Quasiclassical quantum defect theory and the spectrum of highly excited rubidium atoms

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We report on a significant discrepancy between recently published, highly accurate variational calculations and precise measurements of the spectrum of Rydberg states in ⁸⁷Rb on the energy scale of fine splitting. Introducing a modified effective single-electron potential, we determine the spectrum of the outermost bound electron from a standard WKB approach. Overall very good agreement with precise spectroscopic data is obtained.

can be found in [5].

term

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A table of the parameters $a_1(l)$, $a_2(l)$, $a_3(l)$, $a_4(l)$, α_c , and $r_c(l)$

excitation spectrum of the outermost electron of ⁸⁷Rb, it has

been suggested [4] to superimpose *a posteriori* a spin-orbit

 $\widetilde{V}_{\rm SO}(r;j,l) = \frac{V_{\rm SO}(r;j,l)}{[1 - \alpha^2 V_{\rm eff}(r;l)]^2}$

on the potential $V_{\text{eff}}(r; l)$, which then influences the spectrum

accessible to the outermost electron. Here

In an attempt to also describe the fine splitting of the

I. INTRODUCTION

The spectrum of the outermost bound electron of an alkali atom such as ⁸⁷Rb is hydrogenlike but lacks the n^2 degeneracy of the eigenstates labeled by the principal quantum number *n* of the pure Coulomb potential [1,2]:

$$E_{n,l} = -\frac{1}{(n-\delta_l)^2}.$$
 (1)

This effect is the well-known quantum defect δ_l , resulting from the interaction of the outermost electron with the ionic core of the atom and the nucleus. In a refined version of the statistical Thomas-Fermi theory [3], an effective potential determining the interaction between the outermost electron and the nucleus can be modeled accurately by a spherically symmetric potential $V_{\text{eff}}(r; l)$ depending on the distance *r* from the center and depending on the orbital angular momentum $l \in \{0, 1, 2, ..., n - 1\}$ [2,4,5]:

$$V_{\rm eff}(r;l) = -2\left[\frac{Z_{\rm eff}(r;l)}{r} + V_{\rm pol}(r;l)\right].$$
 (2)

Here the function $Z_{\text{eff}}(r;l)$ represents a position-dependent weight function that interpolates the value of the charge between unity for large r and charge number Z near to the nucleus for $r \rightarrow 0$, and $V_{\text{pol}}(r;l)$ represents a short-range interaction taking into account the static electric polarizability of the ionic core [1,6].

Overall good agreement with spectroscopic data of alkali atoms (but discarding the fine splitting) has been reported in [5], choosing

$$Z_{\rm eff}(r;l) = 1 + (Z-1)e^{-ra_1(l)} - re^{-ra_2(l)} \left[a_3(l) + ra_4(l)\right]$$
(3)

and

$$V_{\text{pol}}(r;l) = \frac{\alpha_c}{2} \frac{1 - \exp\left[-\left(\frac{r}{r_c(l)}\right)^6\right]}{r^4}.$$
 (4)

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 $E_{n,j,l}$ on the scale of fine splitting and the orbitals $\psi_{n,j,l}(r)$

(5)

$$Y_{\rm SO}(r;j,l) = \alpha^2 \frac{1}{r} \frac{\partial V_{\rm eff}(r;l)}{\partial r} g(j,l), \qquad (6)$$

and $\alpha = \frac{\lambda_C}{a_B} \simeq \frac{1}{137.036}$ denotes the fine-structure constant, and

$$g(j,l) = \begin{cases} 0 & \text{if } l = 0, \\ \frac{j(j+1) - l(l+1) - \frac{3}{4}}{2} & \text{if } l \ge 1, \end{cases}$$
(7)

where $j \in \{l - \frac{1}{2}, l + \frac{1}{2}\}$. To determine those orbitals (with principal quantum number $n = n_r + l + 1$ and radial quantum number $n_r \in \mathbb{N}_0$), a normalizable solution to the Schrödinger eigenvalue problem for the radial wave function $U_{n,j,l}(r) = rR_{n,j,l}(r)$ and associated eigenvalues $E_{n,j,l} < 0$ is required:

$$\left[-\frac{d^2}{dr^2} + \frac{l(l+1)}{r^2} + \widetilde{V}(r;j,l) - E_{n,j,l}\right] U_{n,j,l}(r) = 0, \quad (8)$$

where

$$\widetilde{V}(r; j, l) = V_{\text{eff}}(r; l) + \widetilde{V}_{\text{SO}}(r; j, l)$$
(9)

denotes the effective single-electron potential.

