

S, P, and D states of two-electron ions via Z-dependent perturbation theory

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A perturbation study of singly excited *S, P*, and *D* states of two-electron atoms has been made. Perturbation expansion coefficients for the nonrelativistic energies are computed and estimates of the relativistic corrections to low orders in *Z* are also obtained. A simple one-electron Lamb-shift correction is then combined with these results to yield theoretical transition energies in excellent agreement with experimental data for HeI through FeXXV.

I. INTRODUCTION

In a recent application of *Z*-dependent perturbation theory,¹ the authors made use of two-electron, first-order wave functions to obtain accurate transition energies for several *S, P*, and *D* states of three-electron ions. One aspect of the work was a simplified procedure for obtaining relativistic corrections to the energy.² The accuracy of these transition energies indicated that this approach to the relativistic corrections might be usefully applied to the corresponding two-electron states. In the latter case, third and higher-order nonrelativistic energy coefficients could be more easily computed, rather than extracted from the experimental data for the first few members of an isoelectronic sequence, as was done in Refs. 1 and 2. Further, very accurate variational calculations for some of these two-electron states are available^{3–6}; in particular, relativistic corrections from these calculations^{3,4} permit an evaluation of the accuracy to be expected from the simple approach used here.

For the present paper, the accuracy of the two-electron, first-order wave functions has been improved and the computations extended to higher order. In addition to the relativistic corrections, estimates of the Lamb shift are included. These results are then combined to yield theoretical estimates of transition energies for singlet and triplet *NS, NP*, and *ND* states through *N*=9. Finally, these theoretical values are compared with the available experimental data for HeI through FeXXV. The theoretical transition energies obtained form a consistent set of data for all states of the He isoelectronic sequence considered here.

II. METHOD**A. The nonrelativistic energy**

The Hamiltonian for a nonrelativistic two-electron atom in charge-scaled atomic units is

$$H = H_0 + \frac{1}{Z} H_1 , \quad (1)$$

where

$$H_0 = -\frac{1}{2}(\nabla_1^2 + \nabla_2^2) - \frac{1}{r_1} - \frac{1}{r_2} , \quad (2)$$

$$H_1 = \frac{1}{r_{12}} \quad (3)$$

are the unperturbed Hamiltonian and perturbation, respectively. The wave function and energy are then written as

$$\Psi = \sum_{n=0} Z^{-n} \psi_n , \quad (4)$$

$$E = \sum_{n=0} Z^{2-n} \epsilon_n . \quad (5)$$

The ψ_n are obtained by a variational-perturbation procedure⁷ using trial functions of the form

$$\begin{aligned} \psi_n^{NL} = & (1 \pm P_{12})(A + r_{12}B) \\ & \times e^{-k_n(r_1 + r_2/N)} P_L(\cos\theta_2) , \end{aligned} \quad (6)$$

with *A* and *B* both of the form

$$\sum_{abl} C_{abl}^{(n)} r_1^a + l r_2^b + l + L P_l(\cos\theta_{12}) . \quad (7)$$

For the calculations reported here the expansions for *A* usually contain about 200 terms, while those

for B contain 50 or fewer terms; the expansion for B contains all terms such that $a+b+2l \leq 6$ or 4 while A contains terms with $a+b+2l \leq 12$, but emphasizing those terms with higher powers of r_2 (i.e., $b > a$). Note that for D states this form of the trial function includes only $(s-d)$ -type terms—no $(p-p)$ terms. Also included in the trial functions for these excited states are the zero-order functions for less excited states of the same symmetry. With the inclusion of these terms the optimized ψ_n yield a rigorous upper bound⁸ for ϵ_{2n} and an estimate of ϵ_{2n+1} .

The nonlinear parameter k_n was optimized for ψ_1 only, and then the same basis was used for all the higher-order wave functions. Although less accurate than individually optimizing the k_n ,⁹ this procedure yields higher-order results with very little additional computational effort. Thus the order to which each state was calculated was limited only by the increasing numerical inaccuracy introduced by this simpler approach. It should be noted that this can be a severe limitation, and more accurate results obtained without such restrictions were used where available.^{8(c)}

Since the final results are presented in terms of transition energies to the 1^1S ground state of the ion, the nonrelativistic contribution to the transition energy is written as

$$T_{nr} = \sum_{n=0}^{n_f} e_n Z^{2-n}, \quad (8)$$

where $e_n = \epsilon_n^{NL} - \epsilon_n^{1^1S}$, and the ground-state expansion coefficients are taken from the work of Aashamar.¹⁰ The truncation point n_f , was determined by the accuracy of the excited-state expansion coefficients.

B. Relativistic corrections

The approach followed here was identical to that of Ref. 1. The relativistic correction is given by the Breit formula¹¹

$$E_{rel} = \alpha^2 \left[-\frac{1}{4} \langle p_1^4 \rangle + \pi Z \langle \delta(\vec{r}_1) \rangle + \pi \langle \delta(\vec{r}_{12}) \rangle \right], \quad (9)$$

with α the fine-structure constant. Note that no spin-orbit or spin-spin operators are included. The triplet states are thus to be compared to a multiplet average of the experimental fine-structure levels, whenever those are available. The first two terms of Eq. (9) are evaluated to first order in Z , the third

term to zero order in Z , giving the first two terms of the Z expansion of E_{rel} ,

$$E_{rel} = \alpha^2 (Z^4 b_0 + Z^3 b_1). \quad (10)$$

However the actual expression used is¹²

$$E_{rel} = \alpha^2 (Z - \sigma)^4 b_0, \quad (11)$$

which, with $\sigma = -b_1/(4b_0)$, reproduces Eq. (10) through Z^3 while providing some contributions of order Z^2 and beyond. In the above,

$$b_0 = -\frac{1}{8} - \frac{(8N - 6L - 3)}{8N^4(2L + 1)} + \frac{\delta_{L,0}}{2N^3} \quad (12)$$

and the b_1 are listed in Table V of Ref. 1. However, since the final results are expressed as transition energies, the contribution of the Breit formula was given by

$$T_{rel} = \alpha^2 (Z - \sigma)^4 b_0^{NL} - \alpha^2 \sum_{n=0} b_n^{1^1S} Z^{4-n}, \quad (13)$$

where the b_0^{NL} in Eq. (13) and in the expression for σ differs from that of Eq. (12) by the elimination of the term $-\frac{1}{8}$. Similarly, the $b_0^{1^1S}$ used in Eq. (13) was $-\frac{1}{8}$ rather than the correct value of $-\frac{1}{4}$. This procedure gave a slight improvement in T_{rel} . The higher-order $b_n^{1^1S}$ are also taken from the work of Aashamar.¹⁰

C. Lamb shift

The Lamb-shift correction is of order Z^4 and was neglected in Ref. 1 where the nonrelativistic energy was sufficiently inaccurate to make its inclusion unnecessary. For the present two-electron study, however, inclusion of the Lamb shift is essential, since for many of the excited states the nonrelativistic energies are well converged. In obtaining estimates of the Lamb shift a simple one-electron approximation was used,

$$E_L = E_L(1S, Z) + E_L(NL, Z^*) \quad (14)$$

with

$$E_L(NS, Z) = \frac{Z^4}{N^3} \left[-2 \ln(\alpha Z) - \ln(K_{N,0}) + \frac{19}{30} \right] \quad (15)$$

for S states and

$$E_L(NL, Z) = -\frac{Z^4}{N^3} \ln(K_{N,L}) \quad (16)$$

for P and D states. $K_{N,L}$ is the average excitation potential¹³ of the one-electron ion and $Z^* = Z - 1 + (3 - L)/N^2$ represents an *ad hoc* screening for the excited electron. No screening was used for the $1S$ electron so that the contribution of E_L to the transition energy would go to the right limit for large N . The ground-state Lamb shift necessary for the transition energy was taken from Aashamar.¹⁰

D. Mass polarization

In the present calculations, the mass-polarization correction for the excited state is ignored, and only the contribution of the ground state to the transition energy is included. The latter is almost constant with increasing Z with a value of $\sim 7 \text{ cm}^{-1}$ for $Z=6$ through 10 (Ref. 10). In contrast, the mass-polarization correction for S (Ref. 3) and D (Ref. 14) states is quite small and decreases rapidly as N increases. For these states, then, the ground-state correction is sufficient and, indeed, only makes a noticeable contribution for low- Z ions. This is not the case for the P states, however, where the absolute value of the polarization correction for the excited states is increasing with Z although still decreasing rapidly with N (Ref. 3).

III. RESULTS

The nonrelativistic energy expansion coefficients for the states considered here are listed in Tables I–VI. The higher-order coefficients listed are based on 100-term expansions for the perturbation wave functions using high-precision (28-decimal digit) arithmetic since, given the number of states considered here, calculations with the 200+ terms used for the first-order wave functions would have been prohibitive in terms of computer time. For the 2^3S and $2^{1,3}P$ states, the higher-order coefficients listed are taken from the earlier work of Sanders and Scherr,^{8(c)} where the k_n for the higher-order wave functions are individually optimized. Values of these coefficients from other theoretical studies of these states are listed for comparison where these are available.

Tables VII and VIII list values of $\alpha^{-2}T_{\text{rel}}$ given by Eq. (13) for those S and P states studied by Accad, *et al.*⁴ Comparison with the latter authors' ac-

curate calculations permits an evaluation of the accuracy obtained with the approximation used here for the relativistic correction.

