# Multiconfiguration Dirac-Hartree-Fock energy levels and transition probabilities for W XXXVIII

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Energies, lifetimes, and wave-function compositions have been computed for all levels of  $4p^64d$ ,  $4p^64f$ , and  $4p^54d^2$  using single and double excitations from a multireference set (SD-MR) to generate expansions for the multiconfiguration Dirac-Hartree-Fock (MCDHF) approximation. An extended version of the general relativistic atomic structure package, GRASP2K, was used to deal with configuration state functions with as many as six open shells and with configurations containing as many as three *f* electrons. *E*1, *E*2, *M*1 transition probabilities are reported for transitions between the levels as supplemental material. Results are compared with other theory and with experiment, when available.

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## I. INTRODUCTION

Tungsten (W) is of great interest as a plasma wall material in the development of future tokamaks. Needed are atomic data for highly ionized tungsten atoms that may appear as impurities in hot plasmas. The feasibility of accurate theoretical studies for complex atoms with open f shells was investigated recently by Gaigalas *et al.* [1] and applied to W<sup>+24</sup>. Without experimental data, the accuracy of transition data could only be related to the gauge dependence of the theoretical results.

Prominent lines in electron-beam ion-trap (EBIT) spectra of Rb-like  $W^{+37}$  to Cu-like  $W^{+45}$  ions have been measured for wavelengths and intensities by Utter *et al.* [2]. In their paper the identification of lines depended on theoretical predictions of wavelengths from *ab initio* calculations performed by Fournier [3] using a fully relativistic parametric potential code [4,5]. Quantum electrodynamic (QED) corrections were not included.

With some experimental data now available, the Rb-like spectrum of  $W^{+37}$  is an excellent test case for theory. The  $4p^{6}4d$  ground state and 10 excited levels of the  $4p^{6}4f$  and  $4p^54d^2$  configurations that can be reached through electric dipole (E1) transitions from the ground configuration have been classified by Kramida and Shirai [6] from available observed wavelengths [7] through a least-squares process developed by Cowan [8]. Thus some energy levels and their uncertainty are accurately known. All configurations are part of the n = 4 complex with unfilled 4d and 4f subshells and there are strong interactions within this complex. Fournier [3] included a small set of configurations in his wave-function expansion with very few from the n = 4complex. Bogdanovich *et al.* [9] applied a quasirelativistic (QR) method that included an algorithm for deciding on the important correlation corrections to these levels of W<sup>+37</sup> and reported their lifetimes. Not included were other quantum electrodynamic (QED) corrections.

In the present paper we report all the energy levels of the even parity  $4p^{6}4d$  ground configuration as well as the  $4p^{5}4d^{2}$ 

and  $4p^{6}4f$  odd parity excited configurations. Transition probabilities (*E*1, *E*2, *M*1) for transitions between these levels are also reported. Accuracy is estimated by comparing the computed energy levels with those derived from observation and the agreement of length and velocity line strengths for *E*1 and *E*2 transition probabilities. The calculations were performed using a version of the general relativistic atomic structure package, GRASP2K [10]. In a revised version [11], the calculations of the spin-angular parts of matrix elements relied on a second quantization method in coupled tensorial form and quasispin technique [12,13]. This revision not only overcame some limitations on the complexity of the cases that could be executed successfully by GRASP2K but also increased the efficiency and the speed of angular integrations.

### **II. METHOD OF CALCULATIONS**

The GRASP2K computer package and its revision are based on the multiconfiguration Dirac-Hartree-Fock (MCDHF) approach taking into account relativistic and QED corrections [13–15].

In this approach, an atomic state function (ASF) of parity *P* and total angular momentum *J*,  $\Psi(\gamma PJ)$ , is given by a linear combination of configuration state functions (CSFs) with the same parity,  $\Phi(\gamma_i PJ)$ , that is,

$$\Psi(\gamma P J) = \sum_{i} c_i \Phi(\gamma_i P J).$$
(1)

The multiconfiguration energy functional is based on the Dirac-Coulomb Hamiltonian, namely (in a.u.),

$$H_{\rm DC} = \sum_{j=1}^{N} [c \boldsymbol{\alpha}_j \, \boldsymbol{p}_j + (\beta_j - 1)c^2 + V(r_j)] + \sum_{j < k} \frac{1}{r_{jk}}, \qquad (2)$$

where  $\alpha$  and  $\beta$  are 4 × 4 Dirac matrices, **p** is the momentum operator,  $V(r_j)$  and  $1/r_{jk}$  are the electrostatic electron-nucleus and electron-electron interactions, respectively. In all the calculations reported here, the nuclear charge distribution was modeled by the two-component Fermi function. A variational method was used to optimize the large and small components

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TABLE I. Summary of the extended optimal level MCDHF calculations performed indicating the range of eigenvalues and the size of the interaction matrix for each J of the group.

J	Parity	Eigenvalues	Size
3/2, 5/2	+	1,1	17 538, 27 462
1/2, 3/2	_	1-7, 1-11	40 963, 82 461
5/2,7/2	_	1-12, 1-10	111 273, 111 868
9/2, 11/2	_	1-5, 1-2	78 363, 41 334

of the radial functions that define the CSFs and their expansion coefficients in the wave function.

In the present case, there may be several CSFs that are important components to the wave-function expansion, defined in terms of orbitals with principal quantum number  $n \leq 4$ . These CSFs define a multireference set (MR). The most important corrections to the wave function arise from single and double excitations (SD) from orbitals occupied in the MR set to unfilled subshells. Because W<sup>+37</sup> is so highly ionized, the most important contributions are CSFs within the n = 4complex that interact with one or more members of the MR set. Included in the complex are CSFs with singly occupied or even unoccupied 4s subshells. Thus the excitation process needs to include excitations from the  $4s^2$  subshell. The Ni-like  $1s^2 \dots 3d^{10}$  core was treated as an inactive core. When the SD excitations are restricted to n = 4 orbitals, we shall refer to the calculation as an n = 4 active set (AS) calculation. When the maximum principal quantum number is extended to n = 5and n = 6, further corrections are included and the expansions increase in size significantly. Because the  $4p^{6}4f$  and  $4p^{5}4d^{2}$ CSFs only interact for J = 5/2 or 7/2, optimization was done in groups as reported recently for Mo VI [16], namely: (i) J = 1/2, 3/2; (ii) J = 5/2, 7/2; (iii) J = 9/2, 11/2. In this extended optimal level (EOL) calculation, the same radial functions are used in the wave-function expansion for all levels of the group.

As a final step, a relativistic configuration interaction (RCI) calculation was performed to include the transversephoton (Breit) interaction describing the transversely polarized photon contribution to the electron-electron interactions in the Coulomb gauge, the vacuum polarization (VP), and the self-energy (SE) corrections. Because the decrease in energy in going from n = 4 to n = 5 calculations was found to be rather small and contributions from n = 6 were expected to be even smaller, the rather large n = 6 expansions for odd parity were condensed to eliminate CSFs with an expansion coefficient less than 0.00001 from the Dirac-Coulomb (DC) wave functions. Table I summarizes the calculations performed for the four groups of levels by showing the J values and parity of each group, the levels that were included in the optimization process, and the size of the wave-function expansion for the n = 6 results reported as the final results for this paper.

Table II shows the contributions to the excitation energy arising from different parts of the Hamiltonian for several levels. As expected, the DC contribution is the largest, with the Breit (TP) interaction giving a significant correction. The vacuum polarization and self-energy corrections are of lesser importance. The DC contribution changes the most as the orbital set is expanded. Other corrections vary by lesser amounts. The effect of triple and quadruple excitations (TQ) from the reference set were neglected in this study. The total sum of these contributions is compared with experimental energy level [19]. These were derived from observed wavelengths by a least-squares fitting process that predicts energy levels and their uncertainties [6]. Table II also reports the difference between and the experimental and the computed energy levels (Total). In all cases, the difference is larger in magnitude than the uncertainty in the energy level and, in most cases, decreases as larger orbital sets are used in the expansion.

