

**Estimates of rates for dissociative recombination of  $\text{NO}_2^+ + e^-$  via various mechanisms**Daniel J. Haxton<sup>1</sup> and Chris H. Greene<sup>2</sup><sup>1</sup>*Lawrence Berkeley National Laboratory, Chemical Sciences, Berkeley, California 94720, USA*<sup>2</sup>*Department of Physics, Purdue University, West Lafayette, Indiana 47907, USA*

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We estimate rates for the dissociative recombination (DR) of  $\text{NO}_2^+ + e^-$ . Although accurate excited-state potential-energy curves for the excited states of the neutral are not available, we estimate that the  $1^2\Phi_g$  and the  $1^2\Pi_g$  states of the neutral may intersect the ground-state cation potential-energy surface near its equilibrium geometry. Using fixed nuclei scattering calculations, we estimate the rate for direct DR via these states and find it to be significant. We also perform approximate calculations of DR triggered by the indirect mechanism, which suggest that the indirect DR rate for  $\text{NO}_2$  is insignificant compared to the direct rate.

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**I. INTRODUCTION**

$\text{NO}_2$  is a fundamental molecule in physical chemistry that is present in appreciable quantities in the Earth's atmosphere. It has remained one of the very few simple triatomics for which the dissociative recombination process has received virtually no attention from either experiment or theory. The present work is intended to provide a pilot study of this molecule to build on other recent progress in understanding triatomic and polyatomic dissociative recombination (DR) [1]. Vital information about  $\text{NO}_2$  has been gleaned from extensive pioneering studies by Grant and coworkers [2–4]. These publications studied in depth the Rydberg spectroscopy of low-lying vibrational states and found that both predissociation and autoionization processes contribute to the lifetime of these states, in varying ratios depending upon the vibrational state in question. One key finding in particular [4] is that bending excitation enhances predissociation. Moreover, these studies provide evidence that the Born-Oppenheimer potential-energy surfaces of certain valence electronic states of the neutral cross that of the ground state of the cation near the Franck-Condon region of the latter.

The characteristics of molecular Rydberg states are pertinent to the process of DR [5–7],  $AB^+ + e^- \rightarrow A + B$ . Predissociation of Rydberg states by valence states indicates that the direct mechanism [8,9] can contribute to DR; the direct mechanism is competitive only when some excited Born-Oppenheimer electronic states of the neutral are isoenergetic with the ground state of the cation within the Franck-Condon region of the latter. Such crossings of potential-energy surfaces are responsible both for predissociation of Rydberg states and dissociative recombination via the direct mechanism.

The observation of vibrational autoionization in  $\text{NO}_2$  indicates that the indirect mechanism [10] may also be important. Most studies thus far on DR via the indirect mechanism (for example, Refs. [11–14]) have concentrated on molecules with light atomic constituents. We anticipate that indirect DR proceeds violently and may lead to the severing of multiple chemical bonds in the target [15,16].  $\text{NO}_2^+$  is made of heavy atomic constituents that are quite strongly bonded, having double bonds between the nitrogen and each oxygen, and is therefore a test case for this picture.

Recently, considerable theoretical progress has been made in understanding the indirect mechanism and accurately calcu-

lating rates for it [14,17]. We apply recent theories [18] to the DR of  $\text{NO}_2^+$  in reduced dimensionality in an effort to quantify any possible contribution of the indirect mechanism to the DR rate. Grant's studies show that vibrational autoionization is significant for low-lying vibrational states of the cation; thus, the implication of Grant's experiments to the question of DR is unclear.

**II. STRUCTURE CALCULATIONS FOR THE DIRECT MECHANISM**

Several authors have looked at the high-lying electronic states of  $\text{NO}_2$ , although none has explicitly addressed the difficult question of the energy of the high-lying neutral excited states relative to that of the cation. Katagiri and Kato [19] performed structure calculations on  $\text{NO}_2$  in the course of a treatment of the photodissociation of this molecule. They analyzed the six lowest doublet electronic states of the neutral (i.e., lowest at the equilibrium geometry of the neutral). Kurkal *et al.* [20] calculated potential-energy surfaces for the lowest two. Takeshita and Shida [21] calculated many low-lying electronic states of the cation in the course of a study of photoionization of  $\text{NO}_2$ , and also analyzed the vibrational structure.

