Strong competition between velocity-changing and phase- or state-changing collisions in H₂ spectra perturbed by Ar

P. Wcisło,^{1,*} F. Thibault,² H. Cybulski,¹ and R. Ciuryło¹

¹Institute of Physics, Faculty of Physics, Astronomy and Informatics, Nicolaus Copernicus University, Grudziadzka 5/7, 87-100 Torun, Poland ²Institut de Physique de Rennes, UMR CNRS 6251, Université de Rennes 1, Campus de Beaulieu, Bâtiment 11B, F-35042 Rennes, France (Descined 9, January 2015, architect 18, Mar 2015)

(Received 8 January 2015; published 18 May 2015)

A collisional inhomogeneous broadening of the $H_2 Q(1)$ line perturbed by Ar was observed for the first time 25 years ago. Several attempts were made to explain the line broadening from *ab initio* calculations, which, however, resulted in fundamental discrepancies between the theory and experiment. To resolve this problem we investigate two possible sources of these differences. First, we repeat the *ab initio* calculations of the broadening and shifting speed dependence using in the scattering calculations a highly accurate *ab initio* H_2 -Ar interaction potential. Second, we replace the previous phenomenological models of velocity-changing collisions with a more physical one based on the interaction potential. This allows us to properly reproduce the experimental broadening over a wide range of temperatures and pressures. We show that this abnormal broadening is affected by strong competition between the velocity-changing collisions and speed-dependent collisional shifting, especially at high pressures.

DOI: 10.1103/PhysRevA.91.052505

PACS number(s): 33.70.Jg

I. INTRODUCTION

Inhomogeneous collisional broadening of the isotropic (polarized) Raman Q(1) line of the 0–1 band of H₂ perturbed by Ar, originating from a strong speed dependence of the collisional line shift, was first observed by Farrow et al. [1]. Further measurements of the broadening and shifting of this line over a wide range of pressures and temperatures were carried out by Berger et al. [2] and Chaussard et al. [3]. Many attempts were made to describe the characteristics of the shape of this line in both phenomenological [1,3-5] and ab initio ways [6–10]. The phenomenological models fitted to experimental data reproduced correctly the line shape and its features (shift, width, and asymmetry) in a wide pressure and temperature range. However, the ab initio descriptions originating from the H₂-Ar interaction, where none of the parameters was fitted to experimental data, underestimated the line broadening significantly [7–9]. Recently, Tran *et al.* [10] have extended the approach of Waldron et al. [9] using a biparametric Keilson-Storer model to handle the velocitychanging collisions. Despite the fact that the parameters of the biparametric Keilson-Storer model were determined from reliable ab initio classical molecular dynamics simulations (CMDS), the apparent linewidth at high densities reported in [10] was still considerably smaller than the experimental width.

In this article we propose an approach for calculating spectral line shapes based on the *ab initio* interaction potential. Using this methodology, we obtain agreement between the theoretical and experimental apparent widths of the H₂ fundamental Q(1) line perturbed by Ar, hence resolving a problem that has remained unexplained for 25 years. We treated the phase- or state-changing collisions in a fully *ab initio* way by solving the close-coupling equations for the quantum scattering problem [11–13]. The obtained line broadening and, in particular, its shifting exhibit strong dependence

on the absorber (H₂) speed. In contrast to the previous studies [1,3-5,10] the velocity-changing collisions are not described in a phenomenological way but are introduced by means of the *ab initio* interaction potential [14,15]. Following Refs. [16,17], we demonstrate that even the use of the simplest hard-sphere approximation of the *ab initio* potential in the speed-dependent billiard-ball profile (SDBBP) [18] leads to a correct reproduction of the experimental spectra.

