be approximated by two straight lines, the light elements lying on a steeper line through the origin, while the heavy elements lie on a less steep line with a positive intercept. Increasing k_1 has the opposite effect. A variation in k_1 of $\pm 0.2 \times 10^{12}$ cm⁻¹, or in V of ± 2 Mev, begins to produce appreciable bending. A reduction in K, with fixed k_1 , introduces a curvature in the radius line, the center being pulled down and the two ends raised. The curvature becomes noticeable if K is reduced to less than $K = 1.9 \times 10^{12}$ cm⁻¹, however K can be almost doubled before the opposite curvature becomes very pronounced. For example, $K = 3.0 \times 10^{12}$ cm⁻¹ gives an about equally good straight line, $R = 1.39 \text{A}^{\frac{1}{2}}$ $\times 10^{-13}$ cm. The total cross-section measurements thus determine the potential fairly well, but are quite insensitive to the absorption coefficient. Measurements of σ_a and of the differential diffraction scattering are required for a better evaluation of K. It should be noted that while k_1 and K are determined directly from the cross sections, the evaluation of V depends also on the energy of the incident neutrons. Cook et al. state that the energy of the neutrons detected in their experiment may be a little lower than 90 Mev, lying somewhere between 80 and 90 Mev. If we took E = 80 Mev, we would find V = 28.8 Mev.

For $K = 2.2 \times 10^{12}$ cm⁻¹, the values of KR range from 0.58 for Li to 1.87 for U. It will be seen from Fig. 1 that the nuclear opacity, $\sigma_a/\pi R^2$, would vary from 0.52 for Li to 0.88 for U. It will also be seen that over this range of values of KR it would be expected that σ_d will be nearly twice as large as σ_a .

If one plots the angular distribution of the diffraction scattering given by (9) (i.e., $d\sigma_d(\theta)/d\sigma_d(0)$ versus $kR \sin\theta$) one finds curves for the heaviest nuclei which are indistinguishable from that for an opaque nucleus (Eq. (4)), at least as far as the first minimum of the diffraction pattern. For the lighter nuclei, the form of the curve is closely the same, but with an altered scale of abscissa, corresponding to using an effective radius somewhat smaller than the true radius. The increase in the half width of the diffraction peak is zero for KR=1.78 (Pb), 3.7 percent for KR=1.20 (Cu), 6.2 percent for KR=0.90 (Al) and 9.6 percent for KR=0.63 (Be). Measurements of the diffraction scattering and of the absorption are now in progress in this laboratory.

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He³ Isotopic Abundance*

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The isotopic abundance of He³ in one sample each of "well" helium and "atmospheric" helium has been measured by detecting the He³(n,p)H³ disintegrations induced by thermal neutrons. The helium gas was put into a proportional counter, the disintegration rate compared to that with nitrogen in the counter, and the He³ content deduced from the known ratio of the He³ and N disintegration cross sections.

INTRODUCTION

PREVIOUS measurements^{1, 2} have indicated that He³ is present in natural helium in amounts of the order of one part in 10^6 to 10^7 , and that the abundance varies by more than a factor of 10 depending on the source of the helium. He³ concentration determination is of special current interest in connection with nuclear investigations of interactions between elementary nuclei, and in connection with investigations of the thermodynamic behavior of He³ and He⁴ at temperatures of liquid helium.

In the present work measurements were made of the isotopic abundance of He³ in two samples of natural helium: one from wells near Amarillo, Texas, and one from air reduction processing. The presence of He³ was detected by counting ionization pulses arising from the disintegration products of the reaction $\operatorname{He}^{3}(n,p)\operatorname{H}^{3}$ induced by thermal neutrons. The cross section for this reaction is about 5000 barns,³ which is sufficiently large to make it possible to detect the He³ in natural helium samples. Data were also taken with nitrogen in the counter, in which case one detects the $N^{14}(n,p)C^{14}$ disintegrations. Since the protons from the He³ and N reactions have very closely the same range, the "wall effect" corrections will be similar for the two cases. If counting is done on nitrogen and on

³ J. H. Coon and R. A. Nobles, Phys. Rev. 75, 1358 (1949).

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¹L. T. Aldrich and A. O. Nier, Phys. Rev. **70**, 983 (1946); **74**, 1225 (1948); **74**, 1590 (1948). ²L. W. Alvarez and R. Cornog, Phys. Rev. **56**, 613 (1939);

^a L. W. Alvarez and R. Cornog, Phys. Rev. 56, 613 (1939); 56, 379 (1939).

	Partial press. p of gas in atmos.	Press. of added argon in atmos.	Counter voltage	Approx. total number dis- intergrations counted	Additive correction Σw	Corrected counts per min. =D
N ₂ -run 1 run 2 "Well" helium	0.258 0.258 7.47	1.71 1.71 0.27 0.65	3725 3725 2010	12,000 12,000 7000	137.0 111.0 0.32 0.54	$\begin{array}{c} 1314\\ 1243 \end{array} Av. = 1278\\ 9.15\\ 10.55 \end{array}$
Pure argon	1.143	0.05	2670	9000	0.54	10.55

TABLE I. Determination of disintegration rates.

helium with the same neutron flux, then the disintegration rates D taking place within the sensitive volume of the counter will be related by:

$$D_{\rm N}/D_{\rm He} = (\sigma_{\rm N} n_{\rm N}/\sigma_{\rm He} n_{\rm He}), \qquad (1)$$

where σ is the disintegration cross section and *n* the number of atoms per cm³. If *I* is the relative concentration of He³ in the helium and *p* is pressure, then the above ratio may be written

$$D_{\rm N}/D_{\rm He} = (\sigma_{\rm N}/\sigma_{\rm He^3}) \cdot (2p_{\rm N}/Ip_{\rm He}).$$
(2)

Using the ratio σ_N/σ_{He^3} elsewhere reported,³ we can



FIG. 1. Differential pulse height distribution curve obtained with a thin deposit of normal uranium mounted on the inner wall of the counter. The two peaks correspond to the alphaparticle groups from U²³⁴ and U²³⁸. 3.0 atmos. argon pressure.



