Multielectron effects in strong-field ionization of the oriented OCS molecule

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We present theoretical calculations of orientation-dependent total ionization yields from the highest occupied molecular orbitals of the oriented OCS molecule by solving the time-dependent Schrödinger equation in three dimensions. The calculations were performed within the single-active-electron approximation including multielectron polarization. The multielectron polarization term was represented by an induced dipole term which contains the polarizability of the OCS⁺ cation parallel to the laser polarization. Upon accounting for multielectron polarization, the calculated total ionization yields and their orientation dependence are in good agreement with experimental data.

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I. INTRODUCTION

Recent advances in strong-field physics allow identification of multielectron effects in strong-field ionization experiments [1-4] and in high-order harmonic generation experiments [5,6]. In parallel to experimental advances, theoretical models are currently being developed to account for multielectron polarization (MEP) (see, for example, Refs. [7,8]). All-electrons methods such as time-dependent Hartree-Fock theory and density functional theory (TDDFT) [9] have been used to model MEP in strong-field ionization of small atoms and molecules [2,10]. In addition to these methods, a time-dependent configuration-interaction singles (TDCIS) formalism was developed [11]. Unlike the common single-active-electron (SAE) approach, multichannel processes can be described with the TDCIS formalism. In Ref. [12], an approach based on a multielectron wave function ansatz was used for modeling of strong-field ionization from multielectron systems. This so-called time-dependent resolution in ionic states (TDRIS) approach uses field-free multielectron orbitals to describe the neutral and lowest few ionic states and allows study of multielectron effects during strong-field ionization, recollision, and high-harmonic generation. In Ref. [13], for example, the TDRIS method was successfully used to model angle-resolved strong-field ionization from the polyatomic 1,3-butadiene molecule. Notice that in the case of 1,3-butadiene, the ionization yield does not follow the orbital structure. These methods provide essential information on the contributions of different orbitals to the ionization process, electronic channel-coupling effects, and MEP effects caused by the interaction of the outgoing electron with the electron density of the parent ion.

Strong-field ionization of the OCS molecule has received considerable attention in the past decade and can be considered as an important test case for the elucidation of MEP effects [10,14–18]. For the oriented OCS molecule, while the total ionization yield (TIY) is dominated by ionization from the degenerate highest occupied molecular orbitals (HOMOs) (see Fig. 1), the ionization process is strongly affected by MEP [10]. For the OCS molecule, a series of experimental measurements was conducted for the investigation of the TIYs and their orientation dependence [10,14,15,17,18]. In Ref. [14], the experiments were conducted using linearly polarized laser pulses with 30-fs duration at 800-nm wavelength and intensities of 1.5×10^{14} and 1.8×10^{14} W/cm². In these experiments, the TIY was measured as a function of the orientation angle β (see Fig. 1, which illustrates the orientation of the OCS molecule relative to the linear laser polarization). The minimum TIY was obtained with the laser polarization parallel to the molecular axis at $\beta = 0^{\circ}$, whereas the maximum TIY was obtained with the laser polarization perpendicular to the molecular axis at $\beta = 90^{\circ}$. While the TIY is dominated by ionization from the HOMO of the OCS molecule, the orientation dependence of the TIYs presents a challenge to ionization models. For example, the molecular Ammosov-Delone-Krainov model [19] and the strong-field approximation (SFA) [20-22] predict that the orientation dependence of the TIY should follow the orbital structure. Hence in these models, the TIY should peak at β close to 35° (see the electron density plot of the HOMO in Fig. 1). Even the weak-field asymptotic theory of tunneling, which includes dipole effects [23], does not capture the experimental result and predicts a maximum at $\beta \simeq 30^{\circ}$ [16]. Likewise the Stark-corrected molecular tunneling theory [1,24] does not reproduce these experimental results for the OCS molecule: The Stark-corrected molecular tunneling theory predicts the maximum ionization yield at $\beta = 45^{\circ}$ [25]. The results of Ref. [14] suggest that the angular behavior of the TIYs in linearly polarized fields could not be explained by ionization rates only. In fact, to map out the instantaneous ionization rates of the OCS molecule, circularly polarized laser pulses were used in previous studies [1,25] because both rescattering with the parent ion and the influence of excited molecular states is minimized for this choice of polarization.

