Atomic Hyperfine Structure. IV Positive and Negative First-Row lons

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Using restricted Hartree-Fock (RHF) wave functions and two previously defined types of configuration interaction (CI) wave functions, the polarization wave function and the first-order wave function, we have made ab initio calculations of the hyperfine structure (hfs) of: ${}^{2}P C^{\dagger}$; P and ¹D N⁺; ⁴S, ²D, and ²P O⁺; ³P and ¹D F⁺; ²P Ne⁺; ³P and ¹D B⁻; ⁴S, ²D, and ²P C⁻; ³P and ¹D N⁻, ²P O⁻. For the polarization wave functions, the calculated parameters $\langle r_l^{-3} \rangle$, $\langle r_{S}^{-3} \rangle$, and $\langle r_{a}^{-3} \rangle$ are expected to be within 2% of the exact values, and the calculated spin densities are expected to be between 55% and 80% of the exact values. All calculated spin densities except that for ${}^{3}P$ B⁻ are positive, and the polarization spin density is usually slightly larger than the first-order value. For all positive ions and neutral atoms, the relationship $\langle {\gamma_q}^{-3} \rangle$ < $\langle {\gamma_l}^{-3} \rangle$ < $\langle {\gamma_l}^{-3} \rangle_{\rm RHF}$ < $\langle {\gamma_g}^{-3} \rangle$ holds. Calculated diamagnetic susceptibilities and nuclear magnetic shielding constants are reported for both the ions and the corresponding neutral atoms, using the present CI wave functions. Finally, we tabulate the angular integration coefficients needed to compute $\langle {r_S}^{-3}\rangle$ and $\langle {r_q}^{-3}\rangle,$ and discuss a procedure used in this work to compact the form of a CI wave function.

INTRODUCTION

This is the last of a series of papers, $1, 2$ the purpose of which is to predict atomic hyperfine structure (hfs) from ab initio configuration interaction (CI) wave functions, the polarization wave function (Ref. la), and the first-order wave function (Ref. 1b). The polarization wave function includes the restricted Hartree-Fock (RHF) configuration plus all singly-excited configurations $X_i - Y_i$, where X_i is a RHF occupied orbital and Y_i is an orbital not occupied in the RHF approximation. The firstorder wave function includes all the configurations in the polarization function plus double excitations of the type $X_i X_j - X_k X_l$ and $X_i X_j - X_k Y_l$.

In the previous papers in this series, it was shown that, in general, the polarization wave function yields hfs parameters in better agreement with experiment than does the more accurate (in an energy sense) first-order wave function. To be specific, it was found (Ref. 1a) that polarization function values of the hfs parameters $\langle r_1^{-3} \rangle$ and $\langle r_s^{-3} \rangle$ were within 2% of experiment³ for both ³P Q and ²P F. Spin densities $|\Psi(0)|^2$ calculated from polarization wave functions were between 54 and 75% of experiment.

Since the publication of the first two' papers in this series, measurements of the hfs constants A_2 and A_1 have been made⁴ for ³P C¹³. Although a rigorous determination of the three magnetic hfs parameters awaits the determination of a third constant, a preliminary analysis' indicates that $\langle \eta^{-3} \rangle$ and $\langle \eta^{-3} \rangle$ differ by 2% from the polarization function values (Ref. 1a), and that the experimental spin density is 120% of the polarization function spin density. These measurements⁴ give further evidence that polarization wave functions provide a powerful theoretical tool for the analysis of atomic hfs.

Since the previous papers^{1, 2} contain detailed comparisons of the different theoretical approaches used in the study of hfs, we only note here that the only other method which yields hfs parameters as accurately as the polarization wave function is the Brueckner- Goldstone many- body perturbation theory⁶ (MBPT). Kelly's results⁶ indicate that effective values of $\langle r^{-3} \rangle$ from MBPT are in about the same agreement with experiment as are polarization function values, but that spin densities can be more accurately predicted using MBPT. However, computation of a polarization wave function and hfs parameters requires about 1 h of card punching time and less than 10 min of IBM 360/65 computer time. An MBPT hfs study' of the same atom, however, can require 2 years of work by a competent graduate student.

