Shift and broadening of hyperfine components of the first doublet of cesium perturbed by foreign gases

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The collision broadening and shift of the hyperfine structure components of the ground-state splitting for the D_1 (8943 Å) and D_2 (8521 Å) cesium lines by noble gases (He, Ne, Ar, Kr, and Xe) and by light molecular gases (H₂ and N₂) under pressures not greater than 150 Torr and at a temperature of 295 K are investigated. The Lennard-Jones (12-6) interatomic potential constants are calculated for these systems, using the semiclassical theory of Lindholm-Foley (for Cs with Ne, Ar, Kr, Xe, and N₂ systems) and the quantum theory of Baranger (for Cs with He and H₂ systems), by comparing with our experimental results in fine structure. These potential constants are employed for the determination of the shift and broadening in hyperfine structure, using the hyperfine theory of collision broadening.

I. INTRODUCTION

The spectral lines of an atom experience modifications in their profiles when the atom is perturbed by the presence of other atoms or molecules.1 A great many experimental studies of the shift, broadening, and asymmetry of the first resonance doublets of cesium perturbed by foreign gases have been made.2 But there are few experimental results of the hyperfine structure of cesium: The broadening of the 4555 Å line by argon³ and of the second doublet ($\lambda = 4555$ and 4593 Å) by helium, argon, and xenon4 have been investigated. The broadening and shift of this doublet has been measured under pressure of neon, argon, and xenon.⁵ For the first doublet we have measured the shift of the hfs (hyperfine-structure) components of the ground state for the D_1 (8943 Å) and D_2 (8521 Å) lines in the presence of noble gases and light molecular gases (hydrogen, nitrogen) at pressures not greater than 150 Torr. The results for the light noble gases helium, neon, and argon have been reported in a previous paper.6

In this paper we present our measurements for krypton, xenon, hydrogen, and nitrogen as foreign gases as well as new values for helium, neon, and argon. These latter values correspond to the ones previously reported by us6 corrected with a more accurate deconvolution method of the instrumental function from the experimental profiles. The method is described in this paper. We also present here the values of the interatomic potential constants for cesium-noble gas and cesium-light molecular gas, calculated for a Lennard-Jones (12-6) potential using the shift and broadening in fine structure, derived from our experimental results in hfs. Moreover, we have calculated with these interatomic potential the shift and broadening of the hfs components of the D_1 line.

II. EXPERIMENTAL METHOD

The absorption coefficient of the cesium vapor with and without foreign gas has been measured in order to make a comparative study on the influence of the foreign gases. The profile of the absorption coefficient has been obtained by the double measurement method, which allows the obtaining of comparable emission and transmission recordings with a common origin. One can obtain the $\alpha(\sigma)$ absorption coefficient profiles by the expression

$$\alpha(\sigma) = (1/l) \ln \left[I_0(\sigma) / I(\sigma) \right], \tag{1}$$

where l is the length of the absorption cell, $I(\sigma)$ and $I_0(\sigma)$ are the spectral distributions of the transmitted and emitted light, respectively. These spectral distribution profiles were measured with a piezoelectrically scanned Fabry-Perot spectrometer with synchronous detection, described in Ref. 6.

For small absorption, such that Xl < 0.4, the absorption coefficient can be expressed by a convolution of the instrumental function of the spectrometer $W(\sigma)$, and the true absorption coefficient $X(\sigma)$ (Ref. 7).

$$\alpha(\sigma) = W(\sigma) \otimes X(\sigma); \tag{2}$$

in our case $Xl \sim 0.3$, so that the condition is fulfilled

The true absorption coefficient is the convolution of the $D(\sigma)$ inhomogeneous process function (Doppler broadening) and the $\pi(\sigma)$ homogeneous process function (pressure effects, resonant broadening, and natural width). Therefore, if the Doppler broadening is a Gaussian function and $\pi(\sigma)$ is a Lorentzian function, the experimental absorption coefficients may be considered as Voigt profiles. Moreover, the instrumental function $W(\sigma)$ of our Fabry-Perot spectrometer may be approximated

