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Band Structure and Fermi Surface of ReO₃

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The nonrelativistic augmented-plane-wave (APW) method has been applied to calculate the band structure of ReO₃. An important feature of this calculation is that it includes corrections to the usual "muffintin" approximation. Because of convergence difficulties, the APW calculation has been carried out only at symmetry points in the Brillouin zone. These results have been used in conjunction with the Slater-Koster linear-combination-of-atomic-orbitals interpolation scheme to determine the band structure and Fermi surface throughout the Brillouin zone. The Fermi energy occurs in the t_{2g} manifold of the rhenium 5d bands. The calculation predicts a Fermi surface consisting of two closed electron sheets centered about Γ , plus a third electron sheet which is open along (100). The tight-binding parameters, which affect both the oxygenrhenium (2p-5d) energy separation and the corresponding bandwidths, have been adjusted to fit the optical and Fermi-surface data for ReO_3 . Spin-orbit effects for the rhenium 5d bands have been included by means of a spin-orbit parameter ξ_{5d} . The optimum value for this parameter has been determined by detailed comparisons between the theoretical and experimental Fermi-surface areas. The final results agree to within 10%.

I. INTRODUCTION

`HE transition-metal oxides exhibit a rich variety of electrical and magnetic properties which are interesting from a fundamental as well as a practical point of view. These properties include the semiconductor-to-metal transitions in the lower oxides of titanium and vanadium¹ as well as the various types of magnetic ordering which are found to exist in other examples of these materials.2,3

Compared to these more exotic transition-metal oxides, the electrical and magnetic properties of rhenium trioxide (ReO₃) can only be described as "ordinary." ReO₃ is nonmagnetic and exhibits metallic-type conductivity down to liquid-helium temperatures. The importance of ReO3 arises from the growing wealth of experimental data relating to the electronic energy bands. The optical measurements of Feinleib et al.⁴ provide band-structure information over a 22-eV energy range. Recently, Marcus⁵ has observed de Haas-van Alphen oscillations in single-crystal samples of this compound. These results provide important and detailed information concerning the ReO₃ band structure at the Fermi energy. More recently, magnetothermal oscillations have been observed in ReO₃ samples by Graebner.⁶ His data are in basic agreement with Marcus's de Haas-van Alphen results.

The purpose of the present investigation is to determine a theoretical model for the ReO₃ band structure which is consistent with both the Fermi surface and the optical data. The calculations leading to this model have been carried out in three stages. The first stage involves a nonrelativistic augmented-plane-wave (APW) calculation for ReO₃ at symmetry points in the Brillouin zone. This calculation assumes an ad hoc potential which is derived from *neutral* atom charge densities. An important feature of this calculation is that it includes corrections to usual muffin-tin approximation to the crystal potential.7

In the second stage of the calculation, the Slater-Koster LCAO (linear-combination-of-atomic-orbitals) interpolation scheme has been fitted to the APW results at symmetry points to determine the energy bands throughout the Brillouin zone.8 This fitting procedure has included states arising from the rhenium 5d and

P. D. DeCicco, Phys. Rev. 153, 931 (1967).
 J. C. Slater and G. F. Koster, Phys. Rev. 94, 1498 (1954).

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¹ F. J. Morin, Phys. Rev. Letters **3**, 34 (1959). ² J. B. Goodenough, *Magnetism and the Chemical Bond* (Wiley-Interscience, Inc., New York, 1963). ⁸ Transition Metal Compounds, edited by E. Schatz (Gordon and Breach Science Publishers, Inc., New York, 1964). ⁴ J. Feinleib, W. J. Scouler, and A. Ferretti, Phys. Rev. **165**, 765 (1968).

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⁵ S. M. Marcus, Phys. Letters 27A, 584 (1968).

⁶ J. E. Graebner (unpublished).

oxygen 2p bands. The bands arising from the rhenium 6s-6p and the oxygen 2s atomic levels have been omitted from this interpolation calculation. The effects of spinorbit coupling among the rhenium 5d states have been taken into account by means of a spin-orbit parameter ξ_{5d} . The results of these calculations are in good qualitative agreement with the Fermi-surface and optical data.

In the final stage of the calculation, the LCAO or tight-binding parameters which affect the energy separation between the (lower) oxygen 2p and (higher) rhenium 5d energy band states as well as the corresponding bandwidths have been adjusted in accordance with the optical⁴ and cyclotron mass⁵ data. The Fermisurface areas obtained from this adjusted band structure agree with the experimental results to within 10%.

The general arrangement of this paper is as follows. The details of the calculation are described in Sec. II. This discussion includes a general description of the ReO_3 structure, the calculation of the one-electron potential, the corrections to the muffin-tin potential, and the details of the tight-binding calculation. The results of the calculation are presented in Sec. III, including energy-band, density-of-states, and Fermisurface information. These results are compared with experiment in Sec. IV. The last section contains a general discussion of the present model for the ReO3 band structure and its relationship to previous models that have been proposed for ReO3 and structurally related materials.

II. DETAILS OF THE CALCULATION

A. ReO₃ Structure

An important aspect of ReO₃ is determined by its similarity to the perovskites ABC_3 . The simple cubic ReO₃ unit cell is shown in Fig. 1(a). The more heavily shaded Re atom (ion) is located at the origin, while the more lightly shaded O atoms (ions) occupy positions at the face centers. In the perovskites, the Re and O atoms are replaced by the B and C atoms, respectively, while the A atoms are situated at the corners of the cube.



Another related class of materials is represented by the tungsten bronzes, including compounds such as Na_xWO₃.

According to Fig. 1(a), the ReO₃ unit cell contains a single ReO_3 complex. The bravais lattice is simple cubic; the appropriate Brillouin zone is shown to the right in Fig. 1(b), where the standard notation is used to designate various symmetry points (Γ, X, M, R) and lines $(\Delta, \Sigma, \Lambda, S, T, Z)$.⁹ The space group for the ReO₃ structure is O_h^1 , which is a symmorphic space group. The irreducible representations for this space group are available in the literature.⁹ The rhenium atom is situated at a site with full cubic (O_h) point symmetry. The oxygen atoms reside at sites with tetragonal (D_{4h}) point symmetry.

In sorting out the energy-band results, it is useful to know which spherical harmonics centered about the rhenium and oxygen sites contribute to energy band states of a given symmetry. This information is readily determined by considering the transformation properties of Bloch-type functions formed from s, p, and datomic-type orbitals centered about the rhenium and oxygen sites, respectively. Using familiar grouptheoretical techniques, the appropriate symmetry of the energy-band states which arise from these atomic levels can be determined. This information is summarized in Table I for the ReO₃ structure.

ReO₃ contains a single rhenium atom per unit cell. This atom produces a single s band, three p bands, and five d bands, as expected. This is consistent with the results shown in Table I when the degeneracies of the various irreducible representations are taken into account. On the other hand, the three oxygen atoms in the ReO_3 unit cell produce three *s* bands, nine *p* bands, and a possible 15 *d*-band states. Because of oxygen's position in the periodic table, the oxygen d states represent highly excited levels, both in the atom as well as in the solid.

B. APW Calculation

The ReO₃ structure is not ideally suited to a straightforward application of the APW method, which involves the muffin-tin potential approximation. This is due to the fact that ReO_3 is a rather open or looselypacked structure when compared to the fcc or hcp structures, for example.¹⁰ However, DeCicco has suggested a means for correcting these deficiencies in the method.⁷ He suggests decomposing the crystal potential $V(\mathbf{r})$ into a muffin-tin part $V_m(\mathbf{r})$ plus corrections $V_{\Delta}(\mathbf{r})$, so that

$$V(\mathbf{r}) = V_m(\mathbf{r}) + V_\Delta(\mathbf{r}). \tag{1}$$

The muffin-tin potential $V_m(\mathbf{r})$ is one which is spherically symmetric inside the APW sphere centered about

^{50, 58 (1936).} ¹⁰ L. F. Mattheiss, J. H. Wood, and A. C. Switendick, Methods Comput. Phys. 8, 64 (1968).



FIG. 1. (a) Unit cell for the ReO₃ structure. (b) Brillouin zone for the simple cubic bravais lattice.

⁹L. P. Bouckaert, R. Smoluchowski, and E. Wigner, Phys. Rev.

