be likely to be confused with the total transition. B. Hamermesh reports no structure in the high energy end of the capture γ -ray spectrum of mercury.⁹ This would indicate $J=0$ for the capture state, since in cases where the total transition is allowed, it is observed to be 10 to 20 percent of the total intensity. The present work, however, favors $J=1$.

¹ W. W. Havens and J. Rainwater, Phys. Rev. 70, 154 (1946).

² T. Brill and H. V. Lichtenberger, Phys. Rev. 72, 585 (1947).

² E. Fermi and L. Marshall, Phys. Rev. 71, 666 (1947).

⁴ Harris, Langsdorf, and Seidl,

On the Viscosity of Gaseous He'

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'HE effect of the similarity of particles on their gas-kinetic properties was studied' some time ago and in particular, attention was drawn to the influence exerted by the presence of nuclear spins. Since then, the discovery of He³ has offered an opportunity to apply these theoretical concepts (with proper modifications) to a case in which sizable effects may well be observed rather conveniently.

If we denote collision cross sections for particles with symmetric or antisymmetric wave functions of their relative motion by σ_S and σ_A , respectively, and the corresponding cross section for the collision of two otherwise equivalent but dissimilar particles by σ_D , then one can easily show that $\sigma_S + \sigma_A = 2\sigma_D$. In the case of He³, the cross section for the collision of two atoms is, obviously,

$\frac{1}{4} \sigma_S + \frac{3}{4} \sigma_A = \sigma_D - \frac{1}{2} (\sigma_S - \sigma_D).$

Since² $\sigma_S > \sigma_D$ we find that the nuclear spin of He³ raises the viscosity of He' with respect to He4 very appreciably (quite apart from the influence of the masses).

We shall present the quantitative details of this effect and some related phenomena in ^a following paper. '

¹ O. Halpern and E. Gwathmey, Phys. Rev. 52, 944 (1937).
² Mott and Massey, *Theory of Atomic Collisions* (Oxford University
Press, London, 1933), p. 231.
² Figure 1 on p. 951 of reference 1 was based on the assumpt

Further Evidence for a Two Quantum Transition in Molecular Spectroscopy*

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T has been pointed out¹ that an extra line group was observed \blacksquare in a study by the molecular beam electric resonance method of transitions between the electrical quadrupole hyperfine structure levels of Rb⁸⁵F. This extra line group occurred exactly at one-half the frequency of the normal transition $(J=1, F_1=\frac{3}{2})\rightarrow (J=1,$ $F_1=7/2$, in which **J** is the molecular rotational angular momentum and $\mathbf{F}_1 = \mathbf{I}_1 + \mathbf{J}$, \mathbf{I}_1 being the spin of the Rb⁸⁵ nucleus. It was suggested that this line group might arise as a two quantum transition in which two half-frequency quanta supply the energy for the transition. The theory of this two quantum transition was presented as a second-order time-dependent perturbation.

A simple extension of this theory indicates that the two quantum transition should be observed if two separate frequencies $(f_1$ and $f_2)$ are applied, provided $f_1+f_2=\Delta E/h$, in which ΔE is the energy difference associated with the normal transition. We arranged experimentally to apply two frequencies in the C-field region and did indeed observe the peak of the zeroth vibrational line to occur

TABLE L

Mc/sec	Mc/sec	$+r2$ Mc/sec
3.928	2.400	6.328
3.828	2.500	6.328
3.628	2.700	6.328
3.528	2.800	6.328

as predicted. The pairs of frequencies at which this transition occurred are indicated in Table I. The rf field intensity associated both with f_1 and f_2 was \sim 4 volts/cm. A weak static C-field of \sim 14.2 volts/cm was applied. The frequency for the normal transition is 6.328 Mc/sec and for the extra'line 3.160 Mc/sec.

It was observed further that, with $f_1=3.528$ Mc/sec and $f_2=2.800$ Mc/sec and the rf field intensities as above, the line intensity increased by a factor of two when the static field was reduced from 14.2 volts/cm to 2.0 volts/cm, and the line was present at the higher intensity when no static field was applied. This increase in line intensity is in qualitative agreement with the theory; no adequate quantitative theory of line intensities has been worked out. Also, the presence of the line at zero applied static field, which we were not able to observe previously, is strong evidence in support of the two quantum theory. No known change was made in the apparatus which would account for our present success in observing the line at zero static field; it is likely that whether or not the line is present at zero static field depends critically upon interfield conditions because of nonadiabatic transitions, and these conditions are not easily controllable.

It is still not understood why other half-frequency lines are not observed. Further search was made with higher rf field intensities (from 5 to 15 volts/cm) at static fields from 0 to 8 volts/cm for half-frequency lines in the spectrum of Rb⁸⁵F, Rb⁸⁷F, and K³⁹F, but with negative results.

