Calculation of Stark resonance parameters for valence orbitals of the water molecule

Susana Arias Laso and Marko Horbatsch

Department of Physics and Astronomy, York University, Toronto, Ontario, Canada M3J 1P3 (Received 6 October 2016; published 14 November 2016)

An exterior complex scaling technique is applied to compute Stark resonance parameters for two molecular orbitals $(1b_1$ and $1b_2)$ represented in the field-free limit in a single-center expansion. For electric dc field configurations that guarantee azimuthal symmetry of the solution the calculation is carried out by solving a two-dimensional partial differential equation in spherical polar coordinates using a finite-element method. The resonance positions and widths as a function of electric field strengths are shown for field strengths starting in the tunneling ionization regime and extending well into the over-barrier ionization region.

DOI: [10.1103/PhysRevA.94.053413](https://doi.org/10.1103/PhysRevA.94.053413)

I. INTRODUCTION

The study of water molecule vapor in strong fields has been an area of recent interest, particularly in the field of intermediate-energy ion-molecule collisions [\[1–7\]](#page-4-0), where capture and direct ionization processes compete. Most of the theoretical studies are in the context of an independentelectron approximation which involves a molecular orbital (MO) representation. The first studies of the properties of molecular orbitals exposed to laser pulses have also been performed [\[8,9\]](#page-4-0).

For the problem of strong electric dc fields numerous investigations have been carried out for the hydrogen molecular ion, where interesting structures occur in the resonance width as a function of internuclear separation [\[10–12\]](#page-4-0). This also carries over to the case of low-frequency ac fields, i.e., infrared laser fields, with the help of the Floquet method [\[13\]](#page-4-0). Experimentally, the investigation of water vapor is challenging but possible [\[14–17\]](#page-4-0). Due to the importance of the water molecule in applied fields (e.g., radiation therapy) one should expect more work to be carried out in this research area in the near future.

From the point of view of a theoretical description, the subject is challenging due to the multicenter nature of the combined Coulomb interactions. Thus, many investigations in the context of electron or positron or ion scattering are carried out in single-center approximations for the molecular target. For a number of situations this approach appears to be successful in that ionization (and capture) cross sections are obtained which agree reasonably with experiment. This is in part the case since experiments are usually not sensitive to the orientation of the molecule during the collision.

A satisfactory description of the molecular structure of H_2O was obtained within the independent-electron approximation by the self-consistent field (SCF) or variational Hartree-Fock method with multicenter Slater orbitals [\[18\]](#page-4-0). The direct application of these orbitals for collisional or strong-field studies represents significant computational and methodological challenges. An application of the SCF method using a single-center Slater basis is available [\[19–21\]](#page-4-0) and has been used in some collision studies [\[22\]](#page-4-0). The molecular orbitals for water from [\[21\]](#page-4-0) have been compared to experimental electron spectroscopy studies [\[23\]](#page-4-0) and also to more sophisticated calculations and were found to give reasonable agreement with observations. For further studies of state-of-the-art spectroscopy and calculations we refer the reader to [\[24\]](#page-5-0).

In order to use a variational SCF solution for collisional or strong-field perturbation studies one faces the need to define a consistent molecular Hamiltonian. In the present work we limit ourselves to two of the three molecular valence orbitals of the water molecule where the wave functions are dominated by a single angular momentum symmetry $(1b_1 \approx 2p_x)$ or $1b_2 \approx 2p_y$; here we assume that the protons are in the *y*-*z* plane, as shown in Fig. [1.](#page-1-0) For our approximate treatment it is straightforward to obtain an effective single-electron potential for each orbital. The calculation of Stark resonance parameters for these approximate single-center molecular orbitals represents a first step to be followed by more ambitious modeling to be carried out in the future.

We present a complex scaling approach to study the effect of an external dc field on the $1b_1 \approx 2p_x$ and $1b_2 \approx 2p_y$ MOs of H2O, where the problem is expressed as a system of coupled partial differential equations (PDEs) with an effective potential that reflects the binding properties of each orbital. This paper is organized as follows: The problem as formulated in spherical polar coordinates is introduced in Sec. II, with some technical details concerning the construction of the electronic potential being described in Sec. [II A.](#page-1-0) The exterior complex scaling formalism and the implementation are explained in Sec. [II C.](#page-2-0) The numerical results are presented in Sec. [III,](#page-2-0) followed by conclusions in Sec. [IV.](#page-4-0) Atomic units $(\hbar = m_e = e = 4\pi\epsilon_0 =$ 1) are used throughout.

