PHYSICAL REVIEW A VOLUME 28, NUMBER 1 JULY 1983

Rotating harmonic oscillator: Its general solution and the lack of ground-state energy equipartition

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Analytically and numerically we give the general solution of the rotating harmonic oscillator. Our solution verifies the resolution of a disagreement in the literature, and can be understood physically. We point out that the solution demonstrates the incorrectness of the "energy equipartition theorem" for the ground state of all quantum oscillators.

The rotating —harmonic-oscillator (RHO) potential

 $v(x) = \frac{1}{2}$ $\frac{2}{3}$ $\frac{1}{2}$ $\frac{2}{3}$ $\frac{1}{2}$ $\frac{1}{2}$ $\frac{2}{3}$

$$
V(r) = \frac{1}{2} m \omega^2 (r - r_0)^2 = \hbar \omega \left[\frac{1}{2} (z - z_0)^2 \right] , \qquad (1)
$$

$$
z = \left[\frac{m \omega}{r} \right]^{1/2} r = ar, \quad z_0 = ar_0 , \qquad (2)
$$

has a history which can arguably be traced back a century in the development of quantum theory.^{$1-10$} Even so, the properties of the potential continue to be studied $11 - 16 -$ this study even including a disagreement on the properties of its solutions.¹²⁻¹⁶ solutions. $12 - 16$

In this Brief Report we will give the general solution showing how the eigenvalues change continuously from the three-dimensional harmonic oscillator ($z_0 \rightarrow 0$) to the RHO "diatomic molecule" $(z_0 \gg 0)$. In doing so we can verify "diatomic molecule" $(z_0 > 0)$. In doing so we can verify
"diatomic molecule" $(z_0 > 0)$. In doing so we can verify
the proper resulution^{13,15,16} of the above disagreemen which resolution can be understood on physical grounds.

To define the problem, start with the Schrödinger equation. Reduce out the angular variable, write the wave function as $\chi(r) = r\psi(r)$, change variables to z, and write all energies in terms of units of $\hbar \omega$:

$$
\left(\frac{d^2}{dz^2} - (z - z_0)^2 - \frac{l(l+1)}{z^2} + 2\epsilon\right) x = 0
$$
 (3)

First consider the large- z_0 case. Physically, this case is important. If the RHO potential is taken as a model diatomic-molecule potential, z_0 can be written as

$$
z_0 = 5.446 \, 114 \, (\mu_a k_{ic})^{1/2} r_A \quad , \tag{4}
$$

$$
k_{ic} = 1/\lambda = \nu_{ic}/c = \omega_{ic}/2\pi c \quad , \tag{5}
$$

where μ_a is the reduced mass in atomic mass units, k_{ic} is

the wave number for transition to the first vibrationally excited state in 1000 cm⁻¹, and r_A is the equilibrium distance of the atoms in angstroms. Thus, even for the hydrogen molecule, z_0 is 6.01, and for other molecules it is larger. Now by writing the function of z in Eq. (3) as a McLaurin series about z_0 , we can approximate the Schrödinger equation as

$$
\left[\frac{d^2}{dY^2} + Y^2 + \left(\frac{2\epsilon - \delta^2 - l(l+1)/(z_0 + \delta)^2}{\left[1 + 3l(l+1)/(z_0 + \delta)^4\right]^{1/2}}\right)\right]X = 0 \quad , \quad (6)
$$

$$
Y = (z - z_0 - \delta) [1 + 3l(l+1)/(z_0 + \delta)^4]^{1/2} , \qquad (7)
$$

$$
\delta = l(l+1)/z_0^3 \quad . \tag{8}
$$

For $z_0 \gg 0$ the effect of the boundary condition at $z = 0$ is negligible, so this is a one-dimensional harmonic oscillator. Thus we have the approximate eigenvalues $(l = 0, 1, 2...;$ $\nu = 0, 1, 2, \ldots$)

$$
\epsilon = (\nu + \frac{1}{2}) \left[1 + \frac{3I(I+1)}{(z_0 + \delta)^4} \right]^{1/2} + \frac{\delta^2}{2} + \frac{I(I+1)}{2(z_0 + \delta)^2} \tag{9}
$$

$$
\approx (\nu + \frac{1}{2}) + \frac{l(l+1)}{2z_0^2} + (\nu + \frac{1}{2})\frac{3l(l+1)}{2z_0^4} + \cdots \qquad (10)
$$

In essence, the above is a variant of what Schrödinger did, 8 exhibiting the first three terms in a power series in z_0^{-2} . In he 1930's, papers appeared^{9, 10} using WKB and perturbation methods which carried out this power series higher, agreeng up to order $(z_0^{-2})^4$.