Finally, Tables IX–XIV present the transition energies for all S , P , and D states considered here for all ions from He I through Fe XXV inclusive, together with accurate experimental values for these transition energies. In many cases these experimental values are taken from recent critical compilations of the available data by the National Bureau of Standards Atomic Energy Levels Data Center, and thus include some theoretical values. In the tables, such theoretical values are indicated by parenthesis. For the most highly ionized S and P states, the listed values are almost exclusively theoretical values taken from the work of Ermolaev and Jones.¹⁵

IV. DISCUSSION

A. The nonrelativistic energy

For Z -dependent perturbation theory to yield accurate nonrelativistic energies, the ϵ_n must be known with sufficient accuracy and through a sufficiently high order. In particular, ϵ_2 must be known to high accuracy—indeed, for large Z , accurate values of ϵ_2 and ϵ_3 are enough to yield acceptable values of the nonrelativistic energy. Generally, the ϵ_n computed here and presented in Tables I–VI fulfill these requirements. Note particularly the agreement with the results of Blanchard,¹⁶ where the ϵ_n for the S and P states are extracted from the accurate variational calculations of Accad *et al.*⁴ The agreement is particularly striking for the ϵ_2 , but is also remarkably good for the higher-order coefficients, which have been computed here with more limited expansion sets for the ψ_n . Also included in the tables are the results of Aashamar *et al.*^{17,18} for the $N=2$ and $3S$ and P states and the results of Blanchard and Drake⁶ for the $N=3$ for $4D$ states. These results are obtained via variational-perturbation wave functions of a complexity similar to that of the present ψ_n . Indeed the $2, 3^3S \epsilon_n, n \geq 4$ of Ref. 17 are somewhat superior and the $3D \epsilon_n$ of Ref. 6 clearly superior to those of the present study. The $4D$ results of Ref. 6 appear to be of similar accuracy to those of the present paper. Note that, generally, for the present results the D -state calculations are not of as high a quality as the S - and P -state calculations. Finally, the tables also include some ϵ_2 computed nonvariationally by Ivanova *et al.*¹⁹

TABLE I. The ϵ_n in a.u. for the N^1S states.^a

n	2^1S	3^1S	4^1S	5^1S	6^1S	7^1S	8^1S	9^1S
2	-0.114 510 11	-0.049 107 87	-0.028 163 33	-0.018 338 00	-0.012 900 44	-0.009 569 86	-0.007 379 95	-0.005 863 40
	-0.114 510 15 ^b	-0.049 061 36 ^c	-0.028 156 ^b	-0.018 334 ^b				
	-0.114 487 03 ^c	-0.049 16 ^d	-0.028 27 ^d	-0.018 50 ^d				
	-0.114 50 ^d							
3	0.009 322 78	-0.001 268 78	-0.000 750 20	-0.000 394 61	-0.000 177 94	-0.000 250 93	-0.000 259 30	-0.000 121 82
	0.009 328 0 ^b	-0.000 944 33 ^c	-0.000 84 ^b	-0.000 47 ^b				
	0.009 224 00 ^c							
4	-0.001 283 61	-0.001 448 29	-0.000 845 18	0.000 229 50	-0.000 227 54	-0.000 136 09	-0.000 164 76	0.001 941 98
	-0.001 286 ^b	-0.001 5 ^b	-0.000 8 ^b	-0.000 5 ^b				
	-0.001 084 86 ^c	0.000 249 81 ^c						
5	0.006 202 11	0.005 433 45	0.004 443 93	0.007 925 95	0.001 452 99	0.000 368 93	0.000 686 80	
	0.006 2 ^b	0.005 ^b	0.002 ^b	0.001 ^b				
	0.005 978 67 ^c	0.010 601 90 ^c						
6	-0.001 454 88	0.003 330 65						
	-0.001 5 ^b	0.013 939 79 ^c						
	-0.001 327 80 ^c							
7	-0.003 485 32	0.005 985 52						
	-0.005 ^b	0.018 265 91 ^c						
	-0.003 742 10 ^c							
8	-0.000 911 64							
	-0.001 999 19 ^c							

^aThe first line in each entry of the table is from the present study. Additional lines are from earlier theoretical studies.^bP. Blanchard, Ref. 16.^cK. Aashamar, G. Lyslo, and J. Midtdal, Ref. 18.^dA. N. Ivanova, U. I. Safronova, and V. N. Kharitonova, Ref. 19.

TABLE II. The ϵ_n in a.u. for the N^3S states.^a

<i>n</i>	2^3S	3^3S	4^3S	5^3S	6^3S	7^3S	8^3S	9^3S
2	-0.04740930	-0.03231754	-0.02130158	-0.01484695	-0.01088005	-0.00829604	-0.00652712	-0.00526595
	-0.04740930 ^b	-0.03231754 ^c	-0.02130154 ^d	-0.01484686 ^d				
	-0.04741 ^e	-0.03234 ^e	-0.02137 ^e	-0.01494 ^e				
3	-0.00487228	-0.00213684	-0.00110711	-0.00063917	-0.00040021	-0.00026502	-0.00019180	-0.00005812
	-0.00487228 ^b	-0.00213686 ^c						
4	-0.00345757	-0.00151798	-0.00072011	-0.00039195	-0.00023351	-0.00014590	-0.00007712	0.00171877
	-0.00345780 ^b	-0.00151885 ^c	-0.000721 ^d	-0.000394 ^d				
5	-0.00203006	-0.00078181	-0.00036834	-0.00020055	-0.00012389	-0.00008462	0.00038706	
	-0.00202985 ^b	-0.00078672 ^c	-0.000374 ^d	-0.00020 ^d				
6	-0.00128709	-0.00043620	-0.00023556	-0.00010530	-0.00005267	0.00004398		
	-0.00128713 ^b	-0.00046563 ^c	-0.00020 ^d	-0.0001 ^d				
7	-0.00087145	-0.00012939		-0.00005512	-0.00002139			
	-0.00087142 ^b	-0.00027207 ^c						
8	-0.00061790	0.00046822		-0.00000518	0.00018622			
	-0.00061823 ^b	-0.00017508 ^c						
9	-0.00045412			-0.000033566				
	-0.00045443 ^b							

^aThe first line in each entry of the table is from the present study. Additional lines are from earlier theoretical studies.

^bK. Aashamar, J. Middal, and G. Lyslo, Ref. 17(a).

^cK. Aashamar, J. Middal, and G. Lyslo, Ref. 17(b).

^dP. Blanchard, Ref. 16.

^eA. N. Ivanova, U. I. Safronova, and V. N. Kharitonova, Ref. 19.

TABLE III. The ϵ_n in a.u. for the N^1P states.^a

<i>n</i>	2^1P	3^1P	4^1P	5^1P	6^1P	7^1P	8^1P	9^1P
2	-0.157 028 66	-0.060 752 36	-0.032 981 98	-0.020 788 93	-0.014 315 01	-0.010 457 05	-0.007 970 10	-0.006 268 64
	-0.157 028 59 ^b	-0.060 752 9 ^b	-0.032 977 ^b	-0.020 787 ^b				
	-0.157 021 50 ^c	-0.060 725 67 ^c						
	-0.157 03 ^d	-0.063 91 ^d						
3	0.026 106 28	-0.000 275 21	-0.000 654 10	-0.000 425 04	-0.000 426 51	-0.000 491 20	-0.000 417 51	-0.000 371 65
	0.026 059 59 ^c	-0.000 257 ^b	-0.000 73 ^b	-0.000 48 ^b				
		-0.000 094 28 ^c						
4	0.005 788 76	0.004 219 06	0.001 549 06	0.004 503 35	0.002 887 10	0.002 070 01	0.000 402 37	0.000 142 70
	0.005 914 90 ^c	0.003 77 ^b	0.001 5 ^b	0.000 7 ^b				
		0.005 003 55 ^c						
5	-0.005 041 31	0.005 096 54	0.005 644 46	-0.000 382 40	0.019 114 53	0.014 991 66	0.000 857 40	0.000 809 83
	-0.005 242 15 ^c	0.006 9 ^b	0.003 ^b	0.002 ^b				
		0.008 791 65 ^c						
6	-0.007 066 59	-0.005 012 94		-0.003 698 86				
	-0.006 889 44 ^c	-0.010 ^b						
		0.004 363 84 ^c						
7	-0.001 259 46	-0.004 680 79		-0.001 162 50				
	-0.001 333 82 ^c	0.002 ^b						
		0.016 433 51 ^c						
8	0.003 405 12							
	0.003 424 61 ^c							
9	0.001 920 93							
	0.001 953 97 ^c							

^aThe first line in each entry of the table is from the present study. Additional lines are from earlier theoretical studies.^bP. Blanchard, Ref. 16.^cK. Aashamar, G. Lyslo, and J. Midtdal, Ref. 18.^dA. N. Ivanova, U. I. Safronova, and V. N. Kharitonova, Ref. 19.