The lowest two states are the ground  $1^2A_1$  state, which has a bent equilibrium geometry, and the  $1^2B_1$  state; these both become the  $X^2\Pi_u$  state in  $d_{\infty H}$  geometry. The next four states are relevant to the present study. They are the first and second states of  $^2B_2$  and  $^2A_2$  symmetry. The  $1^2A_2$  and  $^2B_2$  become the  $1^2\Phi_g$  state at  $d_{\infty H}$  geometry, and the  $2^2A_2$  and  $^2B_2$  become the  $1^2\Pi_g$  state at  $d_{\infty H}$  geometry.

The ground-state configuration of  $\text{NO}_2^+$  (using labels appropriate to  $c_{2v}$  point-group symmetry) is  $(1a_1^2 2a_1^2 1b_2^2 3a_1^2 2b_2^2 4a_1^2 3b_2^2 5a_1^2 1b_1^2 4b_2^2 1a_2^2)$ , which is just like the  $\text{CO}_2$  molecule; the neutral has the 23rd electron in the  $6a_1$  orbital. The  $1^2A_2$  and  $^2B_2$  states are described by single excitations from the neutral,  $1a_2$  and  $4b_1 \rightarrow 6a_1$ , respectively, while the  $2^2A_2$  and  $^2B_2$  are described by  $4b_1$  and  $1a_2 \rightarrow 2b_1$ , respectively. Relative to the cation, therefore, all of these states are described by double excitations from the  $\text{NO}_2^+ + e^-$  continuum, and we expect them to exist as Feshbach resonances for regions of nuclear geometry in which they are electronically unbound. As such, they provide good candidates for direct DR.

### III. STRUCTURE CALCULATIONS

We have performed a set of bound-state electronic structure calculations using the COLUMBUS quantum chemistry suite [22–26] to analyze the crossings of the potential-energy surfaces of these states with that of the cation. These are performed at the configuration-interaction with all singles and doubles (CISD) level of theory and Dunning’s augmented correlation-consistent polarized valence triple zeta (aug-cc-pvtz) basis set [27] using a complete active space self-consistent field (CASSCF) reference space with orbitals obtained from a multiconfiguration self-consistent field (MCSCF) calculation. We include 14 valence molecular orbitals, leaving out only the highest energy  $5b_2$  ( $6\sigma_u$ ) orbital; we exclude the  $1-3a_1$  and  $1-2b_2$  orbitals from the active space, keeping them doubly occupied, giving CAS(12/9) and (13/9) for the cation and neutral calculations.

For the MCSCF calculation that defines the orbitals, we minimize a weighted average of the ground electronic state cation energy and the energies of the first seven states of the neutral (i.e., the states listed above plus the  ${}^2\Sigma_g^+$ ). In this calculation, the weighted average of the energies of the included states is minimized with respect to variations of the orbital and configuration coefficients. The balancing of the  $(N-1)$ - and  $N$ -electron energies is a difficult problem, but crucial to the question of DR via the direct mechanism. In order to provide an estimate of the uncertainty in these relative energies, we vary the weighted average; for the seven neutral states, we use a factor of 1, and for the cation state, we use factors of 3.5, 7, and 14 in the weighted average.

The results are shown in Fig. 1. We find that the potential-energy surfaces of the four high-lying electronic states listed above—the  $1$  and  $2 {}^2A_2$  and  $2 {}^2B_2$  states, which correspond to the  ${}^2\Phi_g$  and  ${}^2\Pi_g$  states at linear geometry—appear to have crossings with the cation potential-energy surfaces within the Franck-Condon region of the latter.

This bound-state structure calculation produces not only eigenvalues approximating the location of metastable states but also discretized continuum states that are unconverged artifacts [28]. One spurious root is removed from our plot of bending potential curves in Fig. 1.

