# The Spins of Li<sup>6</sup> and B<sup>10</sup> in the Shell Model

NISSAN ZELDES Department of Physics, Hebrew University, Jerusalem, Israel (Received December 18, 1952)

The intermediate coupling approximation succeeds in explaining the different spins of Li<sup>6</sup> and B<sup>10</sup>, where the LS coupling and the jj coupling fail.

### INTRODUCTION

HE calculations of the energy levels of Li<sup>6</sup> and  $B^{10}$  in the  $p^n$  shell model have shown that in LS coupling the ground states of these nuclei belong to J=1, and in *jj* coupling to  $J=3.^{1}$  The experimental spins are 1 for Li<sup>6</sup> and 3 for B<sup>10</sup>. It seemed probable, therefore, that an intermediate coupling approximation would succeed,<sup>2</sup> where the extreme couplings failed.

In order to see this, the matrices of the binding energy of these two nuclei for J=1 and for J=3 had to be calculated in this approximation, and the levels with highest binding energy compared.

In the present model and approximation the energy, apart from the central field energy which is common to all the configuration and which can be overlooked. therefore, when comparing the levels, consists of two parts considered together as a perturbation on the central field energy: the energy of the nuclear forces proper, which is taken as the sum of the interaction energies between all pairs of the nucleons, and the spin-orbit interaction which is written analogously to the atomic case. The elements of the energy matrix will therefore be linear combinations of three parameters,  $F_0$ ,  $F_2$ , and  $\zeta_p$ , where the F's are Slater's generalized parameters and  $\zeta_p$  is the spin-orbit parameter. Therefore, the energy values will be, in appropriate units, functions of two ratios only.

If the binding energies are described by surfaces in 3-dimensional space, the experimental spins will show that for  $\zeta_p$  very large as compared with the F's the surface for J=3 will be higher than that for J=1, or  $E_3 > E_1$ , while for  $\zeta_p$  small the contrary happens. There is, therefore, a line in the finite part of the plane, where  $E_1 = E_3$ . And in order that both spins will be accounted for by the theory, the line for Li<sup>6</sup> should be on the side of the higher  $\zeta$  with respect to that for B<sup>10</sup>, at least for some values of  $F_2/F_0$ .

In the case of Li<sup>6</sup> the matrix is of order three for J=1 and reduces to one element for J=3; therefore, one can find the equation of the curve in terms of the above ratios of the parameters explicitly. In B<sup>10</sup> the matrices are of the tenth order, and so an explicit equation cannot be found; but from the equation for Li<sup>6</sup>, the value of the critical  $\zeta_0$ , for which  $E_1 = E_3$  (in terms of  $F_0$  and  $F_2$ ), has to be substituted for  $\zeta$  in the matrices of B<sup>10</sup>, the eigenvalues of which may be expressed, in arbitrary units, as functions of the ratio  $F_2/F_0$ . Varying this ratio one obtains each time two numerical matrices the eigenvalues of which are to be compared.

Between which values is the ratio to be varied? Owing to the rarity of experimental material, very little is known of Slater's nuclear parameters. They are known to be positive by definition, and the ratio  $F_2/F_0$ is known to depend upon the range of the nuclear forces, being small for long-range forces, and reaching the value 0.2 for  $\delta$ -interaction. As the range of the nuclear forces is only little known, we explored all the interval between 0 and 0.2.

# THE INTERMEDIATE COUPLING APPROXIMATION

## 1. The Energy Matrices

Rosenfeld shows<sup>3</sup> that, if the interaction between two nucleons is given by a linear combination of the interactions of Wigner, Majorana, Bartlett, and Heisenberg, with the same distance dependence in each, the coefficients of the combination are approximately determined by the saturation requirements of the nuclear forces. The combination which results is

$$V = -\frac{2}{15}V_W + \frac{14}{15}V_M + \frac{7}{15}V_B - \frac{4}{15}V_H, \qquad (1)$$

and we shall suppose this form for the nuclear potential. The part of the energy due to the nuclear forces is then found from the tables of Racah.<sup>4</sup>

The spin-orbit interaction has the form

$$\zeta_p \Sigma_i (\mathbf{I}_i \cdot \mathbf{s}_i), \qquad (2)$$

and the calculation reduces to that of  $\Sigma_i(\mathbf{l}_i \cdot \mathbf{s}_i)$ . A general formula for the matrices of such operators has been given by Racah.<sup>5</sup> In the practical utilization of it we used further the property of the coefficients of fractional parentage to decompose into factors, each of which depends only on part of the quantum numbers specifying the states.<sup>6</sup> These factors had already been calculated by Racah,7 and were used by us for the

<sup>&</sup>lt;sup>1</sup> E. Feenberg, Phys. Rev. 76, 1275 (1949).

<sup>&</sup>lt;sup>2</sup> G. Racah and N. Zeldes, Phys. Rev. 79, 1012 (1950).