TABLE IV. The ϵ_n in a.u. for the N^3P states.^a

n	2^3P	3^3P	4^3P	5^3P	6^3P	7^3P	8^3P	9^3P
2	-0.07299898	-0.04185634	-0.02564455	-0.01716325	-0.01225572	-0.00917752	-0.00712363	-0.00566623
	-0.07299897 ^b	-0.0418563 ^b	-0.0256444 ^b	-0.0171631 ^b				
	-0.07299675 ^c	-0.04183663 ^c						
	-0.07299 ^d	-0.04487 ^d						
3	-0.01658530	-0.00335117	-0.00132102	-0.00066706	-0.00042441	-0.00030236	-0.00023989	-0.00020986
	-0.01659432 ^c	-0.003352 ^b	-0.001323 ^b	-0.000670 ^b				
		-0.00321765 ^c						
4	-0.01035309	-0.00271013	-0.00101074	-0.00023399	0.00016845	0.00080699	0.00053394	0.00007408
	-0.01033961 ^c	-0.00272 ^b	-0.00103 ^b	-0.00051 ^b				
		-0.00211062 ^c						
5	-0.00542484	-0.00149737	-0.00026171	0.00183929	0.00309077	0.00968105	0.00543481	0.00288299
	-0.00543613 ^c	-0.0015 ^b	-0.0006 ^b	-0.0003 ^b				
		0.00055089 ^c						
6	-0.00202080	-0.00073841	0.00199945					
	-0.00201702 ^c	-0.0008 ^b	-0.0004 ^b					
		0.00473645 ^c						
7	0.00023827	-0.00028675						
	0.00023649 ^c	0.01185275 ^c						
8	0.00161032	0.00484730						
	0.00161229 ^c	0.02182211 ^c						
9	0.00226962	0.00294620						
	0.00227854 ^c	0.03026480 ^c						

^aThe first line in each entry of the table is from the present study. Additional lines are from earlier theoretical studies.^bP. Blanchard, Ref. 16.^cK. Aashamar, G. Lyslo, and J. Midtdal, Ref. 18.^dA. N. Ivanova, U. I. Safronova, and V. N. Kharitonova, Ref. 19.

TABLE V. The ϵ_n in a.u. for the N^1D states.^a

<i>n</i>	3^1D	4^1D	5^1D	6^1D	7^1D	8^1D	9^1D
2	-0.057 387 77 -0.057 482 2 ^b -0.057 31 ^c	-0.032 110 73 -0.032 121 6 ^b	-0.020 455 82	-0.014 155 71	-0.010 371 24	-0.007 921 53	-0.006 244 52
3	0.005 538 59 0.005 974 4 ^b	0.002 486 95 0.002 444 71 ^b	0.001 308 43	0.000 852 70	0.000 674 28	0.000 083 74	0.000 032 03
4	-0.008 121 58 ^b	-0.002 882 32 -0.002 636 34 ^b	-0.001 224 59	0.000 190 03	0.001 559 31	0.000 145 14	0.000 057 84
5	0.005 564 48 ^b	0.002 445 71 0.000 644 30 ^b	0.002 327 04	0.006 076 91		-0.000 234 54	-0.000 037 85
6	-0.002 318 28 ^b	0.002 134 41 0.001 025 81 ^b					
7	0.000 816 29 ^b						
8	0.000 145 93 ^b						
9	-0.000 664 68 ^b						

^aThe first line in each entry of the table is from the present study. Additional lines are from earlier theoretical studies.^bP. Blanchard and G. W. F. Drake, Ref. 6.^cA. N. Ivanova, U. I. Safronova, and V. N. Kharitonova, Ref. 19.TABLE VI. The ϵ_n in a.u. for the N^3D states.^a

<i>n</i>	3^3D	4^3D	5^3D	6^3D	7^3D	8^3D	9^3D
2	-0.054 615 28 -0.054 619 3 ^b -0.054 59 ^c	-0.030 684 33 -0.030 671 37 ^b	-0.019 675 45	-0.013 691 03	-0.010 075 30	-0.007 723 88	-0.006 108 03
3	-0.000 724 88 -0.000 731 80 ^b	-0.000 567 16 -0.000 666 06 ^b	-0.000 337 24	-0.000 199 89	-0.000 105 24	-0.000 173 68	-0.000 138 33
4	0.000 073 74 ^b	0.000 141 74 0.000 439 07 ^b	0.000 087 83	0.000 090 04	0.000 401 74	0.000 169 72	0.000 155 23
5	-0.000 087 86 ^b	0.000 088 86 -0.000 343 91 ^b	0.000 085 30	0.000 340 35		0.000 001 30	0.000 178 41
6	0.000 177 09 ^b	-0.000 043 03 -0.000 392 42 ^b	0.000 102 26				
7	0.000 069 71 ^b	0.000 095 09 0.001 828 46 ^b					
8	-0.000 294 96 ^b						
9	0.000 098 88 ^b						

^aThe first line in each entry of the table is from the present study. Additional lines are from earlier theoretical studies.^bP. Blanchard and G. W. F. Drake, Ref. 6.^cA. N. Ivanova, U. I. Safronova, and V. N. Kharitonova, Ref. 19.

TABLE VII. T_{rel}/α^2 in a.u. for some S states.^a

Z	2^1S	3^1S	4^1S	5^1S	2^3S	3^3S	4^3S	5^3S
2	1.448 69	1.464 65	1.470 73	1.473 21	1.274 31	1.422 36	1.454 68	1.465 52
	1.441 82	1.464 11	1.470 68	1.473 20	1.311 64	1.431 02	1.457 82	1.466 97
3	3.609 75	3.933 37	4.041 86	4.085 20	2.590 90	3.672 68	3.940 04	4.035 51
	3.578 02	3.935 58	4.044 40	4.086 83	2.690 42	3.697 34	3.949 26	4.039 83
4	10.364 61	12.130 09	12.699 34	12.925 06	7.273 35	11.324 71	12.381 60	12.769 04
	10.294 30	12.142 04	12.708 51	12.930 45	7.463 1	11.373 01	12.399 88	12.777 65
5	27.066 77	32.855 03	34.686 35	35.409 90	20.119 91	31.028 81	33.962 23	35.053 18
	26.945 4	32.884 23	34.706 30	35.421 01	20.428 0	31.108 39	33.992 55	35.067 53
6	61.133 09	75.575 88	80.094 59	81.876 03	47.992 17	72.102 87	78.713 31	81.194 29
	60.948 4	75.629 92	80.129 39	81.894 7	48.446 6	72.221 40	78.758 65	81.215 81
7	122.043 18	152.427 27	161.863 65	165.578 46	99.814 48	146.531 69	159.514 11	164.417 33
	121.782 9	152.513 67	161.917 28	165.606 3	100.443 4	146.696 82	159.577 46	164.447 48
8	221.339 22	278.210 59	295.780 88	302.690 65	186.573 78	268.966 80	292.091 64	300.865 77
	220.991 4	278.336 9	295.857 3	302.729 3	187.405 3	269.186 23	292.176 01	300.906 01
9	372.625 95	470.393 92	500.481 30	512.304 51	321.319 54	456.726 48	495.020 63	509.601 53
	372.178 4	470.567 5	500.584 3	512.355 3	322.381 7	457.007 89	495.129 0	509.653 32
10	591.570 63	749.112 03	797.447 62	816.430 37	519.163 75	729.795 65	789.723 46	812.604 94
	591.011 2	749.340 3	797.581 1	816.494 6	520.484 7	730.146 75	789.858 9	812.669 75

^aFor each entry, the first line is the result obtained in the present paper. The second line is from Accad *et al.*, Ref. 4.

B. The relativistic correction

The relativistic correction as applied here via Eq. (11) is in error by terms of order Z^2 . Thus, the relative error should decrease as Z^{-2} . For low Z , the total relativistic correction itself is rather small, $\sim 100 \text{ cm}^{-1}$ for $Z=4$, the error being a few cm^{-1} , at most. In contrast, the nonrelativistic transition energy for $Z=4$ is $\sim 10^6 \text{ cm}^{-1}$ with an error of at most a few cm^{-1} . For larger values of Z , then, the accuracy of the nonrelativistic transition energy will increase and the error in the relativistic correction will determine the overall accuracy of the calculation. The present relativistic correction is compared with the earlier calculations of Accad *et al.*⁴ in Tables VII and VIII. Unfortunately, the tables indicate that the relative difference between the present results and those of Ref. 4 decreases roughly as Z^{-1} for the 3S states, or remains roughly constant for the 1S and $^{1,3}P$ states. This holds true for all states ($N=2$ to 5) of a particular symmetry, although the relative difference drops by about a fac-

tor of 10 from $N=2$ to $N=5$. It appears that the variational data of Ref. 4 have an error which is growing more rapidly than that of the Z -dependent data, i.e., more rapidly than Z^2 . Thus, the accuracy of the present method is indicated only by a comparison of the $Z=2$ data, for example, where the variational calculations are surely of high accuracy. From that point the relative error must necessarily be decreasing as Z^{-2} . It is also not possible to make useful comparisons with the high- Z extrapolations of Ermolaev and Jones,¹⁵ since these are extrapolations based on the calculations of Accad *et al.*,³ and display similar behavior.