In Table III we present the computed energy levels and their lifetimes, along with the composition of the wave function in jj as well as LSJ coupling.

Tables IV and V report electric dipole (*E*1) transitions from the  ${}^{2}D_{3/2}$  and  ${}^{2}D_{5/2}$  levels, respectively, to the odd levels.

A similar table for the numerous E2 and M1 transitions between the odd parity levels is included with this paper as Supplemental Material [17].

### **III. COUPLINGS AND WAVE-FUNCTION COMPOSITION**

The MCDHF calculations were performed with the wave function expanded in a basis of CSFs defined in terms of orbitals that are coupled in *jj* coupling. Usually results are presented in LSJ notation. In fact, the MCHF-MCDHF database [18] is designed only for this notation. If we make the assumption that the  $nl_{-}$  and nl MCDHF radial functions are the same, it is possible to transform the wave function from the *jj*-coupling scheme to *LSJ* coupling. The revised GRASP2K contains a program, JJ2LS, that rapidly transforms 99% of the wave-function composition of a large expansion to LSJ: A complete transformation is also possible but may take considerably more computer time. The contribution of a CSF to the composition is the square of the expansion coefficient in an orthonormal basis. Table III shows the larger contributions of the 49 levels considered in this paper in both *jj* and *LSJ* coupling.

In Table III, the *jj*-coupling notation of a CSF denotes the total J for a multiply occupied shell in parentheses, whereas the resultant of coupling two subshells or, in general, a subconfiguration and a subshell, is given in square brackets. The orbital  $4d_{-}$ , for example, is a  $4d_{3/2}$  orbital whereas 4d is  $4d_{5/2}$ . In the LSJ notation, for brevity, CSFs for the  $4p^54d^2$  configuration are denoted only by the LS term and seniority of the  $4d^2$  subshell and the final LS term. Only in one case is the largest component of the wave function significantly greater in LSJ rather than in jj, namely, for the  $4p^{5}4s^{2}({}_{2}{}^{3}F)^{4}G^{o} J = 9/2$  level at 1 382 620 cm<sup>-1</sup> where the largest component in *j* is 50% whereas the largest in *LSJ* is 70%. In others they are similar or LSJ is considerably less. Fournier [3] designated labels simply by the occupation of the different subshells in *jj* coupling. Without the intermediate quantum numbers such a designation is not always unique.

A wave function or a corresponding energy level is often designated as the label of the CSF with the largest expansion coefficient. As shown in the case of Mo VI [16], labels determined in this manner may not be unique. An algorithm has been proposed for defining unique labels. Basically, for a given set of wave functions for the same J and parity, the CSF with largest expansion coefficient is used as the label

TABLE II. Dirac-Coulomb (DC), transverse photon (TP), vacuum polarization (VP), and self-energy (SE) contributions to the energy (in  $cm^{-1}$ ) as a function of the orbital set. The sum (Total) is compared with experiment and the difference (Diff.) is reported. All energies are relative to the ground state.

	DC	Breit (TP)	VP	SE	Total	Expt. [6,19]	Diff.
	(i) $4p^{6}4d^{2}D J$	= 5/2					
n = 4	1 57 565.02	-3498.50	0.60	237.28	154 304.45	$154640\pm120$	-340
n = 5	1 58 136.24	-3377.23	-1.56	236.91	154 994.36	$154640\pm120$	350
n = 6	1 57 502.42	-3384.70	-1.43	237.07	154 353.36	$154640\pm120$	-290
	(ii) $4p^54d^2({}_2{}^3F)$	$^{4}D J = 3/2$					
n = 4	1 225 740.23	-2629.03	25.15	-1456.09	1 221 680.26	$1227640\pm210$	-5960
n = 5	1 225 850.85	-2698.77	28.52	-1456.03	1 221 724.57	$1227640\pm210$	-5920
n = 6	1 225 589.32	-2685.36	26.36	-1456.45	1 221 473.87	$1227640\pm210$	-6170
	(iii) $4p^54d^2({}_2^3F$	) ${}^{4}D \ J = 1/2$					
n = 4	1 226 132.50	-3044.01	25.43	-1437.62	1 221 676.30	$1227640\pm210$	-5960
n = 5	1 226 557.12	-3099.04	28.66	-1437.66	1 222 049.08	$1227640\pm210$	-5590
n = 6	1 226 380.29	-3091.54	26.51	-1438.11	1 221 877.15	$1227640\pm210$	-5760
	(iv) $4p^54d^2({}_2{}^1G)$	) ${}^{2}F J = 5/2$					
n = 4	1 523 453.12	-6672.53	20.39	-855.28	1 515 945.70	$1508470\pm100$	7480
n = 5	1 519 890.96	-6760.73	23.38	-849.68	1 512 303.93	$1508470\pm100$	3830
n = 6	1 518 779.48	-6655.53	19.95	-850.99	1 511 292.91	$1508470\pm100$	2820
	(v) $4p^54d^2({}_2{}^3F)$	${}^{4}\!F \ J = 3/2$					
n = 4	1 557 569.50	-6449.14	42.77	-1227.78	1 549 935.35	$1542600\pm500$	7300
n = 5	1 556 800.29	-6473.29	43.23	-1227.15	1 549 143.08	$1542600\pm500$	6500
n = 6	1 555 763.55	-6391.42	41.16	-1227.61	1 548 185.68	$1542600\pm500$	5600
	(vi) $4p^54d^2({}_2^3F)$	) $^{2}D J = 5/2$					
n = 4	1 755 868.66	-9303.69	45.57	-950.21	1745660.33	$1731460\pm130$	14 200
n = 5	1 751 747.88	-9201.11	45.46	-928.33	1 741 663.90	$1731460\pm130$	10 200
n = 6	1 750 071.01	-9055.42	42.42	-924.32	1 740 133.69	$1731460\pm130$	8670
	(vii) $4p^{6}4f^{2}F$	J = 5/2					
n = 4	1 782 471.08	-8579.18	39.86	-400.93	1 773 530.83	$1758100\pm180$	15 400
n = 5	1775341.78	-8688.25	43.08	-424.31	1 766 272.30	$1758100\pm180$	8200
n = 6	1 773 419.84	-8560.95	38.57	-427.11	1 764 470.35	$1758100\pm180$	6400
	(viii) $4p^64f \ ^2F$	J = 7/2					
n = 4	1 797 776.71	-9776.58	25.23	-254.51	1 787 770.85	$1769600\pm500$	18 200
n = 5	1 788 853.62	-9797.36	28.91	-267.50	1778817.67	$1769600\pm500$	9200
n = 6	1 786 887.52	-9645.09	24.29	-267.98	1776998.74	$1769600\pm500$	7400
	(ix) $4p^54d^2({}_2^3F)$	) ${}^{4}G J = 5/2$					
n = 4	2 036 806.50	-11 442.54	212.98	-875.65	2 024 701.29	$2014500\pm300$	10 200
n = 5	2 032 712.91	-11 386.54	212.83	-879.36	2 020 659.84	$2014500\pm300$	6200
n = 6	2 031 249.97	-11 305.08	211.07	-880.79	2019275.17	$2014500\pm300$	4800
	(x) $4p^54d^2({}_2{}^3F)$	$^{2}D J = 3/2$					
n = 4	2 213 377.53	-11 531.25	230.28	-964.93	2 201 111.63	$2184310\pm220$	16 800
n = 5	2 212 133.39	-11 494.28	226.73	-963.92	2 199 901.92	$2184310\pm220$	15 590
n = 6	2 210 115.18	-11341.36	224.93	-963.84	2 198 034.91	$2184310\pm220$	13720
	(xi) $4p^54d^2({}_2{}^1G)$	) ${}^{2}F J = 7/2$	<b>•</b> ••				
n = 4	2 340 324.95	-15 454.28	233.54	-659.27	2 324 444.94	$2325500\pm500$	-1100
n=5	2 335 255.64	-15 349.49	229.57	-659.03	2319476.69	$2325500 \pm 500$	-6000
n = 6	2 3 3 3 3 8 6.3 2	-15 233.26	227.53	-658.16	2317722.43	$2325500\pm500$	-7800

for the function containing this largest component. Once a label is assigned, the corresponding CSF is removed from consideration in the determination of the next label [20]. The last remaining label for a wave function may be based on a contribution that is exceedingly small as seen for

the level at 1 742 142 cm<sup>-1</sup> labeled  $4p^54d^2({}_2{}^3P) {}^2D_{3/2}$  in Table III. The Atomic Spectra Database (ASD) [19] reports the leading percentage of the wave-function composition from calculations using the Cowan code [6,8]. Comparing the composition in Table III with ASD, we note that the

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TABLE III. MCDHF energies (in cm<sup>-1</sup>), lifetimes (in s), and wave-function composition of levels in W XXXVIII.