	Г	X	M	R
		Rhen	ium states	
S	Γ_1	X_1	M_1	R_1
Þ	Γ_{15}	$X_{4'} + X_{5'}$	$M_{4'} + M_{5'}$	R_{15}
d	$\Gamma_{12}+\Gamma_{25'}$	$X_1 + X_2 + X_3 + X_5$	$M_1 + M_2 + M_3 + M_5$	$R_{12} + R_{25'}$
		Oxv	gen states	
S	$\Gamma_1 + \Gamma_{12}$	$X_1 + X_2 + X_{4'}$	$M_1 + M_{5'}$	R_{15}
Þ	$2\Gamma_{15} + \Gamma_{25}$	X1+X5+X3'+X4'+2X5'	$M_1 + M_2 + M_3 + M_4$	$R_1 + R_{12} + R_{25'} + R_{15'}$
			$+M_5+M_{4'}+M_{5'}$	
d	$\Gamma_1 + \Gamma_2 + 2\Gamma_{12}$	$2X_1 + 2X_2 + X_3 + X_4 + 2X_5$	$M_1 + M_2 + M_3 + M_5 + M_{1'} + M_{2'}$	$R_{2'} + R_{12'} + 2R_{15} + 2R_{25}$
	$+2\Gamma_{25'}+\Gamma_{15'}$	$+X_{1'}+X_{3'}+X_{4'}+X_{5'}$	$+M_{3'}+M_{4'}+3M_{5'}$	

TABLE I. Spherical harmonic content of ReO₃ energy-band states at the symmetry points Γ , X, M, and R in the Brillouin zone. Entries include states arising from s, p, and d atomic-type orbitals centered about the rhenium and oxygen sites, respectively.

each atomic site and constant throughout the remainder of the unit cell. Although the sphere radii are arbitrary, the convergence of the method is improved by making the spheres as large as possible (i.e., touching).¹⁰

If $V_{\Delta}(\mathbf{r})$ is zero within each sphere, these corrections can be incorporated *exactly* into the APW method by adding a single additional term to the usual APW matrix elements, namely, the Fourier coefficient of $V_{\Delta}(\mathbf{r})$:

$$V_{\Delta}(\mathbf{K}_{j}-\mathbf{K}_{i}) = \Omega^{-1} \int_{\text{cell}} \exp[-i(\mathbf{k}+\mathbf{K}_{j})\cdot\mathbf{r}] V_{\Delta}(\mathbf{r})$$
$$\times \exp[i(\mathbf{k}+\mathbf{K}_{i})\cdot\mathbf{r}] d^{3}r, \quad (2)$$

where Ω is the unit cell volume, **k** is a wave vector in the first Brillouin zone, and **K**_l equals a vector of the reciprocal lattice multiplied by 2π . In general, $V_{\Delta}(\mathbf{r})$ is not zero within the APW spheres; DeCicco has suggested two approximate methods for estimating the effects of such corrections under these circumstances.

The sphere radii used in the present APW calculation for ReO₃ are listed in Table II, along with the appropriate lattice parameter.¹¹ These sphere radii were chosen to produce touching oxygen and rhenium spheres along $\langle 100 \rangle$ and to optimize the convergence of both the oxygen 2p and rhenium 5d bands.¹⁰ Despite this precaution, the APW calculation requires approximately 250 APW functions in order to yield energy-band results which converge to within about 0.001 Ry. This fact precludes the possibility of carrying out APW calculations at general points in the Brillouin zone and even inhibits accurate calculations along symmetry lines. As a result, we have confined the APW calculations to the symmetry points Γ , X, M, and R, where group-theoretical techniques permit one to factor the resulting secular equations and reduce them to more manageable size.

The crystal potential $V(\mathbf{r})$ was approximated by an *ad hoc* potential derived from Hartree-Fock-Slater (HFS) atomic charge densities.¹² The Coulomb poten-

tial $V_c(\mathbf{r}_i)$ at an arbitrary point \mathbf{r}_i in the unit cell was calculated by summing directly the contributions from neighboring Re and O atoms. The total charge density $\rho(\mathbf{r}_i)$ was calculated in this same manner. The total potential $V(\mathbf{r}_i)$ contained the Coulomb term $V_c(\mathbf{r}_i)$ plus the Slater exchange contribution, $-6[3\rho(\mathbf{r}_i)/8\pi]^{1/3}$, or

$$V(\mathbf{r}_{i}) = V_{c}(\mathbf{r}_{i}) - 6[3\rho(\mathbf{r}_{i})/8\pi]^{1/3}.$$
 (3)

The spherically symmetric portions of the muffin-tin potential were determined by calculating $V(\mathbf{r}_i)$ along three independent directions originating from the rhenium and oxygen sites, respectively, and extending to the appropriate sphere radii. A weighted average of these results determined the spherically symmetric portions of $V_m(\mathbf{r})$. To determine the constant portion of $V_m(\mathbf{r})$ outside the spheres, V_{mo} , the potential $V(\mathbf{r}_i)$ was averaged over a uniformly distributed three-dimensional grid of points \mathbf{r}_i equivalent to 17 576 points in the entire unit cell.

The differences between $V(\mathbf{r}_i)$ and V_{mo} were used to calculate the Fourier coefficients $V_{\Delta}(\mathbf{K}_l)$. This was done by direct summation,

$$V_{\Delta}(\mathbf{K}_l) = N^{-1} \sum_{n=1}^{N} e^{i\mathbf{K}_l \cdot \mathbf{r}_n} [V(\mathbf{r}_n) - V_{mo}], \qquad (4)$$

where N is the total number of points in the unit cell at which $V(\mathbf{r}_n)$ was calculated. Those Fourier coefficients for which $\mathbf{K}_l = (2\pi/a)(l_1, l_2, l_3)$ and $0 \le (l_1, l_2, l_3) \le 6$ were calculated and included in the APW calculation. The Fourier series converged rather slowly; typical values of $V_{\Delta}(\mathbf{K}_l)$ ranged from -0.1276 Ry for $\mathbf{K}_1 = (2\pi/a)$ (1,0,0) to -0.0006 Ry for $\mathbf{K}_{34} = (2\pi/a)(6,6,6)$.

In the present APW calculations for ReO₃, the corrections to the muffin-tin potential inside the APW spheres have been neglected. The effect of these cor-

TABLE II. Summary of the structure-related parameters involved in the APW calculation for ReO₃.

¹¹ W. B. Pearson, in *A Handbook of Lattice Spacings and Structures of Metals and Alloys* (Pergamon Press, Inc., New York, 1958). ¹² F. Herman and S. Skillman, *Atomic Structure Calculations* (Prentice-Hall, Inc., Englewood Cliffs, N. J., 1963).

rections on the band structure is expected to be small. The corresponding corrections outside the APW spheres are quite large in ReO₃, ranging from -2.05 Ry in the region where the Re and O spheres contact to +0.68 Ry at $\frac{1}{2}(a,a,a)$. This peak-to-peak variation is roughly six times larger than that found by DeCicco in his KCl calculation.⁷

C. Tight-Binding Calculation

The convergence difficulties resulting from the application of the APW method to transition-metal compounds such as ReO₃ limit the usefulness of this method for determining the details of the band structure and Fermi surface throughout the Brillouin zone. As a result, we have applied the LCAO or tight-binding interpolation scheme of Slater and Koster for this purpose.⁸ This scheme is readily adapted to include the effects of spin-orbit coupling, as we shall show presently.

The LCAO scheme has been applied by Kahn and Leyendecker to determine the band structure for the structurally related compound $SrTiO_{3}$.¹³ In the present application of this method, the APW results at Γ , X, M, and R have been used to determine the various transfer integrals which enter the scheme as disposable parameters. In the Kahn-Leyendecker calculation, these parameters were estimated from a variety of sources, including experimental data, model calculations, and symmetry arguments. More recently, a complete LCAO tight-binding formalism for the ReO₃ structure has been worked out by Honig *et al.*¹⁴ This formalism has been developed independently by the present author (though in less complete form) during the course of the present investigation.

The detailed form of the LCAO tight-binding interpolation scheme which has been applied to the present ReO₃ band-structure calculations is contained in Table III. This application of the method involves a total of 14 basis functions. These include five rhenium 5dorbitals and nine oxygen 2p orbitals, as specified in Table III A. The nonvanishing d-d, p-p, and p-dHamiltonian matrix elements are contained in Table III B-III D, respectively.

