

with liquid nitrogen were employed to detect the radiations. The beta-crystal was so thin (*ca* 20 mg/cm²) that its efficiency for gamma-rays was negligible. The coincidence circuit employed blocking oscillator pulse shaping and a 6BN6 mixing stage. Various lengths of terminated RG 65/U cable were used to introduce the delays.

A plot of the coincidence rate *versus* delay time is shown in Fig. 2. The dotted curve is that of Co⁶⁰ and is typical of substances

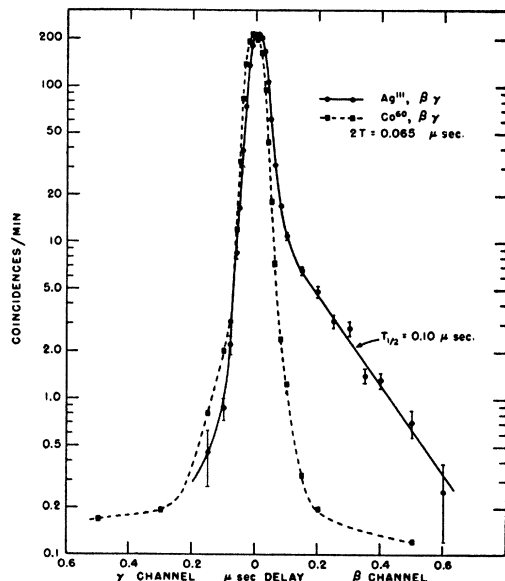


FIG. 2. Delayed coincidences in 7.5-day Ag¹¹¹.

showing no measurable gamma-delay. The solid curve is that observed with a purified source⁴ of Ag¹¹¹. The main body of the transitions shows no delay greater than 0.02 μsec. A small fraction of the transitions are delayed with a half-life of 0.10 ± 0.02 μsec. Since nearly equal counting efficiencies for the two gammas are expected, it may be calculated that the ratio of the undelayed to delayed transitions is about 6:1. This agrees well with the ratio of 8:1 for the 340-keV to 243-keV transitions found by Johansson.

The identification of the 243-keV level in Cd¹¹¹ formed by beta-decay of Ag¹¹¹ with the 0.09-μsec level observed in the K-capture decay of In¹¹¹ and in the 48-min isomeric transition of Cd¹¹¹* thus appears to be justified.

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⁴ Obtained through the courtesy of the Isotopes Division of the Oak Ridge National Laboratory.

The Thermal Neutron Capture Cross Section of Co⁵⁹

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A DISCREPANCY has existed in the values published for the thermal neutron capture cross section of Co⁵⁹. The activation method used by Seren, Friedlander, and Turkel¹ yielded a value of 21.7 ± 4.3 barns. The "pile oscillator" or "danger coefficient" method gave values of 34.3 ± 0.7 barns,² 34.2 ± 3.4 barns,³ and 38.2 ± 3.0 barns.⁴ This discrepancy suggested either an error in the pile activation determination or possibly a short-lived isomer of Co⁶⁰ that had not been detected.

We have determined the activation cross section of Co⁵⁹ for thermal neutrons, using the NRX heavy-water reactor. The

thermal neutron cross section of Au—93 barns—was used as a standard.³ The Au and Co were irradiated simultaneously in a position where the neutron distribution was known to be predominantly thermal. The contribution of epi-cadmium neutrons was determined by an irradiation under cadmium and small corrections were made for this (2.2 percent for Au and 0.2 percent for Co).

The Co metal (99.92 percent pure) was irradiated as a weighed disk 0.3 mm thick. A disk was cut out of a sheet of gold leaf of known weight and area, floated on water, and lifted onto an aluminum backing. The water was then evaporated, and the area of the disk, and hence its weight, was determined under a microscope.

After a measured irradiation period of about 20 hours the Co and Au samples were dissolved in *aqua regia* and made up to standard volumes. From these solutions known aliquots were withdrawn and activity measurements made several hours later. The absolute disintegration rates of these aliquots were measured by 4π proportional counting and by coincidence counting techniques. The relative values were confirmed by an end window counter.

Half-lives of 5.26 years⁵ and 2.69 days⁶ for Co⁶⁰ and Au¹⁹⁸, respectively, were used in making corrections for decay and for lack of saturation of bombardment.

The thermal neutron capture cross section for Co⁵⁹ leading to the formation of Co⁶⁰ was found to be 34.2 ± 1.4 barns. This result is the mean of 6 activations. A large proportion of the probable error is introduced by variations in the thickness of the gold leaf used. The cross section leading to the formation of the 10.7-minute isomer is 0.66 barns,¹ and thus only about 2 percent of the thermal neutron captures lead to the short-lived isomer. Of this, less than 10 percent⁷ decays independently to Ni⁶⁰, leaving a maximum of only 0.2 percent not included in the ultimate production of 5.26 year Co⁶⁰. The cross section as measured can then be taken equal to the total absorption cross section of Co⁵⁹, with an error of less than 0.07 barns.

It will be noted that the activation cross section is now in excellent agreement with the results obtained by the "pile oscillator" method.

When this work was completed Dr. B. W. Sargent kindly drew our attention to an unpublished report by Jones, *et al.*,⁸ which gives a value of 33 barns for Co⁵⁹ obtained by the activation method. Agreement with our value is satisfactory.

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⁷ Deutsch, Elliott, and Roberts, Phys. Rev. **68**, 193 (1945).

⁸ Jones, Clark, and Overman, MonC 398 (unpublished).

The Excitation Functions for B¹⁰(d, α)Be⁸

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THE excitation functions for the ground-state group and the first excited state group for B¹⁰(d, α)Be⁸ have been determined by bombarding thin targets (375 μg/cm²) of isotopically enriched boron (96 percent B¹⁰) and of normal boron with deuterons from the Bartol van de graaff generator, and observing the reaction α-particles with an argon-filled proportional counter, biased to count the α's at the end of their range, at 90° to the incident beam. The isotopic assignment of groups was first made by observing the α-particle groups from the target of 96 percent B¹⁰ (Fig. 1) and from a normal boron target (Fig. 2) bombarded with 1.20-MeV deuterons and noting the change in intensity with change in isotopic ratio. As seen in the curves, groups 1, 2, and 3 are due to the B¹⁰ reaction, and group 4 is due to B¹¹. This confirms