parities as explained below. The arguments are based on two assumptions: (1) Gamow-Teller selection rules govern the beta-decay process, and (2) the nuclear shell model¹ is a reliable guide in the region $Z \sim 40$ and $N \sim 50$.

The nucleus 40Zr90 contains closed shells of 40 protons and 50 neutrons. Thus, I=0 and the parity is even in the ground state.

39 Y90 is an odd-odd nucleus. It is characterized by a $(3p)^{-1}(4d)^1$ configuration in the notation of Feenberg and Hammack. Thus, the ground state parity is odd.

38Sr⁹⁰ is an even-even nucleus. It is characterized by a $(3p)^{-2}(4d)^2$ configuration. In the ground state the parity is even and, most probably, I=0.

Both beta-transitions are associated with change of parity. Thus by G-T rules, both are first-forbidden or both are thirdforbidden. The ft product for the first transition is $\sim 10^9$ and for the second $\sim 10^8$. These values are appropriate for firstforbidden transitions with $\Delta I = \pm 2^{(1-2)}$, but are too small for third-forbidden. Consequently, I = 2 in the ground state of Y⁹⁰.

Thus the two beta-transitions of Fig. 1 should produce peculiar energy distributions similar to those found in Y⁹¹, Cs137, K42, and Rb86 by various investigators.2-7 In such cases the energy distribution differs⁸ from the allowed form by a factor $G \sim (W_0 - W)^2 + (W^2 - 1)$.

A carrier-free sample of Sr⁹⁰ was obtained from Oak Ridge. This sample had been aged to permit an associated 55-day activity of Sr⁸⁹ to die out. Three spectra were measured: (1) Sr^{90} and Y^{90} in equilibrium together, (2) Sr^{90} alone, and (3) Y^{90} alone. For the latter two runs, the isotopes were chemically separated by a method due to Kurbatov and Kurbatov.9 Samples were prepared for the spectrometer by evaporation from solution on thin Zapon films, following the insulin technique of Langer.¹⁰

The spectra were measured in the double-focusing spectrometer described by Kurie, Osoba, and Slack.11 In the interests of higher counting rates, a wide counter window (0.25 in.) was used, which dropped the resolving power to 1 percent. The window was Zapon film of 3- to 4-kev stopping power, and was supported lengthwise by a single 5-mil wire. A pressure regulator¹² was used to avoid loss of counter pressure caused by window leakage.

The results are shown as FK (Fermi-Kurie) plots in Fig. 2 for Sr⁹⁰ and in Fig. 3 for Y⁹⁰. As indicated in Fig. 2, the Sr⁹⁰ data was obtained by subtracting the Y⁹⁰ distribution curve from that for an equilibrium mixture of both Sr⁹⁰ and Y⁹⁰.



FIG. 2. Sr^{90} FK plots. "Allowed" plot (upper) and "forbidden" plot (lower), with ordinates as indicated.



FIG. 3. V⁹⁰ FK plots. "Allowed" plot (upper) and "forbidden" plot (lower), with ordinates as indicated.

The upper curve in each figure (labeled $[N/p^2 F]^{\frac{1}{2}}$) is a conventional or "allowed" FK plot, i.e., computed as though the transition were allowed. The lower curve in each figure (labeled $[N/Gp^2F]^{\frac{1}{2}}$) is a "forbidden" FK plot, i.e., computed as though the transition were first-forbidden, with $\Delta I = \pm 2$ (yes). The near-straightness of the latter pair of plots confirms the assignment of spins and parities shown in Fig. 1.13

- * Assisted by the joint program of the ONR and the AEC.
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 * E. Feenberg and K. C. Hammack, Phys. Rev. 75, 1964 (1949).
 * F. B. Shull and E. Feenberg, Phys. Rev. 75, 1768 (1949).
 * L. M. Langer and H. C. Price, Jr., Phys. Rev. 75, 1109 (1949).
 * A. C. G. Mitchell and C. L. Peacock, Phys. Rev. 75, 1272 (1949).
 * J. S. Osoba, Phys. Rev. (in press).
 * A. C. G. Mitchell (private communication); see also Zaffarano, Kern, and Mitchell, Phys. Rev. 74, 682 (1948), especially Fig. 4.
 * K. Siegbahn, Arkiv. f. Mat., Astr. o. Fys. 34B, No. 4 (1946).
 * E. J. Konopinski and G. E. Uhlenbeck, Phys. Rev. 60, 308 (1941).
 * J. D. Kurbatov and M. N. Kurbatov, J. Phys. Chem. 46, 441 (1942).
 * L. Langer, Rev. Sci. Inst. 20, 216 (1948).
 * Ter-Pogossian, Townsend, and Robinson (to be published in Rev. Sci. Inst.).