### IV. SCATTERING CALCULATIONS FOR THE DIRECT MECHANISM

Scattering calculations are performed using the UK polyatomic  $R$ -matrix code [29]. The GTOBAS [30] code is used to obtain a set of diffuse Gaussians used for the exterior part of the wave function. The same 14 orbital valence reference space defines the target states of the cation. These orbitals emerge from an improved virtual orbital calculation [31] using neutral Hartree-Fock orbitals. The Fock operator of which the improved virtual orbitals are eigenfunctions is defined as an average of that for the neutral and the cation.

All four states corresponding to the  $1 {}^2\Pi_g$  and  $1 {}^2\Phi_g$  states at linear geometry are observed as Feshbach resonances in these calculations. As might be expected, because they couple to the  $p$ -wave continuum, the components correlating with the  ${}^2B_2$  point group take on a larger width than do their  ${}^2A_2$  counterparts, which couple only to partial waves with

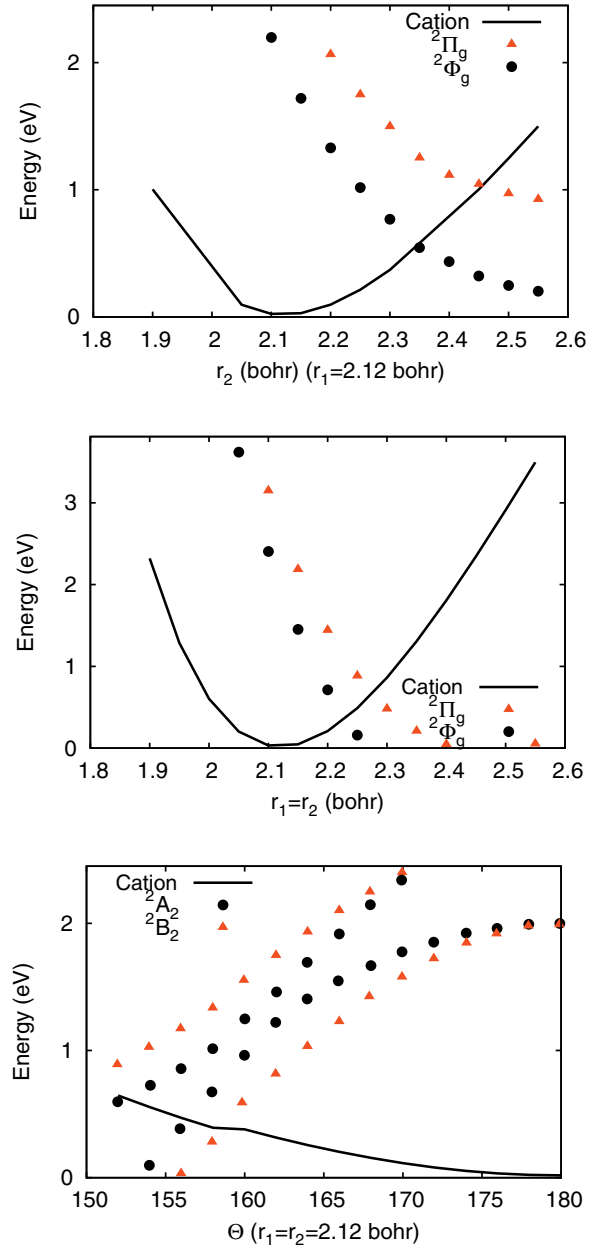


FIG. 1. (Color online) Structure calculations for the direct mechanism as described in the text.

$l = 2$  and higher in  $c_{2v}$  symmetry. As discussed above, the lower-energy  $1 {}^2\Phi_g$  state (the  $1 {}^2A_2$  and  $1 {}^2B_2$  states) becomes isoenergetic with the cation closer to the equilibrium geometry of the neutral than is the  $1 {}^2\Pi_g$  state.