Proper ab initio calculations of molecular spectral line shapes are crucial for diminishing the systematic errors in ultra-accurate Doppler-limited molecular spectroscopy. We show that the use of the appropriate line-shape model allows errors in the validation of quantum-chemical calculations of potential-energy surfaces (PESs) to be eliminated. On the other hand, the use of advanced line-shape models is necessary in order to increase the accuracy in the determination of line positions and intensities. Such accurate measurements are used for testing the relativistic and quantum electrodynamics theories in molecular systems, particularly for H_2 [19–21], which is an excellent playground for such purposes. In the ultra-accurate measurements of line positions the collision-induced shift is measured as a function of pressure and extrapolated to the zero pressure [19-21]. However, at room temperature the Doppler width of H₂ lines is of the order of 0.1 cm^{-1} [19–21], while the recently achieved accuracy of the determination of the H_2 (3–0) S(3) line position is 5×10^{-5} cm⁻¹ [21]. Therefore at this level of accuracy it is crucial to properly describe the competition between the speed-dependent collisional shifting and velocitychanging collisions, which is responsible for the formation of the asymmetric line shape and final determination of the line position. The improvement of the transition frequency determination by the use of a more advanced line-shape model including the speed-dependent collisional shift was clearly demonstrated by Sinclair et al. [22]. The authors were able to reduce the discrepancies between the experimental results and Wolniewicz [23] ab initio calculations by more than one order of magnitude compared to earlier results [24]. The ultra-accurate metrology of H₂ has been already used

^{*}piotr.wcislo@fizyka.umk.pl

in the search for exotic physics beyond the standard model [25].

Clearly, H₂ lines perturbed by Ar are of particular interest from the point of view of spectral line-shape theory because in this system the differences between the real line shape and the ordinary Voigt profile are especially exaggerated. The minimum of the apparent linewidth as a function of pressure caused by Dicke narrowing is very pronounced. A high perturber to absorber mass ratio makes the velocity-changing collisions very different from the soft-collision limit [26,27]. Moreover the collisional broadening and shifting depend very strongly on the absorber speed. A large spread of collisional shifts for different classes of absorber speed leads to inhomogeneous broadening of the line as well as its asymmetry. All this makes the H₂-Ar system very demanding for testing line-shape models and easily reveals any imperfections of the treatment used. Therefore the line profile successfully applied in this case should work even better for other systems important for metrological applications and atmospheric research, helping to reduce the systematic errors.

Spectral line-shape studies play an important role in the spectroscopic determination of the Boltzmann constant and Doppler-width thermometry [28,29]. Actually, the uncertainty in the description of spectral line shape is seen as a main source of systematic errors in this approach. Finally, the determination of atmospheric greenhouse-gas concentrations at the subpercent level, necessary for reliable monitoring of climate change, also requires a more rigorous approach to line-shape modeling [30].

We show that, for the H₂-Ar system, the broadening is significantly influenced by the competition between phaseor state-changing and velocity-changing collisions, which is strong and atypical. This effect can be properly handled only when both the phase- or state- and velocity- or speedchanging collisions are described correctly, especially at high pressures, well above the Dicke minimum. It should be emphasized that for H₂-Ar the inhomogeneous broadening caused by the speed dependence of collisional shifting is much larger than the ordinary collisional broadening. However, this inhomogeneous component is significantly reduced by frequent speed-changing collisions. Therefore, we have two large effects (the inhomogeneous collisional broadening and the speed-changing collisions) which almost cancel each other. Consequently, even small imperfections in the description of either can result in a large error in the estimation of the apparent linewidth. All these points lead us to the conclusion that a proper description of the competition between phaseor state-changing and velocity- or speed-changing collisions is crucial for the determination of linewidth in the investigated case.

The previously used phenomenological methods of retrieving the pressure broadening parameter γ [2,3] from experimental spectra were strongly affected by an improper model of the velocity-changing collisions. Hence the comparison of the experimental pressure-broadening coefficient γ with the *ab initio* close-coupling calculations resulted in huge discrepancies [9,10]. Finally, in contrast to the earlier studies [7–10], we show that the imperfections of previously used PESs [31] were not responsible for such a huge underestimation of γ .