<sup>&</sup>lt;sup>3</sup> L. Rosenfeld, Nuclear Forces (Interscience Publishers, New York, 1948), p. 234. <sup>4</sup> G. Racah, Helv. Phys. Acta 23, 229 (1950). <sup>5</sup> G. Racah, Phys. Rev. 63, 367 (1943), Eq. (23). <sup>6</sup> G. Racah, Phys. Rev. 76, 1352 (1949); Princeton Notes, 1051 - 66 (curved) is held in the second second

<sup>&</sup>lt;sup>7</sup>G. Racah, 1195. Rev. 76, 1532 (1949); Frinceton Notes, 1951, p. 66 (unpublished).
<sup>7</sup>G. Racah (private communication). Mean while extensive tables of these coefficients were published by H. A. Jahn in Proc. Roy. Soc. (London) 205, 192 (1951).

				(a) 6	$0 \Sigma_i (\mathbf{l}_i \cdot \mathbf{s}_i)$	for $J = 1$				
5751	$(222)_{13S}$	(321)	(321)	(321)	(321)	A <sup>a</sup>	$B^{a}$	(420)	(420)	(420)
	3	-•1	<i>D</i>	-• <i>T</i>	<i>D</i>	<i>T</i>			<i>D</i>	<i>D</i>
(222) <sup>13</sup> S	0	$30(3)^{\frac{1}{2}}$	0	50(3)	0	0	0	0	0	0
(321) <sup>15</sup> P	$30(3)^{\frac{1}{2}}$	0	15(3)	0	-3(15	) 1 0	0	-50	2(35)	0
(321) <sup>15</sup> D	0	$15(3)^{\frac{1}{2}}$	0	$45(3)^{\frac{1}{2}}$	0	0	0	0	0	-30
(321) <sup>13</sup> $P$	$50(3)^{\frac{1}{2}}$	0	$45(3)^{\frac{1}{2}}$	0	-(15)	-60(2	$2)^{\frac{1}{2}} = 0$	-30	$-6(35)^{\frac{1}{2}}$	0
(321) <sup>13</sup> D	0	$-3(15)^{\frac{1}{2}}$	0	$-(15)^{\frac{1}{2}}$	0	0	12(30)	0	0	$18(5)^{\frac{1}{2}}$
A <sup>11</sup> $P$	0	0	0	$-60(2)^{\frac{1}{2}}$	0	0	0	0	0	$30(6)^{\frac{1}{2}}$
$B \stackrel{11}{\longrightarrow} P$	0	0	0	0	12(30)	3 0	0	$40(2)^{\frac{1}{2}}$	2(70)	0
(420) <sup>13</sup> S	0	-50	0	-30	0	. 0	$40(2)^{\frac{1}{2}}$	0	0	0
(420) <sup>13</sup> D	0	2(35)*	0	-6(35)*	0	0	2(70)*	0	0	6(105)*
(420) <sup>13</sup> $D'$	0	0	-30	0	18(5)	a 30(6)	)* 0	0	6(105)*	0
				(b) 21	10 $\Sigma_i(\mathbf{l}_i \cdot \mathbf{s}_i)$	for $J=3$				
	(222)	(321)	(321)	(321)	A	В	(420)	(420)	(420)	(420)
$\Sigma TSL$	`17 <i>S</i> ´	15P'	15D'	$^{13}D'$	$^{11}F$	$^{11}F$	$^{13}D'$	13D'	13F'	$^{13}G$
(222) 17S	0	210(2) <sup>1</sup> / <sub>2</sub>	0	0	0	0	0	<b>0</b>	0	0
(321) <sup>15</sup> <i>P</i>	$210(2)^{\frac{1}{2}}$	0	$105(2)^{\frac{1}{2}}$	$-63(15)^{\frac{1}{2}}$	0	0	$42(35)^{\frac{1}{2}}$	0	0	0
(321) <sup>15</sup> D	0	105(2)*	0	0	0	0	0	$-70(6)^{\frac{1}{2}}$	70(3)	0
(321) <sup>13</sup> D	0	$-63(15)^{\frac{1}{2}}$	0	0	0	-42(30)	0	$-42(5)^{\frac{1}{2}}$	-42(10)	0
$A \stackrel{11}{\longrightarrow} F$	0	0	0	0	0	0	0	70(6)*	-70(3)	0
$B$ $^{11}F$	0	0	0	$-42(30)^{\frac{1}{2}}$	0	0	18(70)*	0	0	-90(7)*
(420) <sup>13</sup> D	0	42(35)*	0	0	0	18(70)*	0	$-14(105)^{\frac{1}{2}}$	16(210)*	0
(420) <sup>13</sup> D'	0	0	$-70(6)^{\frac{1}{2}}$	$-42(5)^{\frac{1}{2}}$	70(6)*	0	$-14(105)^{\frac{1}{2}}$	0	0	0
(420) <sup>13</sup> <i>F</i>	0	0	70(3)*	$-42(10)^{\frac{1}{2}}$	$-70(3)^{\frac{1}{2}}$	0	16(210)*	0	0	$-45(21)^{\frac{1}{2}}$
(420) <sup>13</sup> G	0	0	0	0	0	-90(7) <sup>1</sup>	0	0	-45(21)*	0

TABLE I. The matrix of  $\Sigma_i(\mathbf{l}_i \cdot \mathbf{s}_i)$  of  $p^6$  in the LS scheme.

