

Decay of $\text{Ho}^{163}\dagger$

P. K. HOPKE*, J. S. EVANS†, AND R. A. NAUMANN§

Frick Chemical Laboratory, Palmer Physical Laboratory, and Princeton-Pennsylvania Accelerator, Princeton, New Jersey

(Received 20 February 1968)

A sample of Ho^{163} was prepared by neutron irradiation of Er^{162} followed by chemical purification and isotope separation. This nuclide Ho^{163} appears to decay by electron capture from the M and higher shells. A half-life of 33 ± 23 yr has been estimated by counting the M Auger electrons and M x rays from a sample over a period of two years. A search for L x rays has been made. The M -to- L capture ratio is greater than 1040. From this datum the Q value for electron capture can be deduced to be 9.1 ± 1.5 keV.

HOLMIUM-163 is one of six unstable neutron deficient nuclei (Ti^{44} , Ho^{163} , Pt^{193} , Hg^{194} , Pb^{202} , and Pb^{205}) for which K electron capture appears to be energetically impossible. However, Ho^{163} is the only nucleus in this group which also cannot capture from the L shell. This nucleus must have a very small Q value for electron capture and a very long half-life.

A sample of erbium enriched in the mass-162 isotope was irradiated for six months at the Materials Testing Reactor to produce the material used in these studies. After chemical fractionation of the products using ion-exchange chromatography, a mass spectrum of the holmium fraction revealed the presence of a long-lived isotope with mass number 163. In order to study the radiations accompanying the electron-capture decay of this isotope to stable Dy^{163} , a sample free from radioactive Ho^{166} was obtained by dispersing a portion of this holmium fraction using the isotope-separation facility at the Niels Bohr Institute in Copenhagen. The Ho^{163} was collected on 10-mil aluminum foil at the full 50 kV separator accelerating voltage. When this sample was examined using a windowless proportional counter, only very soft radiations of approximately 1.3 keV could be detected. These radiations have been assigned as Auger electrons and x radiations following M electron capture.^{1,2}

A method for investigating possible decay of this sample has been devised.³ The shape of the Fe^{55} K x-ray spectrum observed in a windowless flow proportional counter of the Sugarman type has been investigated as a function of the applied voltage, using a counting gas with composition 90% argon, 10% methane. A 200-V interval (2100 to 2300 V) has been found over which ideal proportional operation was closely approximated. In this region the shape of the observed pulse-height spectrum remained constant (superposable) when

plotted using logarithmic axes for both the number of events and the corresponding pulse amplitude. Since the relative shape of the spectrum remained invariant in this voltage range, a reproducible measure of the counting rate could be found by recording a spectrum while the counter is operated within the 200-V interval and then summing the events contained in all channels between predetermined upper and lower limits. These limits were located as the channels where the events accumulated equaled some specified fraction of the events recorded in the peak of the spectrum.

Spectra due to similarly mounted sources of Fe^{55} , Zn^{65} , and Ho^{163} were observed under these conditions for 24 months. Typical spectra obtained are shown in Fig. 1 using linear axes. The shaded portion shown in

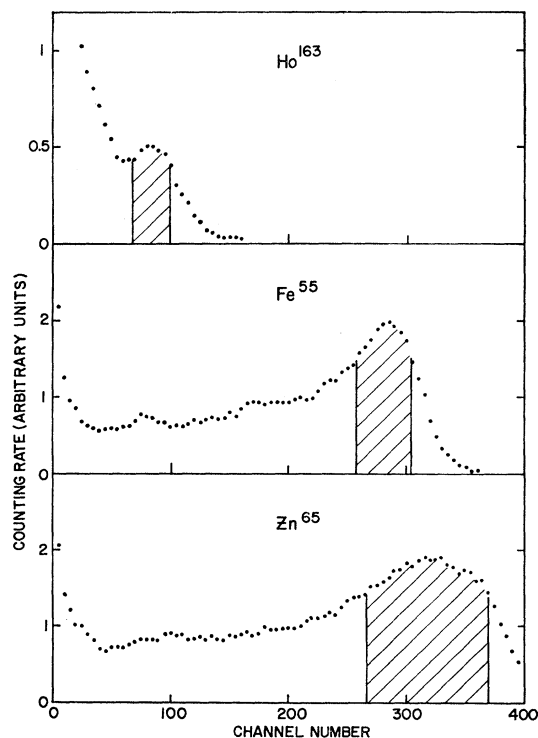


FIG. 1. X-ray spectra of Ho^{163} , Fe^{55} , and Zn^{65} taken with a windowless proportional counter. The Zn^{65} spectrum was recorded with one-half the amplifier gain used for the other spectra.

