Eigenenergies and oscillator strengths for the Hulthén potential

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A detailed study of the bound-state properties of the Hulthén potential is presented. Accurate eigenenergies are obtained for the Hulthén potential by numerical integration of the Schrödinger equation. One-parameter variational calculations are carried out. The variational results are practically identical to the exact energies, except in the high-screening region. The critical screening parameter is calculated for various values of l for $n \leq 10$ by a numerical solution of the wave equation. The energy eigenvalues obtained by a variety of methods are compared and discussed. The variational wave functions are employed to calculate absorption oscillator strengths for $1s \rightarrow 2p$, $1s \rightarrow 3p$, and $2p \rightarrow 3d$ transitions.

I. INTRODUCTION

The Hulthén potential¹ is one of the important shortrange potentials in physics. The potential (in atomic units) is given by

$$V(r) = -Z\delta e^{-\delta r} / (1 - e^{-\delta r}) .$$
⁽¹⁾

where Z is a constant and δ is the screening parameter. If the potential is used for atoms, the Z is identified with the atomic number. This potential is a special case of the Eckart potential.²

The Hulthén potential has been used in nuclear and particle physics,³ atomic physics,⁴⁻⁶ solid-state physics,^{7,8} and chemical physics,⁹ and its bound-state and scattering properties have been investigated by a variety of techniques.¹⁰⁻³¹ Hulthén wave functions have been used in solid-state physics problems.^{7,32} Hulthén-like wave functions have found use in investigations on atomic problems.^{33,34} The Eckart potential has been investigated,³⁵ and other extended versions of the Hulthén potential have also been proposed.^{36,37}

A great advantage with the Hulthén potential is the fact that the Schrödinger equation for this potential can be solved in a closed form for s waves.^{1,5,14} For $l \neq 0$, a number of methods have been employed to evaluate bound-state energies.²⁴⁻³¹ However, no "exact" values obtained from a numerical integration of the Schrödinger equation have been available to assess the accuracy of the various methods. We present such values in Sec. II of this paper and compare them with those obtained by several other methods. For a number of purposes it is convenient to have the wave function in an analytical form. Greene and Aldrich³³ have given a method for generating pseudo-Hulthén wave functions for $l \neq 0$ states. In Sec. III we apply the variational method, using these wave functions to obtain energies for $l \neq 0$ states. Using only a single parameter in the wave functions, we are able to obtain energies that are almost equal to the exact energies, except in the high-screening region. We also suggest a modification for these wave functions and present the results for the 2p level obtained from the

modified wave function. An important quantity of interest for the Hulthén potential (and for other similar screened potentials) is the critical screening parameter δ_c , which is that value of δ for which the binding energy of the level in question becomes zero. In Sec. IV we present values of δ_c for $n \leq 10$ obtained from a numerical solution of the Schrödinger equation. The eigenenergies obtained by various methods are compared in Sec. V. Using the analytical wave functions of Sec. II, we obtain expressions for the absorption oscillator strength for three transitions in Sec. VI and numerical results are presented for a few values of δ . Throughout this paper we shall take Z = 1 and shall use atomic units in which $\hbar = m = e = 1$.

II. NUMERICALLY DETERMINED EIGENENERGIES

Eigenenergies for the Hulthén potential were determined for certain values of δ for $n \leq 6$ and $l \neq 0$ by numerical integration of the Schrödinger equation using Numerov's method and a logarithmic mesh. The results are presented in column 3 of Table I, along with values obtained by several other methods.

The normalized probability density distribution $[r^2R^2(r)]$ for the 3*p* state for $\delta = 0.025$ (low screening) and 0.15 (high screening) are shown in Fig. 1. It will be noticed that as δ increases, the density distribution tends to spread out to larger values of *r*, and the peak values are reduced. Analogous results were found for the static screened Coulomb potential by Rogers *et al.*³⁸

It will be noticed from Table I that as δ increases the splitting between different *l* levels (for a given *n*) increases. For low values of the principal quantum number, the ordering of the levels is the same as in the case of the Coulomb potential, but as one goes to higher values of *n*, it is found that there is a considerable amount of level crossing in the vicinity of E = 0. This effect is shown for n = 7 and 8 levels in Fig. 2. At high values of *n*, the level ordering is quite complex. We shall return to this point when we consider δ_c in Sec. IV.

Patil²⁶ has analytically investigated the behavior of the energy levels for the Hulthén potential near the critical

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screening parameter and found that for $l \ge 2$,

$$E = \operatorname{const} \times (1/\delta - 1/\delta_c) + O[(1/\delta - 1/\delta_c)]^2$$
 (2)

in the vicinity of δ_c . It was of interest to examine the range of δ in which the above relation is applicable. Equation (2) indicates that a plot of $E/(1/\delta - 1/\delta_c)$ versus $(1/\delta - 1/\delta_c)$ should be linear over the range of applicability. We made such plots for n = 7 and l = 2, 3, 4, 5, and 6. It was found that the linearity holds only over a very small interval of δ . We estimate that this range is from δ_c to approximately $0.99\delta_c$.

For s states, from the solution of the Schrödinger equation one can obtain

$$\frac{E}{E_0} = \left[1 - \frac{\delta}{\delta_c}\right]^2,\tag{3}$$

where E_0 is the energy for $\delta = 0$. Thus there is a univer-

TABLE I. Eigenenergies as obtained by various methods. A negative sign before the energy has been omitted everywhere.

State	δ	Numerical integration	Variational [Eq. (7)]	Variational [Eq. (9)]	Lai and Lin (Ref. 24)	Patil (Ref. 26)	Tang and Chan (Ref. 29)	Matthys and De Meyer (Ref. 31)
2 <i>p</i>	0.025	0.112 760 5	0.112 760 5	0.112 760 5		0.112 76		0.112 760 4
1	0.050	0.101 042 5	0.101 042 5	0.101 042 5	0.101 043	0.101 04	0.101 042 4	0.101 042 5
	0.075	0.089 847 8	0.089 847 8	0.089 847 8		0.089 85		
	0.100	0.079 179 4	0.079 179 4	0.079 179 4	0.079 179	0.079 18	0.079 179 4	0.079 179 4
	0.150	0.059 441 5	0.059 441 5	0.059 441 5		0.059 445		0.059 441 5
	0.200	0.041 886 0	0.041 886 0	0.041 885 9	0.041 886	0.041 895	0.041 885 7	0.041 886 0
	0.250	0.026 611 1	0.026 610 8	0.0266100				
	0.300	0.013 790 0	0.013 787 8	0.013 783 8	0.013 790			0.013 790 0
	0.350	0.003 793 1	0.003 773 4	0.003 753 4	0.003 779	0.038 375		
3 <i>0</i>	0.025	0.043 706 9	0.043 706 9		0.043 707	0.043 708 5		0.043 707 1
r	0.050	0.033 164 5	0.033 164 5		0.033 165	0.033 185	0.033 165 18	0.033 165 0
	0.075	0.023 939 7	0.023 939 7			0.024 016 5		
	0.100	0.016 053 7	0.016 053 7		0.016 054	0.01622	0.016 067 72	0.016 053 7
	0.150	0.004 466 3	0. 004 466 0		0.004 466	0.046 995		0.004 466 4
3.1	0.025	0.043.603.0	0.043.603.0		0.043.603	0 043 602 5		0.043.603.0
54	0.025	0.032 753 2	0.032 753 2		0.032 753	0.032745	0 032 753 2	0.032 753 2
	0.050	0.023.030.7	0.023.030.7		0.052755	0.022.99	0.0527552	0.052 755 2
	0.100	0.014 484 2	0.014 484 2		0 014 484	0.014 39	0.014 484 2	0.014 484 2
	0.150	0.001 396 6	0.001 389 4		0.001 391	0.001 375 5		0.001 396 5
	0.005	0.010.040.0	0.010.010.0		0.010.040	0.010.05		0.010.040.0
4 <i>p</i>	0.025	0.019 948 9	0.019 948 9		0.019 949	0.019 95	0.011.070.5	0.019 949 0
	0.050	0.0110582	0.011 058 2		0.011058	0.0110/5	0.0110725	0.011 058 3
	0.075	0.004 621 9	0.004 621 9		0.004 622	0.004 058 5		0.004 622 4
	0.100	0.000 / 33 0	0.000 733 2		0.000 / 34	0.000732		
4 <i>d</i>	0.025	0.019 846 2	0.019 846 2		0.019 846	0.019 845		0.019 846 2
	0.050	0.010 667 4	0.010 667 4		0.010 667	0.01068	0.010 669 0	0.010 667 4
	0.075	0.003 834 5	0.003 834 4		0.003 834	0.003 875		0.003 834 6
4 <i>f</i>	0.025	0.019 691 1	0.019 691 1		0.019 691	0.019 69		0.019 691 1
	0.050	0.010 062 0	0.010 062 0		0.010062	0.010 045	0.010 062 0	0.010 061 9
	0.075	0.002 556 3	0.002 555 7		0.002 556	0.002 557		0.002 556 3
50	0.025	0 009 403 6					0.009.408.7	
Sр	0.025	0.009 403 0					0.009 408 /	
	0.050	0.002 047 0						
5d	0.025	0.009 303 7					0.009 305 0	
	0.050	0.002 313 1						
5 <i>f</i>	0.025	0.009 152 1					0.009 152 3	
	0.050	0.001 783 5						
50	0.025	0 008 946 5					0.008.046.5	
Jg	0.025	0.008 940 9					0.008 940 5	
	0.050	0.0010139						
6р	0.025	0.004 154 8						
6d	0.025	0.004 060 6						
6 <i>f</i>	0.025	0.003 916 8						
_6g	0.025	0.003 720 1						



