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### Magnetic Moment of Os<sup>189</sup>†

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A NUCLEAR induction signal of Os<sup>189</sup> with a width of 10 gauss, measured between the minima of the differentiated dispersion mode, has been detected in molten OsO<sub>4</sub>. The liquid phase of this particular compound was chosen because the Os<sup>189</sup> nucleus has a large electric quadrupole moment,  $Q = (+2.0 \pm 0.8) \times 10^{-24}$  cm<sup>2</sup>,<sup>1</sup> and it is therefore necessary to place the nucleus in a symmetric molecular configuration in order to minimize interactions which broaden the resonance line. It is presumed that in our sample, the osmium nucleus is located at the centroid of a tetrahedron having oxygen atoms at its vertices.

The resonant frequency was compared to that of Cl<sup>35</sup> in pure TiCl<sub>4</sub> with the result

$$\nu(\text{Os}^{189})/\nu(\text{Cl}^{35}) = 0.791896 \pm 0.000093. \quad (1)$$

With the spin of Os<sup>189</sup> assumed to be 3/2,<sup>1</sup> and with the known values of the frequency ratios  $\nu(\text{Cl}^{35})$  in TiCl<sub>4</sub>/ $\nu(\text{Cl}^{35})$  in RbCl,<sup>2</sup>  $\nu(\text{Cl}^{35})$  in RbCl/ $\nu(\text{H}^2)$ ,<sup>3</sup>  $\nu(\text{H}^2)/\nu(\text{H}^1)$ ,<sup>4</sup> the value of the magnetic moment was found to be, with  $\mu(\text{H}^1) = 2.79268$ ,<sup>5</sup>

$$\mu(\text{Os}^{189}) = +0.650655 \pm 0.000081 \text{ nm}. \quad (2)$$

The positive sign in (2) was verified by comparing the sign of the Os<sup>189</sup> signal with that of O<sup>17</sup> in the same compound. Sign comparisons were also made with H<sup>2</sup> and N<sup>14</sup>. The earlier determination of  $\mu(\text{Os}^{189}) = +0.70 \pm 0.09$  nm by Murakawa and Suwa<sup>1</sup> is in agreement with the sign and the more precise value of Eq. (2).

In view of the fact that the single-particle shell model<sup>6</sup> predicts that, for the case of a positive magnetic moment, only  $p_{1/2}$ ,  $f_{5/2}$ , and  $h_{9/2}$  states are available between the magic numbers 82 and 126, it seemed to us of interest to check the previous spin determination by an independent method. Accordingly, the heights  $h$  and line widths  $\Delta H$  of Os<sup>189</sup> and Cl<sup>37</sup> signals were compared.

The reference Cl<sup>37</sup> signal from a 7.10 molar LiCl solution containing 0.0075 molar Mn(NO<sub>3</sub>)<sub>2</sub> was used, and care was taken to ascertain that both line shapes represented the nonsaturated slow passage case. If the frequency, the rf field intensity, the  $Q$  of the receiver coil, the filling factor, and the sweep field remain unchanged for the measurements of both signals, the application of the phenomenological equations<sup>7</sup> gives

$$I(\text{Os})[I(\text{Os})+1] = I(\text{Cl})[I(\text{Cl})+1] \\ \times \frac{h(\text{Os})}{h(\text{Cl})} \frac{N(\text{Cl})}{N(\text{Os})} \frac{\gamma(\text{Cl})}{\gamma(\text{Os})} \left( \frac{\Delta H(\text{Os})}{\Delta H(\text{Cl})} \right)^2. \quad (3)$$

With the known value of 3/2 for  $I(\text{Cl})$ ,<sup>8</sup> the measured values of the number of nuclei  $N$  in each sample, and our experimental results for the other ratios, we find

$$I(\text{Os}^{189}) = 1.45 \pm 0.13, \quad (4)$$

thus verifying the value 3/2.

Signals were not observable in solidified OsO<sub>4</sub> or in powdered Os metal which has a crystal structure of hexagonal symmetry, presumably because of an unfavorable ratio  $T_2/T_1$ .

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### Experimental Study of the $\mu^-$ Meson Mass and the Vacuum Polarization in Mesonic Atoms\*

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STUDIES of the x-rays emitted in transitions of mesons between atomic orbits about nuclei have been extended<sup>1</sup> to  $3D-2P$  and  $4F-3D$  transitions in a variety of elements for both  $\pi^-$  and  $\mu^-$  mesons. Particular attention has been paid to  $\mu^-$  mesonic transitions having energies below 90 kev, using thin filters between the anticoincidence counter and the NaI crystal of the scintillation spectrometer.<sup>2</sup> Because of the large and rapid change in absorption cross section at the photoelectric "K edge" energy and the precise knowledge of