			Level	Lifetimes	Composition			
Label	J	Pos	$(\mathrm{cm}^{-1})$	(s)	<i>jj</i> coupling	LSJ coupling		
$4p^6 4d^2D$	3/2	1			<b>98</b> $4p_{-}^{2}4p^{4}4d_{-}$	98		
-	5/2	1	154 353	$2.53 \times 10^{-5}$	<b>98</b> $4p_{-}^{2}4p^{4}4d$	98		
$4p^5 4d^2({}_2^3F) {}^4D^o$	3/2	1	1 221 474	$8.10 \times 10^{-10}$	87 $4p_{-}^{2}4p_{-}^{3}4d_{-}^{2}(2)$	<b>42</b> + <b>23</b> ( ${}_{2}{}^{3}P$ ) <sup>4</sup> D		
1 (2 )	1/2	1	1 221 877	$2.54 \times 10^{-10}$	87 $4p^2 4p^3 4d^2(2)$	84 + 11( ${}^{1}D$ ) <sup>2</sup> P		
	5/2	2	1 354 946	$1.53 \times 10^{-8}$	<b>84</b> $4p^2 4p^3 4d_{[0]}4d$	<b>33</b> + 22( ${}^{3}P$ ) <sup>4</sup> $P$ + 15( ${}^{3}P$ ) <sup>4</sup> $D$		
	7/2	9	2 267 787	$3.10 \times 10^{-12}$	<b>85</b> $4p_4p^44d^2(4)$	45 + 31( ${}_{2}^{3}F$ ) <sup>2</sup> F + 18( ${}_{2}^{3}F$ ) <sup>4</sup> F		
${}^{4}F^{o}$	5/2	1	1 249 069	$3.91 \times 10^{-10}$	<b>93</b> $4p^2 4p^3 4d^2(2)$	<b>36</b> + <b>18</b> $({}_{2}^{1}D)^{2}D$ + <b>12</b> $({}_{2}^{3}F)^{4}G$ + <b>10</b> $({}_{2}^{3}F)^{4}D$		
	$\frac{1}{7/2}$	2	1 382 497	$2.93 \times 10^{-10}$	<b>71</b> $4p^2 4p^3 4d_{-}[1]4d$	$35 + 26({}^{3}P)^{4}D + 16({}^{3}F)^{4}D + 10({}^{1}D)^{2}F$		
	3/2	5	1 548 186	$2.16 \times 10^{-12}$	<b>59</b> $4p^2 4p^3 4d_{[3]}4d$	$32 + 22({}^{3}P)^{2}D + 12({}^{3}F)^{2}D$		
	9/2	3	1 576 287	$1.19 \times 10^{-5}$	<b>92</b> $4p_{-}^{2}4p^{3}4d^{2}(4)$	<b>38</b> + <b>29</b> $({}_{2}^{1}G)^{2}G$ + <b>18</b> $({}_{2}^{3}F)^{2}G$		
$^{2}G^{o}$	7/2	1	1 264 378	$1.95 \times 10^{-8}$	<b>97</b> $4n^2 4n^3 4d^2(2)$	$+13({}_{2}{}^{1}G)^{2}H$ 31 + 32({}^{3}F)^{4}G + 22({}_{2}{}^{1}D)^{2}F		
0	9/2	5	2 310 262	$1.93 \times 10^{-7}$ $1.34 \times 10^{-7}$	<b>95</b> $4 p - 4p^4 4d^2(4)$	$31 + 32({}_{2}T) + 32({}_{2}D)T$ $31 + 23({}_{3}C)^{2}H + 17({}_{3}F)^{4}G$		
$(4n^5 A d^2 (-^{3}P)) {}^{4}P^{0}$	3/2	2	1 318 542	$1.34 \times 10^{-9}$ 5 74 × 10^{-9}	<b>82</b> $A n^2 A n^3 A d^2 (0)$	31 + 23(20) H + 17(21) 0 $28 + 21(-15)^2 P + 16(-3P)^4 D$		
+p $+a$ $(21)1$	5/2	2	1 401 387	$3.74 \times 10^{-10}$	<b>57</b> $4n^2 4n^3 4d$ [1] $4d$	$41 + 17({}^{3}F)^{2}F + 15({}^{1}G)^{2}F$		
	$\frac{3}{2}$	5	2 1 4 2 2 4 8	$4.51 \times 10^{-11}$	<b>95</b> $4p_4p_4d_{-}[1]4d$	$41 + 17(_2T)T + 15(_2O)T$ $46 + 31(_3D)^2S + 20(_3D)^4D$		
$4 n^5 4 d^2 (3E) 4C^0$	0/2	1	2 142 240	$2.87 \times 10^{-5}$	<b>50</b> $4p_{-}4p_{-}4a_{-}[2]4d_{-}$	40 + 31(2F) + 20(2F) D $70 + 15(3F)^2C + 13(3F)^4E$		
4p 4u (2r) 0	9/2	1	1 540 051	$2.90 \times 10^{-5}$	<b>50</b> $4p_4p_4a_{-}[5]4a$ <b>03</b> $4n^2 4n^3 4d^2(4)$	70 + 13(2F) G + 13(2F) F 57 + 41( $^{1}C$ ) <sup>2</sup> H		
	5/2	2	1 340 931	$2.23 \times 10^{-13}$	<b>93</b> $4p_4p^4 4a^4$ (4) <b>84</b> $4p_4 4a^2$ (2)	$57 + 41(2G) \Pi$ 20 + 29(3E) <sup>2</sup> E + 14(1D) <sup>2</sup> E		
	3/2	9	2019273	$8.90 \times 10^{-10}$	<b>64</b> $4p_4p_4a_(2)$ <b>67</b> $4p_4p_4a_(2)$	$39 + 28(_{2}T)^{-}T + 14(_{2}D)^{-}T$		
4 = 5 + 4 + 2 (10) + 200	2/2	0	2 1 10 127	$1.37 \times 10^{-10}$	<b>97</b> $4p_4p_4a_{2}$	$40 + 19(_{2}T)^{-}G + 14(_{2}T)^{-}T$ $12 + 40(_{3}E)^{4}D + 14(_{3}E)^{2}D + 11(_{1}D)^{2}D$		
$4p^{*} 4a^{-}({}_{2}^{*}D)^{*}D^{*}$	5/2	3	1 590 088	$1.24 \times 10^{-9}$	<b>63</b> $4p_{-}^{2}4p_{-}^{2}4a_{-}[1]4a$	$13 + 40({}_{2}F)D + 14({}_{2}F)D + 11({}_{2}D)P$ $24 + 19({}_{3}D)4D + 15({}_{3}E)2D + 12({}_{3}D)2D$		
4 = 5 + 12(10) 2110	$\frac{3}{2}$	0	1 333 6/2	$1.08 \times 10^{-3}$	<b>05</b> $4p_{-}^{2}4p_{-}^{2}4a_{-}^{2}(2)$ <b>03</b> $4\pi^{2}4\pi^{3}4d$ [2] $4d$	$24 + 18({}_{2}P)^{P} + 15({}_{2}P)^{P} + 13({}_{2}P)^{P} - 13({}$		
$4p^{*} 4a^{-}({}_{2}^{*}G)^{-}H^{*}$	0/2	1	1415015	$2.80 \times 10^{-7}$	<b>95</b> $4p^{-}4p^{-}4a_{-}[5]4a$	$57 + 41({}_{2}T)G$		
200	9/2	4	2 102 095	$1.44 \times 10^{-6}$	<b>95</b> $4p_4p_4a_{2}$	$45 + 19(2G)^2G + 10(2F)^2G + 10(2F)G$		
-G.	1/2	3	1419324	$3.38 \times 10^{-5}$	$+17 4p_{-}^{2}4p_{-}^{3}4d_{-}[3]4d$ +17 4p <sup>2</sup> 4p <sup>3</sup> 4d_[2]4d	$15 + 29(2F)^{-}G + 15(2F)^{-}F + 15(2F)^{-}F + 14(2F)^{-}F$		
					r = r r m - l - l m	$+15({}_{2}{}^{3}F){}^{4}F$ $+11({}_{2}{}^{1}G){}^{2}G$		
	9/2	2	1 474 980	$1.51 \times 10^{-4}$	<b>47</b> $4p_{-}^{2}4p^{3}4d_{-}[3]4d$	<b>38</b> + <b>24</b> $({}_{2}{}^{3}F){}^{4}F$ + <b>18</b> $({}_{2}{}^{3}F){}^{2}G$		
4 5 4 12 ( 3 D) 200	1.70	2	1 401 711	0.62 10-6	$+40 4p_{-}^{2}4p_{-}^{3}4a_{-}[2]4a$	$+17(2^{\circ}G)^{\circ}H$		
$4p^{3} 4d^{2}({}_{2}{}^{3}P) {}^{2}S^{0}$	1/2	2	1 431 711	$9.63 \times 10^{-6}$	<b>80</b> $4p_{-}^{2}4p_{-}^{3}4d_{-}[2]4d$	$29 + 33({}_{2}P)^{2}P + 19({}_{2}D)^{2}P$		
$^{-}D^{\circ}$	1/2	4	1 453 386	$5.09 \times 10^{-6}$	<b>59</b> $4p_{-}^{2}4p_{-}^{3}4d_{-}[2]4d$ + <b>20</b> $4p_{-}^{2}4p_{-}^{3}4d_{-}[3]4d$	$20 + 13({}_{2}^{*}D)^{2}F + 13({}_{2}^{*}G)^{2}G + 12({}_{2}^{*}G)^{2}F + 11({}_{2}^{3}F)^{4}D$		
	5/2	4	1458234	$1.46 \times 10^{-10}$	<b>55</b> $4p^2 4p^3 4d$ [3]4d	$36 + 25({}^{3}P)^{2}D + 14({}^{1}D)^{2}D$		
	3/2	8	2 134 981	$1.02 \times 10^{-11}$	$944p 4p^44d$ [2]4d	$37 + 16({}_{2}^{3}F)^{2}D + 10({}_{2}^{3}P)^{2}D + 10({}_{3}^{3}F)^{4}F$		
	1/2	6	2 144 727	$5.78 \times 10^{-13}$	<b>87</b> $4 p 4 a^4 4 d^2(0)$	$37 + 34({}^{3}P)^{2}P + 21({}^{1}S)^{2}P$		
$4S^{o}$	3/2	4	1 465 969	$3.70 \times 10^{-9}$	<b>60</b> $4n^2 4n^3 4d$ [2] $4d$	$30 + 25({}^{3}P)^{2}D + 17({}^{3}P)^{4}P$		
$4n^5 4d^2({}_{2}^{1}G) {}^{2}F^{o}$	5/2	5	1 511 293	$6.00 \times 10^{-12}$	<b>39</b> $4n^2 4n^3 4d$ [2] $4d$	$41 + 274 n^{6} 4 f^{2} F + 13(s^{3} F)^{2} F$		
<i>ip</i> iu (20) <i>i</i>	5/2	5	1511295	0.00 × 10	$+274 p^2 4 p^4 4 f$	<b>H</b> + <b>2</b> + <b>p</b> + <b>j</b> + <b>i 3</b> (21)1		
					$+134n^24n^34d$ [1]4d			
	7/2	10	2 317 722	$5.26 \times 10^{-13}$	<b>79</b> $4n$ $4n^44d$ [1] $4d$	53 + 29( ${}^{1}G)^{2}G$		
$4n^5 4d^2({}_{2}^{1}D) {}^{2}F^{o}$	$\frac{7}{2}$	5	1 558 637	$5.20 \times 10^{-11}$ 5.07 × 10 <sup>-11</sup>	<b>68</b> $4n^2 4n^3 4d^2(2)$	$33 + 29({}^{3}P){}^{4}D + 4n{}^{6}4f{}^{2}F$		
p in $(2D)$ i	5/2	12	2 325 721	$2.76 \times 10^{-10}$	<b>97</b> $4n$ $4n^44d^2(2)$	$35 + 24({}^{3}P)^{2}D + 18({}^{3}P)^{4}D$		
$2\mathbf{p}^{o}$	1/2	3	1 587 523	$5.26 \times 10^{-12}$	$74 4 n^2 4 n^3 4 d^2(2)$	$66 + 14({}^{3}P)^{4}P + 10({}^{3}P)^{2}S$		
1	3/2	10	2 304 542	$5.20 \times 10^{-12}$ 5.73 × 10 <sup>-12</sup>	<b>90</b> $4n$ $4n^44d^2(2)$	$32 + 26({}_{2}^{1}D)^{2}D + 12({}_{2}^{3}P)^{4}P + 11({}_{2}^{3}P)^{4}D$		
$4n^5 4d^2({}^{3}P) {}^{2}P^{o}$	$\frac{3}{2}$	4	1 621 723	$1.60 \times 10^{-12}$	<b>52</b> $4n^2 4n^3 4d$ [3]4d	$52 + 23({}^{3}P)^{4}D + 20({}^{3}P)^{2}S$		
	3/2	11	2 321 806	$4.71 \times 10^{-13}$	<b>84</b> $Ap$ $Ap^4Ad$ [1] $Ad$	52 + 25(27) B + 20(27) S $52 + 19(^{3}P)^{4}S + 10(^{3}P)^{4}P$		
$4n^5 Ad^2({}^3F) {}^2F^o$	7/2	6	1645220	$4.71 \times 10^{-11}$ $4.23 \times 10^{-11}$	<b>50</b> $4 n^2 4 n^3 4 d^2(4)$	$17 + 164n^{6}4f^{2}F + 15(^{1}D)^{2}F + 14(^{3}F)^{4}D$		
<i>ip</i> iu (21) 1	//2	0	1013220	1.25 × 10	$+29 4 n^2 4 n^3 4 d^2(2)$	$+12({}_{2}^{3}P)^{4}D + 11({}_{2}^{1}G)^{2}G$		
	5/2	7	1740134	$1.07 \times 10^{-12}$	$37 4 n^2 4 n^3 4 d^2(4)$	$4 + 30({}^{3}F)^{2}D + 16({}^{3}F)^{4}F + 104p^{6}4f^{2}F$		
	5/2	,	1710151	1.07 / 10	$+184p^24p^34d^2(2)$	$+10({}_{2}^{3}F){}^{4}D + 9({}_{2}^{1}D){}^{2}F + 5({}_{2}^{1}D){}^{2}D$		
					$+144p^24n^34d$ [1]4d	$+ 5({}_{2}^{3}P)^{4}D$		
$4n^5 4d^2(a^1S)^2P^o$	3/2	6	1679761	$5.97 \times 10^{-10}$	<b>66</b> $4n^2 4n^3 4d^2(0)$	$50 + 14({}^{3}P)^{2}D + 10({}^{3}P)^{4}S$		
P 10 (0.5) 1	1/2	7	2 395 225	$3.27 \times 10^{-11}$	<b>94</b> $4n$ $4n^4 4d^2(0)$	$75 + 13(^{3}P)^{4}D$		
$4p^5 4d^2({}_2^3P) {}^2D^o$	3/2	, 7	1742142	$1.58 \times 10^{-12}$	<b>58</b> $4p^2 4p^3 4d^2(2)$	$0.4 + 20({}_{0}{}^{1}S)^{2}P + 17({}_{3}{}^{2}P)^{4}S + 16({}_{3}{}^{3}P)^{2}P$		
r (21) 2	-/-				$+264p_{-}^{2}4p^{3}4d^{2}(0)$	$+13({}_{2}{}^{1}D){}^{2}D + 12({}_{2}{}^{3}P){}^{4}D + 8({}_{2}{}^{3}P){}^{4}P$		
					$+64p_{-}^{2}4p^{3}4d_{-}[1]4d$	$+ 4({}_{2}{}^{3}F){}^{4}F + 0.5({}_{2}{}^{3}F){}^{2}D$		
	5/2	10	2 125 762	$1.72 \times 10^{-11}$	<b>95</b> 4 <i>p</i> _4 <i>p</i> <sup>4</sup> 4 <i>d</i> _[2]4 <i>d</i>	$14 + 18({}_{2}^{3}F)^{4}F + 16({}_{2}^{1}D)^{2}F + 14({}_{2}^{3}F)^{4}G$		