Although no experiments have yet been carried out to determine the hfs parameters of positive and negative first-row atomic ions, it now appears likely⁸ that for at least one system, O^+ , paramagnetic-resonance experiments can soon be undertaken. It is hoped that the present calculations

 $\overline{1}$

will stimulate further experimental studies. The field of hfs is one that provides considerable opportunity for interaction between theory and experiment, in order both to understand the applicability of certain models of electronic structure and to predict nuclear moments.

In addition to the hfs parameters reported here, diamagnetic susceptibilities and nuclear magnetic shielding constants were also calculated. These quantities are reported for both the ions and the neutral atoms of the first row.

During this study, a procedure was developed to substantially simplify the form of a CI wave function after said wave function has been determined. This procedure is discussed, and also reported are the angular integration coefficients required to compute $\langle r_s^{-3} \rangle$ and $\langle r_a^{-3} \rangle$.

COMPACTING OF CONFIGURATION INTERACTION WAVE FUNCTIONS

The problem of obtaining as accurate as possible a wave function using as few configurations as possible has long been of interest in atomic and molecular physics. In particular, considerable advances⁹ have been made using Löwdin's idea
of natural orbitals.¹⁰ However, there is a fur of natural orbitals.¹⁰ However, there is a furthe simple concept which can be used to greatly simplify the form of a complicated CI wave function. When one makes a serious effort to obtain a large percentage of the correlation energy of an atom,

there are many orbital occupancies which give rise to more than one L-S eigenfunction of the rise to more than one $L-S$ eigenfunction of the proper symmetry.¹¹ Here we point out that the wave function can be expressed in a form in which the total number of configurations is just the number of orbital occupancies. This can be a drastic simplification, since for as simple an atom as oxygen, the important $1s^2 2s 2p^3 d_1 f_1$ orbital occupancy gives rise to 18 linearly independent ${}^{3}P$ configurations.

A CI wave function can be written

$$
\Psi = \sum_{\alpha, i} C_{\alpha i} \Phi_{\alpha i}, \qquad (1)
$$

where $\Phi_{\alpha i}$ is the *i*th configuration corresponding to orbital occupancy α . $\Phi_{\alpha i}^{}$ has the form

$$
\Phi_{\alpha i} = \sum_{j}^{n} b_{\alpha i j} D_{\alpha j} , \qquad (2)
$$

where $D_{\alpha j}$ is the jth determinant arising from the ath orbital occupancy. So if there are m_{α} configurations $\Phi_{\alpha 1}$, $\Phi_{\alpha 2}$, ..., $\Phi_{\alpha m \alpha}$ corresponding to orbital occupancy α , then all m_{α} of these configurations are expressed in terms of the same Slater determinants $D_{\alpha 1}$, $D_{\alpha 2}$, ..., $D_{\alpha n_{\alpha}}$. Given the coefficients $C_{\alpha i}$ in Eq. (1), it is possible to compact these m_{α} configurations $\Phi_{\alpha i}$ into a single configuration Φ_{α} .

The contribution of these m_Q configurations to the final wave function is, from Eqs. (1) and (2) ,

$$
C_{\alpha 1} \sum_{j=1}^{n_{\alpha}} b_{\alpha 1j} D_{\alpha 1j} + C_{\alpha 2} \sum_{j=1}^{n_{\alpha}} b_{\alpha 2j} D_{\alpha j} + \cdots + C_{\alpha m_{\alpha}} \sum_{j=1}^{n_{\alpha}} b_{\alpha m_{\alpha}j} D_{\alpha j} = \sum_{j=1}^{n_{\alpha}} \sum_{k=1}^{m_{\alpha}} C_{\alpha k} b_{\alpha k j} D_{\alpha j} \tag{3}
$$

If we let

$$
S_{\alpha} = \left(\sum_{j=1}^{n_{\alpha}} \sum_{k=1}^{m_{\alpha}} C_{\alpha k} b_{\alpha k j}\right)^2
$$

 n_{α} n_{α}

the final wave function can be written in the simplified form

$$
\Psi = \sum_{\alpha} S_{\alpha}^{1/2} \Phi_{\alpha} \quad , \tag{4}
$$

where

$$
\Phi_{\alpha} = \sum_{j=1}^{n_{\alpha}} t_{\alpha j} D_{\alpha j} , \qquad (5)
$$

in which

$$
t_{\alpha j} = S_{\alpha}^{-1/2} \sum_{k=1}^{m_{\alpha}} C_{\alpha k} b_{\alpha k j} . \tag{6}
$$

The compacted wave function of Eq. (4) is identical to that of Eq. (1), but the compact wave function has only one configuration for each orbital occupancy. In our computer programs, ¹² the compact wave function is used to compute expectation values. For the ${}^{2}D$ state of C^- , for example, the 195-configuration wave function of Eq. (1) becomes a 65-configuration wave function in Eq. (4) .

