the cooperative effort of the entire staff of the 60-inch cyclotron group, and the continued interest of Professor Glenn T. Seaborg and Professor Ernest O. Lawrence in this program.

- \* Part of this work was done under the auspices of the U. S. Atomic
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  <sup>1</sup> L. W. Alvarez, Phys. Rev. 58, 192 (1940).
  <sup>2</sup> R. Condit, Phys. Rev. 62, 301 (1942).
  <sup>3</sup> York, Hildebrand, Putnam, and Hamilton, Phys. Rev. 70, 446 (1946).
  <sup>4</sup> Miller, Hamilton, Putnam, Haymond, and Rossi, Phys. Rev. 80, 486 (1950).
- <sup>4</sup> Miller, frammon, 2 and 1950).
  <sup>5</sup> J. F. Miller, Phys. Rev. 83, 1261 (1951).
  <sup>6</sup> Ghiorso, Thompson, Street, and Seaborg, Phys. Rev. 81, 154 (1951).
  <sup>7</sup> C. A. Tobias, Phys. Rev. 85, 764 (1952).
  <sup>8</sup> Chackett, Fremlin, and Walker, Proc. Phys. Soc. (London) A66, 495 (1952).

- <sup>8</sup> Chackett, Freimin, and Freimin, Nature 171, 189 (1953).
  <sup>9</sup> D. Walker and J. H. Fremlin, Nature 171, 189 (1953).
  <sup>10</sup> L. D. Wyly and A. Zucker, Phys. Rev. 89, 524 (1953).
  <sup>11</sup> J. M. Hollander, University of California Radiation Laboratory Unclassified Report UCRL-1396, July, 1951 (unpublished).
  <sup>12</sup> R. R. Wilson, Phys. Rev. 56, 459 (1939).
  <sup>13</sup> L. Marquez and I. Perlman, Phys. Rev. 78, 189 (1950).

## Reactions of U<sup>238</sup> with Cyclotron-Produced Nitrogen Ions<sup>\*</sup>

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HE acceleration of  $N^{14}(+6)$  ions with the Berkeley Crocker Laboratory 60-inch cyclotron<sup>1</sup> has made it possible to study nuclear reactions of these ions with U238.

The following transmutation products have been observed: 99247(?), 99246, Cf244, Cf246, Cf247(?), Cf248, Bk243, and other berkelium isotopes not yet identified. The identification of the elements was definitely established by their carrying on lanthanum fluoride precipitates and by their order of elution from a Dowex-50 ion exchange column.

The observed nuclear properties of these nuclides are summarized in Table I.

The nuclides Cf<sup>244</sup>, Cf<sup>246</sup>, Cf<sup>248</sup>, Bk<sup>243</sup>, and Bk<sup>245</sup> have previously been observed in this laboratory.2-4

The yields of the transcurium nuclides were low even though bombardment currents of 0.1 microampere of  $N^{14}(+6)$  ions of energy greater than 100 Mev were available. In three separate experiments a total of 40 alpha-emitting atoms of the 7.3-minute TABLE I. Nuclides produced by U238 plus N14 ions.

Nuclide	Half-life	Radiation	Alpha energy (Mev)	Remarks
99247(?) 99246	7.3 min minutes	EC(?), α EC	7.35	Observed only through growth of its 1.5-day Cf <sup>246</sup> daughter
Cf <sup>244</sup> Cf <sup>246</sup> Cf <sup>247</sup> (?)	45 min 35.7 hr ∼2.7 hr	α, EC(?) α EC	7.15 6.75	
Cf <sup>248</sup> Bk <sup>243</sup>	225 day 4.6 hr	α ΕC, α	6.26 6.72 (30%) 6.55 (53%) 6.20 (17%)	
Bk	days	EC	0.20 (11 /6)	Observed K x-rays; probably unresolved mixture of Bk <sup>245</sup> and Bk <sup>246</sup>

isotope of element 99 were observed to decay in the ion exchange column fraction immediately preceding californium, namely the eka-holmium position. Thus, the element identification is certain though the mass number can only be inferred on the basis of nuclear systematics. By observations of the abundant fission product activity it was found that almost all of the nuclear reactions of nitrogen ions with U238 resulted in fission much as in the case of carbon-ion bombardment of the same nucleus.

It is a pleasure to acknowledge the continued help and encouragement of Professor Joseph G. Hamilton, Director of the Crocker Laboratory. Our grateful thanks are extended to William B. Jones and the members of the 60-inch cyclotron operating crew for their cooperation in making the many bombardments necessary for this work. Special thanks are due Dr. Gregory Choppin for his valuable assistance with some of the chemical separations. It is a privilege to acknowledge that this work was accomplished with the always helpful guidance of Professor Glenn T. Seaborg. The continued interest and encouragement of Professor Ernest O. Lawrence is gratefully acknowledged.

\* This work was performed under the auspices of the U. S. Atomic Energy Commission.
<sup>1</sup> Rossi, Jones, Hollander, and Hamilton, preceding letter [Phys. Rev. 93, 256(L) (1954)].
<sup>2</sup> Hollander, Perlman, and Seaborg, Revs. Modern Phys. 25, 469 (1953).
<sup>3</sup> E. K. Hulet, Ph.D. thesis, University of California Radiation Laboratory Unclassified Report UCRL-2283, July, 1953 (unpublished).
<sup>4</sup> There is unpublished information relevant to element 99 at the University of California, Argonne National Laboratory, and Los Alamos Scientific Laboratory. Until this information is published the question of the first preparation should not be prejudged on the basis of this paper.