Potential energies and cross sections for MgH+

R. E. Olson

Molecular Physics Laboratory, SRI International, Menlo Park, California 94025

B. Liu

IBM Research Laboratory, San Jose, California 95193 (Received 13 March 1979)

Configuration-interaction (CI) potential curves have been calculated for the ten lowest singlet molecular states of MgH⁺. The $X^{\perp}\Sigma$, $A^{\perp}\Sigma$, and $B^{\perp}\Pi$ molecular states are calculated to be bound by 1.93, 1.93, and 0.18 eV at equilibrium separations 3.16, 3.84, and $4.41a₀$, respectively. These potential-well characteristics are in reasonable accord with analyses performed on spectroscopic data. The CI potential curves were employed in a three-channel scattering calculation to estimate $H^+ + Mg$ electron-capture cross sections at velocities (0.1–1.4) \times 10⁸ cm/sec. Capture with excitation of the ion dominates with H + Mg⁺ (3p) being the major electron-capture channel at 1×10^8 cm/sec. At higher velocities, a classical-trajectory Monte-Carlo method was used to calculate both the electron-capture and impact-ionization across sections for the H^+ + Mg collision pair. The calculated electron-capture cross sections are in reasonable agreement with the recent measurements of Morgan and Eriksen.

I. INTRODUCTION

Recently, there has been interest in measuring the single- and double-electron-capture cross sections for collisions of protons with magnesium.^{1,2} ng
sec
1,2 This work is of both applied and fundamental interest. The cross sections are useful in the development of ion sources to produce intense beams of H⁻ for heating and fueling tokamak plasmas. On the fundamental side, it is of interest to understand the mechanism for electron capture in collisions of protons with magnesium,

$$
H^+ + Mg \rightarrow H + Mg^+ \tag{1}
$$

In particular, a Mg-atom target is a convenient prototype for understanding electron capture by protons from the alkaline-earth series.

To be able to calculate the low-energy cross sections for (1), it is first necessary to determine the singlet'molecular curves arising from the lowlying states of MgH^+ (H^+ + Mg is the second excited ${}^{1}\Sigma$ state of MgH⁺). These curves allow us to estimate the cross sections and also shed light on the collision pathways for the double-electroncapture process

$$
H^+ + Mg \rightarrow H^- + Mg^{2+}, \qquad (2)
$$

which previously was proposed to be especially efficient in this system. '

Also, we have applied the classical-trajectory Monte Carlo method to the H^+ + Mg collision system to calculate the high-energy $(E \ge 10 \text{ keV})$ electron-capture cross sections for (1), and the impact-ionization cross sections for

$$
H^+ + Mg \to H^+ + Mg^+ + e \ . \tag{3}
$$

The calculated cross sections are compared with experimental data to assess the accuracy of the theoretical techniques.

II. POTENTIAL-ENERGY CALCULATIONS

Our calculations on the MgH' system neglected all the spin-interaction terms in the Hamiltonian and assumed the Born-Oppenheimer separation of nuclear and electronic wave functions. Approximations to the electronic wave functions and energies were calculated using the configuration-interaction (Cl) method. All calculations were performed using ALCHEMY, a system of programs for the calculation of molecular wave functions developed by Bagus, Liu, McLean, and Yoshimine at IBM-San Jose.

The Slater-type-function basis sets used in the MgH' calculations are given in Table I. The $7s-3p$ Mg basis of Clementi and Roetti⁴ was augmented with $4s$, $3p$, and $3d$ functions that were optimized for the lowest self-consistent-field (SCF) energies of these atomic states. The hydrogen basis is exact through $n=2$, with additional s, \hat{p} , and \hat{d} functions to provide flexibility for the molecular calculations.

The Mg basis yields an SCF energy for the Mg atom of -199.61439 hartrees, which is very close to the restricted Hartree-Fock value⁵ of -199.6154 hartrees. The SCF dipole polarizability for Mg was calculated by a finite-perturbation method to be 80.3 a_0^3 , while a fit to the long-range region of the H^+ + Mg potential curve yielded a CI polarizability of $70.6a_0^3$. These values may be compared to the accepted value' from Reinsch and Meyer's atomic CI calculation⁷ of $71.3a_0^3$. The H-atom

TABLE II. Representative values of the calculated potential energies for MgH' in units of hartrees (1 hartree=27.212 eV, 199.0 hartrees has been added to

basis yields the exact energies for the $H(1s)$ and $H(n=2)$ states and produces an SCF dipole polarizability of $4.40a_0^3$, which may be compared with the exact value of $4.5a_0^3$.