			I evel	Lifetimes		Composition
Label	J	Pos	$(cm^{-1})$	(s)	<i>jj</i> coupling	LSJ coupling
						$+12({}_{2}^{3}F)^{2}D + 10({}_{2}^{3}P)^{4}D$
$4p^{6} 4f^{2}F^{o}$	5/2	8	1764470	$1.21 \times 10^{-12}$	<b>53</b> $4p_{-}^{2}4p^{4}4f_{-}$	$53 + 18({}_{2}^{1}G)^{2}F$
	$\frac{1}{7/2}$	7	1776999	$1.75 \times 10^{-12}$	<b>64</b> $4p^2 4p^4 4f$	$33 + 29({}_{2}^{3}P)^{4}D + 134p^{6}4f^{2}F$
$4p^5 4d^2({}_2^3F)^2D^o$	3/2	9	2 198 035	$3.11 \times 10^{-13}$	87 $4p_{-}4p^{4}4d_{-}^{2}(2)$	$44 + 17({}_{2}^{1}D)^{2}D + 12({}_{2}^{3}F)^{4}F$
· ·· ·	5/2	11	2302432	$4.00\times10^{-13}$	<b>87</b> $4p_{-}4p^{4}4d_{-}[1]4d$	$34 + 12({}_{2}^{3}P)^{2}D + 10({}_{2}^{3}F)^{4}D + 10({}_{2}^{3}F)^{4}F$

