

TABLE III. Values of the integrals  $R_2$  for positive ions, neutral atoms, and negative ions with  $d^n$  configurations (in  $\text{cm}^{-1}$ ).

Configuration	State	Positive Ion		Neutral Atom		Negative Ion	
		System	$R_2$	System	$R_2$	System	$R_2$
$3d^2$	${}^3_2F$	Sc <sup>+</sup>	0.69	Ti	1.35	Sc <sup>-</sup>	0.56
$3d^3$	${}^4_3F$	Ti <sup>+</sup>	1.05	V	1.81	Ti <sup>-</sup>	0.95
$3d^4$	${}^5_4D$	V <sup>+</sup>	1.45	Cr	2.32	V <sup>-</sup>	1.36
$3d^5$	${}^6_5S$	Cr <sup>+</sup>	1.92	Mn	2.91	Cr <sup>-</sup>	1.84
$3d^6$	${}^5_4D$	Mn <sup>+</sup>	2.37	Fe	3.47	Mn <sup>-</sup>	2.29
$3d^7$	${}^4_3F$	Fe <sup>+</sup>	2.91	Co	4.12	Fe <sup>-</sup>	2.83
$3d^8$	${}^3_2F$	Co <sup>+</sup>	3.53	Ni	4.86	Co <sup>-</sup>	3.44
$3d^9$	${}^2_1D$	Ni <sup>+</sup>	4.21	Cu	5.68	Ni <sup>-</sup>	4.13
$3d^{10}$	${}^1_0S$	Cu <sup>+</sup>	4.99	Zn	6.59	Cu <sup>-</sup>	4.90

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<sup>1</sup>The dipole-dipole electron-spin coupling constants for the same systems considered here, and some iso-electronic series, have been evaluated by J. Thorhallsson and S. Fraga (unpublished results).

<sup>2</sup>See, e.g., G. Malli and S. Fraga, *Theoret. Chim. Acta* (Berlin) **6**, 278 (1966).

<sup>3</sup>A. M. Sessler and H. M. Foley, *Phys. Rev.* **92**, 1321 (1953).

<sup>4</sup>L. Armstrong, Jr., *Phys. Rev.* **170**, 122 (1968).

<sup>5</sup>L. Armstrong, Jr., *J. Math. Phys.* **7**, 1891 (1966).

<sup>6</sup>See, e.g., B. R. Judd, *Operator Techniques in Atomic Spectroscopy* (McGraw-Hill Book Co., New York, 1963).

<sup>7</sup>G. Racah, *Phys. Rev.* **63**, 367 (1943).

<sup>8</sup>C. W. Nielson and G. F. Koster, *Spectroscopic Coefficients for the  $p^n$ ,  $d^n$ , and  $f^n$  Configurations* Massachusetts Institute of Technology Press, Cambridge, Mass., 1963).

<sup>9</sup>E. Clementi, *Tables of Atomic Functions* (International Business Machines Corp., San Jose, Calif., 1965).

<sup>10</sup>G. Malli, *Can. J. Phys.* **44**, 3121 (1966).

<sup>11</sup>See, e.g., C. Fisk and S. Fraga, *Can. J. Phys.* **46**, 1140 (1968).

<sup>12</sup>G. Malli and S. Fraga, *Theoret. Chim. Acta* (Berlin) **7**, 80 (1967); J. Thorhallsson, C. Fisk, and S. Fraga, *J. Chem. Phys.* **48**, 2925 (1968).

<sup>13</sup>S. Fraga and G. Malli, *J. Chem. Phys.* **46**, 4754 (1967).

## Accurate Analytical Self-Consistent-Field Wave Functions for $\text{Ag}^+$

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Self-consistent-field calculations by the analytical expansion method were carried out for the ground state of  $\text{Ag}^+$  ( $Z=47$ ),  $4d^{10} {}^1S$ . All the expansion basis sets were selected so as to satisfy precisely the cusp condition. The deviations from the tail nodal extinction for the radial functions  $P_{i\lambda}(r)$  do not exceed 0.00008. The results represent the most accurate analytical wave functions for  $\text{Ag}^+$  available.

### INTRODUCTION

Accurate analytical self-consistent-field (SCF)

wave functions for the silver ion are needed for certain applications in solid-state physics.<sup>1,2</sup> Also, the knowledge of such wave functions will

shift the frontier<sup>3</sup> of accurate analytical calculations further into the periodic system of atoms. (Furthermore, the actual accuracy achieved in this work for the tail nodal extinction has been unprecedented for such a heavy atomic system).