The calculation includes nearest-neighbor rheniumrhenium (d-d) interactions, nearest-neighbor and second nearest-neighbor oxygen-oxygen (p-p) interactions, and nearest-neighbor oxygen-rhenium (p-d) interactions. According to the results of Table III, these interactions are represented by a total of 19 parameters; six parameters (D_1-D_6) describe the d-d interactions, 11 (A_1-C_4) represent the p-p interactions, and two (P_1,P_2) include p-d hybridization effects in the model. The 11 p-pparameters can be described more precisely. The parameters A_1-A_3 enter the diagonal matrix elements of the oxygen $2p\sigma$ orbitals, while B_1-B_4 enter the di-

TABLE III. Tabulation	of nonzero	tight-binding	Hamiltonian
matrix elements, as used	in the ReO	3 interpolation	calculation,
where $\xi = k_x a$, $\eta = k_y a$, and	$\zeta = k_z a$.	-	

А. Т	Tight-bindi	ng basis functions	
	No.	Origin	Function
Rhenium d	1	(0,0,0)	xy
	2	(0,0,0)	yz
	3	(0,0,0)	ZX
	4	(0,0,0)	$3z^2 - r^2$
	5	(0,0,0)	$x^2 - y^2$
Oxygen p	6	$\frac{1}{2}a(1,0,0)$	x_1
	1	$\frac{1}{2}a(1,0,0)$	\mathcal{Y}_1
	8	$\frac{1}{2}a(1,0,0)$	z_1
	9 10	$\frac{1}{2}a(0,1,0)$	x_2
	10	$\frac{1}{2}a(0,1,0)$	
	12	$\frac{1}{2}a(0,1,0)$	42 Xo
	13	$\frac{1}{2}a(0,0,1)$	A 3 No
	14	$\frac{1}{2}a(0,0,1)$	y 3
B Rher	ium-rhenii	$\frac{2a(0,0,1)}{1m(d,d)}$ interaction	~3
$D_1 \equiv E_{m_1 m_2}(0)$	$(10^{11}-110^{11})$	$D_4 \equiv E_{2a2} = 2_{2a2}$	² (000).
$D_2 \equiv E_{xy, xy} (10)$)0) .	$D_{4} = E_{3z^{2}-r^{2}}, s_{z}^{3z^{2}}$ $D_{5} \equiv E_{3z^{2}-r^{2}}, s_{z}^{2}$	$r^{2}(001)$.
$D_3 \equiv E_{xy,xy} (00)$)1).	$D_6 = E_{x^2 - y^2} x^2 - x^2$	$u^2(001)$.
$H_{1,1} = D_1 + 2D_0$	rose+rosm	$+2D_{2}\cos c$	
$H_{1,1} = D_1 + 2D_2(0)$ $H_{2,2} = D_1 + 2D_2(0)$	$\cos_2 + \cos_2$	$+2D_2 \cos \xi$	
$H_{2,2} = D_1 + 2D_2($ $H_{2,2} = D_1 + 2D_2($	$\cos(+\cos)$	$(+2D) \cos^2 $	
$H_{4,4} = D_4 + 2D_5(2)$	$\frac{1}{2}\cos^{2}(1+\cos^{2}$	$(c_{0}) + c_{0} = c_{0}$	os (+ cos n)
$H_{5,5} = D_4 + \frac{3}{2}D_5(0)$	$\cos \varepsilon + \cos \eta$	$+2D_{6}(\frac{1}{4}\cos\xi+\frac{1}{4}\cos\xi)$	$\cos(1 + \cos(7))$ $\cos(1 + \cos(7))$
$H_{4,5} = \frac{1}{2}\sqrt{3}(D_5 - 1)$	$D_{\rm f}$) ($-\cos\xi$	$+\cos n$	00.7 1 00037
C. Oxy	vgen-oxvge	n $(p-p)$ interaction	S
$A_1 \equiv E_{m_1,m_2}(000).$	$B_1 \equiv F_2$	$(1000) C_1 =$	$F_{\rm max}(\frac{1}{2},\frac{1}{2},0)$
$A_{1} = E_{x_{1}, x_{1}}(000),$ $A_{2} = E_{x_{1}, x_{1}}(100)$	$D_1 = D_1$ $B_0 \equiv E_1$	$(100), C_1 = C_2 = C_2$	$E_{x_1, x_2(2, 2, 0)},$
$A_{3} \equiv E_{x_{1},x_{1}}(010),$	$B_2 \equiv E_1$ $B_3 \equiv E_2$	$(1, y_1(100)), C_2 \equiv (0, y_1, y_1(100)), C_3 \equiv (0, y_1, y_1(100)), C_3 \equiv (0, y_1, y_1(100)), C_4 \equiv (0, y_1(100))), C_4 \equiv (0, y_1(100)), C_4 \equiv (0, y_1(100))), C_4 \equiv (0, y_1(100)), C_4 \equiv (0, y_1(100))), C_4 \equiv (0, y_1(100))), C_4 \equiv (0, y_1(100))), C_4 \equiv (0, y_1(100))), C_4$	$E_{x_1, y_2}(\frac{1}{2}, 0),$
	$B_4 \equiv E_1$	$C_4 \equiv C_4 =$	$E_{z_1, z_2}(\frac{1}{2}, \frac{1}{2}, 0),$
$H_{6,6} = A$	$1+2A_{2}\cos^{2}$	$\xi + 2A_{2}(\cos n + \cos c)$	·)
$H_{7,7} = B$	$1 + 2B_2 \cos^2 \theta$	$\xi + 2B_3 \cos (2B_4 + 2B_4)$	COSC
$H_{8,8} = B$	$_{1}+2B_{2}\cos \theta$	$\xi + 2B_3 \cos(\xi + 2B_4)$	cosn
$H_{9,9} = B$	$_1+2B_2\cos \theta$	$q+2B_3\cos\xi+2B_4$	cosζ
$H_{10,10} = A$	$_{1}+2A_{2}\cos \theta$	$\eta + 2A_3(\cos\zeta + \cos\xi)$)
$H_{11,11} = B$	$_1+2B_2\cos \theta$	$\eta + 2B_3 \cos\zeta + 2B_4 d$	cosţ
$H_{12,12} = B$	$_1+2B_2\cos \theta$	$+2B_3\cos\xi+2B_4\cos\xi$	$\cos\eta$
$H_{13,13} = B$	$_1+2B_2\cos \theta$	$x+2B_3\cos\eta+2B_4\cos\eta$	cosξ
$H_{14,14} = A$	$_{1}+2A_{2}\cos \theta$	$\zeta + 2A_{3}(\cos\xi + \cos\eta)$)
$H_{6,9} = 4C_1 \cos^2 2$	$\xi \cos^{\frac{1}{2}\eta},$	$H_{8,12} = -4C_3 s$	
$H_{6,10} = -4C_2 \sin \frac{1}{2}$	$\xi \sin \frac{1}{2}\eta$,	$H_{8,14} = 4C_1 c$	$\cos\frac{1}{2}\xi\cos\frac{1}{2}\zeta$
$H_{6,12} = 4C_1 \cos^2 \frac{1}{2}$	$\xi \cos^{\frac{1}{2}\zeta}$,	$H_{9,12} = 4C_4$	$\cos\frac{1}{2}\eta \cos\frac{1}{2}\zeta$,
$H_{6,14} = -4C_2 \sin \frac{1}{2}$	$\xi \sin \frac{1}{2} \zeta$,	$H_{10, 13} = 4C_1$	$\cos\frac{1}{2}\eta \cos\frac{1}{2}\zeta$,
$H_{7,9} = -4C_3 \sin \frac{1}{2}$	$\xi \sin \frac{1}{2} \eta$,	$H_{10,14} = -4C_2$	$\sin\frac{1}{2}\eta \sin\frac{1}{2}\zeta$,
$H_{7,10} = 4C_1 \cos^2 2$	$\xi \cos^{\frac{1}{2}\eta},$	$H_{11,13} = -4C_3$	$\sin\frac{1}{2}\eta \sin\frac{1}{2}\zeta$,
$H_{7,13} = 4C_4 \cos \frac{1}{2}$	$\xi \cos_2 \zeta$,	$H_{11,14} = 4C_1$	$\cos{\frac{1}{2}\eta} \cos{\frac{1}{2}\zeta}$.
$H_{8,11} = 4C_4 \cos \frac{1}{2}$	ξ COS ₂ η,		
D. Oxy	gen-rheniu	m $(p-d)$ interaction	ns
$P_1 \equiv E_{xy}$	$_{2,xy}(0\frac{1}{2}0),$	$P_2 \equiv E_{z_3, 3z^2 - r^2} (00^{\frac{1}{2}})$	·).
$H_{7,1} = 2iP$	1 SIN [±] ξ,	$H_{9,1} = 2iP$	$1 \sin \frac{1}{2} \eta$,
$H_{11,2} = 2iP$	1 SIN [±] η,	$H_{13,2} = 2iP$	$1 \sin \frac{1}{2} \zeta$
$H_{8,3} = 2iP$	1 SINžξ,	$H_{12,3} = 2iP$	1 SINŻζ,
$H_{6,4} = -iP_2$	$\sin_2\xi$,	$\Pi_{10,4} = -\imath P_2$	$\sin_2\eta$,
$H_{4,4} = 2tP_{1,4}$	$2 \sin 2$,	$\Pi_{6,5} = V3tP_2$	sill ₂ ζ,
$11_{10,5} = -\sqrt{3}11$	$2 \sin 2\eta$.		