In the CI calculations the core orbitals, K and L shells of Mg, were kept fully occupied and all single and double excitations of the two valence electrons were allowed. The CI calculations used

FIG. 1. CI potential energies for several low-lying singlet molecular states of MgH⁺. The ${}^{1}\Sigma$ states are denoted by solid lines and the ¹II states by dashed lines.

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^c Pseudopotential calculation of Numrich and Truhlar (Ref. 8).
^dT₀ calculated from $E_{\text{spectro}}(Mg^*(3^2P) - Mg^*(3^2S)) + D_0(K^{12}D) - D_0(A^{12} \text{ or } B^{111}).$

^bSpectroscopic values from Rosen (Ref. 9).

an N-particle basis set of orthonormal-configuration-state functions constructed from the SCF occupied and virtual orbitals calculated on MgH^{2+} (² Σ). The dimensions of the CI calculations were 314 and 250 for the ${}^{1}\Sigma$ and ${}^{1}\Pi$ molecular states, respectively.

Potential points were calculated for 10 molecular states and 46 internuclear separations from $R = 2.0$ to $30a_0$. The results of these calculations are plotted in Fig. 1, but because of the large mass of data, only an abbreviated list of the numerical values is given in Table II. The potentialwell characteristics of the three lowest states are given in Table III along with a comparison to spectroscopic data^{8,9} and pseudopotential calculations. ' The potential-well positions are in reasonable agreement with spectroscopic data $(\Delta R_e \leq 0.12a_o)$, and the dissociation energies for the X and A states are within 10% of the values obtained in a RKR analysis by Numrich and Truhlar. ' In Table III we also compare our CI calculations to the pseudopotential calculations of Numrich and Truhlar⁸ who adjusted a polarization potential to best reproduce their RKR analysis on the $A¹\Sigma$ state; the agreement is fair for the $A¹\Sigma$ state but poor for the $X^1\Sigma$ state.

We can also attempt to independently establish error limits on the $X^1\Sigma$ and $A^1\Sigma$ dissociation energies. Recently, we have completed extensive calculations on the NaH isoelectronic system where we have performed CI calculations with excitations from only the valence shell and also from both the valence- and L-electron shells. The inclusion of the core-valence correlation energies, as compared to the valence calculation, increased the dissociation energies of the $X^1\Sigma$ and $A^1\Sigma$ states by ~0.05 eV and decreased the equilibrium separations by $\sim 0.05a_0$. The best computed D_e are also within 0.02 eV of the spectroscopic values. Reinsch and Meyer⁶ have shown that the core-valence correlation effects on dipole polarizability is greater for Na than Mg. Hence, for the isoelectronic MgH' system, our best estimate of the $X^1\Sigma$ and $A^1\Sigma$ dissociation energies is 1.98 eV, with conservative limits of ± 0.05 eV. The B ^III state is very weakly bound, Table III, and probably is uncertain by $\pm 50\%$.

III. CROSS SECTONS

The low-energy electron-capture cross sections for reaction (1) were calculated using the A, C, and $D^{1}\Sigma$ potential curves of Fig. 1. The potential curves indicate that the dominant products of the electron capture are the excited states $H + Mg^{+}(3p)$ and $H + Mg⁺(4s)$. Three-state perturbed-stationary-state calculations employing straight-line

trajectories¹⁰ were used along with the prescriptions derived by Melius and Goddard¹¹ (MG) to estimate the pertinent $\partial/\partial R$ radial coupling elements. As pointed out by MG, it is advantageous to use a molecular representation to describe the scattering process when there are more than two active channels.

In order to obtain the $\partial/\partial R$ parameters defined by MG, it is first necessary to construct diabatic potentials which are consistent with the adiabatic values. For our system, the A and C adiabatic molecular curves were decomposed to obtain a diabatic coupling matrix element H_{AC} by assuming diabatic A and C states which had the calculated $\Delta V(\infty)$ separation and employing a point-chargeinduced dipole interaction $(\alpha_d/2R^4)$ for the R dependence of the curves. The calculated dipole polarizabilities are given in Sec. II. We then ob-'tained a H_{AC} whose exponential factor was e^{-0} . in a.u. The region of maximum coupling, R_M , occurs at the location where $2H_{AC}(R) = \Delta V_{\text{dip}}(R)$. For this system, we find $R_M \approx 6.8a_0$. The analytical form¹¹ of this $\partial/\partial R$ matrix element is then given by