FIG. 1. $r^2 R^2(r)$ vs r for the 3p state for two values of the screening parameter. The solid-line curve is for $\delta = 0.025$ (low screening) and the dashed-line curve is for $\delta = 0.15$ (high screening).

sal relation between the reduced energy and the reduced screening parameter for s states. It was of interest to examine whether such a relationship is also valid for $l\neq 0$ states. In Fig. 3 we show E/E_0 versus δ/δ_c for p states for n=2 to 8, except n=6. It will be noted that while there is no universal curve for all n, the results for higher values of n rapidly approach an asymptotic curve.



FIG. 2. The energy eigenvalue as a function of δ for various values of *l* for n = 7 (solid line curves) and n = 8 (dashed-line curves) in the vicinity of E = 0.



FIG. 3. The ratio E/E_0 as a function of δ/δ_c for 2p, 3p, 4p, 5p, 7p, and 8p states. To avoid overlapping, 6p state has not been shown and 7p has been shown by a dashed-line curve.

III. VARIATIONAL CALCULATION

For $l \neq 0$ states, Greene and Aldrich³³ have developed a method to generate pseudo-Hulthén wave functions. They replace

$$V_{H}(r) = -\frac{\delta e^{-\delta r}}{1 - e^{-\delta r}} + \frac{l(l+1)}{2r^{2}}$$
(4)

by an effective potential

$$V_{H}^{\text{eff}}(\mathbf{r}) = -\frac{\delta e^{-\delta \mathbf{r}}}{1 - e^{-\delta \mathbf{r}}} + \frac{l(l+1)}{2} \left(\frac{\delta}{1 - e^{-\delta \mathbf{r}}}\right)^{2} e^{-\delta \mathbf{r}} .$$
 (5)

For small δr , Eq. (5) is a good approximation to Eq. (4), and it has the advantage that the Schrödinger equation for this potential is solvable analytically. The resulting wave functions can be written in the form

$$\chi(r) = e^{-a\delta r} \sum_{i=0}^{\infty} C_i (1 - e^{-\delta r})^{i+l+1} , \qquad (6)$$

where a and the C_i 's depend on the state. If δ is replaced by a variational parameter μ , wave functions of the form (6) can be expected to be a reasonable approximation for $l\neq 0$ states for the Hulthén potential. Greene and Aldrich³³ used such wave functions for $l\neq 0$ states of the static screened Coulomb potential. Here we shall employ them for the Hulthén potential itself.

The unnormalized wave functions for the various states considered here are as follows:

$\chi_{2p}(r) = e^{-a_2 \mu r} S^2$,	
$\chi_{3p}(r) = e^{-a_3\mu r} \left[S^2 - \right]$	$\left[\frac{1}{2}+\frac{1}{6\mu}\right]S^3\right],$
$\chi_{3d}(r) = e^{-a_3\mu r} S^3$,	

$$\chi_{4p}(r) = e^{-a_4\mu r} \left[S^2 - \left[1 + \frac{1}{4\mu} \right] S^3 \right],$$
(7)

$$+ \left[\frac{1}{10} + \frac{1}{20\mu}\right] \left[1 + \frac{1}{4\mu}\right] S^{4},$$

$$\chi_{4d}(r) = e^{-a_{4}\mu r} \left[S^{3} - \left[\frac{1}{2} + \frac{1}{12\mu}\right] S^{4}\right],$$

$$\chi_{4f}(\mathbf{r}) = e^{-a_4\mu r} S^4 \; .$$

Here

$$S = (1 - e^{-\mu r}) ,$$

$$a_2 = \frac{1}{2\mu} - 1 ,$$

$$a_3 = \frac{1}{3\mu} - \frac{3}{2} ,$$

and

$$a_4 = \frac{1}{4\mu} - 2 \quad .$$

The variational energy is given by

$$E = \frac{\int_{0}^{\infty} \chi \left[-\frac{1}{2} \frac{d^{2}}{dr^{2}} - \frac{\delta e^{-\delta r}}{1 - e^{-\delta r}} + \frac{l(l+1)}{2r^{2}} \right] \chi dr}{\int_{0}^{\infty} \chi^{2} dr}$$
(8)

The integration of the second term in the numerator of (8) is, however, analytically not tractable. Hence the second derivative was obtained analytically and the integration was carried out numerically. The energies thus obtained are shown in column 4 of Table I and the best values of μ in Table II. The wave functions for the 3p and 4p states with the given values of μ are not exactly orthogonal to the wave function of the corresponding lower states, but the deviation is quite small except when δ is close to δ_c .

The difficulty concerning the analytic integration of the second term in the numerator of Eq. (8) can be avoided if we slightly modify the assumed wave function. This is

achieved by multiplying the wave functions given in (7)
by the factor $(1-e^{-\delta r})/(1-e^{-\mu r})$. This factor is almost
equal to 1 and has only a small effect on the shape of the
wave function. To examine whether the modified wave
function gives satisfactory results, calculations were car-
ried out on the 2p state. Thus the unnormalized modified
χ is

$$\chi(r) = e^{-a_2\beta r} (1 - e^{-\beta r}) (1 - e^{-\delta r}) , \qquad (9)$$

where β is a variational parameter, similar to μ . The corresponding expression for the energy can be evaluated to be

$$E = [(1-2\beta)(1-2\beta+\delta)(1-2\beta+2\delta)(1-\beta)(1-\beta+\delta)(1-\beta+2\delta)(1+\delta)(1+2\delta)] \\ \times \{4\delta^{2}\beta^{2}[4\delta^{4}+(-18\beta+18)\delta^{3}+(-66\beta+28\beta^{2}+33)\delta^{2}+(-72\beta+66\beta^{2}+24-18\beta^{3})\delta \\ +33\beta^{3}-18\beta^{3}+6-24\beta+4\beta^{4}]\}^{-1}-(\frac{1}{2})\delta^{2}\beta^{2}[-4\delta^{4}+(20\beta-20)\delta^{3}+(22\beta-20\beta^{2}+3)\delta^{2} \\ +(24+8\beta^{2}+4\beta^{3}-36\beta)\delta+23\beta^{2}-28\beta-6\beta^{3}+10] \\ \times [(1-\beta)(1-2\beta+\delta)(1-\beta+\delta)(1-\beta+2\delta)(1+\delta)(1-2\beta+2\delta)(1+2\delta)]^{-1} \\ +(4\beta-2\delta-2)\ln(1-2\beta+\delta)+(2\beta-4\delta-2)\ln(1-\beta+2\delta)-(2\delta+2)\ln(1+\delta) \\ +(4\delta-4\beta+4)\ln(1-\beta+\delta)+(2\delta-2\beta+1)\ln(1-2\beta+2\delta)+(2\delta+1)\ln(1+2\delta) \\ +(1-2\beta)\ln(1-2\beta)+(2\beta-2)\ln(1-\beta).$$
(10)