TABLE III. (Continued).

ground state is only 98%  $4p^{6}4d$  in the present work whereas it accounts for 100% of the wave-function compositions in ASD. In the MCDHF expansion most of the remaining composition is accounted for by the configurations  $4p^{5}4d4f$  and  $4p^{4}4d^{3}$ , configurations not included in the calculations for the data given in ASD [6,19].

Table VI compares the present *jj* and *LSJ* compositions with ASD compositions, obtained from a least-squares fitting of observed wavelengths using the Cowan code [6], for three levels. In the first example, a major contribution is from  $4p^{6}4f$ and its composition is essentially the same in all cases. In the second, there appears to be agreement, but for the third, there is a difference in the coupling of the dominant component. One of the features of  $4p^54d^2$  CSFs and their interaction matrix is that there are many different couplings for the same final J value. In fact, Table I indicates that there are 11 different couplings for J = 3/2. It is not clear that the two very different methods would yield the same composition. Whereas the present method is an *ab initio* method, the other is semiempirical. Table III includes an energy level that could be a contender for the  $4p^54d^2({}_2^1D)(1/2,2)$  label used in ASD, namely  $4p^54d^2({}_2^1D) {}^2P_{3/2}$  with a computed energy of 2 304 542  $\text{cm}^{-1}$ . However, the line with the largest measured intensity [2] was observed at 45.7810(22) Å and was classified as a 3/2-3/2 transition from the ground state to  $4p_4p_4^4d_2^2$ . The wavelength for a transition from the ground state to the observed level at 2 184 310 cm<sup>-1</sup> would be 45.7810 Å. According to Table IV for E1 transition probabilities, the upper level with the largest transition rate from the ground state is  $4p^54d^2({}_2^3F) {}^2D_{3/2}$  with a computed wavelength of 45.495 Å, in excellent agreement with observation. Thus we believe there is a difference in the label with ASD as well as a difference in composition. Tables II and III are based on this assumption.

## IV. COMPARISON WITH OTHER THEORY AND EXPERIMENT

For transition probability calculations the important accuracy indicators [21] are the accuracy of the transition energies and the agreement in the length and velocity forms of the line strength for electric multipole transitions.

Figure 1 compares quasirelativistic (QR) [9], parametric potential (PP) [3], and observed [6] energy levels relative to present values as a function of the computed energies. For plotting purposes, the position of the  $4d \ ^2D_{5/2}$  level has been moved to an energy within the graph. The observed fine-structure splitting is somewhat larger than our computed value but the latter is in considerably better agreement with

observed than the other theories. With the exception of the highest observed level,  $4p^54d^2({}_2{}^{1}G) {}^{2}F_{7/2}$ , all other levels from all theories are higher than observed, with present values being higher by less than 0.5%. In nearly all cases the present energies are in better agreement with ASD than QR results which, in turn, are considerably better than the earlier PP values. The latter did not include a sufficient number of configurations from the n = 4 complex, particularly the configurations that result from  $4p^2 \rightarrow 4d^2$  excitation from CSFs in the multireference set. For higher levels, the PP values are too high by about 20 000 cm<sup>-1</sup>. From the present work, the least accurate energy is for the  $4p^54d^2({}_2{}^3F) {}^2D_{3/2}$  level for which there is a disagreement in the label with ASD. The present label agrees with the QR label [9].

Table VII compares PP [3] and present wavelengths for some observed wavelengths. Present results are a considerable improvement and are in error by  $\leq 0.7\%$ .

Another factor in the accuracy of a transition probability,  $A_{ki}$ , is the accuracy of the line strength. In Tables IV and V,  $\delta T$  represents the discrepancy in the length and velocity forms of the transition probability when observed transition energies or wavelengths are used. When transition energies are sufficiently accurate, this is equivalent to the discrepancy in the line strength, namely  $|S_l - S_v| / \max(S_l, S_v)$ . This accuracy indicator varies considerably from one transition to the next. In these tables the *LSJ* labels are used so that it is possible to identify transitions that are essentially "*LS* forbidden" with



FIG. 1. (Color online) Comparison of quasirelativistic (QR) [9], parametric potential (PP) [3] energy levels with values derived from observation [6,19] as a function of the energy. The energy position of  $4p^{6}4d^{2}D_{5/2}$  has been moved to a position on the graph.

