# Total electron density from the *s*-electron density

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Based on a theorem of Theophilou [Philos. Mag. B **69**, 771 (1994)], the total electron density in a central field atom is determined from the Kohn-Sham *s*-electron density. Exchange-only density-functional calculations for atoms are presented. [S1050-2947(98)10105-1]

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## I. INTRODUCTION

In a recent paper [1], it was shown for a bare Coulomb potential energy  $-Ze^2/r$  that the total electron density for an arbitrary number of closed shells is given by

$$n(r) = \frac{2Z}{a_0} \int_r^\infty n_s(r) dr,$$
 (1)

where  $n_s$  is the electron density of *s* shells. Equation (1) can also be transformed into the form

$$\frac{\partial n(r)}{\partial r} = -\frac{2Z}{a_0} n_s(r), \qquad (2)$$

which is the spatial generalization of Kato's theorem [2].

Equation (1) shows that the total electron density n can be calculated if the *s*-electron density  $n_s$  is known. The total density can be determined from  $n_s$ , not only for a bare Coulomb field. In a spherically symmetric system one can always determine n if  $n_s$  is known. Of course, the relation between n and  $n_s$  is not so simple as in the bare Coulomb field [Eqs. (1) and (2)].

Let us consider a noninteracting electron system in a central potential v(r). If the *s* density  $n_s$  is given, and is *v* representable, the number  $N_s$  of the *s* electrons can be computed by integrating  $n_s$  over space. Applying the minimum principle for the  $N_s$ -electron functional  $\langle \Phi | \hat{H} | \Phi \rangle$ , where

$$\hat{H} = \sum_{i=1}^{N_s} \left[ \frac{-\nabla_i^2}{2m} + v(r_i) \right], \tag{3}$$

and the Slater determinant  $|\Phi\rangle$  is constructed solely from spin orbitals belonging to the subspace characterized by the angular momentum quantum number l=0, one can obtain the inequalities

$$\langle \Phi_0 | \hat{T} | \Phi_0 \rangle + \langle \Phi_0 | \hat{V} | \Phi_0 \rangle < \langle \Phi_0' | \hat{T} | \Phi_0' \rangle + \langle \Phi_0' | \hat{V} | \Phi_0' \rangle,$$
(4)

$$\langle \Phi_0' | \hat{T} | \Phi_0' \rangle + \langle \Phi_0' | \hat{V}' | \Phi_0' \rangle < \langle \Phi_0 | \hat{T} | \Phi_0 \rangle + \langle \Phi_0 | \hat{V}' | \Phi_0 \rangle,$$
(5)

where  $\hat{T} = \sum_{i=1}^{N_s} (\nabla_i^2 / 2m)$ ,  $\hat{V} = \sum_{i=1}^{N_s} v(r_i)$ ,  $\hat{V}' = \sum_{i=1}^{N_s} v'(r_i)$ , vand v' are different external potentials, and  $|\Phi_0\rangle$  and  $|\Phi'_0\rangle$  are the ground-state Slater determinants belonging to the potentials v and v', respectively. Using the equalities

$$\langle \Phi_0 | \hat{V} | \Phi_0 \rangle = \int n_s v \, dr, \tag{6}$$

$$\langle \Phi_0 | \hat{V}' | \Phi_0 \rangle = \int n_s v' dr, \qquad (7)$$

$$\langle \Phi_0' | \hat{V} | \Phi_0' \rangle = \int n_s' v \, dr, \qquad (8)$$

$$\langle \Phi_0' | \hat{V}' | \Phi_0' \rangle = \int n_s' v' dr.$$
(9)

Eqs. (4), and (5) lead to the inequality

$$\int [n'_{s}(r) - n_{s}(r)][v(r) - v'(r)] > 0; \qquad (10)$$

therefore, there is a single external potential which corresponds to the given  $n_s$ , i.e., there exists a one-to-one correspondence between the potential and the *s*-state density. This theorem was stated and proved by Theophilou [3], not only for l=0 but also for arbitrary l. In a recent paper [4], Theophilou and Gidopoulos also proved the following theorem which may be regarded as the generalization of the aforementioned statement: "For a non-interacting system of N electrons, the part of the density due to the spin orbitals that transforms according to a certain irreducible representation of a group, determines uniquely the ground state." Using the Kohn-Sham method for handling the many-electron problem, it can be stated that in a spherically symmetric system the Kohn-Sham s-density uniquely determines the Kohn-Sham potential and the ground state. In this paper, a numerical iterative method is proposed to construct the Kohn-Sham potential and the total density in the knowledge of the s-electron density.