~ This research has been supported in part by the ONR. [~] V. Hughes and L. Grabner, Phys. Rev. 79, 314 (1950); V. Hughes and L. Grabner, Phys. Rev. 79, 829 (1950).

Threshold for Photoneutron Reaction in U^{238}

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ECENT experimental neutron binding energy measurement in the region of lead give energy differences between the radioactive series. Additional binding energy measurements to bridge the radioactive series in the region of uranium would serve as valuable checks in closing energy cycles. The binding energy of a neutron in U²³⁸ is of particular interest, since some uncertainty exists in the energy difference between the $4n+1$ and $4n+2$ radioactive series.²

X-rays produced by the 22-Mev betatron at the University of Illinois were used to measure the threshold for the photoneutron reaction U²³⁸(γ , n)U²³⁷. The energy scale is determined relative to the threshold for the 10-minute activity from the Cu⁶³(γ , n)Cu⁶² reaction which is taken as 10.9 Mev. The experimental arrangements are similar to those described elsewhere.³

Five 150-mg samples of uranium oxide were bombarded at different energies in a probe inside the x-ray donut for periods of time ranging from four to thirty-four hours. The uranium was depleted in U²³⁴ and U²³⁵ to decrease neutron fission activity and to reduce the growth of the 25-hour UY activity after chemical purification. The bombarded uranium was chemically purified to constant specific activity. Fifty-mg samples of U_3O_8 were prepared for counting by repeatedly painting and igniting many small portions of an organic uranium solution. The U²²⁷ activity was measured with a thin-window helium-filled Geiger tube with

sufficient aluminum absorber to cut out the uranium alphaparticles. The U²³⁷ counting rate was determined by using the least squares method to extrapolate the linear growth curve of the uranium daughters to an accurately controlled zero time (time of chemical purification). Such extrapolations with similar unirradiated uranium gave a background of 2 c/min (above counter background). The counting rate of U^{237} for each sample varied from 6 to 100 c/min. This activity decayed with a 6.6-day half-life.

The excitation function could be approximated by a parabola; hence it was possible to estimate the threshold from a linear extrapolation of the square root of the U²³⁷ yield plotted as a function of x-ray energy. The results are shown in Fig. 1, where

FIG. 1. Square root of the U²⁸⁷ yield plotted as a function of x-ray energy.

the yield is expressed in c/min of U²³⁷ for 50 mg of U₃O_s given constant bombardment. Analysis of the data gives a threshold of 5.97 \pm 0.10 Mev for the U²³⁸(γ , n) U²³⁷ reaction.

British workers' obtained a neutron emission threshold of 5.85 ± 0.15 Mev for natural uranium by an indirect measurement of the neutrons produced in photodisintegration. As Parsons and Collie4 point out, their method of measurement does not yield a clearly defined threshold since the end product is not identified, and the source of the neutrons is uncertain. However, the close agreement between the two values leads one to believe that they were measuring the neutrons from the U²³⁸(γ , n)U²³⁷ reaction.

This work was supported in part by the joint program of the AEC and ONR.
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8 McElhinney, Hanson, Becker, Duffield, and Diven, Phys. Rev. 75, 542 (1949). ⁶ R. W. Parsons and C. H. Collie, Proc. Phys. Soc. (London} 63k, 839 (195O}.

Lithium Ammonium Tartrate Monohydrate, A New Ferroelectric Crystal*

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N an investigation of the electrical properties of a series of tartrates we have found that $LiNH₄C₄H₄O₆·H₂O$ is ferroelectric. (In a private communication to the author, Matthias has reported his parallel discovery of the ferroelectric behavior of this crystal.¹) It appears that isomorphous salts such as LiRb- and LiTl-tartrate \dot{H}_2O are also ferroelectric. The electrical behavior of the LiNH4 salt is reported in the present letter.

The crystals are orthorhombic, and the dielectric constants in

all three axial directions are small (between 8 and 10) at room temperature. Upon lowering the temperature, ϵ_a and ϵ_c remain essentially constant, whereas ϵ_b shows a sudden and sharp peak at about 106'K (Fig. 1). Below this temperature we find a spon-

FIG. 1. Dielectric constant ϵ_b versus temperature.

taneous electric polarization which establishes this transition as a Curie point (Θ) .

The hysteresis loops (Fig. 2) show the onset of a spontaneous

FIG. 2. Hysteresis loops versus temperature,

polarization P_s at the Curie point and its increase when the temperature is lowered. The saturated value at low temperatures is about $P_s = 0.21 \times 10^{-6}$ coulomb/cm².

The crystal is strongly piezoelectric, with a modulus d_{25} of about 20×10^{-8} cgs at room temperature. The piezoelectric resonance frequencies show the usual sharp minimum at the Curie point (Fig. 3).Using these measurements of the dielectric constant

FIG. 3. Resonance frequency and piezo constant d28 versus temperature.