TABLE II. Best values of the parameters μ and	β
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State	δ	μ	β
2 <i>p</i>	0.025	0.025 004	0.024 979
	0.050	0.050 040	0.050 001
	0.075	0.075 140	0.075 028
	0.10	0.100 337	0.100 092
	0.15	0.151 192	0.150 468
	0.20	0.203 037	0.201 518
	0.25	0.256 595	0.253 946
	0.30	0.313 364	0.309 228
	0.35	0.377 907	0.371 766
3 <i>p</i>	0.025	0.025 011	
	0.050	0.050 094	
	0.075	0.075 333	
	0.100	0.100 851	
	0.150	0.153 820	
3 <i>d</i>	0.025	0.025 030	
	0.050	0.050 251	
	0.075	0.075 901	
	0.100	0.102 364	
	0.150	0.162 638	
4 <i>p</i>	0.025	0.025 021	
	0.050	0.050 180	
	0.075	0.075 726	
	0.100	0.102 906	
4 <i>d</i>	0.025	0.025 057	
	0.050	0.050 438	
	0.075	0.077 150	
4 <i>f</i>	0.025	0.025 109	
	0.050	0.050 924	
	0.075	0.079 170	

The energies obtained by minimizing this expression with respect to β are shown in the fifth column of Table I, and the corresponding values of β in Table II.

TABLE III. Critical screening parameter for various states of the Hulthén potential.

