Ground states of H, He, . . . , Ne, and their singly positive ions in strong magnetic fields: The high-field regime

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The electronic structure of the ground state and some excited states of neutral atoms with the nuclear charge numbers $1 \leq Z \leq 10$ and their singly positive ions are investigated by means of our two-dimensional mesh Hartree-Fock method for strong magnetic fields $0.5 \le \gamma \le 10\,000$. For $\gamma = 10\,000$ the ground-state configurations of all the atoms and ions considered are given by fully spin-polarized configurations of single-electron orbitals with magnetic quantum numbers ranging from $m=0$ to $m=-N+1$, where N is the number of the electrons. Focusing on the fully spin-polarized situation, we provide critical values of the magnetic-field strength for which crossovers with respect to the spatial symmetries of the ground state take place. It is found that the neutral atoms and singly charged positive ions with $2 \leq Z \leq 5$ have one fully spin-polarized groundstate configuration, whereas for 6<*Z*<10 one intermediate fully spin-polarized configuration with an orbital of $2p_0$ type occurs.

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I. INTRODUCTION

The behavior and properties of atoms in strong magnetic fields is a subject which has attracted the interest of many researchers. Partially this interest is motivated by the astrophysical discovery of strong fields on white dwarfs and neutron stars $[1-3]$. On the other hand, the competition of the diamagnetic and Coulomb interaction, characteristic of atoms in strong magnetic fields, causes a rich variety of complex properties which are of interest on their own.

Investigations on electronic structure in the presence of a magnetic field appear to be quite complicated due to the intricate geometry of these quantum problems. Most of the investigations in the literature focus on the hydrogen atom (for a list of references see, for example, Refs. $[4–7]$). These studies provided us with a detailed understanding of the electronic structure of the hydrogen atom in magnetic fields of arbitrary strengths. As a result the absorption features of certain magnetic white dwarfs could be explained, and this allowed for a modeling of their atmospheres (see Ref. $[8]$ for a comprehensive review of atoms in strong magnetic fields and their astrophysical applications up to 1994, and Ref. $[9]$ for a more recent review on atoms and molecules in external fields). On the other hand, there are a number of magnetic white dwarfs whose spectra remain unexplained and cannot be interpreted in terms of magnetized atomic hydrogen. Furthermore new magnetic objects have been discovered (see, for example, Reimers *et al.* [10] in the course of the Hamburg European Southern Observatory survey) whose spectra await to be explained. Very recently significant progress has been achieved with respect to the interpretation of the observed spectrum of the prominent white dwarf GD229 which shows a rich spectrum ranging from the UV to the near IR. Extensive and precise calculations on the helium atom provided data for many excited states in a broad range of field strengths $[11]$. A comparison of the stationary transitions of the atom with the positions of the absorption edges of the observed spectrum yielded strong evidence of the existence of helium in the atmosphere of GD229 $[12]$.

For atoms with several electrons there are two decisive factors which enrich the possible changes in the electronic structure with varying field strength compared to the oneelectron system. First we have a third competing interaction, the electron-electron repulsion, and second the different electrons ''feel'' very different Coulomb forces, i.e. possess different one particle energies; consequently the regime of intermediate-field strengths appears to be the sum of the intermediate regimes for the separate electrons.

There exist a number of investigations on two-electron atoms in the literature (see Ref. $[11]$ and references therein). Focusing on systems with more than two electrons, however, the number of investigations is very small $[13–19]$. Some of them use the adiabatic approximation in order to investigate the very high field regime. These works contain a number of important results on the properties and structure of several multielectron atoms. Though very useful for high fields, the adiabatic approach hardly allows one to describe the electronic structure with decreasing field strength: particularly the core electrons of multielectron atoms ''feel'' a strong nuclear attraction which can be dominated by the external field only for very high-field strengths. In view of this there is a need for further quantum-mechanical investigations on multielectron atoms, particularly in the intermediate- to highfield regime.

The ground states of atoms in strong magnetic fields have different spatial and spin symmetries in different regions of the field strengths. We encounter, therefore, a series of changes i.e. crossovers with respect to their symmetries with varying field stength. The simplest case is a helium atom, which possesses two ground-state configurations: the singlet

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zero- and low-field ground state $1s²$, and the fully spinpolarized high-field ground state $1s2p_{-1}$. In the Hartree-Fock approximation the transition point between these configurations is given by the field strength $\gamma=0.711$. (If not indicated otherwise, in the following we use atomic units for all quantities. In particular, the magnetic field $\gamma = B/B_0$ is measured in units $B_0 = \hbar c / e a_0^2 = 2.3505 \times 10^5$ T = 2.3505 $\times 10^{9}$ G.) In previous works we investigated the series of transitions of the ground-state configurations for the complete range of field strengths for the lithium $\lceil 18 \rceil$ and carbon [19] atoms as well as the ion $Li⁺$ [18]. The evolution and appearence of these crossovers and the involved configurations become more and more intricate with an increasing number of electrons of the atom. Currently the most complicated atomic system with a completely known sequence of ground-state electronic configurations for the whole range of magnetic field strengths is the neutral carbon atom $[19]$. Its ground state experiences six crossovers involving seven different electronic configurations which belong to three groups of different spin projections $S_z = -1$, -2 , and -3 onto the magnetic field. This series of ground state configurations was extracted from results of numerical calculations for more than 20 electronic configurations selected via a detailed analysis on the basis of general energetical arguments. The picture of these transitions is especially complicated at relatively weak and intermediate fields. Due to this circumstance a comprehensive investigation of the structure of the ground states of atoms is a complex problem which has to be solved for each atom separately. On the other hand, the geometry of the atomic wave functions is simplified for sufficiently high magnetic fields: Beyond some critical field strength the global ground state is given by a fully spin-polarized configuration. This allows us to push the current state of the art and to study the ground states of the full series of neutral atoms and singly charged positive ions with $Z \le 10$, i.e., the sequence H, He, Li, Be, B, C, N, O, F, and Ne in the domain of high magnetic fields. For the purpose of this investigation we define the high-field domain as the one where the groundstate electronic configurations are fully spin polarized [the fully spin polarized (FSP) regime. The latter fact supplies an additional advantage for calculations performed in the Hartree-Fock approach, because our one-determinantal wave functions are eigenfunctions of the total spin operator S^2 . Starting from the high-field limit we will investigate the electronic structure and properties of the ground states with decreasing field strength until, we reach the first crossover to a partially spin-polarized (PSP) configuration; i.e., we focus on the regime of field strengths for which fully spin polarized configurations represent the ground state.

II. METHOD

The numerical approach applied in the present work coincides with that of our previous investigations $[17–19]$. References $[7,17,18,20]$ contained more details of the mesh techniques. We solve the electronic Schrödinger equation for atoms in a magnetic field under the assumption of an infinitely heavy nucleus (see below for comments on finite nuclear mass corrections) in the (unrestricted) Hartree-Fock approximation. The solution is established in cylindrical coordinates (ρ, ϕ, z) , with the *z* axis oriented along the magnetic field. We prescribe to each electron a definite value of the magnetic quantum number m_{μ} . Each one-electron wave function Ψ_{μ} depends on the variables ϕ and (ρ, z) as follows:

$$
\Psi_{\mu}(\rho,\phi,z) = (2\pi)^{-1/2} e^{-im_{\mu}\phi} \psi_{\mu}(z,\rho), \qquad (1)
$$

where μ indicates the numbering of the electrons. The resulting partial differential equations for $\psi_{\mu}(z,\rho)$ and the formulas for the Coulomb and exchange potentials were presented in Ref. $[20]$.

