

mass as a function of θ , the angle in the center of mass:

$$(d\sigma/d\Omega)_{\text{cm}} = (3.20 \pm 0.78)(0.071 \pm 0.068 + \cos^2\theta) \times 10^{-29} \text{ cm}^2 \text{ sterad}^{-1}.$$

The total cross section for the mesons in the peak is therefore $(1.62 \pm 0.49) \times 10^{-28} \text{ cm}^2$. This suggests that the meson comes off almost entirely in a P -wave, and since the majority of the mesons of the entire spectrum are in the peak, it would follow that the total spectrum is approximately in a P -wave.

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The Radioactivity of Barium 140

J. M. CORK, J. M. LEBLANC, A. E. STODDARD, D. W. MARTIN,
C. E. BRANYAN, AND W. J. CHILDS

University of Michigan, Ann Arbor, Michigan*

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RADIOACTIVE barium of half-life about 13 days was first noted¹ as a product of the bombardment of uranium by neutrons, even before the phenomenon of nuclear fission was recognized. Subsequent studies² have shown the activity to be in Ba^{140} , in which the beta-decay to La^{140} is complex and accompanied

TABLE I. Summary of electron lines.

Electron energy (kev)	Relative intensity	Interpretation	Gamma-energy (kev)
23.3	20	$L_1^1 (Z=57)$	29.6
23.7	2	L_2^1	29.6
24.1	1	L_3^1	29.6
28.2	10	M^1	29.6
29.3	5	N^1	29.6
79.8	1	K^2	118.5
93.1	4	K^3	131.8
123.3	10	K^4	162.0
156.0	5	L^4	162.3
160.8	2	M^4	162.2
265.5	4	K^5	304.2
382.8	1	K^6	421.5
397.1	1	K^7	435.8
498.0	4	K^8	536.7
530.0	1	L^8	536.3

by gamma-emission. Three gamma-rays had been reported³ with energies of 0.16, 0.30, and 0.54 Mev.

A continued study of the fission product as supplied by the Oak Ridge National Laboratory, using photographic magnetic spectrometers, leads to a more accurate evaluation of energies and shows the existence of certain previously unreported gamma-rays. The barium radioactivity will usually be in equilibrium with the daughter, radioactive lanthanum. It appears, however, that in the original chemical separation of Ba^{140} , the La^{140} is carried down in excess of the equilibrium amount. This leads to a change in the relative intensity with time of the electron lines due to La^{140} as compared with the electron lines due to Ba^{140} and thus aids in their identification. The half-life curve of the specimen is complex, showing an initial 41-hour decay before settling into the longer

barium half-life, now found to be 13.4 days. The K - L - M differences of the electron energies also make it possible, in most cases, to distinguish those electron lines associated with each activity.

A summary of the electron energies (believed to be accurate to ± 0.2 percent) together with an arbitrary estimate of their relative intensities, and proposed gamma-origin is presented in Table I. A

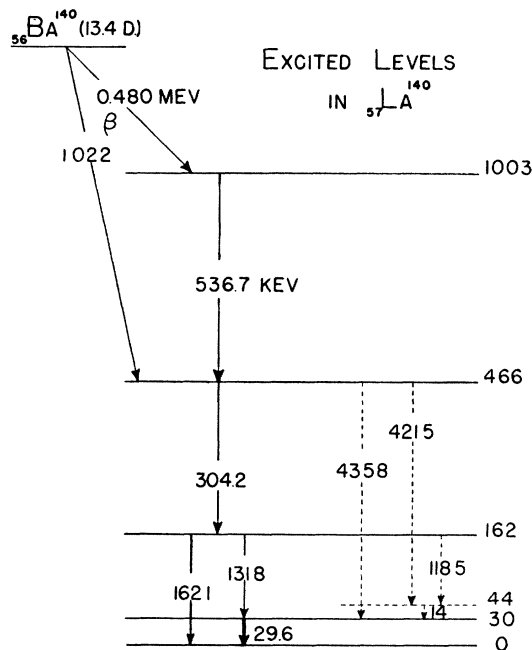


FIG. 1. Energy levels in La^{140} following beta-emission from Ba^{140} .

decay scheme had been proposed by Beach *et al.*, using their three observed gamma-energies, in which an unobserved gamma of 76 kev would have been required. It is now quite certain this gamma-ray does not exist. The observed gamma-energies do, however, fit very satisfactorily a modification and enlargement of the level scheme as shown in Fig. 1. The gamma-rays of greater intensity are represented as transitions with darker lines. The transitions shown as dotted lines are less certain, since only the " K " electron line was observed for each of these gamma-rays and there is some possibility that any or all of this activity is in the daughter product. In order to complete the scheme a gamma-ray of energy 14 kev would be required. This energy is slightly below the limit of the spectrometers.

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Gamma-Radiation from Lanthanum 140

J. M. CORK, A. E. STODDARD, J. M. LEBLANC, C. E. BRANYAN,
D. W. MARTIN, AND W. J. CHILDS

University of Michigan, Ann Arbor, Michigan*

(Received June 28, 1951)

A PREVIOUS study of the radioactivity from La^{140} (41.4 hr) showed¹ the presence of twelve low energy gamma-rays, with an indication of others at higher energy. A contemporary report noted² the beta-decay of La^{140} to be complex with energies of 1.32, 1.67, and 2.26 Mev; but only five gamma-rays, mainly at higher energy, were found. The present spectrometric investigation,