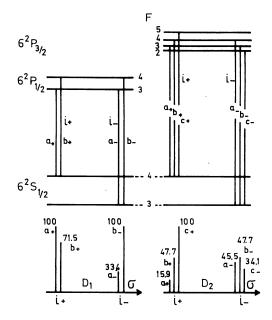


FIG. 1. Hyperfine pattern of $6^2S_{1/2}$ - $6^2P_{1/2}$, $6^2S_{1/2}$ - $6^2P_{3/2}$ transition. The theoretical intensity is given for each of their hyperfine components.

by a Voigt function, and an experimental determination of this function has been made by us. The instrumental function for the first optical doublet of cesium is characterized by the following widths:

Instrumental Voigt function

	D_1	D_2
width $(10^{-3} \text{ cm}^{-1})$	49.1	51.1
Lorentzian component	10.3	10.3
Gaussian component	43.3	45.3

Finally, with these considerations the experimental absorption coefficient can be expressed by means of a Voigt function.

To obtain the homogeneous process contribution $\pi(\sigma)$, one must perform deconvolution of $\pi(\sigma)$, $W(\sigma)$, and $D(\sigma)$. We have used a deconvolution method consisting in applying an analytic approximation to the Voigt function. Pyrex glass absorption cells of 5-cm width with cesium and foreign gases at pressures of 10, 20, 50, 100, and 150 Torr were used. The measurements were made for a temperature of 295 ± 1 K.

III. RESULTS

We have determined the shift and broadening of the hfs components (i_+, i_-) corresponding to the ground-state splitting of the D_1 , D_2 lines of cesium (Fig. 1), starting from the experimental absorption coefficients for the test cells of cesium-buffer gas. The absorption coefficient of a gasless cesium cell has been taken as a reference.

As the hfs components (i_+, i_-) of the D_1 , D_2 lines are complex spectral lines, it has been necessary to carry out an analysis of the absorption coefficient profiles.7 The absorption coefficient of the gasless cesium cell (reference coefficient) for each hfs component (i_+, i_-) of the lines D_1 , D_2 is obtained by the addition of the absorption coefficient of hfs components corresponding to the excited state. For the hfs components i_+, i_- of the line D_2 , the width of the absorption coefficient of each hfs component a_{\pm} , b_{\pm} , c_{\pm} is ~52.8×10⁻³ cm⁻¹; much higher than the hfs splitting of the centers of gravity of the a_{\pm} , b_{\pm} , c_{\pm} components (~7.6×10⁻³ cm⁻¹). In this case the absorption coefficient of the components i_+, i_- of the line D_2 can be considered as a Voigt profile.¹⁰ For the line D_1 , the width of the absorption coefficient of the hfs components a_{\pm} , b_{\pm} is ~50.7×10⁻³ cm⁻¹; it is of the same order as the hfs splitting of the centers of gravity (~40.6×10⁻³ cm⁻¹). Each one of the components i_+, i_- of the line D_1 is then obtained by addition of two shifted Voigt profiles.

The Voigt profile corresponding to the hfs components $(a_{\pm}, b_{\pm}, c_{\pm})$ of the absorption coefficient has a Gaussian component, which is the sum of the Doppler broadening and the Gaussian component of the instrumental function. For our experimental conditions, the Lorentzian component is the sum of the natural width¹¹ ($\sim 10^{-3}$ cm⁻¹), the resonant broadening [$\sim 10^{-7}$ cm⁻¹, for an atomic density of 3×10^{14} atom cm⁻³ at T = 295 K (Ref. 12)], and the Lorentzian component of the instrumental function.

The analysis of the hfs components i_+, i_- of the absorption coefficient of cesium with gas cells has been made by fitting to each component i_+, i_- a Voigt profile for the line D_2 , and an addition of two Voigt profiles for the line D_1 . The Lorentzian contribution, due only to broadening by pressure, is obtained by deconvolution of the absorption coefficient of the cesium-gas system from the one of the gasless cesium. The shifts have been evaluated by considering the gravity center of the i_+, i_- components of the absorption coefficient of cesium with gas respective to the gasless cesium.