The one-particle equations for the wave functions $\psi_{\mu}(z,\rho)$ are solved by means of the fully numerical mesh method described in Refs. $[7,20]$. The feature which distinguishes the present calculations from those described in Ref. [20] is the method for the calculation of the Coulomb and exchange integrals. In the present work as well as in Refs. $|17-19|$, we obtain these potentials as solutions of the corresponding Poisson equation.

Our mesh approach is flexible enough to yield precise results for arbitrary field strengths. Some minor decrease of the precision appears in very strong magnetic fields. With respect to the electronic configurations possessing high absolute values of magnetic quantum numbers of outer electrons, some minor computational problems arose also at lower field strengths. Both these phenomena are due to a large difference with respect to the binding energies $\epsilon_{B\mu}$ of one-electron wave functions belonging to the same electronic configuration

$$
\epsilon_{B\mu} = (m_{\mu} + |m_{\mu}| + 2s_{z\mu} + 1)\gamma/2 - \epsilon_{\mu},\qquad(2)
$$

where ϵ_{μ} is the one-electron energy and $s_{z\mu}$ is the spin *z* projection. The precision of our results also depends, of course, on the number of mesh nodes, and can be improved upon in calculations with denser meshes. Most of the present calculations are carried out on sequences of meshes, with the maximal number of nodes being 65×65 .

III. RELEVANT PROPERTIES IN THE HIGH-FIELD REGIME

In this section we provide some qualitative considerations on the problem of the ground states of multielectron atoms in the high-field limit. These considerations present a starting point for the combined qualitative and numerical considerations given in Sec. IV. At very high-field strengths the nuclear attraction energies and Hartree-Fock potentials (which determine the motion along the z axis) are small compared to the interaction energies with the magnetic field (which determines the motion perpendicular to the magnetic field and is responsible for the Landau zonal structure of the spectrum). Thus in the limit ($\gamma \rightarrow \infty$), all the one-electron wave functions of the ground state belong to the lowest Landau zones, i.e., $m_{\mu} \le 0$ for all the electrons, and the system must be fully spin polarized, i.e. $s_{z\mu} = -\frac{1}{2}$. For the Coulomb central field the one-electron levels form quasi-onedimensional Coulomb series with the binding energy E_B

 $=1/2n_z^2$ for $n_z>0$, whereas $E_B(\gamma \to \infty) \to \infty$ for $n_z=0$, where n_z is the number of nodal surfaces of the wave function crossing the *z* axis. In the limit $\gamma \rightarrow \infty$ the ground-state wave function must be formed from tightly bound singleelectron functions with $n_z=0$. The binding energies of these functions decrease as $|m|$ increases and, thus, the electrons must occupy orbitals with increasing $|m|$ starting with *m* $=0.$

In the language of the Hartree-Fock approximation the ground-state wave function of an atom in the high-field limit is a fully spin-polarized set of single-electron orbitals with no nodal surfaces crossing the *z* axis, and with nonpositive magnetic quantum numbers decreasing from $m=0$ to $m=$ $-N+1$, where *N* is the number of electrons. For the carbon atom, mentioned above, this Hartree-Fock configuration is $1s2p_{-1}3d_{-2}4f_{-3}5g_{-4}6h_{-5}$, with $S_z = -3$. For the sake of brevity, in the following we shall refer to these groundstate configurations in the high-field limit, i.e. the configuration generated by the tightly bound hydrogenic orbitals $1s, 2p_{-1}, 3d_{-2}, 4f_{-3}, \ldots$, as $|0_N\rangle$. The $|0_N\rangle$ states possess the complete spin polarization $S_z = -N/2$. Decreasing the magnetic-field strength, we can encounter a series of crossovers of the ground-state configurations associated with transitions of one or several electrons from orbitals, with maximal values for $|m|$, to other orbitals with a different spatial geometry of the wave function but the same spin polarization. This means the first few crossovers can take place within the space of fully spin-polarized configurations. We shall refer to these configurations by noting only the difference with respect to the $|0_N\rangle$ state. This notation can, of course, also be extended to nonfully spin-polarized configurations. For instance, the state $1s^2 2p_{-1}3d_{-2}4f_{-3}5g_{-4}$ with $S_z = -2$ of the carbon atom, will be briefly refered to as $\langle 1 \rangle$, since the default is the occupation of the hydrogenic series $1s, 2p_{-1}, 3d_{-2}, \ldots$ and only deviations from it are recorded by our notation.

In the following considerations we shall often refer to subsets of electronic states which possess different spin polarizations. As indicated above, we will denote the set of electronic states with $S_z = -N/2$ as the FSP subset. Along with the global ground state it is expedient also to consider what we call local ground states, which are the energetically lowest states with some definite degree of the spin polarization. For the purpose of the present work we need to know the local ground state of the subset of electronic states with $S_z = -N/2+1$ (which is the only partially spin-polarized subset considered in this paper, and which is refered to as subset PSP) in the high-field regime. This knowledge is necessary for the evaluation of the point of the crossover between the FSP and PSP ground states, i.e., for the determination of the critical field strengths at which the global ground state changes its spin polarization from $S_z = -N/2$ to $S_z = -N/2 + 1$. For sufficiently high fields the $|1s^2\rangle$ state is the local ground state of the PSP subset of electronic states.

IV. GROUND-STATE ELECTRONIC CONFIGURATIONS IN THE HIGH-FIELD REGIME

Let us start with the high-field limit and the state $|0_N\rangle$, and subsequently consider possible ground-state crossovers which occur *with decreasing magnetic-field strength*. In the high-field regime we have, per definition, only crossovers due to changes of the spatial orbitals, and no spin-flip crossovers. According to the goals of the present work we investigate the possible global ground-state configurations belonging to the subset FSP and determine the transition points to the subset PSP. Since the detailed study of the latter subset of states for arbitrary field strengths goes beyond the scope of the present work, we consider first only the $|1s^2\rangle$ state of this subset, which is the local ground state of the subset PSP for sufficiently strong fields. Then we investigate FSP ground states with decreasing field strength until we reach the point of crossover with the energy of the configuration $|1s^2\rangle$. Subsequently we need to consider other electronic configurations of the PSP set in order to determine the complete picture of the energy levels as a function of the field strength near the spin-flip crossover and, possibly, to correct the position of this point (the latter is necessary if the state $|1s^2\rangle$ is not the lowest one of the subset PSP at the spin-flip point).