II. PDE APPROACH TO THE PROBLEM IN SPHERICAL POLAR COORDINATES

The set of single-center wave functions introduced by Moccia [\[19–21\]](#page-4-0) is taken as a starting point in the present approach. The general expression for the basis functions is defined by a Slater-type orbital [\[19\]](#page-4-0),

$$
f_{n,l,m}(\zeta,r,\theta,\phi) = \sqrt{\frac{(2\zeta)^{2n+1}}{(2n)!}} r^{n-1} \exp(-\zeta r) S_{l,m}(\theta,\phi), \quad (1)
$$

where the angular part $S_{l,m}(\theta, \phi)$ represents real spherical harmonics. The expansion coefficients and nonlinear coefficients {*ζi*}, determined by Roothaan's self-consistent-field procedure [\[21](#page-4-0)[,25\]](#page-5-0), were used to construct a reduced form of the radial functions that describe all the molecular orbitals, in particular the $1b_1$ and $1b_2$ $1b_2$ states. In Fig. 2 we depict schematically how the orbitals are approximated as $|m| = 1$ eigenstates of spherically symmetric potentials. For the purpose of this study,

FIG. 1. Schematic display of the $1b_1 \approx 2p_x$ (shown in yellow along the *x* axis) and $1b_2 \approx 2p_y$ (shown in blue along the *y* axis) molecular orbitals. Also indicated (in green on the *y*-*z* plane) is the orientation of the protons. The *z* axis (in red) is the direction of the external electric field of strength *F*0.

the expansion of the orbital functions was truncated at $n = 2$ with their associated (n,l,m) combinations.

The framework for the present study involves the construction of the electronic potential as an effective orbital-dependent potential $V_{\text{eff}}(r)$ extracted from the single-center Moccia wave functions. Then we apply exterior complex scaling (ECS) to determine the numerical solution of the problem associated with a molecular orbital in the presence of a strong electric dc field applied along the *z* direction. A PDE needs to be solved to determine the resonance position and width.

The Schrödinger equation for the bound-state problem is expressed in spherical polar coordinates as

$$
\left[-\frac{1}{2}\frac{d^2}{dr^2} + \frac{\hat{L}^2}{2r^2} + V_{\text{eff}}(r) \right] \psi = E \psi, \tag{2}
$$

where \hat{L}^2 is the orbital angular momentum operator.

Figure 1 shows a scheme of the geometry of the system, where the orientation of the $1b_1$ and $1b_2$ MOs is represented

FIG. 2. Electronic effective potential in atomic units for the $1b_1 \approx$ $2p_x$ (black) and $1b_2 \approx 2p_y$ [blue (gray)] MOs of the H₂O molecule. The solid lines give the potential as derived from (2) using the SCF orbitals and eigenenergies, while the dashed lines show the potentials after the Latter correction is applied. The dot-dashed lines show the eigenenergies obtained from the Moccia wave functions [\[21\]](#page-4-0).

with respect to the plane where the protons are located. The direction of the applied electric field along \hat{z} is included as well.

A. Construction of an electronic effective potential

The effective potential is obtained in two steps: first, the orbital energy and the STO are inserted into the single-electron Schrödinger equation (2), which is then solved for $V_{\text{eff}}^{(1)}(r)$. In the second step we apply the so-called Latter correction [\[26\]](#page-5-0) to ensure that the asymptotic behavior of the potential is correct, i.e., proportional to −1*/r*:

$$
V_{\rm eff}(r) = \begin{cases} V_{\rm eff}^{(1)}(r) & \text{for } r < r_0 \\ -1/r & \text{for } r > r_0 \end{cases}.
$$
 (3)

The point r_0 is determined from $V_{\text{eff}}^{(1)}(r_0) = -1/r_0$ and is found to be sufficiently large that the original self-consistent field orbital energy used to derive $V_{\text{eff}}^{(1)}(r)$ is close to the eigenenergy of (2), with at least two significant digits of agreement, with $V_{\text{eff}}(r)$ given by (3).

In Fig. 2, we show a comparison of the effective potential $V_{\text{eff}}^{(1)}(r)$ (solid lines), with black representing the 1*b*₁ MO and blue (gray) representing the $1b_2$ MO, derived from the Moccia wave functions representing the $1b_1$ and $1b_2$ MOs [\[21\]](#page-4-0), and the transformed electronic potential $V_{\text{eff}}(r)$ (dashed lines) after the Latter correction was implemented. The effective potentials for the $1b_1$ and $1b_2$ MOs are given as the shallower and deeper curves. As Fig. 2 illustrates, one drawback of the method is that the effective potential is orbital dependent. A direct consequence is that the value of $r = r_0$, which sets the position in *r* where the Coulombic tail is imposed, differs between the MOs, being almost two times larger for the $1b_2$ compared to the $1b_1$ MO.

B. H₂O in an external electric dc field

Now that we have obtained an effective potential to define the field-free Schrödinger equation (2) for an orbital obtained in the SCF method, we proceed with the problem of the molecule exposed to a strong dc field.

