ments of the bomb. Evidence of other lines is present but confused by near coincidence with lines arising from the beryllium. For this reason, they are not included in Table I.

As a result of stimulating conversations with our colleague, W. H. Zachariasen, we propose the following simple model to explain this rather unusual transformation. Cerium, which is the first atom in the periodic table to permit the existence of a 4felectron in the free gaseous state, when condensed exhibits the tendency to become perverted from a trivalent to a quadrivalent state. For instance, CeO₂ is the stable oxide. Apparently, the application of 12,000 atmospheres of pressure is sufficient to evoke this transformation and the 4f electron is literally squeezed into a 5d state. No other model appears capable of explaining the enormous contraction from one closely packed structure to an identical closely packed structure. Quantitatively, this model appears to be supported by the lattice constants observed. Thus, according to W. H. Zachariasen,9 the ionic radii of trivalent and quadrivalent cerium with a coordination number of 12 should be 1.85 ± 0.01 A and 1.71 ± 0.02 A, respectively; the lattice constants observed correspond to ionic radii of 1.82A and $1.71\pm0.02A$ to which a small correction of about 1.5 percent should be applied for normal compressibility by comparison with the data on lanthanum and praseodymium. If we assume that the phase observed by Trombe and Foex⁵ is also a collapsed face-centered structure, the Clapyron equation permits us to make a rough estimate of the energy level separation between the 4f and 5d states. Such an estimate yields a value of approximately 0.04 ev. In addition, such a change in the electronic configuration of cerium would explain qualitatively the slight decrease in the change of the paramagnetic susceptibility with temperature.¹⁰ Quantitative agreement is not to be expected since the low temperature transformation probably does not go to completion.

The hypothesis proposed above also explains why the transformation, unique to cerium, is absent in lanthanum and praseodymium. In the former, the 5d state is known to be the more stable, while in the latter, the 4f state is probably sufficiently stable to resist perturbation by pressure in the experimentally available range.

Several related observations deserve mention herein. Below the transformation pressure, the cerium was often partially converted to an alternative form (tentatively identified as a close-packed hexagonal form) which occasionally persisted even at the highest pressures. In view of the lack of reproducibility in the formation of this phase, we believe it to be a metastable form created by the action of unavoidable but variable friction transmitted by the binding cement. Upon sudden release of the high pressure, this form (unmistakably identifiable as close-packed hexagonal) invariably was present and gradually transformed over a period of days to the original face-centered cubic modification. After several weeks the cerium oxidizes and only CeO2 lines are observed in the diffraction pattern.

In agreement with Bridgman,⁴ we observe that below the transformation pressure the compressibility of Ce progressively increases with increasing pressure, accounting for an appreciable fraction (over $\frac{1}{3}$) of the over-all volume change. We believe that this progressive increase in compressibility, terminating in a decisive collapse, is responsible for an increase in amplitude of thermal vibration in the lattice and accordingly explains the anomalous increase of resistance with pressure in Ce, previously reported by Bridgman.3

Attempts to produce the transformation in rather impure Ce (containing at least 0.2 percent Fe) existing in the hexagonal modification at atmospheric pressure were unsuccessful.

⁷ The authors are indebted to Professor Norman Nachtrieb for per-forming the chemical analyzes. ⁸ To be described elsewhere.

¹⁰ Private communication.
 ¹⁰ C. H. La Blanchetais, Comptes Rendus 220, 392 (1945).

The Frequency v_2 in the Infra-Red Spectrum of CH₄

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HE infra-red spectrum of methane has been investigated at length by Cooley1 and Nielsen and Nielsen,2 and the complex structure of the band ν_4 has been admirably analyzed by Jahn³ and Childs and Jahn⁴ in a series of papers. They accounted for the rotational structure of ν_4 on the basis that there exists a resonance Coriolis interaction between the frequency ν_4 and the optically inactive frequency, ν_2 . This analysis has been verified by Shaffer, Nielsen, and Thomas⁵ who have presented a theory of the rotation-vibration problem of the tetrahedrally symmetric XY₄ type of molecule complete to second order of approximation.

The Coriolis interaction affects the vibrations somewhat in the following manner. If v_4 is larger than v_2 , the effective moments of inertia in these states will be, respectively, larger and smaller than the moment of inertia in the normal vibrational state. This should give rise to convergence of the rotational lines in ν_4 toward higher frequencies, and in the frequency ν_2 , if it were active, toward lesser frequencies. If, on the other hand, ν_4 is smaller than ν_2 , a situation just the reverse of the above should prevail.

The Coriolis perturbation will cause a mixing of the wave functions of the two states so that, if the two frequencies are near enough alike, it should be possible to observe ν_2 . This is indeed the case in the molecules SiH_4 and GeH_4 where the two frequencies are separated by 65 cm⁻¹ and 114 cm⁻¹, respectively. Tindal, Straley, and Nielsen⁶ have reported on the activity of the band ν_2 , induced by this Coriolis interaction, for these two molecules. The frequency ν_2 has never been observed in the spectrum of CH₄ principally because it overlaps almost exactly the region of the strong water vapor band. An indication of its existence is nevertheless suggested by the data of Coblentz.7

This region has been explored with our vacuum infra-red grating spectrometer, and the band ν_2 has now been observed. The band center has been determined to lie at about 1533.6 cm⁻¹ while ν_4 lies at 1305.9 cm⁻¹. The Q branches of these two bands do fold away from each other as is to be expected from the theory outlined above, and the ν_2 band is considerably weaker in intensity.

Complete details of these measurements will be published at a later date.

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Note on Measurements of Ambipolar Diffusion in Helium*

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THE discussion presented in the last paragraph of the article "Measurements of ambipolar diffusion in helium" 1 is in error. The mobility of He⁺⁺ should be $\sqrt{2}$ times the mobility of He⁺ if charge transfer is neglected, and thus would not explain the difference between our measurements and those of Tyndall and Powell. Meyerott² has suggested that the theoretical mobility of

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