## Lyman-series emission after valence and core excitation of water vapor

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We report Lyman-series emission cross sections of neutral hydrogen dissociation fragments after valence (15–34 eV) and inner-shell (533–542 eV) excitation of water vapor with monochromatic synchrotron radiation as functions of the exciting-photon energy. In the valence excitation energy region, the thermodynamical limits of the production of the differently excited hydrogen fragments are directly observed and absolute emission cross sections are determined. For resonant inner-shell excitations, the fluorescing excited hydrogen state is found to be strongly dependent on the molecular or Rydberg-like character of the excitation.

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## I. INTRODUCTION

Because of its unique role in nature, the structure and dynamics of the water molecule have always been intensely investigated in molecular physics. Processes following photon irradiation are of particular interest in atmospheric physics and astrophysics, radiation biology, and fundamental photochemistry [1]. Depending on the incident photon energy, photodissociation, photoexcitation, or photoionization may occur, and the latter two processes are possibly followed or accompanied by photodissociation into neutral or ionic fragments. Recent progress in experimental techniques on liquid jets in high vacuum [2,3] and theoretical and experimental investigations on water clusters [4–8] call for a complete understanding of the decay dynamics of the isolated molecule.

The electronic configuration of the water molecule in its ground state is

$$1a_1^2 2a_1^2 1b_2^2 3a_1^2 1b_1^2 ({}^1A_1)$$

with an outer-valence ionization potential (of the  $1b_1$  orbital) of 12.62 eV and binding energies of the inner-valence orbitals  $2a_1$ ,  $1b_2$ , and  $3a_1$  of 32.6, 18.6, and 14.8 eV, respectively [9]. The core orbital  $1a_1$  is basically of O 1s character and has an ionization potential of 539.8 eV [10]. In the outer-valence energy region, an overview of absolute absorption cross sections is given in Refs. [11–13], showing a complex structure of Rydberg series excitations for all valence orbitals.

Whereas processes with electron emission (or ion formation) are relatively easy to investigate by guiding the charged particles with suitable electric or magnetic fields to a detector, neutral fragments are much more difficult to detect. If these fragments are excited, however, fluorescence spectrometry can be used to detect their relaxation. This technique is particularly powerful when combined with excitation by monochromatized synchrotron radiation. This enables one to extract absolute and, if dispersed fluorescence is measured, even final dissociationstate selective cross sections [14–21].

Reference [22] gives an overview of energetic formation limits for excited fragments after excitation of water molecules. These thermodynamical limits can be calculated as the sum of the dissociation energy and the respective energy stored in the excited fragment. Below and slightly above the first ionization threshold, the only observable fluorescence stems from OH\* transitions in the ultraviolet  $(A \rightarrow X, C \rightarrow X, C \rightarrow A)$  and visible  $(B \rightarrow A)$  range and from H<sub>2</sub>O<sup>+</sup>  $(A \rightarrow X)$  in the visible range [22]. The onset of fluorescence from neutral hydrogen atoms at 15.3 eV is the sum of the OH + H dissociation energy (5.1 eV) and the energy of the lowest electronic excitation of ground-state hydrogen, which corresponds to the L<sub> $\alpha$ </sub> transition from n = 2 to n = 1at 10.2 eV (corresponding to 121.57 nm). An overview of the thermodynamical production limits of excited hydrogen after dissociation following photoexcitation and subsequent fluorescence decays by Lyman or Balmer emission is given in Table I.

Fluorescence emission from neutral hydrogen after photoexcitation of water molecules was also the subject of several earlier investigations. Using synchrotron radiation, the absolute emission cross sections of  $L_{\alpha}$  fluorescence in the valence excitation energy range from 15 to 40 eV were determined more than 30 years ago [23]. Emission spectra and emission cross sections of  $L_{\alpha}$  and the first three Balmer transitions  $(H_{\alpha}, H_{\beta}, H_{\gamma})$  for some exciting-photon energies were measured in early experiments [24,25] in the excitation range from 15 to 22 eV, and a discussion of dissociation processes leading to Lyman- and Balmer-series emission is given. No formation of excited hydrogen fragments for exciting-photon energies above 22 eV was observed in these early experiments. More recently,  $L_{\alpha}$  emission on a relative scale was reported up to 60 eV [26] and the resonances leading to the formation of excited hydrogen were discussed [27]. Relative emission functions and absolute cross sections are also available for the Balmer series up to 40 eV [22,28,29].