Sci. Inst.). ¹³ Similar results and conclusions are reported by L. J. Laslett and E. Jensen and by L. M. Langer (private communications).

The Forbidden Beta-Decay of Sr^{89*}

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HE isotope Sr⁸⁹ decays by β^- emission to Y⁸⁹ with a half-life of 4.75.106 seconds. The energy release, including rest mass, is 3.93 mc², so that the maximum electron kinetic energy is 1.50 Mev. No gamma-radiation is observed. The ft product for the beta-transition is $4.9 \cdot 10^8$.

Using the nuclear shell model notation of Feenberg and Hammack,¹ the Sr⁸⁹ ground state is characterized by a $(3p)^{-2}(4d)^1$ configuration, so its parity is even. The ground state of Y^{89} is characterized by a $(3p)^{-1}$ configuration, so its parity is odd. The beta-transition between ground states, therefore, involves a parity change, and the transition is first-forbidden by Gamow-Teller rules. The large ft value favors a spin change of 2 units.1,2

A beta-transition for which $\Delta I = \pm 2$ (yes) yields an electron energy distribution which differs from that for an allowed transition by a factor $G \sim (W_0 - W)^2 + (W^2 - 1)$, if it is assumed that Gamow-Teller rules apply to the beta-process.³ Such distributions have been observed previously for Y91, Cs137, Rb⁸⁶, Sr⁹⁰, Y⁹⁰, and K⁴².^{2, 4-10} If the above reasoning is correct, the Sr⁸⁹ spectrum should display a similar forbidden "shape."

The spectrum was measured in the magnetic double-focusing spectrometer.¹¹ A source of Sr⁸⁹, containing also some Sr⁹⁰



FIG. 1. Sr⁸⁹ FK plots. "Allowed" plot (upper) and "Forbidden" plot (lower), with ordinates as indicated.

and Y90, was obtained from Oak Ridge. Contributions to the spectrum from Sr⁹⁰ and Y⁹⁰ have been subtracted from the data. The data are shown in Fig. 1 in the form of FK (Fermi-Kurie) plots. The ordinate for the upper curve ("allowed" plot) is $(N/p^2F)^{\frac{1}{2}}$ and for the lower ("forbidden" plot) is $(N/Gp^2F)^{\frac{1}{2}}$. No attempt has been made to use a closer approximate form of G because of the relatively high value of $W_{0.6}$ The upper curve shows the characteristic upward bulge for energies higher than $W_0/2$. The lower curve is satisfactorily straight, and confirms the prediction made above.¹²

Goldsmith and Inglis¹³ list the spin of Y^{89} as 1/2(?), (original source of this value is not mentioned). If true, the spin of Sr⁸⁹ ground state is 5/2. These spin values are in harmony with the nuclear shell model.

* Assisted by the joint program of the ONR and the AEC.
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* E. Feenberg and K. C. Hammack, Phys. Rev. 75, 1964 (1949).
* F. B. Shull and E. Feenberg, Phys. Rev. 75, 1768 (1949).
* E. J. Konopinski and G. E. Uhlenbeck, Phys. Rev. 60, 308 (1941).
* L. M. Langer and H. C. Price, Jr., Phys. Rev. 75, 1109 (1949).
* A. C. G. Mitchell and C. L. Peacock, Phys. Rev. 75, 1272 (1949).
* J. S. Osoba, Phys. Rev. (in press).
* Braden, Slack, and Shull (preceding letter).
* Zaffarano, Kern, and Mitchell, Phys. Rev. 74, 682 (1948), and private communication from Dr. Mitchell.
* K. Siegbahn, Arkiv. f. Math., Astr. o. Fysik 34B, No. 4 (1946).
* Drivate communications from L. J. Laslett and E. Jensen.
* Kurie, Osoba, and Slack, Rev. Sci. Inst. 19, 771 (1948).
* Similar results have been observed by L. J. Lazlett and L. M. Langer (private communication).
* H. H. Goldsmith and D. R. Inglis, The Properties of Atomic Nuclei I., Brookhaven National Lab. (1948).