The values obtained for the shift and broadening for each pressure value of the perturbing gases have been fitted by a least-squares method to a straight line, ¹³ obtaining a standard deviation of the slope between 1.0 to 3.0%. Each straight line crosses the origin. The values of the shifts and broadenings are given in Table I and Table II, respectively.

As our results are for hyperfine structure and those of other authors for fine structure, we thought it was useful to calculate the shift and

TABLE I.	Summary of	results for the	e sh ift s of all hfs	components of	f the D_1 , D_2	lines of Ca	3
in He, Ne, A					-	-	

	hfs components of the	shift (10 ⁻³ cm ⁻¹ /	Torr; $T = 295 \text{ K}$
System	ground-state splitting	$D_1(8943 \text{ Å})$	$D_2(8521 \text{ Å})$
Не	i_+	0.17 ± 0.03	0.08 ± 0.01
	<i>i_</i>	$\textbf{0.25} \pm \textbf{0.04}$	0.18 ± 0.04
Ne	i_+	-0.09 ± 0.02	-0.10 ± 0.01
	<i>i_</i>	-0.12 ± 0.04	-0.21 ± 0.01
Ar	i_+	-0.26 ± 0.03	-0.23 ± 0.02
	<i>i_</i>	-0.32 ± 0.03	-0.28 ± 0.03
Kr	i_+	-0.09 ± 0.05	-0.28 ± 0.02
	i_	-0.29 ± 0.01	-0.24 ± 0.05
Xe	i_+	-0.24 ± 0.05	-0.33 ± 0.02
	i_	-0.30 ± 0.03	-0.26 ± 0.03
H_2	i_+	0.001 ± 0.008	-0.35 ± 0.01
-	<i>i_</i>	0.148 ± 0.004	0.46 ± 0.01
N_2	i_+	-0.21 ± 0.02	-0.18 ± 0.07
-	i_	-0.28 ± 0.03	-0.30 ± 0.06

broadening in fine structure from our values in hfs. We have hence calculated the shifts of the centers of gravity of the D_1 , D_2 lines, obtained by Jackson's method. 4 We obtained the broadenings by the weighted mean of the broadenings of the hyperfine components i_+ , i_- of each one of the lines. These values are shown in Table III together with those of other authors. These experimental and theoretical values correspond to a linear extrapolation to our experimental conditions from the values given by these authors for higher pressures, except for the values of Granier and collaborators¹⁵ given for low pressure (≤3 relative density). The extrapolation to the temperature range of interest in our measurements has been obtained using the expression given by Kielkopf.16

IV. INTERATOMIC POTENTIALS

We present here the calculation of the interatomic potential constants for Cs-gas systems with a Lennard-Jones $(C_{12}r^{-12}-C_6r^{-6})$ potential using our experimental results in fine structure. The Lindholm-Foley theory¹⁷ has been used to relate the shift (β) and broadening (2γ) to the interatomic potential constants by the expressions¹⁸

$$\beta = 2N\overline{v} \int_{0}^{\infty} \sin\eta(\rho) \rho \, d\rho \,, \tag{3}$$

$$2\gamma = 4\pi N \overline{v} \int_0^{\infty} \left[1 - \cos\eta(\rho)\right] \rho \, d\rho \,, \tag{4}$$

where ρ is the impact parameter of the collision, \overline{v} is the mean relative velocity, N is the number

TABLE II. Summary of results for the broadening of the hfs components of the ground-state splitting of the D_1 , D_2 lines of Cs in He, Ne, Ar, Kr, Xe, H_2 , and N_2 .