† Work supported by the U. S. Atomic Energy Commission.

* AEC Special Fellow in Nuclear Science and Engineering.

† Present address: Lawrence University, Appleton, Wis.

§ Currently on leave at the Niels Bohr Institute, Copenhagen, Denmark.

¹ R. A. Naumann, M. C. Michel, and J. C. Power, *J. Inorg. Nucl. Chem.* **15**, 195 (1960).

² R. A. Naumann, M. C. Michel, and J. C. Power, *Bull. Am. Phys. Soc.* **6**, 73 (1961).

³ J. S. Evans, thesis, Princeton University, 1965, PPAD-572F (unpublished).

each spectrum represents the spectral region summed to investigate the decay of these samples. For Fe⁵⁵ and Zn⁶⁵, those channels where the counting rate was greater than 75% of the peak height were included in the summation. For Ho¹⁶³, channels with a counting rate greater than 80% of the peak height have been included in the summation.

The results from approximately 24 months of observation are shown in Fig. 2. The points have been fitted to exponential decay curves using a least-squares method in which all points have been given equal weight. The values of the half-lives have been computed from the slopes of the lines, and errors have been computed from the scatter of experimental points with respect to the least-squares fitted curve. These values are given in Table I together with the accepted half-lives for Fe⁵⁵ and Zn⁶⁵. A comparison of these values indicates the reliability of our counting method for low-energy radiations.

TABLE I. Half-life determinations.

Nuclide	Half-life	Reported half-life
Zn ⁶⁵	264±9 days	245.0 ±0.8 days ^a
		243.5 ±0.8 days ^b
		246.4 ±2.2 days ^c
		245.7 ±1.1 days ^d
		249.7 ±1.4 days ^e
Fe ⁵⁵	2.41±0.12 yr	2.94±0.03 yr ^f
		2.60±0.02 yr ^g
Ho ¹⁶³	33±23 yr	

^a J. Tabailem, J. Phys. Radium **14**, 553 (1953).
^b K. W. Geiger, Phys. Rev. **105**, 1539 (1957).
^c H. W. Wright *et al.*, Nucl. Sci. Eng. **2**, 427 (1957).
^d H. T. Easterday and R. L. Smith, Nucl. Phys. **20**, 155 (1960).
^e I. I. Agiribiceanu and V. Tatu, Compt. Rend. **252**, 3979 (1961).
^f G. L. Brownell and C. J. Maletskos, Phys. Rev. **80**, 1102 (1950).
^g R. P. Schuman, M. E. Jones, and A. C. Mewherter, J. Inorg. Nucl. Chem. **3**, 160 (1956).

Possible changes in the Ho¹⁶³ source during the observation period must be considered with respect to the decay we observe in this sample. Two possible alterations are (a) surface loss of active material, and (b) diffusion of the Ho¹⁶³ atoms into the interior of the aluminum backing plate. Using the Aarhus University computer program based on Linhard's theory of the ranges of heavy ions in matter,⁴ one calculates that 50-keV Ho¹⁶³ ions penetrate aluminum to a mean depth of 6 μg/cm² with a rms deviation of 1.5 μg/cm².⁵ This computation implies that the bulk of the Ho¹⁶³ is located several hundred atom diameters below the surface, so that surface loss appears unlikely. The diffusion of rare earths through aluminum at room temperature is negligible.

Accordingly, we believe that the decrease of Ho¹⁶³ activity that we have observed is due to radioactive

⁴ J. Lindhard and M. Scharff, Phys. Rev. **124**, 128 (1961); J. Lindhard, M. Scharff, and H. E. Schiøtt, Kgl. Danske Videnskab. Selskab, Mat. Fys. Medd. **33**, No. 14 (1963).

⁵ G. Sorensen (private communication).

