties are estimated by statistics error perturbation, fit quality, and 3D-reconstruction stability.

The laboratory-frame angular distribution of the photoelectrons emitted by randomly oriented molecules excited by circularly polarized light is described by the well-known formula for the differential photoionization cross section [1,45,46]:

$$\frac{d\sigma^{\pm}}{d\Omega} = \frac{\sigma}{4\pi} \left[1 \pm \beta_1 P_1(\cos\theta) - \frac{1}{2} \beta_2 P_2(\cos\theta) \right]. \quad (1)$$

Here, “±” stands for the positive and negative helicity of the circularly polarized radiation, $P_L(\cos\theta)$ are the Legendre polynomials, and θ is the angle between the direction of the propagation of the exciting radiation and the direction of the emission of photoelectrons encompassed in the solid angle $d\Omega$. The total photoionization cross section σ and parameters $\beta_L (L = 1, 2)$ can be computed via the following

equations [47,48]:

$$\begin{aligned} \sigma &= \sum_{\Lambda_0 \Lambda_1} \sum_{\ell m q} |\langle \Lambda_1 \varepsilon \ell m | \mathbf{d}_q | \Lambda_0 \rangle|^2, \\ \beta_L &= \frac{1}{\sigma} \sum_{\Lambda_0 \Lambda_1} \sum_{\ell m q} \sum_{\ell' m' q'} (i)^{\ell+\ell'} (-1)^{\ell'+m+q} \\ &\quad \times e^{i(\delta_{\ell m} - \delta_{\ell' m'})} \sqrt{\frac{3L(8+L)}{2}} (2\ell+1)(2\ell'+1) \\ &\quad \times \begin{pmatrix} \ell & \ell' & L \\ -m & m' & q-q' \end{pmatrix} \begin{pmatrix} 1 & 1 & L \\ q & -q' & m'-m \end{pmatrix} \\ &\quad \times \begin{pmatrix} \ell & \ell' & L \\ 0 & 0 & 0 \end{pmatrix} \langle \Lambda_1 \varepsilon \ell m | \mathbf{d}_q | \Lambda_0 \rangle \langle \Lambda_1 \varepsilon \ell' m' | \mathbf{d}_{q'} | \Lambda_0 \rangle^*. \end{aligned} \quad (2a)$$

In these equations, $\langle \Lambda_1 \varepsilon \ell m | \mathbf{d}_q | \Lambda_0 \rangle$ is the dipole transition matrix element for the ionization of the electronic state Λ_0 by the linearly ($q = 0$) or circularly ($q = \pm 1$) polarized light, which results in the population of the final ionic state Λ_1 and emission of the photoelectron partial wave $\varepsilon \ell m$ with the kinetic energy ε , angular momentum ℓ , and its projection m (quantum numbers m and q are defined in the molecular quantization frame). Further on, $\delta_{\ell m}$ is the phase shift of the partial electron wave, and summations over indices $\Lambda_{0,1}$ must be performed over all degenerate electronic states.

The photoionization transition amplitudes were calculated in the present work by the single center (SC) method and code [49,50], which provides an accurate description of the partial photoelectron continuum waves in molecules. It was already successfully applied to study angle-resolved spectra of diatomic molecules [47,51–53], weakly bound dimers [54], polyatomic molecules [48,55,56], small metallic clusters [50], and even chiral molecules [31,57]. Briefly, a molecular orbital is represented in the SC method with respect to a single center of the molecule via an expansion in terms of spherical harmonics. The wave functions of a photoelectron in the continuum are sought as numerical solutions of the system of coupled one-particle equations for the radial partial electron waves with accurate molecular field potentials, provided by all nuclei and occupied electronic shells of a molecule. Thereby, the main physical mechanism for handing a chiral asymmetry of the molecular ion potential over to the outgoing photoelectron wave by multiple scattering effects is accurately described by the method. The numerical scheme implies noniterative accounting for the nonlocal exchange Coulomb interaction of a photoelectron with a molecular core, which makes the computational procedure robust [49,50].