C. Lamb shift

The Lamb-shift contribution to the transition energies must be included in the present calculations since, even for low values of Z , this contribution is of greater magnitude than the error introduced by the approximations leading to Eq. (11) for the rela-

TABLE VIII. T_{rel}/α^2 in a.u. for some P states.^a

Z	2^1P	3^1P	4^1P	5^1P	2^3P	3^3P	4^3P	5^3P
2	1.463 84	1.470 48	1.473 26	1.474 49	1.450 03	1.464 50	1.470 62	1.473 17
	1.436 09	1.461 27	1.469 16	1.472 2	1.503 04	1.480 96	1.477 62	1.476 79
3	3.895 01	4.029 64	4.083 10	4.106 09	3.785 18	3.983 25	4.062 58	4.095 89
	3.653 19	3.955 01	4.051 35	4.090 33	4.005 68	4.045 33	4.088 07	4.108 29
4	11.886 79	12.622 01	12.909 00	13.031 61	11.516 52	12.466 88	12.840 39	12.997 51
	11.079 58	12.382 42	12.808 98	12.983 1	11.775 20	12.526 64	12.863 1	13.007 1
5	32.004 56	34.416 53	35.350 17	35.747 84	31.127 20	34.050 41	35.188 29	35.667 36
	30.129 96	33.874 20	35.125 9	35.639 6	30.803 71	33.910 96	35.123 8	35.630 1
6	73.377 05	79.397 78	81.716 86	82.702 81	71.663 73	78.684 54	81.401 53	82.546 04
	69.785 34	78.377 18	81.297 0	82.500	69.398 27	77.926 40	81.071 2	82.369
7	147.695 73	160.365 42	165.229 69	167.295 20	144.735 34	159.135 01	164.685 78	167.024 78
	141.590 25	158.653 3	164.528	166.957	138.178 99	157.042 1	163.785 5	166.552
8	269.214 65	292.934 00	302.019 53	305.874 18	264.513 88	290.982 49	301.156 93	305.445 29
	258.652 14	290.279 5	300.934	305.351	250.078 70	286.468 0	299.226	304.439
9	454.750 43	495.532 94	511.127 46	517.739 38	447.733 72	492.622 49	509.841 07	517.099 75
	440.641 1	491.647 3	509.542	516.975	420.342 78	484.155 0	506.681	515.227
10	723.682 17	789.406 48	814.504 73	825.140 80	713.691 76	785.265 35	812.674 48	824.230 74
	703.789 9	783.963 5	812.287	824.071	666.529 15	770.794 5	806.517	821.045

^aFor each entry, the first line is the result obtained in the present paper. The second line is from Accad *et al.*, Ref. 4.

TABLE IX. Transition energies in cm^{-1} for the N^1S states.^a

Z	2^1S	3^1S	4^1S	5^1S	6^1S	7^1S	8^1S	9^1S
2	166 281.0	184 882.4	191 002.5	193 642.1	195 142.2	195 970.3	196 471.8	196 822.0
	166 277.6 ^b	184 864.9 ^b	190 940.3 ^b	193 663.6 ^b	195 115.0 ^b	195 979.0 ^b	196 534.9 ^b	196 913.0 ^b
3	491 376	558 783	581 619	591 993	597 595	600 926	603 070	604 538
	(490 079) ^c	558 779 ^c	581 590 ^c	591 984 ^c	597 574 ^c	600 925 ^c		
4	981 198	1 127 721	1 177 952	1 200 950	1 213 367	1 220 808	1 225 618	1 228 912
	981 177 ^d	1 127 705 ^d	1 177 945 ^d					
5	1 635 745	1 891 817	1 980 125	2 020 679	2 042 603	2 055 769	2 064 290	2 070 127
	1 635 714 ^f	1 891 784 ^f						
6	2 455 078	2 851 192	2 988 265	3 051 318	3 085 436	3 105 946	3 119 225	3 128 324
	2 455 024 ^g	(2 851 180) ^g						
7	3 439 311	4 006 010	4 202 542	4 293 043	4 342 043	4 371 516	4 390 602	4 403 685
8	4 588 562	5 356 432	5 623 131	5 746 032	5 812 603	5 852 660	5 878 604	5 896 391
	4 587 340 ^h	5 356 420 ^h						
9	5 902 992	6 902 671	7 250 256	7 410 516	7 497 351	7 549 614	7 583 466	7 606 679
		6 903 580 ⁱ						
10	7 382 751	8 644 927	9 084 132	9 286 713	9 396 507	9 462 600	9 505 410	9 534 772

TABLE IX. (Continued.)

Z	2^1S	3^1S	4^1S	5^1S	6^1S	7^1S	8^1S	9^1S
11	9 028 074 (9 027 981) ^j	10 583 499 (10 583 431) ^j	11 125 077 (11 124 986) ^j	11 374 953 (11 374 868) ^j	11 510 402 (11 510 320) ^j	11 591 951 (11 591 874) ^j	11 644 772 (11 644 747) ^j	11 681 006 (11 680 966) ^j
12	10 839 156 (10 839 030) ^k	12 718 647 (12 718 550) ^k	13 373 369 (13 373 240) ^k	13 675 516 (13 675 400) ^k	13 839 323	13 937 954	14 001 839	14 045 668
13	12 816 278 (12 816 130) ^l	15 050 727 (15 050 610) ^l	15 829 389 (15 829 230) ^l	16 188 796 (16 188 660) ^l	16 383 666	16 501 008	16 577 011	16 629 161
14	14 959 688	17 580 065	18 493 483	18 915 145	19 143 788	19 281 474	19 370 649	19 431 846
15	17 269 724	20 307 090	21 366 110	21 855 035	22 120 165	22 279 830	22 383 233	22 454 203
16	19 746 675	23 232 183	24 447 676	25 008 882	25 313 218	25 496 498	25 615 186	25 696 658
17	22 390 939	26 355 848	27 738 719	28 377 236	28 723 505	28 932 041	29 067 073	29 159 775
18	25 202 881	29 678 559	31 239 743	31 960 615	32 351 548	32 586 982	32 739 419	32 844 081
19	28 182 847	33 200 766	34 951 232	35 759 513	36 197 846	36 461 825	36 632 728	36 750 081
20	31 331 335 (31 330 780) ^m	36 923 102 (36 922 600) ^m	38 873 859 (38 873 200) ^m	39 774 620	40 263 099	40 557 272	40 747 706	40 878 483
21	34 648 828	40 846 186	43 008 282	44 006 611	44 547 987	44 874 009	45 085 039	45 229 974
22	38 135 748 (38 135 000) ⁿ	44 970 567 (44 969 900) ⁿ	47 355 089 (47 354 200) ⁿ	48 456 087	49 053 119	49 412 647	49 645 341	49 805 169
23	41 792 626 (41 791 780) ^o	49 296 925 (49 296 140) ^o	51 915 005 (51 913 970) ^o	53 123 791	53 779 246	54 173 943	54 429 370	54 604 829
24	45 619 980	53 825 926	56 688 740	58 010 451	58 727 104	59 158 635	59 437 868	59 629 696
25	49 618 409 (49 617 300) ^p	58 558 336 (58 557 300) ^p	61 677 109 (61 675 800) ^p	63 116 902	63 897 536	64 367 573	64 671 686	64 880 624
26	53 788 464	63 494 866	66 880 873	68 443 921	69 291 331	69 801 549	70 131 618	70 358 409

^aFor each entry, the first line is the result obtained in the present paper, the second line (if present) is from experiment, except where parentheses indicate a theoretical value.

^bW. C. Martin, J. Phys. Chem. Ref. Data 2, 257 (1973).

^cC. E. Moore, *Atomic Energy Levels*, Natl. Bur. Stds. (U.S.) Circ. No. 467 (U.S. GPO, Washington D.C., 1949).

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^eM. Eiedlsberg, J. Phys. B 5, 1031 (1972).

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^hC. E. Moore, *Atomic Energy Levels*, Natl. Stds. Ref. Data Ser. (National Bureau of Standards, Washington, D.C., 1979), Vol. 3, Sec. 8.

ⁱA. M. Cantu, E. Jannitti, and G. Tondello, J. Opt. Soc. Am. 64, 699 (1974).

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^kW. C. Martin and R. Zalubas, J. Phys. Chem. Ref. Data 9, 1 (1980).

^lW. C. Martin and R. Zalubas, J. Phys. Chem. Ref. Data 8, 817 (1979).

^mJ. Sugar and C. Corliss, J. Phys. Chem. Ref. Data 8, 865 (1979).

ⁿC. Corliss and J. Sugar, J. Phys. Chem. Ref. Data 8, 1 (1979).

^oJ. Sugar and C. Corliss, J. Phys. Chem. Ref. Data 7, 1191 (1978).

^pC. Corliss and J. Sugar, J. Phys. Chem. Ref. Data 6, 1253 (1977).