In the exciting-photon energy region of core excitations, dissociation processes following Auger decays and leading to fluorescence are discussed in detail in Ref. [30]. There the fluorescence excitation function was measured for  $L_{\alpha}$  and some emission features of OH, OH<sup>+</sup>, O, and O<sup>+</sup>. It was found to resemble the general molecular excitation pattern, mirroring the unoccupied orbital structure of the water molecule with changing intensity ratios for certain excitations. By detecting the Balmer-series ( $\alpha - \delta$ ) emission functions, the production functions of higher excited hydrogen atoms ( $n \ge 3$ ) have been determined [20]. For the molecular resonances  $4a_1$ 

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TABLE I. Thermodynamical production limits *L* of excited hydrogen fragments after dissociation of water molecules into  $OH(X^2\Pi) + H(n)$ , the denotation  $L_x$ ,  $H_x$  of the subsequent Lyman and Balmer decays, and the corresponding fluorescence wavelengths  $\lambda_{fl}$ .

n	L (eV)	Lyman		Balmer	
		$\overline{L_x}$	$\lambda_{fl}$ (nm)	$\overline{\mathbf{H}_{x}}$	$\lambda_{fl}$ (nm)
1	5.1				
2	15.3	$L_{\alpha}$	121.6		
3	17.2	$L_{\beta}$	102.6	$H_{\alpha}$	656.3
4	17.9	$L_{\nu}^{r}$	97.3	$H_{\beta}$	486.1
5	18.2	Ĺδ	95.0	$H_{\nu}$	434.1
6	18.3	$L_{\varepsilon}$	93.8	$\dot{H_{\delta}}$	410.2
$\infty$	18.7		91.2		364.6

and  $2b_2$ , it has been suggested that the main channel for the production of excited hydrogen dissociation fragments is  $H_2O^+ \rightarrow H(n \leq 3) + OH^+$ , where the initial water cation is formed by an Auger decay of the resonance. For higher, Rydberg-like excitations, the Rydberg character was found to be conserved during the (resonant) Auger decay and the dissociation, leading to high-*n* excited hydrogen fragments.

However, these experiments and those carried out in the valence excitation range do not provide complete information about the production of excited hydrogen atoms. For all  $n \ge 3$ states, the branching ratios for the different decay paths have not been experimentally determined because only the Balmer series was observed. Measuring the Lyman series is experimentally difficult because there is no transparent material in this photon energy region (except for  $L_{\alpha}$ ), requiring differential pumping stages. In the present work, we will add experimental information about Lyman-series emission cross sections. Using dispersed photon-induced fluorescence spectrometry [31], absolute fluorescence emission cross sections not only for the  $L_{\alpha}$  emission but also for the Lyman-series  $(\alpha - \delta)$  lines in the valence excitation range are determined. The Lyman-Balmer branching ratios are computed and compared to theory. For the same lines, relative emission functions after inner-shell excitation of water close to the O 1s-electron ionization threshold are given.