Now consider the small- z_0 case. There, one can use standard first-order perturbation theory about the $z_0 = 0$ three-dimensional —harmonic-oscillator limit to obtain $(l = 0, 1, 2, \ldots, n = l, l + 2, l + 4, \ldots)$

$$
\epsilon = (n + \frac{3}{2}) + \frac{1}{2}z_0^2 - z_0 \sum_{k=0}^{(n-l)/2} (-1)^k \frac{\Gamma(\frac{1}{2}(n-l) - k - \frac{1}{2})}{\Gamma(\frac{1}{2}(n-l) - k + 1)} \frac{\Gamma(l+k+2)}{\Gamma(l+k+\frac{3}{2})} \frac{1}{\Gamma(-k-\frac{1}{2})\Gamma(k+1)} \tag{11}
$$

For the general solution, first consider $l = 0$. Changing variables in Eq. (3) to $y = z - z_0$, one has the standard onedimensional —harmonic-oscillator equation, but with the boundary conditions $\chi(\infty) = 0$ and $\chi(-z_0) = 0$. Such a solution is well known analytically. It is the parabolic cylinder function $D_{\mu}(y\sqrt{2})$, with eigenvalue $\mu + \frac{1}{2}$ deter-

mined by the boundary condition $D_{\mu}(-z_0\sqrt{2}) = 0^{17}$ The eigenvalues $\epsilon_{l=0}(z_0)$ are shown as the thick lines of Fig. 1.

For arbitrary l , one can numerically integrate the Schrödinger Eq. (3) to obtain the eigenvalues. Start by observing that at the origin and infinity, the wave function behaves as $(z)^{l+1}$ and $exp[-\frac{1}{2}(z-z_0)^2]$, respectively.

$$
\underline{.8} \qquad \qquad 4
$$

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FIG. 1. Eigenvalues of the RHO potential as a function of z_0 . The thick lines are for $l = 0$, thin lines for $l = 2$, and dashed lines for $l = 4$. The heavy circles show first-order - perturbation-theory predictions for $l = 0$ about the $z_0 = 0$ three-dimensional – harmonicoscillator limit. [See Eq. (11).]

Thus, by using the standard numerical-integration techniques of splitting the Schrödinger equation into two firstorder differential equations satisfying the above boundary conditions, integrating out from zero, and demanding that the wave function be small for $z \gg z_0$, one can obtain the eigenvalues for all l and z_0 . We show these eigenvalues for $l = 2$ and 4 in Fig. 1 and $l = 1$ and 3 in Fig. 2.

Figures 1 and 2 allow us to see the general properties of the solutions. For the transition from the small- z_0 to the large- z_0 case, the solutions change as

$$
\lim_{z_0 \to 0} \text{Curve}(n,l) \to \lim_{z_0 \to \infty} \text{Curve}(\nu = \frac{1}{2}(n-l),l) \quad . \tag{12}
$$

By Eq. (12) we mean that in the two limits of z_0 shown, the two eigenvalue levels connect. This is reminiscent of the transition from L, S to J, S coupling.

Our results verify the conclusions of Refs. 13, 15, and 16 in the disagreement¹²⁻¹⁶ mentioned above. Contrary to Refs. 12 and 14, for $l \neq 0$, there is a continuous change in all the eigenvalues which is dependent on the value of z_0 .

FIG. 2. Eigenvalues of the RHO potential as a function of z_0 . The thick lines are for $l = 1$ and the thin lines for $l = 3$. The heavy circles show first-order – perturbation-theory predictions for $l = 1$ about the $z_0 = 0$ three-dimensional - harmonic-oscillator limit. [See Eq. (11).] The small squares are specific values for $l = 1$ given in Table 2 of Ref. 13.