		Broadening (10 ⁻³ cm	$^{-1}/\text{Torr}; T = 295 \text{ K}$
System	hfs components	D_1 (8943 Å)	$D_2(8521 \text{ Å})$
Cs-He	$i_{\scriptscriptstyle +}$	0.66 ± 0.03	0.92 ± 0.05
	i_	0.62 ± 0.06	0.86 ± 0.06
Cs-Ne	i_+	0.10 ± 0.01	0.39 ± 0.04
	<i>i_</i>	0.57 ± 0.04	0.30 ± 0.03
Cs-Ar	i_+	0.69 ± 0.03	0.73 ± 0.03
	i_	0.62 ± 0.05	0.79 ± 0.02
Cs-Kr	i_+	0.70 ± 0.10	0.40 ± 0.04
	i_	0.61 ± 0.06	0.33 ± 0.02
Cs-Xe	i ,	0.74 ± 0.11	1.89 ± 0.10
	i_	0.69 ± 0.06	1.95 ± 0.30
$Cs-H_2$	i ₊	1.44 ± 0.28	2.00 ± 0.32
-	i_	1.25 ± 0.13	1.98 ± 0.21
$Cs-N_2$	i ,	1.01 ± 0.33	1.19 ± 0.22
-	i_	1.02 ± 0.13	1.38 ± 0.31

TABLE III. Shift (β/N) and broadening (γ/N) constants for D_1 , D_2 lines of the Cs in He, Ne, Ar, Kr, Xe, H₂, and N₂. (β/N) and γ/N are in units for 10^{-21} cm⁻¹ (atom/cm³)⁻¹). (ist denotes isotropic potential; ans, anisotropic potential).

Perturbing	D ₁ (89-	D_1 (8943 Å)		21 Å)		
gas	β/N	γ/N	β/N	γ/N	References	
He	6.72 ± 1.02	19.49 ± 1.37	3.94 ± 1.71	27.20 ± 1.72	Present work	
	7.18	24.18	1.14	24.18	S. Y. Ch'en ^{a,c}	
	4.76	26.01	0.56	24.05	Ref. 2(b) ^a	
	0.616	15.78	0.616	15.78	Ref. 16 ^b	
	12.95	20.47			Ref. 23(b) ^b	
Ne	-2.93 ± 0.09	10.29 ± 0.87	-4.33 ± 0.33	10.51 ± 1.10	Present work	
	-1.86	8.78	-3.72	8.78	Ref. 2(d) ^a	
	-4.4 0	8.41	-4.4 0	8.41	Ref. 16 ^b	
\mathbf{Ar}	-8.87 ± 0.41	20.13 ± 0.23	-7.54 ± 0.33	23.21 ± 0.11	Present work	
	-8.85	21.22	-7.99	21.22	Ref. 2(a) ^a	
	-6.28	19.16	-6.28	19.16	Ref. 16 ^b	
			-7. 88	21.20	S.Y. Ch'en	
	-7.44	29.75			Ref. 23(b) ^{b,c}	
			-4.91	19.32	Ref. 15 ^b (ist)	
			-6.40	22.69	(ans)	
Kr	-2.69 ± 0.1	20.16 ± 2.59	-8.10 ± 1.13	11.18 ± 0.98	Present work	
	-7.44	10.41	-7.81	10.42	Ref. 2(c) ^a	
	-7.44	18.60	-7.44	18.60	J. Duperier ^{a, d}	
	-6.28	19.02	-6.28	19.02	Ref. 16 ^b	
	-4.48	17.85	-4.07	17.48	Ref. 15 b(ist)	
			-5.0 2	18.97	(ans)	
Xe	-8.42 ± 1.11	21.84 ± 2.64	$\boldsymbol{-9.07} \pm 0.84$	58.62 ± 6.26	Present work	
	-9.30	18.23	-9.29	8.55	Ref. 2(e) and 2(f) ^a	
	-11.90	31.99			Ref. 23(a) ^b	
	-8.55	18.60	-8.56	18.60	Ref. $2(c)^a$	
	-6.96	20.95	-6.96	20.95	Ref. 16 ^b	
	-5.21	22.98	-5. 32	22.95	Ref. 15 ^{b(ist)}	
			-5.09	20.35	(ans)	
, $\mathbf{H_2}$	2.29 ± 0.19	$\textbf{41.08} \pm \textbf{6.30}$	$\textbf{1.81} \pm \textbf{0.26}$	47.44 ± 8.10	Present work	
N_2	-7.38 ± 0.11	30.93 ± 5.71	-7.25 ± 0.24	39.38 ± 9.73	Present work	
	-6.42				E. Bernabeu ^{a, e}	

a Experimental value.

density of the buffer gas, and $\eta(\rho)$ is the phase change caused by a collision.