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FIG. 1. A cross section of the electron density for the HOMO of the OCS molecule in the molecular-frame xz plane. The HOMO in the yz plane can be obtained by a 90° rotation around the molecular axis. The OCS molecule is fixed at the orientation angle β relative to the linear laser polarization ε .

From the previous discussion, it is evident that an accurate description of strong-field ionization of the oriented OCS molecule by linearly polarized light requires a theory which accounts for multielectron effects and propagation of the outgoing electron in the combined potential of the remaining cation and the laser pulse. In Ref. [10], a set of measurements was presented for the OCS molecule using 800-nm linearly polarized laser pulses with 30-fs duration and peak intensities of 4.5×10^{13} , 7×10^{13} , and 2×10^{14} W/cm². In addition to the experimental measurements, the authors presented calculations of TIYs within the framework of TDDFT. The computational results were in a good agreement with the experimental measurements and explained the failure of the molecular tunneling theory to describe the orientation dependence of the TIYs by a multichannel ionization mechanism, which was revealed by the localization of a hole on the sulfur atom following strong-field ionization in the parallel orientation (at $\beta = 0^{\circ}$), a mechanism that cannot be captured by ionization models such as single-electron tunneling theory or the SFA. Although the TDDFT method captures essential parts of the dynamics of strong-field ionization of the oriented OCS molecule, it can suffer from inherent problems such as the inability to distinguish between single- and doubleionization yields, and it can be challenged in the accuracy of results at low intensities due to electron leakage [10]. In addition, it can be hard to obtain a good representation of contributions from electronically excited states in the multiphoton ionization process. This latter point is noteworthy in light of the recent results of Ref. [18], in which strong-field ionization of the oriented OCS molecule revealed a contribution from resonant excitation in the photoelectron spectra and angular distributions at different laser intensities and wavelengths.

In view of the interests and challenges in the response of the OCS molecule to an intense laser pulse, we find it interesting to revisit this problem theoretically with an improved TDSE methodology, which can handle MEP within the SAE approximation [8]. This approach will be computationally much cheaper than any multielectron methodology and will allow a clear and physically transparent identification of the influence of the MEP term. Moreover, in cases where MEP is of central importance in the dynamics such as in the OCS molecule, the performance of the present SAE approximation is expected to be of the same quality as, if not superior to, all-electron methods for the following reason: In all-electron methods, which inevitably introduce approximations for the electron-electron correlation, the accuracy of the polarizability of the ion is determined by the level of theory implemented in the method. By contrast, in the SAE model, the polarizability of the ion is an input parameter which can be chosen as the most accurate experimental or theoretical value available. Here, we calculate TIYs from the HOMO of the oriented OCS molecule as a function of the orientation angle, using the TDSE within the SAE approximation and including MEP corrections. The calculations were performed for the OCS molecule in linearly polarized laser pulses which contain five optical cycles at 800-nm wavelength and with peak intensities of 8×10^{12} , 2.2×10^{13} , and 4.5×10^{13} W/cm². The MEP was represented using an induced-dipole term based on the polarizability of the OCS⁺ ion. When the MEP term is included, the TDSE calculations predict the maximum TIY at the orientation $\beta = 90^{\circ}$ in agreement with the experimental measurements of Refs. [10,14].

The theoretical and computational models are presented in Sec. II, followed by results and discussion in Sec. III and conclusions in Sec. IV. Atomic units are used throughout unless otherwise stated.

II. THEORETICAL AND COMPUTATIONAL MODELS

This section is divided into three parts. In the first part, we describe the procedures for generating the SAE potential and the wave function for the HOMO of the OCS molecule. In the second part, we give a brief description of our TDSE method. In the final part, we describe how to include MEP in the TDSE method within the SAE approximation.