(4) ANGULAR INTEGRALS FOR HYPERFINE STRUCf URE

In order to evaluate the orbital and spin-dipolar terms of the hfs Hamiltonian, it is necessary to compute the integrals

$$
E(l_i m_i; l_j m_j) =
$$

$$
\int_0^{2\pi} \int_0^{\pi} Y_{l_i m_i}(\theta, \phi) (3 \cos^2 \theta - 1)
$$

$$
\times Y_{l_m}(\theta, \phi) \sin \theta \, d\theta \, d\phi \quad . \tag{7}
$$

Since the values of these angular integration coefficients do not appear in the literature, it should be useful to tabulate them here. Using

 $(3\cos^2\theta-1)=\left(\frac{16}{5}\pi\right)^{1/2}Y_{20}(\theta, \phi)$

and the standard formula¹³ for the integral of the product of three spherical harmonics, we find

$$
\begin{split} E(l_{\,}m_{\,i};l_{\,j}m_{\,j}) = 2[\;(2l_{\,j}+1)/(2l_{\,i}+1)\,]^{1/2} \\ \times C(l_{\,j}\,2l_{\,i}\,;\;m_{\,j}0m_{\,i}) C(l_{\,j}\,2l_{\,i}\,;000)\;\;,\;\;(8) \end{split}
$$

where the $C(l_1 l_2 l_3; m_1 m_2 m_2)$ are the Clebsch-Gordan coefficients as defined by Rose.¹³ Using tabulated values¹⁴ of the Clebsch-Gordan coefficients, we have evaluated all possible E coefficients for s, p , d, and f electrons. The nonvanishing angular integrals are given in Table I.

WAVE FUNCTIONS AND ENERGIES

The first-order wave functions used here were reported previously in a study¹⁵ of the electron affinities of first- row atoms. Polarization functions were calculated in the present work from the
previously reported basis sets.¹⁵ Energies of previously reported basis sets. Energies of

RHF, ¹⁶ polarization, and first-order wave function are given in Table II.

TABLE II. Calculated energies in hartrees, for RHF, polarization, and first-order wave functions. RHF energies are those of Clementi (Ref. 16).

State	RHF	Polarization	First-order
^{2}P C ⁺	$-37,29221$	$-37,318,34$	$-37,363$ 87
3P N^+	-53.887 97	-53.93377	-53.959 27
\mbox{N}^+ 1_D	-53.80735	$-53,854$ 78	-53.881 07
\mathbf{N}^+ ^{1}S	$-53,690$ 08	$-53,690$ 92	-22 $-53,776$
σ^* 4_S	-74.37255	-74.428 90	$-74,433$ 67
\cdot O ⁺ ^{2}D	-74.23329	-74.293 25	-11 -74.298
^{2}P Ω^+	-74.14204	-74.17161	-74.218 22
3P F^+	$-98,83161$	$-98,88175$	$-98,884$ 82
$^{\rm 1}D$ \mathbf{F}^+	$-98,728,46$	$-98,778,43$	-98.78167
^{1}S F^+	$-98,575,57$	-98.576 44	$-98,645$ 33
^{2}P Ne^+	-127.81765	$-127,848$ 43	-127.84998
3P B^-	-24.51919	-24.545 40	$-24,564$ 93
1D B^-	-24.49048	-24.515 79	-24.535 48
^{1}S $\, {\bf B}^-$	-24.43950	-24.453 36	-24.494 87
^{4}S C^-	-37.708 78	$-37,748$ 44	-37.754 82
^{2}D C^{-}	-37.64252	-37.684 26	-37.69175
^{2}P C^{-}	$-37,600$ 84	-37.624 74	-37.65355
3P N^-	$-54,32189$	-54.362 87	$-54,366,50$
1D N^-	$-54,266,87$	$-54,304$ 89	-54.30922
1S N^-	$-54,186$ 82	-54.187 74	-54.233 90
2 $\bm{\mathcal{P}}$ O^{-}	$-74,78948$	-74.816 05	-74.817 78

TABLE I. Nonvanishing values of the coefficients $E(l_i m_i; l_j m_j) = \int_0^{2\pi} \int_0^{\pi} Y_{l_i m_i}(\theta, \phi) (3 \cos^2 \theta - 1) Y_{l_j m_j}(\theta, \phi) \sin \theta d\theta d\phi$.