102.000 0002.000 000200.500 0000.500 000210.376 9360.377 367300.222 2220.222 222310.186 4860.186 017320.157 6620.157 990400.125 0000.125 000410.110 4910.109 826420.097 5640.097 256430.086 4050.086 727500.080 0000.080 000510.072 8630.072 275520.066 1080.065 616530.059 9730.059 838540.054 5050.054 790600.055 5560.055 556610.051 5790.031 105620.047 6610.047 169630.043 9750.043 670640.040 5850.040 547650.037 5040.037 747700.040 8160.040 816710.038 3980.038 024720.035 9480.032 324740.029 2660.027 584800.031 2500.031 250810.029 6810.029 385820.028 0580.027 681850.024 45910.024 691910.023 6210.023 374860.022 1370.011 38090 </th <th>n</th> <th>l</th> <th>δ_c (Numerical)</th> <th>δ_c [Patil, Eq. (12)]</th>	n	l	δ_c (Numerical)	δ_c [Patil, Eq. (12)]
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	1	0	2.000 000	2.000 000
21 0.376936 0.377367 30 0.222222 0.222222 31 0.186486 0.186017 32 0.157662 0.157990 40 0.125000 0.125000 41 0.110491 0.109826 42 0.097564 0.097256 43 0.086405 0.086727 50 0.080000 0.080000 51 0.072863 0.072275 52 0.066108 0.065616 53 0.059973 0.059838 54 0.051579 0.051105 62 0.047661 0.047169 63 0.043975 0.043670 64 0.040855 0.040647 65 0.037504 0.037747 70 0.040816 0.040816 71 0.033581 0.033234 74 0.027379 0.027584 80 0.031250 0.31250 81 0.029286 0.027681 83 0.026459 0.026122 84 0.024926 0.024691 91 0.022497 0.022384 92 0.022497 0.022160 87 0.020621 0.031396 99 0.0123264 0.021038 99 0.0123264 0.021038 99 0.016	2	0	0.500 000	0.500 000
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	2	1	0.376 936	0.377 367
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	3	0	0.222 222	0.222 222
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	3	1	0.186 486	0.186 017
400.125 0000.125 000410.110 4910.109 826420.097 5640.097 256430.086 4050.086 727500.080 0000.080 000510.072 8630.072 275520.066 1080.065 616530.059 9730.059 838540.054 5050.054 790600.055 5560.055 556610.051 5790.051 105620.047 6610.047 169630.043 9750.043 670640.040 5850.040 547650.037 5040.037 747700.048 160.040 816710.038 3980.038 024720.032 9480.035 508730.033 5810.033 234740.031 3530.031 172750.029 2840.029 296810.029 6810.023 374800.031 2500.031 250810.022 46910.024 691910.023 6210.023 374860.022 1240.022 166870.020 2780.020 031950.019 3200.019 072930.013 2370.018 180970.012 36210.023 384900.024 6910.022 17993 </td <td>3</td> <td>2</td> <td>0.157 662</td> <td>0.157 990</td>	3	2	0.157 662	0.157 990
41 0.110491 0.109826 42 0.097564 0.097256 43 0.086405 0.086727 50 0.080000 0.080000 51 0.072863 0.072275 52 0.066108 0.0656166 53 0.059973 0.059838 54 0.054505 0.054790 60 0.055556 0.055556 61 0.051579 0.051105 62 0.047661 0.047169 63 0.043975 0.043670 64 0.040585 0.0405477 65 0.037504 0.037747 70 0.040816 0.040816 71 0.038988 0.038024 72 0.035948 0.033234 74 0.031353 0.031172 75 0.029284 0.029296 76 0.027379 0.027584 80 0.031250 0.31250 81 0.029681 0.023374 86 0.022124 0.022160 87 0.02046459 0.026122 84 0.024691 0.024691 91 0.023621 0.023384 90 0.021372 0.02103848 90 0.021372 0.02103848 90 0.022478 0.0200311950 93 0.021372 0.02103848 9	4	0	0.125 000	0.125 000
42 0.097564 0.097256 43 0.086405 0.086727 50 0.080000 0.080000 51 0.072863 0.072275 52 0.066108 0.065616 53 0.059973 0.059838 54 0.054505 0.054790 60 0.055556 0.055556 61 0.051579 0.051105 62 0.047661 0.047169 63 0.043975 0.043670 64 0.040585 0.040547 65 0.037504 0.037747 70 0.040816 0.040816 71 0.038988 0.038024 72 0.035948 0.035508 73 0.033581 0.031234 74 0.031250 0.031172 75 0.029284 0.029296 76 0.027379 0.027584 81 0.029681 0.023374 86 0.022124 0.022160 87 0.0206459 0.026122 84 0.024926 0.024691 91 0.023621 0.023374 93 0.021372 0.021064 91 0.022678 0.020001 95 0.019230 0.019072 96 0.018237 0.018186 97 0.016	4	1	0.110491	0.109 826
43 0.086405 0.086727 50 0.080000 0.080000 51 0.072863 0.072275 52 0.066108 0.065616 53 0.059973 0.059838 54 0.054505 0.054790 60 0.055556 0.055556 61 0.051579 0.051105 62 0.047661 0.047169 63 0.043975 0.043670 64 0.040585 0.0405477 65 0.037504 0.037747 70 0.040816 0.040816 71 0.038988 0.038024 72 0.035948 0.031233 74 0.031353 0.031172 75 0.029284 0.029296 76 0.027379 0.027584 80 0.031250 0.031250 81 0.0246459 0.026122 84 0.024926 0.024691 85 0.023478 0.023374 86 0.022179 0.022160 87 0.0224671 0.022160 91 0.0224971 0.022164 92 0.0224971 0.021038 90 0.0123621 0.023344 92 0.0224971 0.021064 94 0.020278 0.020031 95 0.019230 0.019072 96<	4	2	0.097 564	0.097 256
50 0.080000 0.080000 51 0.072863 0.072275 52 0.066108 0.065616 53 0.059973 0.059838 54 0.054505 0.054790 60 0.055556 0.055556 61 0.051579 0.051105 62 0.047661 0.047169 63 0.043975 0.043670 64 0.040855 0.040547 65 0.037504 0.037747 70 0.040816 0.040816 71 0.038988 0.038024 72 0.035948 0.035508 73 0.033581 0.033234 74 0.031250 0.031172 75 0.029284 0.029296 76 0.027379 0.027584 80 0.031250 0.031250 81 0.029681 0.023374 85 0.023478 0.023374 86 0.022124 0.022166 87 0.020664 0.021038 90 0.024691 0.024691 91 0.022621 0.021064 94 0.020278 0.020001 95 0.019230 0.019072 96 0.018237 0.018180 97 0.017330 0.017340 101 0.016	4	3	0.086 405	0.086 727
51 $0.072 863$ $0.072 275$ 52 $0.066 108$ $0.065 616$ 53 $0.059 973$ $0.059 838$ 54 $0.054 505$ $0.054 790$ 60 $0.055 556$ $0.055 556$ 61 $0.051 579$ $0.051 105$ 62 $0.047 661$ $0.047 169$ 63 $0.043 975$ $0.043 670$ 64 $0.040 885$ $0.040 547$ 65 $0.037 754$ $0.037 747$ 70 $0.040 816$ $0.040 816$ 71 $0.038 398$ $0.038 024$ 72 $0.035 948$ $0.033 508$ 73 $0.033 581$ $0.032 508$ 74 $0.031 353$ $0.031 172$ 75 $0.022 284$ $0.029 2966$ 76 $0.027 379$ $0.027 584$ 80 $0.031 250$ $0.031 250$ 81 $0.029 681$ $0.022 385$ 82 $0.026 681$ $0.022 374$ 86 $0.022 124$ $0.022 160$ 85 $0.023 478$ $0.023 374$ 86 $0.022 1372$ $0.021 064$ 91 $0.022 627$ $0.022 179$ 93 $0.021 372$ $0.021 064$ 97 $0.019 230$ $0.019 072$ 96 $0.018 237$ $0.018 180$ 97 $0.017 340$ $0.016 575$ <	5	0	0.080 000	0.080 000
52 $0.066\ 108$ $0.065\ 616$ 53 $0.059\ 973$ $0.059\ 838$ 54 $0.054\ 505$ $0.055\ 9838$ 60 $0.055\ 556$ $0.055\ 556$ 61 $0.051\ 579$ $0.051\ 105$ 62 $0.047\ 661$ $0.047\ 169$ 63 $0.043\ 975$ $0.043\ 670$ 64 $0.040\ 585$ $0.040\ 547$ 65 $0.037\ 504$ $0.037\ 747$ 70 $0.040\ 816$ $0.040\ 816$ 71 $0.038\ 998$ $0.038\ 024$ 72 $0.033\ 948$ $0.033\ 508$ 73 $0.033\ 581$ $0.033\ 234$ 74 $0.031\ 353$ $0.031\ 172$ 75 $0.029\ 284$ $0.029\ 296$ 76 $0.027\ 379$ $0.027\ 584$ 80 $0.031\ 250$ $0.031\ 250$ 81 $0.029\ 681$ $0.029\ 385$ 82 $0.026\ 681$ $0.029\ 385$ 82 $0.024\ 695$ $0.024\ 691$ 83 $0.024\ 691$ $0.024\ 691$ 91 $0.023\ 621$ $0.023\ 374$ 86 $0.021\ 278$ $0.020\ 031$ 97 $0.017\ 303$ $0.017\ 350$ 98 $0.016\ 227$ $0.021\ 075$ 98 $0.016\ 237$ $0.019\ 072$ 96 $0.018\ 237$ $0.018\ 180$ 97 $0.017\ 303$ $0.017\ 350$ 98 $0.016\ 323$ <td< td=""><td>5</td><td>1</td><td>0.072 863</td><td>0.072 275</td></td<>	5	1	0.072 863	0.072 275
53 0.059973 0.059838 54 0.054505 0.054790 60 0.055556 0.055556 61 0.051179 0.051105 62 0.047661 0.047169 63 0.043975 0.043670 64 0.040585 0.040547 65 0.037504 0.037747 70 0.040816 0.040816 71 0.038998 0.038024 72 0.035948 0.035508 73 0.033581 0.033234 74 0.029284 0.029296 76 0.027379 0.027584 80 0.031250 0.031250 81 0.029681 0.029385 82 0.028058 0.027681 83 0.026459 0.026122 84 0.024926 0.024691 87 0.020864 0.021038 90 0.024691 0.024691 91 0.023621 0.023384 92 0.022497 0.022179 93 0.021372 0.021064 94 0.020278 0.020031 95 0.019230 0.019072 96 0.018237 0.018180 97 0.017303 0.017350 98 0.016030 0.015851 100 0.020	5	2	0.066 108	0.065 616
54 0.054505 0.054790 60 0.055556 0.055556 61 0.051579 0.051105 62 0.047661 0.047169 63 0.043975 0.043670 64 0.040585 0.040547 65 0.037504 0.037747 70 0.040816 0.040816 71 0.038398 0.038024 72 0.035948 0.0335508 73 0.033581 0.033234 74 0.031353 0.011172 75 0.022284 0.022296 76 0.027379 0.027584 80 0.031250 0.031250 81 0.029681 0.029385 82 0.028058 0.027681 83 0.026459 0.026122 84 0.024926 0.024691 85 0.023478 0.023374 86 0.022124 0.022160 87 0.020664 0.021038 90 0.024691 0.024691 91 0.023621 0.023384 92 0.022497 0.022179 93 0.021372 0.010644 94 0.020070 0.020001 100 0.020000 0.020000 101 0.016427 0.016575 100 0.0	5	3	0.059 973	0.059 838
