may be observable in polar semiconductors. Unfortunately ΔE_t has been computed¹ only at $T=0^{\circ}$, so that the form of the dependence of the effective gap width on temperature is not known.

The situation seems somewhat more hopeful in the case of optical excitation. The frequency of maximum absorption will be $\hbar\omega = E_G - 2\Delta E_0$. It will be noted that the effective energy shift ΔE_0 here differs from ΔE_t , for the energy shift ΔE_t includes a contribution from the alteration of the form of the phonon; but, by the Franck-Condon principle, this contribution is absent in an optical process. Thus ΔE_0 must be computed by finding the change in energy of an electron in the field of a phonon, assuming the phonon states to be unaffected by the presence of the electron. Using the methods employed in the treatment of electrical breakdown in polar crystals,² we have computed³ ΔE_0 as a function of temperature (ω_t is restrahl frequency, ϵ_0 and ϵ_s are optical and static dielectric constants):

$$(\Delta E_0)^3 = \frac{m\epsilon^4 \omega_i^2}{2} \left(\frac{\epsilon_s}{\epsilon_0}\right) \left(\frac{1}{\epsilon_0} - \frac{1}{\epsilon_s}\right)^2 \left(\exp\frac{\hbar\omega_i\epsilon_s^{\frac{1}{2}}}{kT\epsilon_0^{\frac{1}{2}}} + 1\right)^2 \left(\exp\frac{\hbar\omega_i\epsilon_s^{\frac{1}{2}}}{kT\epsilon_0^{\frac{1}{2}}} - 1\right)^{-2}.$$

It may be hoped that the polaron effect can thus be observed as a dependence of the frequency of maximum absorption ω on T, and it is, in fact, observed that ω decreases with increasing T in polar crystals, but is virtually constant in non-polar crystals. Unfortunately there is a second effect which, in polar crystals, decreases the gap width with increasing temperature. This effect, which has been treated theoretically by Radkowsky,⁴ arises from the fact that an electronic state subject to a collision frequency ν_c has a half-width $\frac{1}{2}h\nu_c$, resulting in a decrease in gap width of approximately $\hbar \nu_c$. Recomputing Radkowsky's results in accordance with the methods of reference 2 one obtains

$$2\Delta E = \hbar\nu_c = \frac{2^{\frac{1}{2}}m^{\frac{1}{2}}e^2\omega_t^{\frac{1}{2}}\epsilon_{\delta}^{\frac{1}{4}}}{\hbar^{\frac{1}{2}}\epsilon_0^{\frac{1}{4}}} \left(\frac{1}{\epsilon_0} - \frac{1}{\epsilon_s}\right) \left(\exp\frac{\hbar\omega_t\epsilon_s^{\frac{1}{2}}}{kT\epsilon_0^{\frac{1}{2}}} - 1\right)^{-1}.$$

These results may be compared with some experimental observations⁴ which very likely represent the rate of shift of ω with T:

$(d\hbar\omega/dT)$ in	KI	KI	RbBr	RbBr
10⁻₄ ev/°K	(296°)	(766°)	(360°)	(766°)
Observed	11.9	9.4	7.4	7.4
Polaron effect	4.4	3.3	5.0	4.0
Radkowsky effect	7.9	8.2	11.2	11.5

It would thus appear that the shift of optical absorption frequency with temperature allows, in principle, an observation of the polaron effect, but that experimental observations and theoretical treatments must be improved before truly quantitative conclusions can be drawn.

* This work was partially supported by the Bureau of Ships, U. S. Navy, ¹ Fröhlich, Pelzer and Zienau, Phil. Mag. **41**, 221 (1950). ² H. Callen, Phys. Rev. **76**, 1394 (1949). ³ We had computed ΔE_0 , ΔE_1 and the effective mass prior to the appear-ance of reference 1, using conventional perturbation theory and the methods of reference 2. Our results for ΔE_1 and effective mass agree with those of F.P.Z. except insofar as the results of those authors are improved by a novel variation of the perturbation technique. We have therefore recast the calculation of ΔE_0 so as to take advantage of this improvement in the nerturbation method. perturbation method. ⁴ A. Radkowsky, Phys. Rev. 73, 749 (1948).