FIG. 2. Pulse height distribution curves obtained by counting N¹⁴(n,p)C¹⁴ disintegrations. The plotted points represent the number of pulses in a 5-volt pulse height interval and the point is plotted at the center of the interval. The vertical lines indicate standard deviations computed by taking the square root of the total number of counts obtained and adjusting this to the counting rate.

TABLE II. He³/He⁴ Ratio $\times 10^{-7}$.

	Present measurement	Aldrich and Nier
"Well" helium (Amarillo, Texas)	1.73	1.4
"Atmospheric" helium (Airco)	13.0	12.0
Beryl crystals (Amarillo, Texas)		0.6, 1.4, 1.4, 2.0
Radioactive ores		<0.3

therefore obtain I by determining the counting rates at measured pressures.

APPARATUS AND PROCEDURES

The techniques and apparatus used were essentially the same as those reported in the article by Coon and Nobles in this issue of the Physical Review. There were a few exceptions as follows. A normal uranium alpha-particle emitter was used inside the counter instead of plutonium. A calcium purifier was attached directly to the counter in the manner reported by Klema and Barschall.⁴ This purifier was desirable since the effect of a small N₂ impurity in the helium would not be distinguishable from the effect of He³. The counter gas flowed continually over calcium turnings in the purifier at about 300°C. The effectiveness in removing nitrogen was tested by contaminating a pure argon filled counter with a small amount of air, and observing the drop in neutron induced counting rate as the calcium removed the nitrogen.

RESULTS

The pulse height distribution curves are shown in Figs. 1 to 4. The curves show counting data taken both with and without a cadmium cover over the counter. The count with cadmium is subtracted as a background except in the small pulse region where gamma-rays are detected, and where neutron capture gammas from the cadmium interfere. From Fig. 2 it is seen that in the case of nitrogen the count with cadmium is a very small fraction of the count without cadmium (about one part in 800)

⁴E. D. Klema and H. H. Barschall, Phys. Rev. **63**, 18 (1943).

except in the region of small pulses. In the case of helium (see Figs. 3 and 4) the count without cadmium is relatively high because of the presence of a few residual high energy neutrons getting to the region of the counter in spite of the large amount of graphite between the counter and the fast neutron source. The He⁴ recoils from these fast neutrons are appreciable in number because of the tremendous abundance of He⁴ as compared to He³. Figure 3 also shows the cadmium difference curve obtained with 1.95-atmosphere pure argon in the counter. This count with argon serves as a background to be subtracted from the cadmium difference curves gotten with helium.

The number of disintegrations D taking place within the sensitive volume of the counter was obtained from

$$D = \sum C_1 - \sum C_2 - \sum A + \sum w,$$

where $\sum C_1$ is the counting rate without cadmium summed over all channels counted; $\sum C_2$ is the counting rate with cadmium summed over all channels counted; A is the background as determined with pure argon, this also being a cadmium difference count; and $\sum w$ is the "wall effect" correction determined by adding up the number of pulses under the extrapolated lower end of the corrected pulse height distribution curves. The correction $\sum w$ accounts at least in part for all small pulses whether they are due to "wall effect" or to other causes which reduce their size so much that they cannot be counted in the lowest pulse height counting channel.

Inserting values for p and D from Table I into relation (2), using³ $\sigma_{\text{He}^3}/\sigma_{\text{N}} = 2860$, and solving for I gives values listed in Table II, along with values obtained by Aldrich and Nier.¹ Differences between the two sets of values are probably not significant, especially in view of the variations which Aldrich and Nier find in the He³ abundance for samples of different origin.

For correlating the He³ abundance with the source of helium, the mass spectrograph is better than the present counter technique because of the large quantity of gas necessary in the counter method. For analysis of the He³ content in enriched samples the counter technique may offer some advantages in ease and accuracy, depending of course on available facilities.

The accuracy of the present measurement is estimated at ± 15 percent, which does not include the error in the value of $\sigma_{H_e^3}/\sigma_N$ which is about ± 5 percent. Of the 15 percent error, approximately 6



FIG. 3. Pulse height distribution curves obtained with "well" helium. The plotted points represent the number of pulses in a 10-volt pulse height interval and the point is plotted at the center of the interval. The difference curve, C_1-C_2-A , corresponds to the number of He³(n,p)H³ disintegrations.



FIG. 4. Pulse height distribution curves obtained with "atmospheric" helium. The plotted points represent the number of pulses in a 10-volt pulse height interval and the point is plotted at the center of the interval. The difference curve, $C_1 - C_2 - A$, corresponds to the number of He³(n,p)H³ disintegrations.

percent is statistical error determined by considering the number of counts obtained. The correction $\sum w$ which involves extrapolation of the curve to zero pulse height may introduce 5 percent error. Background fluctuations from spurious electrical disturbances assume more importance because of the low counting rates. There is an undetermined effect of capture gammas from neutron capture in cadmium; these gammas cause a larger number of small pulses when cadmium surrounds the counter than when it does not.

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