### V. ESTIMATE OF DIRECT DR CROSS SECTION USING THE REFLECTION PRINCIPLE

It is possible to estimate the magnitude of the direct DR rate for  $\text{NO}_2 + e^-$  via the components of the  $1 {}^2\Phi_g$  state, which crosses the cation state closer to its Franck-Condon region. To accomplish this, three-dimensional complex-valued splined representations of the location and width are developed for this  ${}^2B_2$  Feshbach resonance. The location and width

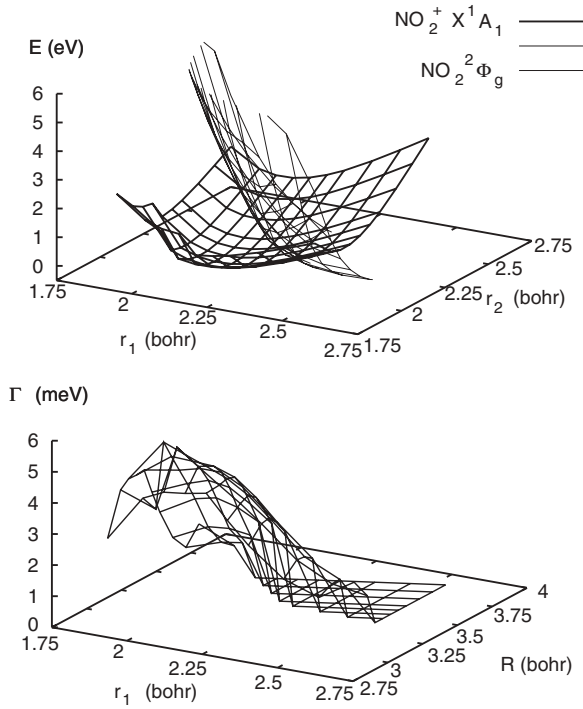


FIG. 2. Location (top) and width (bottom) of the  $1^2B_2$  ( $1^2\Phi_g$ ) Feshbach resonance for linear geometry as a function of bond lengths.

of the calculated resonance at linear geometry are shown in Fig. 2.

The present estimate is based on the standard first-order perturbative expression [8] for the DR cross section, with the assumption that the survival probability is close to unity here. We use the reflection principle [32] approximation to the bound-continuum matrix element,

$$\sigma_{\text{DR}}(E) = \frac{4\pi^2}{k_i^2} \sum_f |\langle \Psi_i | V_{\bar{a}} | \chi_f(E) \rangle|^2, \quad (1)$$

$$\approx \int d\vec{R} \delta(E - \langle \hat{T} \rangle - V(\vec{R})) |\Psi_i(\vec{R}) V_{\bar{a}}|^2. \quad (2)$$

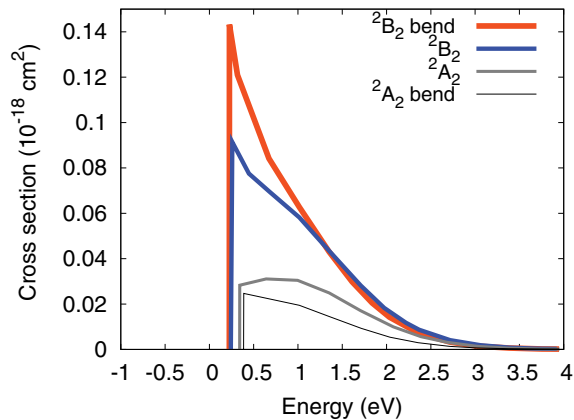


FIG. 3. (Color online) Direct DR cross section vs incident electron energy calculated for DR via the  $1^2\Phi_g$  state.

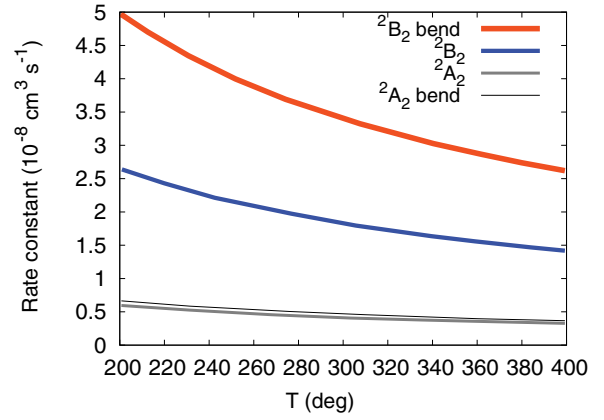


FIG. 4. (Color online) Thermally averaged rate coefficient for direct DR via the  $1^2\Phi_g$  state.