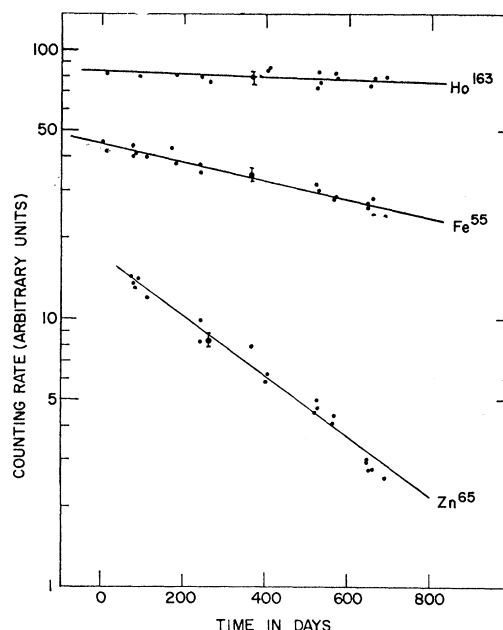


FIG. 2. Decay curves for Ho¹⁶³, Fe⁵⁵, and Zn⁶⁵. The lines are the least-squares fit to the data.

decay of the sample in accord with a half-life of 33±23 yr. In any event this figure must represent a lower limit of the half-life for the decay of this species.

In order to set limits for the Ho¹⁶³ electron-capture Q value, a search has been made for x rays indicative of L electron capture. A cleaved cylindrical NaI(Tl) detector 3.1 mm diam by 6 mm in thickness with a 0.127-mm-thick beryllium window contained in a lead shield has been used. Spectra covering the range 6 to 20 keV were recorded with the source placed directly on the counter window for 142 h. A background was taken for a similar period. An upper limit of 550 counts/h above background was observed in the region of 7 keV in this period. The integrated counting rate of the Ho¹⁶³ source above background observed in the proportional counter was approximately 44 000 counts/h in the M Auger-x-ray region. By assuming an L fluorescence yield $\omega_L=0.200$, an M fluorescence yield $\omega_M=0.015$,⁶ and the same geometrical efficiencies for the x-ray and gas proportional counters, one determines a lower limit for the M -to- L capture ratio of 1040. Since the attenuation of the Ho¹⁶³ M Auger and x radiation due to the 6-μg/cm² penetration depth in aluminum has been neglected, this limit is conservative.

An upper limit for Q , the electron-capture energy for Ho¹⁶³, may now be estimated using the allowed electron-capture result.

$$\lambda_{M_1}/\lambda_{L_1} = [(Q - W_{M_1}) / (Q - W_{L_1})]^2 |g_{M_1}/g_{L_1}|^2,$$

where $\lambda_{M_1}/\lambda_{L_1}$ is the M -to- L electron-capture ratio. One

⁶ R. W. Fink, R. C. Jopson, H. Mark, and C. D. Swift, Rev. Mod. Phys. **38**, 513 (1966).

TABLE II. $\log ft$ values for allowed unhindered β transitions $\frac{5}{2}^- [523] \rightarrow \frac{7}{2}^- [523]$.

Parent			Daughter				
Nucleus	Spin	Process	Nucleus	Spin	State populated (keV)	Q (keV)	$\log ft$
$^{64}\text{Gd}_{97}^{161}$	$(\frac{5}{2}^-)$	β^-	$^{65}\text{Tb}_{96}^{161}$	$(\frac{7}{2}^-)$	418	2020	4.9
$^{67}\text{Ho}_{94}^{161}$	$(\frac{7}{2}^-)$	EC	$^{66}\text{Dy}_{95}^{161}$	$(\frac{5}{2}^-)$	26	1100 ^a	6.0 ^a
$^{68}\text{Er}_{95}^{163}$	$(\frac{5}{2}^-)$	EC, β^+	$^{67}\text{Ho}_{96}^{163}$	$(\frac{7}{2}^-)$	0	1210 \pm 6	4.8
$^{68}\text{Er}_{97}^{165}$	$(\frac{5}{2}^-)$	EC	$^{67}\text{Ho}_{98}^{165}$	$(\frac{7}{2}^-)$	0	371 \pm 5	4.62 \pm 0.02
$^{70}\text{Yb}_{97}^{167}$	$(\frac{5}{2}^-)$	EC, β^+	$^{69}\text{Tm}_{98}^{167}$	$(\frac{7}{2}^-)$	293	1970 \pm 30	4.5
$^{67}\text{Ho}_{96}^{163}$	$(\frac{7}{2}^-)$	EC	$^{66}\text{Dy}_{97}^{163}$	$(\frac{5}{2}^-)$	0		

^a The decay energy has been estimated in the *Nuclear Data Sheets* from systematics. The resulting $\log ft$ value seems to be too high.

estimates the ratio of the densities of the large components of the Dirac wave functions for the M_1 and L_1 electron shells $|g_{M_1}/g_{L_1}|^2 = 0.220$ by linear interpolation of the values given in the tables of Robinson.⁷ Using the electron binding energies $W_{M_1} = 2.05$ keV and $W_{L_1} = 9.05$ keV,⁸ one finds $Q \leq 9.1 \pm 1.5$ keV.