II. THE MODEL OF THE LINE SHAPE

A line shape can be calculated via an integral over the active molecule velocity \vec{v} ,

$$I(\omega) = \frac{1}{\pi} \operatorname{Re} \int d^3 \vec{\mathrm{v}} f_m(\vec{\mathrm{v}}) h(\omega, \vec{\mathrm{v}}), \qquad (1)$$

where $f_m(\vec{v})$ is the Maxwellian velocity distribution and where the function $h(\omega, \vec{v})$ is a solution of the transport-relaxation equation [32,33],

$$1 = \left\{ \Gamma(v) - i [(\omega - \omega_0) - \vec{k} \cdot \vec{v} - \Delta(v)] - \hat{S}_{VC}^f \right\} h(\omega, \vec{v}),$$
(2)

where $(\omega - \omega_0)$ is the light detuning, $\vec{k} \cdot \vec{v}$ is the Doppler shift, \hat{S}_{VC}^f is the collision operator describing molecule velocity changes, and $\Gamma(v)$ and $\Delta(v)$ are the speed-dependent collisional width (half width at half maximum, HWHM) and shift [34,35], respectively. The physical interpretation of $h(\omega, \vec{v})$ is that the product $f_m(\vec{v})h(\omega, \vec{v})$ is proportional to the velocity distribution of an optical coherence. We solve Eq. (2) by decomposing the operator and functions in Burnett's basis [18,32,36]. This allows the integral equation (2) to be converted into a set of algebraic equations [33].

Equation (2) implies that the phase- or state- and velocitychanging collisions are not correlated. As was pointed out in Ref. [10], this assumption is justified by the fact that the rate of the velocity-changing collisions is more than two orders of magnitude greater than the rate of phase or state changes. Nevertheless, a similar approach to the line-shape calculations can be used in the case where the correlations between the phase- or state- and velocity-changing collisions [37,38] should be taken into account (see Refs. [39–41]).

III. PHASE- OR STATE-CHANGING COLLISIONS

First, to describe the phase- or state-changing collisions we have performed close-coupling calculations with the MOLSCAT scattering code [11]. The vibrational motion of H_2 is taken into account while Ar is considered a structureless particle. We have used Ben-Reuven's formalism [12,13] to obtain, from the scattering S matrix, the generalized spectroscopic cross sections, which depend on the relative collisional kinetic energy. The real and imaginary parts of such cross sections, thermally averaged over the perturber velocity [42,43], provide the speed-dependent collisional broadening $\gamma(v) = \Gamma(v)/n$ and shifting $\delta(v) = \Delta(v)/n$ coefficients, where *n* is the molecular number density (see Fig. 1). The pressure broadening and shifting coefficients, for a given temperature, are defined as $\gamma = \langle \gamma(v) \rangle$ and $\delta = \langle \delta(v) \rangle$, where $\langle \cdots \rangle$ is an average over a Maxwellian velocity distribution. The quantum scattering calculations were performed on our highly accurate PES calculated using the spin-restricted coupled-cluster singles and doubles including noniterative triple excitations [CCSD(T)] method with the large augmented correlation-consistent polarized valence quadruple zeta (aug-cc-pCVQZ) basis and the 332211 midbond basis set, taking into account stretching of the H_2 bond [44]. In the *ab initio* calculations all electrons were correlated; the MOLPRO program 2010.1 [45] and 2012.1 [46] versions were employed. We have also repeated the closecoupling calculations using the semiempirical PES of Bissonnette et al. [31]. We find that the use of our PES [44] leads to



FIG. 1. (Color online) Speed-dependent collisional broadening $\gamma(v)$ and shifting $\delta(v)$ vs the speed of the active molecule for the isotropic Raman Q(1) line of the 0–1 band of H₂ perturbed by Ar. The blue (dark gray) and red (light gray) lines correspond to T = 296 K and T = 795 K, respectively. The vertical dashed lines indicate the most probable speeds at these two temperatures. The blue (dark gray) and red (light gray) dotted lines represent the Maxwellian speed distribution for T = 296 K and T = 795 K, respectively.

only slight corrections in the generalized spectroscopic cross sections for line broadening, which are even less consistent with the broadening retrieved from experimental spectra [2,3]. For instance, for the collision kinetic energy of 215 cm⁻¹ the cross sections for broadening obtained with our PES [44] and the PES by Bissonnette *et al.* [31] are 0.300 and 0.323 Å², respectively. The broadening coefficient γ corresponding to room temperature obtained from these two PESs is about 1.3×10^{-3} cm⁻¹/amagat, which is consistent with previously reported *ab initio* calculations [7,9,10]. This value, however, is almost two times smaller than those phenomenologically derived from experimental spectra [2,3].