A highly accurate variational calculation of the excitation spectrum of the outermost electron of ⁸⁷Rb has been carried out recently [7], in which the authors expand the radial wave function of the Schrödinger eigenvalue problem (8) in a basis spanned by 500 Slater-type orbitals (STOs). On the other hand, modern high-precision spectroscopy of Rydberg levels of ⁸⁷Rb has been conducted recently. Millimeter-wave spectroscopy employing selective field ionization allows for precise measurements of the energy differences between Rydberg

State $ n,l=1\rangle$	Expt. [11]	Expt. [8]	Theory [7]	Theory (this work)
8P	$565.1(4) \times 10^3$	NA	602.04×10^{3}	567.75×10^{3}
10P	$219.1(4) \times 10^{3}$	NA	231.87×10^{3}	218.77×10^{3}
30P	NA	4246.30(5)	4500.50	4246.46
35P	NA	2566.41(32)	2717.41	2566.28
45P	NA	1144.09(13)	1217.24	1143.95
55P	NA	605.77(7)	644.81	605.68
60P	NA	460.76(5)	480.32	460.68

TABLE I. Fine splitting $\Delta E_{n,l=1}$ for P states in megahertz.

levels [8]. An independent approach is to perform purely optical measurements on absolute Rydberg level energies by observing electromagnetically induced transparency (EIT) [9,10]. However, there is a systematic discrepancy between variational calculations and the spectroscopic measurements of the fine splitting,

$$\Delta E_{n,l} = E_{n,l+\frac{1}{2},l} - E_{n,l-\frac{1}{2},l}, \qquad (10)$$

as shown in Tables I and II. Given the fact that the error bars of the independent experiments [8,10] are below 1.1 MHz down to 20 kHz, and on the other hand considering the high accuracy of the numerical calculations presented in [7], such a discrepancy between experiment and theory is indeed significant.

So, what could be the reason for the reported discrepancies? First, it should be pointed out that in the variational calculations [7] a slightly different potential was used, that is,

$$V(r; j, l) = V_{\text{eff}}(r; l) + V_{\text{SO}}(r; j, l).$$
(11)

Certainly, within the first-order perturbation theory there exists no noticeable discrepancy in the spectrum of the outermost electron on the fine-splitting scale, when taking into account the spin-orbit forces with $V_{SO}(r; j,l)$ instead of working with $\widetilde{V}_{SO}(r; j,l)$. This is due to the differences being negligible for $r > Z\alpha^2$. However, since $V_{SO}(r; j,l)$ eventually dominates even the contribution of the centrifugal barrier term $\frac{l(l+1)}{r^2}$ within the tiny region $0 < r \leq \alpha^2 Z$, a subtle problem with a non-normalizable radial wave function $U_{n,j,l}(r)$ emerges when attempting to solve the Schrödinger eigenvalue problem for any l > 0 with the potential $V_{SO}(r; j,l)$. Such a problem is absent when one works with $\widetilde{V}_{SO}(r; j,l)$ [4].

A variational calculation with the potential (11) employing N = 500 normalizable STOs as basis functions thus engenders a systematic (small) error of the matrix elements calculated in [7] on the fine-splitting scale. When employing substantially more STOs this error would certainly become larger. With N =

500 STOs the discrepancy of these theoretical results with the high-precision spectroscopic data, as shown in Tables I and II, is far too large to be corrected by simply replacing $V_{SO}(r; j, l)$ with $\tilde{V}_{SO}(r; j, l)$. Hence another explanation is required.

