

## Isotope Shift in the $^{58}\text{Ce II}$ Spectrum\*

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(Received September 30, 1953)

The spectrum of Ce II from enriched samples of  $\text{Ce}^{136}$  and  $\text{Ce}^{138}$  has been studied and compared with the same spectrum from natural Ce. Only a negligible isotope shift, certainly less than 6 percent of that between the isotopes with 82 and 84 neutrons, has been found between the isotopes with 78, 80, and 82 neutrons.

THE isotope shift between  $\text{Ce}^{140}$  and  $\text{Ce}^{142}$  has been measured by Brix and Frank<sup>1</sup> in some of the lines  $4f^26s\ ^4H$ ,  $^2H$ ,  $^4F-4f^26p\ ^4I^\circ$ ,  $^4H^\circ$ ,  $^2H^\circ$ ,  $^4G^\circ$ . An attempt was made by Murakawa and Ross<sup>2</sup> to measure a shift between  $\text{Ce}^{138}$  and  $\text{Ce}^{140}$  in the  $4f^26s\ ^4H_{9/2}-4f^26p\ ^4I_{9/2}^\circ$  line by using an enriched (4.4 percent) sample of  $\text{Ce}^{138}$ . Murakawa and Ross estimated an upper limit of this shift at about the half of the  $\text{Ce}^{140}-\text{Ce}^{142}$  shift, which was found by Brix and Frank to be  $0.054\pm 0.003\text{K}$ ,<sup>3</sup> with the  $\text{Ce}^{142}$  towards lower frequency.

A new attempt has been made to measure not only the  $\text{Ce}^{138}-\text{Ce}^{140}$  shift, but also the  $\text{Ce}^{136}-\text{Ce}^{140}$  shift, in the lines where Brix and Frank measured the shift between  $\text{Ce}^{140}$  and  $\text{Ce}^{142}$ . Enriched samples<sup>4</sup> of  $\text{Ce}^{138}$  (13.1 percent) and  $\text{Ce}^{136}$  (first 22.3 percent, later 30 percent) were available.

The experimental setup consisted of 3 parallel hollow cathode tubes<sup>5</sup> and a modified Steinheil three-prism spec-

trograph (designed and built by Mr. George Streander at Wisconsin) with a Perot-Fabry interferometer in the parallel beam between the collimator and the prisms. The spacers used in the interferometer had the sizes: 15, 18, 19, 20, 25, 27, 30, and 40 mm. The coating on the plates was silver in most of the exposures and an aluminum-magnesium alloy in the others. The reflectivity exceeded 90 percent.

The shift between  $\text{Ce}^{140}$  and  $\text{Ce}^{142}$  could be measured in all the lines with all three samples and the value agrees with the value found by Brix and Frank. It was, however, impossible to find any structure from the  $\text{Ce}^{138}$  and  $\text{Ce}^{136}$  isotopes, which lines seem to coincide with the  $\text{Ce}^{140}$  lines. Not even a line broadening was detected. A quantitative measurement of the lines gave the following values for the shifts:

$$\text{Ce}^{136}-\text{Ce}^{140}: 0.0003\pm 0.0032\text{K} \text{ (33 measurements),}$$

$$\text{Ce}^{138}-\text{Ce}^{140}: 0.0002\pm 0.0035\text{K} \text{ (12 measurements),}$$

with the lighter isotopes in each case towards higher frequency. The uncertainty includes the largest deviation from the average.

The author wishes to express his sincere appreciation to Dr. J. E. Mack for his continued interest in the experiment.

\* Supported by the U. S. Office of Naval Research.

<sup>1</sup> P. Brix and H. Kopfermann, *Z. Physik* **127**, 289 (1950).

<sup>2</sup> K. Murakawa and J. S. Ross, *Phys. Rev.* **83**, 1272 (1951).

<sup>3</sup> 1K(kayser)  $\equiv 1\text{ cm}^{-1}$ , following a recommendation of the international Joint Commission for Spectroscopy [see *J. Opt. Soc. Am.* **43**, 410 (1953)].

<sup>4</sup> Produced by the Y-12 plant, Carbide and Carbon Chemicals Corporation, and obtained by allocation from the U. S. Atomic Energy Commission.

<sup>5</sup> H. Arroe and J. E. Mack, *J. Opt. Soc. Am.* **40**, 386 (1950).