In the present work, photoionization transition amplitudes were computed in the frozen-core Hartree-Fock approximation at the equilibrium internuclear geometry of the ground electronic state of R-trifluoromethyloxirane, optimized at the (2,2)-CASSCF/6-31G(*d*, *p*) level of theory. Molecular orbitals of the occupied electronic shells, computed by the PC GAMESS (general atomic and molecular electronic structure system [58]), version of Granovsky [59], were represented relative to the nuclear charge center of gravity (chosen as the molecular center) by expansions over spherical harmonics with $\ell, |m| < 60$. The SC expansions of the continuum partial waves were

restricted by partial harmonics with $\ell_\varepsilon, |m_\varepsilon| < 35$. Finally, multipole expansions of the exchange Coulomb interaction of the photoelectron with the core electrons [50] were restricted by harmonics with $k, |q| < 15$.

III. RESULTS AND DISCUSSION

The present angle-resolved spectra for the *K*-shell photoionization of oxygen and fluorine atoms are collected in Figs. 1 and 2, respectively. These figures depict (from the top to bottom) the total cross section σ , dichroic parameter β_1 , and angular distribution parameter β_2 computed as functions of the electron kinetic energy up to 20 eV. For the photoelectron energies up to 16 eV, the computed σ , β_1 , and β_2 parameters are compared with the presently measured respective values. The experimental cross sections, determined on the relative scale by normalization of the total electron signal to the data-acquisition time, gas pressure, and photon flux, were set in Figs. 1 and 2 on the absolute scale with the help of the present theory. No further normalization to compare the computed and measured dimensionless β_1 and β_2 parameters is required.

For higher photoelectron energies from 10 to 16 eV, the recorded velocity map images were partially compromised in

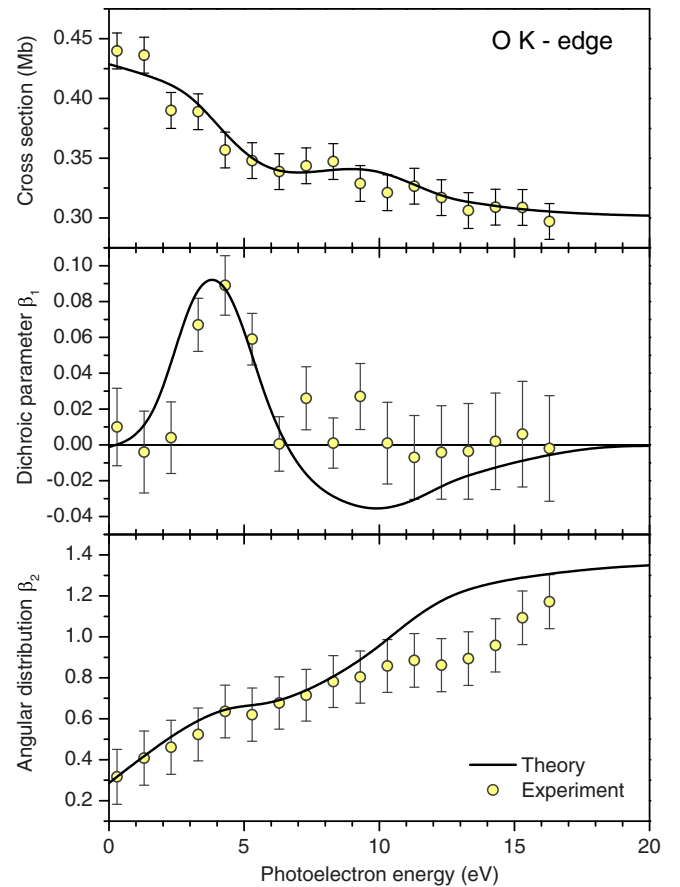


FIG. 1. Presently measured (circles with error bars) and computed (solid curves) cross section (uppermost panel), dichroic parameter β_1 (middle panel), and angular distribution parameter β_2 (lowermost panel) for the O *1s* photoionization of R-trifluoromethyloxirane as functions of the photoelectron energy.