Let us consider the ground-state transitions within the subset FSP with decreasing field strength. The first of these transitions occurs when the binding energy associated with the outermost orbital $(m_N = -N+1)$ becomes less than the binding energy of one of the orbitals with n_z $>$ 0. Due to the circumstance that all the orbitals with n_z $>$ 0 are not occupied in the high-field ground-state configuration, it is reasonable to expect the transition of the outermost electron to one of the orbitals with $m=0$ and either $n_z=1$ (i.e., the 2 $p₀$ orbital) or $n_z = 2$ (i.e., the 2*s* orbital). The decision between these two possibilities cannot be made on the basis of qualitative arguments. For the hydrogen atom or hydrogenlike ions in a magnetic field, the $2p_0$ orbital is more strongly bound than the 2*s* orbital for any field strength. On the other hand, owing to the electronic screening of the nuclear charge in multielectron atoms in field-free space, the 2*s* orbital tends to be more tightly bound than the $2p_0$ orbital. Thus we have two competing mechanisms, and numerical calculations are required for the decision between the possible $|0_N\rangle$ - $|2p_0\rangle$ and $|0_N\rangle$ - $|2_s\rangle$ crossovers to a new local FSP ground state. Our results of the calculations for the $|25\rangle$ states of both neutral atoms and positive ions presented below in Sec. V show that the $|2s\rangle$ state becomes more tightly bound than the $|2p_0\rangle$ state only for rather weak-field strengths, where this state cannot pretend to be the ground state of the corresponding atom or ion due to the presence of more tightly bound nonfully spin-polarized states. As a result the first intermediate ground state of the subset FSP, i.e., the state beside the $|0_N\rangle$ state which might be involved in the first crossover of the ground state with decreasing field strength, is the $|2p_0\rangle$ state. Calculations for the subset PSP (see below) show, indeed, that this state is the global ground state in a certain regime of field strengths for the neutral atoms with $Z \ge 6$, i.e., C, N, O, F and Ne, as well as their positive ions C^+ , N^+ , O^+ , F^+ , and Ne⁺. For the atoms He, Li, Be, and B ($Z \le 5$) as well as for the ions Li^+ , Be⁺, and B⁺ the $|1s^2\rangle$ state becomes more tightly bound than $|0_N\rangle$ for fields stronger that those associated with the $|0_N\rangle$ - $|2p_0\rangle$, crossover and the $|2p_0\rangle$ state never becomes the global ground state of these atoms and ions.

FIG. 1. The total energies (in atomic units) of the relevant states of the neon atom under consideration for the determination of the ground-state electronic configurations for the high-field regime.

Thus both neutral atoms and positive ions A^+ with $Z \le 5$ have only one fully spin-polarized ground-state configuration $|0_N\rangle$, which represents the global ground state above some critical field strength.

The question about a possible second intermediate fully spin-polarized ground state occurring with a further decreasing field strength arises for neutral atoms and positive ions with $Z \ge 6$ which possess the intermediate fully spinpolarized ground state $|2p_0\rangle$. This state could be either a state containing an additional orbital with $n_z=1$, which would result in the $|2p_03d_{-1}\rangle$ configuration, or a state with an additional *s*-type orbital, i.e., $|2s2p_0\rangle$. The third possibility of the simultaneous transition of the electron with the magnetic quantum number $m_{N-1} = -N+2$ to the 3*d*₋₁ orbital and the electron in the $2p_0$ orbital to the 2*s* orbital, which gives the $|2s3d_{-1}\rangle$ configuration, can be excluded from the list of possible ground-state configurations without a numerical investigation. The reason for this is that the $3d_{-1}$ orbital for any field strength is more weakly bound than the $2p_0$ orbital, and thus the $|2s2p_0\rangle$ configuration possesses a lower energy than the $|2s3d_{-1}\rangle$ configuration for arbitrary magnetic field strengths. When comparing the configurations $|2s2p_0\rangle$ and $|2p_03d_{-1}\rangle$ we can make use of what we have learned (see above) from the competing $|2p_0\rangle$ and $|2s\rangle$ configurations for higher field strengths: The 2*s* orbital is energetically preferable at weak magnetic fields, whereas the $3d_{-1}$ orbital yields energetically lower configurations in the strong-field regime. Thus we perform calculations for the $|2p_03d_{-1}\rangle$ configuration for many field strengths, and then perform calculations at much fewer field strengths to check the energy of the $|2s2p_0\rangle$ configuration in order to obtain the correct lowest energy and state of the FSP set.

The behavior of the energy levels described in the previous paragraph is illustrated in Fig. 1. In this figure the energy curves for four possible fully spin-polarized electronic configurations and two energy curves for the PSP subset of the neon $(Z=10)$ atom are presented. This figure shows, in particular, the energy curve of the high-field ground state $|0_N\rangle$ which intersects with the curve $E_{|2p_0\rangle}(\gamma)$ at $\gamma=159.138$. The latter energy remains the lowest in the FSP subset until the intersection of this curve with $E_{|2p_03d_{-1}\rangle}(\gamma)$ at γ $=40.537$. This intersection occurs at higher field strength than the intersection of the curves $E_{|2p_0\rangle}(\gamma)$ and $E_{|1s^2\rangle}(\gamma)$, which is at γ =38.060. On the other hand, the control calculations for the $|2s2p_0\rangle$ state, not presented in Fig. 1, show that its total energy for γ =38.060 is larger than the energy $E_{|2p_03d_{-1}}$. According to the previous argumentation, this means that the $|2s2p_0\rangle$ state is not the global ground state of the Ne atom for any magnetic-field strengths. Furthermore the $|2p_03d_{-1}\rangle$ state is a candidate for becoming the global ground state of the neon atom in some bounded regime of the field strength. However, we have not yet performed (see below) a detailed investigation of the lowest-energy curves of the PSP subset which is essential to take a definite decision on the global ground-state configurations. For neutral atoms with $6 \le Z \le 9$ and positive ions A^+ with $6 \le Z \le 10$, the energies of the $|2p_03d_{-1}\rangle$ and $|2s2p_0\rangle$ states at the points of intersections of the curves $E_{|2p_0\rangle}(\gamma)$ and $E_{|1s^2\rangle}(\gamma)$ are higher than the energies of the $|2p_0\rangle$ and $|1s^2\rangle$ states. This leads to the conjecture that no neutral atoms with $Z \leq 10$ and positive ions with $Z \le 10$ can possess more than two different fully spin-polarized ground-state configurations in the complete range of field strengths.

The above concludes our considerations of the fully spinpolarized ground-state configurations. To prove or refute the above conjecture we have to address the question of the lower boundary of the fully spin-polarized domain, i.e., the lowest field strength, at which a fully spin-polarized state represents the ground state of the atom considered. It is evident that this boundary value of the field strength is given by the crossover from a fully spin-polarized ground state to a nonfully spin-polarized ground state with decreasing field strength.

First of all we have to check if the state $|1s^2\rangle$ has the lowest energy of all the states of subset PSP at the point of intersection of the curve $E_{11s^2}(\gamma)$, with the corresponding

Ζ	$ 0_N\rangle$ - $ 1s^2\rangle$	$ 0_N\rangle$ - $ 2p_0\rangle$	$ 2p_0\rangle$ - $ 1s^2\rangle$		$ 2p_0\rangle - 1s^2 2p_0\rangle$ $ 2p_0\rangle - 2p_03 d_{-1}\rangle$ $ 2p_03 d_{-1}\rangle - 1s^2\rangle$	
2	0.711					
3	2.153					
$\overline{4}$	4.567	2.529	4.765451			
5	8.0251	7.923	8.0325			
6	12.577	18.664	12.216			
7		36.849	17.318	17.398		
8		64.720	23.3408	23.985		
9		104.650	30.285	31.735	22.744	30.6125
10		159.138	38.151	40.672	40.537	38.060

TABLE I. Magnetic-field strengths γ (a.u.) for energy-level crossovers in neutral atoms. Ground-state crossovers are underlined.

energy curve for the local ground-state configuration of subset FSP. Following our considerations for the fully spinpolarized case, we can conclude that calculations have to be performed first of all for the $|1s^22p_0\rangle$ and $|1s^22s\rangle$ states.