60 0.055556 0.055556 61 0.051579 0.051105 62 0.047661 0.047169 63 0.043975 0.043670 64 0.040585 0.040547 65 0.037504 0.037747 70 0.040816 0.040816 71 0.038398 0.038024 72 0.035948 0.035508 73 0.033581 0.033234 74 0.031353 0.031172 75 0.029284 0.029296 76 0.027379 0.027584 80 0.031250 0.031250 81 0.029681 0.029385 82 0.028658 0.027681 83 0.026459 0.026122 84 0.024926 0.024691 85 0.023478 0.023374 86 0.022124 0.022160 87 0.020864 0.021038 90 0.024691 0.024691 91 0.023621 0.023384 92 0.022497 0.022179 93 0.021372 0.021064 94 0.020278 0.020000 100 0.019230 0.019049 97 0.017303 0.017350 98 0.016427 0.016575 100 0.016	5	4	0.054 505	0.054 790
61 0.051579 0.051105 62 0.047661 0.047169 63 0.043975 0.043670 64 0.040585 0.040547 65 0.037504 0.037747 70 0.040816 0.040816 71 0.038398 0.038024 72 0.035948 0.035508 73 0.033581 0.033234 74 0.031353 0.031172 75 0.029284 0.029296 76 0.027379 0.027584 80 0.031250 0.031250 81 0.029681 0.029385 82 0.028058 0.027681 83 0.026459 0.026122 84 0.024926 0.024691 85 0.023478 0.023374 86 0.022124 0.022160 87 0.020864 0.021038 90 0.024691 0.024691 91 0.023621 0.023384 92 0.022497 0.022179 93 0.021372 0.01031250 98 0.016427 0.016575 100 0.020000 0.020000 101 0.016233 0.017330 98 0.016427 0.018180 97 0.017303 0.017350 98 0.0	6	0	0.055 556	0.055 556
62 0.047661 0.047169 63 0.043975 0.043670 64 0.040585 0.040547 65 0.037504 0.037747 70 0.040816 0.040816 71 0.038398 0.038024 72 0.035948 0.035508 73 0.033581 0.033234 74 0.031353 0.031172 75 0.029284 0.029296 76 0.027379 0.027584 80 0.031250 0.031250 81 0.029681 0.029385 82 0.028058 0.027681 83 0.026459 0.026122 84 0.024926 0.024691 85 0.023478 0.023374 86 0.022124 0.022160 87 0.020644 0.021038 90 0.024691 0.024691 91 0.023621 0.023384 92 0.022497 0.022179 93 0.021372 0.021064 94 0.020070 0.019072 96 0.018237 0.018180 97 0.016323 0.017350 98 0.016427 0.016575 100 0.020000 0.019049 102 0.018433 0.015178 104 0.01	6	1	0.051 579	0.051 105
63 0.043975 0.043670 64 0.040585 0.040547 65 0.037504 0.037747 70 0.040816 0.040816 71 0.038398 0.038024 72 0.035948 0.035508 73 0.033581 0.033234 74 0.031353 0.031172 75 0.029284 0.029296 76 0.027379 0.027584 80 0.031250 0.031250 81 0.029681 0.029385 82 0.028058 0.027681 83 0.026459 0.026122 84 0.024926 0.024691 85 0.022174 0.022160 87 0.022621 0.023374 86 0.022124 0.022160 87 0.022621 0.023384 92 0.022497 0.022179 93 0.021372 0.021064 94 0.020278 0.020001 95 0.019230 0.019072 96 0.016227 0.016575 100 0.020000 0.020000 101 0.019240 0.019049 102 0.018433 0.018165 103 0.017615 0.017340 104 0.016630 0.015851 106 0	6	2	0.047 661	0.047 169
64 0.040585 0.040547 65 0.037504 0.037747 70 0.040816 0.040816 71 0.038398 0.038024 72 0.035948 0.035508 73 0.033581 0.033234 74 0.031353 0.031172 75 0.029284 0.029296 76 0.027379 0.027584 80 0.031250 0.031250 81 0.029681 0.029385 82 0.026459 0.026122 84 0.024926 0.024691 85 0.022124 0.022160 87 0.020864 0.021038 90 0.024691 0.024691 91 0.023621 0.023374 86 0.022124 0.022160 87 0.020864 0.021038 90 0.024691 0.024691 91 0.023621 0.023384 92 0.022497 0.022179 93 0.021372 0.021064 94 0.020000 0.019072 96 0.018237 0.018180 97 0.017303 0.017350 98 0.016427 0.016575 100 0.020000 0.020000 101 0.019240 0.019049 102 0.018433 0.018165 103 <td>6</td> <td>3</td> <td>0.043 975</td> <td>0.043 670</td>	6	3	0.043 975	0.043 670
65 $0.037\ 504$ $0.037\ 747$ 70 $0.040\ 816$ $0.040\ 816$ 71 $0.038\ 398$ $0.038\ 024$ 72 $0.035\ 948$ $0.035\ 508$ 73 $0.033\ 581$ $0.033\ 234$ 74 $0.031\ 353$ $0.031\ 172$ 75 $0.029\ 284$ $0.029\ 295$ 80 $0.031\ 250$ $0.031\ 250$ 81 $0.029\ 681$ $0.029\ 385$ 82 $0.028\ 058$ $0.027\ 681$ 83 $0.026\ 459$ $0.026\ 122$ 84 $0.024\ 926$ $0.024\ 691$ 85 $0.023\ 478$ $0.023\ 374$ 86 $0.022\ 124$ $0.022\ 160$ 87 $0.020\ 664$ $0.021\ 038$ 90 $0.024\ 691$ $0.024\ 691$ 91 $0.023\ 621$ $0.023\ 384$ 92 $0.022\ 179$ $0.022\ 179$ 93 $0.021\ 372$ $0.021\ 064$ 94 $0.020\ 278$ $0.020\ 031$ 95 $0.019\ 230$ $0.019\ 072$ 96 $0.018\ 237$ $0.018\ 180$ 97 $0.017\ 303$ $0.017\ 350$ 98 $0.016\ 427$ $0.018\ 657$ 100 $0.020\ 000$ $0.020\ 000$ 101 $0.018\ 433$ $0.018\ 165$ 103 $0.017\ 615$ $0.017\ 340$ 104 $0.016\ 810$ $0.015\ 851$ 106 $0.013\ 902$ <td>6</td> <td>4</td> <td>0.040 585</td> <td>0.040 547</td>	6	4	0.040 585	0.040 547
70 0.040816 0.040816 71 0.038398 0.038024 72 0.035948 0.035508 73 0.033581 0.033234 74 0.031353 0.031172 75 0.029284 0.029296 76 0.027379 0.027584 80 0.031250 0.031250 81 0.029681 0.029385 82 0.028058 0.027681 83 0.026459 0.0266122 84 0.024926 0.024691 85 0.023478 0.023374 86 0.022124 0.022160 87 0.020664 0.021038 90 0.024691 0.024691 91 0.023621 0.023384 92 0.022497 0.022179 93 0.021372 0.021064 94 0.020278 0.020001 95 0.019230 0.019072 96 0.018237 0.018180 97 0.017303 0.017350 98 0.016427 0.016575 100 0.020000 0.020000 101 0.019240 0.019049 102 0.018433 0.018165 103 0.017533 0.015851 106 0.015283 0.015851 107	6	5	0.037 504	0.037 747
71 $0.038 \ 398$ $0.038 \ 024$ 72 $0.035 \ 948$ $0.035 \ 508$ 73 $0.033 \ 581$ $0.033 \ 234$ 74 $0.031 \ 353$ $0.031 \ 172$ 75 $0.029 \ 284$ $0.029 \ 296$ 76 $0.027 \ 379$ $0.027 \ 584$ 80 $0.031 \ 250$ $0.031 \ 250$ 81 $0.029 \ 681$ $0.029 \ 385$ 82 $0.028 \ 058$ $0.027 \ 681$ 83 $0.026 \ 459$ $0.026 \ 122$ 84 $0.024 \ 926$ $0.024 \ 691$ 85 $0.023 \ 478$ $0.023 \ 374$ 86 $0.022 \ 124$ $0.022 \ 160$ 87 $0.020 \ 664$ $0.021 \ 038$ 90 $0.024 \ 691$ $0.024 \ 691$ 91 $0.023 \ 621$ $0.023 \ 384$ 92 $0.022 \ 497$ $0.022 \ 179$ 93 $0.021 \ 372$ $0.021 \ 064$ 94 $0.020 \ 278$ $0.020 \ 031$ 95 $0.019 \ 230$ $0.019 \ 072$ 96 $0.018 \ 237$ $0.018 \ 180$ 97 $0.017 \ 303$ $0.017 \ 350$ 98 $0.016 \ 427$ $0.016 \ 575$ 100 $0.020 \ 000$ $0.020 \ 000$ 101 $0.019 \ 240$ $0.019 \ 049$ 102 $0.018 \ 433$ $0.018 \ 165$ 103 $0.017 \ 515$ $0.016 \ 571$ 105 $0.016 \ 630$ $0.015 \ 85$	7	0	0.040 816	0.040 816
72 0.035948 0.035508 73 0.033581 0.033234 74 0.031353 0.031172 75 0.029284 0.029296 76 0.027379 0.027584 80 0.031250 0.031250 81 0.029681 0.029385 82 0.028058 0.027681 83 0.026459 0.026122 84 0.024926 0.024691 85 0.023478 0.023374 86 0.022124 0.022160 87 0.020664 0.021038 90 0.024691 0.024691 91 0.023621 0.023384 92 0.022497 0.022179 93 0.021372 0.021064 94 0.020278 0.020031 95 0.019230 0.019072 96 0.018237 0.018180 97 0.017303 0.017350 98 0.016427 0.016575 100 0.020000 0.020000 101 0.019240 0.019049 102 0.018433 0.018165 103 0.017615 0.017340 104 0.0166150 0.015851 106 0.015283 0.015178 107 0.014573 0.014546 108 <td< td=""><td>7</td><td>1</td><td>0.038 398</td><td>0.038 024</td></td<>	7	1	0.038 398	0.038 024
73 0.033581 0.033234 74 0.031353 0.031172 75 0.029284 0.029296 76 0.027379 0.027584 80 0.031250 0.031250 81 0.029681 0.029385 82 0.028058 0.027681 83 0.026459 0.026122 84 0.024926 0.024691 85 0.023478 0.023374 86 0.022124 0.022160 87 0.020864 0.021038 90 0.024691 0.024691 91 0.023621 0.023384 92 0.022497 0.022179 93 0.021372 0.021064 94 0.020278 0.020031 95 0.019230 0.019072 96 0.018237 0.018180 97 0.017303 0.017350 98 0.016427 0.016575 100 0.020000 0.020000 101 0.019240 0.019049 102 0.018433 0.018165 103 0.017615 0.017340 104 0.016810 0.015851 106 0.015283 0.015178 107 0.014573 0.014546 108 0.013902 0.013963	7	2	0.035 948	0.035 508
74 0.031353 0.031172 75 0.029284 0.029296 76 0.027379 0.027584 80 0.031250 0.031250 81 0.029681 0.029385 82 0.028058 0.027681 83 0.026459 0.026122 84 0.024926 0.024691 85 0.023478 0.023374 86 0.022124 0.022160 87 0.020864 0.021038 90 0.024691 0.024691 91 0.023621 0.023384 92 0.022497 0.022179 93 0.021372 0.021064 94 0.020278 0.020031 95 0.019230 0.019072 96 0.018237 0.018180 97 0.017303 0.017350 98 0.016427 0.016575 100 0.020000 0.020000 101 0.019240 0.019049 102 0.018433 0.018165 103 0.017615 0.017340 104 0.016300 0.015851 106 0.015283 0.015178 107 0.014573 0.014546 108 0.013902 0.013396	7	3	0.033 581	0.033 234