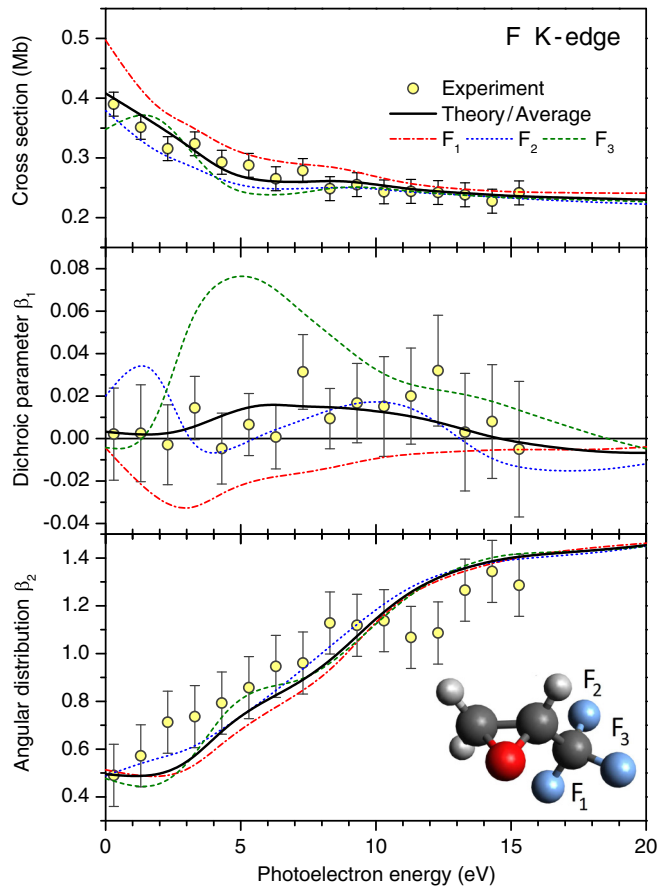


FIG. 2. Presently measured (circles with error bars) and computed (curves) cross sections (uppermost panel), dichroic parameters β_1 (middle panel), and angular distribution parameters β_2 (lowermost panel) for the F $1s$ photoionization of R-trifluoromethyloxirane as functions of the photoelectron energy. Results computed for the individual fluorine atoms F_i , as enumerated in the inset in the lowermost panel, are shown by broken curves, whereas the final data, averaged over three atoms, are drawn as solid curves.

the forward direction. Therefore, only part of the data was taken into account for the β_1 determination. The procedure of reconstructing data with a reduced image area and, therefore, reduced statistical validity was carefully cross-checked with different transformation methods [38–44] and is taken into account in the determination of uncertainties, which can be clearly seen in the middle panels of Figs. 1 and 2 in the energy range of 10–16 eV. This problem did not affect the accuracy of the measured β_2 parameter, since the main dipole contributions of the electron signal, which are important for the determination of positive β_2 values, were always in the detection area.

As discussed above, dichroic parameters were determined from the difference between the two normalized photoelectron signals corresponding to the use of two opposite circular polarizations of exciting synchrotron radiation. This procedure automatically implies an almost full elimination of the extrinsic artificial signals deposited in each spectrum as a symmetric background and the residual mesh-type pattern. On the contrary, these signals cannot be excluded from the data

analysis when determining angular distribution parameter out of the single spectrum acquired with the linearly polarized radiation. This led to larger error bars for the measured β_2 values compared to those for β_1 (see lowermost panels of Figs. 1 and 2). Finally, the one-particle frozen-core Hartree-Fock approximation, used in the present calculations, does not include electron correlations and core relaxation effects. Neglecting these effects may result in an underestimation of the absolute values of the total photoionization cross section very close to the ionization threshold by up to a factor of 2 [60–62].