When an electric field is applied in the \hat{z} direction, $\hat{F} = F_0 \hat{z}$, the separation-of-variables ansatz as applied to the Schrödinger equation

$$
\psi(r,\theta,\phi) = \psi(r,\theta) \exp(im\phi) \tag{4}
$$

leads to a PDE that represents the Stark problem for an H_2O orbital:

$$
-\frac{1}{2}\frac{\partial^2 \psi}{\partial r^2} - \frac{1}{2r^2} \left(\frac{\cos \theta}{\sin \theta} \frac{\partial \psi}{\partial \theta} + \frac{\partial^2 \psi}{\partial \theta^2} \right) + \left[\frac{m^2}{2r^2 \sin^2 \theta} + V_{\text{eff}}(r) - E + F_0 r \cos \theta \right] \psi = 0.
$$
 (5)

Here the complex eigenenergy *E* contains the information about the resonance position (real part), i.e., E_R and width Γ (imaginary part is $-\Gamma/2$), and may be expressed as

$$
E = E_R + iE_I = E_R - i\frac{\Gamma}{2}.
$$
 (6)

The imaginary part Γ is related to the lifetime of the decaying state τ via $\Gamma \tau = 1$. For the 1*b*₁ and 1*b*₂ orbitals we have $m = \pm 1$, respectively. The presence of the effective potential $V_{\text{eff}}(r)$ makes this problem challenging in the sense that it is not possible to obtain separable solutions like for the hydrogen atom in which a pure Coulomb potential leads to separability in parabolic coordinates [\[27\]](#page-5-0). It is then necessary to generate a more general solution by solving the PDE numerically, e.g., by applying a finite-element method.

C. Exterior complex scaling

The ionization regime of the water molecule will be described by means of a non-Hermitian Hamiltonian that reveals discrete resonance eigenvalues containing information about the quasibound states that tunnel through the barrier or escape over the potential barrier for strong fields. Among the different techniques implemented to compute the resonance energies established methods are the complex scaling [\[28–30\]](#page-5-0) and exterior complex scaling [\[31\]](#page-5-0); the latter was introduced as an extension of the former method. These have been widely used in scattering problems [\[29,30\]](#page-5-0) and also in time-dependent Schrödinger equation problems for strong fields $[32-34]$. For our aim of studying the field-ionization properties of H_2O orbitals, we implement a modified exterior complex scaling technique in which the radial coordinates are extended into the complex plane by a phase factor, which is turned on gradually beyond some distance from the origin. This method allows us to address the tunneling and over-barrier ionization problem by avoiding the complication of describing quasibound states with outgoing waves for $r \to \infty$.

In the present work the complex scaling transformation is given by

$$
r \to r \exp[i\chi(r)],\tag{7}
$$

where $\chi(r)$ is defined as a function of the *r* coordinate with the purpose of making the scaling gradually effective from some vicinity of $r = r_s$ on,

$$
\chi(r) = \frac{\chi_s}{1 + \exp\left[-\frac{1}{\Delta r}(r - r_s)\right]}.\tag{8}
$$

For given r_s one has to choose Δr to be sufficiently small, so that the function $\chi(r)$ starts from small values at $r = 0$. For large *r* it reaches the value *χ*s.

The set of possible values for the asymptotic scaling angle χ_s and the parameters r_s and Δr , which control where and how quickly the scaling is turned on, was explored in detail in order to establish how sensitive the PDE solutions are and to test the effectiveness of the complex scaling technique to absorb the outgoing wave. Numerous tests were carried out to ensure that the "exact" results of Telnov [\[27\]](#page-5-0) for atomic hydrogen orbitals including 2*p* were reproduced.

In order to investigate the effects of the dc field on the H_2O orbital energies it is necessary to consider the extra terms that the exterior complex scaling (7) introduces in the Schrödinger equation [\(5\)](#page-1-0). Additionally, we need to turn the scaling on only in the regime $r > r_0$ [Eq. [\(3\)](#page-1-0)], such that we have a simple Coulomb potential in the scaling region. In order to make use of standard finite-element methods, the complex-valued wave function is separated into real and imaginary parts, such that a system of coupled differential equations is obtained as follows:

$$
-\frac{1}{2}\frac{\partial^2 \psi_R}{\partial r^2} - \frac{1}{2r^2} \left(\frac{\cos \theta}{\sin \theta} \frac{\partial \psi_R}{\partial \theta} + \frac{\partial^2 \psi_R}{\partial \theta^2} \right) + \left[\frac{m^2}{2r^2 \sin^2 \theta} + V_{\text{eff}}^R(r)c_2 - V_{\text{eff}}^I(r)s_2 - E_R c_2 + E_I s_2 + F_0 r \cos \theta c_3 \right] \psi_R + \left[-V_{\text{eff}}^R(r)s_2 - V_{\text{eff}}^I(r)c_2 + E_R s_2 + E_I c_2 - F_0 r \cos \theta s_3 \right] \psi_I = 0, -\frac{1}{2}\frac{\partial^2 \psi_I}{\partial r^2} - \frac{1}{2r^2} \left(\frac{\cos \theta}{\sin \theta} \frac{\partial \psi_I}{\partial \theta} + \frac{\partial^2 \psi_I}{\partial \theta^2} \right) + \left[\frac{m^2}{2r^2 \sin^2 \theta} + V_{\text{eff}}^R(r)c_2 - V_{\text{eff}}^I(r)s_2 - E_R c_2 + E_I s_2 + F_0 r \cos \theta c_3 \right] \psi_I + \left[V_{\text{eff}}^R(r)s_2 + V_{\text{eff}}^I(r)c_2 - E_R s_2 - E_I c_2 + F_0 r \cos \theta s_3 \right] \psi_R = 0.
$$
 (9)

The labels *R* and *I* stand for real and imaginary parts, respectively; also, the notation $\{c_k, s_k\}$ is introduced to represent $\{\cos[k\chi(r)], \sin[k\chi(r)]\}$, respectively, with $k = 2,3$ and $\chi(r)$ given in (8). Note that the effective potential has real and imaginary parts on account of the coordinate transformation (7).