II. QUASICLASSICAL APPROACH AND FINE SPLITTING OF THE HIGHLY EXCITED ⁸⁷Rb

In 1941 alkali atoms had already been studied in the context of modern quantum mechanics in the seminal work by Mayer [3], who emphasized the exceptional role of the l = 1 and l = 2 orbitals. According to Mayer, the outermost electron of an alkali atom is governed by an effective *r*-dependent charge term

$$Z_{\rm eff}(r) = 1 + (Z - 1)F(r), \tag{12}$$

where the function F(r) has been determined by employing the semiclassical statistical Thomas-Fermi approach to the many-electron-atom problem, posing the boundary conditions as $\lim_{r\to 0} F(r) = 1$ and $\lim_{r\to\infty} F(r) = 0$. As discussed by Schwinger [12], this approach ceases to be valid in the innershell region $Z^{-1} < r < Z^{-\frac{1}{3}}$ of the atom. Therefore, taking into account the fine splitting in the spectrum of the outermost electron of alkali atoms *a posteriori* by simply adding the phenomenological spin-orbit term (5) to (2), resulting in the effective single-electron potential (9), seems to be questionable on general grounds in that inner-shell region.

On a more fundamental level, the treatment of relativistic effects in multi-electron-atom spectra requires an *a priori* microscopic description based on the well-known Breit-Pauli Hamiltonian [13,14]:

$$\mathcal{H} = \mathcal{H}_{\rm nr} + \mathcal{H}_{\rm rs} + \mathcal{H}_{\rm fs}.$$
 (13)

Here \mathcal{H}_{nr} is the ordinary *nonrelativistic* many-electron Hamiltonian, while the *relativistic corrections* are represented by the perturbation operators \mathcal{H}_{rs} and \mathcal{H}_{fs} . The perturbation term \mathcal{H}_{rs}

TABLE II. Fine splitting $\Delta E_{n,l=2}$ for D states in megahertz.

State $ n,l=2\rangle$	Expt. [11]	Expt. [8]	Expt. [10]	Theory [7]	Theory (this work)
8D	$30.4(4) \times 10^3$	NA	NA	113.17×10^{3}	36.42×10^{3}
10D	$14.9(2) \times 10^3$	NA	NA	52.05×10^{3}	16.56×10^{3}
30D	NA	452.42(18)	452.5(11)	1447.53	456.13
35D	NA	279.65(10)	280.4(11)	894.84	281.52
45D	NA	128.33(4)	127.8(11)	407.64	128.98
55D	NA	69.17(2)	69.4(11)	223.71	69.47
57D	NA	61.98(2)	62.2(11)	197.39	62.24

contains all the relativistic perturbations like *mass correction*, one- and two-body *Darwin terms*, and further the *spin-spin contact* and *orbit-orbit* terms, which all commute with the total angular momentum **L** and total spin **S**, thus effectuating only small *shifts* of the spectrum of the nonrelativistic Hamiltonian \mathcal{H}_{nr} . The perturbation operator \mathcal{H}_{fs} , on the other hand, breaks the rotational symmetry. It consists of the standard *nuclear spin-orbit*, the *spin-other-orbit*, and the *spin-spin dipole* interaction terms, which all commute with $\mathbf{J} = \mathbf{L} + \mathbf{S}$, but not with **L** or with **S** separately, thus inducing the fine splitting of the nonrelativistic spectrum.

Although the proposed functional form of the potential (11) is highly plausible on physical grounds outside the inner-core region $r > Z^{-\frac{1}{3}}$, *prima facie* it appears to be inconsistent to lump the aforementioned relativistic many-body forces, spin-other-orbit and spin-spin dipole interaction, into an effective single-electron potential of the functional form (11), so that it provides an accurate description also for small distances $Z^{-1} < r < Z^{-\frac{1}{3}}$.

In the absence of a better microscopic theory for an effective single-electron potential $V_{\text{eff}}(r; j, l)$ describing the fine splitting of the spectrum of the outermost electron in the alkali atoms, we introduce a *cutoff* at a distance $r_{\text{so}}(l)$ with $Z^{-1} < r_{\text{so}}(l) < Z^{-\frac{1}{3}}$ so that the effective single-electron potential is now described by the following modified potential:

$$\widetilde{V}_{\text{mod}}(r; j, l) = \begin{cases} V_{\text{eff}}(r; l) & \text{if } 0 \leq r \leq r_{\text{so}}(l), \\ V_{\text{eff}}(r; l) + V_{\text{SO}}(r; j, l) & \text{if } r > r_{\text{so}}(l). \end{cases}$$
(14)