A. Single-active-electron potential and the HOMO of the OCS molecule

The SAE potential describing the OCS molecule was obtained from quantum chemistry calculations following the procedure described in Ref. [26]. The molecule is placed along the molecular-frame z axis such that the center-of-mass coincides with the origin and the oxygen (sulfur) atom points in the positive (negative) z direction. The SAE potential of the OCS molecule was expanded in partial waves as $V(\vec{r}) =$ $\sum_{l,m=0}^{l_{\text{max}}} V_{l0}(r) Y_{l0}(\theta, \phi)$ where m = 0 since the molecule is linear and the potential is invariant under rotations around the molecular axis. The expansion of the molecular potential was truncated at $l_{\text{max}} = 20$. Based on our SAE potential for the OCS molecule, the HOMO of the OCS molecule is a π orbital with energy -0.35 a.u., which is slightly above a reference orbital energy of -0.42 a.u. as obtained from quantum chemistry calculations using standard quantum chemistry software (GAMESS [27]). Since the SAE potential of the OCS molecule underestimates the magnitude of the binding energy of the HOMO orbital, the TDSE calculations will be performed at

somewhat lower intensities, compared with the experimental measurements of Ref. [10], in order for the theoretical results to be in the same ionization regime as the experimental measurements.

We applied the split-operator spectral method [28] to obtain the wave function corresponding to the HOMO of the OCS molecule. We started the time propagation from an initial guess wave function, denoted by $\psi(\vec{r}, t = 0)$, and used the SAE potential for the OCS molecule. For the initial state, we need a state that includes the π orbital symmetry and we chose the hydrogenic $2p_x$ state for convenience. We performed fieldfree propagation for a time duration of 1000 a.u. and saved the wave packet every 1 a.u. of time. From the time-dependent wave function obtained in this manner, $\psi(\vec{r}, t)$, an autocorrelation function $\mathcal{P}(t) = \langle \psi(\vec{r}, t = 0) | \psi(\vec{r}, t) \rangle$ was calculated, and a bound state spectrum $\mathcal{P}(E)$ was obtained as

$$\mathcal{P}(E) = \frac{1}{T} \int_0^T dt \, w(t) \, \exp(iEt) \, \mathcal{P}(t), \tag{1}$$

where w(t) is the Hanning window function [28]. Once the HOMO energy E_{HOMO} is well resolved in the $\mathcal{P}(E)$ spectrum, the corresponding wave function can be constructed and normalized as follows:

$$\psi_{\text{HOMO}}(\vec{r}) = \frac{1}{T} \int_0^T dt \, w(t) \, \exp(iE_{\text{HOMO}}t) \Psi(\vec{r}, t).$$
 (2)

We used the wave function of Eq. (2) as the initial state for the solution of the TDSE.

B. Summary of the TDSE calculations

The TDSE methodology was thoroughly discussed elsewhere [29]. Here, we give a brief overview of the method. The TDSE is solved for the active (HOMO) electron within the SAE approximation. The time-dependent wave function $\psi(\vec{r}, t)$ is expanded in spherical harmonics $Y_{lm}(\Omega)$ for the angular degrees of freedom and a radial grid for the timedependent reduced radial wave functions, $f_{lm}(r, t)$, i.e.,

$$\psi(\vec{r},t) = \sum_{lm} \frac{f_{lm}(r,t)}{r} Y_{lm}(\Omega).$$
(3)

The TDSE is propagated in the length gauge (LG) with a combined split-operator [28] Crank-Nicolson method. In the LG, the interaction of the laser field with the active electron is given by $V_{\text{Ext}}(\vec{r}, t) = \vec{E}(t) \cdot \vec{r}$. The electric field $\vec{E}(t)$, linearly polarized along the laboratory-frame *z* axis, is defined as

$$\vec{E}(t) = -\partial_t A(t)\hat{z} = -\partial_t \left(\frac{E_0}{\omega}\sin^2(\pi t/\tau)\cos(\omega t + \phi)\right)\hat{z}, \quad (4)$$

where $\bar{A}(t)$ is the vector potential, E_0 is the field amplitude, ω is the frequency, and ϕ is the carrier-envelope phase (CEP) for a laser pulse with duration τ . The construction of the laser electric field from the above vector potential ensures that there are no unphysical DC components in the field [30].