The PDE system (9) is solved numerically on a rectangular mesh defined by the (r, θ) coordinates, which take values in the domains $[\epsilon, r_{\text{max}}]$ and $[\eta, \pi - \eta]$, respectively. The parameters ϵ and η that define the coordinate ranges were chosen to be of the order of 10^{-3} a.u., and the *r* coordinate extends to $r_{\text{max}} = 20$ a.u. In order to find a correct set of $\psi_{R(I)}(r,\theta)$ solutions it is essential to impose proper boundary conditions that ensure the wave functions vanish at the limits of the mesh. For the $|m| = 1$ states we impose the condition $\psi_{R(I)}(\epsilon, \theta) = \epsilon \sin(\theta) = \epsilon P_1^1(\theta)$, which is consistent with the assumption that at small $r = \epsilon$ the lowest term in an expansion in associated Legendre polynomials dominates and behaves like $Ar^2 \sin(\theta)$.

We implemented a two-parameter root search for ${E_R, E_I}$ by solving the PDE as if it were an inhomogeneous problem. We pick a location in the (r, θ) plane where the probability amplitude is expected to be large and vary ${E_R, E_I}$, i.e., effectively the complex energy *E* to maximize the amplitude.

III. STARK RESONANCE PARAMETERS

We explored the influence of a set of parameters involved in the two-dimensional problem (9) on the eigenvalue $E_R + iE_I$ which describes the ionization process as an exponential decay in time in terms of the resonance position and half width. In addition to testing the code against known results for atomic hydrogen [\[27\]](#page-5-0), we have performed systematic studies of our results for the H_2O orbitals against a number

FIG. 3. Resonance position (in atomic units) as a function of the external field strength F_0 (in atomic units) for the $1b_1$ (red circles) and $1b_2$ (blue triangles) MOs of H_2O .

of parameters in order to assess the accuracy of these results. One parameter concerns the limiting resolution with which the finite-element method proceeds (the MAXCELLSIZE parameter in the *Mathematica* 10 implementation of NDSOLVE, which we call Δ). For values $\Delta < 0.02$ a.u. we find stability in the eigenvalues (real and imaginary parts) of two to three significant digits. For the results quoted below we then applied the more stringent criterion of $\Delta = 0.01$ a.u.

The second parameter which was investigated is the range where the complex scaling function sets in, i.e., r_s and Δr in Eq. [\(8\)](#page-2-0). For the scaling method to work we require scaling to set in for $r > r_0$ when the effective potential represents a simple Coulomb tail, which in practice is satisfied by $r_s > 2r_0$. We also need $\Delta r < 2$ a.u. to guarantee smooth turn-on in this region. Small values of Δr pose challenges for the automated finite-element method since in the limit of $\Delta r \rightarrow 0$ one would need to implement the derivative discontinuity in the solution as discussed by Scrinzi [\[34\]](#page-5-0). We find stable results for the real and imaginary parts of the eigenenergies at the level of three significant digits for the range $10 < r_s < 15$ a.u. Larger values would require an increase in the computational domain beyond $r = 20$ a.u.

Finally, another systematic that was explored is the choice of the ultimate scaling angle reached at large *r*, namely, the value of χ _s in [\(8\)](#page-2-0). For an accuracy demand of three significant digits and the other parameters chosen in the ranges described above stability in the resonance widths is achieved for 0*.*6 *< χ*^s *<* 1*.*2 rad.

In Fig. 3 we show the resonance position E_R as obtained from the present calculations for the weakly bound $1b_1$ and the strongly bound $1b_2$ valence orbitals as a function of applied electric field strength F_0 . In the limit of zero field the calculation reproduces the SCF eigenvalues of Moccia [\[21\]](#page-4-0). The field has to be strong (in comparison with atomic hydrogen results for 2*p* orbitals [\[27\]](#page-5-0)) in order to change the resonance position appreciably. For the more deeply bound $1b_2$ orbital the shift in resonance position saturates with field strength.

In Fig. 4 the resonance widths are shown for both orbitals as functions of external field strength $F₀$. The graphs display threshold behavior at the weaker field strengths. As expected

FIG. 4. Resonance width (in atomic units) as a function of the external field strength F_0 (in atomic units) for the $1b_1$ (red circles) and $1b_2$ (blue triangles) MOs of H_2O . For comparison, atomic hydrogen H(1*s*) ionization rates from Refs. [\[27,35\]](#page-5-0) are shown as crosses.

we find a lower threshold (critical field strength) for the more weakly bound $1b_1$ orbital. Interestingly, however, at a field strength of about $F_0 = 0.3$ a.u. the values for the widths cross; that is, the deeper bound $1b_2$ orbital displays a larger ionization rate as the field strength is increased further.