The choice [2]

$$r_{so} (l = 1) = 0.029\,483 \times r_c (l = 1) = 0.044\,282\,5,$$

$$r_{so} (l = 2) = 0.051\,262 \times r_c (l = 2) = 0.249\,572\,0,$$
(15)

gives a surprisingly accurate description of the fine splitting in the spectroscopic data for *all* principal quantum numbers n (see Fig. 1, Tables I and II). By choosing larger values for $r_{so}(l)$ than stated in (15), the calculated fine splitting is too small compared to experiment, and vice versa, by choosing smaller values for $r_{so}(l)$ we find the calculated fine splitting is too large compared to experiment.

The calculation of the spectrum of the outermost bound electron is then reduced to solving the radial Schrödinger equation (8) with the modified potential $\tilde{V}_{mod}(r; j, l)$. The resulting spectrum is actually hydrogenlike, that is,

$$E_{n,j,l} = -\frac{1}{(n - \Delta_{j,l})^2},$$
(16)

where $\Delta_{j,l}$ denotes a quantum defect also comprising the fine splitting. In actual fact the quantum defect describes a reduction of the number of nodes n_r of the radial wave function for l = 0, 1, 2 as a result of the short-range interaction of the outermost electron with the ionic core of the atom. Because the higher the orbital angular momentum quantum number l, the lower the probability of the electron being located near to the center, it is clear that the quantum defect decreases



FIG. 1. (Color online) Reduced fine splitting in THz (a) for P states (cf. Table I) and (b) for D states (cf. Table II). The blue line corresponds to the theory from this work [see Eq. (17)]. The blue circles show the recent numerically calculated results from [7]. The red symbols denote experimental data from Refs. [11] (squares), [8] (crosses), and [10] (circles). The error bars for the experimental data are given in Tables I and II.

rapidly with increasing orbital angular momentum *l*. Therefore $\Delta_{j,l}$ is only notably different from zero for l = 0, 1, 2.

Writing $\Delta_{j,l} = \delta_l + \eta_{j,l}$ with $\eta_{j,l} \ll \delta_l$, the fine splitting to leading order in α^2 is

$$\Delta E_{n,l} = 2 \frac{\eta_{l-\frac{1}{2},l} - \eta_{l+\frac{1}{2},l}}{(n-\delta_l)^3}.$$
(17)

The quasiclassical momentum $p \equiv \sqrt{-Q}$ of the bound electron depending on energy E < 0 with orbital angular momentum l > 0, total angular momentum $j = l \pm \frac{1}{2}$, and taking into account the Langer shift $l(l + 1) \rightarrow (l + \frac{1}{2})^2$ in the centrifugal barrier [15,16] is then given by

$$Q(r; j, l, E) = \frac{\left(l + \frac{1}{2}\right)^2}{r^2} + \widetilde{V}_{\text{mod}}(r; j, l) - E.$$
(18)

For l = 0 the centrifugal barrier term and the spin-orbit potential are absent.

Considering high excitation energies E < 0 of the bound outermost electron, i.e., a principal quantum number $n \gg 1$, the respective positions of the turning points $r^{(\pm)}$ are given approximately by

$$r^{(-)} = \frac{\left(l + \frac{1}{2}\right)^2}{1 + \sqrt{1 + \left(l + \frac{1}{2}\right)^2 E}} \quad \text{if } l \ge 3,$$

$$r^{(+)} \simeq \frac{1}{-E} \left[1 + \sqrt{1 + \left(l + \frac{1}{2}\right)^2 E}\right] \quad \text{if } l \ge 1,$$
(19)

where $0 < l \ll \frac{1}{\sqrt{-E}}$. Of course, for l = 0 only a single (large) turning point $r^{(+)} = \frac{2}{-E}$ exists due to the absence of the centrifugal barrier. However, the lower turning points $r^{(-)}$ are strongly modified for l = 1,2 compared to the pure Coulomb potential case, taking into account the core polarization. For l = 1,2 the relation $r^{(-)}(l) \simeq 0.02 \times r_c(l)$ holds; that is, $r^{(-)}(l = 1) \simeq 0.03472$ and $r^{(-)}(l = 2) \simeq 0.12827$ [2]. While the analytic formula for the lower turning points $r^{(-)}(l)$ in (19), being valid for $3 \leq l \ll n$, depends only weakly on the principal quantum number n, we find from numerical calculations with the potential (14) that this also applies for l = 1,2 and $n \ge 8$. Since the cutoff $r_{so}(l)$ in (15) is substantially above those values of the lower turning points $r^{(-)}(l)$, a quasiclassical calculation of the fine-split spectrum of the bound outermost electron is reliable.