The ground vibrational state is approximated as a Gaussian in the normal modes, but the asymmetric stretch has been omitted despite having calculated three-dimensional surfaces. This omission allows us to treat the  $^2A_2$  and  $^2B_2$  components of the  $1^2\Phi_g$  state separately.

Figures 3 and 4 shows the results using a two-dimensional vibrational treatment in both the bend and symmetric stretch coordinates, and a one-dimensional treatment including the bend degree of freedom only. The static degree(s) of freedom are fixed at their equilibrium values. Evidently, the cross sections via  $^2B_2$  are higher; this follows from the larger electronic decay width of this state.

We expect the bending degree of freedom to be most important for the direct DR mechanism because it is in that degree of freedom that we find crossings of the neutral and cation states close to the minimum of energy of the Born-Oppenheimer potential-energy surface. Indeed, the cross sections obtained from the treatment with only the bending mode are similar to those calculated using two degrees of freedom, validating this hypothesis. These results should be regarded as approximate because there is considerable uncertainty in the exact location of the cation-neutral crossings.

## VI. INDIRECT MECHANISM FORMULATION OF DIRECT DR VIA RYDBERGS

We follow the method originally outlined in Ref. [18]. This method uses Siegert states in place of the bound vibrational states of the usual vibrational frame transformation. These vibrational states  $\chi_i$  are not orthogonal with respect to the usual Hermitian inner product. In the present case, as in our previous work on the high-energy peak of  $\text{HeH}^+$  [17], we include cation channels that are energetically open. An improved formula is used to calculate the unitary defect  $D_i$  of the  $S$  matrix for an electron incident in channel  $i$ , that is, not

$$D_i = 1 - \sum_j |S_{ij}|^2, \quad (3)$$

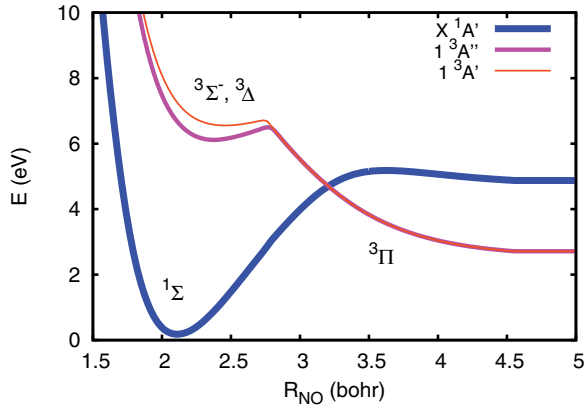


FIG. 5. (Color online) Born-Oppenheimer electronic states of  $\text{NO}_2^+$  included in the present indirect calculation: energies for  $r_2 = 2.12a_0$ , linear geometry, with respect to  $r_1$ .

but

$$D_i = 1 - \sum_{jk} S_{ij} U_{jk} S_{ik}^*, \quad (4)$$

$$U_{jk} = \int dR \chi_i(R) \chi_j(R)^*. \quad (5)$$

Following the previous work using our method, we define the DR cross section as proportional to the unitary defect of the frame-transformed  $S$  matrix.

Molecular orbitals are calculated for the cation using the MOLPRO quantum chemistry suite [33]. The calculation treats 16 electrons in 11 orbitals, with only the  $1s$  orbitals constrained to be doubly occupied, and minimizes the average energy of the  $1A'$ ,  $3A'$ , and  $3A''$  states with respect to variations in the orbital and configuration coefficients. The first is the ground state at the equilibrium geometry. The next are the lowest triplet states  $A'$  and  $A''$  symmetry. Within the Franck-Condon region in  $d_{\infty H}$  geometry, these are the  $1^3\Sigma_u^-$  and one component of the  $1^3\Delta_u$  state. These states change discontinuously where they are crossed by the  $1^3\Pi_g$  state, with each crossing being an actual intersection due to symmetry. The potential-energy curves of these states are plotted in Fig. 5. We use these orbitals in another calculation using the UK  $R$ -matrix codes, including these three target states in the scattering calculation. We obtain the  $R$  matrix from these calculations and solve for the multichannel quantum defect theory (MQDT)  $S$  matrix. We find that the calculation is very well converged with the three target states and a sizable virtual orbital basis included for penetrating configurations. We include  $s$ -,  $p$ -,  $d$ -, and  $f$ -wave scattering orbitals and obtain a multichannel  $S$  matrix with all such partial waves for each of the three included target states.

