Infrared Dielectric Dispersion and Lattice Dynamics of Uranium Dioxide and Thorium Dioxide

J. D. AXE AND G. D. PETTIT

IBM Watson Research Center, Yorktown Heights, New York (Received 27 April 1966; revised manuscript received 28 June 1966)

Measurements at room temperature of the infrared reflectivity of single crystals of ThO₂ and UO₂ have been carried out and analyzed by Kramers-Kronig relations. A single strong resonance is found in each case with $\omega_{TO} = 279(\pm 2) \text{ cm}^{-1}$, $\omega_{LO} = 568(\pm 4) \text{ cm}^{-1}$, $\epsilon_0 - n^2 = 15.5 \text{ for ThO}_2$; $\omega_{TO} = 278(\pm 2) \text{ cm}^{-1}$, $\omega_{LO} = 556(\pm 4) \text{ cm}^{-1}$, $\omega_{LO} = 556(\pm 4) \text{ cm}^{-1}$, $\omega_{LO} = 568(\pm 4) \text{ cm}^$ cm^{-1} , $\epsilon_0 - n^2 = 17.7$ for UO₂ (TO=transverse optic; LO=longitudinal optic). The best classical dispersionformula fit to the reflectivity yields the following parameters: for ThO₂, $\omega_0 = 282.7$ cm⁻¹, $\epsilon_0 - n^2 = 14.85$, $\overline{\gamma} = 16.2 \text{ cm}^{-1}$; for UO₂, $\omega_0 = 283.2 \text{ cm}^{-1}$, $\epsilon_0 - n^2 = 15.80$, $\overline{\gamma} = 18.5 \text{ cm}^{-1}$. The low-frequency (0.3-Mc/sec) dielectric constant of ThO₂ was measured to be 18.9 ± 0.4 . For UO₂ the mode frequencies are in excellent agreement with those obtained by neutron spectroscopy, and can be reconciled with existing infrared absorption data by taking into account particle-size corrections which can shift the polar mode frequency by a factor $\sim (\epsilon_0 + 2/n^2 + 2)^{1/2} \sim 1.8$ for small spherical samples. The above data are used to evaluate (Szigeti-type) effective charges ($Z'_{Th}^{+4}=2.33$, $Z'_{U}^{+4}=2.42$) and to discuss several approximate relations between the long-wave elastic and optical properties of the fluorite (CaF2-type) lattice. Absorption bands were measured in the two-phonon-combination region in UO2 and their assignment is discussed.

I. INTRODUCTION

 $\mathbf{B}^{\mathrm{ECAUSE}}$ of its technological importance, those bulk properties of uranium dioxide (UO₂) of engineering interest are being extensively studied. Because of its relatively simple structure, UO₂ is of interest from more fundamental points of view as well. Recently Dolling et al.¹ have made a study of the lattice dynamics of UO₂ by inelastic neutron scattering, and are continuing to study antiferromagnetic spin waves by the same techniques.² This paper, which concerns itself with the dielectric response of the UO₂ lattice, supplements to a certain extent the above study. It also supplements optical studies in the near-infrared³ and visible spectral regions.⁴⁻⁶ For purposes of comparison we have also made similar measurements on thorium dioxide (ThO_2) , which is isomorphous with UO_2 and proves to have very similar dielectric properties.

In the following section we present the experimental results which were obtained primarily from an analysis of the specular reflectivity of the two materials in the infrared region. This is a very effective technique for obtaining optical constants in lossy spectral regions.

In the Sec. III these results are analyzed in some detail to obtain such parameters as the (Szigeti) effective charge and force constants which are of more fundamental importance in characterizing the lattice dynamics. The interrelationships between various optical, dielectric, and elastic properties of UO₂ and ThO₂ are discussed from relations previously derived from a simple shell-model treatment of the long-wavelength

properties of the fluorite lattice.7 (The fluorite CaF₂ structure is isomorphous with that of UO_2 and ThO_2 .) An elementary discussion of two-phonon absorption processes is attempted in this section also.