The TDSE calculations were performed at a laser frequency of $\omega = 0.057$ a.u., corresponding to a wavelength of 800 nm, and a CEP value of $\phi = -\pi/2$, and the pulses contained five cycles with field amplitudes ($E_0 = 0.015, 0.025,$ and 0.036 a.u.) corresponding to peak intensities of 8×10^{12} , 2.2×10^{13} , and 4.5×10^{13} W/cm². Since we consider total ionization yields and the laser pulse is sufficiently long to overcome significant short-pulse effects, the choice of CEP is not expected to affect the results. We chose somewhat lower peak intensities than what is reported experimentally in order to achieve convergence of the TDSE calculations and to be in the same ionization regime as the experimental measurements [10], in particular since our SAE potential underestimates the ionization potential of the HOMO for the OCS molecule (see Sec. II A). In the TDSE calculations, the radial grid contained 8192 points and extended to 320 a.u. The size of the angular basis set was limited by setting $l_{max} = 60$ in the partial wave expansion of the wave function [see Eq. (3)]. The calculations were performed at orientation angles in the range $\beta = 0^{\circ}-90^{\circ}$ with a step of 15°. The results have been checked for convergence by varying l_{max} .

The TIYs were produced by projecting the wave packet at the end of the laser pulse on Coulomb scattering states in the asymptotic region (r > 20 a.u.), an approach that was validated in Ref. [31] and recently applied for the molecular hydrogen ion [32,33] and the polar CO molecule [8].

C. Including the MEP term in the TDSE calculations

Theory accounting for the effect of MEP in strong-field ionization was developed in Refs. [1,24,34–36]. To include MEP in our TDSE method, we adopted the approach laid out in Ref. [36] (see also Ref. [8]). For the polar OCS molecule, the potential describing the interaction of the active electron with the core, including the MEP term and the time-dependent external field, is given asymptotically as [24,34,37]

$$V_{\rm eff}(\vec{r},t)|_{r\to\infty} = \vec{r} \cdot \vec{E}(t) - \frac{Z}{r} - \frac{(\vec{\mu}_p + \vec{\mu}_{\rm ind}) \cdot \vec{r}}{r^3} + \cdots, \quad (5)$$

where Z = 1 is the charge of the cation, and $\vec{\mu}_p$ and $\vec{\mu}_{ind}$ are the permanent and induced dipoles of the OCS⁺ cation. The MEP term is defined as $-\vec{\mu}_{ind} \cdot \vec{r}/r^3 = -\alpha_{||}\vec{E}(t) \cdot \vec{r}/r^3$, where $\alpha_{||}$ is the static polarizability of the OCS⁺ ion parallel to the laser polarization axis. The contribution of the first hyperpolarizability to a higher-order term in Eq. (5) is not considered here because the hyperpolarizability components for the OCS⁺ ion are small and their effect on the potential is negligible. A cutoff radius is chosen close to the core at a radial distance of

$$r_c = \alpha_{||}^{1/3},\tag{6}$$

such that the MEP cancels the external field at $r \leq r_c$ [34–36]. Implementing the MEP term in the TDSE method is straightforward in the LG. In our approach, upon including the MEP term, the interaction term is expressed in the LG at each radial point *r* as [8]

$$V_{\rm LG}^{\rm Ext}(r,t) = \begin{cases} \left(1 - \frac{\alpha_{||}}{r^3}\right) \vec{E}(t) \cdot \vec{r}, \ r > r_c, \\ 0, \ r \leqslant r_c. \end{cases}$$
(7)

The interaction term is zero at $r \leq r_c$ because the external field is counteracted by the MEP term [34,35]. Note that the permanent dipole term of Eq. (5) is already part of the field-free SAE potential of the OCS molecule.