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TABLE IV. MCDHF transition data for E1 transitions from $4p^{6}4d^{2}D_{3/2}$ : levels and 2J for lower level i, upper level k,	wavelength $\lambda$
(in Å), line strength S (length form), oscillator strength f (length form), transition rate $A_{ki}$ (length form) in s <sup>-1</sup> .	c

Lev	vels	,	2J					
i	k	i	k	λ (Å)	S	f	$A_{ki}$ (s <sup>-1</sup> )	$\delta T$
$\frac{1}{4n^{6}4d}$	$-4n^54d^2(s^3F)$							
$^{2}D$	${}^{4}D^{o}$	3	3	81.868	$1.297 \times 10^{-3}$	$4.811 \times 10^{-3}$	$1.197 \times 10^{9}$	0.092
_	_	3	1	81.841	$2.130 \times 10^{-3}$	$7.907 \times 10^{-3}$	$3.937 \times 10^{9}$	0.095
$^{2}D$	${}^{4}F^{o}$	3	5	80.060	$3.532 \times 10^{-3}$	$1.340 \times 10^{-2}$	$2.324 \times 10^{9}$	0.052
$4p^{6}4d$ -	$-4p^{5}4d^{2}(2^{3}P)$							
$^{2}D$	<sup>4</sup> <i>P</i> <sup>o</sup>	3	3	75.841	$1.149 \times 10^{-5}$	$4.600 \times 10^{-5}$	$1.334 \times 10^{7}$	0.508
$4p^{6}4d -$	$-4p^{5}4d^{2}({}_{2}{}^{3}F)$				5			
<sup>2</sup> D	<sup>4</sup> <i>D</i> <sup>o</sup>	3	5	73.804	$7.725 \times 10^{-5}$	$3.179 \times 10^{-4}$	$6.489 \times 10^{7}$	0.135
$4p^{6}4d -$	$-4p^{5}4d^{2}({}_{2}^{1}D)$	-						
<sup>2</sup> D	<sup>2</sup> D <sup>0</sup>	3	3	71.629	$4.476 \times 10^{-3}$	$1.898 \times 10^{-2}$	$6.169 \times 10^{9}$	0.028
$4p^{6}4d -$	$-4p^{5}4d^{2}({}_{2}{}^{3}P)$		_		<b>2</b> 004 40 5		<b>a</b> (a) ( ) (a)	
$^{2}D$	${}^{4}P^{o}$	3	5	71.358	$2.804 \times 10^{-5}$	$1.194 \times 10^{-4}$	$2.606 \times 10^{7}$	0.218
$^{2}D$	$^{2}S^{o}$	3	1	69.847	$2.239 \times 10^{-8}$	$9.736 \times 10^{-8}$	$6.656 \times 10^{4}$	0.950
$^{2}D$	${}^{4}D^{o}$	3	5	68.576	$6.544 \times 10^{-3}$	$2.898 \times 10^{-2}$	$6.852 \times 10^{9}$	0.001
$^{2}D$	${}^4S^o$	3	3	68.214	$5.901 \times 10^{-5}$	$2.628 \times 10^{-4}$	$9.417 \times 10^{7}$	0.179
$4p^{6}4d -$	$-4p^54d^2({}_2{}^1G)$							
$^{2}D$	$^{2}F^{o}$	3	5	66.169	$1.428 \times 10^{-1}$	$6.553 \times 10^{-1}$	$1.664 \times 10^{11}$	0.004
$4p^{6}4d -$	$-4p^54d^2({}_2{}^3F)$							
$^{2}D$	${}^{4}F^{o}$	3	3	64.592	$2.455\times10^{-1}$	1.155	$4.615\times10^{11}$	0.011
$4p^{6}4d -$	$-4p^54d^2({}_2{}^1D)$							
$^{2}D$	$^{2}D^{o}$	3	5	64.364	$9.554 \times 10^{-5}$	$4.509 \times 10^{-4}$	$1.210 \times 10^{8}$	0.108
$^{2}D$	$^{2}P^{o}$	3	1	62.991	$4.688\times10^{-2}$	$2.260\times10^{-1}$	$1.900\times10^{11}$	0.027
$4 p^{6} 4 d -$	$-4p^54d^2(2^3P)$							
$^{2}D$	$^{2}P^{o}$	3	1	61.663	$1.449 \times 10^{-1}$	$7.139 \times 10^{-1}$	$6.262 \times 10^{11}$	0.018
$4 n^6 4 d =$	$4n^54d^2(a^1S)$							
$^{2}D$	$2\mathbf{p}^{o}$	3	3	59 532	$1.732 \times 10^{-4}$	$8.839 \times 10^{-4}$	$4.159 \times 10^{8}$	0.032
1 61 1	1 4 54 12 ( 3T)	5	5	57.552	1.752 × 10	0.057 × 10	4.157 × 10	0.052
$4p^{\circ}4a - 2p$	$-4p^{3}4d^{2}(2F)$	2	5	57 167	$2.484 \times 10^{-1}$	1 212	4 420 × 1011	0.012
-D	$F^{\circ}$	3	5	57.407	$2.484 \times 10^{-1}$	1.515	$4.420 \times 10^{11}$	0.012
$4p^{6}4d -$	$-4p^{5}4d^{2}({}_{2}{}^{3}P)$				-		0	
$^{2}D$	$^{2}D^{o}$	3	3	57.401	$7.383 \times 10^{-5}$	$3.907 \times 10^{-4}$	$1.977 \times 10^{8}$	0.267
$4p^{6}4d -$	$-4p^{6}4f$							
$^{2}D$	$^{2}F^{o}$	3	5	56.674	$3.190 \times 10^{-1}$	1.710	$5.918  imes 10^{11}$	0.013
$4p^{6}4d -$	$-4p^54d^2({}_2{}^3F)$							
$^{2}D$	${}^{4}G^{o}$	3	5	49.523	$4.039 \times 10^{-1}$	2.478	$1.123\times10^{12}$	0.007
$4 p^{6} 4 d -$	$-4p^54d^2({}_2{}^3P)$							
$^{2}D$	$^{2}D^{o}$	3	5	47.042	$2.032 \times 10^{-3}$	$1.312 \times 10^{-2}$	$6.592 \times 10^{9}$	0.005
$^{2}D$	${}^{4}D^{o}$	3	3	46.839	$1.691 \times 10^{-2}$	$1.096 \times 10^{-1}$	$8.334 \times 10^{10}$	0.003
$^{2}D$	${}^{4}P^{o}$	3	1	46 680	$3494 \times 10^{-3}$	$2.274 \times 10^{-2}$	$3480 \times 10^{10}$	0.010
$^{2}D$	${}^{4}D^{o}$	3	1	46.626	$1.731 \times 10^{-1}$	1.128	$1.730 \times 10^{12}$	0.010
$4 n^{6} 4 d -$	$-4n^54d^2(a^3F)$							
$^{2}D$	$^{2}D^{o}$	3	3	45 495	$5.950 \times 10^{-1}$	3 973	$3.200 \times 10^{12}$	0.003
D	D	3	5	43 432	$1.379 \times 10^{-4}$	$9.643 \times 10^{-4}$	$5.200 \times 10^{8}$ 5.683 × 10 <sup>8</sup>	0.005
4 641	4 54 2 (10)	5	5	13.132	1.577 × 10	9.015 X 10	5.005 X 10	0.020
$4p^{\circ}4d - 2p^{\circ}$	$-4p^{3}4d^{2}(2^{4}D)$	2	2	12 202	$2 460 - 10^{-3}$	1 700 10-?	1.505 1.010	0.007
<i>-</i> Д	<sup>2</sup> P <sup>3</sup>	3	3	43.393	$2.460 \times 10^{-5}$	$1.722 \times 10^{-2}$	$1.525 \times 10^{10}$	0.007
$4p^{6}4d -$	$-4p^{5}4d^{2}({}_{2}{}^{3}P)$	-			4			
²D	<sup>2</sup> P <sup>0</sup>	3	3	43.070	$5.718 \times 10^{-4}$	$4.032 \times 10^{-3}$	$3.625 \times 10^{9}$	0.056
$4p^{6}4d -$	$-4p^54d^2({}_2{}^1D)$							
$^{2}D$	$^{2}F^{o}$	3	5	42.997	$7.497 \times 10^{-6}$	$5.297 \times 10^{-5}$	$3.185 \times 10^{7}$	0.002
$4p^{6}4d -$	$-4p^54d^2(_0{}^1S)$							
$^{2}D$	$^{2}P^{o}$	3	1	41.750	$2.193 \times 10^{-3}$	$1.596\times10^{-2}$	$3.053 \times 10^{10}$	0.009

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TABLE V. MCDHF transition data for E1 transitions from $4p^{6}4d^{2}D_{5/2}$ : levels and 2J for lower level <i>i</i> , upper level <i>k</i> , wavelength $\lambda$ (in	Å),
line strength S (length form), oscillator strength f (length form), transition rate $A_{ki}$ (length form) in s <sup>-1</sup> .	