This treatment is based on the three electronic state MQDT  $S$  matrix calculated at zero energy in the  $1^1A'$  channel (such that the triplet channels are closed) at the equilibrium geometry

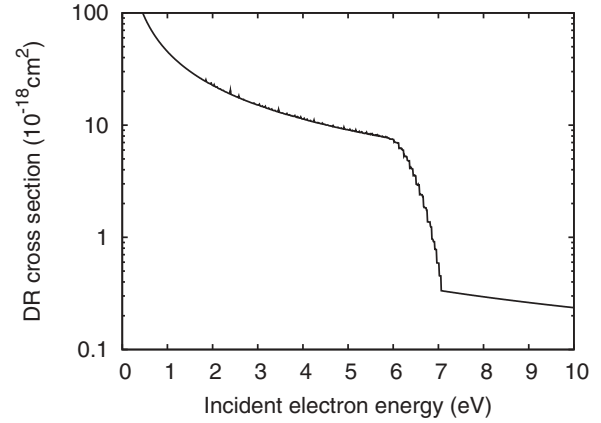


FIG. 6. Dissociative recombination cross section from a one-dimensional (1D) frame-transformation calculation using energy- and  $R$ -independent quantum defects.

of the cation. A very simple treatment is adopted in which the  $R$  dependence of the quantum defect is neglected (and the energy dependence is likewise neglected). Such a treatment was sufficient to yield a sensible calculation of the high-energy DR peak of  $\text{HeH}^+$  [17]. We include only the motion of single bond stretch at linear geometry. The results are shown in Fig. 6.

## VII. CONCLUSION

We have presented estimates of dissociative recombination rates of  $\text{NO}_2^+$ . Several results analyzing the contribution of direct and indirect processes to dissociative recombination of  $\text{NO}_2^+ + e^-$  were presented. These results compared the relative contribution of the direct and indirect mechanisms that drive dissociative recombination. These calculations in reduced dimensionality indicate that the direct DR mechanism is likely to dominate the indirect, the former of which is driven by intersections of neutral Born-Oppenheimer potential-energy curves with that of the cation. We find that the  $1^2\Phi_g$  and the  $1^2\Pi_g$  electronic states are good candidates for states that drive the direct mechanism. There is considerable uncertainty in the present results, both due to the difficulty in locating the aforementioned crossings and due to the reduced dimensionality of the calculations. However, these results provide guidance for experimental studies of dissociative recombination in these molecules and also provide information that is useful in the description of the decay channels of Rydberg states of the molecule, which is a topic that has received significant interest in the literature.

## ACKNOWLEDGMENT

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- [1] N. Douguet, A. E. Orel, C. H. Greene, and V. Kokoouline, *Phys. Rev. Lett.* **108**, 023202 (2012).  
 [2] F. Campos, Y. Jiang, and E. Grant, *J. Chem. Phys.* **93**, 7731 (1990).

- [3] G. Bryant, Y. Jiang, M. Martin, and E. Grant, *J. Chem. Phys.* **101**, 7199 (1994).  
 [4] H. Matsui and E. Grant, *J. Chem. Phys.* **104**, 42 (1995).