One of the reasons for undertaking this work was to attempt to clear up an apparent discrepancy between the existing far-infrared absorption data on dispersed finely powdered samples⁸ of UO₂ and the neutronscattering data as to the frequency of the fundamental polar mode. Our measurements confirm the neutron results,¹ and we present arguments in support of the conclusion that the shifted absorption frequency reported above is the predictable result of sample sizes small in comparison with the wavelength of the exciting radiation. Under such circumstances, the peaks of the absorption spectra are no longer simply interpretable, at least in terms of resonant frequencies for bulk material.9-11 The notion of size- and shape-dependent normal modes of this type is not new. Frölich⁹ predicted such effects several years ago and related effects have been demonstrated.^{10,12}

II. EXPERIMENTAL

Well characterized single-crystal samples of high quality were used. They were prepared by fusion in an electric-arc furnace and subsequently treated with moist hydrogen at 1750°C. The materials had 99% theoretical density and the oxygen/metal ratio for the UO₂ was

151 676

¹ G. Dolling, R. A. Cowley, and A. D. B. Woods, Can. J. Phys. **43**, 1397 (1965). ² R. A. Cowley and G. Dolling, Bull. Am. Phys. Soc. **11**, 109

^{(1966).} J. L. Bates, Nucl. Sci. Eng. 21, 26 (1965).

⁴ R. J. Ackermann, R. J. Thorne, and G. H. Winslow, J. Opt.

Soc. Am. 49, 1107 (1959) ⁵ A. Companion and G. H. Winslow, J. Opt. Soc. Am. 50,

^{1042 (1960).} ⁶ D. M. Gruen, J. Am. Chem. Soc. 76, 2117 (1954).

⁷ J. D. Axe, Phys. Rev 139, A1215 (1965).

⁸ M. Tsuboi, M. Terada, and T. Shimanouchi, J. Chem. Phys. 36, 1301 (1962).

⁶ H. Frölich, *Theory of Dielectrics* (Oxford University Press, New York, 1948), 2nd ed., pp. 153-155.

¹⁰ M. Hass, Phys. Rev. Letters 13, 429 (1964).

^{1995 (1961)}

¹² D. W. Berreman, Phys. Rev. 130, 2193 (1963).



FIG. 1. The infrared reflectivity of ThO2. The circles represent measured points. The solid line represents the reflectivity calculated from the best fitting classical dispersion formula [Eq. (1)]. Note the change of energy scale at 650 cm⁻¹.

2.001 or better. These samples were prepared at Battelle-Northwest under the direction of Dr. H. J. Anderson.

For reflectivity measurements surfaces of both materials were prepared by standard metallographic procedures, the final polish being on $.05-\mu$ alumina. The reflectivity was measured at room temperature and near-normal incidence using point-by-point comparison with an aluminized surface in a Perkin-Elmer model 301 spectrometer. The frequency range between 60 and 2500 cm⁻¹ was covered and the spectral slit widths varied between 1 and 6 cm⁻¹ (see Figs. 1 and 2). The low-frequency dielectric constant (300 kc/sec) for ThO₂ was determined by measuring the capacitance of a several mm-thick disk of sample material in a conventional Scherring bridge circuit. The conductivity of the

TABLE I. Some quanitities related to the elastic and infrared dielectric properties of ThO₂ and UO₂.

	ThO_{2}	UO_2
Kramers-Kronig analysis		
$\omega_{\rm TO}(\rm cm^{-1})$	$279(\pm 2)$	$278(\pm 2)$
$\omega_{\rm LO}(\rm cm^{-1})$	$568(\pm 4)$	$556(\pm 4)$
$(\epsilon_0 - n^2)$	15.5	17.7
Classical dispersion analysis		
ω_0 (cm ⁻¹)	282.7	283.2
$(\epsilon_0 - n^2)$	14.85	15.80
n^2	4.86	5.51
$\overline{\gamma}(\mathrm{cm}^{-1})$	16.2	18.5
Measured dielectric constants		
€o	$18.9(\pm 0.4)$	24^{a}
n^2	$4.30(\pm 0.05)^{b}$	5.3 ^b
(ϵ_0/n^2)	4.39	4.53
$(\omega_{\rm LO}/\omega_{\rm TO})^2$	4.14	4.00
Effective charge $Z'_{M^{4+}(e)}$	2.42	2.33
$R_{M0}'(10^4 \text{ cm} - \text{dyn}^{-1})$	-21.2	-23.1
$\beta_{\rm obs}(10^{-12} {\rm cm}^2 - {\rm dyn})$	0.518°	0.472^{d}
β_{calo}	•••	0.478