Le	vels	-	2J					
i	k	i	k	λ (Å)	S	f	$A_{ki}$ (s <sup>-1</sup> )	$\delta T$
$4 p^{6} 4 d$ -	$-4p^54d^2(r^3F)$							
$^{2}D$	${}^{4}D^{o}$	5	3	93.710	$6.217 \times 10^{-5}$	$2.015 \times 10^{-4}$	$3.827 \times 10^{7}$	0.333
$^{2}D$	${}^{4}F^{o}$	5	5	91.348	$5.235 \times 10^{-4}$	$1.741 \times 10^{-3}$	$2.319 \times 10^{8}$	0.099
$^{2}D$	$^{2}G^{o}$	5	7	90.088	$1.484  imes 10^{-4}$	$5.003 \times 10^{-4}$	$5.139 \times 10^{7}$	0.020
$4p^{6}4d$ -	$-4p^{5}4d^{2}({}_{2}{}^{3}P)$							
$^{2}D$	$^{4}P^{o}$	5	3	85.897	$2.013\times10^{-4}$	$7.120\times10^{-4}$	$1.609 \times 10^{8}$	0.113
$4p^{6}4d$ -	$-4p^54d^2({}_2{}^3F)$							
$^{2}D$	${}^{4}D^{o}$	5	5	83.292	$7.734 \times 10^{-7}$	$2.820 \times 10^{-6}$	$4.520 \times 10^{5}$	0.179
$^{2}D$	${}^{4}F^{o}$	5	7	81.424	$7.276 \times 10^{-3}$	$2.714\times10^{-2}$	$3.414 \times 10^{9}$	0.040
$4p^{6}4d$ -	$-4p^54d^2(^1_2D)$							
$^{2}D$	$^{2}D^{o}$	5	3	80.533	$1.972 \times 10^{-3}$	$7.437 \times 10^{-3}$	$1.912 \times 10^{9}$	0.026
$4p^{6}4d$ -	$-4p^54d^2({}_2^3P)$							
$^{2}D$	${}^{4}P^{o}$	5	5	80.190	$3.502 \times 10^{-3}$	$1.327 \times 10^{-2}$	$2.294 \times 10^{9}$	0.048
$4p^{6}4d$ -	$-4p^54d^2({}_2{}^1G)$							
$^{2}D$	$^{2}G^{o}$	5	7	79.053	$4.708 \times 10^{-7}$	$1.809 \times 10^{-6}$	$2.414 \times 10^{5}$	0.585
$4p^{6}4d$ -	$-4p^54d^2({}_2{}^3P)$							
$^{2}D$	${}^{4}D^{o}$	5	7	76.980	$3.535 \times 10^{-5}$	$1.395  imes 10^{-4}$	$1.963 \times 10^{7}$	0.077
		5	5	76.694	$1.825 \times 10^{-6}$	$7.229 \times 10^{-6}$	$1.366 \times 10^{6}$	0.994
$^{2}D$	${}^{4}S^{o}$	5	3	76.242	$1.817  imes 10^{-4}$	$7.239  imes 10^{-4}$	$2.077 \times 10^{8}$	0.025
$4p^{6}4d$ -	$-4p^54d^2({}_2{}^1G)$							
$^{2}D$	$^{2}F^{o}$	5	5	73.695	$1.925  imes 10^{-4}$	$7.936  imes 10^{-4}$	$1.624 \times 10^{8}$	0.257
$4p^{6}4d$ -	$-4p^54d^2({}_2{}^3F)$							
$^{2}D$	${}^{4}F^{o}$	5	3	71.745	$1.186 \times 10^{-3}$	$5.020 \times 10^{-3}$	$1.626 \times 10^{9}$	0.023
$4p^{6}4d$ -	$-4p^54d^2(^1_2D)$							
$^{2}D$	$^{2}D^{o}$	5	5	71.463	$5.107  imes 10^{-4}$	$2.171 \times 10^{-3}$	$4.726 \times 10^{8}$	0.042
$^{2}D$	$^{2}F^{o}$	5	7	71.211	$2.814 \times 10^{-2}$	$1.200 \times 10^{-1}$	$1.974 \times 10^{10}$	0.008
$4p^{6}4d$	$-4p^54d^2({}_2{}^3F)$							
$^{2}D$	$^{2}F^{o}$	5	7	67.075	$2.820 \times 10^{-2}$	$1.277 \times 10^{-1}$	$2.366 \times 10^{10}$	0.013
$4p^{6}4d$	$-4p^54d^2({}_0{}^1S)$							
$^{2}D$	$^{2}P^{o}$	5	3	65.556	$7.008 \times 10^{-4}$	$3.247 \times 10^{-3}$	$1.260 \times 10^{9}$	0.010
$4p^{6}4d$	$-4p^54d^2({}_2{}^3F)$							
$^{2}D$	$^{2}F^{o}$	5	5	63.060	$3.672 \times 10^{-1}$	1.769	$4.945 \times 10^{11}$	0.011
$4p^{6}4d$	$-4p^54d^2({}_2^3P)$							
$^{2}D$	$^{2}D^{o}$	5	3	62.981	$3.124 \times 10^{-1}$	1.507	$6.334 \times 10^{11}$	0.024
$4p^{6}4d$	$-4p^{6}4f$							
$^{2}D$	$^{2}F^{o}$	5	5	62.107	$1.688  imes 10^{-1}$	$8.255 \times 10^{-1}$	$2.379 \times 10^{11}$	0.012
		5	7	61.628	$5.272 \times 10^{-1}$	2.598	$5.704 \times 10^{11}$	0.016
$4p^{6}4d$	$-4p^54d^2({}_2{}^3F)$							
$^{2}D$	${}^4\!G^o$	5	5	53.622	$3.718 \times 10^{-4}$	$2.106 \times 10^{-3}$	$8.144 \times 10^{8}$	0.018
	5	5	7	50.922	$6.850 \times 10^{-4}$	$4.086 \times 10^{-3}$	$1.314 \times 10^{9}$	0.023
$4p^{\circ}4d$	$-4p^{3}4d^{2}({}_{2}{}^{3}P)$	_	_	<b>T</b> O <b>TO T</b>			<b>7</b>	0.014
$^{2}D$	$^{2}D^{o}$	5	5	50.725	$1.989 \times 10^{-2}$	$1.191 \times 10^{-1}$	$5.146 \times 10^{10}$	0.011
<sup>2</sup> D	$^{4}D^{o}$	5	3	50.489	$3.644 \times 10^{-5}$	$2.193 \times 10^{-2}$	$1.434 \times 10^{10}$	0.031
$4p^{\circ}4d$	$-4p^{3}4d^{2}({}_{2}F)$	E	2	40.021	$2 (00 10^{-3})$	1 (10 10-?	1 1 20 1010	0.049
-D 2D	$^{2}D^{a}$	5	3	48.931	$2.608 \times 10^{-3}$ 1.247 × 10 <sup>-1</sup>	$1.619 \times 10^{-2}$	$1.128 \times 10^{10}$	0.048
$\frac{D}{2}$	$^{2}D^{o}$	5	5	47.510	$1.347 \times 10^{-1}$ 7.460 × 10 <sup>-1</sup>	8.049 × 10 -	$3.221 \times 10^{12}$ 2 500 × 10 <sup>12</sup>	0.003
$\int D dn^6 dd$	$-4n^{5}4d^{2}(a^{1}D)$	5	5	40.555	7.409 × 10	4.075	2.300 × 10	0.004
$\frac{1}{2}$ D	- +p +u (2D) $2p^{o}$	5	3	46 508	$3.165 \times 10^{-2}$	$2.067 \times 10^{-1}$	$1.594 \times 10^{11}$	0.024
$4n^{6}4d$	$-4n^{5}4d^{2}(a^{1}G)$	5	5	+0.500	5.105 × 10	2.007 × 10	1.574 × 10	0.024
$^{2}D$	${}^{2}F^{o}$	5	7	46,224	$7.413 \times 10^{-1}$	4,871	$1.901 \times 10^{12}$	0.001
$\frac{1}{4}p^{6}4d$	$-4p^{5}4d^{2}(r^{3}P)$	2						0.001
$^{2}D$	$^{2}P^{o}$	5	3	46.137	$4.110 \times 10^{-1}$	2.706	$2.120 \times 10^{12}$	0.011
$4p^{6}4d$	$-4p^54d^2({}_2^1D)$							
$^{2}D$	$^{2}F^{o}$	5	5	46.054	$1.035 \times 10^{-3}$	$6.830 \times 10^{-3}$	$3.580 \times 10^{9}$	0.024

TABLE VI. Comparison of present wave-function composition with ASD [19].

