

Laser-induced fluorescence spectroscopy of the Hg₃ excimer

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We report what we believe to be the first observation of a laser-induced fluorescence band, excited by absorption from a low-lying state of the Hg₃ excimer using a "pump-and-probe" method. A study of the fluorescence intensity as a function of the time separation between the "pump" and "probe" laser pulses provides clear evidence for the presence of both Hg₃ and Hg₂ molecules in the laser-excited mercury vapor. The fluorescence may be due to trimer emission or to dimer emission following dissociation of a trimer state.

Although the existence of molecular mercury emission bands at 3350 and 4850 Å has been known for over 50 years, they have, until recently, been attributed to the Hg₂ dimer,¹ particularly since their persistence times at low Hg-vapor densities were found to be identical and equal to about 50 ns.² More recent experiments, employing laser spectroscopy³ and flash photolysis,⁴ as well as theoretical calculations,^{5,6} led to the conclusion that the 4850-Å continuum could not be emitted by Hg₂ and that it must be due to the radiative decay of a low-lying state of Hg₃. This has been assigned as an AO_u^\pm state by Callear and Lai,⁴ who also designate it as the lower state of a series of unstructured absorption bands ranging from 2900 to 4650 Å. We will use their assignment, although no other evidence is available to support it. There have been no reports of fluorescence from excited Hg₃ other than the 4850 Å, and Callear and Lai were not able to suggest any assignments for the other Hg₃ levels. The most recently observed laser-induced fluorescence (LIF) spectra of mercury excimers arise exclusively from various excited states of Hg₂.⁷⁻⁹ In this communication, we present the first evidence of fluorescence emitted from a more highly excited state of Hg₃ which was populated by absorption from the Hg₃ AO_u^\pm state using "pump and probe" techniques. Figure 1 shows a partial energy-level diagram for Hg₂ based on the calculations of Baylis¹⁰ and Mies *et al.*,⁵ and on our experimental results.^{8,9} The energy level corresponding to the lowest Hg₃ state shown in the figure conforms to the data of Drullinger *et al.*³

The arrangement of the apparatus has been described elsewhere.^{8,9} Mercury vapor, at a density of 5×10^{17} cm⁻³ and contained in a quartz cell, was irradiated with 2660-Å pulses from the fourth harmonic of a Nd yttrium-aluminum-garnet (YAG) laser, resulting in a population of Hg₂ AO_g^\pm molecules. After a delay which was variable from 50 ns to 50 μs, pulses from a N₂ laser-pumped dye laser, directed collinearly with the Nd:YAG pulses, were used to probe the AO_g^\pm molecules as well as the collisionally produced Hg₃ molecules. The resulting LIF spectrum was observed at right angles to the laser beams, resolved with a monochromator and detected with an EMI 9816QB photomultiplier. A Biomation model 6500 transient digitizer, controlled by a PET microcom-

puter, together constituted a signal integrating and/or averaging data acquisition system.

The experiment consisted of three separate parts. Firstly, monochromator scans of the time-integrated fluorescence were made with various pump-probe delays. Secondly, the intensity of the time-integrated fluorescence detected at a particular monochromator setting was measured as a continuous function of pump-probe delay. Finally, time-resolved fluorescence measurements were made at various monochromator settings, both with and without probe laser pulses. The monochromator and probe laser could be scanned in constant wavelength increments which were multiples of $\frac{1}{2}$ and $\frac{1}{28}$ Å, respectively. The pump-probe delay was produced by a computer controlled delay generator and could be programmed to values ranging from 50 ns to 1.8 ms in steps of 5.0 ns.