¹³ A. H. Kahn and A. J. Leyendecker, Phys. Rev. 135, A1321 (1964).

¹⁴ J. M. Honig, J. O. Dimmock, and W. H. Kleiner, J. Chem. Phys. (to be published).

TABLE IV. Dependence of particular energy-band states on individual tight-binding parameters.

No.	State	A_1	A_2	A_3	B_1	B_2	B_3	B_4	C_1	C_2	C_3	C_4	D_1	D_2	D_3	D_4	D_5	D_6	P_1	P_2
1	Γ_{25}	Vor	Vez	Var	Yes	Yes	Yes	Yes	Var			Yes								
1	$\Gamma_{15} \\ \Gamma_{25'}$	res	res	res	res	res	res	res	res			res	Yes	Yes	Yes					
1	Γ_{12}	¥7	37	\$7												Yes	Yes	Yes		Vez
1	$X_1 \\ X_2$	y es	y es	y es												Yes	Yes	Yes		res
1	X_3				**		**						Yes	Yes	Yes				T 7	
2	$X_{5} X_{3'}$				Yes Ves	Yes Ves	Yes Ves	Yes Ves				Ves	Yes		Yes				Yes	
1	$X_{4'}$	**			Yes	Ŷes	Yes	Ŷes				Yes								
$\frac{2}{2}$	$X_{5'}$ M_1	Yes Ves	Yes Ves		Yes	Yes	Yes	Yes	Yes	Ves						Ves	Ves	Ves		Ves
$\tilde{2}$	M_2	Yes	Yes							Yes						Ŷes	Yes	Yes		Yes
2	M_3				Yes Vec	Yes Ves	Yes	Yes			Yes Ves		Yes	Yes	Yes				Yes	
2	M_5				Yes	Yes	Yes	Yes			105		Yes		Yes				Yes	
1	$M_{4'}$	Yes	Yes	Yes	Vea	Vea	Vea	Ver												
1	$R_1^{15'}$	Yes	Yes	Yes	res	res	res	res		Yes										
2	R_{12}	Yes	Yes	Yes	X 7	37	37	37		Yes	37		T 7	*7	3.7	Yes	Yes	Yes	17	Yes
$\frac{2}{1}$	$K_{25'} R_{15'}$				Y es Yes	Y es Yes	y es Yes	Y es Yes			Y es Yes		Yes	Yes	Yes				<u>x</u> es	

agonal matrix elements of the $2p\pi$ orbitals. The parameters C_1-C_4 introduce off-diagonal nearest-neighbor π - π (C_3,C_4), σ - σ (C_2), and σ - π (C_1) interactions in the Hamiltonian matrix.

By applying the appropriate unitary transformations, the 14×14 Hamiltonian matrix of Table III can be reduced to a considerable extent at the symmetry points Γ , X, M, and R. From the results of Table I, we verify that the dimensions of these reduced submatrices are either 1×1 or 2×2. For the 1×1 cases, the energy of the particular state is linearly related to the tightbinding parameters. The 2×2 solutions are nonlinear functions of the parameters.

The dependence of particular energy levels for the ReO₃ structure on the various tight-binding parameters of Table III is summarized in Table IV. The number of states of a particular symmetry is indicated in the first column. Those parameters which are involved in determining the energy of a given energy-band state are indicated by the word yes. The actual coefficients are readily determined from the results of Table III and the appropriate unitary transformations.¹³

According to the results of Table IV, a total of 32 energy levels are available for determining the 19 LCAO parameters. Twelve states occur only once; these provide 12 linear relationships between the energy and the parameters. Ten additional linear equations can be obtained by applying the diagonal sum rule to those states which occur twice. These 22 linear equations (which include only 16 parameters, since C_1 , P_1 , and P_2 are not involved) are not linearly independent. They can be made linearly independent by eliminating the parameter D_2 . When this is done, the remaining 15 parameters can be determined by a least-squares-fitting procedure. Finally, the last four parameters (C_1 , D_2 , P_1 , and P_2) are determined from the nonlinear equations, which represent the difference in energy between those states which occur twice in Table IV.

Only the magnitudes of C_1 , P_1 , and P_2 are determined by this procedure, since the equations involve the squares of these parameters. Symmetry arguments based on the two-center approximation to these parameters suggest positive signs for C_1 and P_1 and a negative sign for P_2 .

This tight-binding scheme is readily extended to include the effects of spin-orbit coupling. Such effects are important only for the rhenium 5*d* states. For example, Herman and Skillman estimate that the spinorbit effects for oxygen 2p states are smaller by at least an order of magnitude.¹² Spin-orbit coupling between the rhenium 5*d* states is introduced by means of an additional parameter, the spin-orbit parameter ξ_{5d} . The appropriate matrix elements of the spin-orbit operator among *d* states have been tabulated by Friedel *et al.*¹⁵ These effects double the size of the secular equation, increasing it from a 14×14 to a 28×28 equation. As we shall describe later, the final value for the spin-orbit parameter ξ_{5d} was determined by optimizing the theoretical fit to the experimental Fermi-surface data.

III. RESULTS

The general nature of the APW results and the influence of the muffin-tin corrections on the ReO₃ band structure are illustrated in Figs. 2(a) and 2(b). Here we plot the energy bands along the Δ direction of the Brillouin zone. The results of Fig. 2(a) were obtained using the muffin-tin potential $V_m(\mathbf{r})$ while those of Fig. 2(b) involved $V_m(\mathbf{r}) + V_{\Delta}(\mathbf{r})$. The energy scale is relative to the constant portion of the muffin-tin potential V_{mo} . Since the APW calculations were carried out only at

¹⁵ J. Friedel, P. Lenglart, and G. Leman, J. Phys. Chem. Solids **25**, 781 (1964).





x



symmetry points, the connecting bands are meant to illustrate the connectivity required by the compatibility relations rather than the detailed band shapes.

In Figs. 2(a) and 2(b), the lowest nine bands correspond to oxygen 2p states. This is easily established by comparing with the results of Table I. These oxygen 2p bands are separated by an energy gap from three distinct groups of bands. These include the rhenium 5dbands plus some s-p or nearly-free-electron-type bands. The rhenium 5d bands are separated into t_{2g} ($\Gamma_{25'}$; X_3, X_5) and e_g ($\Gamma_{12}; X_2, X_1$) manifolds. The nearly-freeelectron bands connect $\Gamma_1, X_{4'}$, and presumably the highest X_1 state in Fig. 2(a).

From these results, it is clear that the corrections to the muffin-tin potential have an important effect on the ReO₃ band structure. These corrections shift groups of bands by energies as large as ± 0.25 Ry. These shifts reduce the oxygen-rhenium (2*p*-5*d*) band gap by almost 50% and place the *s*-*p* band above the *e_q* manifold. These corrections also increase the magnitude of the oxygen-rhenium (2*p*-5*d*) hybridization interaction. This affects the oxygen 2*p* bandwidth as well as the corresponding bandwidths of the *t*_{2*q*} and *e_q* manifolds.

In order to indicate the over-all effect of oxygenrhenium hybridization on the ReO₃ conduction bands, an additional APW calculation involving $V_m(\mathbf{r}) + V_{\Delta}(\mathbf{r})$ has been carried out. In this calculation, the potential within the oxygen muffin-tin spheres was set equal to zero (V_{mo}) . The corresponding energy-band results along the Δ direction are shown in Fig. 2(c). As expected, the nine oxygen 2p bands have disappeared. In addition, this modification to the oxygen potential has produced important changes in the conduction bands. The rhenium 6s-6p bands $(\Gamma_1,\Gamma_{15}; X_1,X_{4'})$ have shifted to lower energies so that Γ_1 lies slightly below the bottom of the 5d bands. Equally interesting and important is the fact that the ordering of the Γ_{12} and $\Gamma_{25'}$ states has been reversed and both the e_g and t_{2g} bandwidths have been reduced significantly by this modification.

The results of APW calculations at the symmetry points Γ , X, M, and R involving $V_m(\mathbf{r})+V_{\Delta}(\mathbf{r})$ [Fig. 2(b)] have been used to determine the LCAO tightbinding parameters defined in Table III, following the procedures described in Sec. II C. The values obtained for these parameters are listed in Table V under the "Fitted" heading. The two-center representation of these parameters is contained in the last column. Here, the subscripts 1 and 2 refer to nearest- and nextnearest-neighbor interactions.