For the impact approximation, and assuming that the atoms move in classical straight paths one obtains

$$\eta(\rho) = \frac{63\pi}{256} \frac{C_{12}}{\hbar \overline{\nu}} \rho^{-11} - \frac{3\pi}{8} \frac{C_6}{\hbar \overline{\nu}} \rho^{-5}.$$
 (5)

The validity range of the impact approximation for the Lennard-Jones (12-6) potential is determined by¹⁹

$$\gamma \ll (\overline{v}/2\pi\sqrt{\pi c})(C_6/C_{12})^{1/6}$$
. (6)

For neutral atom pairs $C_6 \sim 10^{-58} \ {\rm erg \, cm^6}$, $C_{12} \sim 10^{-102} \ {\rm erg \, cm^{12}}$, and $\overline{v} \sim 10^5 \ {\rm cm \, sec^{-1}}$. With these values, we find that the condition (6) is $\gamma \ll 6.45 \ {\rm cm^{-1}}$. This condition is largely fulfilled by all our experimental results.

The path described by the colliding atoms can influence the results for shift and broadening. In particular, if the interaction potential between these atoms is much smaller than the kinetic energy, the path can be considered as straight. This condition is verified in the applicability range of semiclassical theory.

A qualitative calculation can be made in order to

^bTheoretical value.

^cS. Y. Ch'en, in Proceedings of the International Conference on Optical Pumping and Atomics Lines Shape, edited by T. Skalinski (Panstwowe Wydawnictwo Naukowe, Warsaw, 1969), p. 403.

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TABLE IV. Su	\mathbf{mmary} of results for C_6 and	$\operatorname{id} C_{12} \operatorname{for} D_1, D_2 \operatorname{lines} C_1$	of Cs in He, Ne, Ar, Kr,
$Xe, H_2, and N_2.$	$(C_6 \text{ and } C_{12} \text{ are in units of }$	$10^{-58} \text{ erg cm}^6 \text{ and } 10^{-10}$	2 erg cm 12).

	D ₁ (8	943 Å)	D_1 (8)	521 Å)	
System	C 6	C_{12}	C ₆	C ₁₂	
Cs-He	0.87 ± 0.26	-0.245 ± 0.31	0.66 ± 0.28	-3.38 ± 0.61	
	0.37		0.42		Ref. 22 ^a
	0.76	40.0	-0. 32	-2.12	Ref. 21 ^a
	-1.04	-0.4 3	-0.4 3		Ref. 23 ^b
Cs-Ne	1.57 ± 0.28	34.0 ± 5.6	$\textbf{0.64} \pm \textbf{0.13}$	$\textbf{0.76} \pm \textbf{0.36}$	
	0.71		0.81		Ref. 22 ^a
	1.66	340.0	-0.52	-19.67	Ref. 21 a
	0.91	11.0	1.2	11.0	S. Y. Ch'en ^{b, c}
Cs-Ar	5.40 ± 0.38	84.22 ± 15.1	4.73 ± 0.65	55.20 ± 8.60	
	2.94		3.35		Ref. 22 ^a
	7.49	143.76	-0.41	-3.41	Ref. 21 ^a
	20.0	8400			S. Y. Ch'en ^{b, c}
			3.85	7.69	Ref. 23 ^b
Cs-Kr	13.16 ± 0.53	16 360	7.29 ± 0.42	461.8 ± 20.5	
	4.44		7.14		Ref. 22a
	12.19	243.86	0.55	3.279	Ref. 21 ^a
Cs-Xe	9.18 ± 0.31	308.22	249.27	$9.89{\times}10^6$	
	7.19		11.59		Ref. 22 ^a
	20.90	423.348	3.74	-2.74	Ref. 21 ^a
	9.2	11.8			Ref. 23 ^b
$\mathtt{Cs-H}_2$	-2.16 ± 0.71	29.52 ± 82	3.35 ± 0.64	56.9 ± 10.1	
$Cs-N_2$	26.91 ± 0.43	26 760	$\textbf{15.10} \pm \textbf{0.51}$	17460	

a Theoretical values.