TABLE X. Transition energies in cm^{-1} for the N^3S states.^a

Z	2^3S	3^3S	4^3S	5^3S	6^3S	7^3S	8^3S	9^3S
2	159 855.8 159 856.1 ^b	183 236.5 183 236.9 ^b	190 295.3 190 298.2 ^b	193 347.0 193 347.1 ^b	194 935.5 194 936.2 ^b	195 866.4 195 868.4 ^b	196 379.2 196 461.4 ^b	196 786.7 196 862.0 ^b
3	476 034 476 046 ^c	554 754 554 761 ^c	579 981 579 982 ^c	591 185 591 184 ^c	597 122 597 122 ^c	600 644 600 641 ^c	602 873	604 414
4	956 518 956 502 ^d	1 121 200 1 121 184 ^d	1 175 313 1 175 295 ^d	1 199 629 1 199 659 ^e	1 212 608	1 220 343	1 225 304	1 228 700
5	1 601 566 1 601 540 ^f	1 882 772 1 882 746 ^f	1 976 465 1 976 436 ^f	2 018 844	2 041 552	2 055 119	2 063 855	2 069 827
6	2 411 309 2 411 262 ^g	2 839 608 2 839 562 ^g	2 983 576 (2 983 541) ^g	3 048 965 (3 048 927) ^g	3 084 089 (3 084 048) ^g	3 105 109 (3 105 066) ^g	3 118 668 (3 118 635) ^g	3 127 938
7	3 385 882 3 385 890 ^h	3 991 872	4 196 818	4 290 170	4 340 398	4 370 492	4 389 922	4 403 211
8	4 525 412 4 524 640 ⁱ	5 339 729 5 338 820 ⁱ	5 616 366 5 616 100 ⁱ	5 742 636	5 810 660	5 851 449	5 877 800	5 895 830
9	5 830 061 5 830 600 ^j	6 883 388 6 883 220 ^k	7 242 446 7 242 400 ^k	7 406 594	7 495 107	7 548 214	7 582 539	7 606 031
10	7 299 976 7 300 900 ^l	8 623 048	9 075 269	9 282 263	9 393 959	9 461 009	9 504 360	9 534 037
11	8 935 388 (8 935 337) ^m	10 559 009 (10 558 946) ^m	11 115 156 (11 115 065) ^m	11 369 970 (11 369 887) ^m	11 507 550 (11 507 469) ^m	11 590 169 (11 590 091) ^m	11 643 598 (11 643 558) ^m	11 680 184 (11 680 134) ^m
12	10 736 488 10 730 000 ^l	12 691 526 (12 691 440) ⁿ	13 362 381 (13 362 260) ⁿ	13 669 997 (13 669 880) ⁿ	13 836 164	13 935 980	14 000 541	14 044 760
13	12 703 551 (12 703 460) ^o	15 020 957 (15 020 850) ^o	15 817 327 (15 817 170) ^o	16 182 737 (16 182 600) ^o	16 380 198	16 498 841	16 575 589	16 628 167
14	14 836 821 14 840 000 ^p	17 547 624	18 480 338	18 908 542	19 140 009	19 279 111	19 369 104	19 430 764
15	17 136 627	20 271 954	21 351 873	21 847 884	22 116 072	22 277 270	22 381 564	22 453 035
16	19 603 255 19 210 000 ^p	23 194 327	24 432 336	25 001 176	25 308 808	25 493 740	25 613 394	25 695 403
17	22 237 095	26 315 246	27 722 266	28 368 971	28 718 776	28 929 082	29 065 157	29 158 433
18	25 038 507 25 030 000 ^p	29 635 182	31 222 166	31 951 785	32 346 495	32 583 820	32 737 379	32 842 652
19	28 007 830	33 154 584	34 932 518	35 750 112	36 192 467	36 458 459	36 630 564	36 748 566
20	31 145 557 31 180 000 ^p	36 874 084 (36 873 600) ^q	38 853 995 (38 853 340) ^q	39 764 642	40 257 390	40 553 699	40 745 418	40 876 880
21	34 452 161	40 794 297	42 987 255	43 996 048	44 541 944	44 870 227	45 082 627	45 228 285

TABLE X. (Continued.)

Z	2^3S	3^3S	4^3S	5^3S	6^3S	7^3S	8^3S	9^3S
22	37 928 062 (37 927 500) ^r	44 915 772 (44 915 100) ^r	47 332 885 (47 332 000) ^r	48 444 933	49 046 739	49 408 654	49 642 806	49 803 394
23	41 573 783 (41 573 150) ^s	49 239 187 (49 238 420) ^s	51 891 609 (51 890 590) ^s	53 112 039	53 772 525	54 169 735	54 426 711	54 602 967
24	45 389 835	53 765 207	56 664 136	57 998 093	58 720 037	59 154 210	59 435 086	59 627 748
25	49 376 810 (49 376 000) ^t	58 494 594 (58 493 600) ^t	61 651 281 (61 650 000) ^t	63 103 929	63 890 119	64 362 928	64 668 780	64 878 589
26	53 535 254 53 604 000 ^p	63 428 060	66 853 804	68 430 325	69 283 558	69 796 681	70 128 588	70 356 286

^aFor each entry, the first line is the result obtained in the present paper, the second line (if present) is from experiment, except where parentheses indicate a theoretical value.

^bW. C. Martin, J. Phys. Chem. Ref. Data **2**, 257 (1973).

^cC. E. Moore, *Atomic Energy Levels*, Natl. Bur. Stds. (U.S.) Circ. No. 467 (U.S. GPO, Washington, D.C., 1949).

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^hC. E. Moore, *Atomic Energy Levels*, Natl. Stds. Ref. Data Ser. (National Bureau of Standards, Washington, D.C., 1971), Vol. 3, Sec. 4.

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^tC. Corliss and J. Sugar, J. Phys. Chem. Ref. Data **6**, 1253 (1977).

TABLE XI. Transition energies in cm^{-1} for the N^1P states.^a

Z	2^1P	3^1P	4^1P	5^1P	6^1P	7^1P	8^1P	9^1P
2	171 134.4 171 135.0 ^b	186 233.5 186 209.5 ^b	191 423.4 191 492.8 ^b	194 093.4 193 942.6 ^b	195 344.8 195 275.0 ^b	196 099.9 196 079.2 ^b	196 469.5 196 601.5 ^b	196 835.0 196 959.8 ^b
3	501 807 501 810 ^c	561 759 561 730 ^c	582 813 582 820 ^c	592 695 592 660 ^c	597 979 597 940 ^c	601 167 601 180 ^c	603 188 603 281 ^c	604 625 604 704 ^c
4	997 474 997 466 ^e	1 132 411 1 132 390 ^d	1 179 890 1 179 830 ^d	1 201 986 1 201 894 ^d	1 213 954 1 213 931 ^d	1 221 176 1 221 135 ^d	1 225 840	1 229 069
5	1 658 012 1 657 975 ^f	1 898 259 1 898 230 ^f	1 982 807 1 982 761 ^f	2 022 076 2 022 044 ^f	2 043 401 2 043 360 ^f	2 056 270	2 064 610	2 070 352
6	2 483 443 2 483 371 ^g	2 859 416 2 859 375 ^g	2 991 698 2 991 710 ^g	3 053 088 3 053 044 ^g	3 086 452 3 086 439 ^g	3 106 584 (3 106 541) ^g	3 119 643 (3 119 619) ^g	3 128 618

TABLE XI. (*Continued.*)

Z	2^1P	3^1P	4^1P	5^1P	6^1P	7^1P	8^1P	9^1P
7	3 473 877 3 473 790 ^h	4 016 042 4 016 390 ^h	4 206 735 4 206 810 ^h	4 295 195 4 296 100 ⁱ	4 343 280 4 343 300 ⁱ	4 372 293 4 372 500 ⁱ	4 391 118 4 392 700 ⁱ	4 404 047
8	4 629 450 4 629 200 ^j	5 368 306 5 368 550 ^j	5 628 097 5 628 100 ^j	5 748 574 5 748 450 ^j	5 814 067 5 813 950 ^j	5 853 580 5 852 740 ^j	5 879 220	5 896 823
9	5 950 352 5 950 900 ^k	6 916 431 6 915 600 ^k	7 256 013 7 255 800 ^j	7 413 459 7 411 800 ⁱ	7 499 047 7 498 500 ⁱ	7 550 680 7 546 000 ⁱ	7 584 184 7 587 200 ⁱ	7 607 183
10	7 436 768 7 437 700 ^l	8 660 626 8 660 300 ^j	9 090 701 9 071 900 ^l	9 290 070 9 313 600 ^l	9 398 441	9 463 815	9 506 233	9 535 350
11	9 088 979 9 087 600 ^m	10 601 207 10 610 000 ^m	11 132 489 (11 132 393) ⁿ	11 378 737 (11 378 646) ⁿ	11 512 584 (11 512 505) ⁿ	11 593 322 (11 593 248) ⁿ	11 645 705 (11 645 667) ⁿ	11 681 661 (11 681 612) ⁿ
12	10 907 233 10 907 000 ^m	12 738 446 12 738 000 ^m	13 381 658 13 381 000 ^m	13 679 748 13 681 000 ^m	13 841 763 13 840 000 ^m	13 939 488 (13 939 380) ^o	14 002 886 (14 003 830) ^o	14 046 404 (14 046 330) ^o
13	12 891 870 12 891 900 ^m	15 072 721 15 072 700 ^m	15 838 598 15 838 600 ^m	16 193 496 16 196 000 ^m	16 386 377 16 392 000 ^m	16 502 713 16 504 000 ^m	16 578 180 16 585 000 ^m	16 629 982 16 640 000 ^m
14	15 043 201 15 040 000 ^m	17 604 374 17 603 000 ^m	18 503 663 18 503 000 ^m	18 920 341 18 921 000 ^m	19 146 785	19 283 359	19 371 947	19 432 757
15	17 361 637 17 364 000 ^m	20 333 858 20 333 000 ^m	21 377 321 21 370 000 ^m	21 860 757 21 856 000 ^m	22 123 466 22 120 000 ^m	22 281 906 22 300 000 ^m	22 384 668	22 455 211
16	19 847 545 19 852 000 ^m	23 261 575 23 258 000 ^m	24 459 988 24 456 000 ^m	25 015 165 25 010 000 ^m	25 316 844 25 320 000 ^m	25 498 778	25 616 769	25 697 770
17	22 501 407 22 503 000 ^m	26 388 056 26 399 000 ^m	27 752 214	28 384 123	28 727 480	28 934 540	29 068 815	29 160 998
18	25 323 677 25 320 000 ^p	29 713 799 29 860 000 ^p	31 254 512	31 968 152	32 355 898	32 589 718	32 741 333	32 845 425
19	28 314 796 28 321 000 ^m	33 239 286	34 967 379	35 767 753	36 202 603	36 464 816	36 634 828	36 751 556
20	31 475 368 31 480 000 ^m	36 965 179 36 990 000 ^p	38 891 500 38 880 000 ^p	39 783 623	40 268 297	40 560 541	40 750 009	40 880 100
21	34 805 980 34 823 000 ^m	40 892 127	43 027 548	44 016 444	44 553 664	44 877 580	45 087 562	45 231 746
22	38 307 172 38 313 000 ^m	45 020 716 (45 021 900) ^q	47 376 125 (47 376 200) ^q	48 466 824	49 059 319	49 416 547	49 648 105	49 807 111
23	41 979 595 41 976 000 ^m	49 351 661 (49 353 420) ^r	51 937 971 (51 938 260) ^r	53 135 514	53 786 016	54 178 201	54 432 397	54 606 955
24	45 823 893 45 890 000 ^s	53 885 665	56 713 811	58 023 250	58 734 496	59 163 285	59 441 182	59 632 024

