

Rydberg-atom collisions with SF₆ and CCl₄ at very high n

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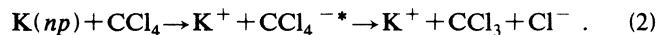
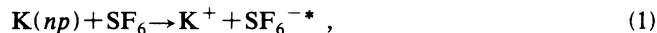
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Rate constants for Rydberg-atom destruction through electron transfer in collisions between $K(np)$ Rydberg atoms and SF₆ and CCl₄ have been measured for values of principal quantum number n up to $n \sim 400$ using an apparatus specifically developed for very-high- n collision studies. Analysis of the data using the free-electron model provides cross sections for electron attachment to SF₆ and CCl₄ at electron energies down to $\sim 10 \mu\text{eV}$, which are far below those accessible using any alternate technique. The measured cross sections are inversely proportional to electron velocity and are consistent with the Wigner threshold law for s -wave attachment.

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In recent years studies of Rydberg-atom collisions have been used to obtain information on electron attachment to electronegative molecules at subthermal electron energies [1–6]. Rydberg-atom collisions are frequently discussed using the free-electron model, in which it is assumed that the separation between the Rydberg electron and its associated ionic core is so large that both do not interact simultaneously with a target molecule [7]. At sufficiently large values of principle quantum number n , reactions involving attaching targets are dominated by the binary Rydberg electron-target interaction. Capture of the excited electron by the target molecule leads to Rydberg-atom destruction and the formation of a positive-negative ion pair. Previous studies, however, have been limited to Rydberg atoms with values of $n \lesssim 100$. In the present work we have extended measurements to $n \sim 400$ using an apparatus specifically developed for very-high- n collision studies and have investigated Rydberg-atom destruction in collisions between $K(np)$ Rydberg atoms and SF₆ and CCl₄ via the electron transfer reactions



Analysis of the data provides cross sections for electron attachment to SF₆ and CCl₄ at electron energies down to $\sim 10 \mu\text{eV}$, which are far below those accessible using any alternate technique, and demonstrates that, at threshold, electron capture is an s -wave process.

Collision studies at very high n have not been undertaken previously because of the significant experimental obstacles involved. These stem from the extreme sensitivity of very-high- n atoms to even very small external fields and the small oscillator strengths associated with their excitation. We have, however, overcome these difficulties using the apparatus shown in Fig. 1.

Potassium atoms contained in a tightly collimated thermal-energy ($\sim 300^\circ\text{C}$) beam are photoexcited to a selected np state by a crossed, weakly focused laser beam. Excitation occurs at the center of an interaction region

defined by three pairs of planar copper electrodes, each $\sim 10 \times 10 \text{ cm}^2$. The use of large electrodes well separated from the excitation volume minimizes the effects of stray patch fields due to nonuniformities in the electrode surfaces. Any small residual electric fields that remain are cancelled by application of small bias potentials to the various electrodes. Magnetic fields are eliminated by use of a μ -metal shield that surrounds the interaction region.

Excitation is accomplished using a frequency-doubled Coherent CR699-21 Rh6G dye laser. The output of the laser is formed into a series of pulses of $\sim 2 \mu\text{sec}$ duration with a pulse repetition frequency of $\sim 5\text{--}10 \text{ kHz}$ using an acousto-optic modulator. To achieve stable excitation and reproducible laser scanning, the long-term drift in the laser output frequency is controlled using a technique that employs a scanning Fabry-Pérot étalon and stabilized He-Ne laser [8]. Rydberg-atom production is monitored by field ionization [9,10]. A voltage pulse is applied to the lower electrode and the resulting ions or electrons are detected using a particle multiplier.

