

FIG. 1. Temperature dependence of the magnetic field required at fixed
frequency for the Na²⁸ nuclear resonance in the metal. ΔH_r is the difference
in applied fields; the field required for resonance in the metal is theoretical curve representing the effect of thermal expansion on para-magnetic susceptibility.

Fig. 1. The sample, from National Distillers Chemical Corporation, was a 25 percent sodium dispersion in Nujol mineral oil; particle sizes ranged from 1 to 60μ , averaging 15μ . ΔH_r is the applied field, H_a , for resonance in aqueous NaCl, less that required for the metal. The resonance shift, $\sigma = 10^4 \times \Delta H_r / H_a$, is relatively insensitive to temperature, varying by 5 percent from 210' to 425'K. Moreover, as suggested by Dr. C. P. Slichter in one of our discussions, most of the temperature dependence can be ascribed to volume effects. The alkali metals show a small increase with temperature in paramagnetic susceptibility, which Stoner attributes' to thermal expansion, on the basis of a free electron model. By analogy, the resonance shift would be $\sigma_T=(V_T/V_0)^{\frac{3}{2}}\sigma_0$, where the subscripts T and 0 refer to the given and a reference temperature, respectively, and V is the atomic volume. The dashed line in Fig. 1 represents this function, the reference temperature being chosen to be 25°C. The value of σ_0 is taken to be 9.9, the experimental relative value of σ at 25°C. It is seen that the values computed at higher temperatures with the equation $\sigma_T = (V_T/V_0)^3 \times 9.9$ are lower than the observed σ by an amount that is proportional to the temperature effect on σ . This indicates that $\sigma=9.9$ measured with respect to the NaCl solution is not the absolute σ value. The experimental values should all be increased by 2, and the theoretical equation changed to $\sigma_T = (V_T/V_0)^3 \times 11.9$, to put everything on an absolute scale and get the two curves to agree. The hysteresis in the resonance shift below the melting point probably results from super-cooling. The low temperature increase starting at 250'K appears to be real; similar results have been obtained for Li'.

The temperature dependence of the Na²³ resonance line width in the metal is given in Fig. 2. The transition at 190'K is presumably

FIG. 2. The Na²² nuclear magnetic resonance line-width transition in the metal. The transition is consistent with an activation energy for self-
diffusion of 9.5 ±1.5 kcal/g-atom. The measure of line-width used here is

associated with self-diffusion. An activation energy for selfdiffusion of 9.5 ± 1.5 kcal/g-atom is estimated⁶ from the temperature and slope of the transition. The high temperature apparent line width of about 0.05 gauss is determined by field inhomogeneities and modulation effects. An independent analysis of the diffusion process is provided by the relaxation time measurements of Norberg and Slichter, which are presented and discussed in the accompanying letter. '

A comparison of Figs. 1 and 2 suggests that the low temperature increase in the resonance shift is perhaps associated with the diffusion process. For Na²³ the increase in resonance shift starts at $255\textdegree K$, a temperature about $65\textdegree$ above the line-width transition; in the case of Li' the increase starts at a temperature about 85' above the line-width transition at 255°K. The time average magnetic resonance field at the nuclei could change as the diffusion rate becomes less than the Larmor frequency. In fact, the relaxation time data⁷ and also the line-width observations for Na²³ indicate the diffusion frequency is about 1 Mc at 250'K.

Various room temperature observations have been made. The $Cu⁶³$ and $Cu⁶⁵$ resonance shifts in the metal with respect to $Cu₂Br₂(s)$ were observed to be¹ 23.46 \times 10⁻⁴, within experimental error of ± 0.5 percent. Also, a chemical shift⁴ of 0.56×10^{-4} was found between $Cu_2Cl_2(s)$ and $Cu_2Br_2(s)$ for Cu^{63} and Cu^{65} with an experimental error of ± 5 percent; Cu₂Br₂(s) required the higher applied magnetic field. The Al^{27} resonance in Al^{+3} aqueous solution and in anhydrous $AICl_3(s)$ occurred in applied fields 0.81 and 1.32×10^{-4} higher than in AlO₂⁻ solution. Such chemical effects are an additional problem in the theoretical analysis² of resonance shifts in metals. The 20 percent factor suggested above for Na²³ may reflect chemical effects; also, the temperature dependence itself complicates the issue.

Sodium, aluminum, and copper crystallize in cubic lattices for which electric quadrupolar broadening of the resonance line should be absent.⁸ However, detailed line shape plots for the Na²³ dispersion at 85° K and for Al²⁷, Cu⁶³, and Cu⁶⁵ in the powdered metals at room temperature gave experimental second moments of about 0.95, 11.7, $\overline{7}$.2, and $\overline{5}$.8 gauss² compared with theoretical second moments on the basis of magnetic dipolar broadening alone³ of 0.66, 7.5, 5.1, and 3.4 gauss², respectively. The high experimental values may result from anisotropy in the conduction electron shift in resonance frequency. Also, such anisotropy might generate electrical field gradients at the nuclei and lead to quadrupolar broadening. Experiments designed to determine the relative importance of the two factors include observation of metals with nuclear spin of $\frac{1}{2}$, the investigation of the dependence of second moments on apphed field, and single-crystal studies.

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1 W. D. Knight, Phys. Rev. 76, 1259 (1949).

1 Townes, Herring, and Knight, Phys. Rev. 77, 852 (1950).

³ Townes, Herring, and Knight, Phys. Rev. 77, 852 (1951).