The numerical calculations show that, for atoms with *Z* ≤ 6 and ions with $Z \leq 7$, the $|1s^2\rangle$ state becomes the ground state while lowering the spin polarization from the maximal absolute value $S_z = -N/2$ to $S_z = -N/2+1$. For heavier atoms and ions we first remark that the $|1s^2\rangle$ state is not the energetically lowest one in the PSP subset at magnetic fields at which its energy becomes equal to the energy of the lowest FSP state. For these atoms and ions the $1s^22p_0$ state lies lower than the $|1s^2\rangle$ state at these field strengths. One can see this behavior for the neon atom in Fig. 1. The second possible PSP local ground state $|1s^22s\rangle$ (not presented in Fig. 1) proves to be less tightly bound at these fields. In what follows, these facts allow a definite clarification of the picture of the global ground-state configuration in the high-field regime. For atoms with $Z \ge 7$ and positive ions with $Z \ge 8$, the intersection points between the $\left| 1s^2 2p_0 \right\rangle$ state and the energetically lowest state in the FSP subspace have to be calculated. As a result, the spin-flip crossover occurs at higher fields than what it would be if $|1s^2\rangle$ was the lowest state in the PSP subspace. In particular, the spin-flip crossover for the neon atom is found to be slightly higher than the point of the crossover $|2p_0\rangle - |2p_03d_{-1}\rangle$, and, therefore, this atom, in the framework of the Hartree-Fock approximation, has only two fully spin polarized configurations, like other neutral atoms and positive ions with $6 \le Z \le 10$. The above conjecture is therefore refuted, and the FSP $|2p_03d_{-1}\rangle$ never represents the global ground-state configuration in the highfield regime for all neutral atoms and positive ions with *Z* ≤ 10 . It should be noted that the situation with the neon atom can be regarded as a transient one due to the closeness of the intersection $|2p_0\rangle$ - $|2p_03d_{-1}\rangle$ to the intersection $|2p_0\rangle$ - $|1s^22p_0\rangle$. This means that we can expect the configuration $|2p_03d_{-1}\rangle$ to be the global ground state for the sodium atom $(Z=11)$. In addition, an investigation of the neon atom carried out on a more precise level than the Hartree-Fock method could also introduce some corrections to the picture described above.

After obtaining new spin-flip points for atoms with 7 \leq Z \leq 10 and ions with 8 \leq Z \leq 10 (which are transition points between the $|2p_0\rangle$ and $|1s^22p_0\rangle$ states) one has to check them with respect to the next (in the order of decreasing field strengths) possible PSP local ground-state configurations. Analogously to the FSP subset these configurations are $|1s^22p_03d_{-1}\rangle$ and $|1s^22s2p_0\rangle$. Numerical calculations show that their energies lie higher than the energy of the $|1s^22p_0\rangle$ configuration at the spin-flip points, and they are therefore excluded from the list of global ground states considered here.

The final picture of the crossovers of the global groundstate configurations is presented in Tables I (for the neutral atoms) and II (for the positive ions A^+). The corresponding values of the field strengths belonging to the point of crossover are underlined in these tables. The field strengths for other close-lying crossovers which actually do not affect the

TABLE II. Magnetic-field strengths γ (a.u.) for energy-level crossovers in positive ions A^+ . Ground-state crossovers are underlined.

Ζ	$ 0_N\rangle$ - $ 1s^2\rangle$	$ 0_N\rangle$ - $ 2p_0\rangle$	$ 2p_0\rangle$ - $ 1s^2\rangle$	$ 2p_0\rangle$ - $ 1s^22p_0\rangle$	$ 2p_0\rangle$ - $ 2p_03d_{-1}\rangle$
3	2.0718				
$\overline{4}$	4.501	1.464			
5	7.957	5.575			
6	12.506	14.536	12.351		
7		30.509	17.429		
8		55.747	23.434	23.849	
9		92.624	30.364	31.612	
10		143.604	38.220	40.559	33.353

FIG. 2. The magnetic-field strengths $(a.u.)$ corresponding to crossovers of energy levels in neutral atoms as functions of the nuclear charge. The filled symbols mark crossovers between global ground-state configurations.

ground-state configuration for *Z*.10.

scenario of the changes of the global ground state are also presented in these tables. In a graphical form these results are illustrated in Figs. 2 (neutral atoms) and 3 (ions). There we show the critical field strengths belonging to the crossovers of selected states of the atoms (ions) as functions of the nuclear charge. The filled symbols mark the crossovers of the energy levels which correspond to the actual transitions of the ground-state configurations, whereas the analogous nonfilled symbols correspond to magnetic-field strengths of the crossovers not associated with changes in the ground state. In these figures one can see the dependencies of the field strengths for various types of crossovers on the charge of the nucleus. In particular, one can see many significant crossovers for $Z=10$ lying very close to each other on the γ axis. This peculiarity, in combination with the behavior of the curve $\gamma(Z)$ for the $|2p_0\rangle - |2p_03d_{-1}\rangle$ crossover, allows one to expect the configuration $|2p_03d_{-1}\rangle$ to become a

Some summarizing remarks with respect to the global ground-state configurations in the high-field regime are in order. Atoms and positive ions with $Z \le 5$ have one groundstate configuration $|0_N\rangle$. Atoms and ions with $6 \le Z \le 10$ possess two high-field configurations. The C atom $(Z=6)$ plays an exceptional role in the sense that it is the only atom which shows the ground state crossover $|2p_0\rangle$ - $|1s^2\rangle$ involving the $|1s^2\rangle$ state as a global ground state.

V. NUMERICAL RESULTS AND DISCUSSION

Tables III–X contain numerical values of the total energies of the neutral atoms and positive ions obtained in our Hartree-Fock calculations. Tables III, IV, V, and VI contain the energies of the neutral atoms in states $|0_N\rangle$, $|2p_0\rangle$, $|1s^2\rangle$,

FIG. 3. Same as Fig. 2, but for the singly positive ions.

and $|2s\rangle$, respectively. The analogous results for the A^+ ions are presented in Tables VII, VIII, IX and X (the results are for states $|0_N\rangle$, $|2p_0\rangle$, $|1s^2\rangle$, and $|2s\rangle$). The energies associated with the points of crossover for the global ground state both in neutral atoms and in their singly positive ions are presented in Table XI. These energy values provide us with the ionization energies at the transition points. Combined with the data of the previous tables, they provide the behavior of the ionization energies of the atoms and the total energies of the atoms and positive ions in the complete highfield region.

In Fig. 4 we present the ionization energies of neutral atoms divided by the ionization energy of the hydrogen atom as a function of the magnetic-field strength. All curves for multielectron atoms at γ <600 lie lower than the curve for hydrogen at the corresponding field strengths. But for γ >1500 the ionization energies of all atoms exceed the ionization energy for the hydrogen atom. Moreover, with growing nuclear charge we observe a stronger increase of the ionization energy for stronger fields accompanied by a shift of the starting point for the growth to the regime of stronger magnetic fields. This strengthening of the binding of the multielectron atoms at strong magnetic fields may be considered a hint of increasingly favorable conditions for the formation of the corresponding negative ions.