The Rydberg-atom signal observed as the laser is scanned over a frequency interval corresponding to excitation of Rydberg states with $n \sim 410$ is shown in Fig. 2. The well-resolved excitation spectrum demonstrates that

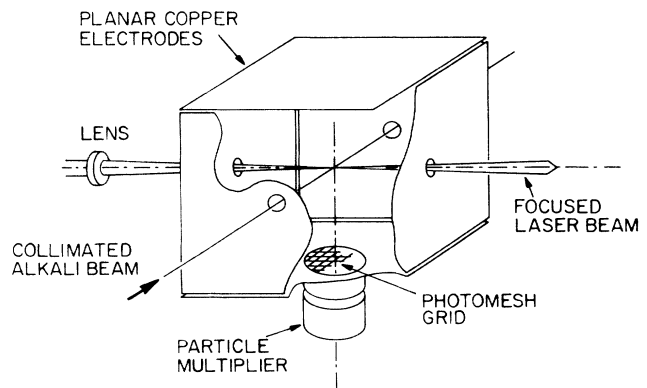


FIG. 1. Schematic diagram of the apparatus.

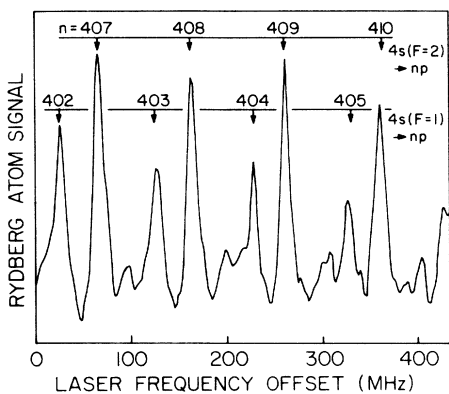


FIG. 2. Rydberg-atom production observed in the vicinity of $n \sim 410$.

stray residual fields in the excitation volume are very small. The effect of external electric fields is illustrated in Fig. 3, which shows, for several values of applied external field, the Rydberg-atom production observed as the laser is scanned over a frequency interval corresponding to excitation of states with $n \sim 270$. It is apparent that the excitation efficiency is dramatically reduced by the presence of even a very small external field. The $m_l = 0$ states are, however, more sensitive to the presence of the field.

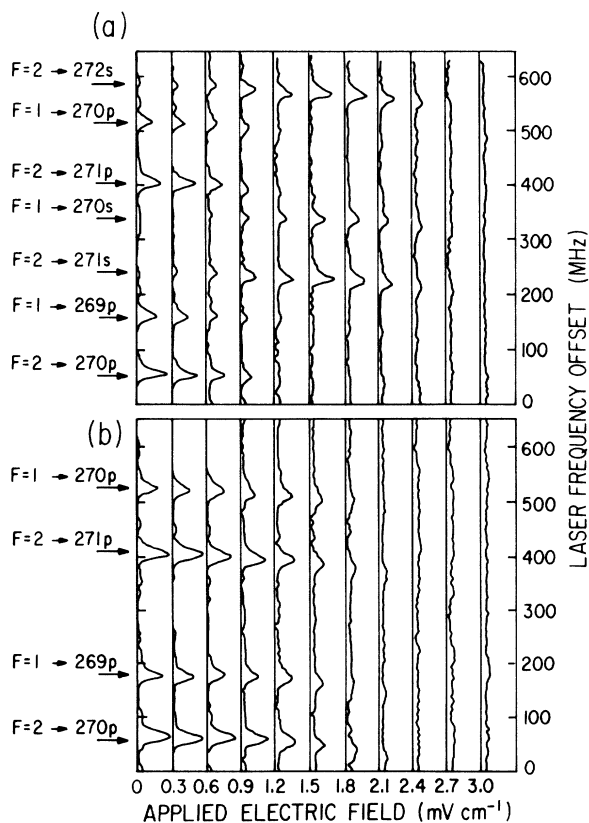


FIG. 3. The effect of applied external electric fields on the excitation of (a) $m_l = 0$ and (b) $|m_l| = 1$ states in the vicinity of $n \sim 270$.

Indeed, at “intermediate” external fields, excitation of s -like states made accessible through Stark mixing becomes dominant, as has been noted previously in studies at lower n [11,12]. The external fields that reduce the np excitation efficiency by one-half were measured as a function of n and are shown in Fig. 4. For reference, the solid line indicates the electric field at which states from adjacent hydrogenic Stark manifolds first cross. It is apparent that at $n = 400$ external fields of only $\sim 100 \mu\text{V cm}^{-1}$ are sufficient to cause a significant reduction in excitation efficiency. This, coupled with the narrow width of the various excitation peaks present in Fig. 2, indicates that stray electric fields in the excitation volume are reduced to $\lesssim 50 \mu\text{V cm}^{-1}$ using the present apparatus.