### **II. EXPERIMENT**

Owing to the large difference between the exciting-photon energy ranges for valence and inner-shell excitations, two experiments were performed separately. Both of them were carried out with an established setup for photon-induced fluorescence spectrometry (PIFS) [31] using a commercial McPherson 1 m normal-incidence spectrometer, equipped with a gold-coated 1200 lines/mm grating to disperse the fluorescence. In the low-energy region, the measurements were performed at the 10 m NIM beam line of undulator U125-2 of the Helmholtz-Zentrum Berlin (HZB), i.e., BESSY II [32]. Two energy regions of the linear horizontally polarized synchrotron radiation were scanned by varying the photon energy stepwise: between 15 and 22 eV, in 25 meV steps, in first order of the beam-line monochromator and between 21 and 34 eV, in 100 meV steps, in second order to optimize the photon flux in the respective energy region. With an exit slit width of 200  $\mu$ m, a photon bandwidth of 20 meV was achieved at 19 eV in first order and at 25 eV in second order. A target cell with pinholes for the synchrotron beam and a 2 mm slit towards the fluorescence monochromator was used. Inside the target cell, a static pressure of 0.2 mbar was maintained using an inlet valve with a reservoir pressure of about 32 mbar (vapor pressure of water at about 25 °C). The water was degassed by repeated freeze and thaw cycles.

In the high exciting-photon energy region, the measurements were performed at the beam line P04 of the PETRA III storage ring at DESY, Hamburg. The energy range 533–542 eV was scanned in 100 meV steps with an exit slit width of 1500  $\mu$ m, resulting in 100 meV bandwidth. Here, no target cell was used but a gas jet created by supersonic expansion of water vapor at about 50 °C through a nozzle of 150  $\mu$ m diameter. The expansion chamber was separated from the interaction chamber by a 1 mm skimmer.

In both experiments, the dispersed fluorescence photons were detected with an "open-face" stack of two microchannel plates (MCPs) without photocathode. An operating voltage of -1150 V was applied to the front of each MCP with respect to its rear side. Additionally, a -190 V voltage between the anode and the rear side of the second MCP and a -170 V voltage between both MCPs were applied, resulting in a potential of -2660 V at the front of the first MCP with respect to the anode, which was kept at ground potential. By that, electrons were rejected from reaching the MCP. Ions were rejected by a mesh in front of the first MCP, set to +200 V. A position-sensitive wedge and strip anode behind the second MCP was used to detect the dispersed fluorescence position resolved.

#### **III. RESULTS AND DISCUSSION**

The absolute emission cross section of  $L_{\alpha}$  fluorescence as a function of the exciting-photon energy is shown in Fig. 1.



FIG. 1. (Color online) Absolute  $L_{\alpha}$  emission cross sections after excitation of water molecules with synchrotron radiation as a function of the exciting-photon energy. Black squares: data measured in this work in first order of the beam-line monochromator, normalized to the photon flux and calibrated to Ref. [23]. Red circles: measured in this work in second order of the beam-line monochromator, normalized to the photon flux. Blue line: data from Ref. [26].



FIG. 2. (Color online) Lyman-series emission after valence excitation of gaseous water with synchrotron radiation. (a) Absolute cross sections for emission after excitation into  $1b_2^{-1}$  states, measured in first order of the beam-line monochromator. Black squares:  $L_{\alpha}$ ; red circles:  $L_{\beta}$ ; green triangles:  $L_{\gamma}$ ; blue triangles:  $L_{\delta}$ ; cyan diamonds:  $Ly_{\varepsilon}$ ; magenta triangles:  $L_{>\varepsilon}$ . Literature values of thermodynamical onsets are indicated by black bars [25]. Inset: A cutout of the spectrally resolved fluorescence (without  $L_{\alpha}$  at 121.6 nm). The Lyman transitions contributing to the cross-section curves are labeled. In ">  $\varepsilon$ ," all unresolved transitions higher than  $L_{\varepsilon}$  are integrated. (b) Absolute cross sections for emission after excitation into doubly excited and  $2a_1^{-1}$  states, measured in second order of the beam-line monochromator.

Data shown as black squares were obtained in first order of the beam-line monochromator and are normalized to the photon flux (see Sec. II). They are calibrated to the known cross section of Ref. [23] at an exciting-photon energy of 17.5 eV. Using the same procedure, the data from Ref. [26] were absolutely calibrated up to 60 eV. In turn, data shown as red circles, which were obtained in second order of the beam-line monochromator up to 33 eV, are calibrated to the value of the curve of Ref. [26] at an exciting-photon energy of 31 eV. Above 33 eV, no reasonable results were obtained in the present experiment owing to extremely weak photon flux. From approximately 28 eV on, the beam-line intensity starts to decrease substantially and statistics become worse. However, within the uncertainty (discussed below), our relative cross sections agree with Ref. [26].