The Hg₃ AO_u^\pm state is produced through collisional processes which involve the initial formation of Hg₂ in the $D1_u$ state by photoassociation of a ground-state atom pair. The $D1_u$ excimers collisionally relax to the AO_g^\pm metastable states which participate in the formation of

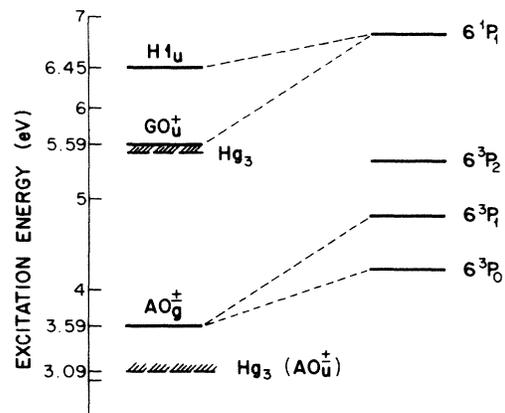


FIG. 1. Energy-level diagram showing the pertinent Hg₂ and Hg states. The locations of the Hg₃ (AO_u^\pm) state (Ref. 3) and of the higher Hg₃ states as suggested by the results of this experiment are also indicated.

Hg_3 excimers through three-body collisions involving two ground-state mercury atoms.^{3,11} The trimer population time dependence is a function of the density and collision rate,¹¹ and its radiative decay gives rise to the 4850-Å band. The tunable probe laser may then be used to excite either the dimers or trimers to a higher state whose decay produces a spectrum characteristic of the two states involved in the fluorescence transition.

Figure 2 shows scans of the time-integrated LIF spectrum obtained at three different delay times of the probe pulse and with the probe wavelength set to 4335 Å. The peak near 1996 Å is due to the $H1_u \rightarrow X0_g^+$ emission from Hg_2 ,⁹ and its intensity remains constant within a factor of 2 for pump-probe delays in the range 0–8 μs . However, the intensity of the broad LIF band centered near 2170 Å increases significantly with the probe pulse delay. Since the time-integrated intensity of an LIF band is proportional to the population density of the state absorbing the probe laser pulse, the different time behavior clearly shows that it is not the $\text{Hg}_2 A0_g^\pm$ states which are the absorbing species leading to the 2170 Å emission.

In order to identify the states associated with the 2170-Å band, the two additional experiments were carried out and their results are presented together in Fig. 3. The intensities of the 4850- and 3350-Å bands were measured as a function of time following the pump laser pulse and without using the probe laser. These data are shown in Figs. 3(a) and 3(b) and represent the time dependence of the $\text{Hg}_3 A0_u^\pm$ and $\text{Hg}_2 A0_g^\pm$ populations, respectively. In the final experiment, the LIF intensities of two bands were scanned in relation to the time delay between the

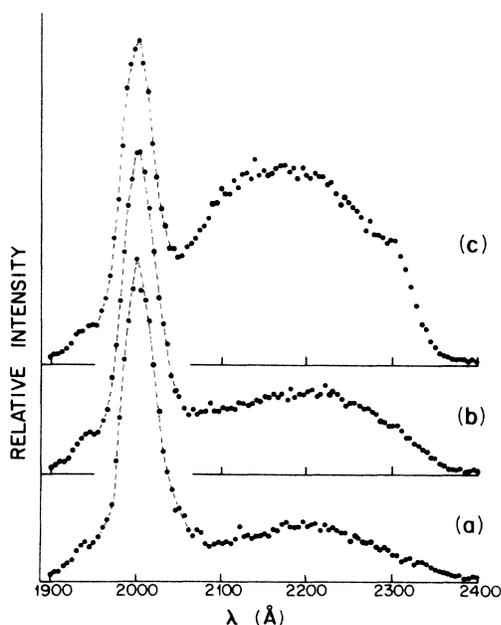


FIG. 2. 2170-Å LIF band of Hg_3 recorded at various pump-probe pulse delay times: (a) 150 ns; (b) 500 ns; (c) 9 μs . All intensities have been normalized to that of the 1996-Å peak due to $H1_u \rightarrow X0_g^+$ emission from Hg_2 .