These results suggest, in contrast to those reported in Refs. [7,9,10], that the huge discrepancy between theoretically and experimentally determined line broadening is not caused by imperfections in PESs. It is very unlikely that two independent PESs, one *ab initio* [44] and one semiempirical [31], would lead to the same cross sections if either or both of them were incorrect.

IV. VELOCITY-CHANGING COLLISIONS

The second effect of the collisions, described by the \hat{S}_{VC}^{I} operator in Eq. (2), are the changes in the H₂ velocity in both its amplitude and direction. The velocity-changing collisions have two consequences on molecular spectra. First, the changes in the molecule velocity can reduce the effective Doppler shift and narrow the Gaussian component of the

profile. This confinement narrowing is well known as the Dicke effect [47]. Second, dominating at higher pressures, the inhomogeneous broadening originating from the speed dependence of the collisional shift is narrowed by speed class exchanges [16]. To understand the source of the huge discrepancies reported in Refs. [7-10], first, it is crucial to realize that this narrowing increases for larger changes in the molecule speed under the collisions. It was shown [16,17] (by a comparison of the results with those obtained from CMDS) that the changes in the speed are significantly overestimated by the simple phenomenological models such as hard- or soft-collision models [27,48]. Therefore, when analyzing their experimental data with a hard-collision model, Berger *et al.* [2] introduced an extra parameter to reduce the effective rate of speed changes. However, the value of the fitted pressure broadening coefficient γ was almost twice as large as the theoretical predictions [7,9,10]. Chaussard et al. [3] analyzed their experimental spectra with a model combining both hard and soft collisions [49–51]. The soft-collision model was incorporated in an approximated way [52], which neglected collisionally induced speed-class exchanges in relation to phase- or state-changing collisions (see also discussion in Refs. [18,53]). However, the authors of Ref. [3] obtained almost the same value for the pressure-broadening coefficient γ as in Ref. [2]. Recently, Tran *et al.* [10] have made an attempt to resolve this problem by applying the more advanced phenomenological biparametric three-dimensional Keilson-Storer model, in which the memory parameters were adjusted to reproduce not only velocity but also speed autocorrelation functions deduced from CMDS [54]. Despite the qualitative improvement of the velocity-changing model, this approach resulted in only slightly better agreement with measurements; at high pressures and room temperature, the experimental linewidth was still about 40% larger than the model prediction.

In this study, we replace the phenomenological description of velocity-changing collisions with a more physical approach based on the quantum-mechanical *ab initio* PES. We represent the H₂-Ar interaction by the hard-sphere potential adjusted such that the hard-sphere wall intersects the isotropic component of the PES at the mean collision energy $\epsilon = k_B T$, where *T* is the temperature and k_B is the Boltzmann constant. Note that the potential-well depth is much smaller than ϵ . The hard-sphere approximation for H₂-Ar was verified by comparing hard-sphere collision kernels and relaxation times with those obtained from CMDS [17].

V. RESULTS

To validate our model of the velocity-changing collisions we directly compared the apparent HWHM and the shift of our calculated SDBBP [18] with the experimental values by Berger *et al.* [2] and Chaussard *et al.* [3] (see Fig. 2). Note that, usually, the term HWHM is attributed to one component of the total spectral line shape like the Lorentz or Doppler component (see Sec. II). Here the term HWHM is used for the spectral line shape as a whole.

The experimental data [2,3] were collected for a mixture of 95% Ar and 5% H_2 , taken into account in our calculations by considering the collision operator as a weighed sum of H_2 - H_2



FIG. 2. (Color online) The apparent half width at half maximum (HWHM) and the shift (center of gravity) of the spectral line shape as a function of number density for the isotropic Raman Q(1) transition of the 0–1 band of H₂ perturbed by Ar. The solid lines represent the results of our calculations using SDBBP [18], while the points correspond to the experimental values [2,3]. The dashed and dotted lines represent our theoretical calculations in which the velocity-changing-collision model was replaced with phenomenological softand hard-collision models, respectively. The concentration of Ar was 95% (in the experiment as well as in our calculations).