The asymptotic quantum numbers $[523] \frac{7}{2}^-$ have been assigned to the last odd proton in the ground state of Ho^{163} .⁹ The measured nuclear spins of Ho^{161} and Ho^{165} of $\frac{7}{2}$ support this assignment.¹⁰ Similarly the asymptotic quantum numbers $[523] \frac{5}{2}^-$ have been assigned to the last odd neutron in the ground state of Dy^{163} , corresponding to a rotational sequence based on the ground state.¹¹ The electron capture then involves the transition $[523] \frac{7}{2}^-$ to $[523] \frac{5}{2}^-$ or $\Delta N = \Delta N_3 = \Delta \Lambda = 0$, that is, an allowed unhindered transition in the classification of Alaga.¹² The known $\log ft$ values for β processes involving these proton and neutron Nilsson states are collected in Table II. An average $\log ft$ value of 4.7 is found.

A theoretical estimate of the half-life of Ho^{163} is possible using the equation for the electron-capture rate for

an allowed (Gamow-Teller) transition given in Table I of Robinson.⁷

$$\lambda = \frac{4}{3} g_A^2 \langle \sigma \rangle^2 (n_{-1} g_{-1}^2 q_{-1}^2 + n_1 f_1^2 q_1^2),$$

where we have used Konopinski's¹³ notation. The value for the coupling constant is $g_A = g_A^C = 1.63 \times 10^{-49}$ erg cm^3 .¹⁴ We have made the approximation that the β moment is given by its value in mirror nuclei,

$$\langle \sigma \rangle = (-1)^{I+\frac{1}{2}-L} [(2I'+1)/(L+\frac{1}{2})]^{1/2},$$

$$\langle \sigma \rangle^2 = 12/7,$$

where q_{-1}^2 and q_1^2 are the squares of the neutrino energies for capture from the M_1 and M_{11} shells, respectively. We have used the wave functions of Brewer, Harmer, and Hay¹⁵ for $Z=65$. The error incurred by this approximation should be small compared to the error in the β moment. The wave functions are $g_{-1}^2 = 8.78 \times 10^{-5}$ and $f_1^2 = 3.57 \times 10^{-7}$. A value of the half-life $T_{1/2} = 59$ yr was calculated. This value is comparable to the value of 33 ± 23 yr found by direct observation.

We would like to thank Ottar Skilbreid for performing the isotopic separation and Dr. G. Sorensen for the range calculation that we have reported.

⁷ B. L. Robinson, Nucl. Phys. 64, 197 (1965).

⁸ C. M. Lederer, J. M. Hollander, and I. Perlman, *Table of Isotopes* (John Wiley & Sons, Inc., New York, 1967), 6th ed.

⁹ B. R. Mottelson and S. G. Nilsson, Kgl. Danske Videnskab. Selskab, Mat. Fys. Skrifter I, No. 8 (1959).

¹⁰ I. Kindgren, in *Alpha-, Beta-, and Gamma-Ray Spectroscopy*, edited by K. Siegbahn (North-Holland Publishing Co., Amsterdam, 1965).

¹¹ K. Takahashe, J. Phys. Soc. Japan 17, 1229 (1963).

¹² G. Alaga et al., Kgl. Danske Videnskab. Selskab, Mat. Fys. Medd. 29, No. 9 (1955).

¹³ E. S. Konopinski, *The Theory of Beta Radioactivity* (Oxford University Press, London, 1966).

¹⁴ R. K. Bardin, C. A. Barnes, W. A. Fowler, and P. A. Seeger, Phys. Rev. 127, 583 (1962).

¹⁵ H. R. Brewer, D. S. Harmer, and D. H. Hay, Georgia Institute of Technology, GIT-ES-B151-1 (unpublished).