75 $0.029 284$ $0.029 296$ 76 $0.027 379$ $0.027 584$ 80 $0.031 250$ $0.031 250$ 81 $0.029 681$ $0.029 385$ 82 $0.028 058$ $0.027 681$ 83 $0.026 459$ $0.026 122$ 84 $0.024 926$ $0.024 691$ 85 $0.023 478$ $0.023 374$ 86 $0.022 124$ $0.022 160$ 87 $0.020 864$ $0.021 038$ 90 $0.024 691$ $0.024 691$ 91 $0.023 621$ $0.023 384$ 92 $0.022 497$ $0.022 179$ 93 $0.021 372$ $0.021 064$ 94 $0.020 278$ $0.020 031$ 95 $0.019 230$ $0.019 072$ 96 $0.018 237$ $0.018 180$ 97 $0.017 303$ $0.017 350$ 98 $0.016 427$ $0.016 575$ 100 $0.020 000$ $0.020 000$ 101 $0.019 240$ $0.019 049$ 102 $0.018 433$ $0.018 165$ 103 $0.017 615$ $0.017 340$ 104 $0.016 810$ $0.015 571$ 107 $0.014 573$ $0.014 546$ 108 $0.013 902$ $0.013 953$ 109 $0.013 268$ $0.013 396$	7	4	0.031 353	0.031 172
76 $0.027 379$ $0.027 584$ 80 $0.031 250$ $0.031 250$ 81 $0.029 681$ $0.029 385$ 82 $0.028 058$ $0.027 681$ 83 $0.026 459$ $0.026 122$ 84 $0.024 926$ $0.024 691$ 85 $0.023 478$ $0.023 374$ 86 $0.022 124$ $0.022 160$ 87 $0.020 864$ $0.021 038$ 90 $0.024 691$ $0.024 691$ 91 $0.023 621$ $0.023 384$ 92 $0.022 497$ $0.022 179$ 93 $0.021 372$ $0.021 064$ 94 $0.020 278$ $0.020 031$ 95 $0.019 230$ $0.019 072$ 96 $0.018 237$ $0.018 180$ 97 $0.017 303$ $0.017 350$ 98 $0.016 427$ $0.016 575$ 100 $0.020 000$ $0.020 000$ 101 $0.019 240$ $0.019 049$ 102 $0.018 433$ $0.018 165$ 103 $0.017 615$ $0.017 340$ 104 $0.016 810$ $0.015 851$ 106 $0.015 283$ $0.015 178$ 107 $0.014 573$ $0.014 546$ 108 $0.013 902$ $0.013 396$	7	5	0.029 284	0.029 296
80 $0.031 250$ $0.031 250$ 81 $0.029 681$ $0.029 385$ 82 $0.028 058$ $0.027 681$ 83 $0.026 459$ $0.026 122$ 84 $0.024 926$ $0.024 691$ 85 $0.023 478$ $0.023 374$ 86 $0.022 124$ $0.022 160$ 87 $0.020 864$ $0.021 038$ 90 $0.024 691$ $0.024 691$ 91 $0.023 621$ $0.023 384$ 92 $0.022 497$ $0.022 179$ 93 $0.021 372$ $0.021 064$ 94 $0.020 278$ $0.020 031$ 95 $0.019 230$ $0.019 072$ 96 $0.018 237$ $0.018 180$ 97 $0.017 303$ $0.017 350$ 98 $0.016 427$ $0.016 575$ 100 $0.020 000$ $0.020 000$ 101 $0.019 240$ $0.019 049$ 102 $0.018 433$ $0.018 165$ 103 $0.017 615$ $0.017 340$ 104 $0.016 810$ $0.016 571$ 105 $0.016 030$ $0.015 851$ 106 $0.015 283$ $0.015 178$ 107 $0.014 573$ $0.014 546$ 108 $0.013 902$ $0.013 396$	7	6	0.027 379	0.027 584
81 $0.029\ 681$ $0.029\ 385$ 82 $0.028\ 058$ $0.027\ 681$ 83 $0.026\ 459$ $0.026\ 122$ 84 $0.024\ 926$ $0.024\ 691$ 85 $0.023\ 478$ $0.023\ 374$ 86 $0.022\ 124$ $0.022\ 160$ 87 $0.020\ 864$ $0.021\ 038$ 90 $0.024\ 691$ $0.024\ 691$ 91 $0.023\ 621$ $0.023\ 384$ 92 $0.022\ 497$ $0.022\ 179$ 93 $0.021\ 372$ $0.021\ 064$ 94 $0.020\ 278$ $0.020\ 031$ 95 $0.019\ 230$ $0.019\ 072$ 96 $0.018\ 237$ $0.018\ 180$ 97 $0.017\ 303$ $0.017\ 350$ 98 $0.016\ 427$ $0.016\ 575$ 100 $0.020\ 000$ $0.020\ 000$ 101 $0.019\ 240$ $0.019\ 049$ 102 $0.018\ 433$ $0.018\ 165$ 103 $0.017\ 615$ $0.017\ 340$ 104 $0.016\ 810$ $0.016\ 571$ 105 $0.016\ 030$ $0.015\ 851$ 106 $0.015\ 283$ $0.015\ 178$ 107 $0.014\ 573$ $0.014\ 546$ 108 $0.013\ 902$ $0.013\ 396$	8	0	0.031 250	0.031 250
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	8	1	0.029 681	0.029 385
83 0.026459 0.026122 84 0.024926 0.024691 85 0.023478 0.023374 86 0.022124 0.022160 87 0.020864 0.021038 90 0.024691 0.024691 91 0.023621 0.023384 92 0.022497 0.022179 93 0.021372 0.021064 94 0.020278 0.020031 95 0.019230 0.019072 96 0.018237 0.018180 97 0.017303 0.017350 98 0.016427 0.016575 100 0.020000 0.020000 101 0.019240 0.019049 102 0.018433 0.018165 103 0.017615 0.017340 104 0.016810 0.016571 105 0.016030 0.015851 106 0.015283 0.015178 107 0.014573 0.014546 108 0.013902 0.013396	8	2	0.028 058	0.027 681
84 0.024926 0.024691 85 0.023478 0.023374 86 0.022124 0.022160 87 0.020864 0.021038 90 0.024691 0.024691 91 0.023621 0.023384 92 0.022497 0.022179 93 0.021372 0.021064 94 0.020278 0.020031 95 0.019230 0.019072 96 0.018237 0.018180 97 0.017303 0.017350 98 0.016427 0.016575 100 0.020000 0.020000 101 0.019240 0.019049 102 0.018433 0.018165 103 0.017615 0.017340 104 0.016810 0.016571 105 0.016030 0.015851 106 0.015283 0.015178 107 0.014573 0.014546 108 0.013902 0.013396	8	3	0.026 459	0.026 122
85 $0.023 478$ $0.023 374$ 86 $0.022 124$ $0.022 160$ 87 $0.020 864$ $0.021 038$ 90 $0.024 691$ $0.024 691$ 91 $0.023 621$ $0.023 384$ 92 $0.022 497$ $0.022 179$ 93 $0.021 372$ $0.021 064$ 94 $0.020 278$ $0.020 031$ 95 $0.019 230$ $0.019 072$ 96 $0.018 237$ $0.018 180$ 97 $0.017 303$ $0.017 350$ 98 $0.016 427$ $0.016 575$ 100 $0.020 000$ $0.020 000$ 101 $0.019 240$ $0.019 049$ 102 $0.018 433$ $0.018 165$ 103 $0.017 615$ $0.017 340$ 104 $0.016 810$ $0.016 571$ 105 $0.016 030$ $0.015 851$ 106 $0.015 283$ $0.015 178$ 107 $0.014 573$ $0.014 546$ 108 $0.013 902$ $0.013 396$	8	4	0.024 926	0.024 691
86 $0.022 124$ $0.022 160$ 87 $0.020 864$ $0.021 038$ 90 $0.024 691$ $0.024 691$ 91 $0.023 621$ $0.023 384$ 92 $0.022 497$ $0.022 179$ 93 $0.021 372$ $0.021 064$ 94 $0.020 278$ $0.020 031$ 95 $0.019 230$ $0.019 072$ 96 $0.018 237$ $0.018 180$ 97 $0.017 303$ $0.017 350$ 98 $0.016 427$ $0.016 575$ 100 $0.020 000$ $0.020 000$ 101 $0.019 240$ $0.019 049$ 102 $0.018 433$ $0.018 165$ 103 $0.017 615$ $0.017 340$ 104 $0.016 810$ $0.016 571$ 105 $0.016 030$ $0.015 851$ 106 $0.015 283$ $0.015 178$ 107 $0.014 573$ $0.014 546$ 108 $0.013 902$ $0.013 396$	8	5	0.023 478	0.023 374
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	8	6	0.022 124	0.022 160
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	8	7	0.020 864	0.021 038
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	9	0	0.024 691	0.024 691
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	9	1	0.023 621	0.023 384
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	9	2	0.022 497	0.022 179
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	9	3	0.021 372	0.021 064
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	9	4	0.020 278	0.020 031
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	9	5	0.019 230	0.019072
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	9	6	0.018 237	0.018 180
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	9	7	0.017 303	0.017 350
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10 1 0.019 240 0.019 049 10 2 0.018 433 0.018 165 10 3 0.017 615 0.017 340 10 4 0.016 810 0.016 571 10 5 0.016 030 0.015 851 10 6 0.015 283 0.015 178 10 7 0.014 573 0.014 546 10 8 0.013 902 0.013 953 10 9 0.013 268 0.013 396	10	0	0.020 000	0.020 000
10 2 0.018 433 0.018 165 10 3 0.017 615 0.017 340 10 4 0.016 810 0.016 571 10 5 0.016 030 0.015 851 10 6 0.015 283 0.015 178 10 7 0.014 573 0.014 546 10 8 0.013 902 0.013 953 10 9 0.013 268 0.013 396	10	1	0.019 240	0.019 049
10 3 0.017 615 0.017 340 10 4 0.016 810 0.016 571 10 5 0.016 030 0.015 851 10 6 0.015 283 0.015 178 10 7 0.014 573 0.014 546 10 8 0.013 902 0.013 953 10 9 0.013 268 0.013 396	10	2	0.018 433	0.018 165
10 4 0.016 810 0.016 571 10 5 0.016 030 0.015 851 10 6 0.015 283 0.015 178 10 7 0.014 573 0.014 546 10 8 0.013 902 0.013 953 10 9 0.013 268 0.013 396	10	3	0.017 615	0.017 340
10 5 0.016 030 0.015 851 10 6 0.015 283 0.015 178 10 7 0.014 573 0.014 546 10 8 0.013 902 0.013 953 10 9 0.013 268 0.013 396	10	4	0.016 810	0.016 571
10 6 0.015 283 0.015 178 10 7 0.014 573 0.014 546 10 8 0.013 902 0.013 953 10 9 0.013 268 0.013 396	10	5	0.016 030	0.015 851
10 7 0.014 573 0.014 546 10 8 0.013 902 0.013 953 10 9 0.013 268 0.013 396	10	6	0.015 283	0.015 178
10 8 0.013 902 0.013 953 10 9 0.013 268 0.013 396	10	7	0.014 573	0.014 546
10 9 0.013 268 0.013 396	10	8	0.013 902	0.013 953
	10	9	0.013 268	0.013 396