We carried out the SCF calculations for the ground state of  $\text{Ag}^+$ ,  $4d^{10}$ ,  $^1\text{S}$ , by means of the analytical expansion method.<sup>4</sup> The computation was done with a program<sup>4</sup> for the IBM line of large-memory computers.<sup>5</sup> All of the orbital exponents  $\xi_{\lambda p}$  of the basis functions (except those restricted by the cusp condition) were subjected to repeated and exhaustive optimizations.

### RESULTS AND DISCUSSION

The terminology, the atomic units,<sup>6</sup> and the notation<sup>4</sup> used are the usual ones.<sup>7</sup>

The most important results are given in Tables I and II. The numerical tabulations of the orbital wave functions, etc., are given in Tables III-V<sup>8</sup>; these tables are deposited at the Library of Congress<sup>8</sup>; in these tables the radial part of the orbital  $\varphi_{i\lambda\alpha}(r, \theta, \varphi)$  multiplied by  $r$  is denoted by  $P_{i\lambda}(r)$ . [The numerical Tables III and IV<sup>8</sup> of  $P_{i\lambda}(r)$  present a direct demonstration of the satisfaction of the tail nodal extinction.]

We assume that the radial wave functions  $P_{i\lambda}(r)$  represent the Hartree-Fock values to about two to three decimal places. The Hartree-Fock total

TABLE I. The total energy  $E$  (a.u.), the virial theorem and the orbital energies  $\epsilon$  (a.u.) for  $\text{Ag}^+$ ,  $4d^{10}$ ,  $^1\text{S}$ .

Quantity	Value
$E$	-5197.478 3
$E_p/E_k$	-2.000 005 2
$\epsilon$	
1s	-914.121
2s	-135.161
3s	-26.200
4s	-4.285
2p	-125.465
3p	-22.229
4p	-2.964
3d	-14.962
4d	-0.824

energy is represented to about five significant figures. The registered deviations from the tail nodal extinction for the radial functions  $P_{i\lambda}(r)$  never exceed 0.000 08 (Tables III and IV<sup>8</sup>). (To achieve such high accuracy for a system with 46 electrons required additional laborious optimizations on certain well chosen and coupled exponents  $\xi_{\lambda p}$ .) Since the cusp condition<sup>9</sup> is precisely satisfied for all the orbitals by a special choice of the

TABLE II. The orbital exponents  $\xi_{\lambda p}$  of the basis functions and the eigenvectors of coefficients  $C_{i\lambda p}$  for  $\text{Ag}^+$ ,  $4d^{10}$ ,  $^1\text{S}$ .

Basis function	Exponents	Eigenvectors			
		1s	2s	3s	4s
1s	47.000 00	0.988 74	-0.318 28	0.137 97	-0.057 43
3s	52.776 82	0.009 65	0.000 49	0.001 21	0.004 71
3s	40.614 96	0.008 73	0.058 79	-0.034 66	-0.010 39
3s	30.234 48	-0.001 25	0.430 73	-0.171 66	0.138 25
3s	22.157 46	0.000 70	0.553 31	-0.462 89	0.075 30
3s	16.060 27	-0.000 33	0.034 28	-0.022 36	0.236 20
3s	10.837 37	0.000 07	0.010 93	0.866 20	-0.838 49
3s	8.863 55	-0.000 04	-0.005 67	0.348 51	0.064 31
4s	5.436 19	0.000 01	0.000 37	0.010 11	0.899 87
4s	3.543 49	-0.000 00	-0.000 16	-0.001 03	0.247 00
		2p	3p	4p	
2p	23.500 00	0.842 23	-0.377 76	0.148 83	
4p	28.812 65	0.068 80	-0.034 64	0.010 05	
4p	21.425 75	0.132 18	0.053 64	-0.018 04	
4p	13.849 32	0.004 97	0.629 08	-0.330 81	
4p	9.723 49	-0.000 21	0.413 10	-0.181 19	
4p	5.312 43	-0.000 06	0.007 48	0.813 95	
4p	3.264 36	-0.000 03	0.000 20	0.348 15	
		3d	4d		
3d	15.666 67	0.432 05	-0.138 36		
5d	26.702 05	0.015 26	-0.006 41		
5d	15.717 64	0.323 98	-0.089 99		
5d	10.877 49	0.330 50	-0.089 20		
5d	7.367 87	0.038 92	0.292 59		
5d	4.456 64	0.001 68	0.559 24		
5d	2.509 96	-0.000 33	0.353 05		
5d	1.101 40	0.000 08	0.018 92		

basis set,<sup>9</sup> the cusp values are not included in the tables. The virial theorem<sup>10</sup> is satisfied to almost six significant figures (Table I).