## A. Exciting-photon energy range between 15 and 22 eV: Dissociation of $1b_2^{-1}$ states into $OH(X^2\Pi) + H(n)$ fragment pairs

Figure 2(a) shows the absolute Lyman-series fluorescence emission cross sections as functions of the exciting-photon energy emitted from neutral excited hydrogen dissociation fragments after valence electron excitations of water with energies between 15 and 22 eV. The thermodynamical limits for the dissociation of water into  $OH(X^2\Pi) + H(n)$  fragment pairs are listed in Table I. All observed Lyman lines start emitting at their respective thermodynamical limits for the formation of the corresponding excited fragment. The observed energetic onset of the L<sub> $\alpha$ </sub> line agrees with observations in previous publications [22,25] and the observed onsets for L<sub> $\beta$ </sub> and L<sub> $\gamma$ </sub> agree with the observed onsets of H<sub> $\alpha$ </sub> and H<sub> $\beta$ </sub>, respectively [22], as these lines are branched transitions from the same initial states.

The simultaneous observation of the  $L_{\alpha}$  line together with the higher members of the Lyman series in the dispersed fluorescence spectra enabled a determination of the absolute emission cross sections of the higher Lyman-series lines [Fig. 2(a)], calibrated to the known  $L_{\alpha}$  emission cross section [23–25]. The uncertainty of the literature values is given as 30% [23]. The uncertainty caused by the spatial response of different detector areas for position-sensitive detection of the dispersed fluorescence was determined by measuring the intensity of the same line at different positions and is 20%. No correction of the spectral detection efficiencies was attempted here, as the quantum efficiency of bare MCPs is assumed to be constant within the Lyman-series range [33,34]. Considering Gaussian uncertainty propagation, the total uncertainty for the absolute cross-section values is therefore in the range of 40%. The relative uncertainties within one dispersed fluorescence emission function are essentially the statistical uncertainties of the measurement and are typically at 5% in the range from 15 to 22 eV and at 20% above 22 eV. The inset of Fig. 2(a) shows a cutout of a typical fluorescence spectrum. Integration of all Lyman transitions yields a maximum Lyman emission cross section of about 3.7 Mb at an exciting-photon energy of 18.5 eV. By comparing it with the known absorption cross section in the respective exciting-photon energy range (about 20 Mb [11]), we find that about 20% of the absorption probability results in the formation of an excited hydrogen dissociation fragment.

The maximum values of the determined absolute cross sections of the Lyman lines are compared with those of the corresponding Balmer lines [29], and the intensity ratio

$$R_{\sigma} = \frac{\sigma_{\rm Lyman}}{\sigma_{\rm Lyman} + \sigma_{\rm Balmer}} = \frac{\sigma_{\rm L}}{\sigma_{\rm L} + \sigma_{\rm H}}$$
(1)

is calculated and listed in Table II. From theory, for a given n, the branching ratio  $R_{p,n}$  can be computed from the probabilities for the respective Lyman and Balmer transitions [35]:

$$R_{p,n} = \frac{p_{\text{Lyman}}}{p_{\text{Lyman}} + p_{\text{Balmer}}} = \frac{p_{\text{L}}}{p_{\text{L}} + p_{\text{H}}}.$$
 (2)

It should be noted that the  $L_{\alpha}$  emission cross-section function in the exciting-photon energy range from its onset to about 17.2 eV represents the formation probability of H(2p) excited fragments from the dissociating excitations of H<sub>2</sub>O via the dissociation channel OH( $X^2\Pi$ ) + H (n = 2) alone. Above 17.2 eV, the measured cross sections are not equivalent to the formation probability of the corresponding excited hydrogen fragment. For all  $n \ge 3$ , fluorescence cascades are possible, populating states with lower n.