For collision experiments, excitation occurs (in zero electric field) in the presence of a target gas. Target-gas densities $\lesssim 10^{11} \text{ cm}^{-3}$ are employed to ensure single-collision conditions. Rate constants k_d for Rydberg-atom destruction in collisions are determined by measuring the time evolution of the Rydberg-atom population $N(t)$ in the interaction region. $N(t)$ is given to a good approximation by

$$N(t) = N(0)e^{-t/\tau}, \quad (3)$$

where

$$\frac{1}{\tau} = \rho k_d + \frac{1}{\tau_{\text{eff}}} \quad (4)$$

$N(0)$ is the number of Rydberg atoms present at $t = 0$ (i.e., at the end of the laser excitation pulse), τ_{eff} is the effective Rydberg-atom lifetime in the absence of the target gas, and ρ is the target-gas density (measured by an ionization gauge calibrated against a capacitance manometer). In practice, the probability that a Rydberg atom is created during any laser pulse is small ($\lesssim 0.01$) and the time evolution of the population is obtained by applying a field ionizing pulse at a variable delay after each laser

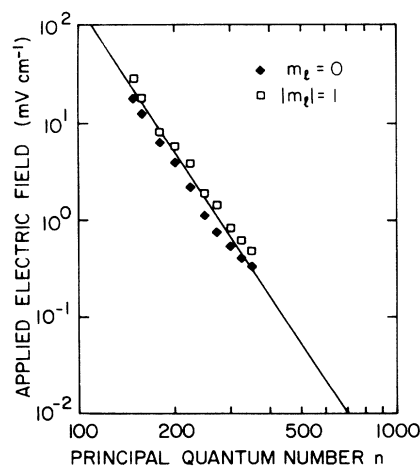


FIG. 4. Applied external electric field required to reduce the np excitation efficiency by one-half. The solid line shows, for each value of n , the electric field at which states from adjacent hydrogenic Stark manifolds first cross.

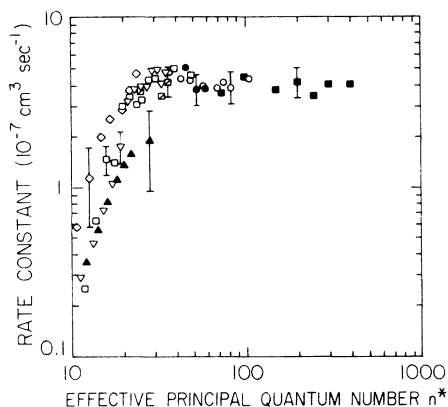


FIG. 5. Rate constants for Rydberg-atom destruction (k_d) and for free-ion production (k_i) in Rydberg atom collisions with SF_6 . ■, k_d - $\text{K}(np)$ (present data); ○, k_d - $\text{Rb}(nd)$ (Ref. [2]); ●, k_i - $\text{Rb}(ns)$ (Ref. [2]); □, k_i - $\text{Xe}(nf)$ (Ref. [1]); □, k_i - $\text{K}(nd)$ (Ref. [1]); ▽, k_i - $\text{Na}(np)$ (Ref. [16]); ◇, k_i - $\text{Ne}(ns)$ (Ref. [15]); ▲, k_i - $\text{Ne}(nd)$ (Ref. [15]).

pulse and accumulating data following many laser pulses. Values of $1/\tau$ are measured at several different target-gas densities ρ , and k_d is obtained from a fit to the data using Eq. (4).