A preliminary and orientational numerical SCF calculation *without* exchange on  $\text{Ag}^+$  was published by Black.<sup>11</sup> The numerically represented orbitals  $P_{i\lambda}(r)$  of this work (Tables III and IV<sup>8</sup>) disagree considerably with those by Black (after normalization). A large disagreement is, of course, typical, when comparing calculations with and without exchange.<sup>3,12</sup>

A numerical SCF calculation *with* exchange on  $\text{Ag}^+$  was published by Worsley.<sup>13</sup> The numerically represented orbitals  $P_{i\lambda}(r)$  of this work (Tables III and IV<sup>8</sup>) usually agree with those by Worsley to at least two decimal places.<sup>14</sup>

Earlier an approximate analytical calculation for  $\text{Ag}^+$  was carried out, with minimum basis sets for the *s* and *p* symmetries, and with four basis functions for the *d* symmetry (exponents

optimized).<sup>15</sup> We found that at least for the 3*d* and 4*d* orbitals the numerically represented functions  $P_{i\lambda}(r)$  agree with the more accurate ones of this work to almost two decimal places. For the *s* and *p* symmetries the shortcomings (due to their minimum set description) are, of course, much more pronounced, in comparison with an accurate description used in this work.

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<sup>1</sup>Some of the currently active studies involving  $\text{Ag}^+$  are related to the investigation of photographic processes and of the relevant ionic crystals. [See, e.g., J. Malinowski, *Contemp. Phys.* **8**, 285 (1967); H. E. Spencer, L. E. Brady, and J. F. Hamilton, *J. Opt. Soc. Am.* **57**, 1020 (1967); H. E. Spencer and D. H. Shares, *ibid.*, **57**, 508 (1967).]

<sup>2</sup>J.-L. Calais, private communication. ( $\text{Ag}^+$  wave functions are needed for currently active studies on silver halides; the deviations of their elastic constants from Cauchy relations are quite unusual, in comparison with alkali halides.) [See also, G. Petterson, J. Vallin, J.-L. Calais, and K. Mansikka, *Arkiv Fysik* **34**, 371 (1967); J. Vallin, *ibid.* **34**, 367 (1967).]

<sup>3</sup>C. C. J. Roothaan and M. Synek, *Phys. Rev.* **133**, A1263 (1964).

<sup>4</sup>C. C. J. Roothaan and P. S. Bagus, in *Methods in Computational Physics*, edited by B. Alder, S. Fernbach, and M. Rotenberg (Academic Press Inc., New York, 1963), Vol. 2.

<sup>5</sup>The original computer program was somewhat modified.

<sup>6</sup>D. R. Hartree, *The Calculation of Atomic Structures* (John Wiley & Sons, Inc., New York, 1957).

<sup>7</sup>M. Synek, A. E. Rainis, and E. A. Peterson, *J. Chem.*

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<sup>8</sup>Tables III and IV of the radial wave functions  $P_{i\lambda}(r)$ , and Table V of the total electronic density, numerically tabulated to five decimal places, are not printed in this article. These tables may be obtained by ordering NAPS Document 00197 from ASIS National Auxiliary Publications Service, c/o CCM Information Sciences, Inc., 22 West 34th Street, New York, New York 10001; remitting \$1.00 for microfiche or \$6.50 for photocopies.

<sup>9</sup>P.-O. Löwdin, *Phys. Rev.* **94**, 1600 (1954); C. C. J. Roothaan, L. M. Sachs, and A. W. Weiss, *Rev. Mod. Phys.* **32**, 186 (1960); C. C. J. Roothaan and P. S. Kelly, *Phys. Rev.* **131**, 1177 (1963); **133**, 11 (E) (1964).

<sup>10</sup>P.-O. Löwdin, *J. Mol. Spectry.* **3**, 46 (1959).

<sup>11</sup>M. M. Black, *Mem. Proc. Manchester Lit. Phil. Soc.* **79**, 29 (1935).

<sup>12</sup>M. Synek, A. E. Rainis, and C. C. J. Roothaan, *Phys. Rev.* **141**, 174 (1966).

<sup>13</sup>B. H. Worsley, *Proc. Roy. Soc. (London)* **A247**, 390 (1958).

<sup>14</sup>Due to different scales, it is convenient to start the comparison for the values  $r=0.010\,043$  and  $r=0.999\,10$  of Worsley.

<sup>15</sup>M. Synek and F. Schmitz, *Phys. Letters* **27A**, 349 (1968).