If these parameters are introduced into the Hamiltonian matrix of Table III and the resulting secular equation is diagonalized, the energy-band curves shown in Fig. 3(a) are obtained. The solid lines correspond to the LCAO results while the open circles represent the APW values. The LCAO interpolation scheme succeeds in ordering correctly all states at symmetry points as a function of energy. The poorest fit involves the strongly hybridized states such as X_1 , M_1 , M_2 , and R_{12} . In these cases, the interpolated results are in error by as much as 0.04 Ry. In general, the fit is accurate to better than 0.01 Ry. A detailed comparison between the APW and LCAO results at symmetry points in the Brillouin zone is contained in Table VI.



FIG. 3. (a) Tight-binding energy bands for ReO_3 obtained by fitting APW results at symmetry points. (b) Adjusted band structure for ReO_3 .

TABLE V. Values for the fitted and adjusted ReO₃ tight-binding parameters (in rydbergs). Column 4 contains the two-center approximation to these parameters.

			Two-center
	Fitted	Adjusted	approximation
A_1	-0.2761	-0.2761	рo
A_2	0.0258	0.0258	$(p p \sigma)_2$
A_3	-0.0034	-0.0034	$(p p \pi)_2$
B_1	-0.2294	-0.2294	po
B_2	-0.0191	-0.0191	$(pp\pi)_2$
B_3	0.0049	0.0049	$(pp\sigma)_2$
B_4	-0.0001	-0.0001	$(pp\pi)_2$
C_1	0.0097	0.0097	$\frac{1}{2}(pp\sigma)_1+\frac{1}{2}(pp\pi)_1$
C_2	0.0142	0.0142	$\frac{1}{2}(pp\sigma)_1 - \frac{1}{2}(pp\pi)_1$
C_3	0.0034	0.0034	$\frac{1}{2}(pp\sigma)_1-\frac{1}{2}(pp\pi)_1$
C_4	-0.0037	-0.0037	$(pp\pi)_1$
D_1	0.0592	-0.1408	d_0
D_2	0.0000	0.0000	$(dd\pi)_1$
D_3	-0.0001	-0.0001	$(dd\delta)_1$
D_4	0.4218	0.2218	d_0
D_5	-0.0241	-0.0241	$(dd\sigma)_1$
D_6	0.0006	0.0006	$(dd\delta)_1$
P_1	0.1324	0.1200	$(pd\pi)_1$
P_2	-0.2210	-0.2450	$(pd\sigma)_1$
ξ_{5d}		0.0350	- m •

The density-of-states curve for this band structure is represented by the histogram to the right. This curve has been obtained by sampling the equivalent of 64 000 uniformly distributed points in the Brillouin zone. The actual calculations were confined to 1/48 of the Brillouin zone, as shown in Fig. 1(b). The density-of-states curve exhibits a considerable amount of structure, particularly in the oxygen 2p valence bands. A single peak due to the flat Z_2 band dominates the density-ofstates curve in the conduction band.

The ReO₃ Fermi energy is indicated by the horizontal dashed line. It occurs in the t_{2g} manifold originating from $\Gamma_{25'}$. These bands do not quite overlap the e_g bands

TABLE VI. Comparison between the calculated APW and fitted LCAO tight-binding results at symmetry points in the Brillouin zone.

State	APW	LCAO	State	APW	LCAO
Γ_1	0.6532	•••	M_2	0.8105	0.7748
Γ_{12}	0.3511	0.3514	M_1	0.4868	0.5057
$\Gamma_{25'}$	0.0546	0.0591	M_{3}	0.3275	0.3209
Γ_{15}	-0.1929	-0.1979	M_5	0.2291	0.2299
Γ_{25}	-0.2502	-0.2434	M_4	-0.1864	-0.1878
Γ_{15}	-0.3110	-0.3129	$M_{4'}$	-0.2177	-0.2109
			$M_{5'}$	-0.2701	-0.2773
X_1	0.6260	0.6457	M_{5}	-0.3500	-0.3517
$\overline{X_2}$	0.3563	0.3490	M_{3}	-0.4740	-0.4765
$\overline{X_5}$	0.2300	0.2296	M_1	-0.4797	-0.4943
X_3	0.0638	0.0593	M_2	-0.5714	-0.5512
$X_{5'}$	-0.1999	-0.1989	-		
$X_{3'}$	-0.2599	-0.2631	R_{12}	0.8030	0.7757
$X_{5'}$	-0.2823	-0.2833	$R_{25'}$	0.3215	0.3212
$X_{4'}$	-0.3028	-0.2924	$R_{15'}$	-0.1887	-0.1873
X_5	-0.3533	-0.3520	R_1	-0.4207	-0.4277
X_1	-0.5321	-0.5393	$R_{25'}$	-0.4726	-0.4762
-			R_{12}^{20}	-0.5740	-0.5411



FIG. 4. Conduction bands in ReO3 including spin-orbit coupling.

which evolve from Γ_{12} . According to the results of Fig. 2(c), the energy splitting between Γ_{12} and $\Gamma_{25'}$ is due to oxygen-rhenium (2s-5d) hybridization effects. The present calculations predict a splitting of about 0.3 Ry or 4 eV.

The t_{2g} and e_g bandwidths are due almost entirely to (p-d) hybridization effects, which are very strong in ReO₃. This is reflected by the large values for P_1 and P_2 in Table V. Typical bandwidths expected from rhenium-rhenium (d-d) overlap effects are much smaller, as the results of Fig. 2(c) indicate.

The energy-band results shown in Fig. 3(b) were obtained using the parameters listed in Table V under the "Adjusted" heading (but neglecting spin-orbit effects). As described in the following sections, these adjustments were introduced in order to provide a more accurate representation of both the experimental Fermi surface as well as the optical data. The main effect of these adjustments on the band structure is to decrease the energy gap between the oxygen 2p and rhenium 5d bands and to accentuate the hybridization effects. The density-of-states curve shown to the right in Fig. 3(b) represents 125 000 points in the Brillouin zone. It shows slightly less scatter than the corresponding curve in Fig. 3(a), but still reflects the same general features.





FIG. 6. Three-dimensional sketch of the $\alpha(a)$, $\beta(b)$, and $\gamma(c)$ sheets of the ReO₃ Fermi surface.

The effects of spin-orbit coupling on the rhenium 5dconduction bands are shown in Fig. 4 for $\xi_{5d} = 0.030$ Ry. As expected, spin-orbit effects reduce many degeneracies at symmetry points and eliminate most band-crossings along symmetry lines. Spin-orbit effects split the $\Gamma_{25'}$ state into a lower fourfold degenerate Γ_{8+} state and an upper doubly-degenerate Γ_{7+} state. Neglecting the interaction with the higher Γ_{8+} state which originates from Γ_{12} , the $[E(\Gamma_{7+})-E(\Gamma_{8+})]$ splitting is $\frac{3}{2}\xi_{5d}$.¹⁶ Similar splittings occur at the symmetry point R; however, these are reduced by oxygen-rhenium (2p-5d)hybridization effects. This hybridization is sufficiently strong to transfer reduced spin-orbit splittings to the oxygen 2p bands. Except along the Λ direction, all the bands shown in Fig. 4 are nondegenerate (neglecting spin). The lower and upper bands drawn along the Λ direction actually correspond to pairs of bands. These pairs are nearly degenerate and are separated by energies in the 0.0001-0.004 Ry range.

Central cross sections of the ReO₃ Fermi surface that are obtained from the band structure of Fig. 4 are shown in Fig. 5. The rectangular portion of this figure (ΓMRX) corresponds to a central {110} cross section. The two triangular sections (ΓMX and XRM) represent central {100} planes, one through Γ and the other through X.

The three sheets of the ReO₃ Fermi surface are labelled α , β , and γ , respectively. Each sheet corresponds to an electronlike surface centered about Γ . The α and β sheets are closed, while the γ sheet is open along (100). Because of the nearly degenerate bands along the Λ direction in Fig. 4, the β and γ sheets nearly touch along (111). The α Fermi surface sheet contains 0.093 electrons per rhenium atom. The β and γ sheets contain 0.171 and 0.736 electrons per atom, respectively.

Three-dimensional sketches of these Fermi-surface sheets are shown in Figs. 6(a)-6(c), respectively. The α sheet is rather circular in the {100} planes and slightly squared-off in the {110} planes. The reverse is true for the β sheet. According to Figs. 5 and 6(c), the open sheet consists of three intersecting cylinders. Two closed extremal orbits which exist on γ when the magnetic field is along [001] are shown in Fig. 6(c). The electronlike orbit labelled γ_1 exists on the arms of the cylinder and is centered about X. The holelike orbit labelled γ_2 , which is closed in the extended zone scheme, is centered about the point M. Another extremal orbit γ_3 [which is not shown in Fig. 6(c)] exists on the γ sheet when the magnetic field is near $\langle 111 \rangle$.