estimate the range of applicability of the straightpath approximation. The orbit will be significantly curved if the impulse of the force exerted between the colliding atoms becomes of the order of the momentum of either of them. The relevant ratio p is therefore

$$p \sim (6/\mu)(\hbar/\bar{v}^4 C_6)^{1/5}$$
, (7)

if $p \sim 1$ the path is significantly curved and it is necessary to use the quantum theory of Baranger.²⁰ If $p \ll 1$ the path is practically unaltered and the semiclassical theory can be used. Substituting in expression (7) the typical values of C_6 for the systems of our interest leads to the following values:

$$\frac{\text{Gas}}{p}$$
 He Ne Ar Kr Xe H₂ N₂ 0.2 0.06 0.01 0.009 0.006 0.3 0.02.

From these considerations, we analyze our experimental results using the semiclassical theory of Lindholm-Foley for all the systems studied by us except for Cs-He, and Cs-H₂ where we use the quantum theory of Baranger. In Table IV we show the calculated values of the interatomic potential

constants for the D_1 , D_2 lines of Cs interacting with the different buffer gases. For Cs-noble gas systems our values of the C_6 constant for the D_1 line are in agreement with the theoretical values obtained by Baylis²¹; for the D_2 line such values are in agreement with the theoretical values of Mahan²² and the empirical values of Jacobson.²³ The values of the constant C_{12} are in disagreement among all the authors, 21,22,23 but the values of this constant have very little influence on the shifts and broadenings of the lines in the theories employed by us.

V. SHIFT AND BROADENING OF THE HYPERFINE COMPONENTS OF D_1 LINE

The general theory of collision broadening and shift of the hfs components of atomic spectral lines has been developed by Omont²⁴ and Rebane.²⁵ The shift and broadening of the hyperfine components are determined by the eigenvalues of the collisional relaxation matrix of the electronic-nuclear multipole moments. If we consider the transition $j = \frac{1}{2} \longrightarrow j' = \frac{1}{2}$, $(D_1 \text{ line})$ we find for the impact ap-

^b Experimental value.

^cS. Y. Ch'en, in *Proceedings of the International Conference on Optical Pumping and Atomics Lines Shape*, edited by T. Skalinski (Panstwowe Wydawnictwo Naukowe, Warsaw, 1969), p. 403.

TA	BLE V.	Theoretical	and experim	ental values	s of shift ar	d broadening	of hfs components
of D_1	line of	Cs in Cs-gas	system (in	units of 10 ⁻³	cm-1/Tor	r).	

	j	Broadening	Shift		
System	Theoretical ^a	Experimental	Theoreticala	Experimental	
Cs-Ar	0.5	0.66 ± 0.04	-0.18	-0.29 ± 0.03	
Cs-Kr	0.51	0.66 ± 0.08	-0.18	-0.29 ± 0.01	
Cs-Xe	0.57	0.72 ± 0.09	-0.21	-0.27 ± 0.04	

a Reference 25.

proximation two purely electronic constant relaxations, and if very weak magnetic interactions are neglected one obtains24 an equal value for the shift and broadening in both hfs components.

In Table V we show the theoretical values of shift and broadening of the hfs components of the D, line of cesium-heavy noble gas systems calculated by Rebane.25 These values are somewhat lower than the ones calculated by us by a weighted average from our experimental results for the

 (i_+, i_-) components of the D_1 line.²⁶ The experimental shift shown for the Cs-Kr system corresponds only to the i- component, because the experimental values for the two hfs components are in disagreement. For other systems measured by us (Cs-He, Cs-Ne, Cs- H_2 , and Cs- N_2) the values for the shift and broadening of hfs (i_+, i_-) components, are also in disagreement, but they are not shown in the table.

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²⁶This can be due, in part, to the fact that the interatomic potential considered has only a van der Waals term.