 $^{\mathbf{a}}\psi(A) = \left[\psi(411) + \psi(330)\right] / (2)^{\frac{1}{2}}, \psi(B) = \left[\psi(411) - \psi(330)\right] / (2)^{\frac{1}{2}}.$ 

calculation of the spin-orbit matrices for the charge singlets of  $p^6$  with J=1 and J=3, which are given in Tables I(a) and I(b). The states are labeled by  $\Sigma$ , the partition specifying Wigner's supermultiplet to which the state belongs, and by 2T+1, 2S+1 and L, T and S being the isotopic and ordinary spin numbers and Lthe orbital quantum number.  $\alpha = \frac{1}{2}\zeta_p$  is the positive spin-orbit parameter. For  $p^2$  the matrices are well known.8

# 2. The Curve for Li<sup>6</sup>

The energy matrix for the charge singlet with J=1 is

where the  $E_L$  stands for the nuclear binding energy of the corresponding L term without spin-orbit interaction. For J=3 the matrix has only one element:

$$^{13}D = E_D + 2\alpha. \tag{3'}$$

Subtracting this element from the diagonal elements of the matrix for J=1 one obtains a matrix

...

$$\begin{vmatrix} -5\alpha & -(10/3)^{\frac{1}{2}}\alpha & 0 \\ -(10/3)^{\frac{1}{2}}\alpha & E_P - E_D - 2\alpha & 2(2/3)^{\frac{1}{2}}\alpha \\ 0 & 2(2/3)^{\frac{1}{2}}\alpha & E_S - E_D - 2\alpha \end{vmatrix} , \quad (4)$$

all the eigenvalues of which are negative when  $\alpha$  is vary large, while for a small  $\alpha$  there is also a positive one. The transition occurs at the critical value  $\alpha_0$  for which there is a zero eigenvalue, and therefore the determinant (4) vanishes. Developing this determinant we obtain for  $\alpha_0$  the value

$$\alpha_0 = \frac{3(E_D - E_P)(E_S - E_D)}{10E_D - 6E_P - 4E_S},$$
(5)

and inserting the values of the terms in Rosenfeld's mixture (1),

$$E_D = F_0 + F_2, E_P = -(9/5)(F_0 - 5F_2), E_S = F_0 + 10F_2,$$
 (6)

one obtains finally

$$\alpha_0 = \frac{9}{14} \cdot F_2 \cdot \frac{7F_0 - 20F_2}{F_0 - 5F_2}.$$
 (7)

# 3. Limiting Cases

The value of  $\alpha_0$  from Eq. (7) was now substituted for  $\alpha$  in the energy matrices of B<sup>10</sup>, and the highest binding energies of J=1 and J=3 were compared.

The end points of the interval of variation of the ratio  $F_2/F_v$  could be dealt with easier than the inner part of the interval, and so will be considered first:

## A. Long-Range Limit: $F_2 \ll F_0$

Neglecting higher powers of  $F_2/F_0$  we get, from (7),

$$\alpha_0 = (9/2)F_2,\tag{8}$$

and we shall regard  $F_2$  as a perturbation. In this approximation it is sufficient to diagonalize that part of

<sup>&</sup>lt;sup>8</sup> E. U. Condon and G. H. Shortley, Theory of Atomic Spectra (Cambridge University Press, Cambridge, 1935), p. 268.

TABLE II. The nonvanishing elements of the nuclear forces binding energy matrix of  $p^6$  in the LS scheme.