State	$\langle r_l^{-3} \rangle$	—ა $\langle r_{s}$	mа $\langle r_{q}$	$ \Psi(0) ^2$
D_N^-	2.33185	NA	2.33185	0.0
	2.35846	NA	2.07785	0.0
	2.41349	NA	2.15152	0.0
^{2}P O ⁻	4.055 08	4.055 08	4.055 08	0.0
	3.73535	4.33058	3,420 04	0.04533
	3.71193	4.30235	3.45305	0.04719

TABLE III. (continued).

For 2D and 2P seven electron systems, an $\langle r_{\bm q}^{-3}\rangle$ analysis is not applicable, but there is a small electric–quadrupo hyperfine interaction. For these states we give the calculated values of the expectation value $\langle J=L+S, M_A=J|\sum_{i}$ $(3\cos^2\theta_i-1)/r_i^3|J=L+S,M_J=J\rangle$.

hfs Results

Reference 1(a) contains a complete discussion of the formulas needed to predict atomic hyperfine structure. The previous papers^{1, 2} of this series presented calculated expectation values, reduced matrix elements, hfs parameters, and hfs constants A and B . For the sake of brevity, we report here only the hfs parameters $\langle r_1^{-3} \rangle$, $\langle r_s^{-3} \rangle$, $\langle r_q^{-3} \rangle$, and $\left[\Psi(0) \right]^2$. Using previously discussed relationships,^{1, 2} all other hfs information can be derived from these four hfs parameters.

Table III contains values of $\langle r_l^{-3} \rangle$, $\langle r_s^{-3} \rangle$, $\langle r_q^{-3} \rangle$, and $|\Psi(0)|^2$ obtained from RHF, polarization and first-order wave functions.

It is interesting to note that all the calculated spin densities are positive except for ${}^{3}P$ B, where both polarization and first-order spin densities are negative. The polarization-wave-function spin 'density $\vert\Psi(0)\vert^2$ is usually slightly larger than the first-order spin density.

Several interesting relationships can be seen in Table III between the calculated parameters $\langle r_1^{-3} \rangle$, $\langle r_s^{-3} \rangle$, and $\langle r_q^{-3} \rangle$. For both polarization and first-order wave functions, the calculate positive ion parameters obey

$$
\langle r_q^{-3} \rangle < \langle r_l^{-3} \rangle < \langle r^{-3} \rangle_{RHF} < \langle r_s^{-3} \rangle . \tag{9}
$$

It was pointed out in Paper Ithat this relationship also holds for the ground states of the neutral firstrow atoms. The only states for which Eq. (9) does not hold are some of the negative ions, which can have very irregular electron distributions. In paper III it was pointed out that the hfs parameters are not, in general, transferable between different

states of the same atom (e.g., 4S , 2D , and 2P N). This is confirmed for ions in Table III. However, there do seem to be certain consistencies in hfs parameters computed for isoelectronic systems of the same symmetry. For example, for ${}^{2}DO^{+}$, ${}^{2}DN$, and ${}^{2}DC$, polarization and first-order wave functions yield nearly identical calculated electric field gradients. However, for ${}^{2}PO^{+}$, ${}^{2}PN$, and ${}^{2}PC^{-}$, the polarization and first-order values of $\langle (3\cos^2\theta - 1)/$ r^3 differ by more than a factor of 2.

DIAMAGNETIC SUSCEPTIBILITIES AND NUCLEAR MAGNETIC SHIELDING CONSTANTS

In the course of this work we routinely computed two additional properties. The molar diamagnet susceptibility ${}_\chi$ (d) in cm $^3/$ mole is given by 17

$$
\chi^{(d)} = -\frac{1}{6} N \alpha^2 a_0^3 \langle \Psi | \Sigma_i r_i^2 | \Psi \rangle, \qquad (10)
$$

where N is Avogadro's number, α is the finestructure constant, and α_0 is the Bohr radius. The diamagnetic contribution $\sigma^{(d)}$ to the nuclear magnetic shielding is given $by¹⁸$

$$
\sigma^{(d)} = \frac{1}{3} \alpha^2 \langle \Psi | \Sigma_i (1/r_2) | \Psi \rangle \quad . \tag{11}
$$

In Table IV, we tabulate calculated values of χ ^(d) and $\sigma^{(d)}$ for positive ions, neutral atoms, and negative ions of the first row. The RHF values of Malli and Fraga¹⁹ are included for comparison in Table IV.