For a chosen radial quantum number n_r , the associated eigenvalues $E = E_{n,j,l} < 0$ of the outermost electron moving in the potential (14) now follow from the WKB patching condition [17–19]:

$$\nu(j,l,E) \stackrel{!}{=} \begin{cases} n_r + 1 & \text{if } l = 0, \\ n_r + \frac{1}{2} & \text{if } l > 0, \end{cases}$$
(20)

where v(j,l,E) denotes the action integral

$$\nu(j,l,E) = \frac{1}{\pi} \int_{r^{(-)}}^{r^{(+)}} dr \sqrt{-Q(r;j,l,E)}$$
$$= \frac{1}{2\pi} \oint dr \ p(r;j,l,E) \,. \tag{21}$$

Plotting the function v(j,l,E) versus $\frac{1}{\sqrt{-E}}$ for l = 0,1,2 clearly reveals a linear dependence of the form $v(j,l,E) = \frac{1}{\sqrt{-E}} + c(j,l)$ (see Fig. 2).

According to [6], for $A, B, C, D \in \mathbb{R}$, with A > 0, B > 0, C > 0, and $|D| \ll C$, the following equality holds:

$$\frac{1}{2\pi} \oint dr \sqrt{-A + \frac{2B}{r} - \frac{C}{r^2} + \frac{D}{r^3}} = \frac{B}{\sqrt{A}} - \sqrt{C} + \frac{BD}{2C\sqrt{C}}.$$
(22)

For a pure Coulomb potential $A \equiv -E$, $B \equiv 1$, $C \equiv (l + \frac{1}{2})^2$, and $D \equiv \alpha^2 g(j,l)$. The corresponding action integral then reads

$$\nu^{(C)}(j,l,E) = \begin{cases} \frac{1}{\sqrt{-E}} & \text{if } l = 0, \\ \frac{1}{\sqrt{-E}} - \left(l + \frac{1}{2}\right) + \frac{\alpha^2 g(j,l)}{2\left(l + \frac{1}{2}\right)^3} & \text{if } l > 0. \end{cases}$$
(23)

It is thus found from WKB theory that the quantum defect associated with the single-electron potential $\widetilde{V}_{mod}(r; j, l)$ is

$$\Delta_{j,l} = \lim_{E \to 0^{-}} [\nu(j,l,E) - \nu^{(C)}(j,l,E)].$$
(24)



FIG. 2. (Color online) The action integral v(j,l,E) associated with the effective single-electron potential $\widetilde{V}_{mod}(r; j,l)$ vs scaled energy $\frac{1}{\sqrt{-E}}$ for l = 0 (blue line), l = 1 (red line), l = 2 (green line), all for $j = l + \frac{1}{2}$. The curves for $j = l - \frac{1}{2}$ differ only by a tiny shift proportional to α^2 .

Ignoring spin-orbit coupling, i.e., for $\alpha = 0$, one has $\Delta_{j,l} \equiv \delta_l$, the standard quantum defect. For l = 0 the centrifugal barrier and the spin-orbit coupling term (6) are zero, so $\Delta_{j,l} \rightarrow \Delta_{\frac{1}{2},0} \equiv \delta_0$.