IV. CRITICAL SCREENING PARAMETER

For the s states, δ_c is readily obtained from the expression for the energy to be

$$\delta_c = 2/n^2 . \tag{11}$$

For $l\neq 0$ states, δ_c was calculated by numerical solution of the Schrödinger equation and the results for $n \leq 10$ are shown in Table III. For the sake of completeness, the results for l=0 are also shown, obtained from Eq. (11). Patil²⁶ has obtained an approximate expression for the critical strength Z_c . In terms of δ_c , his equation can be written as

$$\delta_c = 1/(n\sqrt{2} + 0.1645l + 0.0983l/n)^2 .$$
 (12)

Values of δ_c obtained from this expression are also shown in Table III. It will be noticed that the results obtained from Patil's equation are in good agreement with the exact values—which is rather remarkable for such a simple equation.

The only other calculation of δ_c for the Hulthén potential is that of Popov and Weinberg,²⁷ who have calculated $n^2\delta_c$ for a few states which have l=n-1 by perturbation series summation. Their values are in good agreement with the exact ones calculated here.

For s states, as Eq. (11) shows, $n^2\delta_c = 2$. For $l \neq 0$ states, we show the product $n^2\delta_c$ as a function of n in Fig. 4. Points for the same l have been joined by lines. It



FIG. 4. $n^2\delta_c$ as a function of *n*. Points for the same *l* have been joined by lines. The numbers on the right of the curves represent *l* values.



FIG. 5. Number of bound states n^* as a function of $1/\delta_c$. The straight line represents a least-squares fit, Eq. (13).

will be noticed that in each case $n^2\delta_c$ increases with n and appears to approach an asymptotic value close to 2.

In the order of decreasing δ_c values, the level order is as follows: 1s, 2s, 2p, 3s, 3p, 3d, 4s, 4p, 4d, 4f, 5s, 5p, 5d, 5f,6s, 5g, 6p, 6d, 6f, 7s, 6g, 7p, 6h, 7d, 7f, 7g, 8s, 8p, 7h, 8d, 7i, 8f, $8g, 9s, 9p, 8h, 8i, 9f, 8k, 9g, 10s, 10p, 9h, 10d, 9i, 10f, \ldots$ It will be noticed that as one goes to higher levels there is an increasing departure from the ordering of levels in a Coulomb potential. This is a consequence of an increasing degree of crossover of levels, which was discussed in Sec. II.

The total number of states n^* which have negative energy at a given δ has been studied for the static screened Coulomb potential by Rogers *et al.*³⁸ In Fig. 5 we show n^* as a function of $1/\delta_c$ for the Hulthén potential. The distribution of points indicates a linear relationship and a least-squares treatment yields the following relation:

$$n^* = 0.8029 / \delta_c + 0.9018$$
 (13)

Equation (13) can be used to calculate the number of states which are bound at a given δ , or one can calculate δ_c for a given n^* .

