

TABLE I. Ratio of e^-/e^+ scattering by platinum.

| Kinetic energy (Mev) | e^-/e^+ exp (percent) | e^-/e^+ (unscreened theory) (percent) |
|----------------------|-------------------------|---|
| 0.7 | 3.15 ± 5 | 2.74 |
| 1.0 | 3.13 ± 4 | 2.90 |
| 1.3 | 3.60 ± 10 | 2.98 |

having suffered a mean scattering of 57.6° , were predominantly single scattered particles. The equipment was best adapted to studying the ratio of electron to positron scattering as a function of Z and the energy. Therefore comparison with theory was made by assuming that the theoretical ratio,⁶ e^-/e^+ , for polystyrene of 1.08 was correct and then using this datum to normalize the apparatus. For heavy nuclei it is necessary to take into account the screening of the nucleus by the atomic electron shells. Calculations for the scattering of electrons by heavy atoms have been made by Mohr,⁷ but we know of no similar calculations for positrons. For 1.0-Mev electrons on platinum one finds from Mohr's theory that the screened nucleus should scatter more strongly than the unscreened by a factor of 1.12. However, positrons will behave somewhat differently because of a tendency to remain farther from the nucleus. There may also be other differences between electrons and positrons in a screened field. Pending the working out of calculations for positrons in a screened field, we can only compare our ratio e^-/e^+ with theory using a Coulomb field. Table I shows that our results agree only approximately with unscreened calculations. Presumably the discrepancy is to be explained by neglect of screening. The experimental errors quoted are just the statistical errors caused by a limited number of counts; we assume that our equipment behaves the same for positrons and electrons.

It is a pleasure to acknowledge the help of the Oak Ridge National Laboratory in providing the Ce^{144} , and we are particularly indebted to Mr. Dean Cowie of the Department of Terrestrial Magnetism for kindly providing the Ga^{66} . We also appreciate the help of Dr. Fulbright and Dr. Goeckermann in preparing the sources. One of us (H.J.L.) is indebted to the DuPont Company and the Hercules Powder Company for fellowships early in the course of this work.

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The Reaction $C^{14}(n, \gamma)C^{15}$

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HUDSPETH, Swann, and Heydenburg have recently reported¹ the production of C^{15} by the reaction $C^{14}(d, p)C^{15}$. They bombarded both "normal" $BaCO_3$ and $BaCO_3$ containing about 40 percent C^{14} and obtained a β -emitter from the latter with an energy of 8.8 Mev and a half-life of 2.4 seconds. This they assigned to C^{15} . Using this data they have calculated the mass of C^{15} to be at least 15.01434 a.m.u. If this transition does not go to the ground state of N^{15} the mass will then be correspondingly greater. The reaction $C^{14}(n, \gamma)C^{15}$ should accordingly take place with an energy release greater than 2 Mev.

We have attempted to prepare C^{15} by the above reaction. If there is to be no doubt as to the proper assignment of the activity, the active and inactive $BaCO_3$ must be shown to be chemically identical, since the short half-life precludes any chemical separa-

tion after the irradiation. To ensure chemical identity both the active (6 atom percent C^{14}) and inactive Na_2CO_3 were reduced to elemental carbon using the same technique and the same reagents.² This reduced the impurities present and guaranteed, if not purity, at least the presence of the same impurities. Both active and inactive samples of elemental carbon were bombarded with neutrons in the rapid action "rabbit" of the Chalk River NRX pile. We have been unable to detect any activity which could be ascribed to C^{15} .

Lithium was bombarded under identical conditions and the resulting Li^8 observed. This is a β -emitter with a half-life of 0.88 second and a maximum energy of 12.7 Mev. Using the value of 33 millibarns³ for the reaction $Li^7(n, \gamma)Li^8$, we have set limits for the capture cross section of C^{14} . If the half-life of C^{15} is 2.4 seconds, then the cross section of C^{14} for thermal neutron capture must be less than 1 microbarn.

A full report of this work is being submitted to the Canadian Journal of Research.

¹ Hudspeth, Swann, and Heydenburg, Phys. Rev. 77, 738 (1950).