The MEP term was implemented in the TDSE calculations in the LG as discussed in detail in Ref. [8]. The polarizability of the OCS^+ ion was taken from the NIST computational



FIG. 2. Orientation-dependent TIYs from the HOMOs of the OCS molecule probed by linearly polarized laser pulses containing five optical cycles at 800 nm with different models of MEP. In panel (a), the isotropic polarizability of the OCS⁺ ion is used at peak laser intensities of 8×10^{12} W/cm² (dashed curve) and 2.2×10^{13} W/cm² (solid curve). In panel (b), the external field is turned off within $r_c = \alpha_{\parallel}^{1/3}$ while neglecting the induced dipole term $[-\alpha_{\parallel}\vec{E}(t)\cdot\vec{r}/r^3]$ at peak laser intensities of 2.2×10^{13} (dashed curve) and 4.5×10^{13} W/cm² (solid curve). In panel (c), the full MEP effect is accounted for based on the parallel polarizability of the OCS⁺ ion (α_{\parallel}) at peak laser intensities of 2.2×10^{13} W/cm² (dashed curve) and 4.5×10^{13} W/cm² (solid curve). The maxima in the TIYs (normalized to unity in the plots for a better comparison) at the low and high intensities, respectively, are as follows: (a) 2.58×10^{-6} and 2.17×10^{-4} (b) 1.15×10^{-3} and 1.83×10^{-2} , and (c) 4.52×10^{-4} and 7.77×10^{-3} .

chemistry database [38], as computed using the BLYP functional of density functional theory and the cc-pVDZ basis set in the geometry of the ion. At this level of theory, the OCS⁺ ion has a static polarizability with the following nonzero components: $\alpha_{xx} = 12.56$, $\alpha_{yy} = 12.96$, and $\alpha_{zz} = 40.72$ a.u., whereas the nonzero components of the dynamic polarizability (computed at 800 nm) are as follows: $\alpha_{xx} = 13.16$, $\alpha_{yy} = 12.57$, and $\alpha_{zz} = 41.78$ a.u. Regarding the polarizability components for the OCS⁺ ion, notice that for the ion, due to spin restrictions, the spin-up and spin-down densities are slightly different, which explains the difference in polarizability components α_{xx} and α_{yy} . The small differences between the static and the dynamic polarizability components suggest that the static polarizability is suitable for the present calculations. For the static polarizability, the spherical component is 21.94 a.u., whereas the anisotropic component is 28.16 a.u. We have checked that calculating the polarizability of the OCS⁺ ion in the geometry of the neutral molecule results in very similar values for the above polarizability components. The polarizability component parallel to the laser polarization axis is defined as

$$\alpha_{||} = \cos^2(\beta)\alpha_{zz} + \sin^2(\beta)\alpha_{xx}, \qquad (8)$$

where β is the orientation angle, defined in Fig. 1. Accordingly, the value of the polarizability parallel to the laser polarization is $\alpha_{||} = \alpha_{zz} = 40.72$ a.u. at $\beta = 0^{\circ}$ ($\beta = 180^{\circ}$) when the laser polarization is parallel to the molecular axis, whereas it is $\alpha_{||} = \alpha_{xx} = 12.56$ a.u. at $\beta = 90^{\circ}$ when the laser polarization is perpendicular to the molecular axis. The large polarizability anisotropy between the orientation angles $\beta = 0^{\circ}$ and 90° is essential in order to correctly capture the MEP effect in strong-field ionization of the OCS molecule.