Figure 5 presents the ionization energies for the $|0_N\rangle$ states for various field strengths depending on the nuclear charge *Z*, i.e., for all atoms H, He, . . . , Ne. All the field strengths presented in this figure are above the first crossover to another global ground-state configuration. Thus the ion-

TABLE IV. Total energies (a.u.) of neutral atoms in strong magnetic fields in the fully spin-polarized $|2p_0\rangle$ states.

Z	$\gamma = 0.5$	$\gamma = 1$	$\gamma = 2$	$\gamma = 5$	$\gamma = 10$	$\gamma = 20$	γ = 50	$\gamma = 100$	$\gamma = 200$
	-0.224760	-0.260007	-0.297711	-0.347618	-0.382650	-0.413378	-0.445685	-0.463618	-0.476532
2	-2.477333	-2.730171	-3.130766	-3.953993	-4.842630	-6.00481	-8.05248	-10.072	-12.588
3	-5.969573	-6.492478	-7.324937	-9.125540	-11.17884	-13.96583	-19.0436	-24.1951	-30.734
	$4 - 11.06254$	-11.89891	-13.22133	-16.10812	-19.51207	-24.27725	-33.2000	-42.4440	-54.368
5		-19.05098	-20.92634	-25.03513	-29.94166	-36.95414	-50.377973	-64.5298	-83.031
6		-28.0195	-30.4938	-35.96012	-42.52774	-52.02820	-70.51870	-90.275	$-116,4070$
7		-38.8370	-41.9590	-48.9040	-57.29256	-69.5147	$-93,6004$	-119.5977	-154.272
8		-51.5182	-55.3413	-63.877	-74.2380	-89.4093	-119.592	-152.453	-196.522
9		-66.0734	-70.6514	-80.8826	-93.3580	-111.6968	-148.4508	-188.7802	-243.1024
10		-82.5108	$-87,8960$	-99.9271	-114.64655	$-136,36054$	-180.1312	-228.500	-293.944

TABLE V. Total energies (a.u.) of neutral atoms in strong magnetic fields in the $|1s^2\rangle$ states.

Z	$\gamma = 0.5$	$\gamma = 1$	$\gamma = 2$	$\gamma = 5$	$\gamma = 10$	$\gamma = 20$	γ =50	$\gamma = 100$	$\gamma = 200$
2	-2.8144511	-2.688885	-2.289145	-0.532445	$+3.110634$	$+11.319608$	$+38.14390$	$+85.00416$	$+181.10639$
3	-7.58789	-7.666532	-7.662455	-6.942304	-4.617769	$+1.705656$	$+24.97942$	$+68.17347$	$+159.57479$
4	-14.82273	-15.16179	-15.57496	-15.91027	-15.04644	-10.97100	$+7.83395$	$+46.25962$	$+131.4188$
5	-24.5395	-25.20257	-26.11859	-27.59737	-28.27946	-26.68603	-13.06555	$+19.65113$	$+97.1970$
6	-36.7864	-37.8130	-39.3061	-42.06081	-44.38721	-45.44649	-37.57176	-11.36933	$+57.3384$
7	-51.5899	-53.0202	-55.1513	-59.3169	$-63,4083$	-67.27185	-65.5935	-46.5970	$+12.1743$
8	-68.967	-70.8400	-73.6672	-79.3704	-85.3599	-92.1817	-97.0777	-85.8840	-38.0379
9	-88.930	-91.2830	-94.8623	-102.2227	-110.2464	-120.1897	-131.9955	-129.1244	-93.0956
	$10 - 111.491$	-114.3575	-118.7427	-127.8738	$-138,0661$	-151.3018	-170.3322	-176.2422	-152.8395

TABLE VI. Total energies (a.u.) of neutral atoms in strong magnetic fields in the $|2s\rangle$ states.

TABLE VII. Total energies (a.u.) of the high-field ground states $|0_N\rangle$ of positive ions A^+ in strong magnetic fields.

Ζ	$\gamma = 0.5$	$\gamma = 1$	$\gamma = 2$	$\gamma = 5$	$\gamma = 10$	$\gamma = 20$	γ =50
2	-2.2346282	-2.4409898	-2.7888422	-3.5438677	-4.3901481	-5.5215956	-7.5463093
3	-5.640062	-6.114623	-6.894080	-8.629427	-10.651315	-13.4297434	-18.525475
4	-10.51258	-11.31312	-12.59206	-15.42817	-18.820184	-23.612005	-32.61959
5	-16.9017	-18.07243	-19.93091	-24.01520	-28.93504	-36.02020	-49.63544
6	-24.8433	-26.4227	-28.9235	-34.40433	-41.01061	-50.62785	-69.44195
7	-34.3550	-36.3826	-39.5839	-46.5957	-55.04672	-67.41737	-91.94699
8		-47.9633	-51.9215	-60.5880	-71.0369	-86.37441	-117.08457
9			-65.9423	-76.3802	-88.9723	-107.4850	-144.8061
10			-81.6509	-93.9710	-108.8443	-130.7348	-175.0743
Ζ	$\gamma = 100$	$\gamma = 200$	$\gamma = 500$	$\gamma = 1000$	$\gamma = 2000$	$\gamma = 5000$	$\gamma = 10000$
2	-9.5605466	-12.071443	-16.2898727	-20.2706955	-25.028351	-32.65713	-39.548989
3	-23.699944	-30.260769	-41.50393	-52.323018	-65.47657	-86.9940	-106.8134
4	-41.93414	-53.90638	-74.73619	-95.07513	-120.11947	-161.7052	-200.5709
5	-63.947265	-82.55711	-115.33672	-147.71743	-187.99221	-255.6619	-319.6394
6	-89.51120	-115.87500	-162.80039	-209.6030	-268.2990	-367.8817	-462.931
7	-118.45429	-153.5960	-216.7194	-280.1976	-360.3670	-497.502	-629.454
8	-150.6447	-195.5087	-276.7565	-359.0516	-463.6191	-643.7651	-818.311
9	-185.9795	-241.4411	-342.6284	-445.780	-577.553	-805.9918	-1028.687
10	-224.3773	-291.251	-414.09358	-540.0501	-701.7295	-983.5779	-1259.8444

TABLE VIII. Total energies (a.u.) of positive ions A^+ in strong magnetic fields in the fully spin polarized $|2p_0\rangle$ states.

