Laser magnetic-resonance measurement of the ${}^{3}P_{1}$ - ${}^{3}P_{2}$ fine-structure splittings in ${}^{17}O$ and ${}^{18}O$

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The ${}^{3}P_{1}$ - ${}^{3}P_{2}$ fine-structure splittings of 17 O and 18 O have been measured by far-infrared laser magnetic-resonance spectroscopy. The signals for 17 O were detected with a sample in its natural abundance (0.038%). The isotopic shifts from the corresponding interval in 16 O are +13.0 MHz for 17 O and +23.5 MHz for 18 O, in accord with theoretical expectations.

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I. INTRODUCTION

The fine-structure intervals of the oxygen atom in its ground ${}^{3}P$ state fall in the far-infrared (FIR) region of the spectrum. The transitions between these levels provide a simple, direct, and sensitive method of monitoring O atoms, both in the atmosphere [1] and in astrophysical sources [2,3]. Although these intervals have been known for a long time from optical measurements [4], they have been measured much more accurately in recent years by direct, FIR techniques. The frequencies of 16 O were first refined by laser magnetic resonance (LMR), by Davies et al. [5] for the $J=0\leftarrow1$ transition, and by Saykally and Evenson [6] for the $J=1\leftarrow2$ transition. These values have been improved still further by Zink et al. [7], who used a tunable FIR source and so were able to dispense with the magnetic field.

In this paper, we report the observation of the LMR spectrum associated with the $J=1\leftarrow2$ transition for the ¹⁷O and ¹⁸O isotopes. The ¹⁷O spectrum was recorded using a sample in natural abundance (0.038%). By using the g_J factors measured in the EPR spectrum of ¹⁶O by Radford and Hughes [8], and the ¹⁷O hyperfine parameters measured by Harvey [9], we have determined much-improved values for the $J=1\leftarrow2$ fine-structure intervals for these isotopes. In the course of these experiments, we have been able to make observations of saturation dips on the LMR signals for ¹⁶O.

II. EXPERIMENTAL DETAILS

The FIR LMR spectrometer has been described in detail elsewhere [10]. Briefly, it consists of a transversely pumped gain cell which is 56 cm long and is separated from the sample region by a 12.5- μ m-thick polypropylene beam splitter set at Brewster's angle. The magnetic field is provided by an electromagnet with an 8-cm gap. The FIR power is coupled out with a movable 45° mirror and detected by a liquid-helium-cooled photoconductive detector. The signal is Zeeman-modulated at 40 kHz and detected with a lock-in amplifier tuned to the same frequency. The magnetic field is controlled by a rotating coil magnetometer, located on the axis of the field and close to the absorption region. The small correction for the difference between the field at the sample and that at the probe position is measured with a proton NMR fluxmeter.

The present measurements were made with the 63.1- μ m line of ¹³CH₃OH pumped by the 9P(12) line of a ¹²C¹⁶O₂ laser. Because our measurement of the fine-structure interval depends directly on the knowledge of the laser frequency as used in the experiment, we have carefully remeasured the frequency as 4751 341.8±0.5 MHz. The oxygen atoms were generated by flowing molecular oxygen at 160 Pa (1.2 Torr) through a microwave discharge.

III. RESULTS AND DISCUSSION

The LMR spectrum of ¹⁶O recorded with the 63.1- μ m laser consists of a single line at about 0.313 T [6]. This line is composed of three unresolved Zeeman components whose separation is less than the Doppler linewidth [14.7-MHz full width at half maximum (FWHM) at 300 K]. By reducing the total pressure to 13 Pa (100 mTorr), lowering the modulation amplitude, and maximizing the intracavity laser power, we have been able to resolve the two stronger Zeeman components as Lamb dips (Fig. 1).