Row and column	The element							
(a)	J = 1							
Diagonal elements								
(222) <sup>13</sup> S	$-(5/3)F_0+(275/6)F_2$							
(321) <sup>15</sup> P	$3 F_0 + 36 F_2$							
(321) <sup>15</sup> D	$3 F_0 + (168/5)F_2$							
(321) <sup>13</sup> P	$(17/15)F_0 + (644/15)F_2$							
$(321)^{-13}D$	$(17/15)F_0 + (2968/75)F_2$							
$A$ $^{\text{II}}P$	$3 F_0 + (231/5)F_2$							
$\overline{B}$ $^{11}P$	$3 F_0 + (243/5)F_2$							
(420) <sup>13</sup> S	$(29/5)F_0 + (487/10)F_2$							
$(420)$ $^{13}D$	$(29/5)F_{0} + (979/25)F_{0}$							
$(420)$ $^{13}D'$	$(29/5)F_0 + (203/5)F_0$							
(120) D								
Nondiago	nal elements							
$(222)^{13}S - (420)^{13}S$	$(3/2)(3)^{\frac{1}{2}}F_{2}$							
$(321)^{13}D - (420)^{13}D$	$(12/25)(21)^{\frac{1}{2}}F_2$							
(b)	J = 3							
Diagona	l elements							
(222) <sup>17</sup> S	$3 F_0 + 30 F_2$							
(321) <sup>15</sup> P	$3 F_0 + 36 F_2$							
(321) <sup>15</sup> D	$3 F_0 + (168/5)F_2$							
$(321)^{13}D$	$(17/15)F_0 + (2968/75)F_2$							
$A^{11}F$	$3 F_0 + (186/5)F_2$							
$\overline{B}$ $^{11}F$	$3 F_0 + (168/5)F_2$							
(420) <sup>13</sup> D	$(29/5)F_0 + (979/25)F_0$							
$(420)$ ${}^{13}D'$	$(29/5)F_0 + (203/5)F_0$							
(420) <sup>13</sup> F	$(29/5)F_0 + (158/5)F_0$							
(120) 13C	$(29/5)F_0 + (116/5)F_0$							
(420) 0	$(23/3)P_0 + (110/3)P_2$							
Nondiagor	nal elements							
$(321)^{13}D - (420)^{13}D$	$(12/25)(21)^{\frac{1}{2}}F_{2}$							
() -	(// - 2							

it follows that this unperturbed value is that of the super-multiplet (420), both for J=1 and for J=3.

The secular equations for this supermultiplet are of order 3 and 4 for J=1 and J=3, respectively. When the value (8) for  $\alpha_0$  is inserted in them and the calculation performed, one obtains

$$E_1 = (29/5)F_0 + 49.13F_2,$$
  

$$E_3 = (29/5)F_0 + 49.77F_2,$$
(9)

and we see that the binding energy for J=3 is higher than for J=1.

# B. Short-Range Limit: $F_0 - 5F_2 \ll F_0$

Here  $\alpha_0$  becomes infinite of the first order, and we need to go over to the jj scheme. The matrix of  $\alpha$  becomes diagonal, and the matrices of the coefficients of  $F_0$  and  $F_2$  are given in Tables III(a), (b) and IV(a), (b), where the states are labeled by the numbers of nucleons having  $j=\frac{1}{2}$  and  $j=\frac{3}{2}$ , though this specification is not complete.

Here we consider  $F_0$  and  $F_2$  as a perturbation to  $\alpha$ . Neglecting  $F_0 - 5F_2$ , we obtain for the eigenvalues in the first approximation of the perturbation theory

$$E_1 = E_3 = 6\alpha_0 + 54F_2, \tag{10}$$

and we have to pass to the second approximation. The usual formula gives

$$E_{1} = 6\alpha_{0} + 54F_{2} + (107/45)(F_{0} - 5F_{2}) + (146/5)(F_{2}^{2}/\alpha_{0}),$$
  

$$E_{3} = 6\alpha_{0} + 54F_{2} + (59/15)(F_{0} - 5F_{2}) + (79/5)(F_{2}^{2}/\alpha_{0}).$$
(11)

the energy matrix which belongs to the maximal unperturbed eigenvalue, and from Tables II(a) and (b)

TABLE III. The matrix of the coefficients of  $F_0/45$  of  $p^6$  in the *jj* scheme.