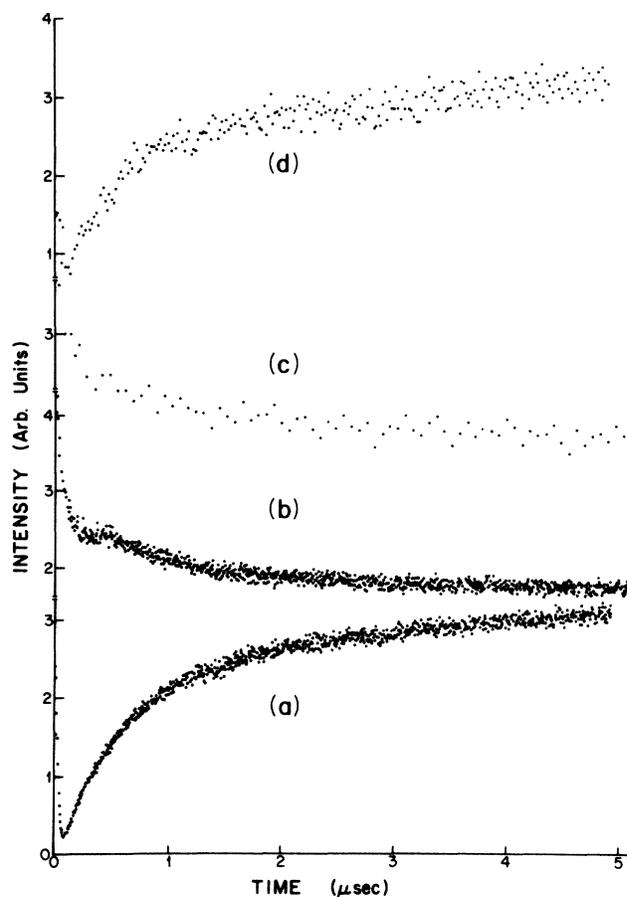


FIG. 3. (a) and (b) show time evolutions of the 4850-Å Hg_3 band and 3350-Å Hg_2 band, respectively, excited by pulses of pump-laser radiation. (c) and (d) show the LIF intensities of the 5130-Å Hg_2 band and 2170-Å Hg_3 band, respectively, in relation to the pump-probe delay.

pump and probe laser pulses. Figure 3(c) shows such a scan with the probe laser set at 4200 Å to excite the $A0_g^\pm \rightarrow J1_u$ transition¹² while detecting the LIF band at 5130 Å, arising from the $J1_u \rightarrow E2_g$ decay. The results of a similar scan of the 2170-Å band are shown in Fig. 3(d). The probe laser was set at 4650 Å in order to reduce fluorescence from dimer bands.^{4,8,12} However, similar scans could be obtained with probe laser wavelengths between 4350 and 5150 Å. The similarities between Figs. 3(a) and 3(d) and between 3(b) and 3(c) are immediately obvious. They confirm that the $\text{Hg}_2 D1_u$ state, which emits the 3350-Å band is in equilibrium with the $\text{Hg}_2 A0_g^\pm$ states. They also show clearly that the initial absorbing state of the 2170-Å LIF band is either the $\text{Hg}_3 A0_u^\pm$ state or one in collisional equilibrium with it. The population of this state initially increases at a rate characterized by its own collisional and radiative decay time.¹¹ At times later than those shown in Fig. 3, all four curves decay at a common rate, which is the collisional decay rate of the $\text{Hg}_2 A0_g^\pm$ metastable reservoir.

While the results of our investigation clearly show that a Hg_3 state is the absorbing species giving rise to an ab-

sorption continuum which ranges from below 4150 to 5100 Å, the designation of the upper state of the strong 2170-Å fluorescence bands is uncertain. The fluorescence decay time of the 2170-Å band is less than 3 ns since it follows the probe pulse time dependence. This result is consistent with a strongly allowed radiative transition or with significant collisional quenching of the excited Hg₃ state. It is also likely that the terminal state for such a transition dissociates into three ground-state Hg atoms as is the case with the 4850-Å band. Alternatively, the excited trimer might rapidly dissociate to a free atom and a Hg₂ G₀_u[±] molecule which emits in the same spectral region and has a short lifetime.⁸ Since the dissociation could populate several vibrational states of the dimer, the Condon diffraction patterns would be averaged out to give

the observed unstructured continuum. At present, we are unable to decide between these two possibilities.

The experiments described here show clearly the existence of an absorption band in the spectral region from 4150 to 5100 Å, arising from an excited Hg₃ molecule. If this state is located at 3.09 eV, the upper Hg₃ state must lie in the vicinity of 5.50–5.60 eV. Two possible sources for the resulting uv fluorescence are suggested, but a definitive answer must await further experimental and theoretical work.

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