and H_2 -Ar collision operators [17]. The speed-dependent collisional broadening and shifting associated with the H₂-H₂ collisions were taken from Refs. [55,56] (see also Appendix C in Ref. [17]). The upper panel in Fig. 2 illustrates that our calculated values of the HWHM of SDBBP agree with the measurements to within experimental noise at both 296 and 795 K. It should be noted that our calculations originate only from an ab initio PES, and no parameter was adjusted or fitted to experimental data. Note also that the speed-dependent collisional broadening and shifting associated with the H2-H2 collisions taken from Refs. [55,56] were determined from the fit to experimental data; however, this has a small impact on our results since the concentration of H_2 is only 5% and the speed dependence of collisional shift is about 2.5 times weaker for the H₂-H₂ collisions than for the H₂-Ar ones [compare the slopes of the $\delta(v)$ curves from Fig. 1 and from Eq. (C4) given in [17] with parameters taken from [55]]. Moreover, for H_2 - H_2 the competition between the collisional speed changes, and speed-dependent collisional shifting is less sensitive to the form of the speed-changing model due to a smaller absorber to perturber mass ratio [57], which is equal to 1 in this case.

To show the importance of a proper description of speedchanging collisions we also simulated the linewidth and shift for soft- and hard-collision models (the soft collisions were incorporated in an exact manner as a limiting case of SDBBP [18] where the perturber to absorber mass ratio approaches zero; see dashed and dotted lines in Fig. 2). The linewidth predicted by these models is much smaller than the experimental one, which has already been demonstrated in Refs. [16,39]. As mentioned earlier, this is a consequence of greatly overestimated speed changes by these models [17]. On the other hand, the line shift is almost independent of the model of the velocity-changing collisions (note the overlap of the solid, dashed, and dotted lines in the lower panel in Fig. 2). Possible sources of discrepancies between simulated and experimental shifts are PES imperfections and neglecting the correlations between phase- or state- and velocity-changing collisions [37,38]. The line shifts determined from the closecoupling calculations with our PES [44] are close to those determined in Ref. [10] with the PES of Bissonnette *et al.* [31], which are in good agreement with the experimental results (see Fig. 3 in Ref. [10]), in contrast to the collisional widths, which are very dependent on the model used to derive them from experimental line shapes. However, intrinsically, for this system, at T = 795 K it is difficult to calculate the line shift with high accuracy because even small differences in the speed-dependent collisional shifting $\delta(v)$ (see lower panel in Fig. 1) lead to large relative differences in the apparent line shift (see lower panel in Fig. 2). The reason for this is the fact that $\delta(v)$ is almost centered around zero and the value of the apparent line shift is much smaller than the range of $\delta(v)$ values. This observation is also consistent with experimental results obtained by Chaussard *et al.* [3] showing that the sign of the line shift changes at high temperature.

VI. CONCLUSIONS

To sum up, in this work we have shown that the SDBBP [18] calculations based on an intermolecular interaction potential [17] explain the physical mechanism responsible for the anomalously high broadening of H₂ transition observed in experiment [3], which was significantly underestimated by previous theoretical studies [7–10]. Moreover our results contradict the conclusions reported in Refs. [7-10], where it was stated that the fundamental discrepancy between theory and experiment originates from some defects in the PESs used to calculate phase- or state-changing collisions. We have demonstrated that the key factor needed to explain the experimental line broadening is a proper description of the velocity-changing collision. In our approach we have replaced the phenomenological models used previously (which overestimate the rate of the speed changes) with a more physical one based on the interaction potential; that is, we approximated the PES with the hard-sphere potential, which has already been shown to properly reproduce the velocity-changing collisions for H₂-Ar and H₂-H₂ [17]. These results are important not only for understanding the principal physical mechanisms behind the formation of inhomogeneously broadened spectra but also for recently rapidly developing ultraprecise optical metrology based on Doppler-limited molecular spectroscopy, where a proper modeling of molecular spectra is necessary to avoid additional systematic errors in fundamental studies [19-21,25,28,29].