There is no solution which is independent of z_0 with eigenvalues $(n + \frac{1}{2})$. From our point of view this makes sense intuitively. As z_0 gets larger, we are changing continuously from the three-dimensional - harmonic-oscillator problem to the one-dimensional - harmonic-oscillator problem. The change is a continuous perturbation from one limit to the other, and is nonsingular.

Further, our figures show where the special-case exact results of Ref. 16 fit in. In Ref. 16 it was shown by continued-fraction methods that for certain specific values of z_0 , the continued-fraction solution terminates to give exact eigenvalues $(\frac{1}{2} + l +$ positive integer). [These methods also specifically exclude the predicted exact solutions of
Refs. 12 and 14, $\epsilon = (l + \frac{3}{2})$, $l \neq 0$, supposedly true for all z_0 .] The above intuitively is not in disagreement with us. For any given (n, l) , as z_0 increases the eigenvalue curves always decrease and thus can cross the value $(\frac{1}{2}+l)$ + positive integer).

In particular, Ref. 16 derives¹⁸

$$
\epsilon = l + \frac{5}{2} \text{ at } z_0 = 1/(l+2)^{1/2} , \qquad (13)
$$

and

$$
\epsilon = l + \frac{7}{2} \text{ at } z_0 = \left(\frac{(4l + 9)}{(l + 2)(l + 3)}\right)^{1/2} \tag{14}
$$

Table I shows the $l = 0, 1, 2,$ and 3 special-case eigenvalues of Eqs. (13) and (14) that are contained in and agree with Figs. 1 and 2.

Finally, this brings up the¹⁹ quantum "folk theorem,"
that a confined oscillator has $\frac{1}{2}$ quantum per degree of freedom as a ground-state energy. For the $z_0 = 0$ threedimensional harmonic oscillator, this is correct. But as z_0 gets larger, the ground-state energy goes down to $\frac{1}{2}$ (as Figs. 1 and 2 show) for all l. Further, if we consider the N-space-dimensional rotating harmonic oscillator, the same phenomena will occur, with the even higher ground-state energy of $N/2$ also going down to $\frac{1}{2}$.
What is going on? Descriptively, in the one-dimensional

problem, the ground-state Gaussian is large, approximately between some $-x_T$ and $+x_T$. In the three-dimensional case, the boundary condition at the origin means that the

TABLE I. Special-case exact eigenvalues of Eqs. (13) and (14) that are contained in our Figs. 1 and 2.

	ϵ	z_0
$\bf{0}$	$rac{5}{2}$	$(\frac{1}{2})^{1/2} = 0.707$
$\mathbf{0}$	$rac{7}{2}$	$(\frac{3}{2})^{1/2}$ = 1.225
1	$rac{7}{2}$	$(\frac{1}{3})^{1/2} = 0.577$
1	$\frac{9}{2}$	$(\frac{13}{12})^{1/2} = 1.041$
$\overline{2}$	$\frac{9}{2}$	$\frac{1}{2} = 0.5$
$\overline{2}$	$\frac{11}{2}$	$(\frac{17}{20})^{1/2}$ = 0.922
3	$\frac{11}{2}$	$(\frac{1}{5})^{1/2} = 0.447$

wave function must go to zero at the origin (this is like the first excited state in the one-dimensional problem) so that the ground-state energy is raised to $\frac{3}{2}$. However, as z_0 gets large, on the left the wave function does not have to go to zero near z_0 . Thus the distance in the radial direction over which the wave function can be large becomes effectively that of a one-dimensional problem (say, from $-x_T+z_0$ to $+x_T+z_0$). This allows the ground-state energy to approach that of a real one-dimensional oscillator.

Physically, one must realize that the above amounts to describing the results of analyzing the expectation values of the kinetic- and potential-energy terms of the Hamiltonian, minimizing the energy with respect to an $x-p$ uncertainty limit. This is the type of analysis that Messiah²⁰ applies to the hydrogen atom, and which must be done before a "theorem" is applied to any particular potential. To show this we repeat Messiah's analysis²⁰ for this system. (Note that all the arguments used to show the incorrectness of this "theorem" could be realized from any three-dimensional potential with many bound states and a minimum away from the origin, such as the Morse potential.)