The angular variation of central extremal areas and cyclotron masses for orbits on the ReO₃ Fermi-surface have been determined via the Mueller inversion scheme.¹⁷ Figure 7 contains the results for magnetic field directions in $(1\overline{10})$ and (001) planes, respectively. The procedures and techniques involved in these calculations will be described in detail elsewhere.¹⁸ The



¹⁷ F. M. Mueller, Phys. Rev. 148, 636 (1966).
¹⁸ L. F. Mattheiss (unpublished).

¹⁶ L. F. Mattheiss, Phys. Rev. 139, A1893 (1965).

present discussion is limited to a brief summary of the methods employed.

The Mueller inversion scheme is used here to convert calculated values of the Fermi radii $[k_{\alpha}(\theta,\varphi), k_{\beta}(\theta,\varphi),$ and $k_{\gamma}(\theta,\varphi)$ with an origin at Γ] to extremal crosssectional areas. Somewhere between 40 and 50 values of $[k_{\nu}(\theta_i,\varphi_i)]^2$ are fit with 12–16 Kubic harmonics using least-squares techniques. By calculating the area $A_{\nu}(\theta,\varphi)$ as a function of energy E, the cyclotron mass,

$$\frac{m_c}{m} = \frac{\hbar^2}{2\pi m} \frac{\partial A\left(\theta,\varphi\right)}{\partial E} \bigg|_{E=E_F},\tag{5}$$

is readily determined by numerical differentiation.

The application of this method to the closed α and β sheets is straightforward. However, the open γ sheet presents more of a problem. Following the suggestion of Zornberg and Mueller,¹⁹ we "close" this sheet by adding a fictitious "tip." Although the resulting surface is highly anisotropic, the inversion scheme seems to work satisfactorily. The fictitious tip is actually determined by the inversion scheme, since the Fermi radii $[k_{\gamma}(\theta_i,\varphi_i)]^2$ are chosen to avoid the neck region entirely. The γ_1 orbits on the open sheet of the Fermi surface are determined in a similar manner, using the point X $(0,0,\pi/a)$ as an origin. In this case, the inversion scheme converts the arms of the γ Fermi-surface sheet into a "capped" cylindrical-type surface. These artificially closed surfaces appear to provide accurate results except when the orbit passes over or near the fictitious regions.

According to the results of Fig. 7, the areas of α and β branches are fairly isotropic. The area of the γ_1 branch increases like sec θ , where θ is measured from [100]. This orbit is expected to disappear at approximately $\theta=45^{\circ}$, when the orbit begins to extend onto the Γ -centered body of the γ Fermi surface sheet. According to the preceding discussion, this is precisely the region where the inversion results become unreliable. We conclude that the γ_1 results are probably not very accurate in this limiting region. The same is true for the γ_3 results shown to the right in Fig. 7. As the magnetic field is rotated from [111] toward either [001] or [110], the orbit extends onto the arms of the γ Fermi-surface sheet and approaches the fictitious tip centered at X.

There are some additional orbits which are not included in the results shown in Fig. 7. The first is represented by γ_2 in Fig. 6(c). By direct calculation, this orbit encloses an area of 0.808 Å⁻² and has a cyclotron mass of -0.87 m. Another possible orbit involves magnetic breakdown from the β to the γ Fermi-surface sheets when the magnetic field is near $\langle 110 \rangle$. The extremal area and cyclotron mass of this orbit are found to be 0.750 Å⁻² and 0.86 m, respectively. A third set of closed orbits can exist on the γ sheet. These are holelike orbits whose plane passes through the symmetry point *R* in Fig. 6. They are analogous to the γ_3 orbits of Fig. 7. We have not calculated either the area or the cyclotron mass of this set of orbits.

IV. COMPARISON WITH EXPERIMENT

The de Haas-van Alphen data of Marcus⁵ and the optical data of Feinleib *et al.*⁴ have been used to adjust the original APW model for the ReO₃ band structure. In general terms, our interpretation of the optical data suggests that the oxygen-rhenium (2p-5d) energy gap shown in Fig. 3(a) should be reduced from 0.25 Ry to about 0.05 Ry. In addition, the cyclotron-mass measurements by Marcus suggest that the $\Gamma_{25'}$ - $R_{25'}$ bandwidth of Fig. 3(a) be increased by 10% or so. The adjusted results of Fig. 3(b) reflect both these modifications to the fitted ReO₃ band structure.

A. Fermi-Surface Data

The de Haas-van Alphen data of Marcus are shown by the open and filled circles in Fig. 8. The solid lines represent the Fermi-surface areas obtained from the adjusted band structure of Fig. 4. In the regions where the γ_1 and β branches cross, Marcus makes no specific assignment of areas (or frequencies) with the individual branches. Consequently, the distinction between the open and filled circles is quite arbitrary in these regions.

It was found that the relative dimensions of the α , β , and γ sheets of the Fermi surface were reasonably sensitive to the value of the spin-orbit parameter ξ_{5d} . According to Marcus, the low-frequency oscillations have the largest amplitude; as a result, we assume that this data is the most reliable. We have determined ξ_{5d} by optimizing the theoretical fit to these data. The value $\xi_{5d}=0.030$ Ry was obtained in this manner (Table V and Fig. 4). With this value for ξ_{5d} , the theoretically determined area as a function of angle for the α sheet of the Fermi surface fits the de Haas-van Alphen data to within 5%, as shown by Fig. 8.

(001)

(110)

0.7



FIG. 8. Comparison between the de Haas-van Alphen results of Marcus (open and filled circles) and the present theoretical model (solid lines).

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¹⁹ E. I. Zornberg and F. M. Mueller, Phys. Rev. 151, 557 (1966).

The corresponding data for the γ_1 branch also agrees with the theoretical curve to within about 5%. The fit to the cross-sectional area of the orbit centered about X on the arms of the γ sheet when the magnetic field is along [001] is particularly good. However, the data appear to rise more rapidly than the theoretical curve as the magnetic field is rotated away from [001] in either the (100) or the (110) plane. This suggests that the arms of the γ sheet are more tapered than the calculations predict.

The fit of the experimental data to the theoretical curve for the β sheet is the least accurate, with deviations as large as 10%. Since these oscillations have the smallest amplitude,⁵ the data is presumably less reliable. In general, the angular variation of the data agrees well with the calculated curve.

Marcus has determined the cyclotron masses for several orbits on the α sheet of the Fermi surface. These are compared with the calculated results in Table VII. For comparison purposes, we include the cyclotron masses predicted by the fitted as well as the adjusted band-structure models for ReO₃. As mentioned previously, the discrepancy between the fitted results and the experimental values was one of the factors which was taken into account in the adjustment procedure.

Additional Fermi-surface data for ReO_3 has been obtained from magnetothermal oscillations by Graebner.⁶ By combining field sweep and rotation data, Graebner is able to determine the angular variation of Fermi-surface areas more precisely. In general, the data are consistent with Marcus's de Haas-van Alphen results.

B. Optical Data

Feinleib *et al.*⁴ have measured the reflectance of ReO₃ over a 22-eV energy range. They have analyzed their reflectance data by the Kramers-Kronig relations to determine the real and imaginary parts of the dielectric constant $\epsilon_1(\omega)$ and $\epsilon_2(\omega)$, respectively. They obtain an $\epsilon_2(\omega)$ curve which exhibits free-electron characteristics below about 2.3 eV, where interband transitions begin. They find principal peaks in ϵ_2 at 4.2, 9.3, and 14 eV, with some smaller structure at 7.0 and 8.5 eV. The structure above 2.3 eV is attributed to interband transitions by these authors.

In the present investigation, we have applied the simplest possible model for interpreting the structure in the ϵ_2 curve for ReO₃. This model ignores the transition-

TABLE VII. Cyclotron masses for α orbits on the ReO₃ Fermi surface with the magnetic field near $\langle 111 \rangle$ in a $\{1\overline{10}\}$ plane.

H	Experiment ^a m_c/m	Fitted m _e /m	$\begin{array}{c} \text{Adjusted} \\ m_{c}/m \end{array}$
(111)-9°	0.68	0.758	0.637
$\langle 111 \rangle$	0.67	0.757	0.635
$\langle 111 \rangle + 17^{\circ}$	0.65	0.759	0.645

* Reference 5.