The middle panel of Fig. 1 demonstrates an overall good agreement between the β_1 values computed and measured for the O K edge. In most of the cases the deviation between theory and experiment is almost within the experimental error bars. Both theory and experiment indicate a sizable chiral asymmetry of $\beta_1 \approx 9\%$ around $\varepsilon = 4$ eV. From the lowermost panel of Fig. 1 one can see that the presently computed and measured angular distribution parameters β_2 agree well, and they start to deviate from each other only for the electron energies above about 10 eV. This might be related to the fact that the presently measured β_2 values are slightly underestimated due to the presence of a symmetric background in the collected electron signal as discussed above.

The presently computed O $1s$ photoionization cross section exhibits a weak feature around the photoelectron kinetic energy of 9 eV. As one can see from Fig. 1, this peak is fully correlated with the change of sign in the computed and measured dichroic parameter β_1 and also with a weak modulation of the computed and measured angular distribution parameter β_2 . We assign this feature to a shape resonance in the O $1s$ photoionization continuum. The presence of this shape resonance could be the main reason for a somewhat larger disagreement between the computed and measured dichroic parameter β_1 around the photoelectron kinetic energy of 9 eV.

The σ , β_1 , and β_2 parameters computed for the individual fluorine atoms are depicted in Fig. 2 by broken curves. The numbering of F_i atoms is indicated in the inset at the bottom of this figure. As one can see, the cross sections σ and angular distribution parameters β_2 , computed for different atoms, are very similar for higher photoelectron energies, and they differ from each other only for electron energies below about 10 eV. On the contrary, dichroic parameters β_1 of the individual fluorines (middle panel) are very different from each other: $\beta_1(F_1)$ is mainly negative; $\beta_1(F_3)$ is mainly positive; whereas $\beta_1(F_2)$ changes its sign as a function of electron kinetic energy. These findings illustrate that the dichroic parameter is much more sensitive to the molecular potential than the angular distribution parameter. The present theory predicts the individual asymmetry to be on the order of about $\beta_1 \approx 8\%$ for the F_3 atom. An understanding of why the $1s$ photoionization of the F_3 atom exhibits the largest PECD effect is a rather involved subject and goes far beyond the present work.

The $1s$ photoelectrons emitted from different fluorine atoms cannot be energetically resolved, since the corresponding lines overlap within their natural lifetime widths. In order to compare the present theoretical and experimental data, the computed β_L parameters need to be averaged over three

fluorine atoms as follows:

$$\sigma = \frac{1}{3}\sigma_{\text{tot}} = \frac{1}{3} \sum_{i=1}^3 \sigma(F_i), \quad (3a)$$

$$\beta_L = \frac{1}{\sigma_{\text{tot}}} \sum_{i=1}^3 \sigma(F_i)\beta_L(F_i). \quad (3b)$$

The average theoretical parameters σ , β_1 , and β_2 are depicted in Fig. 2 by solid curves. Note that the average cross section σ in Eq. (3a) and in the uppermost panel of Fig. 2 differs from the total cross section σ_{tot} by a factor of 3. As a result of this averaging, the computed chiral asymmetry drops significantly down, and it does not exceed now 2%. This is consistent with the present observation, as can be seen from the middle panel of Fig. 2 which demonstrates a very good agreement between the average theoretical and experimental β_1 parameters. Similar to the case of the O K edge, the averaged computed and the measured angular distribution parameters β_2 for the F K edge agree very well for the lower electron energies, and start to deviate for the energies above 10 eV.

IV. CONCLUSION

Dichroic parameters β_1 and angular distribution parameters β_2 for the $1s$ photoionization of O and F atoms in R-trifluoromethyloxirane were measured and computed for different electron kinetic energies above the respective ionization thresholds. The experiment was performed at the BL13-2 beamline of SSRL (SLAC) utilizing variably polarized soft x rays and velocity map imaging spectroscopy. Electronic-structure calculations were carried out by the single center method in the frozen-core Hartree-Fock approximation.