Also shown in Fig. 4 are the widths for the H(1*s*) orbital from Refs. $[27,35]$. They can be compared to the $1b_1$ orbital results since the binding energy is very close in the free-field limit. Since the tunneling barrier is mostly in the asymptotic regime where the potential energy has a $-1/r$ tail, it is not surprising that the widths for $H_2O(1b_1)$ and $H(1s)$ share some similarity in shape. In the tunneling region H(1*s*) has an ionization rate that is larger by about an order of magnitude. In the over-barrier regime, however, the ionization rates come to within a factor of 3. Reasons for why the $1b_1$ water molecular orbital is harder to ionize than H(1*s*) have to do with the different shapes of the orbital density $[m = 1]$ vs the spherical H(1*s*) density] and the substantially more attractive potential at shorter distances.

An examination of contour plots of the densities $\Psi^*\Psi$, as well as of the potential energies $V_{\text{eff}}(r) - F_0 z$ for different field strengths (both as a function of r, θ), allows us to make the following observations. For field strengths $F_0 < 0.1$ a.u. there is a barrier the electrons need to penetrate in order to be ionized. From the potential-energy plot shown in Fig. 3 one can see that for weak fields (small values of F_0) the barrier is longer for the more deeply bound $1b_2$ orbital. This explains why the ionization threshold occurs for $F_0 > 0.1$ a.u. for this orbital, which is about a factor of 2 larger than for the $1b_1$ orbital.

The field-strength region where the ionization rates (resonance widths) display a change in character, i.e., turn over to rise much more gradually with the field strength F_0 , can be characterized as a regime where there is a narrow potential saddle at small θ in the vicinity of $r \approx 3$ a.u., such that electron flux can leave and is then accelerated by the electric field. The crossing of the ionization rates for the $1b_1$ and $1b_2$ orbitals occurs since the saddle in the potential becomes effectively lower at strong fields for the 1*b*² orbital. This can be inferred from the comparison of the two effective potentials, which share the same asymptotic behavior beyond $r = 4.3$ a.u. (see Fig. [2\)](#page-1-0).

TABLE I. Resonance positions and widths for different field strengths (in atomic units). The numbers in parentheses indicate the exponent *k*; that is, the numbers are to be multiplied by 10^k .

F_0	$1b_1$		1b ₂	
	E_R	г	E_R	г
0.1	-0.506	$1.14(-3)$	-0.689	$4.04(-5)$
0.2	-0.525	$2.28(-2)$	-0.718	$1.23(-2)$
0.3	-0.546	$6.74(-2)$	-0.760	$7.51(-2)$
0.4	-0.564	$1.24(-1)$	-0.790	$1.91(-1)$
0.5	-0.580	$1.90(-1)$	-0.796	$3.11(-1)$
0.6	-0.593	$2.61(-1)$	-0.797	$4.11(-1)$

The origin for the different radial dependencies of the effective potential for the two orbitals can be found in the geometry of the water molecule. The weakly bound $1b_1$ orbital has its lobes perpendicular to the plane defined by the location of the three nuclei. Therefore, it is least affected by the two protons. The $1b_2$ orbital explores the potentials due to the protons more strongly in the SCF calculation of Moccia, and therefore, the resulting (in our approximation spherically symmetric) $V_{\text{eff}}(r)$ has a more attractive region in the range $0.7 < r < 4.3$.

We summarize the results in Table I for further reference, i.e., for future comparison with calculations based on other models for the molecular orbitals.

IV. CONCLUSION

We have carried out a study of two of the three valence orbitals of the water molecule, $1b_1$ and $1b_2$, in the presence of an external electric dc field. The tunneling ionization and over-barrier ionization regimes were explored by finding a numerical solution to the PDE system defined by an effective potential obtained from single-center Slater-type orbitals. The exterior complex scaling parameters and a finite-element resolution parameter were optimized to guarantee a minimum of two to three significant digits for the solutions. The resonance parameters that describe the ionization process, resonance position, and width were explored over a wide range of electric field strengths. We demonstrated how an increase of the field strength beyond a critical point in the over-barrier region led to a crossing between the ionization rates of the two orbitals. Additional observations of the effective potential for different field strengths were carried out to shed some light on the interpretation of this behavior.

For practical applications this work needs to be extended to deal with ac (laser) fields. For the hydrogen molecular ion this was done using Floquet theory [13], where some parallels were found between the monochromatic ac and the dc cases. Experimental observations of strong-field ionization of water vapor do exist for short, intense laser pulses [\[36,37\]](#page-5-0). One needs to solve the time-dependent Schrödinger equation for realistic simulations of these [8,9]. For far-infrared fields, however, the Floquet approach can yield results that are not too different from time-dependent solutions involving pulses with a moderate number of laser cycles [\[38\]](#page-5-0).