TABLE XI. (Continued.)

Z	2^1P	3^1P	4^1P	5^1P	6^1P	7^1P	8^1P	9^1P
25	49 840 798 (49 853 100) ^t	58 623 534 (58 627 100) ^t	61 704 478 (61 705 400) ^t	63 130 875	63 905 607	64 372 651	64 675 313	64 883 171
26	54 031 001 54 039 000 ^p	63 566 021 63 810 000 ^p	66 910 750 66 260 000 ^p	68 459 176	69 300 142	69 807 092	70 135 587	70 361 195

^aFor each entry, the first line is the result obtained in the present paper, the second line (if present) is from experiment, except where parentheses indicate a theoretical value.

^bW. C. Martin, J. Phys. Chem. Ref. Data 2, 257 (1973).

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ⁱA. M. Cantu, E. Jannitti, and G. Tondello, J. Opt. Soc. Am. 64, 699 (1974).

^jC. E. Moore, *Atomic Energy Levels*, Natl. Stds. Ref. Data Ser. (National Bureau of Standards, Washington, D.C., 1979), Vol. 3, Sec. 8.

^kN. J. Peacock, M. G. Hobby, and M. Galanti, J. Phys. B 6, L298 (1973).

^lJ. H. Parkinson, Solar Phys. 42, 183 (1975).

^mV. A. Boiko, A. Ya. Faenov, and S. A. Pikuz, J. Quant. Spectrosc. Radiat. Transfer 19, 11 (1978).

ⁿW. C. Martin and R. Zalubas, J. Phys. Chem. Ref. Data 10, 153 (1981).

^oW. C. Martin and R. Zalubas, J. Phys. Chem. Ref. Data 9, 1 (1980).

^pW. M. Neupert, Solar Phys. 18, 474 (1971).

^qC. Corliss and J. Sugar, J. Phys. Chem. Ref. Data 8, 1 (1979).

^rJ. Sugar and C. Corliss, J. Phys. Chem. Ref. Data 7, 1191 (1978).

^sJ. Sugar and C. Corliss, J. Phys. Chem. Ref. Data 6, 317 (1977).

^tC. Corliss and J. Sugar, J. Phys. Chem. Ref. Data 6, 1253 (1977).

TABLE XII. Transition energies in cm^{-1} for the N^3P states.^a

Z	2^3P	3^3P	4^3P	5^3P	6^3P	7^3P	8^3P	9^3P
2	169 088.6 169 087.0 ^b	185 564.0 185 564.7 ^b	191 222.4 191 217.1 ^b	193 787.2 193 800.8 ^b	195 180.0 195 192.9 ^b	196 039.1 196 027.4 ^b	196 557.2 196 566.8 ^b	196 843.5 196 935.4 ^b
3	494 264 494 273 ^c	559 502 559 501 ^c	581 889 581 897 ^c	592 134 592 141 ^c	597 663 597 666 ^c	600 992	603 128	604 559
4	983 380 983 365 ^d	1 128 316 1 128 300 ^d	1 178 193 1 178 174 ^d	1 201 072 1 201 066 ^e	1 213 433	1 220 864	1 225 663	1 228 931
5	1 636 965 1 636 948 ^f	1 892 250 1 892 230 ^f	1 980 318 1 980 295 ^f	2 020 780 2 020 746 ^f	2 042 659	2 055 815	2 064 327	2 070 144
6	2 455 248 2 455 225 ^g	2 851 459 2 851 418 ^g	2 988 408 2 988 359 ^g	3 051 398 3 051 332 ^g	3 085 482 (3 085 435) ^g	3 105 984 (3 105 933) ^g	3 119 257 (3 119 212) ^g	3 128 340
7	3 438 424 3 437 960 ^h	4 006 119 4 006 160 ^h	4 202 641 4 202 620 ^h	4 293 104 4 293 080 ^h	4 342 081 4 342 360 ⁱ	4 371 547	4 390 631	4 403 700
8	4 586 668 4 585 980 ^j	5 356 406 5 355 670 ^j	5 623 194 5 622 600 ^j	5 746 080 5 745 440 ^j	5 812 635 5 811 730 ^j	5 852 687 5 851 890 ^j	5 878 632 (5 877 800) ^j	5 896 407 (5 894 500) ^j
9	5 900 191 5 901 100 ⁱ	6 902 545 6 903 100 ⁱ	7 250 300 7 250 900 ⁱ	7 410 559 7 410 800 ⁱ	7 497 382 7 489 000 ^k	7 549 640	7 583 497	7 606 698

TABLE XII. (*Continued.*)

Z	2^3P	3^3P	4^3P	5^3P	6^3P	7^3P	8^3P	9^3P
10	7 379 191 7 379 000 ^l	8 644 748	9 084 175	9 286 762	9 396 542	9 462 628	9 505 448	9 534 796
11	9 023 957 9 018 800 ^m	10 583 331 (10 583 324) ⁿ	11 125 148 (11 125 103) ⁿ	11 375 020 (11 374 960) ⁿ	11 510 450 (11 510 387) ⁿ	11 591 987 (11 591 920) ⁿ	11 644 820 (11 644 781) ⁿ	11 681 038 (11 680 991) ⁿ
12	10 834 741 10 836 000 ^m	12 718 567 (12 718 550) ^o	13 373 499 (13 373 430) ^o	13 675 619 (13 675 540) ^o	13 839 393	13 938 005	14 001 904	14 045 712
13	12 811 885 12 810 000 ^m	15 050 832 (15 050 810) ^p	15 829 620 (15 829 540) ^p	16 188 955 (16 188 850) ^p	16 383 770	16 501 081	16 577 099	16 629 222
14	14 955 703 14 954 000 ^m	17 580 470	18 493 863	18 915 387	19 143 941	19 281 578	19 370 767	19 431 928
15	17 266 605 17 265 000 ^m	20 307 933	21 366 697	21 855 388	22 120 384	22 279 976	22 383 390	22 454 313
16	19 744 960 19 744 000 ^m	23 233 623	24 448 537	25 009 380	25 313 523	25 496 699	25 615 393	25 696 803
17	22 391 248 22 387 000 ^m	26 358 070	27 739 933	28 377 920	28 723 919	28 932 310	29 067 341	29 159 963
18	25 205 926 25 195 000 ^q	29 681 773	31 241 398	31 961 530	32 352 097	32 587 337	32 739 761	32 844 321
19	28 189 434 28 182 000 ^m	33 205 211	34 953 429	35 760 711	36 198 560	36 462 284	36 633 158	36 750 383
20	31 342 374 31 322 000 ^m	36 929 049 (36 928 570) ^r	38 876 712 (38 876 210) ^r	39 776 159	40 264 012	40 557 857	40 748 240	40 878 859
21	34 665 334 34 664 000 ^m	40 853 935	43 011 917	44 008 556	44 549 136	44 874 743	45 085 696	45 230 436
22	38 158 851 38 126 000 ^m	44 980 453 (44 979 590) ^s	47 359 648 (47 358 890) ^s	48 458 510	49 054 547	49 413 558	49 646 141	49 805 732
23	41 823 578 41 773 000 ^m	49 309 318 (49 308 150) ^t	51 920 644 (51 919 670) ^t	53 126 772	53 780 998	54 175 058	54 430 335	54 605 508
24	45 660 157 45 540 000 ^u	53 841 235	56 695 631	58 014 079	58 729 232	59 159 988	59 439 022	59 630 508
25	49 669 320 (49 666 400) ^v	58 577 006 (58 575 000) ^v	61 685 440 (61 683 900) ^v	63 121 272	63 900 095	64 369 198	64 673 055	64 881 587
26	53 851 756 53 894 000 ^q	63 517 386	66 890 850	68 449 140	69 294 382	69 803 484	70 133 231	70 359 542

^aFor each entry, the first line is the result obtained in the present paper, the second line (if present) is from experiment, except where parentheses indicate a theoretical value.