FIG. 1. Lamb dips recorded in the laser magnetic-resonance spectrum of atomic oxygen at 63.1 μ m. The total sample pressure was 13 Pa (100 mTorr). The transition involved is the magnetic dipole transition between the ${}^{3}P_{1}$ and ${}^{3}P_{2}$ spin components. Three Zeeman components are expected: $M_{J} = -1 \leftarrow -2$, $0 \leftarrow -1$, and $1 \leftarrow 0$ with relative intensities 6:3:1. The weakest of the three falls at almost the same field as the strongest, and so it is not seen in this spectrum.



FIG. 2. The FIR LMR spectrum of ¹⁷O recorded in natural abundance (0.038%) with the 63.1- μ m line of ¹³CH₃OH pumped by the 9P(12) line of the CO₂ laser. The individual ¹⁷O lines are over four orders of magnitude weaker than the corresponding ¹⁶O transition, which is shown in the inset recorded under the same conditions.

The recording of the saturation dips for an atomic magnetic dipole transition in the FIR has not previously been reported.

The ¹⁷O and ¹⁸O spectra were both recorded with Doppler limited linewidths at a total O₂ pressure of 160 Pa (1.2 Torr). The ¹⁷O spectrum, recorded in natural abundance, is shown in Fig. 2. The nuclear hyperfine splitting for ¹⁷O $(I = \frac{5}{2})$ is the greatest in the J = 2 level and is large enough to shift the individual hyperfine lines well outside the main ¹⁶O line shape. Since the line intensity for ¹⁷O is diluted among (2I+1)=6 hyperfine components, the ¹⁶O resonance is 1.61×10^4 times stronger than ¹⁷O and corresponds to a signal-to-noise ratio of 4.8×10^5 with a time constant of 300 ms. Although ¹⁸O is more abundant (0.204%) than ¹⁷O, its LMR spectrum shows no hyperfine structure; the isotope shift from ¹⁶O is



FIG. 3. The LMR spectrum associated with the ${}^{3}P_{1}$ - ${}^{3}P_{2}$ fine-structure transition of atomic oxygen, using a sample enriched in 18 O (approximately 50%). The separation between the two line centers is measured as 1.12 mT after correcting for the overlap of the two line shapes.

TABLE I. Observed resonances associated with the ${}^{3}P_{1}{}^{-3}P_{2}$ fine-structure transition of 17 O recorded with the 63.1- μ m line of 13 CH₃OH. Laser frequency is 4751 341.8±0.5 MHz.

| J | Transition M_J | $M_I^{\rm a}$ | Flux density (mT) | Expt. – Calc. (MHz) |
|-----|------------------|-------------------------|----------------------|------------------------|
| 1←2 | -1←-2 | $-\frac{5}{2}$ | 259.12 | -1.1 |
| | | $-\frac{2}{3}{2}$ | 278.37 | 0.5 |
| | | $-\frac{1}{2}$ | 298.48 | 0.4 |
| | | $\frac{\frac{2}{3}}{2}$ | 340.69 | -0.3 |
| | | $\frac{\frac{2}{5}}{2}$ | 362.56 | 0.3 |
| | 0←-1 | $-\frac{5}{2}$ | 288.04 | 0.3 |
| | | $-\frac{2}{3}{2}$ | 296.76 | 0.4 |
| | | $\frac{3}{2}$ | 325.27 | 0.7 |
| | | $\frac{\frac{2}{5}}{2}$ | 335.72 | -1.2 |

^aThe observed transitions all obey the selection rule $\Delta M_I = 0$.

not large enough to place the signal outside the dominant ${}^{16}O$ line shape. We therefore recorded its spectrum with an enriched sample of ${}^{18}O$ (50%) as shown in Fig. 3. The shift from the ${}^{16}O$ resonance is measured to be 1.12 ± 0.02 mT to lower field, after correcting for the partial overlap of the two line shapes.

The measurements of the resonances for ¹⁷O are given in Table I. The positions of these lines depend on (i) the zero-field separation of the J=2 and 1 fine-structure levels, (ii) the magnetic and electric nuclear hyperfine interactions, and (iii) the g factors for the O atom in the ${}^{3}P_{2}$ and ${}^{3}P_{1}$ levels.