ACKNOWLEDGMENTS

The authors are grateful to F. Chaussard for discussions and for providing the experimental data. P.W. is supported by the National Science Centre, Poland, Project No. DEC-2013/09/N/ST4/00327. The research is a part of the program of the National Laboratory FAMO in Toruń, Poland. The research is partially supported by the Foundation for Polish

- R. L. Farrow, L. A. Rahn, G. O. Sitz, and G. J. Rosasco, Phys. Rev. Lett. 63, 746 (1989).
- [2] J. P. Berger, R. Saint-Loup, H. Berger, J. Bonamy, and D. Robert, Phys. Rev. A 49, 3396 (1994).
- [3] F. Chaussard, X. Michaut, R. Saint-Loup, H. Berger, P. Joubert, B. Lance, J. Bonamy, and D. Robert, J. Chem. Phys. 112, 158 (2000).
- [4] D. Robert and L. Bonamy, Eur. Phys. J. D 2, 245 (1998).
- [5] L. Bonamy, H. Tran, P. Joubert, and D. Robert, Eur. Phys. J. D 31, 459 (2004).
- [6] S. Green, J. Chem. Phys. 93, 1496 (1990).
- [7] S. Green, D. W. Schwenke, and W. M. Huo, J. Chem. Phys. 101, 15 (1994).
- [8] L. Waldron and W.-K. Liu, J. Chin. Chem. Soc. 48, 439 (2001).
- [9] L. Waldron, W.-K. Liu, and R. J. L. Roy, J. Mol. Struct. THEOCHEM 591, 245 (2002).
- [10] H. Tran, F. Thibault, and J.-M. Hartmann, J. Quant. Spectrosc. Radiat. Transfer 112, 1035 (2011).
- [11] J. M. Hutson and S. Green, MOLSCAT computer code, version 14, distributed by Collaborative Computational Project No. 6 of the UK Science and Engineering Research Council, Swindon, 1994.
- [12] A. Ben-Reuven, Phys. Rev. 141, 34 (1966).
- [13] A. Ben-Reuven, Phys. Rev. 145, 7 (1966).
- [14] R. Blackmore, S. Green, and L. Monchick, J. Chem. Phys. 91, 3846 (1989).
- [15] A. May, W.-K. Liu, F. McCourt, R. Ciuryło, J. Sanchez-Fortún Stoker, D. Shapiro, and R. Wehr, Can. J. Phys. 91, 879 (2013).
- [16] R. Ciuryło, D. Lisak, and J. Szudy, Phys. Rev. A 66, 032701 (2002).
- [17] P. Wcisło, H. Tran, S. Kassi, A. Campargue, F. Thibault, and R. Ciuryło, J. Chem. Phys. **141**, 074301 (2014).
- [18] R. Ciuryło, D. A. Shapiro, J. R. Drummond, and A. D. May, Phys. Rev. A 65, 012502 (2002).
- [19] A. Campargue, S. Kassi, K. Pachucki, and J. Komasa, Phys. Chem. Chem. Phys. 14, 802 (2012).
- [20] S.-M. Hu, H. Pan, C.-F. Cheng, Y. R. Sun, X.-F. Li, J. Wang, A. Campargue, and A.-W. Liu, Astrophys. J. 749, 76 (2012).
- [21] C.-F. Cheng, Y. R. Sun, H. Pan, J. Wang, A.-W. Liu, A. Campargue, and S.-M. Hu, Phys. Rev. A 85, 024501 (2012).
- [22] P. M. Sinclair, J. P. Berger, X. Michaut, R. Saint-Loup, R. Chaux, H. Berger, J. Bonamy, and D. Robert, Phys. Rev. A 54, 402 (1996).
- [23] L. Wolniewicz, J. Chem. Phys. 99, 1851 (1993).
- [24] I. Dabrowski, Can. J. Phys. 62, 1639 (1984).
- [25] E. J. Salumbides, J. C. J. Koelemeij, J. Komasa, K. Pachucki, K. S. E. Eikema, and W. Ubachs, Phys. Rev. D 87, 112008 (2013).

Supercomputing (WCSS) is also gratefully acknowledged.