# II. S-STATE DENSITY WITH ONE AND TWO S SHELLS

If the *s*-state density is the 1*s* electron density belonging to the occupation number  $\omega_1$ ,

$$n_s = \omega_1 |u_{1s}|^2, \tag{11}$$

from the Kohn-Sham equation

<u>57</u>

3458

$$\left(-\frac{\nabla^2}{2}+v_{KS}(r)\right)u_{1s}(r)=\varepsilon_{1s}u_{1s}(r),\qquad(12)$$

the Kohn-Sham potential can be trivially constructed:

$$v_{KS}(r) = \varepsilon_{1s} + \frac{1}{2n_s^{1/2}} \nabla^2 n_s^{1/2}.$$
 (13)

The one-electron energy  $\varepsilon_{1s}$  can be determined from the asymptotic decay of  $n_s$ ,

$$n_s \sim \exp\{-2\sqrt{2|\varepsilon_{1s}|}r\}.$$
 (14)

If the s-state density is obtained from two s shells, i.e.,

$$n_s = \omega_1 |u_{1s}|^2 + \omega_2 |u_{2s}|^2, \qquad (15)$$

where  $\omega_1$  and  $\omega_2$  are the occupation numbers, the Kohn-Sham potential can be constructed in the following way: The method developed earlier for the ground state of the Be atom [5] is applied. Following Dawson and March [6], the phase function  $\theta$  is introduced:

$$P_{1s} = \frac{1}{\sqrt{\omega_1}} \varrho_s^{1/2} \cos \theta, \qquad (16)$$

$$P_{2s} = \frac{1}{\sqrt{\omega_2}} \varrho_s^{1/2} \sin \theta, \qquad (17)$$

where  $P_{1s}$  and  $P_{2s}$  are the radial wave functions, and  $\varrho_s = 4r^2 \pi n_s$  is the radial *s*-electron density. Equations (16) and (17) and the radial Kohn-Sham equations

$$P_{is}'' + 2(\varepsilon_{is} - v_{KS})P_{is} = 0, \quad i = 1,2$$
(18)

lead to the relation between the *s*-state electron density  $\varrho_s$  and phase  $\theta$ :

$$\theta'' + \frac{\varrho'_s}{\varrho_s} \theta' = 2\xi \sin(2\theta), \qquad (19)$$

where

$$\xi = \frac{\varepsilon_{1s} - \varepsilon_{2s}}{2}.$$
 (20)

 $\varrho_s$  can be explicitly written as

$$\varrho_s = \frac{1}{\theta'} \exp(2\xi h), \qquad (21)$$

where

$$h(r) = \int \frac{\sin(2\theta)}{\theta'} dr.$$
 (22)

The Kohn-Sham potential takes the form

$$v_{\rm KS} = \frac{1}{4} \frac{\varrho_s'}{\varrho_s} - \frac{1}{8} \left(\frac{\varrho_s'}{\varrho_s}\right)^2 - \frac{1}{2} (\theta')^2 + \xi \cos(2\theta) + \beta,$$
(23)

where

$$\beta = \frac{\varepsilon_{1s} + \varepsilon_{2s}}{2}.$$
 (24)

From Eq. (19),  $\theta$  can be determined from  $\varrho_s$  using the normalization and orthogonality conditions

$$\varrho_s \cos^2 \theta dr = \omega_1, \qquad (25)$$

$$\int \varrho_s \sin^2 \theta dr = \omega_2, \qquad (26)$$

$$\int \varrho_s \sin^2(2\theta) dr = 0.$$
 (27)

Then Eq. (23) gives  $v_{\rm KS}$ . If the Kohn-Sham potential is known, the solutions of the Kohn-Sham equations yield all the occupied wave functions, and the total electron density can be given. Considering the special case of a three-level spherically system, with the obtained  $v_{\rm KS}$  one has just to solve the third radial Kohn-Sham equation

$$P_{2p}'' + 2\left(\varepsilon_{2p} - v_{\rm KS} - \frac{l(l+1)}{2r^2}\right)P_{2p} = 0, \quad l = 1 \quad (28)$$

to obtain the total density.