The present calculations demonstrate strong dispersions of the dichroic parameters β_1 for O and individual F atoms in trifluoromethyloxirane, which for some photoelectron energies reach about 9%. For the oxygen K edge, this theoretical result

is in full agreement with the experiment. In order to compare theoretical and experimental results for the fluorine K edge, the computed data were additionally averaged over the three F atoms. This results in a considerable drop of the maximal value of β_1 down to about 2%, which also agrees with the present observations.

The present study provides opportunities for future investigations of the molecular frame [31] photoelectron circular dichroism in trifluoromethyloxirane. As the next step, we plan to extend it towards exploration of transient chirality accompanying fragmentation dynamics by inner-shell pump-inner-shell probe experiments at x-ray free-electron lasers [63,64].

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- [1] B. Ritchie, *Phys. Rev. A* **13**, 1411 (1976).
 [2] N. Böwering, T. Lischke, B. Schmidtke, N. Müller, T. Khalil, and U. Heinzmann, *Phys. Rev. Lett.* **86**, 1187 (2001).
 [3] I. Powis, *Adv. Chem. Phys.* **138**, 267 (2008).
 [4] L. Nahon and I. Powis, in *Chiral Recognition in the Gas Phase*, edited by A. Zehnacker (CRC Press, Boca Raton, FL, 2010), Chap. 1.
 [5] L. Nahon, G. A. Garcia, and I. Powis, *J. Electron Spectrosc. Relat. Phenom.* **204**, 322 (2015).
 [6] G. A. Garcia, L. Nahon, M. Lebech, J. C. Houver, D. Doweck, and I. Powis, *J. Chem. Phys.* **119**, 8781 (2003).
 [7] S. Turchini, N. Zema, G. Contini, G. Alberti, M. Alagia, S. Stranges, G. Fronzoni, M. Stener, P. Decleva, and T. Prosperi, *Phys. Rev. A* **70**, 014502 (2004).
 [8] S. Stranges, S. Turchini, M. Alagia, G. Alberti, G. Contini, P. Decleva, G. Fronzoni, M. Stener, N. Zema, and T. Prosperi, *J. Chem. Phys.* **122**, 244303 (2005).
 [9] G. A. Garcia, L. Nahon, S. Daly, and I. Powis, *Nat. Commun.* **4**, 2132 (2013).
 [10] M. Stener, G. Fronzoni, D. Di Tommaso, and P. Decleva, *J. Chem. Phys.* **120**, 3284 (2004).
 [11] L. Nahon, G. A. Garcia, C. J. Harding, E. Mikajlo, and I. Powis, *J. Chem. Phys.* **125**, 114309 (2006).
 [12] I. Powis, C. J. Harding, G. A. Garcia, and L. Nahon, *ChemPhysChem* **9**, 475 (2008).
 [13] L. Nahon, L. Nag, G. A. Garcia, I. Myrgorodska, U. Meierhenrich, S. Beaulieu, V. Wanie, V. Blanchet, R. Géneaux, and I. Powis, *Phys. Chem. Chem. Phys.* **18**, 12696 (2016).
 [14] L. Nahon, G. A. Garcia, H. Soldi-Lose, S. Daly, and I. Powis, *Phys. Rev. A* **82**, 032514 (2010).
 [15] S. Daly, I. Powis, and G. A. Garcia, *J. Chem. Phys.* **134**, 064306 (2011).
 [16] D. Catone, M. Stener, P. Decleva, G. Contini, N. Zema, T. Prosperi, V. Feyer, K. C. Prince, and S. Turchini, *Phys. Rev. Lett.* **108**, 083001 (2012).
 [17] A. Giardini, D. Catone, S. Stranges, M. Satta, M. Tacconi, S. Piccirillo, S. Turchini, N. Zema, G. Contini, T. Prosperi,