^bW. C. Martin, J. Phys. Chem. Ref. Data 2, 257 (1973).

^cC. E. Moore, *Atomic Energy Levels*, Natl. Bur. Stds. (U.S.) Circ. No. 467 (U.S. GPO, Washington, D.C., 1949).

^dB. Löfstrand, Ref. 20.

TABLE XII. (*Continued.*)

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^hC. E. Moore, *Atomic Energy Levels*, Natl. Stds. Ref. Data Ser. (National Bureau of Standards, Washington, D.C., 1971), Vol. 3, Sec. 4.
ⁱA. M. Cantu, E. Jannitti, and G. Tondello, J. Opt. Soc. Am. 64, 699 (1974).
^jC. E. Moore, *Atomic Energy Levels*, Natl. Stds. Ref. Data Ser. (National Bureau of Standards, Washington, D.C., 1979), Vol. 3, Sec. 8.
^kN. J. Peacock, M. G. Hobby, and M. Galanti, J. Phys. B 6, L298 (1973).
^lJ. H. Parkinson, Solar Phys. 42, 183 (1975).
^mV. A. Boiko, A. Ya Faenov, and S. A. Pikuz, J. Quant. Spectrosc. Radiat. Transfer 19, 11 (1978).
ⁿW. C. Martin and R. Zalubas, J. Phys. Chem. Ref. Data 10, 153 (1981).
^oW. C. Martin and R. Zalubas, J. Phys. Chem. Ref. Data 9, 1 (1980).
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^qW. M. Neupert, Solar Phys. 18, 474 (1971).
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^sC. Corliss and J. Sugar, J. Phys. Chem. Ref. Data 8, 1 (1979).
^tJ. Sugar and C. Corliss, J. Phys. Chem. Ref. Data 7, 1191 (1978).
^uJ. Sugar and C. Corliss, J. Phys. Chem. Ref. Data 6, 317 (1977).
^vC. Corliss and J. Sugar, J. Phys. Chem. Ref. Data 6, 1253 (1977).

TABLE XIII. Transition energies in cm^{-1} for the N^1D states.^a

Z	3^1D	4^1D	5^1D	6^1D	7^1D	8^1D	9^1D
2	186 075.9	191 470.0	193 877.5	195 279.7	196 027.2	196 497.5	196 858.3
	186 105.1 ^b	191 446.6 ^b	193 918.4 ^b	195 260.9 ^b	196 070.2 ^b	196 595.5 ^b	196 955.5 ^b
3	561 260	582 642	592 508	597 903	601 114	603 179	604 620
	561 276 ^c	582 631 ^c	592 508 ^c	597 877 ^c	601 115 ^c		
4	1 131 474	1 179 523	1 201 747	1 213 837	1 221 104	1 225 803	1 229 044
	1 131 462 ^d	1 179 495 ^e	1 201 743 ^e				
5	1 896 869	1 982 255	2 021 765	2 043 239	2 056 173	2 064 550	2 070 311
	1 896 829 ^f	1 982 223 ^f					
6	2 857 577	2 990 969	3 052 702	3 086 244	3 106 458	3 119 562	3 128 562
	2 857 529 ^g	2 990 923 ^g	(3 052 656) ^g	(3 086 189) ^g	(3 106 407) ^g	(3 119 630) ^g	
7	4 013 770	4 205 840	4 294 734	4 343 028	4 372 141	4 391 020	4 403 978
	4 013 770 ^h	4 205 830 ^h	4 294 670 ^h				
8	5 365 626	5 627 048	5 748 043	5 813 774	5 853 402	5 879 105	5 896 743
	5 365 470 ⁱ	5 626 670 ⁱ	5 748 230 ⁱ	5 813 680 ⁱ			
9	6 913 377	7 254 826	7 412 865	7 498 717	7 550 480	7 584 057	7 607 094
	6 913 510 ^h	7 254 600 ^h	7 412 600 ^h	7 498 700 ^h	7 548 400 ^h	7 584 900 ^h	
10	8 657 243	9 089 396	9 289 421	9 398 081	9 463 596	9 506 096	9 535 254
11	10 597 549 (10 597 475) ^j	11 131 088 (11 131 017) ^j	11 378 046 (11 377 984) ^j	11 512 200 (11 512 137) ^j	11 593 088	11 645 562	11 681 561
12	12 734 584 (12 734 480) ^k	13 380 191 (13 380 090) ^k	13 679 029 (13 678 940) ^k	13 841 364	13 939 245	14 002 743	14 046 303
13	15 068 736	15 837 101	16 192 767	16 385 973	16 502 467	16 578 040	16 629 885
14	17 600 365	18 502 177	18 919 623	19 146 389	19 283 118	19 371 818	19 432 667

TABLE XIII. (Continued.)

Z	3^1D	4^1D	5^1D	6^1D	7^1D	8^1D	9^1D
15	20 329 940	21 375 896	21 860 076	22 123 093	22 281 680	22 384 557	22 455 133
16	23 257 882	24 458 680	25 014 549	25 316 509	25 498 577	25 616 684	25 697 710
17	26 384 742	27 751 087	28 383 606	28 727 203	28 934 376	29 068 765	29 160 963
18	29 711 039	31 253 642	31 967 771	32 355 702	32 589 604	32 741 329	32 845 423
19	33 237 277	34 966 847	35 767 551	36 202 512	36 464 770	36 634 882	36 751 595
20	36 964 143	38 891 402	39 783 649	40 268 338	40 560 579	40 750 134	40 880 189
21	40 892 312	43 027 987	44 016 750	44 553 870	44 877 721	45 087 773	45 231 895
22	45 022 395	47 377 217	48 467 470	49 059 722	49 416 813	49 648 418	49 807 332
23	49 355 137	51 939 842	53 136 565	53 786 655	54 178 616	54 432 829	54 607 260
24	53 891 270	56 716 603	58 024 778	58 735 413	59 163 875	59 441 753	59 632 426
25	58 631 629	61 708 344	63 132 958	63 906 846	64 373 444	64 676 044	64 883 686
26	63 577 002	66 915 855	68 461 900	69 301 754	69 808 121	70 136 501	70 361 839

^aFor each entry, the first line is the result obtained in the present paper, the second line (if present) is from experiment, except where parentheses indicate a theoretical value.

^bW. C. Martin, J. Phys. Chem. Ref. Data **2**, 257 (1973).

^cC. E. Moore, *Atomic Energy Levels*, Natl. Bur. Stds. (U.S.) Circ. No. 467 (U.S. GPO, Washington, D.C., 1949).

^dB. Löfstrand, Ref. 20.

^eM. Eildesberg, J. Phys. B **5**, 1031 (1972).

^fM. Eildesberg, J. Phys. B **7**, 1476 (1974).

^gC. E. Moore, *Atomic Energy Levels*, Natl. Stds. Ref. Data Ser. (National Bureau of Standards, Washington, D.C., 1970), Vol. 3, Sec. 3.

^hA. M. Cantu, E. Jannitti, and G. Tondello, J. Opt. Soc. Am. **64**, 699 (1974).

ⁱC. E. Moore, *Atomic Energy Levels*, Natl. Stds. Ref. Data Ser. (National Bureau of Standards, Washington, D.C., 1979), Vol. 3, Sec. 8.

^jW. C. Martin and R. Zalubas, J. Phys. Chem. Ref. Data **10**, 153 (1981).

^kW. C. Martin and R. Zalubas, J. Phys. Chem. Ref. Data **9**, 1 (1980).

TABLE XIV. Transition energies in cm^{-1} for the N^3D states.^a

Z	3^3D	4^3D	5^3D	6^3D	7^3D	8^3D	9^3D
2	186 099.4	191 433.5	193 906.5	195 259.6	196 052.3	196 495.6	196 857.5
	186 101.7 ^b	191 444.6 ^b	193 917.2 ^b	195 260.2 ^b	196 069.7 ^b	196 595.2 ^b	196 955.3 ^b
3	561 244	582 612	592 503	597 878	601 116	603 178	604 620
	561 245 ^c	582 612 ^c	592 505 ^c	597 876 ^c	601 115 ^c		
4	1 131 400	1 179 471	1 201 721	1 213 807	1 221 094	1 225 798	1 229 041
	1 131 383 ^d	1 179 451 ^e	1 201 702 ^d	1 213 791 ^e			
5	1 896 723	1 982 167	2 021 717	2 043 199	2 056 152	2 064 540	2 070 303
	1 896 685 ^f	1 982 141 ^f	2 021 696 ^f				
6	2 857 350	2 990 839	3 052 629	3 086 191	3 106 427	3 119 545	3 128 549
	2 857 310 ^g	2 990 776 ^g	3 052 589 ^g	(3 086 138) ^g	(3 106 374) ^g	(3 119 507) ^g	
7	4 013 456	4 205 665	4 294 635	4 342 961	4 372 099	4 390 995	4 403 960
	4 013 460 ^h	4 205 820 ^h	4 294 570 ^h	4 343 120 ⁱ			

TABLE XIV. (Continued.)