140 **Ш 120** STA 00 Ч 80 DENSITY 60 40 JOINT 20 12 13 14 3 5 9 10 11 15 16 17 2 4 6 8 ENERGY IN eV FIG. 9. Joint density-of-states curve for the adjusted band structure of Fig. 3(b).

probability matrix elements and involves only a calculation of the joint density of states $J_{n,s}(\omega)$, where

$$J_{n,s}(\omega) = \frac{2}{(2\pi)^3} \int_{\omega_{n,s}(\mathbf{k})=\omega} \frac{ds}{|\nabla_k \omega_{n,s}(\mathbf{k})|}, \qquad (6)$$

 $\hbar\omega_{n,s}(\mathbf{k}) = E_s(\mathbf{k}) - E_n(\mathbf{k}), ds$ is an element of area on the surface for which $\omega_{n,s}(\mathbf{k}) = \omega$, and *n* and *s* refer to occupied and unoccupied states, respectively.²⁰

The joint density-of-states curve which is obtained from the adjusted band structure of Fig. 3(b) is shown in Fig. 9. Spin-orbit effects have been neglected in this calculation. The curve has been obtained by sampling the equivalent of 125 000 uniformly distributed points in the Brillouin zone. These points are identical with those used to determine the density-of-states curve in Fig. 3(b).

The main features of this joint density-of-states curve are a large narrow peak centered near 3.5 eV plus a fairly broad peak extending from approximately 6–12 eV. The general shape of the joint density-of-states curve resembles the valence band density-of-states curve in Fig. 3(b). This similarity may be due to the peak in the conduction-band density of states at 0.075 Ry, which results from the (X_5-M_5) degeneracy. It appears that this peak serves to pick out the structure in the valence-band density of states.

In view of the simplified model introduced here to interpret the ReO₃ optical data, we have concentrated our attention on one feature of the data, namely, the low-energy peak in ϵ_2 at 4.2 eV. The fitted band structure of Fig. 3(a) produced a joint density of states similar to that shown in Fig. 9 except that the first large peak occurred near 6.5 rather than 4.5 eV. The fitted parameters were adjusted to lower this peak by about 2 eV.

The experimental ϵ_2 curve contains additional peaks, one extending from about 8–12 eV and the other centered near 14 eV. In terms of the joint density-ofstates model, these peaks tower over the 4.2-eV peak in

²⁰ David Brust, Phys. Rev. 134, A1337 (1964).

height. The 9.3-eV peak is about three times larger than the 4.2-eV peak. The 14-eV peak is roughly five times larger. According to Fig. 9, the present calculations provide a rather poor representation for this highenergy structure in ϵ_2 . It is possible that variations in the transition-probability matrix elements are responsible for these differences. More accurate dielectric constant calculations involving the techniques proposed by Dresselhaus and Dresselhaus might resolve these difficulties.21

In proposing this interpretation for the ϵ_2 curves for ReO₃, we are encouraged by the fact that a similar interpretation is consistent with Cardona's ϵ_2 curves for the structurally-related compounds SrTiO₃ and BaTiO₃.²² In both materials, Cardona finds a peak in ϵ_2 near 5 eV, with a shoulder on the low-energy side. The shape of this peak is quite similar to the one centered about 4.5 eV in Fig. 9. The high-energy structure in ϵ_2 is also in reasonable agreement with this joint densityof-states curve. Of course, the band structures for SrTiO₃ and BaTiO₃ differ from the present ReO₃ results; however, as we discuss in Sec. V, the band structures for these materials are expected to share many common features.

V. DISCUSSION

The traditional approach to the theory of transitionmetal oxides involves the ligand-field and molecularorbital theories.^{23,24} As a result, it is appropriate to begin this discussion of the ReO3 band structure by establishing a connection between the present energy-band approach and the ligand-field and molecular-orbital theories of transition-metal oxides.

Ligand-field theory treats the transition-metal ions in an electrostatic crystal-field environment which arises from the negatively charged neighboring ligand ions (which are often considered to be negative point charges). This theory neglects completely any covalent bonding or hybridization effects between the transitionmetal ion and its neighbors. In an octahedral environment, the electrostatic repulsion due to the negatively charged ligands causes a splitting of the five d orbitals into two separate groups of levels; these consist of $t_{2g}(xy, yz, \text{ and } zx)$ and $e_g(3z^2-r^2 \text{ and } x^2-y^2)$ orbitals, respectively. Since the e_q orbitals point in the direction of the ligands, they experience greater electrostatic repulsion than the t_{2g} orbitals. As a result, the t_{2g} level is at a lower energy than the e_g level in such an environment. The energy splitting between the t_{2g} and e_g levels is normally denoted by the parameter 10Dq.

On the other hand, the molecular-orbital theory is more realistic in that it treats both the transition-metal ion as well as its neighboring ligands directly in terms of symmetrized combinations of atomic-type orbitals. As a result, this theory includes covalency and hybridization effects. In a typical application, this method would be used to treat a single ReO_6 complex, thereby neglecting the effects of lattice periodicity. When the lattice periodicity is taken into account, this molecularorbital scheme becomes identical with the tight-binding method in solids.

In its simplified form, the results of a molecularorbital calculation are related to the band structure at the center (Γ) and corner (R) of the Brillouin zone. For an octahedrally coordinated system, this is seen by comparing the bonding and antibonding molecular orbitals²⁴ with the corresponding LCAO tight-binding functions.¹³ Therefore, this molecular-orbital approach is meaningful only when the crystal-field splitting 10Dqis large compared to the t_{2g} and e_g bandwidths. The results of Figs. 3(a) and 3(b) suggest that this condition is not satisfied in ReO₃ since the t_{2g} and e_g bandwidths are comparable in magnitude with 10Dq.

The magnitude of 10Dq plays a central role in the ligand-field and molecular-orbital theories. Our discussion is concerned primarily with the strong-field limit, where 10Dq is large compared to atomic multiplet splittings and spin-orbit effects. Attempts to calculate 10Dq via the ligand-field and molecular-orbital theories have enjoyed only limited success. KNiF₃ has been a favorite system for such attempts. Although there is still some controversy, the consensus is that ligand-field effects account for about one-third of 10Dq in KNiF₃ while the remaining two-thirds are attributed to covalency effects.²⁴ Our interpretation of the results shown in Figs. 2(a) and 2(b) involves a similar explanation for the 10Dq splitting in ReO₃. We attribute the 0.2 Ry contribution to 10Dq in the muffin-tin results of Fig. 2(a) to oxygen-rhenium (2s-5d) covalency effects. In Fig. 2(b), the additional contribution of 0.1 Ry to 10Dqis due to the muffin-tin corrections $V_{\Delta}(\mathbf{r})$, which are analogous to the electrostatic ligand-field effects.

The results of Fig. 2(c) emphasize the crucial effect that oxygen-rhenium covalency has on the ReO3 conduction bands. Unlike the fcc and bcc transition metals,^{25,26} the Γ_{12} - $\Gamma_{25'}$ splitting in Fig. 2(b) is not determined by the two-center (d-d) integrals which are responsible for the over-all d bandwidth. These parameters are responsible for the reduced splitting of opposite sign in Fig. 2(c). The fact that Γ_{12} interacts with the oxygen 2s levels is a special feature of the ReO₃ and perovskite structures. It does not occur, for example, in the NaCl or CsCl structures.

The ReO₃ conduction bands also differ from those of the transition metals in that the bottom of the s band Γ_1

²¹ G. Dresselhaus and M. S. Dresselhaus, Phys. Rev. 160, 649

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 ²³ Leslie E. Orgel, An Introduction to Transition-Metal Chemistry (Wiley-Interscience, Inc., New York, 1966), 2nd ed.
 ²⁴ J. Owen and J. H. M. Thronley, in Reports on Progress in Physics, edited by A. C. Stickland (Stonebridge Press, Bristol, England, 1966).

 ²⁵ F. M. Mueller, Phys. Rev. 153, 659 (1967).
 ²⁶ R. A. Deegan, Phys. Rev. 171, 659 (1968).

lies above the *d*-band states. From the results of Fig. 2 and Table I, it is clear that oxygen-rhenium (2s-6s) and (2s-6p) hybridization are responsible for shifting to higher energies the plane-wave-type bands which originate from Γ_1 .