Paramagnetic Resonance and Hyperfine Structure in the Iron Transition Group

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N the paramagnetic ions of the iron group, the orbital angular momentum is generally quenched by the crystalline field, which makes the h.f.s. much narrower than in the free ion and thereby considerably increases the perturbing influence on this h.f.s. of excited states, with unpaired s-electrons and wide h.f.s.

This fact, and the great precision of paramagnetic resonance methods, make it necessary to know the electronic wave function

with greater accuracy than is generally needed in the optical spectroscopy of the free atoms or ions. In fact the h.f.s. of the paramagnetic salts of divalent copper and manganese, calculated by assuming for Cu++ and Mn++ respectively the electronic configurations $3d^{9} {}^{2}D$ and $3d^{5} {}^{6}S$, suggested by the optical properties of these ions, disagree considerably with experiment.

The interaction between the electrons and the nuclear magnetic moment can be written $H = \mathbf{I} \cdot T \cdot \mathbf{S}$ where **I** is the nuclear spin $(\frac{3}{2}$ for Cu, 5/2 for Mn), S the total electronic spin ($\frac{1}{2}$ for Cu⁺⁺, 5/2 for Mn⁺⁺) and T is a tensor representing a generalized anisotopic h.f.s. interval. The principal values of the tensor T are (in 10^{-3} cm⁻¹):

Copper Tutton salts

Exper.¹ $|T_{11}| \simeq |T_{22}|$; 2< $|T_{11}| < 6$ according to the salt; $|T_{33}| \simeq 13$

Theoret.²
$$T_{11} \simeq T_{22} \simeq 13; T_{33} \simeq -6$$

Copper fluosilicate

Exper.³ $|T_{11}| \simeq |T_{22}| = 2.1; |T_{33}| = 2.8$ Theoret.⁴ $T_{11} = T_{22} = T_{33} = 7$

Manganese Tutton salts and fluosilicate

Exper.⁵ $|T_{11}| \simeq |T_{22}| \simeq |T_{33}| \simeq 9$

Theoret. $T_{11} = T_{22} = T_{33} = 0$.

These discrepancies can be explained by assuming that the electronic wave functions contain small admixtures of excited states with unpaired s-electrons, which are also respectively 2D and ⁶S states.

For Mn⁺⁺ no state ⁶S with unpaired 4s-electron can be constructed by exciting one or two 3d-electrons. For Cu++ the first excited state $3d^84s \ ^2D$ gives for T a positive isotropic correction ΔT which further increases the discrepancy.

As a working hypothesis we suggest the promotion of a 3s-electron to a 4s-orbital giving the excited configurations 3s3p63d94s 2D and 3s3p63d54s 6S. Energetically these configurations are far above the ground state and their admixture in this state is small, but this is compensated by the fact that 3s-electron has a very wide h.f.s. For Mn⁺⁺ the 3s-electron gives an isotropic h.f.s. of the right order of magnitude. For Cu⁺⁺ it is possible to make the correction ΔT negative. If we take $\Delta T = -9$ we find the new theoretical values:

Copper Tutton salts

$$T_{11}' \simeq T_{22}' \simeq 13 - 9 = 4; T_{33}' = -6 - 9 = -15$$

Copper fluosilicate

 $T_{11}' \simeq T_{22}' \simeq T_{33}' = 7 - 9 = -2$

which agree reasonably with experiment.

Details of these calculations will be reported elsewhere.

