

Autodetaching state of F^- and autoionizing states of F^\dagger

A. K. Edwards and D. L. Cunningham

Department of Physics and Astronomy, University of Georgia, Athens, Georgia 30602

(Received 1 October 1973)

The collisional excitation method is used to excite an autodetaching state of F^- and several autoionizing states of neutral F. The F^- state lies at 14.85 ± 0.04 eV above the neutral-fluorine ground state and has as its dominant configuration $2p^4(^1D)3s^2^1D$. The observed autoionizing states of fluorine have configurations $(^1D)n'l'$ and $(^1S)nl''$.

In two recent papers^{1,2} we reported measurements on autodetaching states of O^- and Cl^- and autoionizing states of O. The same apparatus and procedure has been used to observe the autoionizing states of neutral fluorine and an autodetaching state of F^- .

An F^- ion beam is extracted from a duoplasmatron, accelerated, momentum analyzed by a magnet, and focussed into a differentially pumped chamber containing helium. The electrons ejected from the region of the F^- on He collisions are energy analyzed to obtain an electron spectrum. The electron spectrum consists of a smoothly varying background due to electrons stripped from the ion beam plus peaks due to electrons ejected from autodetaching states excited in the collisions. Such a peak appears in the spectrum shown in Fig. 1 corresponding to an F^- autodetaching transition at 14.85 ± 0.04 eV. Calculations by Matese *et al.*³

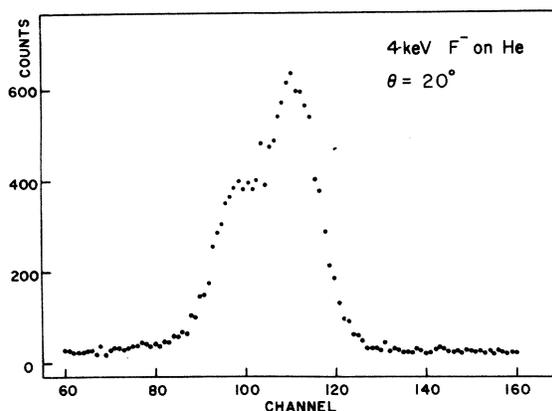


FIG. 1. Electron spectrum produced by collisions of 4-keV F^- on He. The doublet structure is due to the splitting of the neutral fluorine $^2P_{3/2,1/2}$ ground-state term. Each channel corresponds to approximately 4 meV.

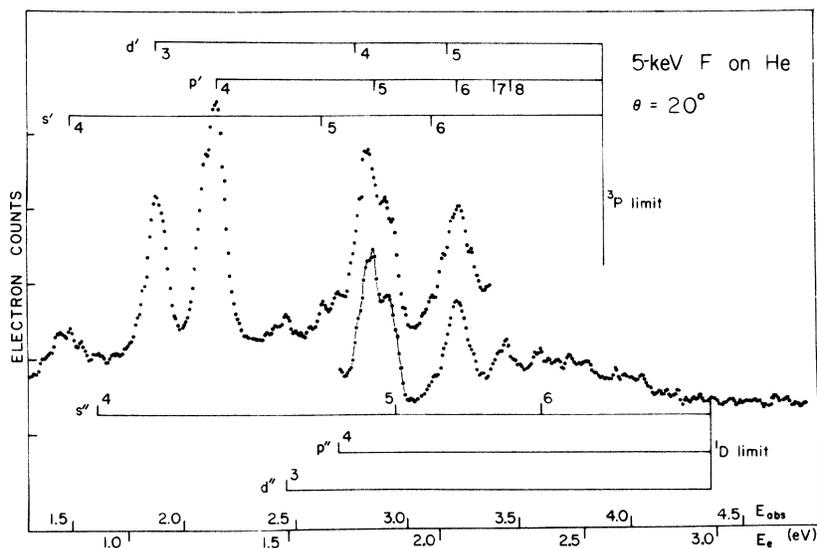


FIG. 2. Electron spectrum produced by collisions of 5-keV neutral F on He. E_{obs} is the transition energy measured in the laboratory frame and E_e is the transition energy in the frame of the neutral-fluorine atom.

TABLE I. Autoionizing levels of fluorine. The energies are measured with respect to the ${}^2P_{3/2}$ ground state of the neutral fluorine atom.

Fluorine level	Huffman <i>et al.</i> , ^a (eV)	Present work (eV) ± 0.05 eV
$({}^1D) 4s' {}^2D$	18.184	18.18
5s'	19.035	19.02
6s'	19.404	19.42
$({}^1D) 4p'$		18.66
5p'		Unresolved
6p'		19.50
$({}^1D) 3d'$	18.466	18.48
	18.489	
4d'	19.145	Unresolved
	19.155	
$({}^1S) 4s'' {}^2S$		20.88
5s''		Unresolved
6s''		22.37
$({}^1S) 3p'' {}^2P$		20.11
4p''		21.67
$({}^1S) 3d'' {}^2D$		21.50

^a Reference 4.

predict the $2p^4({}^1D)3s^2 {}^1D F^-$ state to be at 14.85 eV. The structure in the peak is due to the fluorine 2P ground-state splitting of 0.05 eV. More structure in the electron spectrum occurs about $\frac{1}{2}$ eV lower in energy; however, its position is too uncertain to attempt identification with another F^- state.

In order to observe the autoionizing states of neutral fluorine, a stripping cell is placed directly in front of the collision chamber to form a beam of F atoms. The electron spectrum from collisions of neutral F with helium is measured. Figure 2 shows a typical spectrum. The energy scale is calibrated using the $({}^1D)3d'$ state as measured by Huffman and co-workers⁴ in a photoabsorption experiment. They were able to resolve two terms for this configuration. Our resolution was not good enough to separate the terms, so a mean value for the two was used. For the states with the configuration $({}^1D)nl'$ the transition was into the $F^+({}^3P)$ continuum, and for the $({}^1S)nl''$ configurations it was into the $F^+({}^1D)$ continuum. The energies of the autoionizing levels are listed in Table I.

[†]Work supported in part by the Air Force Office of Scientific Research and the Research Corporation.

¹A. K. Edwards and D. L. Cunningham, *Phys. Rev. A* **8**, 168 (1973).

²D. L. Cunningham and A. K. Edwards, *Phys. Rev. A* **8**, 2960 (1973).

³John J. Matese, Steven P. Rountree, and Ronald J. W. Henry (unpublished). We thank Dr. Henry for the results of these calculations prior to publication.

⁴R. E. Huffman (private communication). We are indebted to Dr. Huffman for communicating these results to us prior to publication.