	Composition
(i) Energy level: 1508470	(J = 5/2)
jj	$39 4p_{-}^{2}4p^{3}4d_{-}[2]4d + 27 4p_{-}^{2}4p^{4}4f_{-}$
	$+13 4p_{-}^{2}4p^{3}4d_{-}[1]4d$
LSJ	$41 ({}_{2}^{1}G) {}^{2}F + 27 4 p^{6}4 f {}^{2}F + 13 ({}_{2}^{3}F) {}^{2}F$
ASD	$45 ({}_{2}^{1}G) (3/2,4) + 28 4p^{6}4f (0,5/2)$
(ii) Energy level: 1542600	(J = 3/2)
jj	$59 4p_{-}^{2}4p^{3}4d_{-}[3]4d$
LSJ	$32 ({}_{2}^{3}F) {}^{4}F + 22 ({}_{2}^{3}P) {}^{2}D + 12 ({}_{2}^{3}F) {}^{2}D$
ASD	$33 \binom{3}{2}F (3/2,3) + 18 \binom{3}{2}P (3/2,2)$
(iii) Energy level: 218431	0 (J = 3/2)
jj	$87 4p_4p_4^4 d_2^2(2)$
LSJ	$44 \left( {}_{2}^{3}F \right) {}^{2}D + 17 \left( {}_{2}^{1}D \right) {}^{2}D + 12 \left( {}_{2}^{3}F \right) {}^{4}F$
ASD	43 $\binom{1}{2}$ (1/2,2) + 28 $\binom{3}{2}$ P) (1/2,2)

Lifetimes (s) Theory [9] Label JPos This work  $4p^5 4d^2({}_2^3F) {}^4D^o$  $8.10\times10^{-10}$  $8.44 \times 10^{-10}$ 3/21  $2.54\times10^{-10}$  $2.44 \times 10^{-10}$ 1/21  $4\mathbf{F}^{a}$  $2.16\times10^{-12}$  $2.16\times10^{-12}$ 3/25  $4p^5 4d^2({}_2^1G) {}^2F^o$ 5  $6.00\times10^{-12}$  $6.50\times10^{-12}$ 5/2 $5.26\times10^{-13}$  $5.17\times10^{-13}$ 7/210  $4p^{6} 4f^{2}F^{o}$  $1.21\times10^{-12}$ 5/28  $1.21 \times 10^{-12}$ 7  $1.72 \times 10^{-12}$ 7/2 $1.75 \times 10^{-12}$  $4p^5 4d^2({}_{2}^{3}F) {}^{4}G^{o}$  $8.90\times10^{-13}$ 5/29  $8.80 \times 10^{-13}$  $^{2}D^{o}$ 9  $3.11 \times 10^{-13}$  $3.03 \times 10^{-13}$ 3/25/211  $4.00 \times 10^{-13}$  $1.02 \times 10^{-12}$ 

TABLE VIII. Computed lifetimes for W XXXVIII.

### **V. CONCLUSIONS**

 $\Delta S \neq 0$  or  $\Delta L \neq 1$ . Large values of  $\delta T$  are associated with the latter. A relative error estimate for an electric multipole (Ek) transition, is  $\delta T + (2k + 1)\delta E$  where  $\delta E$  is the relative error in the transition energy and k is the order of the multipole transition. Many E1 transition probabilites have  $\delta T \leq 0.010-0.03$  or 1-3%. For M1 transitions, the length and velocity values are the same by definition and are not a test of the wave function. For E2 transitions, the Supplemental Material [17] shows that for many LS allowed E2 transitions,  $\delta T \leq 0.05-0.06$  or 5-6%.

Table VIII compares the lifetimes of some levels with QR values computed from *E*1 transition rates. Generally there is excellent agreement in the two sets of results. An exception is the lifetime of the  $4p^54d^2({}_2{}^3F) {}^2D^o$  level. None of these lifetimes of higher levels, are affected by *E*2, *M*1 transitions to lower odd-parity levels to the accuracy reported.

With the revised GRASP2K code, it has been possible to compute the energies of  $4p^{6}4d$ ,  $4p^{6}4f$ , and  $4p^{5}4d^{2}$  levels and all *E*1, *E*2, and *M*1 transitions between them to high accuracy using the SD-MR method. Because the 4*d* and 4*f* subshells of the n = 4 shell are unfilled, the interaction with other configurations from the n = 4 complex are of prime importance. The major components of the wave function have been compared in both *jj* and *LSJ* coupling. Although in *jj* coupling most of the expansions have a dominant component, the *LSJ* label can be used to identify the *LS*-forbidden transitions. The property of being *LS* forbidden explains the larger error estimates for such a transition. Finally, we estimated the lifetime of  $4p^{6}4d^{2}D_{5/2}$  to be 25.2(2) ms.

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TABLE VII. Comparison of computed wavelengths from different theories with observed wavelengths for transitions from  $4p^{6}4d^{2}D$  to selected upper levels.

Upper	$2J_i$	$2J_k$	QR [9]	PP [3]	Expt. [2]	This work
$\overline{4p^64d^2D}$	3	5			646.7(5) <sup>a</sup>	647.86
$4p^5 4d^2({}_2^3F) {}^2D^o$	3	3	45.312	44.766	45.7810(22)	45.495
$4p^5 4d^2({}_2^3P) {}^2P^o$	5	3		45.480	46.0640(55)	46.137
$4p^5 4d^2({}_2^1G) {}^2F^o$	5	7	46.007	45.556	46.0640(55)	46.224
$4p^5 4d^2({}_2^3F) {}^4G^o$	3	5	49.215	49.030	49.6407(54)	49.523
$4p^{6} 4f^{2}F^{o}$	3	5	56.332	56.055	56.8797(41)	56.674
$4p^5 4d^2({}_2^3F) {}^2F^o$	3	5	57.029	56.680	57.7547(14)	57.467
$4p^{6} 4f^{2}F^{o}$	5	7	61.158	60.922	61.9200(83)	61.628
$4p^6 4f {}^2F^o$	5	5		61.340	61.9200(83)	62.107
$4p^5 4d^2({}_2^3F) {}^2F^o$	5	5		62.090	63.4319(119)	63.060
$4p^5 4d^2({}_2^3P) {}^2D^o$	5	3		62.138	63.4319(119)	62.981
$4p^5 4d^2({}_2^3F) {}^4F^o$	3	3	64.029	63.775	64.8250(200)	64.592
$4p^5 4d^2({}_2^1G) {}^2F^o$	3	5	65.703	65.703	66.29 252(147)	66.169
$4p^5 4d^2({}_2^3F) {}^4D^o$	3	1	81.222	81.299	81.4573(89)	81.841
$4p^5 4d^2({}_2^3F) {}^4D^o$	3	3	81.228	81.359	81.4573(89)	81.868

<sup>a</sup>From Radtke et al. [7].

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