- [5] M. Larsson and A. E. Orel, *Dissociative Recombination of Molecular Ions* (Cambridge University Press, New York, 2008).
- [6] M. Larsson, *Annu. Rev. Phys. Chem.* **48**, 151 (1997).
- [7] V. Kokoouline, N. Douguet, and C. H. Greene, *Chem. Phys. Lett.* **507**, 1 (2011).
- [8] T. F. O'Malley, *Phys. Rev.* **150**, 14 (1966).
- [9] T. F. O'Malley and H. S. Taylor, *Phys. Rev.* **176**, 207 (1968).
- [10] S. L. Guberman, *Phys. Rev. A* **49**, R4277 (1994).
- [11] A. Larson and A. E. Orel, *Phys. Rev. A* **72**, 032701 (2005).
- [12] S. Morisset, L. Pichl, A. E. Orel, and I. F. Schneider, *Phys. Rev. A* **76**, 042702 (2007).
- [13] A. Giusti-Suzor, J. N. Bardsley, and C. Derkits, *Phys. Rev. A* **28**, 682 (1983).
- [14] R. Curik and C. H. Greene, *Phys. Rev. Lett.* **98**, 173201 (2007).
- [15] J. Glosik, R. Plasil, V. Poterya, P. Kurdna, J. Ruzs, M. Tichy, and A. Pysanenko, *J. Phys. B* **34**, L485 (2001).
- [16] V. Kokoouline and C. H. Greene, *Phys. Rev. A* **68**, 012703 (2003).
- [17] D. J. Haxton and C. H. Greene, *Phys. Rev. A* **79**, 022701 (2009).
- [18] E. L. Hamilton and C. H. Greene, *Phys. Rev. Lett.* **89**, 263003 (2002).
- [19] H. Katagiri and S. Kato, *J. Chem. Phys.* **99**, 8805 (1993).
- [20] V. Kurkal, P. Fleurat-Lessard, and R. Shinke, *J. Chem. Phys.* **119**, 1489 (2003).
- [21] K. Tkeshita and N. Shida, *J. Chem. Phys.* **116**, 4482 (2002).
- [22] H. Lischka, R. Shepard, F. B. Brown, and I. Shavitt, *International Journal of Quantum Chemistry: Quantum Chemistry Symposium* (Wiley, New York, 1981).
- [23] R. Shepard, I. Shavitt, R. M. Pitzer, D. C. Comeau, M. Pepper, H. Lischka, P. G. Szalay, R. Ahlrichs, F. B. Brown, and J. Zhao, *Int. J. Quantum Chem., Quantum Chem. Symp.* **22**, 149 (1988).
- [24] H. Lischka, R. Shepard, R. M. Pitzer, I. Shavitt, M. Dallos, T. Müller, P. G. Szalay, M. Seth, G. S. Kedziora, S. Yabushita *et al.*, *Phys. Chem. Chem. Phys.* **3**, 664 (2001).
- [25] H. Lischka, T. Müller, P. G. Szalay, I. Shavitt, R. M. Pitzer, and R. Shepard, *Wiley Interdiscip. Rev.: Comput. Mol. Sci.* **1**, 191 (2011).
- [26] H. Lischka, R. Shepard, I. Shavitt, R. M. Pitzer, M. Dallos, T. Müller, P. G. Szalay, F. B. Brown, R. Ahlrichs, H. J. Böhm *et al.*, computer code COLUMBUS, Columbus, An *Ab Initio* Electronic Structure Program (2012), release 7.0.
- [27] J. T. H. Dunning, *J. Chem. Phys.* **90**, 1007 (1989).
- [28] A. U. Hazi and H. S. Taylor, *Phys. Rev. A* **1**, 1109 (1970).
- [29] J. Tennyson and L. A. Morgan, *Philos. Trans. R. Soc. London A* **357**, 1161 (1999).
- [30] F. Alexandre, J. D. Gorfinkiel, L. A. Morgan, and J. Tennyson, *Comput. Phys. Commun.* **144**, 224 (2002).
- [31] W. J. Hunt and W. A. Goddard, III, *Chem. Phys. Lett.* **3**, 414 (1969).
- [32] S.-Y. Lee, R. C. Brown, and E. J. Heller, *J. Phys. Chem.* **87**, 2045 (1983).
- [33] H.-J. Werner, P. J. Knowles, R. Lindh, F. R. Manby, M. Schütz, P. Celani, T. Korona, A. Mitrushenkov, G. Rauhut, T. B. Adler *et al.*, computer code MOLPRO, version 2006.1 (2006), <http://www.molpro.net>.