Z	3^3D	4^3D	5^3D	6^3D	7^3D	8^3D	9^3D
8	5 365 221 5 364 420 ^j	5 626 824 5 626 280 ^j	5 747 917 5 747 420 ^j	5 813 692 5 813 070 ^j	5 853 350 5 853 660 ^j	5 879 074 5 878 400 ^j	5 896 720 5 892 950 ^j
9	6 912 878 6 912 500 ^j	7 254 551 7 254 100 ^j	7 412 710 7 412 100 ^j	7 498 619 7 498 100 ^j	7 550 416 7 549 800 ^j	7 584 017 7 583 300 ^j	7 607 065 7 607 100 ^j
10	8 656 647	9 089 069	9 289 237	9 397 966	9 463 522	9 506 049	9 535 220
11	10 596 855 (10 596 783) ^k	11 130 709 (11 130 639) ^k	11 377 833 (11 377 767) ^k	11 512 068 (11 512 003) ^k	11 593 002	11 645 507	11 681 521
12	12 733 789 (12 733 690) ^l	13 379 759 (13 379 660) ^l	13 678 786 (13 678 690) ^l	13 841 214	13 939 147	14 002 679	14 046 258
13	15 067 841	15 836 614	16 192 494	16 385 805	16 502 358	16 577 969	16 629 833
14	17 599 368	18 501 636	18 919 320	19 146 203	19 282 997	19 371 738	19 432 610
15	20 328 841	21 375 300	21 859 741	22 122 888	22 281 547	22 384 469	22 455 070
16	23 256 680	24 458 029	25 014 184	25 316 286	25 498 432	25 616 588	25 697 641
17	26 383 437	27 750 381	28 383 210	28 726 962	28 934 219	29 068 660	29 160 889
18	29 709 631	31 252 879	31 967 344	32 355 442	32 589 436	32 741 216	32 845 342
19	33 235 765	34 966 029	35 767 093	36 202 233	36 464 589	36 634 769	36 751 514
20	36 962 527	38 890 528	39 783 159	40 268 041	40 560 385	40 750 012	40 880 102
21	40 890 592	43 027 057	44 016 229	44 553 553	44 877 515	45 087 643	45 231 802
22	45 020 571	47 376 231	48 466 918	49 059 387	49 416 596	49 648 280	49 807 233
23	49 353 209	51 938 800	53 135 982	53 786 301	54 178 387	54 432 682	54 607 155
24	53 889 237	56 715 505	58 024 163	58 735 039	59 163 634	59 441 598	59 632 316
25	58 629 493	61 707 190	63 132 311	63 906 454	64 373 190	64 675 880	64 883 570
26	63 574 761	66 914 645	68 461 222	69 301 343	69 807 855	70 136 328	70 361 716

^aFor each entry, the first line is the result obtained in the present paper, the second line (if present) is from experiment, except where parentheses indicate a theoretical value.

^bW. C. Martin, J. Phys. Chem. Ref. Data 2, 257 (1973).

^cC. E. Moore, *Atomic Energy Levels*, Natl. Bur. Stand. (U.S.) Circ. No. 467 (U.S. GPO, Washington, D.C., 1949).

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^gC. E. Moore, *Atomic Energy Levels*, Natl. Stds. Ref. Data Ser. (National Bureau of Standards, Washington, D.C., 1970), Vol. 3, Sec. 3

^hC. E. Moore, *Atomic Energy Levels*, Natl. Stds. Ref. Data Ser. (National Bureau of Standards, Washington, D.C., 1971), Vol. 3, Sec. 4.

ⁱA. M. Cantu, E. Jannitti, and G. Tondello, J. Opt. Soc. Am. 64, 699 (1974).

^jC. E. Moore, *Atomic Energy Levels*, Natl. Stds. Ref. Data Ser. (National Bureau of Standards, Washington, D.C., 1979), Vol. 3, Sec. 8.

^kW. C. Martin and R. Zalubas, J. Phys. Chem. Ref. Data 10, 153 (1981).

^lW. C. Martin and R. Zalubas, J. Phys. Chem. Ref. Data 9, 1 (1980).

TABLE XV. A comparison of some observed and calculated transitions (in cm^{-1}).^a

Transition	$Z = 4^b$	$Z = 5^c$	$Z = 6^d$
$2^1S - 2^1P$	16 275.9 16 276.8 \pm 0.1	22 267 22 261 \pm 1	28 365 28 347 \pm 4
$2^3S - 2^3P$		35 399 35 408 \pm 0.5	43 940 43 963 \pm 1
$4^3P - 5^3D$	23 527.9 23 527.6 \pm 0.6	41 399 41 408 \pm 2	
$4^3S - 5^3P$		44 315 44 307 \pm 2	
$3^3D - 4^3P$		83 595 83 608 \pm 7	
$3^3P - 4^3S$	46 996.4 46 995.4 \pm 0.9	84 215 84 196 \pm 7	
$3^1P - 4^1D$	47 112.6 47 104.5 \pm 0.9	83 996 83 993 \pm 7	
$3^3P - 4^3D$	51 154.0 51 151.7 \pm 0.8	89 917 89 912 \pm 8	
$3^3S - 4^3P$	56 992.2 56 990.1 \pm 1.0	97 546 97 551 \pm 9	
$2^1P - 3^1S$	130 247 130 251 \pm 5	233 805 233 808 \pm 27	
$2^1P - 3^1D$	134 000.5 134 007.3 \pm 1.4	238 857 238 857 \pm 1	374 134 374 158 \pm 7
$2^3P - 3^3S$	137 820.5 137 819.6 \pm 2	245 807 245 797 \pm 30	
$2^3P - 3^3D$	148 020.3 148 018.1 \pm 2	259 758 259 740 \pm 13	
$2^1S - 3^1P$	151 212.4 151 212.3 \pm 2	262 514 262 508 \pm 42	404 338 404 351 \pm 33
$2^3S - 3^3P$	171 798.5 171 798.3 \pm 3	290 684 290 689 \pm 17	440 150 440 156 \pm 10
$2^1P - 4^1D$	182 049 182 047 \pm 10	324 243 324 243 \pm 52	507 526 507 552 \pm 18
$2^3P - 4^3D$	196 091 196 082 \pm 8	345 202 345 256 \pm 58	
$2^3S - 4^3P$		378 752 378 759 \pm 70	577 100 577 097 \pm 17

TABLE XV. (*Continued.*)

Transition	$Z = 4^b$	$Z = 5^c$	$Z = 6^d$
$2^3P - 5^3D$			597 381 597 364 \pm 35
$2^3S - 5^3P$			640 089 640 070 \pm 41

^aFor each entry, the upper line is the result obtained in the present paper. The lower line is the experimental result.

^bB. Löfstrand, Ref. 20.

^cM. Eideberg, J. Phys. B 7, 1476 (1974).

^dB. Edlen and B. Löfstrand, J. Phys. B 3, 1380 (1970).

tivistic correction. The simple one-electron screening approximation presented in Eq. (14) appears to be sufficient to assure that the error in the overall transition energy is essentially of the order of the error in the relativistic correction, or less.

D. Transition energies

The transition energies relative to the ground 1^1S state are presented in Tables IX–XIV. For $Z=2$ the accuracy of the present values is determined almost entirely by the convergence of the expansion for the nonrelativistic energy. This rapidly improves with increasing Z so that, as indicated above, for $Z \geq 4$ the accuracy of the tabulated values is determined by the accuracy of the relativistic correction. For increasing N (degree of excitation) the calculated nonrelativistic energies become less reliable because of the lack of accurate, higher-order ϵ_n , while the relativistic, Lamb shift, and mass-polarization corrections increase in accuracy as the contribution of the “excited electron” becomes less reliable because of the lack of accurate, higher-order ϵ_n , while the relativistic, Lamb shift, and mass-polarization corrections increase in accuracy as the contribution of the “excited electron” becomes less significant and these corrections become simply the corresponding corrections to the ground-state ionization energy. Therefore, for $Z \geq 4$, the least reliable results are the $2S$, $2P$, and $3D$ results. That this is not reflected in the comparisons presented in the tables, where any particular row displays a generally constant difference, suggests that the present results are more accurate than the experimental

data. In any case the agreement throughout the tables with both experimental and theoretical values is fairly uniform—five significant figures, occasionally four or six figures. However, since the energies in Tables IX through XIV are all relative to the 1^1S ground state, the error in the quoted experimental values is in large part due to the difficulty in establishing an accurate value for the energy of the ground state. Thus a more meaningful comparison would make use of the transitions actually observed. Such a comparison is presented in Table XV for $Z=4$, 5, and 6. Whereas differences for those ions in Tables IX through XIV are generally in the range of $20-40 \text{ cm}^{-1}$, differences in Table XV are generally less than 10 cm^{-1} for $Z=4$, 5, less than 20 cm^{-1} for $Z=6$, and generally in reasonable agreement with the experimental error limits. Particularly striking is the agreement with the data of Löfstrand²⁰ for $Z=4$, where the observed wavelengths are given to greater accuracy.

V. CONCLUSIONS

The results of the present study are based on a uniform set of expansion coefficients for the nonrelativistic energy of two-electron ions. Many of these are presented here for the first time. Relativistic and Lamb-shift corrections for these two-electron ions are computed in a particularly simple manner, and are found to yield reliable results. The accuracy to be expected for the transition energies tabulated here has a predictable behavior with respect to Z and N , and the entire set of S , P , and D state results form a consistent set of data.

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