	(a) $J=1$									
	$p_{\frac{3}{2}}^6$	$(p_{\frac{1}{2}}^{5} p_{\frac{1}{2}})_{A}$	$(p_{\frac{3}{2}} p_{\frac{1}{2}})_B$	$(p_{\frac{3}{2}}^4 p_{\frac{1}{2}}^2)_A$	$(p_{\frac{3}{2}} p_{\frac{1}{2}} p_{\frac{1}{2}})_B$	$(p_{\frac{3}{2}}^4 p_{\frac{1}{2}}^2)_C$	$(p_{\frac{3}{2}}^4 p_{\frac{1}{2}}^{12})_D$	$(p_{\frac{3}{2}}^3 p_{\frac{1}{2}}^3)_A$	$(p_{\frac{3}{2}}^3 p_{\frac{1}{2}}^{13})_B$	$p_{\frac{3}{2}}^2 p_{\frac{1}{2}}^4$
P36	107	14(6) <sup>1/2</sup>	56	$28(2)^{\frac{1}{2}}$	$14(3)^{\frac{1}{2}}$	14(14) <sup>1</sup> /2	14(3) <sup>1</sup> / <sub>2</sub>	0	0	0
$(p_{\frac{5}{2}} p_{\frac{1}{2}})_A$	$14(6)^{\frac{1}{2}}$	177	0	$14(3)^{\frac{1}{2}}$	$-28(2)^{\frac{1}{2}}$	$-14(21)^{\frac{1}{2}}$	$14(2)^{\frac{1}{2}}$	0	$14(6)^{\frac{1}{2}}$	0
$(p_{\frac{3}{2}}^5 p_{\frac{1}{2}})_B$	56	0	107	7(2)1	$14(3)^{\frac{1}{2}}$	$-7(14)^{\frac{1}{2}}$	$28(3)^{\frac{1}{2}}$	$14(6)^{\frac{1}{2}}$	42	0
$(p_{\frac{3}{2}}^4 p_{\frac{1}{2}}^2)_A$	$28(2)^{\frac{1}{2}}$	$14(3)^{\frac{1}{2}}$	$7(2)^{\frac{1}{2}}$	149	0	0	0	$14(3)^{\frac{1}{2}}$	$7(2)^{\frac{1}{2}}$	$28(2)^{\frac{1}{2}}$
$(p_{\frac{3}{2}} p_{\frac{1}{2}} p_{\frac{1}{2}})_B$	$14(3)^{\frac{1}{2}}$	$-28(2)^{\frac{1}{2}}$	$14(3)^{\frac{1}{2}}$	0	149	0	0	$28(2)^{\frac{1}{2}}$	$-14(3)^{\frac{1}{2}}$	$-14(3)^{\frac{1}{2}}$
$(p_{\frac{3}{2}}^4 p_{\frac{1}{2}}^2)c$	$14(14)^{\frac{1}{2}}$	$-14(21)^{\frac{1}{2}}$	$-7(14)^{\frac{1}{2}}$	0	0	65	0	$14(21)^{\frac{1}{2}}$	$7(14)^{\frac{1}{2}}$	$-14(14)^{\frac{1}{2}}$
$(p_{\frac{1}{2}}^4 p_{\frac{1}{2}}^2)_D$	$14(3)^{\frac{1}{2}}$	$14(2)^{\frac{1}{2}}$	$28(3)^{\frac{1}{2}}$	0	0	0	205	$-14(2)^{\frac{1}{2}}$	$-28(3)^{\frac{1}{2}}$	$-14(3)^{\frac{1}{2}}$
$(p_{\frac{3}{2}}^{3} p_{\frac{1}{2}}^{3})_{A}$	0	0	$14(6)^{\frac{1}{2}}$	$14(3)^{\frac{1}{2}}$	28(2) <sup>1</sup> / <sub>2</sub>	$14(21)^{\frac{1}{2}}$	$-14(2)^{\frac{1}{2}}$	177	0	$14(6)^{\frac{1}{2}}$
$(p_{\frac{3}{2}}^3 p_{\frac{1}{2}}^3)_B$	0	$14(6)^{\frac{1}{2}}$	42	$7(2)^{\frac{1}{2}}$	$-14(3)^{\frac{1}{2}}$	$7(14)^{\frac{1}{2}}$	$-28(3)^{\frac{1}{2}}$	0	107	56
P3 <sup>2</sup> P3 <sup>4</sup>	0	0	0	$28(2)^{\frac{1}{2}}$	$-14(3)^{\frac{1}{2}}$	$-14(14)^{\frac{1}{2}}$	$-14(3)^{\frac{1}{2}}$	$14(6)^{\frac{1}{2}}$	56	107
					(b) <i>J</i> =3			1		
	₽3 <sup>6</sup>	$(p_{\frac{1}{2}}^{5} p_{\frac{1}{2}})_{A}$	$(p_{\frac{3}{2}}^5 p_{\frac{1}{2}})_B$	$(p_{\frac{3}{2}}^4 p_{\frac{3}{2}}^2)_A$	$(p_{3}^{4} p_{3}^{2})_{B}$	$(p_{\frac{3}{2}}^4 p_{\frac{1}{2}}^2)_C$	$(p_{\frac{3}{2}}^4 p_{\frac{1}{2}}^2)_D$	$(p_{\frac{3}{2}}^3 p_{\frac{1}{2}}^3)_A$	$(p_{\frac{3}{2}}^{3} p_{\frac{1}{2}}^{3})_{B}$	$p_{\frac{3}{2}}^2 p_{\frac{1}{2}}^4$
\$P 36	177	$-12(14)^{\frac{1}{2}}$	$-4(21)^{\frac{1}{2}}$	$14(3)^{\frac{1}{2}}$	$12(7)^{\frac{1}{2}}$	8(21) <sup>1</sup> / <sub>2</sub>	$14(3)^{\frac{1}{2}}$	0	0	0
$(p_{\frac{3}{2}}^{5} p_{\frac{1}{2}})_{A}$	$-12(14)^{\frac{1}{2}}$	177	0	$-2(42)^{\frac{1}{2}}$	$-6(2)^{\frac{1}{2}}$	$-4(6)^{\frac{1}{2}}$	$2(42)^{\frac{1}{2}}$	12	$16(6)^{\frac{1}{2}}$	0
$(p_{\frac{3}{2}} p_{\frac{1}{2}})_B$	$-4(21)^{\frac{1}{2}}$	0	177	$12(7)^{\frac{1}{2}}$	$12(3)^{\frac{1}{2}}$	24	$-12(7)^{\frac{1}{2}}$	16(6)	30	0
$(p_{\frac{3}{2}}^4 p_{\frac{1}{2}}^2)_A$	$14(3)^{\frac{1}{2}}$	$-2(42)^{\frac{1}{2}}$	$12(7)^{\frac{1}{2}}$	219	0	0	0	$-2(42)^{\frac{1}{2}}$	$12(7)^{\frac{1}{2}}$	$14(3)^{\frac{1}{2}}$
$(p_{\frac{4}{2}}^4 p_{\frac{1}{2}}^2)_B$	12(7)3	$-6(2)^{\frac{1}{2}}$	$12(3)^{\frac{1}{2}}$	0	135	0	0	$6(2)^{\frac{1}{2}}$	$-12(3)^{\frac{1}{2}}$	$-12(7)^{\frac{1}{2}}$
$(p_{\frac{3}{2}}^4 p_{\frac{3}{2}}^2)_C$	$8(21)^{\frac{1}{2}}$	$-4(6)^{\frac{1}{2}}$	24	0	0	135	0	$4(6)^{\frac{1}{2}}$	-24	$-8(21)^{\frac{1}{2}}$
$(p_{\frac{3}{2}}^4 p_{\frac{1}{2}}^2)_D$	14(3)	2(42)3	$-12(7)^{\frac{1}{2}}$	0	0	0	219	$-2(42)^{\frac{1}{2}}$	$12(7)^{\frac{1}{2}}$	$-14(3)^{\frac{1}{2}}$
$(p_{3}^{3} p_{3}^{3})_{A}$	0	12	$16(6)^{\frac{1}{2}}$	$-2(42)^{\frac{1}{2}}$	$6(2)^{\frac{1}{2}}$	$4(6)^{\frac{1}{2}}$	$-2(42)^{\frac{1}{2}}$	177	0	-12(14)
$(p_{3}^{3} p_{3}^{3})_{B}$	0	16(6)3	30	$12(7)^{\frac{1}{2}}$	$-12(3)^{\frac{1}{2}}$	-24	$12(7)^{\frac{1}{2}}$	0	177	$-4(21)^{\frac{1}{2}}$
<i>P</i> <sup>2</sup> <i>P</i> <sup>4</sup>	0	0	0	14(3)3	$-12(7)^{\frac{1}{2}}$	$-8(21)^{\frac{1}{2}}$	$-14(3)^{\frac{1}{2}}$	$-12(14)^{\frac{1}{2}}$	$-4(21)^{\frac{1}{2}}$	177