Z	$\gamma = 0.5$	$\gamma = 1$	$\gamma = 2$	$\gamma = 5$	$\gamma = 10$	$\gamma = 20$	γ = 50	$v=100$	$\gamma = 200$
	$3 - 5.450607$	-5.790277	-6.354440	-7.6155748	-9.07498561	-11.052577	-14.60723	-18.1514	-22.5884
	$4 - 10.71847$	-11.39964	-12.50590	-14.969431	-17.896267	-21.986880	-29.579033	-37.35540	-47.2987
	$5 - 17.58187$	-18.62668	-20.31984	-24.06890	-28.57270	-35.01093	-47.26504	-60.06278	-76.6646
	$6 - 26.2094$	-27.6300	-29.9454	-35.09561	$-41,30920$	-50.308209	-67.773425		$-86,30316 - 110,6170$
	$7 - 36.6424$	-38.4731	-41.4488	-48.10689	-56.17490	-67.94617	-91.12203	$-116.03024 - 149.0175$	
8			-54.8620	-63.1295	-73.19399	-87.94947	-117.30577	$-149.19626 - 191.75146$	
9				-80.1774	-92.3729	-110.32119	$-146,3062$	-185.7521	-238.7295
10				-99.2581	-113.7112	-135.0539	-178.0971	-225.6432	-289.8700

ization energies in this figure represent the differences between the energies of the high-field ground states of the neutral atoms and the corresponding singly charged positive ions. The curve for γ =2000 can be considered a prototypical example of the general properties of the dependencies $E_{\text{Ion}}(Z)$. For small values of *Z* this curve shows increasing values for E_{Ion} with increasing *Z*, it has a maximum at *Z* $=$ 5, and for Z > 5 it decreases with increasing *Z*. Analogous curves for lower-field strengths have their maxima at lower values of *Z*. At $\gamma = 1000$ the ionization energy shows its maximal value at $Z=2$, whereas the ionization energies for γ =500 and 200 decrease monotonically with increasing *Z*. On the other hand, for $\gamma = 5000$ and $\gamma = 10000$ we obtain a monotonically increasing behavior of the ionization energy for the whole range $1 \le Z \le 10$ of nuclear charges investigated in the present work. The behavior described above results from a competition of two different physical mechanisms which impact the binding energy of the outermost electron in the high-field ground-state Hartree-Fock configuration. The first mechanism is the lowering of the binding energy of the outermost electron while increasing the absolute value of its magnetic quantum number $|m|$, provided that this electron feels a constant nuclear charge. The latter assumption is a rough approximation of the case of relatively weak fields when the inner $Z-1$ electrons screen the Coulomb field of the nucleus more or less effectively. The second and opposite tendency is associated with the decrease of the efficiency of this screening in extremely strong magnetic fields due to the fact that the geometry of the wave functions tends to be one dimensional in these fields. As a result the effect of the increasing effective nuclear charge exceeds the effect of the growth of |m| with increasing *Z* for the highfield ground-state configurations. Continuing this qualitative consideration we point out that at each fixed γ the influence of the magnetic field on the inner electrons becomes less and less significant as *Z* increases, which is due to the dominance of the Coulomb attraction potential of the nucleus over the magnetic-field interaction. This has to result in a significant screening of the nuclear charge by these electrons. As a result the functions $E_{\text{Ion}}(Z)$ for strong fields defined on the whole interval $1 \leq Z \leq +\infty$ have maxima at some values for *Z* and decrease for sufficiently large values of *Z*.

Next we provide a comparison of the present results with adiabatic Hartree-Fock (HF) calculations which were carried out for multielectron atoms in Refs. $[13,16]$. We compare our results on the Hartree-Fock electronic structure of atoms in strong magnetic fields with results obtained by Neuhauser *et al.* [13] via a one-dimensional *adiabatic* Hartree-Fock approximation. The calculations in this work were carried out for the four field strengths γ =42.544, 212.72, 425.44, and 2127.2. For *Z*<9 and all these field strengths, and for *Z* $=10$ at the three largest values of these fields, the Hartree-Fock wave functions of the ground states are reported to be fully spin polarized with no nodes crossing the *z* axis. This conclusion differs from our result for γ =42.544. According to our calculations, at γ =42.544 the wave functions without nodes crossing the *z* axis represent the ground states of atoms with $Z \le 7$ (i.e., H, He, Li, Be, B, C, and N), whereas for the atoms with $8 \le Z \le 10$ (i.e., O, F, and Ne) the wave functions of the ground states are fully spin polarized with one nodal surface crossing the *z* axis. A numerical comparison of our results with those of Refs. $[13,16]$ is shown in Table XII. All our values lie lower than the values of these adiabatic calculations. Since our total energies are upper bonds to the exact

TABLE IX. Total energies (a.u.) of positive ions A^+ in strong magnetic fields in the $|1s^2\rangle$ states.

Z	$\gamma = 0.5$	$\gamma = 1$	$\gamma = 2$	$\gamma = 5$	$\gamma = 10$	$\gamma = 20$	γ = 50	$\gamma = 100$	$\gamma = 200$
	$3 - 7.217983$	-7.164014	-6.962999	-5.850510	-3.110916	$+3.748961$	$+27.964647$	$+72.09337$	$+164.66867$
	$4 - 14.49163$	-14.70591	-14.95181	-14.96820	-13.75773	-9.217910	$+10.42836$	$+49.70820$	$+135.95916$
	$5 - 24.2429$	-24.78674	-25.54108	-26.71999	-27.08417	-25.06410	-10.65835	$+22.86883$	$+101.46468$
	$6 - 36.5110$	-37.4273	-38.7685	-41.23663	-43.25931	-43.91255	-35.28670	-8.30098	$+61.43184$
	$7 - 51.3324$	-52.6586	-54.6467	-58.5397	-62.33918	-65.81095	-63.40519	-43.64463	$+16.13450$
	$8 - 68.725$	-70.500	-73.1912	-78.6347	-84.3435	-90.78611	-94.97381	-83.03149	-34.19097
-9		-90.962	-94.412	-101.5242	-109.2779	-118.8537	-129.9684	-126.3622	-89.3511
10		-114.054	-118.316	-127.2092	-137.1413	-150.0207	-168.3763	-173.5637	-149.190

Ζ	$\gamma = 0.5$	$\gamma = 1$	$\gamma = 2$	$\gamma = 5$	$\gamma = 10$	$\gamma = 20$	γ = 50	$\gamma = 100$
3	-5.482414	-5.725577	-6.125201	-7.168333	-8.489089	-10.34858	-13.7869	-17.2792
$\overline{4}$	-10.88665	-11.48854	-12.39822	-14.49715	-17.22586	-21.18159	-28.68072	-36.4493
.5	-17.83551	-18.82816	-20.35231	-23.60163	-27.76915	-34.02228	-46.19941	-59.0447
6		-27.9171	-30.0961	-34.69799	-40.35174	-49.03420	$-66,40440$	-85.0457
				-47.83068	-55.11866	-66.30856	-89.28664	-114.3624
8				-62.9964	-72.1475	-85.93862	-114.84057	-146.90998
9				-80.1907	-91.4315	-108.01546	-143.08156	-182.6254
10				-99.4125	-112.9303	-132.5918	-174.0451	-221.4723

TABLE X. Total energies (a.u.) of the positive ions A^+ in strong magnetic fields in the $|2s\rangle$ states.

values we consider our HF results as being closer to the exact values compared to the results of the adiabatic HF calculations. Therefore, on the basis of our calculations combined with the results of Refs. $[13,16]$, we can obtain an idea of the degree of the applicability of the adiabatic approximation for multielectron atoms for different field strengths and nuclear charges. It is well known that the precision of the adiabatic approximation decreases with decreasing field strength. The increase of the relative errors with decreasing field strength is clearly visible in the table. On the other hand, the relative errors of the adiabatic approximation pos-

sess the tendency to increase with growing *Z*, which is manifested by the scaling transformation $E(Z, \gamma) = Z^2 E(1, \gamma/Z^2)$ (see, e.g., Refs. $[8,20]$) well known for hydrogenlike ions. The behavior of the inner electrons is to some extent similar to the behavior of the electrons in the corresponding hydrogenlike ions. Therefore, their behavior is to lowest order similar to the behavior of the electron in the hydrogen atom at magnetic field strength γ/Z^2 , i.e., this behavior can be less accurately described by the adiabatic approximation at large *Z* values. The absolute values of the errors in the total energy associated with the adiabatic approximation are in many

TABLE XI. Total energies $(a.u.)$ of the neutral atoms and ions A^+ at the crossover points of the groundstate configurations.