- P. Decleva, D. Di Tommaso, G. Fronzoni, M. Stener, A. Filippi, and M. Speranza, *ChemPhysChem* **6**, 1164 (2005).
- [18] S. Turchini, D. Catone, G. Contini, N. Zema, S. Irrera, M. Stener, D. Di Tommaso, P. Decleva, and T. Prosperi, *ChemPhysChem* **10**, 1839 (2009).
- [19] M. Tia, B. Cunha de Miranda, S. Daly, F. Gaie-Levrel, G. A. Garcia, I. Powis, and L. Nahon, *J. Phys. Chem. Lett.* **4**, 2698 (2013).
- [20] M. Tia, B. Cunha de Miranda, S. Daly, F. Gaie-Levrel, G. A. Garcia, L. Nahon, and I. Powis, *J. Phys. Chem. A* **118**, 2765 (2014).
- [21] I. Powis, *J. Chem. Phys.* **112**, 301 (2000).
- [22] M. Stener, G. Fronzoni, and P. Decleva, *J. Chem. Phys.* **122**, 234301 (2005).
- [23] U. Hergenhahn, E. E. Rennie, O. Kugeler, S. Marburger, T. Lischke, I. Powis, and G. Garcia, *J. Chem. Phys.* **120**, 4553 (2004).
- [24] C. Lux, M. Wollenhaupt, T. Bolze, Q. Liang, J. Kohler, C. Sarpe, and T. Baumert, *Angew. Chem. Int. Ed.* **51**, 5001 (2012).
- [25] C. S. Lehmann, N. B. Ram, I. Powis, and M. H. M. Janssen, *J. Chem. Phys.* **139**, 234307 (2013).
- [26] M. M. Rafiee Fanoood, M. H. M. Janssen, and I. Powis, *J. Chem. Phys.* **145**, 124320 (2016).
- [27] A. Comby, S. Beaulieu, M. Boggio-Pasqua, D. Descamps, F. Légareé, L. Nahon, S. Petit, B. Pons, B. Fabre, Y. Mairesse, and V. Blanchet, *J. Phys. Chem. Lett.* **7**, 4514 (2016).
- [28] A. Kastner, T. Ring, B. C. Krüger, G. B. Park, T. Schäfer, A. Senftleben, and T. Baumert, *J. Chem. Phys.* **147**, 013926 (2017).
- [29] G. Alberti, S. Turchini, G. Contini, N. Zema, T. Prosperi, S. Stranges, V. Feyer, P. Bolognesi, and L. Avaldi, *Phys. Scr.* **78**, 058120 (2008).
- [30] V. Ulrich, S. Barth, S. Joshi, U. Hergenhahn, E. Mikajlo, C. J. Harding, and I. Powis, *J. Phys. Chem. A* **112**, 3544 (2008).
- [31] M. Tia, M. Pitzer, G. Kastirke, J. Gatzke, H.-K. Kim, F. Trinter, J. Rist, A. Hartung, D. Trabert, J. Siebert, K. Henrichs, J. Becht, S. Zeller, H. Gassert, F. Wiegandt, R. Wallauer, A. Kuhlins, C. Schober, T. Bauer, N. Wechselberger, P. Burzynski, J. Neff, M. Weller, D. Metz, M. Kircher, M. Waitz, J. B. Williams, L. Schmidt, A. D. Müller, A. Knie, A. Hans, L. Ben Ltaief, A. Ehresmann, R. Berger, H. Fukuzawa, K. Ueda, H. Schmidt-Böcking, R. Dörner, T. Jahnke, Ph. V. Demekhin, and M. Schöffler, [arXiv:1609.03828](https://arxiv.org/abs/1609.03828).
- [32] T. Jahnke, Th. Weber, A. L. Landers, A. Knapp, S. Schössler, J. Nickles, S. Kammer, O. Jagutzki, L. Schmidt, A. Czasch, T. Osipov, E. Arenholz, A. T. Young, R. Díez Muiño, D. Rolles, F. J. García de Abajo, C. S. Fadley, M. A. Van Hove, S. K. Semenov, N. A. Cherepkov, J. Rösch, M. H. Prior, H. Schmidt-Böcking, C. L. Cocke, and R. Dörner, *Phys. Rev. Lett.* **88**, 073002 (2002).
- [33] G. A. Garcia, H. Dossmann, L. Nahon, S. Daly, and I. Powis, *Chem Phys Chem* **18**, 500 (2017).
- [34] Webpage of the BL13-2 station of SSRL, <http://www-ssrl.slac.stanford.edu/beamlines/bl13-2/>
- [35] T. Katayama, T. Anniyev, M. Beye, R. Coffee, M. Dell'Angela, A. Föhlisch, J. Gladh, S. Kaya, O. Krupin, A. Nilsson, D. Nordlund, W. F. Schlotter, J. A. Sellberg, F. Sorgenfrei, J. J. Turner, W. Wurth, H. Öström, and H. Ogasawara, *J. Electron Spectrosc. Relat. Phenom.* **187**, 9 (2013).
- [36] R. Carr, J. B. Kortright, M. Rice, and S. Lidia, SLAC-PUB-95-6627, <http://slac.stanford.edu/pubs/slacpubs/6500/slac-pub-6627.pdf>