Measurements Concerning the Vapor-Liquid Equilibrium of Solutions of He³ in He⁴ below 2.19°K

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SMALL vessel having a volume of 0.33 cm3 had condensed in it a known amount of helium in which the He³ concentration was about 5×10^{-4} . The vessel was placed in a bath of normal liquid helium and the vapor pressure difference, Δp , between the vessel and bath was measured while the vapor and liquid in the vessel were effectively stirred. In the top of the vessel we inserted a ground copper plug just below the capillary tube which connects the vessel with the apparatus outside the cryostat. The plug gives the helium film sufficient means for creeping out of the vessel, leaving the He³ below the plug.1 The film helium could be pumped out of the capillary by means of a Toeppler pump, and mass spectrometric analysis of this gas confirmed that no measurable amount of He³ was going out. Thus the pumping effected a very considerable increase in the He³ concentrations inside of the vessel.

If one assumes (a) perfect solution, i.e. independence of the energy of one atom of He³ or He⁴ of the concentration. (b) liquid helium to be a two fluid system with He³ soluble in the normal fluid only, and (c) helium to act as a perfect gas in the vapor phase, then a determination of Δp as a function of the known He³ content at constant temperature (below the λ point) enables one to test the relations which follow.

According to Henry

and

and

$$p_3 = N_3{}^L \pi_3 / (N_3{}^L + N_4{}^{nL}) \tag{1}$$

$$p_4 = N_4^{nL} \pi_4 / (N_3^L + N_4^{nL}) \tag{2}$$

in which p_3 and p_4 are the partial pressures of the two isotopes in the vapor, N_3^L and N_4^{nL} the number of molecules of He³ and normal fluid He⁴ in the liquid, π_3 and π_4 the saturated vapor pressures. We derive from (1) and (2):

$$p_3/p_4 = N_3{}^L\pi_3/N_4{}^{nL}\pi_4 \tag{3}$$

$$\Delta p = p_3 + p_4 - \pi_4 = N_3^L (\pi_3 - \pi_4) / (N_3^L + N_4^{nL}). \tag{4}$$

Accepting provisionally Tisza's relation $N_4^{nL}/N_4^L = \alpha$ $=S/S_{\Lambda}$ (S is the entropy of the pure He⁴ at the temperature used and S_{Λ} its entropy at the λ -point), we were able to calculate from the measured value of Δp at each temperature the absolute and relative concentrations in liquid and vapor, i.e.: $C_L = N_3^L / N_4^L$, $C_V = p_3 / p_4$, $X_L = N_3^L / (N_3^L + N_4^L)$ and $X_V = p_3(p_3 + p_4).$

TABLE I. $T = 1.75^{\circ}$ K.

∆¢ mm Hg	$K_{L}10^{4}$	$X_{V}10^{3}$	$N_3{}^L \cdot 10^8$	$N_3^V \cdot 10^8$	$N_3^{ ext{tot}} \cdot 10^8$
0.16	5	16	435	22	457
0.33	10	35	284	88	372
0.54	17	58	221	172	393
0.67	21	70	189	219	408
0.78	24	80	168	258	426
0.84	26	85	130	280	410
0.91	28	91	84	306	390
1.11	35	112	35	390	425

The so calculated values of X_L and X_V together with the known volumes of vapor and liquid in the vessel made it possible to compute the total number of molecules He³ present in the vessel. We could finally compare this number with our original amount of He³, viz.: 440×10^{-8} mole.

Table I shows in column 6 the number of N_3^{tot} at 1.75°K, which is, considering the accuracy of measurement, quite satisfactory. Similar isotherms were investigated at 2.0 and 1.9°K and analogous results were obtained. Lowering the temperature at constant filling gave again a good confirmation of our assumption. At 1.2°K we reached concentrations in the vapor as high as about 70 percent for a liquid concentration of 2 percent.

On the whole our chief assumption that He³ is soluble in the normal fluid fraction of helium II appears to be remarkably well realized. Details will be published in Physica.

¹ Daunt, Probst, Johnston, Aldrich, and Nier, Phys. Rev. 72, 502 (194.7)

A Search for Crystals That Exhibit Conduction Pulses Under Alpha-Particle Bombardment

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FTER conduction pulses were observed in diamond¹ A FIER conduction pulses were observed in under polonium alpha-particle bombardment, a search was made for other crystals that exhibit this phenomenon. The choice of crystal species that were selected for test was