	(a) Coefficients of $F_2/45$ for $J=1$									
	Þ3 <sup>6</sup>	$(p_{3}^{5} p_{3})_{A}$	$(p_{\frac{1}{2}}^{5} p_{\frac{1}{2}})_{B}$	$(p_{\frac{3}{2}}^4 p_{\frac{1}{2}}^2)_A$	$(p_{\frac{3}{2}}^4 p_{\frac{1}{2}}^2)_B$	$(p_{3}^{4} p_{3}^{2})c$	$(p_{\frac{3}{2}}^4 p_{\frac{1}{2}}^2)_D$	$(p_{\frac{3}{2}}^{3} p_{\frac{3}{2}}^{3})_{A}$	$(p_{\frac{1}{2}}^3 p_{\frac{1}{2}}^3)_B$	p32 p34
P36	1895	$2(6)^{\frac{1}{2}}$	26	$-50(2)^{\frac{1}{2}}$	65(3) <sup>1</sup> / <sub>2</sub>	$-25(14)^{\frac{1}{2}}$	$-25(3)^{\frac{1}{2}}$	0	0	0
$(p_{\frac{3}{2}} p_{\frac{1}{2}})_A$	$2(6)^{\frac{1}{2}}$	1905	$-36(6)^{\frac{1}{2}}$	65(3) <sup>1</sup>	$-31(2)^{\frac{1}{2}}$	$-11(21)^{\frac{1}{2}}$	$-70(2)^{\frac{1}{2}}$	-90	$-25(6)^{\frac{1}{2}}$	0
$(p_{\frac{3}{2}} p_{\frac{1}{2}})_B$	26	$-36(6)^{\frac{1}{2}}$	1994	$-35(2)^{\frac{1}{2}}$	$20(3)^{\frac{1}{2}}$	$35(14)^{\frac{1}{2}}$	22(3)	$-25(6)^{\frac{1}{2}}$	60	0
$(p_{\frac{3}{2}}^4 p_{\frac{3}{2}}^2)_A$	$-50(2)^{\frac{1}{2}}$	$65(3)^{\frac{1}{2}}$	$-35(2)^{\frac{1}{2}}$	1802	0	0	0	$65(3)^{\frac{1}{2}}$	$-35(2)^{\frac{1}{2}}$	$-50(2)^{\frac{1}{2}}$
$(p_{\frac{3}{2}}^4 p_{\frac{1}{2}}^2)_B$	$65(3)^{\frac{1}{2}}$	$-31(2)^{\frac{1}{2}}$	$20(3)^{\frac{1}{2}}$	0	1874	$36(42)^{\frac{1}{2}}$	-72	$31(2)^{\frac{1}{2}}$	$-20(3)^{\frac{1}{2}}$	$-65(3)^{\frac{1}{2}}$
$(p_{\frac{3}{2}}^4 p_{\frac{1}{2}}^2)_C$	$-25(14)^{\frac{1}{2}}$	$-11(21)^{\frac{1}{2}}$	35(14)3	0	36(42)	1898	0	$11(21)^{\frac{1}{2}}$	$-35(14)^{\frac{1}{2}}$	25(14)3
$(p_{\frac{3}{2}}^4 p_{\frac{1}{2}}^2)_D$	$-25(3)^{\frac{1}{2}}$	$-70(2)^{\frac{1}{2}}$	$22(3)^{\frac{1}{2}}$	0	-72	0	1792	$70(2)^{\frac{1}{2}}$	$-22(3)^{\frac{1}{2}}$	$25(3)^{\frac{1}{2}}$
$(p_{\frac{3}{2}}^{3} p_{\frac{1}{2}}^{3})_{A}$	0	-90	$-25(6)^{\frac{1}{2}}$	$65(3)^{\frac{1}{2}}$	$31(2)^{\frac{1}{2}}$	$11(21)^{\frac{1}{2}}$	70(2)3	1905	$-36(6)^{\frac{1}{2}}$	2(6)
$(p_{\frac{3}{2}}^{3} p_{\frac{1}{2}}^{3})_{B}$	0	$-25(6)^{\frac{1}{2}}$	60	$-35(2)^{\frac{1}{2}}$	$-20(3)^{\frac{1}{2}}$	$-35(14)^{\frac{1}{2}}$	$-22(3)^{\frac{1}{2}}$	$-36(6)^{\frac{1}{2}}$	1994	26
$p_{\frac{3}{2}}^2 p_{\frac{3}{2}}^4$	0	0	0	$-50(2)^{\frac{1}{2}}$	$-65(3)^{\frac{1}{2}}$	25(14)	$25(3)^{\frac{1}{2}}$	$2(6)^{\frac{1}{2}}$	26	379
				(b) Coeffici	ents of $F_2/3$	15 for $J=3$				