V. COMPARISON OF EIGENENERGIES

In this section we compare the results obtained by the various methods for the eigenenergies of $l \neq 0$ levels. Besides the results obtained in this paper, we shall consider the results obtained by four other methods. We briefly describe these four other calculations.

Lai and Lin^{24} expanded the Hulthén potential in a series in the form

$$V(r) = -(Z/r) \sum_{k=0}^{\infty} V_k (\delta r)^k .$$
 (14)

By employing the Hellmann-Feynman theorem and the hypervirial theorems they were able to express the bound-state energies in a power series of the screening parameter δ . To improve the accuracy of the results, especially when δ is close to the critical screening parameter δ_c , they formed Padé approximants to the energy series. The results of Lai and Lin are shown in column 6 of Table I.

Patil's²⁶ treatment is based on the fact that the Hulthén potential allows a strong-coupling expansion. He uses an interpolation for the bound-state energies which incorporates the required behavior near the critical strength Z_c and also has a strong-coupling expansion in conformity with the first few known terms of the strong-coupling expansion. Patil's values are given in column 7 of Table I. Roy and Roychoudhury²⁸ and Tang and Chan²⁹ have

Roy and Roychoudhury²⁸ and Tang and Chan²⁹ have applied the shifted 1/N method of Sukhatme and collaborators³⁹ to calculate energies of $l \neq 0$ states of the Hulthén potential. Roy and Roychoudhury²⁸ have used two terms in their calculations, while Tang and Chan²⁹ have used three terms. The results of the latter authors are shown in column 8 of Table I.

Matthys and De Meyer³¹ have studied the $l \neq 0$ states in the context of algebraic perturbation theory using an approach which is based on the Lie algebra of the group SO(2,1). They introduce a so-called tilting transformation that relates between physical states and the group states which constitute a basis of the relevant unitary irreducible representations of SO(2,1). It is shown that all the matrix elements of the energy functional can be expressed in closed form. Then the energy matrix, in which the unknown bound-state energies occur on diagonal as well as on off-diagonal positions, is numerically diagonalized by means of a Gauss-Seidel iteration scheme. The tilting parameter is adjusted in order to accelerate the convergence of the perturbation expansions. Column 9 of Table I shows the eigenenergies obtained by Matthys and De Meyer.³¹

Next we compare the various sets of results. A comparison of the eigenenergies obtained from the wave functions of Eq. (7) with the "exact" values shows that the two are almost the same except in the high-screening region. For the 2p state the values obtained by the modified wave function, Eq. (9), are the same as those obtained from the unmodified one at low screening, but slightly inferior in the high-screening region. Amongst the four sets (columns 6-9 of Table I) of other workers, the results of Matthys and De Meyer³¹ are practically identical to the "exact" results. The next best results are those of Lai and Lin,²⁴ which agree with the exact results to the number of significant figures quoted by these authors, except in 3 cases in the high-screening region. Next in accuracy in the low- and medium-screening regions are the shifted 1/N expansion results of Tang and Chan,²⁹ but no conclusion is possible for the highscreening region as Tang and Chan have not given any

results for such values δ . It has been found⁴⁰ that the shifted 1/N expansion gives poor results for the static screened Coulomb potential in the high-screening region and it is entirely possible that a similar situation may hold for the Hulthén potential.

VI. OSCILLATOR STRENGTHS

There has been no previous calculation of the oscillator strengths for the Hulthén potential. In this section we obtain results for the absorption oscillator strength for $1s \rightarrow 2p$, $1s \rightarrow 3p$, and $2p \rightarrow 3d$ transitions by using the analytical wave functions of Sec. III. The absorption oscillator strength for the $nl \rightarrow n'l'$ transition is given by⁴¹

$$f = \frac{2}{3} \frac{\max(l, l')}{(2l+1)} (E_{n'l'} - E_{nl}) \left| \int_0^\infty \chi_{nl} r \chi_{n'l'} dr \right|^2 .$$
 (15)

The normalization constants for the four required wave functions are as follows:

$$N_{1s} = (4/\delta^2 - 1)^{1/2} \tag{16a}$$

$$N_{2p} = \left[\frac{(1-\mu^2)(1-4\mu^2)}{24\mu^4}\right]^{1/2}$$
(16b)

$$N_{3p} = \left[\frac{2(4-9\mu^2)(4-81\mu^2)}{2187\mu^4}\right]^{1/2}$$
(16c)

$$N_{3d} = \left[\frac{(1 - 9\mu_2^2)(4 - 9\mu_2^2)(4 - 81\mu_2^2)}{196\,830\mu_2^6} \right]^{1/2} .$$
(16d)

In Eq. (16d), μ_2 has been used for μ to distinguish it from the μ for the 2p state in the expression for f for the $2p \rightarrow 3d$ transition. Following Shore and Menzel⁴² we

TABLE IV. Absorption oscillator strengths in atomic units for the Hulthén potential.

Transition	δ	f
1s-2p	0.025	0.4155
•	0.050	0.4134
	0.075	0.4098
	0.10	0.4047
	0.15	0.3897
	0.20	0.3671
	0.25	0.3348
	0.30	0.2886
	0.35	0.2169
1s-3p	0.025	0.078 34
-	0.050	0.076 00
	0.075	0.071 96
	0.10	0.066 00
	0.15	0.046 33
2 <i>p</i> -3 <i>d</i>	0.025	0.6916
-	0.050	0.6785
	0.075	0.6547
	0.10	0.6161
	0.15	0.4400

represent the transition integral by I(nl, n'l'),

$$I(nl,n'l') = \int_0^\infty \chi_{nl} r \chi_{n'l'} dr . \qquad (17)$$

Using wave functions for the 2p, 3p, and 3d states from Eq. (7) and the exact wave function for the 1s state, the transition integrals were evaluated and the results are as follows:

$$I(1s,2p) = 768N_{1s}N_{2p}\mu^{2}\delta \times (-243\delta^{2} - 1458\mu^{2} - 27\delta^{4} + 2187 + 3\delta^{6} + 32\mu^{4}\delta^{2} - 32\mu^{6} - 18\delta^{4}\mu^{2} + 288\mu^{4} + 36\mu^{2}\delta^{2}) \times \{(9-\delta^{2})^{2}[(3-\delta)^{2} - 4\mu^{2}]^{2}[(3+\delta)^{2} - 4\mu^{2}]^{2}\}^{-1},$$

$$I(1s,3p) = N_{1s}N_{3p}\left[\frac{p^{2} + 9\mu^{2}/4 + p}{(p^{2} - 9\mu^{2}/4)^{2}} - \frac{p^{2} + \mu^{2}/4 + p}{(p^{2} - \mu^{2}/4)^{2}} - \frac{q^{2} + 9\mu^{2}/4 + q}{(q^{2} - 9\mu^{2}/4)^{2}} + \frac{q^{2} + \mu^{2}/4 + q}{(q^{2} - \mu^{2}/4)^{2}}\right],$$
(19)

where

$$p = -\frac{4}{3} + \delta/2, \quad q = -\frac{4}{3} - \delta/2$$

and

$$I(2p, 3d) = N_{2p}N_{3d} \left[-\frac{6p\mu_2}{(p^2 - 9\mu_2^2/4)^2} + \frac{6p\mu_2}{(p^2 - \mu_2^2/4)^2} - \frac{10\mu_2}{(\frac{25}{36} - 9\mu_2^2/4)^2} + \frac{10\mu_2}{(\frac{25}{36} - \mu_2^2/4)^2} - \frac{6q\mu_2}{(q^2 - 9\mu_2^2/4)^2} + \frac{6q\mu_2}{(q^2 - \mu_2^2/4)^2} \right],$$
(20)

where $p = -\frac{5}{6} + \mu$ and $q = -\frac{5}{6} - \mu$. In Eq. (20), μ refers to the 2p wave function and μ_2 to the 3d function.

The calculated oscillator strengths are shown in Table IV. It will be noticed that for each of the transitions, as δ increases, at first there is a slow decrease, but near δ_c there is a rapid drop in the value of the oscillator strength.

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