III. RESULTS AND DISCUSSION

In this section, we discuss the effects of MEP on the TIYs for the oriented OCS molecule. The TDSE calculations were performed within the SAE approximation including the effect of MEP. The laser pulses, linearly polarized, contained five optical cycles at a wavelength of 800 nm. The TIYs were calculated at the end of the laser pulses for the degenerate HOMOs and the overall TIY was obtained as the sum of the contributions from the two HOMOs. Notice that at $\beta = 0^{\circ}$, the external laser field runs through the molecular axis, and due to cylindrical symmetry, both HOMOs produce the same TIYs at this orientation. Although the contribution to the overall TIY from the HOMO in the molecular *yz* plane is small, it can affect the position of the orientation angle of maximum ionization yield and that is why its contribution should be considered.

As a first attempt to include MEP effects, the TDSE calculations were conducted using the isotropic polarizability of the OCS⁺ ion, $\alpha = 21.95$ a.u. [38], and its corresponding cutoff radius $r_c = 2.8$ a.u. [see Eq. (6)]. This value of the polarizability and the corresponding r_c were then applied at all orientation angles. The TDSE calculations were conducted at laser intensities of 8×10^{12} and 2.2×10^{13} W/cm². The summed-up TIYs for the degenerate HOMOs are plotted at different orientation angles in Fig. 2(a). From Fig. 2(a), one can see that the maximum ionization yield is obtained at $\beta =$ 30° at the higher laser intensity and 45° at the lower intensity. Clearly, using the isotropic polarizability of the OCS⁺ ion results in a maximum ionization yield at too small orientation angle in comparison with the experimental results [10,14], where the maximum ionization yield is expected at $\beta = 90^{\circ}$. This disagreement indicates that accounting for MEP effects using an isotropic polarizability model is not satisfactory. The results show that the polarizability anisotropy should be considered. Notice that polarizability anisotropy means that the parallel β -dependent polarizability value, α_{\parallel} of Eq. (8), and the corresponding cutoff radius, r_c of Eq. (6), should be considered.

The TDSE calculations were then conducted using the parallel polarizability of the OCS⁺ ion, which accounts for polarizability anisotropy. At first, we considered switching off the external field within a radial cutoff distance of $r_c = \alpha_{||}^{1/3}$ determined based on the polarizability of the OCS⁺ ion along the laser polarization, without including an induced dipole term in the MEP model; i.e., the term $-\alpha_{||}\vec{E}(t) \cdot \vec{r}/r^3$ in Eq. (7) was omitted. The TDSE calculations were conducted



FIG. 3. Orientation-dependent TIYs from the degenerate HOMOs of the OCS molecule probed by linearly polarized laser pulses containing five optical cycles at 800 nm with peak intensities of (a) 8×10^{12} W/cm², (b) 2.2×10^{13} W/cm², and (c) 4.5×10^{13} W/cm². The blue dashed (red dotted) curves denote the contribution from the HOMO in the *xz* plane (*yz* plane), the black curves denote the summed-up TIYs. The maxima in the TIYs (normalized to unity in the plots for a better comparison) are (a) 5.53×10^{-6} , (b) 4.52×10^{-4} , and (c) 7.77×10^{-3} .

at laser intensities of 2.2×10^{13} and 4.5×10^{13} W/cm². The summed-up TIYs for the degenerate HOMOs are presented in Fig. 2(b). From Fig. 2(b) at both intensities, although the ionization yield does not change significantly with the orientation angle, in particular, at the higher intensity, we do not observe a maximum yield at $\beta = 90^{\circ}$. The orientation angle of maximum TIY depends on laser intensity: The maximum yield is observed at $\beta = 60^{\circ}$ at 2.2×10^{13} W/cm² and at $\beta = 30^{\circ}$ at 4.5×10^{13} W/cm². Notice that, in the reported experimental measurements [10,14], the orientation angle of maximum ionization yield ($\beta = 90^{\circ}$) does not depend on laser intensity. A careful analysis of the TIYs and their orientation dependence [see Fig. 2(b)] suggests that, since the induced dipole term, $-\alpha_{\parallel}\vec{E}(t)\cdot\vec{r}/r^3$ in Eq. (7), was neglected in the TDSE calculations, the TIYs calculated at small orientation angles must be significantly overestimated, in particular, at $\beta = 0^{\circ}$. This, in turn, resulted in overestimating the contribution to the TIYs from the HOMO in the yz plane. To investigate this aspect further, our next step is to take the full MEP effect into consideration.