- [37] T. Osipov, D. Rolles, C. Bostedt, J.-C. Castagna, R. Hartmann, J. D. Bozek, I. Schlichting, L. Strüder, J. Ullrich, and N. Berrah, *J. Phys.: Conf. Ser.* **388**, 142025 (2012).
- [38] V. Dribinski, A. Ossadtchi, V. A. Mandelshtam, and H. Reisler, *Rev. Sci. Instrum.* **73**, 2634 (2002).
- [39] M. J. J. Vrakking, *Rev. Sci. Instrum.* **72**, 4084 (2001).
- [40] T. Gerber, Y. Liu, G. Knopp, P. Hemberger, A. Bodi, P. Radi, and Y. Sych, *Rev. Sci. Instrum.* **84**, 033101 (2013).
- [41] E. W. Hansen and P.-L. Law, *J. Opt. Soc. A* **2**, 510 (1985).
- [42] C. Dasch, *Appl. Opt.* **31**, 1146 (1992).
- [43] C. Bordas, F. Paulig, H. Helm, and D. L. Huestis, *Rev. Sci. Instrum.* **67**, 2257 (1998).
- [44] C. E. Rallis, T. G. Burwitz, P. R. Andrews, M. Zohrabi, R. Averin, S. De, B. Bergues, J. Bethany, A. V. Voznyuk, N. Gregerson, B. Gaire, I. Znakovskaya, J. McKenna, K. D. Carnes, M. F. Kling, I. Ben-Itzhak, and E. Wells, *Rev. Sci. Instrum.* **85**, 113105 (2014).
- [45] N. A. Cherepkov, *J. Phys. B* **14**, 2165 (1981).
- [46] N. A. Cherepkov, *Chem. Phys. Lett.* **87**, 344 (1982).
- [47] Ph. V. Demekhin, I. D. Petrov, V. L. Sukhorukov, W. Kielich, P. Reiss, R. Hentges, I. Haar, H. Schmoranzler, and A. Ehresmann, *Phys. Rev. A* **80**, 063425 (2009); **81**, 069902(E) (2010).
- [48] A. Knie, M. Ilchen, Ph. Schmidt, Ph. Reiß, C. Ozga, B. Kambs, A. Hans, N. Müglich, S. A. Galitskiy, L. Glaser, P. Walter, J. Viefhaus, A. Ehresmann, and Ph. V. Demekhin, *Phys. Rev. A* **90**, 013416 (2014).
- [49] Ph. V. Demekhin, A. Ehresmann, and V. L. Sukhorukov, *J. Chem. Phys.* **134**, 024113 (2011).
- [50] S. A. Galitskiy, A. N. Artemyev, K. Jänkälä, B. M. Lagutin, and Ph. V. Demekhin, *J. Chem. Phys.* **142**, 034306 (2015).
- [51] Ph. V. Demekhin, I. D. Petrov, T. Tanaka, M. Hoshino, H. Tanaka, K. Ueda, W. Kielich, and A. Ehresmann, *J. Phys. B* **43**, 065102 (2010).
- [52] Ph. V. Demekhin, I. D. Petrov, V. L. Sukhorukov, W. Kielich, A. Knie, H. Schmoranzler, and A. Ehresmann, *Phys. Rev. Lett.* **104**, 243001 (2010).
- [53] Ph. V. Demekhin, I. D. Petrov, V. L. Sukhorukov, W. Kielich, A. Knie, H. Schmoranzler, and A. Ehresmann, *J. Phys. B* **43**, 165103 (2010).
- [54] H. Sann, C. Schober, A. Mhamdi, F. Trinter, C. Müller, S. K. Semenov, M. Stener, M. Waitz, T. Bauer, R. Wallauer, C. Gohl, J. Titze, F. Afaneh, L. Ph. H. Schmidt, M. Kunitski, H. Schmidt-Böcking, Ph. V. Demekhin, N. A. Cherepkov, M. S. Schöffler, T. Jahnke, and R. Dörner, *Phys. Rev. Lett.* **117**, 263001 (2016).
- [55] E. Antonsson, M. Patanen, C. Nicolas, S. Benkoula, J. J. Neville, V. L. Sukhorukov, J. D. Bozek, Ph. V. Demekhin, and C. Miron, *Phys. Rev. A* **92**, 042506 (2015).
- [56] A. Knie, M. Patanen, A. Hans, I. D. Petrov, J. D. Bozek, A. Ehresmann, and Ph. V. Demekhin, *Phys. Rev. Lett.* **116**, 193002 (2016).
- [57] A. N. Artemyev, A. D. Müller, D. Hochstuhl, and Ph. V. Demekhin, *J. Chem. Phys.* **142**, 244105 (2015).
- [58] M. W. Schmidt, K. K. Baldrige, J. A. Boatz, S. T. Elbert, M. S. Gordon, J. H. Jensen, S. Koseki, N. Matsunaga, K. A. Nguyen, S. Su, T. L. Windus, M. Dupuis, and J. A. Montgomery, Jr., *J. Comput. Chem.* **14**, 1347 (1993).
- [59] <http://classic.chem.msu.su/gran/games/index.html>
- [60] V. L. Sukhorukov, A. N. Hoppersky, I. D. Petrov, V. A. Yavna, and Ph. V. Demekhin, *J. Phys.* **48**, 1677 (1987).