² We are indebted to Professor W. F. Libby for a private communication of his method of reducing CO_2 to C.

³ Hughes, Hall, Egger, and Goldfarb, Phys. Rev. 72, 646 (1947).

Low Energy Electrons at the End of μ -Meson Tracks

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ELECTRON sensitive nuclear track plates (Eastman NTB-3) were exposed to cosmic radiation in the stratosphere by means of meteorological balloons. A systematic search has been made for meson tracks which stopped in the emulsion. 158 mesons were found which fulfilled this requirement. Of these, 36 are σ -mesons, 14 are π - μ -decays, and 4 are π - μ - e -decays. The remaining 104 mesons show no associated tracks other than electron tracks. Low energy electrons (10 kev $< E < 60$ kev) are observed to originate from the ends of 17 of these tracks. Figure 1 shows an example of such an event. The number of low energy electrons from mesons is in rough agreement with the results of Occhialini¹ and of Franzinetti.² In three cases, two low energy electrons are observed to originate from the end of the same meson track. No low energy electrons and high energy decay electrons have been observed from the end of the same track. The spectral energy histogram of 20 low energy electrons, originating from mesons, is shown in Fig. 2.

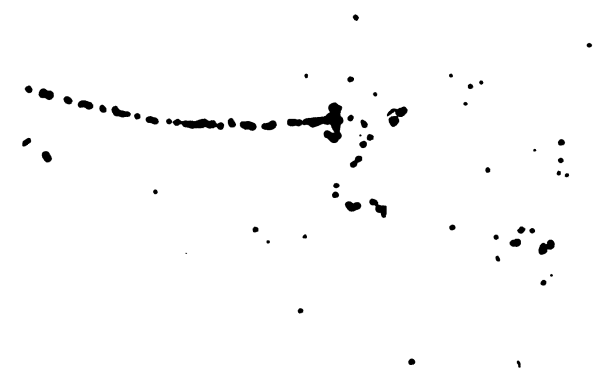


FIG. 1. Mosaic of a meson and two low energy electrons. The meson entered the emulsion from the left. The electron at the very end of the meson doubled back upon itself giving a false impression of the grain density. The electron to the right and below the meson has an energy of 45 kev.

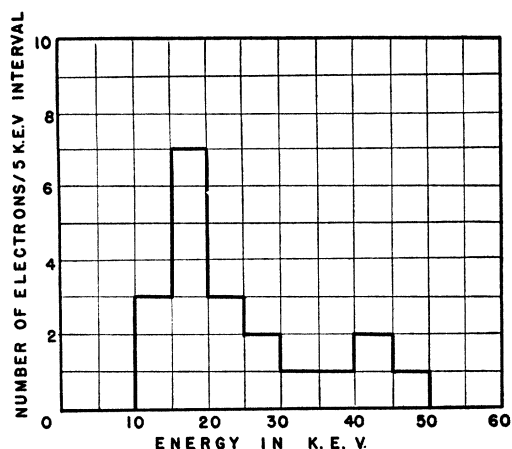


FIG. 2. Spectral energy histogram of 20 low energy electrons from mesons.

The density of random electron tracks in the emulsion has been measured. It is estimated that less than two percent of the mesons ending in the emulsion would appear to have associated electron tracks due to this background.

Assuming that the stopping power of the various constituents of the emulsion, for low energy mesons, is proportional to the atomic number, an estimate can be made of the relative number of mesons ending in gelatine and in silver bromide crystals. Assuming equal numbers of positive and negative μ -mesons,² 40 negative mesons are estimated to have stopped in silver bromide crystals, while 12 stopped in gelatine. If the observed electrons are due to the interaction of the meson with the orbital electrons, it seems improbable, from radiation considerations, that electrons of this energy originate from the lighter elements. The present data indicate that a negative meson, which has been captured in the field of a heavy atom, has a probability of 1/2 of having an associated electron in the 10 to 60 kev range; the probability that it have two such electrons is about 1/12.

¹ Cosynes, Dilworth, Occhialini, and Shoenberg, *Proc. Phil. Soc.* **62**, 801 (1949).

² C. Franzinetti, *Phil. Mag.* **41**, 86 (1950).