ACKNOWLEDGMENT

The financial support from NSERC of Canada is gratefully acknowledged.

- [1] M. Murakami, T. Kirchner, M. Horbatsch, and H. J. Lüdde, [Phys. Rev. A](https://doi.org/10.1103/PhysRevA.85.052713) **[85](https://doi.org/10.1103/PhysRevA.85.052713)**, [052713](https://doi.org/10.1103/PhysRevA.85.052713) [\(2012\)](https://doi.org/10.1103/PhysRevA.85.052713).
- [2] M. Murakami, T. Kirchner, M. Horbatsch, and H. J. Lüdde, [Phys. Rev. A](https://doi.org/10.1103/PhysRevA.86.022719) **[86](https://doi.org/10.1103/PhysRevA.86.022719)**, [022719](https://doi.org/10.1103/PhysRevA.86.022719) [\(2012\)](https://doi.org/10.1103/PhysRevA.86.022719).
- [3] H. Luna, W. Wolff, E. C. Montenegro, A. C. Tavares, H. J. Lüdde, G. Schenk, M. Horbatsch, and T. Kirchner, Phys. Rev. A **[93](https://doi.org/10.1103/PhysRevA.93.052705)**, [052705](https://doi.org/10.1103/PhysRevA.93.052705) [\(2016\)](https://doi.org/10.1103/PhysRevA.93.052705).
- [4] L. Gulyás, S. Egri, H. Ghavaminia, and A. Igarashi, *Phys. Rev.* A **[93](https://doi.org/10.1103/PhysRevA.93.032704)**, [032704](https://doi.org/10.1103/PhysRevA.93.032704) [\(2016\)](https://doi.org/10.1103/PhysRevA.93.032704).
- [5] X. Hong, F. Wang, Y. Wu, B. Gou, and J. Wang, *[Phys. Rev. A](https://doi.org/10.1103/PhysRevA.93.062706)* **[93](https://doi.org/10.1103/PhysRevA.93.062706)**, [062706](https://doi.org/10.1103/PhysRevA.93.062706) [\(2016\)](https://doi.org/10.1103/PhysRevA.93.062706).
- [6] L. F. Errea, C. Illescas, L. Méndez, and I. Rabadán, *Phys. Rev.* A **[87](https://doi.org/10.1103/PhysRevA.87.032709)**, [032709](https://doi.org/10.1103/PhysRevA.87.032709) [\(2013\)](https://doi.org/10.1103/PhysRevA.87.032709).
- [7] S. Nandi, S. Biswas, A. Khan, J. M. Monti, C. A. Tachino, R. D. Rivarola, D. Misra, and L. C. Tribedi, [Phys. Rev. A](https://doi.org/10.1103/PhysRevA.87.052710) **[87](https://doi.org/10.1103/PhysRevA.87.052710)**, [052710](https://doi.org/10.1103/PhysRevA.87.052710) [\(2013\)](https://doi.org/10.1103/PhysRevA.87.052710).
- [8] J. P. Farrell, S. Petretti, J. Förster, B. K. McFarland, L. S. Spector, Y. V. Vanne, P. Decleva, P. H. Bucksbaum, A. Saenz, and M. Gühr, *[Phys. Rev. Lett.](https://doi.org/10.1103/PhysRevLett.107.083001)* **[107](https://doi.org/10.1103/PhysRevLett.107.083001)**, [083001](https://doi.org/10.1103/PhysRevLett.107.083001) [\(2011\)](https://doi.org/10.1103/PhysRevLett.107.083001).
- [9] M. Falge, V. Engel, and M. Lein, [Phys. Rev. A](https://doi.org/10.1103/PhysRevA.81.023412) **[81](https://doi.org/10.1103/PhysRevA.81.023412)**, [023412](https://doi.org/10.1103/PhysRevA.81.023412) [\(2010\)](https://doi.org/10.1103/PhysRevA.81.023412).
- [10] T. Tsogbayar and M. Horbatsch, [J. Phys. B](https://doi.org/10.1088/0953-4075/46/8/085004) **[46](https://doi.org/10.1088/0953-4075/46/8/085004)**, [085004](https://doi.org/10.1088/0953-4075/46/8/085004) [\(2013\)](https://doi.org/10.1088/0953-4075/46/8/085004).
- [11] X. Chu and Shih-I Chu, [Phys. Rev. A](https://doi.org/10.1103/PhysRevA.63.013414) **[63](https://doi.org/10.1103/PhysRevA.63.013414)**, [013414](https://doi.org/10.1103/PhysRevA.63.013414) [\(2000\)](https://doi.org/10.1103/PhysRevA.63.013414).
- [12] M. Plummer and J. F. McCann, [J. Phys. B](https://doi.org/10.1088/0953-4075/29/20/017) **[29](https://doi.org/10.1088/0953-4075/29/20/017)**, [4625](https://doi.org/10.1088/0953-4075/29/20/017) [\(1996\)](https://doi.org/10.1088/0953-4075/29/20/017).