- [61] V. L. Sukhorukov, A. N. Hopersky, and I. D. Petrov, *J. Phys. II France* **1**, 501 (1991).
- [62] U. Arp, B. M. Lagutin, G. Materlik, I. D. Petrov, B. Sonntag, and V. L. Sukhorukov, *J. Phys. B* **26**, 4381 (1993).
- [63] A. A. Lutman, J. P. MacArthur, M. Ilchen, A. O. Lindahl, J. Buck, R. N. Coffee, G. L. Dakovski, L. Dammann, Y. Ding, H. A. Dürr, L. Glaser, J. Grünert, G. Hartmann, N. Hartmann, D. Higley, K. Hirsch, Y. I. Levashov, A. Marinelli, T. Maxwell, A. Mitra, S. Moeller, T. Osipov, F. Peters, M. Planas, I. Shevchuk, W. F. Schlotter, F. Scholz, J. Seltmann, J. Viefhaus, P. Walter, Z. R. Wolf, Z. Huang, and H.-D. Nuhn, *Nat. Photon.* **10**, 468 (2016).
- [64] G. Hartmann, A. O. Lindahl, A. Knie, N. Hartmann, A. A. Lutman, J. P. MacArthur, I. Shevchuk, J. Buck, A. Galler, J. M. Glowonia, W. Helml, Z. Huang, N. M. Kabachnik, A. K. Kazansky, J. Liu, A. Marinelli, T. Mazza, H.-D. Nuhn, P. Walter, J. Viefhaus, M. Meyer, S. Moeller, R. N. Coffee, and M. Ilchen, *Rev. Sci. Instrum.* **87**, 083113 (2016).