	₽\$ <sup>6</sup>	$(p_{\frac{1}{2}} p_{\frac{1}{2}})_A$	$(p_{\frac{1}{2}}^5 p_{\frac{1}{2}})_B$	$(p_{\frac{3}{2}}^4 p_{\frac{1}{2}}^2)_A$	$(p_{\frac{3}{2}}^4 p_{\frac{1}{2}}^2)_B$	$(p_{\frac{1}{2}}^4 p_{\frac{1}{2}}^2)c$	$(p_{\frac{3}{2}}^4 p_{\frac{3}{2}}^2)_D$	$(p_{\frac{3}{2}}^3 p_{\frac{1}{2}}^3)_A$	$(p_{\frac{3}{2}} p_{\frac{1}{2}} p_{\frac{1}{2}})_B$	P3 <sup>2</sup> P3 <sup>4</sup>
\$P 36	10.815	96(14) <sup>1</sup> / <sub>2</sub>	$-94(21)^{\frac{1}{2}}$	$-175(3)^{\frac{1}{2}}$	$-150(7)^{\frac{1}{2}}$	$-100(21)^{\frac{1}{2}}$	$455(3)^{\frac{1}{2}}$	0	0	0
$(p_{\frac{1}{2}}^{5} p_{\frac{1}{2}})_{A}$	96(14) <sup>1</sup> / <sub>2</sub>	10.383	432(6)3	$70(42)^{\frac{1}{2}}$	$-168(2)^{\frac{1}{2}}$	140(6)	$-70(42)^{\frac{1}{2}}$	795	$-200(6)^{\frac{1}{2}}$	0
$(p_{\frac{3}{2}}^{5} p_{\frac{1}{2}})_{B}$	$-94(21)^{\frac{1}{2}}$	432(6)	11.310	210(7) <sup>1</sup> / <sub>2</sub>	$-420(3)^{\frac{1}{2}}$	-588	42(7)	$-200(6)^{\frac{1}{2}}$	570	0
$(p_{3}^{4} p_{3}^{2})_{A}$	$-175(3)^{\frac{1}{2}}$	70(42) <sup>1</sup> / <sub>2</sub>	$210(7)^{\frac{1}{2}}$	10.794	0	0	0	70(42)3	210(7)	$-175(3)^{\frac{1}{2}}$
$(p_{\frac{3}{2}}^4 p_{\frac{3}{2}}^2)_B$	$-150(7)^{\frac{1}{2}}$	$-168(2)^{\frac{1}{2}}$	$-420(3)^{\frac{1}{2}}$	0	10.584	0	$216(21)^{\frac{1}{2}}$	168(2)3	$420(3)^{\frac{1}{2}}$	150(7)3
$(p_{\frac{3}{2}}^4 p_{\frac{3}{2}}^2)c$	$-100(21)^{\frac{1}{2}}$	$140(6)^{\frac{1}{2}}$	- 588	0	0	11.466	432(7) <sup>3</sup>	$-140(6)^{\frac{1}{2}}$	588	100(21)3
$(p_{\frac{1}{2}}^4 p_{\frac{1}{2}}^2)_D$	$455(3)^{\frac{1}{2}}$	$-70(42)^{\frac{1}{2}}$	42(7)	0	$216(21)^{\frac{1}{2}}$	$432(7)^{\frac{1}{2}}$	10.668	$70(42)^{\frac{1}{2}}$	$-42(7)^{\frac{1}{2}}$	$-455(3)^{\frac{1}{2}}$
$(p_{\frac{3}{2}}^{3} p_{\frac{1}{2}}^{3})_{A}$	0	795	$-200(6)^{\frac{1}{2}}$	70(42) <sup>1</sup> / <sub>2</sub>	$168(2)^{\frac{1}{2}}$	$-140(6)^{\frac{1}{2}}$	$70(42)^{\frac{1}{2}}$	10.383	432(6)3	96(14) <sup>3</sup>
$(p_{\frac{1}{2}}^{3} p_{\frac{1}{2}}^{3})_{B}$	0	$-200(6)^{\frac{1}{2}}$	570	$210(7)^{\frac{1}{2}}$	$420(3)^{\frac{1}{2}}$	588	$-42(7)^{\frac{1}{2}}$	$432(6)^{\frac{1}{2}}$	11.310	$-94(21)^{\frac{1}{2}}$
$p_{\frac{3}{2}}^2 p_{\frac{1}{2}}^4$	0	0	0	$-175(3)^{\frac{1}{2}}$	150(7)3	100(21) <sup>1</sup> / <sub>2</sub>	$-455(3)^{\frac{1}{2}}$	96(14) <sup>1</sup> / <sub>2</sub>	$-94(21)^{\frac{1}{2}}$	10.815