n	Decay	σ <sub>L</sub> (Mb)	σ <sub>H</sub> [29] (Mb)	$R_{\sigma}$ Expt.	$R_{p,n}$ Theory [35]	$E_{\text{peak}}$ (eV)	$({}^{df}/{}_{dE})_{em}$ at $E_{\rm peak}$ (10 <sup>-4</sup> eV)
2	$L_{\alpha}/-$	$1.5 \pm 0.8$		1	1	17.4	137
3	$L_{\beta}/H_{\alpha}$	$2.0 \pm 1.0$	0.27	$0.88\pm0.10$	0.89	18.2	178
4	$L_{\gamma}/H_{\beta}$	$0.8 \pm 0.4$	0.06	$0.93\pm0.07$	0.88	18.7	68
5	$L_{\delta}/H_{\nu}$	$0.4 \pm 0.2$	0.024	$0.94 \pm 0.07$	0.88	19.2	34
6	$L_{\varepsilon}/H_{\delta}$	$0.26\pm0.13$	0.012	$0.95\pm0.05$	0.89	19.3	24

TABLE II. Maximum emission cross sections of the Lyman transitions  $\sigma_L$  and determined branching ratios *R* of Lyman and Balmer transitions. Also the oscillator strengths at the energies of maximum emission cross section  $E_{\text{peak}}$  are given.

Additionally, the oscillator strength functions for the Lyman emission  $\left(\frac{df}{dE}\right)_{em}(E)$  can be calculated analogous to Ref. [29] from the absolute emission cross section  $\sigma_{em}(E)$  by

$$\left(\frac{df}{dE}\right)_{em}(E) = \frac{\sigma_{em}(E)}{4\pi^2 \alpha a_0^2 R_y},\tag{3}$$

with the fine-structure constant  $\alpha$ , the Bohr radius  $a_0$ , and the Rydberg energy  $R_y$ . Table II lists the oscillator strengths for the excitation energies  $E_{\text{peak}}$  with maximum emission cross section.

An assignment of the resonant excitations in this excitation energy range, leading to dissociation into excited neutral hydrogen fragments, was described in several previous publications. Excitations were attributed to  $1b_2^{-1}4sa^1$  and  $1b_2^{-1}5sa^1$  states in the range from 15 to 20 eV [11,22]. Recently, an alternative assignment to  $1b_2^{-1}nla_1^1$  and  $1b_2^{-1}nlb_2^1$  was suggested [26].

# B. Exciting-photon energy range between 22 and 33 eV: Neutral dissociation of doubly excited states and dissociative ionization of $2a_1^{-1}$ states into $OH^+(A) + e^- + H(n)$ fragments

Above 22 eV, the formation of excited hydrogen atoms has been investigated using  $L_{\alpha}$  and Balmer-series cross-section functions [26,29] and electron energy loss spectroscopy in coincidence with  $L_{\alpha}$  detection [27]. The cross sections for all transitions show three pronounced features at about 24, 28, and 31 eV. The energetic limit for dissociative ionization with the formation of an excited hydrogen is at 28.2 eV. Therefore, the first two features were assigned to doubly excited states forming excited hydrogen atoms after neutral dissociation [26]. The feature at 31 eV corresponds to single-hole one-electron  $2a_1^{-1}$ states, decaying also by neutral dissociation or dissociative ionization into  $OH^+(A) + e^- + H(n)$ . Absolute cross sections are available for the Balmer series for these excitations [29]. In the present experiment, statistics became worse in this range due to the decrease of the photon flux of the used beam line with increasing exciting-photon energy. Despite that, the general excitation structure of previous works was reproduced, as shown in Fig. 2(b). The doubly excited state at 24 eV was reported to be weak compared to those at higher exciting-photon energies and can therefore not clearly be identified here. However, the intensity between 26 and 33 eV is composed of two maxima and the intensity ratio between the double excitation at 28 eV and the  $2a_1^{-1}$  state at 31 eV agrees reasonably with previous results [26,29]. Above 33 eV, the  $L_{\alpha}$  emission was interpreted to stem from direct dissociative ionization into  $H(2p) + OH^+(X) + e^-$ .