We account for the full MEP effect in the TDSE calculations, where both the short-range cutoff radius of Eq. (6) and the long-range induced-dipole term, $-\alpha_{||}\vec{E}(t) \cdot \vec{r}/r^3$ in Eq. (7), are included based on the β -dependent parallel polarizability of the OCS⁺ cation [see Eq. (8)]. The summed-up TIYs for the degenerate HOMOs are shown in Fig. 2(c). As can be seen, the TIY is maximum at $\beta = 90^{\circ}$, at both laser intensities, in excellent agreement with the experimental findings [14].

In Fig. 3, we compare the orientation-dependent TIYs from the degenerate HOMOs of the OCS molecule at three different intensities of 8×10^{12} , 2.2×10^{13} , and 4.5×10^{13} W/cm². In Fig. 3, in addition to the summed-up TIYs, we show the separate contributions from the degenerate HOMOs: the ionization yields corresponding to the HOMO in the molecular *xz* plane are represented by blue dashed curves, while the yields corresponding to the HOMO in the molecular *yz* plane are represented by red dotted curves. The TDSE calculations were performed using the parallel polarizability of the OCS⁺ ion of Eq. (8). From Fig. 3, one can see that at all intensities the HOMO in the molecular *xz* plane has its maximum yield at $\beta = 90^{\circ}$ with the external field perpendicular to the molecular axis, whereas the HOMO in the molecular *yz* plane has its maximum yield at $\beta = 0^{\circ}$ with the external field parallel to the molecular axis. Moreover, the maximum overall TIY is obtained at $\beta = 90^{\circ}$ at all intensities. While the intensity has no effect on the orientation angle of maximum TIY, it definitely has an effect on the ratio (\mathcal{R}) between the ionization yields at $\beta = 0^{\circ}$ and $\beta = 90^{\circ}$, where \mathcal{R} is defined as $\mathcal{R} =$ $\frac{\text{TIY}(\beta=0^\circ)}{\text{TIY}(\beta=90^\circ)}$ [14]. Based on the results in Fig. 3, the ratio of ionization yields has values of $\mathcal{R} \approx 0.3$ at 8×10^{12} W/cm², $\mathcal{R} \approx$ $0.6 \text{ at } 2.2 \times 10^{13} \text{ W/cm}^2$, and $\mathcal{R} \approx 0.8 \text{ at } 4.5 \times 10^{13} \text{ W/cm}^2$. This trend in \mathcal{R} and its dependence on laser intensity is consistent with both the TDDFT calculations and the experimental measurements on the OCS molecule reported in Ref. [10], in which the ratio between the ionization yields at $\beta = 0^{\circ}$ and $\beta = 90^{\circ}$ increases with increasing intensities. Our estimates of \mathcal{R} at the higher intensities are comparable to the experimental value of 0.65 [14] determined at a laser intensity of $1.5 \times 10^{14} \text{ W/cm}^2$.

IV. CONCLUSION

For the oriented OCS molecule, TIYs were obtained from TDSE calculations for the degenerate HOMOs within the SAE approximation and taking into account MEP. In the treatment of MEP, the induced dipole term was first calculated based on the isotropic polarizability of the OCS⁺ ion, in which case the resultant ionization yields failed to reproduce the experimental measurements [10,14]. However, when the induced dipole term was calculated based on the full polarizability of the OCS⁺ ion, in which the polarizability parallel to the laser polarization and the corresponding cutoff radius depends on the orientation angle, a good agreement was obtained between our calculated TIYs and the experimental measurements [10,14]. The calculations were performed at several intensities and the agreement with the experiment is satisfactory at all considered intensities. These findings show that the SAE methodology can capture fundamental aspects of multielectron dynamics in strong-field ionization of the OCS molecule when suitably extended to account for the rearrangement of the remaining core-electrons as described by the induced dipole potential.

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