TABLE IV. The matrix of coefficients of  $F_2$  of  $p^6$  in the jj scheme.

After inserting for  $\alpha_0$  in the last terms its value in this approximation, which is seen from (7) to be

$$\alpha_0 = \frac{135}{14} \cdot \frac{F_2^2}{F_0 - 5F_2},\tag{12}$$

the final values obtained are

$$E_1 = 6\alpha_0 + 54F_2 + (3649/675)(F_0 - 5F_2),$$
  

$$E_2 = 6\alpha_0 + 54F_2 + (3761/675)(F_0 - 5F_2),$$
(13)

and we see that here too  $E_3 > E_1$ .

#### 4. Numerical Calculations

To see what happens for intermediate ranges of the nuclear forces, we had to diagonalize numerically the matrices of order 10, after inserting for  $\alpha$  the critical value  $\alpha_0$ . We took a set of values of  $F_2/F_0$  between 1/35 and 0.198. For  $F_2/F_0 \leq 0.16$ , the matrices in LS scheme were used, and the diagonalization was made with the electrical network of the Physical Department of the University.<sup>9</sup> In the last five cases,  $F_2/F_0 \geq 0.17$ , the matrices in the jj scheme were used, and the eigen-

TABLE V. Values of  $E_3/E_1$  for  $0 < F_2/F_0 < 0.2$ .

$F_{2}/F_{0}$	$E_1/F_0$	$E_{3}/F_{0}$	$E_{3}/E_{1}$
1/35	7.2585	7.2959	1.0052
0.05	8.4645	8.5380	1.0087
0.08	10.412	10.520	1.0104
0.10	11.994	12.104	1.0092
0.11	12.920	13.025	1.0081
0.12	13.977	14.077	1.0072
0.14	16.741	16.819	1.0047
0.16	21.438	21.488	1.0023
0.17	25.738	25.764	1.00101
0.18	33.869	33.887	1.00053
0.19	57.433	57.443	1.00017
0.195	103.931	103.935	1.000038
0.198	242.915	242.917	1.0000082

values were found by the perturbation formulas up to and including the third order, which sufficed for our purposes, as the rapid convergence of the first three approximations had shown.

The results are given in Table V and are seen to behave quite regularly. It is seen that the theory succeeds in explaining the facts for all the ranges of the nuclear forces.

I thank Professor Racah both for suggesting the problem and for his continual help throughout the work, and Dr. Many for his help in operating the electrical network diagonalizing the numerical matrices.

<sup>&</sup>lt;sup>9</sup> A. Many, Rev. Sci. Instr. 21, 972 (1950); thesis, Jerusalem, 1950 (unpublished).