A comparison with the shape of the Balmer-series emission cross-section curves in this exciting-photon energy range was interpreted as a preferred production of hydrogen fragments with n = 2 compared to higher states  $(n \ge 3)$  [26]. However, in the present work, an intensity ratio of approximately two was found for both  $L_{\beta}$  and  $L_{\gamma}$  with respect to  $L_{\alpha}$  for the excitations at 28 and 31 eV. Apparently, the production cross section of H (n = 3,4) dissociation fragments is much higher than that of H (n = 2) for these excitations. We therefore refine the interpretation in Ref. [26] of the signal above 33 eV: The production rate of hydrogen fragments with n = 3,4 with respect to n = 2 is higher for the excitations at 28 and 31 eV compared to direct dissociative ionization above 33 eV.

## C. Exciting-photon energy range between 533 and 542 eV: Inner-shell excitation

The excitation of inner-shell electrons into unoccupied orbitals and their assignments were also the subject of previous works [36,37]. Fluorescence emission after excitation of water vapor in this energy range was already investigated [20,30,38]. The relative emission excitation functions for  $L_{\alpha}$  and several OH and O fragment transitions were reported [30], as well as for the Balmer series [20]. The intensity ratios of low Lyman  $(L_{\alpha})$  and Balmer  $(H_{\beta})$  lines after different excitations were studied and discussed extensively and can be understood by energetic reasons after spectator Auger decays [20,30,39,40]. To produce excited hydrogen fragments, the internal energy of the Auger final states of H<sub>2</sub>O<sup>+\*</sup> must be higher than the respective energy necessary for dissociation into  $OH^+ +$  $H(n \ge 2)$ , which is 16.5 eV for n = 2 and 17.5 eV for n = 3 [20,29]. From the Auger spectra, it can be deduced that for core excitations into the  $4a_1$ ,  $2b_2$ , and  $3pa_1/3pb_1$  states, only a small fraction of the Auger final states fulfills this condition [39,40] and the intensity relations  $3pa_1/3pb_1 > 2pa_1/3pb_1$  $2b_2 > 4a_1$  can be explained satisfactorily. However, the probability for the production of excited hydrogen fragments after these excitations is rather low. In contrast, excitation into Rydberg-like states close to the core ionization threshold yields a remarkable enhancement of Balmer emission compared to the molecular excitations [20].

The relative emission cross section of the Lyman series is shown in Fig. 3. The energies of the three lowest molecular excitations are labeled and indicated by black dotted lines. The red dashed line indicates the inner-shell ionization threshold. Figures 3(a)-3(d) represent the spectrally resolved Lyman transitions  $L_{\alpha}-L_{\delta}$ . In Fig. 3(e), all unresolved transitions higher than  $L_{\delta}$  are integrated. Our results agree with the literature,



FIG. 3. (Color online) Relative Lyman-series emission intensities as functions of the exciting-photon energy in the H<sub>2</sub>O inner-shell excitation range. (a)  $L_{\alpha}$ , (b)  $L_{\beta}$ , (c)  $L_{\gamma}$ , (d)  $L_{\delta}$ , and (e) integrated intensity over all transitions higher than  $L_{\delta}$ . All panels are set to the same relative scale. Black dotted lines indicate the first three molecular excitations [36]. The red dashed line at the exciting energy of 539.8 eV indicates the molecular inner-shell ionization threshold.

i.e., the production of highly excited hydrogen fragments is enhanced in the energy range of Rydberg-like excitations. Intensity ratios in the  $L_{\beta}$  and  $L_{\gamma}$  excitation function agree with the corresponding curves for  $H_{\alpha}$  and  $H_{\beta}$  of Ref. [20]. For H ( $n \ge 4$ ), Rydberg excitation leads to the highest production cross section in the core excitation range.