Putting units back in, for $r_0 = 0$, the lowest state has $l = 0$. First consider the x direction. The average of the potential energy in the x direction around some classical turning points $\pm x_{\overline{I}}$ is $m\omega^2 x_{\overline{I}}^2/6$. If we take $\Delta x=x_{\overline{I}}$ in the x-p uncertainty relation, this yields a lower bound for Δp of $\hbar /2x_T$.

Take the kinetic energy as $(\Delta p)^2/2m = \hbar^2/(4mx_f^2)$. The sum of this kinetic energy and the average potential energy has a minimum at $x_T^4 = 3\hbar^2/(4m^2\omega^2)$, giving a value for the total minimum energy in this degree of freedom of

$$
E_x = \frac{1}{2}\hbar\omega 3^{-1/2} \quad . \tag{15}
$$

(This answer is not precise, as it need not be.) However, now the total ground-state energy is predicted to be three times E_x , from all the x, y, and z degrees of freedom.

Contrary to this, in the large r_0 limit, the above argument goes through for one degree of freedom (the radial degree), but the angular coordinates give a vanishingly small number. To see this, consider the θ coordinate. (We here ignore the complications of using the rigorous angular operators, 21 to make the arguments simpler.) At the position r_0 , the potential energy is zero. The angle variable can vary over π . Taking this as $\Delta\theta$, the $p_{\theta} - \theta$ uncertainty relation says that $\Delta p_{\theta} \geq \hbar/(2\pi)$. Using Δp_{θ} as p_{θ} in the kinetic, and hence the total, energy gives 22

$$
E_{\theta} = \hbar \omega / (8 \pi^2 z_0^2) \quad , \tag{16}
$$

which is vanishingly small as r_0 gets larger.

This work was supported by the United States Department of Energy.

- 'In 1886, one year after Balmer (Ref. 2) had empirically determined his form of the hydrogen spectra, Deslandres did the same for the diatomic molecule (Ref. 3). Later, in his fundamental oldquantum-theory papers, Bohr was able to solve the hydrogen atom (Ref. 4), but he could not obtain the Deslandres terms for the diatomic molecule (Ref. 5). That remained for the last paper of Schwarzschild (Ref. 6). Finally, although most people known that the first of Schrodinger's basic papers on quantum mechanics solved the hydrogen atom (Ref. 7), few know that in his second paper, he approximately solved the rotating harmonic-oscillator problem for the diatomic molecule (Ref. 8). A more detailed historical account will appear elsewhere.
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- 7 We found it convenient to derive and use a nonstandard representation of the parabolic cylinder function $D_{\mu}(y\sqrt{2})$ to calculate the zeros of $D_{\mu}(-z_0\sqrt{2})$ as a function of z_0 . This representation is

$$
2^{\mu/2} \exp(z^2/2) \sum_{n=0}^{\infty} [(-z^2)^n/\Gamma(n+1)]
$$

×[cos($\pi \mu/2$) $\Gamma(n + \frac{1}{2} + \mu/2)/\Gamma(n + \frac{1}{2})$
+sin($\pi \mu/2$) $\Gamma(n + 1 + \mu/2)/\Gamma(n + \frac{3}{2})$].

- 8 Our Eq. (13) rectifies a crucial misprint three lines below Eq. (23) bur Eq. (13) rectifies a crucial misprint three lines below Eq. (23) of Ref. 16. There, " $(l+2)/2$ " was written as "1/2($l+2$)," with heir $\alpha = 1/2z_0^2$ and $\lambda = \epsilon - \frac{1}{2}$ in our notation. We thank J. P. Ader for pointing this out to us.
- ^{19"}Folk theorem" is a term which mathematical physicists like to use to describe widely known results which may, however, be incorrect. A definition perhaps can be taken from a quote at the end of the preface to R. F. Streater and A. S. Wightman, PCT, Spin & Statistics, & All That (Benjamin, New York, 1964). "... we have eliminated all theorems whose proofs are nonexistent."
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