The measured rate constants k_d for Rydberg-atom destruction in collisions with SF_6 and CCl_4 are shown in Figs. 5 and 6, respectively, as a function of effective principal quantum number n^* , where $n^* \equiv n - \delta$ and δ is the quantum defect. It has been demonstrated previously that for nonpolar attaching targets such as SF_6 and CCl_4 , Rydberg-atom destruction results from Rydberg-electron capture and that, at high n , the rate constants k_d for Rydberg-atom destruction can be equated to the rate constants k_i for the production [1,2] of unbound K^+ - SF_6^- and K^+ - Cl^- ion pairs via reactions (1) and (2), re-

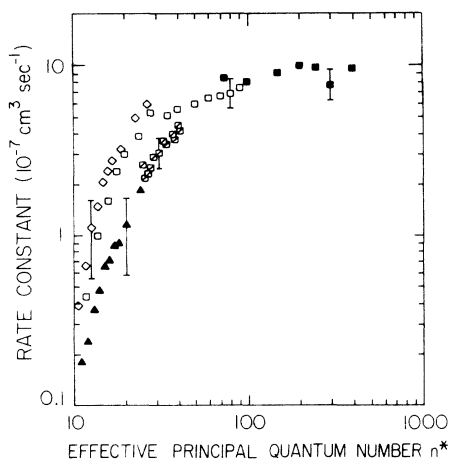
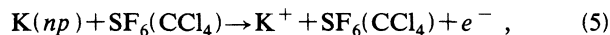


FIG. 6. Rate constants for Rydberg-atom destruction (k_d) and for free-ion production (k_i) in Rydberg atom collisions with CCl_4 . ■, k_d - $\text{K}(np)$ (present data); □, k_i - $\text{K}(nd)$ (Ref. [1]); □, k_i - $\text{Xe}(nf)$ (Ref. [1]); ◇, k_i - $\text{Ne}(ns)$ (Ref. [15]); ▲, k_i - $\text{Ne}(nd)$ (Ref. [15]).

spectively. This implies that the rate constants for collisional destruction through direct collisional ionization, i.e., through the reactions



are negligible in comparison to those for destruction through reactions (1) and (2). To check this, measurements of Rydberg-atom destruction were undertaken at $n \sim 250$ using CO_2 as the target gas. (Electron- CO_2 scattering is characterized by an unusually large momentum-transfer cross section at low energies [13].) The data showed that rate constants for Rydberg-atom destruction through direct collisional ionization are small and provided an upper limit for CO_2 (at $n \sim 250$) of $10^{-8} \text{ cm}^3 \text{ sec}^{-1}$. Theory, however, suggests that rate constant for direct collisional ionization are typically substantially smaller than even this value [14]. Figures 5 and 6 include rate constants for collisional destruction and for the production of free ions measured by earlier workers [1,2,15,16]. Good agreement between the present data and the earlier measurements is observed. At low-to-intermediate values of n the rate constants for the production of free ions decrease due to the increasing electrostatic attraction between the product positive and negative ions and the increasing opacity of the Rydberg-electron cloud. Both of these effects are discussed in detail elsewhere [17–20]. At high n , however, these effects are negligible and the free-electron model asserts that the rate constant for collisional destruction (and free-ion production) via reactions (1) and (2) should equal that for capture, by the target, of free electrons having the same velocity distribution as the Rydberg electron, i.e.,

$$k_d = \int_0^\infty v \sigma_e(v) f(v) dv, \quad (6)$$

where $f(v)$ is the Rydberg-electron velocity distribution (determined by its quantum state) and $\sigma_e(v)$ is the cross section for the capture of free electrons of velocity v . (Even at $n = 400$ the root-mean-square orbital velocity of the excited electron, $\sim 6 \times 10^5 \text{ cm sec}^{-1}$, is much greater than the relative heavy-particle collision velocity, $\sim 6 \times 10^4 \text{ cm sec}^{-1}$.) It is apparent from Figs. 5 and 6 that for both SF_6 and CCl_4 , the rate constants measured at very high n are essentially independent of n , i.e., independent of the Rydberg-electron velocity distribution. Inspection of Eq. (6) shows that this independence requires that the attachment cross section be inversely proportional to electron velocity, i.e., inversely proportional to the square root of the electron energy ϵ . This behavior, $\sigma_e(\epsilon) \propto \epsilon^{-1/2}$, is consistent with the Wigner threshold law for s -wave attachment [21,22]. Analysis of the data shows that, if v is expressed in cm sec^{-1} , $\sigma_e(v)$ is given by $(4.0 \pm 1.0) \times 10^{-7}/v$ and $(8.5 \pm 2.0) \times 10^{-7}/v \text{ cm}^2$ for SF_6 and CCl_4 , respectively, for electron energies $\leq 2 \text{ meV}$. To facilitate comparison with data for attachment of free electrons, velocity-averaged cross sections $\bar{\sigma}_e$ are derived from the Rydberg-atom data by use of the relation $\bar{\sigma}_e = k_{d,i}/v_m$, where v_m is the median velocity of the electrons attached, i.e., the Rydberg-electron velocity, such that integration of Eq. (6) from 0 to v_m yields a value one-half that for integration from 0 to ∞ .