In Ref. [20], this significant increase in the production probability of highly excited hydrogen atoms is explained by the different character of the Rydberg-like excitation. According to their analyses, a highly excited molecular Rydberg state leaves the water molecule in a quasi-doubly-ionized state after a spectator Auger decay. Because of this, the Auger decay causes a collapse of the orbitals and enables a simultaneous shakeup of the spectator electron. Subsequently, the systems dissociates into  $OH^+ + H^+$  and the Rydberg electron attaches to one of the fragments, while conserving its principal and angular quantum numbers. Thereby, the production of excited hydrogen becomes independent of the Auger final state. This model is supported by theoretical calculations of the overlap of the initially excited Rydberg states and final states of the shakeup process. Using fluorescence spectroscopy, a similar behavior of this "spectator dissociation" of Rydberg excitations was previously observed for inner-shell excitations of other molecules [16–18]. Our results corroborate the suggested model and should stimulate further theoretical effort to determine quantitative production cross sections of excited fragments after core excitation of the water molecule.

#### **IV. CONCLUSIONS**

To summarize, we measured relative and absolute emission cross sections for Lyman-series emission after monochromatic valence and inner-shell excitation of gaseous water as functions of the exciting-photon energy. For emission in the range 15–33 eV, absolute emission cross sections were determined and oscillator strengths for the peak intensities were calculated. The results agree reasonably with complementary previous works on Lyman and Balmer emission. In the exciting-photon energy range 22–34 eV, previously reported superexcited states were corroborated. For inner-shell Rydberg excitations, we confirmed a spectator dissociation and gave relative cross sections for the Lyman-series lines. Our results contribute to a completion of dissociation models and branching ratios in the decay processes of (photo)excited gaseous water molecules.

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- [1] B. C. Garrett et al., Chem. Rev. 105, 355 (2005).
- [2] B. Winter, Nucl. Instrum. Methods Phys. Res. A 601, 139 (2009).
- [3] S. Thürmer, M. Ončák, N. Ottosson, R. Seidel, U. Hergenhahn, S. E. Bradforth, P. Slavíček, and B. Winter, Nat. Chem. 5, 590 (2013).
- [4] L. S. Cederbaum, J. Zobeley, and F. Tarantelli, Phys. Rev. Lett. 79, 4778 (1997).
- [5] S. D. Stoychev, A. I. Kuleff, and L. S. Cederbaum, J. Chem. Phys. 133, 154307 (2010).
- [6] S. Barth, M. Ončák, V. Ulrich, M. Mucke, T. Lischke, P. Slavíček, and U. Hergenhahn, J. Phys. Chem. A 113, 13519 (2009).
- [7] T. Jahnke et al., Nat. Phys. 6, 139 (2010).
- [8] M. Mucke et al., Nat. Phys. 6, 143 (2010).
- [9] M. S. Banna, B. H. McQuaide, R. Malutzki, and V. Schmidt, J. Chem. Phys. 84, 4739 (1986).
- [10] R. Sankari *et al.*, Chem. Phys. Lett. **380**, 647 (2003).