- [13] T. Tsogbayar and M. Horbatsch, [J. Phys. B](https://doi.org/10.1088/0953-4075/46/24/245005) **[46](https://doi.org/10.1088/0953-4075/46/24/245005)**, [245005](https://doi.org/10.1088/0953-4075/46/24/245005) [\(2013\)](https://doi.org/10.1088/0953-4075/46/24/245005).
- [14] [U. Werner, K. Beckord, J. Becker, and H. O. Lutz,](https://doi.org/10.1103/PhysRevLett.74.1962) *Phys. Rev.* Lett. **[74](https://doi.org/10.1103/PhysRevLett.74.1962)**, [1962](https://doi.org/10.1103/PhysRevLett.74.1962) [\(1995\)](https://doi.org/10.1103/PhysRevLett.74.1962).
- [15] F. Gobet, B. Farizon, M. Farizon, M. J. Gaillard, M. Carre, M. ´ Lezius, P. Scheier, and T. D. Märk, [Phys. Rev. Lett.](https://doi.org/10.1103/PhysRevLett.86.3751) [86](https://doi.org/10.1103/PhysRevLett.86.3751), [3751](https://doi.org/10.1103/PhysRevLett.86.3751) [\(2001\)](https://doi.org/10.1103/PhysRevLett.86.3751).
- [16] F. Gobet, S. Eden, B. Coupier, J. Tabet, B. Farizon, M. Farizon, M. J. Gaillard, M. Carré, S. Ouaskit, T. D. Märk, and P. Scheier, [Phys. Rev. A](https://doi.org/10.1103/PhysRevA.70.062716) **[70](https://doi.org/10.1103/PhysRevA.70.062716)**, [062716](https://doi.org/10.1103/PhysRevA.70.062716) [\(2004\)](https://doi.org/10.1103/PhysRevA.70.062716).
- [17] H. Luna, A. L. F. de Barros, J. A. Wyer, S. W. J. Scully, J. Lecointre, P. M. Y. Garcia, G. M. Sigaud, A. C. F. Santos, V. [Senthil, M. B. Shah, C. J. Latimer, and E. C. Montenegro,](https://doi.org/10.1103/PhysRevA.75.042711) Phys. Rev. A **[75](https://doi.org/10.1103/PhysRevA.75.042711)**, [042711](https://doi.org/10.1103/PhysRevA.75.042711) [\(2007\)](https://doi.org/10.1103/PhysRevA.75.042711).
- [18] S. Aung, R. M. Pitzer, and S. I. Chan, [J. Chem. Phys.](https://doi.org/10.1063/1.1670368) **[49](https://doi.org/10.1063/1.1670368)**, [2071](https://doi.org/10.1063/1.1670368) [\(1968\)](https://doi.org/10.1063/1.1670368).
- [19] R. Moccia, [J. Chem. Phys.](https://doi.org/10.1063/1.1725489) **[40](https://doi.org/10.1063/1.1725489)**, [2164](https://doi.org/10.1063/1.1725489) [\(1964\)](https://doi.org/10.1063/1.1725489).
- [20] R. Moccia, [J. Chem. Phys.](https://doi.org/10.1063/1.1725490) **[40](https://doi.org/10.1063/1.1725490)**, [2176](https://doi.org/10.1063/1.1725490) [\(1964\)](https://doi.org/10.1063/1.1725490).
- [21] R. Moccia, [J. Chem. Phys.](https://doi.org/10.1063/1.1725491) **[40](https://doi.org/10.1063/1.1725491)**, [2186](https://doi.org/10.1063/1.1725491) [\(1964\)](https://doi.org/10.1063/1.1725491).
- [22] C. C. Montanari and J. E. Miraglia,[J. Phys. B](https://doi.org/10.1088/0953-4075/47/1/015201)**[47](https://doi.org/10.1088/0953-4075/47/1/015201)**, [015201](https://doi.org/10.1088/0953-4075/47/1/015201) [\(2014\)](https://doi.org/10.1088/0953-4075/47/1/015201).
- [23] H. Hafied, A. Eschenbrenner, C. Champion, M. F. Ruiz-López, [C. D. Cappello, I. Charpentier, and P. A. Hervieux,](https://doi.org/10.1016/j.cplett.2007.03.067) Chem. Phys. Lett. **[439](https://doi.org/10.1016/j.cplett.2007.03.067)**, [55](https://doi.org/10.1016/j.cplett.2007.03.067) [\(2007\)](https://doi.org/10.1016/j.cplett.2007.03.067).
- [24] C. G. Ning, B. Hajgato, Y. R. Huang, S. F. Zhang, K. Liu, Z. H. ´ [Luo, S. Knippenberg, J. K. Deng, and M. S. Deleuze,](https://doi.org/10.1016/j.chemphys.2007.09.030) Chem. Phys. **[343](https://doi.org/10.1016/j.chemphys.2007.09.030)**, [19](https://doi.org/10.1016/j.chemphys.2007.09.030) [\(2008\)](https://doi.org/10.1016/j.chemphys.2007.09.030).