The dependence of the quasiclassical momentum $\sqrt{-Q(r; j, l, E)}$ on the scaled distance $\frac{r}{r_c(l)}$ is shown for l = 0, 1, 2 in Fig. 3. Clearly, it is the inner-core region $r^{(-)}(l) < r < r_c(l)$ that provides the main contribution to the quantum defect values. We find, for l = 0, 2, that changing the fitting parameter $a_3(l)$ in (3) from its tabulated value in [5] according to the scaling prescription $a_3(l = 0) \rightarrow 0.814 \times a_3(l = 0)$ and $a_3(l = 2) \rightarrow 0.914 \times a_3(l = 2)$ leads to a slight downward



FIG. 3. (Color online) The quasiclassical momentum $\sqrt{-Q(r; j, l, E)}$ vs scaled distance $\frac{r}{r_c(l)}$ for l = 0 (dashed black), l = 1 (green), and l = 2 (red) for $E = E_{n,j,l}$, corresponding to principal quantum number n = 57 and $j = l + \frac{1}{2}$. The lower turning points $r^{(-)}(l)$ for the case of a pure Coulomb potential, for l > 0 all being greater than $0.6 \times r_c(l)$, are located well outside the core region. The main contribution to the quantum defect values in (24) thus originates from the inner-core region $r < r_c(l)$.

Quantum defect $\Delta_{j,l}$	Expt. [8]	Expt. [10]	Theory [7]	Theory (this work)
$\overline{\Delta_{\frac{1}{2}0}}$	3.1312419(10)	3.13125(2)	3.12791	3.13095
$\Delta_{\frac{1}{2},1}^{2,0}$	2.6549831(10)	NA	2.65795	2.65197
$\Delta_{\frac{3}{2}1}^{2,1}$	2.6417735(10)	NA	2.64399	2.63876
$\Delta_{\frac{1}{2}1}^{2,1} - \Delta_{\frac{3}{2}1}$	0.0132096(14)	NA	0.01396	0.01321
$\Delta_{\frac{3}{2},2}^{2,1}$	1.3478971(4)	1.34789(2)	1.35145	1.34851
$\Delta_{\frac{5}{2}2}^{2,2}$	1.3462733(3)	1.34626(2)	1.34628	1.34688
$\Delta_{\frac{3}{2},2}^{2,2} - \Delta_{\frac{5}{2},2}$	0.0016238(5)	0.00163(3)	0.00517	0.00163

TABLE III. The values of quantum defect $\Delta_{i,l}$ associated with the Rydberg level n = 57 for l = 0, 1, 2.

constant shift of the WKB-quantum defect. As a result of this change, the calculated WKB-quantum defect $\Delta_{l\pm\frac{1}{2},l}$ then agrees well with the spectroscopic data (see Table III). Such a change of $a_3(l)$ does *not* affect the fine-splitting values $\Delta E_{n,l}$ though. We also find that the dependence of the fine splitting $\Delta E_{n,l}$ on the principal quantum number *n* is well described by (17) for all $n \ge 8$ (see Tables I and II).

In actual fact, for $r^{(+)} \gg r^{(-)}$, which is a criterion that is always met for high excitation energies $\sqrt{-E} \simeq 0$ of the outermost electron, the uniform Langer-WKB wave function $U_{n,j,l}^{(WKB)}(r)$ [19,20], with $r^{(+)}$ considered as the only turning point, describes the numerical solution $U_{n,j,l}(r)$ to the radial differential equation (8) under the influence of the effective modified single-electron potential (14) rather accurately [21]. Only very near to the second turning point $r^{(-)}$, at a distance smaller than $r_{so}(l)$, does the Langer-WKB wave function $U_{n,j,l}^{(WKB)}(r)$ cease to be a good approximation to the numerical solution $U_{n,j,l}(r)$ of the radial Schrödinger equation (8) [21].

III. CONCLUSIONS

In this work we reported a significant discrepancy between experiment [8,10] and highly accurate variational calculations [7] of the spectrum of Rydberg states of ⁸⁷Rb on the energy scale of the fine splitting. We discussed that the usual aposteriori adding of the relativistic spin-orbit potential to the effective single-electron potential governing the outermost electron of alkali atoms is indeed inconsistent inside the inner atomic core region. In the absence of a full microscopic theory that lumps all many-body interactions together with the relativistic corrections into an effective single-electron potential in a consistent manner, we suggested a modified effective single-electron potential, cf. (14), that enables a correct description of the spectrum of Rydberg states on the fine-splitting scale in terms of a simple WKB-action integral for all principal quantum numbers $n \ge 8$. Modern precision spectroscopy of highly excited Rydberg states thus enables the probing of the multielectron correlation problem of the ionic core of alkali atoms. This is certainly a fascinating perspective for further experiments and theoretical studies.

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