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Properties of Tl²⁰⁹

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PREVIOUS investigations at this laboratory¹ of the decay products of U²³³ showed that Bi²¹³ undergoes branching disintegrations, 98 percent decaying by beta-emission and two percent by alpha-emission. Canadian workers² who carried out parallel but independent studies also detected such alpha-branching. The product of alpha-disintegration of Bi213 would be Tl209 (neptunium C") which, by analogy with the C" products of the natural radioactive series, should be a beta-emitter with a half-life of only a few minutes. The amount of thallium activity formed by

decay of the milligram amounts of U²³³ used in the earlier investigations was too low to permit detection. However, gram quantities of U²³³ have since become available and with this larger amount of source material it has been possible to observe directly the Tl²⁰⁹ beta-particles.

The thallium was isolated by ether extraction from a solution of Ac²²⁵ obtained from a stock of U²³³ which had decayed for one year. The Ac²²⁵ was separated from the uranium by co-precipitation with lanthanum fluoride and was thoroughly purified by a procedure involving extraction of the Ac225 with a benzene solution of thenoyltrifluoracetone to separate from radium and thorium isotopes,³ followed by addition of lead carrier and precipitation of lead sulfide to remove any extraneous lead, bismuth, or polonium activities.

Because of the short half-life of the thallium a method of rapid separation was necessary. The following procedure proved satisfactory. Ten milligrams of inactive thallium carrier and a small amount of bromine were added to 2 ml of 6N HCl containing the Ac²²⁵. The solution was heated to oxidize the thallium to the extractable +3 state and the excess bromine boiled out. The solution was cooled, 2 ml of ether added and the mixture stirred for 30 sec. The ether phase was transferred to another vessel and stirred for several seconds with an equal volume of ether-saturated 6N HCl. After separation of the phases, the acid layer was drained through a stopcock and the washing repeated two or three times with fresh portions of acid. After the final wash the ether was transferred to a platinum dish and ignited. When the ether was consumed the platinum dish was rapidly cooled and placed immediately in the counter. The total time elapsed from start of the separation to start of counting was three to five minutes.

Decay of the thallium activity, which amounted to several thousand counts per minute initially, was followed at half-minute intervals. A small amount of long-lived activity, 20 to 30 counts/ min, above the natural background, which remained after decay of the thallium was probably due to incomplete chemical separation. A typical decay curve is shown in Fig. 1. The half-life as determined by least-squares analysis of seven such decay curves was 2.20 min., with a probable error of ± 0.07 min.

An aluminum absorption curve of the Tl²⁰⁹ beta-particles (Fig. 2) was obtained by isolating a series of samples and following the





FIG. 2. Absorption of Tl²⁰⁹ beta-particles in aluminum. Feather range, ~8. mg Al/cm² corresponding to E_{max} = 1.8. Half-thickness ~80 mg Al/cm². ~825

decay of each alternately with and without a given absorber. Each sample thus yielded one point on the absorption curve which was the ratio of the extrapolated counting rate at zero time with absorber to that without absorber. A Feather analysis4 of the absorption curve, after subtraction of the hard component, gave a value of about 825 mg of aluminum for the range, corresponding to a maximum energy of 1.8 Mev. The half-thickness was about 80 mg Al/cm².

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Resistivity of Semiconductors Containing Both Acceptors and Donors*

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HE resistivity of impurity semiconductors has previously been explained¹ by the addition of ρ_L , resistivity caused by scattering by lattice ions, and ρ_I , resistivity due to scattering by impurity ions. Since the mean free path associated with lattice scattering² is inversely proportional to temperature,

$$\rho = A T^{3/2} / n, \tag{1}$$

where A is a constant characteristic of the material and n is the density of conduction electrons or holes. By assuming that carriers are subject to Rutherford scattering by impurity ions, it has been shown³ that

$$\rho_I = 9 \times 10^{11} \pi^{3/2} e^2 m^{1/2} 2^{-7/2} \kappa^{-2} (kT)^{-3/2} (N_S/n) \\ \times \ln \{1 + 36 \kappa^2 d^2 (kT)^2 e^{-4}\}, \quad (2)$$

where κ = dielectric constant, e and m electron charge and mass, and d one-half the average distance between the N_S scattering ions per unit volume.