Restriction of Prismatic Punching to a Limited Class of Crystals

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SMAKULA and Klein¹ have observed that when a small single crystal of thallium bromiodide is indented, mounds appear on the other surfaces at points lying in $\langle 100 \rangle$ directions from the indentation. Seitz² has attributed this prismatic punching to the formation of rectangular loops of dislocation, a typical loop lying in two $\{110\}$ planes and two $\{1\bar{1}0\}$ planes, with the slip direction $[001]$. He has discussed the formation of such loops. Whether they arise from internal nuclei, as he suggests, or at the free surface where it meets the indenter, they can geometrically lead to the observed prismatic punching.

Why has this behavior so far been observed only in the thallium halides? It would be geometrically possible for prismatic punching to occur in aluminum, with a loop lying in two $\{111\}$ planes and two $\{1\bar{1}1\}$ planes, with the slip direction $[011]$, but it has not been observed. The simplest explanation seems to be that the slip vectors $\{100\}$ in thallium halides form an orthogonal set. It is not possible to resolve the $[001]$ slip vector into two other permissible slip vectors, and slip must always be parallel to the

axis of the prism. For aluminum the case is different. The slip vector $[011]a/\sqrt{2}$ in the plane (111) may dissociate into the two permissible slip vectors $[\bar{1}01]a/\sqrt{2}$ and $[1\bar{1}0]a/\sqrt{2}$ in the same slip plane. A similar dissociation is commonly observed in the bubble model. Unless slip occurs equally in these two directions the resultant will not be parallel to the axis of the prism. A complicated system of internal strains is then set up, and propagation of glide down the prism ceases. In a homogeneously stressed single crystal, the tendency to dissociation of the slip vector is not very noticeable, because the crystal is already slipping on that glide system for which the resolved shear stress is greatest. In a polycrystal, branching of the slip bands is commonly observed near the grain boundaries. The condition for prismatic punching is that there should be no linear relation connecting the permissible slip vectors.

I am indebted to Professor Seitz for allowing me to see his letter before publication, and to Professor Cottrell for discussion.

¹ A. Smakula and M. W. Klein, *J. Opt. Soc. Am.* **39**, 445, 890 (1949).

² F. Seitz, *Phys. Rev.* **79**, 723 (1950).

Luminescence of Color Centers in Alkali Halides

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WHEN alkali halides are x-rayed various color centers are formed. The one giving the strongest absorption is the *F*-center which is generally agreed to be an electron trapped at a vacant negative ion site. The next strongest absorption band is due to the *M*-center which Seitz¹ has suggested as being due to an electron trapped in a complex of two negative ion vacancies and a positive ion vacancy. Because these color centers are simple cases of impurities in solids there has been much investigation on their optical absorption and photo-conductive properties. The investigation reported here was to determine under what conditions these color centers luminesce and the effect of temperature on the luminescence emission. Gurney and Mott² have suggested

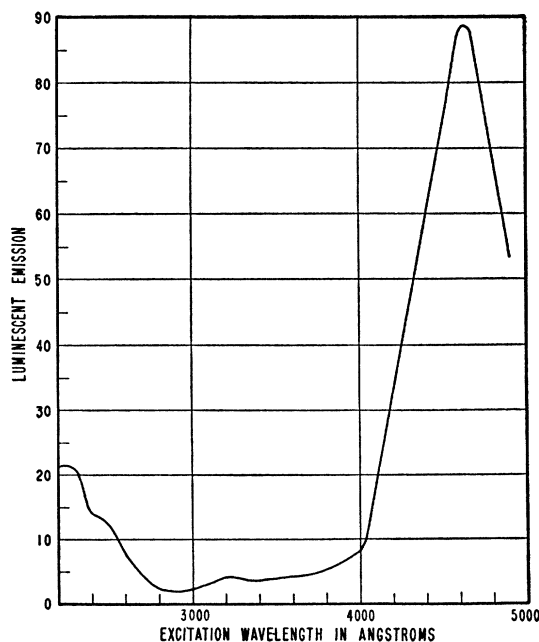


FIG. 1. Excitation spectrum of x-rayed lithium fluoride.