- [11] P. Gürtler, V. Saile, and E. E. Koch, Chem. Phys. Lett. 51, 386 (1977).
- [12] W. F. Chan, G. Cooper, and C. E. Brion, Chem. Phys. 178, 387 (1993).
- [13] G. N. Haddad and J. A. R. Samson, J. Chem. Phys. 84, 6623 (1986).
- [14] H. Hertz, H. W. Jochims, H. Schenk, and W. Sroka, Chem. Phys. Lett. 29, 572 (1974).
- [15] E. D. Poliakoff, in Vacuum Ultraviolet Photoionization and Photodissociation of Molecules and Clusters, edited by N. C. Ng (World Scientific, New York, 1991).
- [16] M. Ukai, S. Machida, K. Kameta, M. Kitajima, N. Kouchi, Y. Hatano, and K. Ito, Phys. Rev. Lett. 74, 239 (1995).
- [17] H. Liebel, S. Lauer, F. Vollweiler, R. Müller-Albrecht, A. Ehresmann, H. Schmoranzer, G. Mentzel, and K.-H. Schartner, and O. Wilhelmi Phys. Lett. A 267, 357 (2000).
- [18] A. Ehresmann, Ph. V. Demekhin, W. Kielich, I. Haar, M. A. Schlüter, V. L. Sukhorukov, and H. Schmoranzer, J. Phys. B 42, 165103 (2009).
- [19] M. Glass-Maujean, Ch. Jungen, G. Reichardt, A. Balzer, H. Schmoranzer, A. Ehresmann, I. Haar, and P. Reiss, Phys. Rev. A 82, 062511 (2010).
- [20] E. M. García, A. Kivimäki, L. G. M. Pettersson, J. Á. Ruiz, M. Coreno, M. de Simone, R. Richter, and K. C. Prince, Phys. Rev. Lett. 96, 063003 (2006).
- [21] M. Meyer, S. Aloise, and A. N. Grum-Grzhimailo, Phys. Rev. Lett. 88, 223001 (2002).
- [22] O. Dutuit, A. Tabche-Fouhaile, I. Nenner, H. Frohlich, and P. M. Guyon, J. Chem. Phys. 83, 584 (1985).
- [23] C. Y. R. Wu and D. L. Judge, J. Chem. Phys. 75, 172 (1981).

- [24] J. E. Mentall, G. R. Möhlmann, and P. M. Guyon, J. Chem. Phys. 69, 3735 (1978).
- [25] C. Y. R. Wu, E. Phillips, L. C. Lee, and D. L. Judge, J. Chem. Phys. 70, 601 (1979).
- [26] M. Nakano, T. Odagiri, T. Tanabe, K. Funatsu, I. H. Suzuki, M. Kitajima, and N. Kouchi, J. Phys. B 43, 215206 (2010).
- [27] T. Tsuchida et al., J. Phys. B 44, 175207 (2011).
- [28] K. Mitsuke, J. Chem. Phys. 117, 8334 (2002).
- [29] M. Kato et al., J. Phys. B 37, 3127 (2004).
- [30] A. Kivimäki et al., J. Phys. B 39, 1101 (2006).
- [31] H. Schmoranzer, H. Liebel, F. Vollweiler, R. Müller-Albrecht, A. Ehresmann, K.-H. Schartner, and B. Zimmermann, Nucl. Instrum. Methods Phys. Res. A 467, 1526 (2001).
- [32] G. Reichardt *et al.*, Nucl. Instrum. Methods Phys. Res. A 467, 462 (2001).
- [33] O. H. W. Siegmund, in *Vacuum Ultraviolet Spectroscopy II*, edited by J. A. R. Samson and D. L. Ederer (Academic, San Diego, 2000).
- [34] *MCP Assembly Technical Information* (Hamamatsu Photonics, Shimokanzo, Japan, 2006).
- [35] H. A. Bethe and E. Salpeter, *Quantum Mechanics of One- and Two-electron Atoms* (Plenum, New York, 1977).
- [36] J. Schirmer, A. B. Trofimov, K. J. Randall, J. Feldhaus, A. M. Bradshaw, Y. Ma, C. T. Chen, and F. Sette, Phys. Rev. A 47, 1136 (1993).
- [37] K. Okada et al., Chem. Phys. Lett. 326, 314 (2000).
- [38] A. Kivimäki, M. de Simone, M. Coreno, V. Feyer, E. M. García, J. Á. Ruiz, R. Richter, and K. C. Prince, Phys. Rev. A 75, 014503 (2007).
- [39] I. Hjelte et al., Chem. Phys. Lett. 334, 151 (2001).
- [40] U. Hergenhahn et al., Chem. Phys. 289, 57 (2003).