The second and third of these quantities have been measured much more accurately by gas-phase EPR. The hyperfine parameters were measured by Harvey [9] and

TABLE II. Parameters for ¹⁷O (in MHz).

| 2 060 069.09ª | | | | |
|-------------------------------|--|--|--|--|
| 4 744 790.52(26) ^b | | | | |
| 1.500 986° | | | | |
| 1.500 921° | | | | |
| 4.738 ^d | | | | |
| -218.569 ^d | | | | |
| -91.7 ^d | | | | |
| -126.6 ^d | | | | |
| -0.5219 ^{d,e} | | | | |
| | $\begin{array}{c} 2060069.09^{a}\\ 4744790.52(26)^{b}\\ 1.500986^{c}\\ 1.500921^{c}\\ 4.738^{d}\\ -218.569^{d}\\ -91.7^{d}\\ -126.6^{d}\\ -0.5219^{d,e}\\ \end{array}$ | | | |

^aValue taken from ¹⁶O (Zink et al.) [7].

^bValue determined from fit of the present data set. The number in parentheses is one standard deviation of the least-squares fit. ^cg factors taken from the EPR measurements of Radford and Hughes [8].

^dHyperfine parameters determined in the EPR experiments of Harvey [9].

^eThe electric-quadrupole parameter B in the model of Cooksy, Hovde, and Saykally [11] is related to the parameter b_J used by Harvey [9] by

$$b_J = B[3X(X+1) - 4J(J+1)L(L+1)]$$

$$\times I(2I+1)/[(J+1)(2J+3)]$$
,

where X = S(S+1) - L(L+1) - J(J+1). Thus for ¹⁷O, $b_1 = -10B$ and $b_2 = 20B$.

the g factors by Radford and Hughes [8]. We have therefore fitted our data using these parameter values to determine the fine-structure splitting for ¹⁷O. The details of the standard Russell-Saunders model used are given in a paper by Cooksy, Hovde, and Saykally [11]. The results of our fit are given in Tables I (residuals) and II (parameters for ¹⁷O); each data point was given unit weight. The ${}^{3}P_{1}-{}^{3}P_{2}$ fine-structure interval is determined to be 4744790.52 ± 0.26 MHz, where the estimated error is one standard deviation of the fit. The corresponding quantity for ¹⁸O is determined from the value for ¹⁶O, 4744777.49±0.16 MHz, measured by Zink et al. [7]. The oxygen atom transition concerned tunes at a rate of 21.008 MHz/mT, so the fine-structure interval for 18 O is determined to be 4744801.0 ± 0.5 MHz. The isotopic shifts in the fine-structure interval are therefore +13.0MHz for 17 O and +23.5 MHz for 18 O.

The identical frequencies were measured by De Natale et al., using tunable FIR methods on isotopically enriched samples [12]. The results of their measurements, which are more accurate than ours since they are made in the absence of a magnetic field, are

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 $v_{J=1\leftarrow 2} = \begin{cases} 4\ 744\ 790.76\pm 0.25\ \text{MHz} & \text{for}^{-17}\text{O} \\ 4\ 744\ 801.37\pm 0.19\ \text{MHz} & \text{for}^{-18}\text{O} \end{cases}.$

The agreement with our measurements is very good.

These are some of the most accurate measurements of the isotopic dependence of atomic-fine-structure intervals that have been made. The shift of ¹⁸O is almost exactly, but not quite, twice that of ¹⁷O, as might be expected. Veseth [13] has calculated the isotopic shifts in the finestructure interval for several light atoms by use of manybody perturbation theory. The value calculated for the ¹⁸O-¹⁶O shift in the ³P₁-³P₂ intervals is +27.5 MHz, in good agreement with the experimental value. The agreement is even more impressive when it is realized that the calculated value is the sum of three larger, counteracting contributions.

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