- [26] S. Chandrasekhar, Rev. Mod. Phys. 15, 1 (1943).
- [27] L. Galatry, Phys. Rev. 122, 1218 (1961).
- [28] L. Moretti, A. Castrillo, E. Fasci, M. D. De Vizia, G. Casa, G. Galzerano, A. Merlone, P. Laporta, and L. Gianfrani, Phys. Rev. Lett. 111, 060803 (2013).
- [29] C. Lemarchand, S. Mejri, P. L. T. Sow, M. Triki, S. K. Tokunaga, S. Briaudeau, C. Chardonnet, B. Darquié, and C. Daussy, Metrologia 50, 623 (2013).
- [30] C. E. Miller, L. R. Brown, R. A. Toth, D. C. Benner, and V. M. Devi, C. R. Phys. 6, 876 (2005).
- [31] C. Bissonnette, C. E. Chuaqui, K. G. Crowell, R. J. Le Roy, R. J. Wheatley, and W. J. Meath, J. Chem. Phys. **105**, 2639 (1996).
- [32] R. Blackmore, J. Chem. Phys. 87, 791 (1987).
- [33] D. A. Shapiro, R. Ciuryło, J. R. Drummond, and A. D. May, Phys. Rev. A 65, 012501 (2002).
- [34] P. R. Berman, J. Quant. Spectrosc. Radiat. Transfer 12, 1331 (1972).
- [35] H. M. Pickett, J. Chem. Phys. 73, 6090 (1980).
- [36] P. Wcisło, A. Cygan, D. Lisak, and R. Ciuryło, Phys. Rev. A 88, 012517 (2013).
- [37] S. G. Rautian and I. I. Sobelman, Usp. Fiz. Nauk 90, 209 (1966) [Sov. Phys. Usp. 9, 701 (1967)].
- [38] L. Demeio, S. Green, and L. Monchick, J. Chem. Phys. 102, 9160 (1995).
- [39] R. Ciuryło, A. Bielski, J. R. Drummond, D. Lisak, A. D. May, A. S. Pine, D. A. Shapiro, J. Szudy, and R. S. Trawiński, *Spectral Line Shapes*, edited by C. A. Back (AIP, Melville, NY, 2002), p. 151.
- [40] R. Wehr, R. Ciuryło, A. Vitcu, F. Thibault, J. Drummond, and A. May, J. Mol. Spectrosc. 235, 54 (2006).
- [41] D. Lisak, J. T. Hodges, and R. Ciuryło, Phys. Rev. A 73, 012507 (2006).
- [42] A. Pine, J. Quant. Spectrosc. Radiat. Transfer 62, 397 (1999).
- [43] R. Wehr, A. Vitcu, R. Ciuryło, F. Thibault, J. R. Drummond, and A. D. May, Phys. Rev. A 66, 062502 (2002).
- [44] H. Cybulski and P. S. Żuchowski (unpublished).
- [45] H.-J. Werner, P. J. Knowles, G. Knizia, F. R. Manby, M. Schütz *et al.*, MOLPRO, version 2010.1, http://www.molpro.net.
- [46] H.-J. Werner, P. J. Knowles, G. Knizia, F. R. Manby, M. Schütz, et al., MOLPRO, version 2012.1, http://www.molpro.net.
- [47] R. Dicke, Phys. Rev. 89, 472 (1953).
- [48] M. Nelkin and A. Ghatak, Phys. Rev. 135, A4 (1964).
- [49] R. Ciuryło, Phys. Rev. A 58, 1029 (1998).
- [50] B. Lance and D. Robert, J. Chem. Phys. **109**, 8283 (1998); **111**, 789 (1999).
- [51] R. Ciuryło, A. Pine, and J. Szudy, J. Quant. Spectrosc. Radiat. Transfer 68, 257 (2001).

- [52] R. Ciuryło and J. Szudy, J. Quant. Spectrosc. Radiat. Transfer 57, 411 (1997).
- [53] R. Ciuryło, R. Jaworski, J. Jurkowski, A. S. Pine, and J. Szudy, Phys. Rev. A 63, 032507 (2001).
- [54] P. Joubert, P. N. M. Hoang, L. Bonamy, and D. Robert, Phys. Rev. A 66, 042508 (2002).
- [55] J. W. Forsman, J. Bonamy, D. Robert, J. P. Berger, R. Saint-Loup, and H. Berger, Phys. Rev. A 52, 2652 (1995).
- [56] H. Tran, J.-M. Hartmann, F. Chaussard, and M. Gupta, J. Chem. Phys. 131, 154303 (2009).
- [57] P. Wcisło and R. Ciuryło, J. Quant. Spectrosc. Radiat. Transfer 120, 36 (2013).