The position and width of the oxygen 2s bands in ReO₃ also suggest strong hybridization effects. On the energy scale of Fig. 2, these bands are centered at -1.4 Ry and have a 0.11 Ry bandwidth. On this same energy scale, the energies of the HFS oxygen 2s and 2patomic levels¹² are -1.2 and -0.12 Ry, respectively. Differences between the muffin-tin and atomic potentials could introduce moderate energy shifts in the solid. However, we note that the top of the oxygen 2p bands in Fig. 2(a) is close to the atomic 2p level while the 2s bands are located approximately 0.2 Ry below the oxygen 2s atomic level. We attribute this shift to oxygen-rhenium 2s-6s, 2s-6p, and 2s-5d hybridization. Those oxygen 2s band states which interact with the rhenium 5d and 6s states $(\Gamma_1, \Gamma_{12}; X_1, X_2; M_1)$ are all grouped within 0.025 Ry of the bottom of the 2s band (Γ_1) , while the remaining states which interact with rhenium 6p levels $(X_{4'}; M_{5'}; R_{15})$ are grouped within 0.005 Ry of the top.

These considerations lead to a closely related question concerning the validity of the ionic model in transitionmetal oxides. As described in Sec. II, the starting point of the present APW calculation corresponds to the nonionic limit since it involves an ad hoc crystal potential which is derived from neutral atom charge densities. We have not attempted to achieve a self-consistent calculation using, for example, the iteration techniques described by DeCicco.⁷ However, we have calculated the charge distribution which is implied by the LCAO eigenvectors. In this calculation, we assume that the orthogonalized orbitals which represent the oxygen 2pand rhenium 5d states are completely localized about the oxygen and rhenium sites, respectively. With this approximation, we can calculate a separate density of states for the oxygen 2p and rhenium 5d states, weighting each eigenvalue by the sums of squares of the appropriate eigenvector components. At the Fermi energy, this calculation implies a charge distribution of (Re^{+4}) $(O^{-4/3})_3$ for the band structure of Fig. 3(a) and (Re^{+3}) $(O^{-1})_3$ for that of Fig. 3(b). These results probably set an upper limit for the ionicity of Re and O in ReO₃.

There has been considerable speculation regarding the nature of the conduction bands in ReO₃ and structurally related compounds such as SrTiO₃ and Na_xWO₃. A good summary of these models is contained in Ref. 4. In molecular-orbital terminology, the present calculations predict a conduction band arising from π bonding between rhenium t_{2g} and oxygen 2p orbitals. The t_{2g} bandwidth results from the fact that this bonding is zero at the center of the Brillouin zone but nonzero at the zone boundaries. These results substantiate the interpretation by Ferretti *et al.*²⁷ of the observed metallic conductivity in ReO₃ in terms of this model.

These calculations for ReO₃ reproduce many of the essential features in the band-structure model for SrTiO₃ that has been derived from a tight-binding calculation by Kahn and Leyendecker.¹³ SrTiO₃ contains one less electron per unit cell than ReO₃. According to the Kahn-Leyendecker model, the 18 valence electrons are just sufficient to fill the nine oxygen 2p bands. These are separated by a 3-eV energy gap from the unoccupied t_{2g} manifold of conduction band states. The Kahn-Leyendecker model anticipates the value of the tight-binding approximation in these materials. It also anticipates the importance of hybridization effects in determining the valence and conduction bandwidths.

The similarity in the conduction bands predicted by the two calculations is quite striking. The main differences are in the t_{2g} and e_g bandwidths and the 10Dqsplitting of $\Gamma_{25'}$ and Γ_{12} . Preliminary APW calculations for SrTiO₃ suggest that the Kahn-Leyendecker model underestimates these parameters. The ReO₃ valence bands differ, however, in several details from those predicted by the Kahn-Levendecker model. To some extent, these differences originate from the sign of their two-center oxygen-oxygen $(p p \sigma)$ and $(p p \pi)$ integrals. These calculations also differ regarding the location in the Brillouin zone of the conduction-band minimum. Kahn and Leyendecker predict a conduction-band minimum at X. The present ReO_3 (and preliminary SrTiO₃) calculations favor a minimum at Γ . These differences depend entirely on the sign of the transfer integral $D_3 \equiv E_{xy,xy}(001)$ of Table III. According to Table V, this integral is equivalent to $(dd\delta)_1$ in the twocenter approximation. The present calculations predict a negative rather than a positive sign for D_2 . This integral is related directly to the energy difference between $\Gamma_{25'}$ and X_3 . In fact,

$$4E_{xy,xy}(001) = E(\Gamma_{25'}) - E(X_3).$$

Using the APW values for $E(\Gamma_{25'})$ and $E(X_3)$ from Table VI, we obtain a value for $E_{xy,xy}(001) = -0.0023$ Ry. This is somewhat larger than the fitted value in Table V, where $D_3 = 0.0001$ Ry. According to the results of Fig. 4, spin-orbit coupling favors a conduction-band minimum at Γ by lowering the energy of Γ_{8^+} with respect to the relativistic counterpart of X_3 .

Our earlier discussion concerning the covalent contribution to the 10Dq splitting in ReO₃ illustrates the limitations of our tight-binding model. The present model represents this 10Dq splitting in terms of a ligand-field effect and ignores the covalent contribution entirely. In principle, the LCAO approach can be generalized to include these effects by adding oxygen 2s and rhenium 6s and 6p bands. Adding the oxygen 2s states increases the size of the (nonrelativistic) Hamil-

²⁷ A. Ferretti, D. B. Rogers, and J. B. Goodenough, J. Phys. Chem. Solids **26**, 2007 (1965).

TABLE VIII. Comparison of oxygen-oxygen (p-p) and oxygenrhenium (p-d) tight-binding parameters (in rydbergs) which have been determined from two separate APW calculations, as described in the text.

	(APW)1	$(APW)_2$
A_1	-0.2761	-0.2443
A_2	0.0258	0.0109
$\overline{A_3}$	-0.0034	-0.0026
B_1	-0.2294	-0.2332
B_{2}	-0.0191	-0.0179
B_3	0.0049	0.0043
B_4	-0.0001	-0.0001
C_1	0.0097	0.0094
$\tilde{C_2}$	0.0142	0.0214
C_3	0.0034	0.0046
C_4	-0.0037	-0.0040
P_1	0.1324	0.1198
P_{\bullet}	-0.2210	-0.2476

tonian matrix from 14×14 to 17×17 . More important, these improvements increase the number of parameters involved. Consequently, we have not attempted to introduce these effects in our model. A fairly direct method has been used to evaluate the physical basis for this tight-binding parameterization scheme. This method has involved an additional APW calculation in which the rhenium 5d bands have been artificially lowered in energy with respect to the oxygen 2p bands. This has been achieved by rigidly shifting the logarithmic derivatives for the rhenium d states to lower energies.¹⁰ If our tight-binding scheme were rigorously correct, only the rhenium-rhenium (d-d) and the oxygen-rhenium (p-d) parameters should be affected. However, the results show that the oxygen-oxygen (p-p) parameters are also affected. The p-p and p-dparameters obtained from this calculation are listed in Table VIII under the heading (APW)₂. These are compared with the previous results of Table V, $(APW)_1$.

The values assumed for the adjusted parameters P_1 and P_2 of Table V were influenced by the results of Table VIII. A larger value for P_1 would further decrease the cyclotron masses of Fig. 7 and Table VII. In connection with our interpretation of the ReO₃ optical data, a larger value would also produce overlap between the t_{2g} and oxygen 2p bands. We have not determined the possible effects of this overlap on the ReO₃ Fermi surface.

Feinleib *et al.*⁴ have proposed a model band structure for ReO₃ which is based on a simplified version of the Kahn-Leyendecker scheme. They assume that the Fermi energy occurs within the manifold of the t_{2g} bands and neglect the e_g and oxygen 2p bands. According to their interpretation, there is no evidence for interband transitions below about 2 eV in their optical data. Since the Kahn-Leyendecker model permits interband transitions at very low energies, they have investigated the possible effects of spin-orbit coupling on these states. Their model involves a spin-orbit parameter which is comparable with the Fermi energy (measured from the bottom of the t_{2g} band). As a result, spin-orbit-induced splittings are large enough to eliminate the low-energy interband transitions.

In the present model for the ReO₃ band structure, spin-orbit coupling produces rather small effects, as shown by the results of Fig. 4. The nearly degenerate bands which occur in the Λ direction indicate the possibility of magnetic breakdown between the β and γ sheets of the Fermi surface when the magnetic field is along [110]. Thus far, there has been no experimental evidence for such magnetic breakdown effects in ReO3. The fitted value for the spin-orbit parameter $\xi_{5d} = 0.030$ Ry is close to the (interpolated) Herman-Skillman atomic spin-orbit parameter for rhenium,¹² namely, 0.032 Ry. Unlike the ReO₃ band model of Feinleib et al., the present model allows interband transitions down to very low energies. However, the results of Fig. 9 suggest that these low-energy transitions will be weak from a density of states point of view.

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