Ζ	γ	Atomic state(s)	$-E(Atomic)$	Ionic state(s)	$-E(A^+)$
2	0.711	$ 0_N\rangle$, $ 1s^2\rangle$	2.76940	$ 0_N\rangle$	2.32488
3	2.153	$ 0_N\rangle$, $ 1s^2\rangle$	7.64785	$ 0_N\rangle$	7.00057
	2.0718	$ 1s^2\rangle$	7.65600	$ 0_{N}\rangle, 1s^{2}\rangle$	6.94440
$\overline{4}$	4.567	$ 0_N\rangle$, $ 1s^2\rangle$	15.9166	$ 0_N\rangle$	15.07309
	4.501	$ 1s^2\rangle$	15.91625	$ 0_N\rangle$, $ 1s^2\rangle$	15.01775
5	8.0251	$ 0_N\rangle$, $ 1s^2\rangle$	28.18667	$ 0_N\rangle$	27.16436
	7.957	$ 1s^2\rangle$	28.17996	$ 0_N\rangle$, $ 1s^2\rangle$	27.10004
6	18.664	$ 0_N\rangle$, $ 2p_0\rangle$	50.9257	$ 0_N\rangle$	49.50893
	14.536	$ 2p_0\rangle$	47.23836	$ 0_N\rangle, 2p_0\rangle$	45.77150
	12.351	$ 2p_0\rangle$	45.07386	$ 2p_0\rangle$, $ 1s^2\rangle$	43.72095
	12.216	$ 2p_0\rangle$, $ 1s^2\rangle$	44.9341	$ 1s^2\rangle$	43.70075
7	36.849	$ 0_N\rangle$, $ 2p_0\rangle$	84.4186	$ 0_N\rangle$	82.58182
	30.509	$ 2p_0\rangle$	79.34493	$ 0_N\rangle, 2p_0\rangle$	77.41246
	17.429	$ 2p_0\rangle$	66.72786	$ 2p_0\rangle$, $ 1s^2\rangle$	65.26170
	17.398	$ 2p_0\rangle$, $ 1s^22p_0\rangle$	66.69306	$ 1s^2\rangle$	65.25362
8	64.720	$ 0_N\rangle$, $ 2p_0\rangle$	130.6806	$ 0_N\rangle$	128.4054
	55.747	$ 2p_0\rangle$	124.1125	$ 0_N\rangle$, $ 2p_0\rangle$	121.69825
	23.985	$ 2p_0\rangle$, $ 1s^22p_0\rangle$	94.3773	$ 2p_0\rangle$	92.78308
	23.849	$ 1s^22p_0\rangle$	94.3336	$ 2p_0\rangle$, $ 1s^22p_0\rangle$	92.62502
9	104.650	$ 0_N\rangle, 2p_0\rangle$	191.8770	$ 0_N\rangle$	189.1446
	92.624	$ 2p_0\rangle$	183.6944	$ 0_N\rangle, 2p_0\rangle$	180.7819
	31.735	$ 2p_0\rangle$, $ 1s^22p_0\rangle$	128.1605	$ 2p_0\rangle$	126.4414
	31.612	$\ket{1s^22p_0}$	128.1125	$ 2p_0\rangle$, $ 1s^22p_0\rangle$	126.2897
10	159.138	$ 0_N\rangle$, $ 2p_0\rangle$	270.220	$ 0_N\rangle$	267.0112
	143.604	$ 2p_0\rangle$	260.2740	$ 0_N\rangle, 2p_0\rangle$	256.8459
	40.672	$ 2p_0\rangle$, $ 1s^22p_0\rangle$	168.4734	$ 2p_0\rangle$	166.6327
	40.559	$\ket{1s^22p_0}$	168.4217	$ 2p_0\rangle$, $ 1s^22p_0\rangle$	166.4863

FIG. 4. Ionization energies of neutral atoms divided by the ionization energy of the hydrogen atom as a function of the magnetic-field strength $(a.u.).$

cases larger than the corresponding values of the ionization energies.

To conclude this section we discuss briefly three issues, which could affect the precision of the results presented above. These issues are electron correlations, effects due to

FIG. 5. Ionization energies of the $|0_N\rangle$ states of the neutral atoms ($1 \le Z \le 10$) for different magnetic-field strengths.

levels. In most cases their influence on the latter values is much smaller due to the fact that they involve differences of total energies for quantum states possessing a similar atomic core. Let us start by addressing the problem of electronic correlations, which is the critical problem for the precision of the Hartree-Fock calculations. The final evaluation of the correlation effects is possible only on the basis of exact calculations going beyond the Hartree-Fock approximation. Therefore, here we can give here only qualitative arguments based on the geometry of the wave function and on existing calculations for less complicated systems. The dependence of the ratio of the correlation energy and the total binding energy for the two ground-state configurations of the helium atom was investigated in Ref. [21]. This ratio for the $1s²$ state decreases with growing γ , from 1.4% at $\gamma=0$ to about 0.6% at γ =100. The same ratio for the $1s2p_{-1}$ state (the high-field ground-state configuration) increases with growing γ . However, for all the field strengths considered it remains essentially smaller than the values for the $1s²$ state. This result for the helium atom in strong magnetic fields allows us to speculate that for the field strengths considered here the correlation energy for atoms and positive ions heavier than the helium atom does not exceed their corresponding values without fields. Due to the similar geometry of the inner shells in the participating electronic configurations, we do not expect a major influence of the correlation effects both on the field strengths of the crossovers of the ground-state configurations within the subsets FSP or PSP and on the ionization energies if the states of a neutral atom and the positive ion belong to the same subset. On the other hand, the properties associated with configurations from different subsets (for instance, values of the spin-flip crossover field strengths) can be affected more strongly by correlation effects.

the finite nuclear mass, and relativistic corrections. For all these effects we have to distinguish between their influence on the total energy and on other quantities like the ionization energy and the field strength for the crossover of the energy

TABLE XII. Absolute values of the total energies (keV) of the high-field ground states of neutral atoms in strong magnetic fields compared with the literature. $B_{12} = B/(10^{12} \text{ G})$. IS, present work. NKL, results by Neuhauser, Koonin, and Langanke [13]. DHG, results by Demeur, Heenen and Godefroid [16]. $|2p_0\rangle$, results for states $|2p_0\rangle$ at the points where they are the ground states.