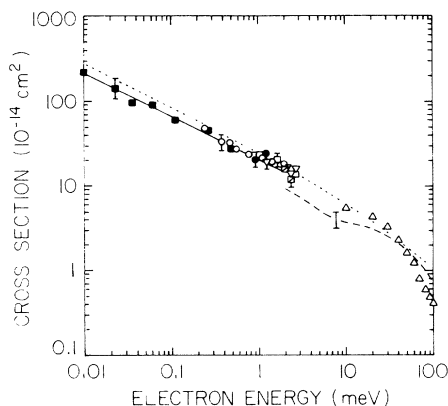


FIG. 7. Cross section for electron attachment to SF₆. ■, $\bar{\sigma}_e$ -K(*np*) present data; □, $\bar{\sigma}_e$ -K(*nd*) (Ref. [1]); ●, $\bar{\sigma}_e$ -Rb(*ns*) (Ref. [2]); ○, $\bar{\sigma}_e$ -Rb(*nd*) (Ref. [2]); △, $\bar{\sigma}_e$ -Xe(*nf*) (Ref. [1]); △, swarm-unfold data (Ref. [23]); - - -, TPSA results (Ref. [22]); ···, Klots (Ref. [26]); —, $\sigma_e(v)$.

The velocity-averaged cross sections $\bar{\sigma}_e$ are shown in Figs. 7 and 8 together with the corresponding cross sections $\sigma_e(v)$. The values of $\bar{\sigma}_e$ are positioned on the electron-energy axis according to the value of $mv_m^2/2$. Figures 7 and 8 also include data for free-electron attachment obtained using both swarm-unfold [23,24] and threshold photoelectron spectroscopy (TPSA) techniques [22,25], together with cross sections for electron capture derived using the theoretical expression for *s*-wave capture given by Klots [26], and polarizabilities of 6.52 and 10.6 Å³ for SF₆ and CCl₄, respectively [27,28]. The agreement among the various data is good.

The present work shows that Rydberg-atom techniques can be used to measure cross sections for electron attachment and examine threshold behavior at ultralow elec-

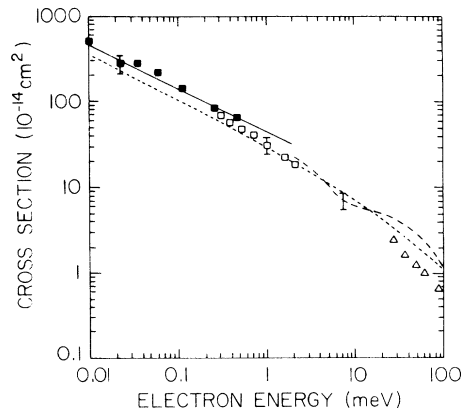


FIG. 8. Cross section for electron attachment to CCl₄. ■, $\bar{\sigma}_e$ -K(*np*) present data; □, $\bar{\sigma}_e$ -K(*nd*) (Ref. [1]); △, swarm-unfold data (Ref. [24]); - - -, TPSA results (Refs. [22,25]); ···, Klots (Ref. [26]); —, $\sigma_e(v)$.

tron energies, $\lesssim 10 \mu\text{eV}$, which are inaccessible using any alternate approach. The data confirm unequivocally that low-energy electron capture by SF₆ and CCl₄ is an *s*-wave process at threshold and demonstrate that Rydberg-atom collisions can be studied at very high *n*. Indeed, since application of the free-electron collision model is most easily justified at very large values of *n*, the study of Rydberg-atom collisions at very high *n* promises to provide information on a wide variety of electron-molecule interactions at ultralow electron energies.

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