Although the theorem of Møller and Plesset²⁰ does not apply to open-shell RHF wave functions,

TABLE IV. Calculated diamagnetic susceptibilitie) and diamagnetic contributions $\sigma^{(d)}$ to the nuclear magnetic shielding. As in Table II, values are given for RHF, polarization, and first-order wave functions. χ (d) is given in 10⁻⁶ cm³/mole and σ ^(d) in 10⁻⁵ a.u. RHF

	values are those of Malli and Fraga (Ref. 19).			1S C	12.153	25.965
		$\sigma^{(d)}$			12,200	25.962
	$\chi^{(d)}$				11.729	26.009
^{2}P C ⁺	6.388	25.107	$4S$ N		9.565	32.547
	6.361	25.106			9.533	32.543
	6.282	25,120			9.573	32.537
3P N ⁺	6.138	31.430	2D N		9.918	32.484
	6.124	31.428			9.852	32.483
	6.108	31.433			9.914	32.476
${}^{1}D$ N ⁺	6.267	31.391	^{2}P N		10.175	32,441
	6.239	31,391			10.204	32.435
	6.228	31.395			10.093	32,455
$^1S \quad N^+$	6.490	31.328	${}^{3}P$ O		8.846	39.511
	6.492	31,328			8.852	39.504
	6.356	31.364			8.874	39.500
$4s$ O ⁺	5.802	38.350	${}^{1}D$ O		9.008	39.468
	5.799	38.346			8,979	39.468
	5.967	38.318			9.000	39.464
^{2}D O ⁺	5.932	38,290	1S O		9.264	39.405
	5.915	38.289			9.271	39.403
	5.928	38,285			9.155	39,435
^{2}P O ⁺	6.026	38,249	^{2}P F		8.110	47.071
	6.029	38.246			8.107	47.067
	5.988	38,261			8.117	47.065
3P F ⁺	5.604	45.766		${}^{3}P$ B ⁻	31.529	20.651
	5.612	45.759			29.487	20.663
	5.616	45.757			30.991	20.650
1_D F^+	5.666	45.726		$1_D B^-$	40.898	20.578
	5.662	45.724			35.928	20.607
	5.668	45.721			38.888	20,584
1_S F^+	5.767	45.667		1S B ⁻¹	29.126	20.622
	5.776	45.663			44.508	20.424
	5.730	45.687			37.087	20.548
^{2}P Ne ⁺	5.352	53.775		$4s$ c	20,830	26.696
	5.354	53.772			20,353	26.700
	5.356	53.771			20.833	26.683
^{2}P B	12.556	20.199	2D C		23.315	26.617
	12.429	20.198			22.700	26.625
	12.276	20.213			23,469	26.602
${}^{3}P$ C	10.930	26.074		^{2}P C ⁻	25.481	26,562
	10.847	26.073			26.539	26.501
	10.843	26.076			25.320	26.579
${}^{1}D$ C	11.354	26.031		${}^{3}P$ N ⁻	17.665	33.205
	11.215	26,033			17.978	33.184
	11.248	26.035			18.044	33.179

 $\sigma^{(d)}$

TABLE IV. (continued).

 $\chi^{(d)}$

a similar theorem does apply to polarization wave functions for open-shell systems. Thus, we can

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IV indicates that for the less complicated expectation values $\langle 1/r \rangle$ and $\langle r^2 \rangle$, RHF, polarization, and first-order wave functions usually yield very similar results (typically the three values agree within $1\%)$. We conclude that even for open-shell systems expectation values such as $\langle 1/r \rangle$ and $\langle r^2 \rangle$ are quite adequately predicted by the RHF method.

rigorously expect polarization wave functions to give expectation values of one-electron operators to "first-order" accuracy. %e have seen that for the one-electron operators that give rise to hfs. polarization wave functions yield results very different from the RHF results. However, Table

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