	$B_{12} = 0.1$			$B_{12} = 0.5$			$B_{12}=1$		$B_{12} = 2.3505$ ($\gamma = 1000$)	$B_{12} = 5$	
Z	IS $(2p_0\rangle)$	IS	NKL	IS	NKL	IS	NKL	IS	DHG	IS	NKL
		0.07781	0.0761	0.13114	0.130	0.16222	0.161	0.20851	0.206	0.25750	0.2550
2		0.26387	0.255	0.46063	0.454	0.57999	0.574	0.76279	0.754	0.96191	0.9580
3		0.54042	0.516	0.96180	0.944	1.22443	1.209	1.63429	1.611	2.08931	2.0760
4		0.89833	0.846	1.61624	1.580	2.07309	2.042	2.79610	2.746	3.61033	3.5840
5		1.33229	1.238	2.41101	2.347	3.10924	3.054	4.22674	4.139	5.49950	5.4560
6		1.83895	1.678	3.33639	3.22	4.31991	4.20	5.90872	5.773	7.73528	7.60
		2.41607	2.17	4.38483	4.22	5.69465	5.54	7.82757		10.29919	10.20
8	3.08253	3.06214	2.71	5.55032	5.32	7.22492	7.02	9.97107		13.17543	13.00
9	3.82966	3.77607	3.36	6.82794	6.51	8.90360	8.63	12.32880		16.34997	16.10
10	4.65087	4.55698		8.21365	7.819	10.72452	10.39	14.89168		19.81072	19.57

Our second issue is the influence of the finite nuclear mass on the results presented above. A discussion of this problem was provided in Ref. [11] and references therein. Importantly there exists a well-defined procedure which tells us how to relate the energies for an infinite nuclear mass to those for a finite nuclear mass. The corresponding equations are exact for hydrogenlike systems, and provide the lowest order mass corrections $O(m/M)$ (*m* and *M* are the electron and total mass, respectively) for general atoms and ions. Essentially they consist of a redefinition of the energy scale (atomic units \rightarrow reduced atomic units, due to the introduction of the reduced mass) and an additional energy shift $-(1/M_0)\gamma(M+S_7)$, where M_0 is the nuclear mass. The first effect can simply be ''included'' in our results by taking the energies in reduced a.u. instead of a.u. The mentioned shift can become relevant for high fields. However, it can easily be included in the total energies presented here. We emphasize that it plays a minor role in the region of the crossovers of the ground-state configurations, and decreases significantly with the increasing mass of the atom (nucleus).

Relativistic calculations for the hydrogen atom and hydrogenlike ions were performed by Lindgren and Virtamo $[22]$ and Chen and Goldman [23]. Our considerations are based on the work of Chen and Goldman $[23]$, which contains results for 1*s* and $2p_{-1}$ states for a broad range of magneticfield strengths. Interpolating their results for the 1*s* state and using well-known scaling transformations, we can conclude that in the least favorable case of $Z=10$ relativistic corrections $\delta E = (E^{\text{relativistic}} - E^{\text{nonrelativistic}})/|E^{\text{nonrelativistic}}|$ have to be of the order 4×10^{-4} for $\gamma=200$ and 2×10^{-4} for $\gamma=10^{4}$. The relativistic corrections for the $2p_{-1}$ state at relatively strong fields appear to be of the same order of magnitude as or smaller than those for the 1*s* state. Thus, making a reasonable assumption that relativistic corrections for both inner and outer electrons are similar to those in hydrogenlike ions with a properly scaled nuclear charge, we can evaluate $|\delta E|$ \leq 4 \times 10⁻⁴ for *Z* = 10 and a lesser amount for lower *Z* values. The same relative correction can also be expected for the ionization energies and energy values used for the determination of the crossovers of the electronic configurations.

VI. SUMMARY

In the present work we have applied our two-dimensional Hartree-Fock method to the magnetized neutral atoms H, He, Li, Be, B, C, N, O, F, and Ne in the high-field regime, which is characterized by fully spin-polarized electronic shells. Additionally we have studied the crossover from fully spinpolarized to partially spin-polarized global ground-state configurations. The highest field strength investigated was γ $=10000$. Our single-determinant Hartree-Fock approach supplies us with exact upper bounds for the total energy. A comparison with adiabatic calculations in the literature shows the decrease of the precision of the adiabatic approximation with growing *Z*.

The investigation of the geometry of the spatial part of the electronic wave function demonstrates that in the high-field limit this wave function is a composition of the lowest Landau orbitals with absolute values of the magnetic quantum number growing from $|m|=0$ up to $|m|=N-1$, where *N* is the number of the electrons: i.e. we have the series 1*s*, $2p_{-1}$, $3d_{-2}$, For atoms with $2 \le Z \le 5$ these states of type $1s2p_{-1}3d_{-2}...$ represent the complete set of fully spin-polarized ground-state configurations. Heavier atoms 6 \leq *Z* \leq 10 have one intermediate ground-state configuration associated with the low-field end of the fully spin polarized region. This state contains one $2p_0$ -type orbital (i.e. the orbital with a negative-*z* parity and $|m|=0$) instead of the orbital with the positive-*z* parity and the maximal value of $|m|$. Extrapolating our data as a function of the nuclear charge *Z*, we expect that a third fully spin-polarized ground-state configuration occurs first for $Z=11$, i.e., the sodium atom. The third configuration is suggested to be the $|2p_03d_{-1}\rangle$ state. The critical field strength which provides the crossover from the partially spin-polarized regime to the fully spin-polarized regime depends sensitively on the changes of the geometry of the wave functions. Indeed a number of different configurations have been selected as candidates for ground states in the crossover regime, and only concrete calculations could provide us with a final decision about the energetically lowest state of the non-fully spin polarized electronic states. Generally speaking, all the spin-flip crossovers mentioned above involve a pairing of the 1*s* electrons, i.e., the pair of orbitals $1s^2$. The carbon atom (*Z*=6) plays an exceptional role since it is the only neutral atom which possesses two fully spin-polarized configurations and the $|1s^2\rangle$ state as a global non-fully spin-polarized ground-state configuration. The spin-flip crossover of the carbon atom preserves the total magnetic quantum number. All other atoms N, O, F, and Ne $(7 \le Z \le 10)$ instead possess the $|1s^2 2p_0\rangle$ configuration as a nonfully spin-polarized ground state for strong fields. We have determined the positions, i.e., field strengths, of the crossovers of the ground states. Beyond this total energies have been provided for many field strengths for several lowlying excited states.

An analogous investigation has been carried out for singly charged positive ions $2 \le Z \le 10$. The structure of the fully spin-polarized ground-state configurations for these ions is the following: The ions with $3 \le Z \le 5$ have one fully spinpolarized ground-state configuration analogous to the highfield limit of the neutral atoms. For $6 \le Z \le 10$, analogously to the neutral atoms, there exist two fully spin-polarized

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ground-state configurations. Depending on the values of the nuclear charge number *Z*, the spin-flip transitions associated with the lowering of the spin polarization with decreasing field strength also lead to wave functions of different spatial symmetries. These data, combined with the data for neutral atoms, allow us to obtain the ionization energies of the atoms. The dependencies of the ionization energies on the nuclear charge at fixed field strength generally exhibit maxima at certain values of *Z*. The positions of these maxima shift to larger values of *Z* with increasing field strength. We provide some qualitative arguments explaining this behavior of $E_{\text{Ion}}(Z)$. Finally, we have made some remarks on the interactions going beyond the present level of investigation, i.e., correlations and finite nuclear mass effects as well as relativistic corrections.

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