#### III. S-STATE DENSITY WITH SEVERAL S SHELLS

Recently, there has been a considerable interest in constructing the Kohn-Sham potential with knowledge of the total density [7–12]. One of the authors [7] also proposed a numerical, iterative method that can be modified to obtain the Kohn-Sham potential, if the *s*-electron density is known. Starting out from a properly chosen starting potential the Kohn-Sham equations

$$\left[-\frac{1}{2}\nabla^2 + v_{\rm KS}(r)\right]u_i(\mathbf{r}) = \varepsilon_i u_i(\mathbf{r}) \tag{29}$$

are solved. Then the s-state density

$$n_s = \sum_{\substack{i\\(l=0)}}^{i} 2u_i^2(\mathbf{r}) \tag{30}$$

is calculated and compared with the input *s* density. If the difference is larger than an appropriately chosen small constant, the potential of the following iteration is constructed. Then the Kohn-Sham equations are solved again, and so on. If convergency is achieved then the true Kohn-Sham potential is obtained, since it is uniquely determined by the *s* state density. In principle, in spherically symmetric systems the atomic number *Z* can be determined from Kato's theorem [2]

$$\left(\frac{\partial n}{\partial r}\right)_{(r=0)} = -2Zn(r=0). \tag{31}$$

The same expression can also be written for  $n_s$ . Consequently, the atomic number Z can be readily obtained from

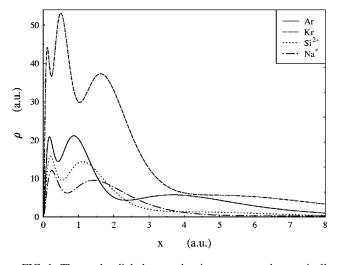


FIG. 1. The total radial electron density  $\rho$  computed numerically from the *s*-state density as the function of *x* for the atoms Ar and Kr, and the ions Na<sup>+</sup> and Si<sup>2+</sup> (in a.u.). *x* is defined as  $x = r/(\mu Z^{1/3})$ , where *Z* is the atomic number,  $\mu = (1/2)(3\pi/4)^{2/3}$ , and *r* is the radial distance from the nucleus.

 $n_s$ . On the other hand, the asymptotic behavior of the Kohn-Sham potential is given by

$$v_{\rm KS}(r) = -\frac{Z - N - 1}{r} \quad \text{as} \ r \to \infty. \tag{32}$$

From this expression, in principle, the total number of electrons N can be extracted. With a knowledge of  $v_{KS}$  and N, the total electron density can be found by solving the Kohn-Sham equations.

# **IV. EXCHANGE-ONLY CALCULATIONS**

In an earlier paper [7], Kohn-Sham one-electron orbitals and energies were determined from Hartree-Fock oneelectron orbitals [13]. It was shown [14,15] that the Kohn-Sham potential leading to the Hartree-Fock density is very close to the exchange-only Kohn-Sham potential, so we can loosely refer to it as the exchange-only case. As a result of this kind of calculation, one obtains the exchange-only oneelectron energies and orbitals, from which the exchange-only  $n_s$  can be constructed. Applying  $n_s$  as an input *s*-electron density following the procedure described above, the exchange-only total electron density *n* can be computed.

This method has been applied to several atoms and ions with closed shells. Some typical results of these calculations are displayed in the figures. Figure 1 presents the total radial electron density  $\varrho = 4 \pi r^2 n$  for the atoms Ar and Kr and the ions Na<sup>+</sup> and Si<sup>2+</sup>. The calculations performed show that the  $d = 100|\varrho - \varrho_{\rm HF}|/\varrho_{\rm HF}$  relative difference between  $\varrho$  and the Hartree-Fock density  $\varrho_{\rm HF}$  is less than 5% for the consid-

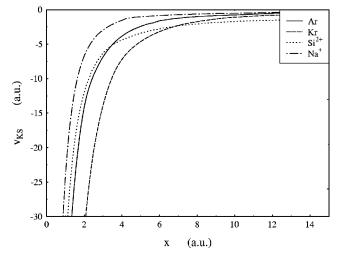


FIG. 2. The Kohn-Sham potential  $v_{\rm KS}$  for the Ar and Kr atoms and the ions Na<sup>+</sup> and Si<sup>2+</sup> (in a.u.).

ered systems, i.e., the Kohn-Sham density is close to the Hartree-Fock density, as expected because of the criterion for self-consistency. The Kohn-Sham potentials are plotted in Fig. 2.

## V. SUMMARY AND DISCUSSION

As we pointed out, the *s*-state contribution to the electron density contains all information about the ground-state system. If the *s*-electron density is given, it is possible to compute the Kohn-Sham potential and the total electron density. In the present work this problem was solved by the aid of a numerical method described above. As an illustration, calculations were performed for spherically symmetric systems in the exchange-only density-functional case.

According to the theorem of Theophilou [3], the total electron density can be determined not only from the s-electron density, but can also be obtained from the p-electron density, a numerical demonstration of which might be the subject of another paper.

The total electron density can be calculated from the d-electron density, and so on, i.e., the l part of the density contains all the information that is needed to obtain the total electron density. These results are valid for noninteracting systems, i.e., for Kohn-Sham systems. The problem of whether a similar theorem exists for interacting system is still an open question (though there is a conjecture [3] that it is probably true).

# ACKNOWLEDGMENTS

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