- [25] C. C. J. Roothaan, [Rev. Mod. Phys.](https://doi.org/10.1103/RevModPhys.23.69) **[23](https://doi.org/10.1103/RevModPhys.23.69)**, [69](https://doi.org/10.1103/RevModPhys.23.69) [\(1951\)](https://doi.org/10.1103/RevModPhys.23.69).
- [26] R. Latter, [Phys. Rev.](https://doi.org/10.1103/PhysRev.99.510) **[99](https://doi.org/10.1103/PhysRev.99.510)**, [510](https://doi.org/10.1103/PhysRev.99.510) [\(1955\)](https://doi.org/10.1103/PhysRev.99.510).
- [27] D. A. Telnov, [J. Phys. B](https://doi.org/10.1088/0953-4075/22/14/002) **[22](https://doi.org/10.1088/0953-4075/22/14/002)**, [L399](https://doi.org/10.1088/0953-4075/22/14/002) [\(1989\)](https://doi.org/10.1088/0953-4075/22/14/002).
- [28] J. Aguilar and J. M. Combes, [Commun. Math. Phys.](https://doi.org/10.1007/BF01877510) **[22](https://doi.org/10.1007/BF01877510)**, [269](https://doi.org/10.1007/BF01877510) [\(1971\)](https://doi.org/10.1007/BF01877510).
- [29] E. Baslev and J. M. Combes, [Commun. Math. Phys.](https://doi.org/10.1007/BF01877511) **[22](https://doi.org/10.1007/BF01877511)**, [280](https://doi.org/10.1007/BF01877511) [\(1971\)](https://doi.org/10.1007/BF01877511).
- [30] B. Simon, [Ann. Math.](https://doi.org/10.2307/1970847) **[97](https://doi.org/10.2307/1970847)**, [247](https://doi.org/10.2307/1970847) [\(1973\)](https://doi.org/10.2307/1970847).
- [31] B. Simon, [Phys. Lett. A](https://doi.org/10.1016/0375-9601(79)90165-8) **[71](https://doi.org/10.1016/0375-9601(79)90165-8)**, [211](https://doi.org/10.1016/0375-9601(79)90165-8) [\(1979\)](https://doi.org/10.1016/0375-9601(79)90165-8).
- [32] F. He, C. Ruiz, and A. Becker, [Phys. Rev. A](https://doi.org/10.1103/PhysRevA.75.053407) **[75](https://doi.org/10.1103/PhysRevA.75.053407)**, [053407](https://doi.org/10.1103/PhysRevA.75.053407) [\(2007\)](https://doi.org/10.1103/PhysRevA.75.053407).
- [33] L. Tao, W. Vanroose, B. Reps, T. N. Rescigno, and C. W. McCurdy, [Phys. Rev. A](https://doi.org/10.1103/PhysRevA.80.063419) **[80](https://doi.org/10.1103/PhysRevA.80.063419)**, [063419](https://doi.org/10.1103/PhysRevA.80.063419) [\(2009\)](https://doi.org/10.1103/PhysRevA.80.063419).
- [34] A. Scrinzi, [Phys. Rev. A](https://doi.org/10.1103/PhysRevA.81.053845) **[81](https://doi.org/10.1103/PhysRevA.81.053845)**, [053845](https://doi.org/10.1103/PhysRevA.81.053845) [\(2010\)](https://doi.org/10.1103/PhysRevA.81.053845).
- [35] V. V. Kolosov, [J. Phys. B](https://doi.org/10.1088/0022-3700/20/11/008) **[20](https://doi.org/10.1088/0022-3700/20/11/008)**, [2359](https://doi.org/10.1088/0022-3700/20/11/008) [\(1987\)](https://doi.org/10.1088/0022-3700/20/11/008).
- [36] D. Mathur, F. A. Rajgara, A. K. Dharmadhikari, and J. A. Dharmadhikari, [Phys. Rev. A](https://doi.org/10.1103/PhysRevA.78.023414) **[78](https://doi.org/10.1103/PhysRevA.78.023414)**, [023414](https://doi.org/10.1103/PhysRevA.78.023414) [\(2008\)](https://doi.org/10.1103/PhysRevA.78.023414).
- [37] M. Wilke, R. Al-Obaidi, A. Moguilevski, A. Kothe, N. Engel, J. Metje, I. Y. Kiyan, and E. F. Aziz, [New J. Phys.](https://doi.org/10.1088/1367-2630/16/8/083032) **[16](https://doi.org/10.1088/1367-2630/16/8/083032)**, [083032](https://doi.org/10.1088/1367-2630/16/8/083032) [\(2014\)](https://doi.org/10.1088/1367-2630/16/8/083032).
- [38] T. Tsogbayar and M. Horbatsch, [J. Phys. B](https://doi.org/10.1088/0953-4075/47/11/115003) **[47](https://doi.org/10.1088/0953-4075/47/11/115003)**, [115003](https://doi.org/10.1088/0953-4075/47/11/115003) [\(2014\)](https://doi.org/10.1088/0953-4075/47/11/115003).