$$
\left\langle A \left| \frac{\partial}{\partial R} \right| C \right\rangle = \frac{\beta}{2(e^u + e^{-u})},\tag{4}
$$

where $u = \beta (R - R_M)$ and β is the exponential decay factor $0.6a_0^{-1}$. Similarly, the $\partial/\partial R$ matrix elements between the C and D states around the avoided crossing at $R_x = 5.25a_0$ may be represented $by¹¹$

$$
\left\langle D\left|\frac{\partial}{\partial R}\right|C\right\rangle = \frac{0.5\alpha}{1 + \alpha^2 (R - R_x)^2} \quad , \tag{5}
$$

where $\alpha = \Delta F / \Delta V(R_x)$. ΔF is the absolute value of the difference in slopes of the diabatic potentials at R_x , and $\Delta V(R_x)$ is the adiabatic-potential separation at R_x . From the potential calculations, the C - D parameter $\Delta V(R_x)$ = 0.0146 hartree and the parameter ΔF was graphically determined to be 0.0338 hartree/ a_0 .

The three-state close-coupling calculations yield the cross sections depicted by the solid circles in Fig. 2. The calculations indicate that at the higher energies, $E \approx 5-10$ keV, electron capture into the $H + Mg^{+}(3p)$ state predominates. However, because of the large energy gap between the A and C states, production of $Mg^+(3p)$ decreases as the collision energy is lowered and coupling to the D and higher molecular states via the avoided crossings dominate. At the low energies we predict that the cross section will plateau around \sim 1
 \times 10⁻¹⁵ cm², which simply reflects the dimen $\times\,10^{-15}$ cm², which simply reflects the dimensio of the repulsive wall of the potential where σ $=\pi R^2$ with $R \approx 3a_0$. Within this radius, several

FIG. 2. Cross sections for H'+Mg collisions. The solid circles are the single-electron-capture cross sections calculated in a three-state model using our potentials, whereas the open circles are the results of our Monte-Carlo calculations for this process; the open squares are the experimental data of Morgan and Eriksen $(Ref. 2)$. The Monte-Carlo calculations for the impactionization process are denoted by the open triangles. Lines have been drawn through the calculated points as a visual aid.

electron-capture channels couple strongly to the initial state so it is reasonable to expect approximately unit probability for electron capture. The calculations also exhibit pronounced oscillatory structure that is due to an extremum in the difference of potential energies¹² at $R = 3a_0$ between the A and C states.

The experimental and calculated cross sections are found to be in very good agreement around the maximum at \sim 8 keV, indicating the A-C interaction has been reasonably accounted for. However, at lower energies the calculated values overestimate the electron capture; this indicates that states lying higher than the D states that populate simple excitation of the Mg atom will be important and must be accounted for in a more extensive (but very difficult) calculation. It should be noted that the experimental values tend to show the low-energy plateau predicted by the calculations, but there is no hint of anypronouncedoscillatory structure.

The cross section calculations were extended into the atomic regime (region where the collision velocity is greater than the orbital velocity of the valence electrons, for Mg $v_e \approx 1.4 \times 10^8$ cm/sec) by means of the three-body three-dimensional (both nuclei and one active electron) classical-trajectory Monte Carlo method. This method has been fully

described previously. 13 For extension to the H † + Mg collision system, an effective charge of 2.85,
determined by using Slater's rules,¹⁴ was used for determined by using Slater's rules,¹⁴ was used for the Mg nucleus, and the calculated cross sections were doubled to account for the two valence electrons. This latter approximation is valid because the transition probabilities are small and hence the unitarity of the S matrix is not severely violated.

The Monte Carlo electron-capture cross sections correlate well with the molecular results and also agree well with the experimental data, as shown in Fig. 2. In addition, impact-ionization cross sections are obtained simultaneously in the Monte Carlo calculation and are given by the open triangles in Fig. 2. At the highest velocities, the calculated impact-ionization cross sections also reflect electron removal from the L shell of Mg; this removal is significant and amounts to 50% of the cross section. The L-shell calculations employed an effective charge for the Mg nucleus of 7.85, again obtained from Slater's rules, with the calculated cross sections multiplied by 8 to reflect the equivalence of the L -shell electrons. The transition probabilities were small $(\leq 2\%)$ and hence the unitarity of the S matrix was reasonably preserved. Unfortunately, there are no experimental data to compare with the ionization-cross-section calculations. Experience with Monte Carlo calculations on similar systems 15 indicates that the Monte Carlo cross sections will have a maximum uncertainty of $\pm 50\%$.