⁴ H. S. Gutowsky McCl

Nuclear Relaxation Times in Metallic Na²³

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VUCLEAR magnetic relaxation times T_1 (spin-lattice) and T_2 ' (enin phase magnetic) T_2' (spin phase-memory) have been measured in metallic sodium as a function of temperature from -58° C to $+80^{\circ}$ C using the spin-echo method.¹ The apparatus was a modified version of

that used by one of the authors to investigate hydrogen-palladium systems.² The sample was a dispersion of sodium in mineral oil as described in the accompanying letter.³

Within experimental error, T_1 appears to be inversely proportional to the absolute temperature, in agreement with the theory of Heitler and Teller' regarding nuclear relaxation by interaction with the conduction electrons. From Korringa's theory5 and Knight's measurement⁶ of the frequency shift between Na²³ resonances in the metal and in a salt, one calculates that T_1 due to conduction electrons should equal 11.7 milliseconds at room temperature, in excellent agreement with the measured value of 9.2 milliseconds. A 20 percent increase in the shift, such as Gutowsky proposes in the accompanying letter,³ would lower the theoretical T_1 to a value even closer to the experimental result,

The T_2' data are shown in Fig. 1. Over the entire temperature range covered, T_2' is temperature dependent and is larger than the rigid lattice value by an order of magnitude. These results are in agreement with those of Gutowsky³ which indicate that the linewidth is narrower than the rigid lattice value for all temperatures above -100° C. The spin echo results show that T_2' increases markedly with temperature from the lowest temperature at which spin echoes were observed (-58°C) to about -25 °C, where T_2' becomes of the order of T_1 . For higher temperatures, T_2' goes as $1/T$ within experimental error and is of the order of T_1 , indicating that spin-lattice relaxation determines the line-width in the high temperature range.

Using a simple theory, the observed low temperature behavior of T_2 ' can be explained as the combined effect of spin-lattice line broadening and nuclear dipolar broadening. T_{2}' is assumed to be given by the equation

$$
1/T_2' = (1/T_2) + 1/T_2{}^a,\t\t(1)
$$

where T_2 is the phase-memory time determined by dipolar interactions alone, and T_2^a is that determined by spin-lattice interactions alone. The expression provides a better 6t to the data than does the addition of second moments.

The large values observed for T_2' indicate that there is a motion of the sodium nuclei. Taking the motion to be self-diffusion of the nuclei, one may calculate the line narrowing on the basis of the theory of Bloembergen, Purcell, and Pound.⁷ The calculation assumes the metal to be a liquid in which the atoms come no closer than a distance r , and predicts an exponential temperature dependence of T_2 according to the law

$$
\frac{1}{T_2} = \frac{4}{5} \frac{\gamma^4 \hbar^2 I (I+1) N e^{E/RT}}{r D_0},
$$
 (2)

where γ is the nuclear gyromagnetic ratio, I is the nuclear spin, N the number of atoms per unit volume, and the diffusion constant, D, has been written in the customary form $D=D_0e^{-E/RT}$. For numerical work r was taken to be the rigid-lattice nearest-neighbor distance 3.7A. The liquid theory should give the temperature dependence of T_2 rather well, but be less reliable in giving the actual magnitude of T_2 .

To compare Eqs. (1) and (2) with experiment, T_2^a was taken to be a constant (10.8 milliseconds), since T_2^a is a much more slowly varying function of temperature than T_2 . The work of Nachtrieb⁸ has been utilized in the comparison. He has measured the coefficient of self-diffusion for sodium by means of radioactive tracers and finds $D=0.320e^{-10650/RT}$ cm²/sec. Extrapolating his measurements to -56° C gives $D=0.74\times10^{-11}$ cm²/sec. The nuclear resonance data at this temperature, where the T_2^a correction is small, yield $D=1.3\times10^{-11}$ cm²/sec. The order of magnitude agreement is considered satisfactory, since the theory is somewhat crude.

To compare the temperature dependence of the nuclear resonance data with Nachtrieb's results, a theoretical T_2' curve was constructed. Equations (1) and (2) and the constant T_2^a were used to calculate a value of T_2 at -56° C from the data. This value of T_2 , together with T_2^a and Nachtrieb's value of the activation energy, determined the theoretical T_2' curve shown in Fig. 1. Curves corresponding to values of activation energy 20 percent

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- 8 ~ g -6 \$ ~THEORETIC+ Ty E [~]l0.65 ^x IO CAL ខ្លួ - ទ ដូ 4 -5 gee THEORETICAL T_e
E = 10.65 x 10 CAL **EXPERIMENTAL** I/T (RECIPROCAL ⁹K x 10³) ------
2 34 3.6 3.8 4.0 2.8 3.0 3.2 3.4 3.6 3.8 4.0 4.2 4.4 4.6 4.<mark>8</mark> I I I I I I I I I

FIG. 1. Temperature dependence of T_2' for metallic Na²³

higher and lower are also shown. The agreement of both the temperature dependence and the order of magnitude of D with Nachtrieb's results confirms the hypothesis of diffusion narrowing. Similar work is currently in progress on the Li' relaxation times.

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¹ E. L. Hahn, Phys. Rev. 80, 580 (1950).

² R. E. Norberg, thesis, University of Illinois (1951), unpublished.

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Nuclear Absorption of Negative Pi-Mesons at Diferent Energies

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LFORD G-5, 400-, and 600-micron nuclear emulsions were exposed to the 70- and 105-Mev external π^- meson beams of the Nevis cyclotron, and the analysis of the interaction with emulsion nuclei was continued.^{1,2} To extend the energy range, additional plates were exposed to the 70-Mev beam behind 8" of lithium plus $\frac{1}{2}$ " of carbon. The energy of the mesons crossing these plates was 30-50 Mev, as determined by plural scattering and grain density

FIG. 1. Backward elastic scattering $(E = 105 \text{ MeV}, \Theta = 173^{\circ})$ with recoil and slow electron at the vertex (observed by Mrs. N. Bernardini).