From Fig. 1 and crude Landau-Zener calculations we can predict the collision mechanism of double-electron capture to form H^- in H^+ + Mg collisions, reaction (2). This cross section will be small, because of competition from single-electron excitation and transfer channels. The H $+$ Mg²⁺ channel is 8.3 eV endoergic and will be produced by two collision mechanisms. One mechanism is simply small-impact-parameter collisions that drive the particles up to the negative-ion state by the series of curve crossings between $R = 3$ and $5a_0$ on the repulsive wall of the potential. This mechanism will be important at high energies, because of the small-impact-parameter collisions required. The dominant mechanism at low energies will be the interaction of the H^+ + Mg initial channel with the attractive $-2/R$ coulomb potential of the H⁻ + Mg⁺ reactants. This is reflected in the adiabatic potentials by the avoided crossings between the C and D states at 5.25 a_0 , another avoided crossing at $11a_0$ between the $H + Mg^{+}(3d)$ and $H(n=2) + Mg^{+}$ states, and the pinching together of the ${}^{1}\Sigma$'s arising from $H(n = 2) + Mg^{+}$ at 12.5a₀.

Using the C-D crossing and the $R = 11a_0$ crossing as the rate determining steps for double-electron capture, we have performed a crude three-state Landau-Zener calculation¹⁶ to test our hypothesis. At 100 eV we obtain a cross section of 6.7×10^{-17} At 100 eV we obtain a cross section of 6.7×10^{-7}
cm², which decreases to 3.1×10^{-17} cm² at 1000 eV. The latter number is in agreement with experiment' along with the energy dependence of the cross sections. Thus, as previously proposed, ' the long-range attractive Coulomb potential of the reactants appears to determine the low-energy double-electron-capture process.

IV. CONCLUDING REMARKS

CI potential curves for the singlet molecular states of MgH' have been calculated and used in the calculation of the single-electron-capture cross section. This cross section is in reasonable agreement with experiment and indicates that the dominant products for collision energies between 50 and 10000 eV are ground-state H and excited levels of Mg'. Similarly, we can conclude from this study that the heavier alkaline earths will behave in the same manner with excited alkalineearth ions and ground-state H preferentially produced after electron capture for $E \le 10$ keV.

Monte Carlo calculations were performed for higher-energy collisions and the electron-capture cross sections are in good agreement with experiment. Impact-ionization cross sections are also presented, and the calculations at the highest energies indicate the importance of removal of L-shell electrons.

Low-energy, $E < 1$ keV, double-electron-capture cross sections were estimated using the CI potentials and confirm that the dominant collision process is population of the long-range $-2/R$ coulomb potential of the $H^- + Mg^{2+}$ products via an initial pseudo-curve-crossing with the H^+ + Mg channel.

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- ¹T. J. Morgan and F. Eriksen, Phys. Lett. 66A, 198 (1978) .
- $2T. J.$ Morgan and F. Eriksen, Phys. Rev. A 19, 1448 (1979).
- 3R. E. Olson, Phys. Lett. 55A, 83 (1975).
- E. Clementi and C. Boetti, At. Data Nucl. Tables 14, 177 (1974).
- 5 C. F. Fischer, At. Data 4 , 301 (1972).
- 6T. M. Miller and B. Bederson, Adv. Mol. Phys. 13, 1 (1978).
- ${}^{7}E$. A. Reinsch and W. Meyer, Phys. Rev. A 14, 915 (1976).
- 8 R. W. Numrich and D. G. Truhlar, J. Phys. Chem. 79 ,

2745 (1975).

- 9 B. Rosen, Spectroscopic Data Relative to Diatomic
- Molecules (Pergamon, New York, 1970), p. 254.
- 10 A. Riera and A. Salin, J. Phys. B 9 , 2877 (1976). ¹¹C. F. Melius and W. A. Goddard, \overline{III} , Phys. Rev. A 10, 1541 (1974).
- 12 R. E. Olson, Phys. Rev. A 2, 121 (1970).
- ${}^{13}R$. E. Olson and A. Salop, Phys. Rev. A 16, 531 (1977).
- 14 H. Erying, J. Walter, and G. E. Kimball, Quantum Chemistry (Wiley, New York, 1964), pp. 162 and 163.
- ${}^{15}R.$ E. Olson, Phys. Rev. A 18 , 2464 (1978).
- 16 D. R. Bates and J. T. Lewis, Proc. Phys. Soc. A 68 , 173 (1955).