* A. Briggs, Report to the International Atomic Energy Agency, Vienna (unpublished).

(unpublished).
b Reference 18.
P. M. Macedo, W. Capps, and J. B. Wachtman, Jr., J. Am. Ceram. Soc. 47, 12 (1964).
d J. B. Wachtman, M. L. Wheat, H. J. Anderson, and J. L. Bates, J. Nucl. Energy (to be published).
e Equation (8).



FIG. 2. The infrared reflectivity of UO_2 . The circles represent measured points. The solid curve represents the reflectivity calculated from the best fitting classical dispersion formula [Eq. (1)]. Note the change of energy scale at $\sim 650 \text{ cm}^{-1}$.

UO₂ was too high to give a meaningful determination of the low-frequency dielectric constant by this method. It is not believed that this conductivity measurably affected the observed reflectivity.

Two different methods were used to derive the optical constants from the measured reflectivity. The first consisted of a least-squares adjustment of a classical dispersion formula13

$$\epsilon(\omega) - n^2 = (\epsilon_0 - n^2)\omega_0^2 / (\omega_0^2 - \omega^2 + i\bar{\gamma}\omega). \tag{1}$$

Here ϵ_0 is the static dielectric constant and n^2 represents the dielectric constant at frequencies well above those to which the lattice can respond but below all characteristic electronic excitations. $\bar{\gamma}$ is a phenomenological, velocity-dependent damping term. The best-fitting reflectivities of this form are shown as the solid lines in Figs. 1 and 2. Similarly, the solid lines in Figs. 3 and 4 represent the real and imaginary parts of the refractive index as given by Eq. (1) for the best-fitting parameters. The parameters thus obtained are given in Table I. The second method made use of the Kramers-Kronig relation between $R(\omega)$, the fraction of reflected energy at a given frequency, and the phase shift, $\theta(\omega)$, which



FIG. 3. The infrared dielectric response of ThO2. The circles represent values calculated by the Kramers-Kronig relation [Éq. (2)]. The solid curve represents the best fitting classical dispersion formula [Eq. (1)].



FIG. 4. The infrared dielectric response of UO2. The circles represent values calculated by the Kramers-Kronig relation [Eq. (2)]. The solid curve represents the best fitting classical dispersion formula [Eq. (1)].

occurs upon reflection¹³

$$\theta(\omega) = \left(\frac{2\omega}{\pi}\right) \int \frac{d\omega'}{(\omega^2 - \omega'^2)} \ln \left[\frac{R(\omega')}{R(\omega)}\right].$$
 (2)

Once $\theta(\omega)$ is known it is possible to rigorously deduce all the optical constants of interest. Both the numerical integration of Eq. (2) and the subsequent calculations of the optical constants were performed for each data point using an IBM 7094 digital computer.14 The results for the real and imaginary dielectric constants computed in this manner are shown in Figs. 3 and 4. It is possible that the large discrepancy between the dielectric response as deduced by the two methods around ω_0 is a measure of the difference between the true frequency-dependent damping function $\gamma(\omega)^{15,16}$ and the constant average value $\bar{\gamma}$, determined by the fit to the reflectivity. As can be seen from Table I, the two methods are substantially in agreement as to both the position and total strength, $\epsilon_0 - n^2$, of the resonance.

At energies greater than about 600 cm⁻¹ it was possible to study the transmission of thin $(x \ge 50 \mu)$ wafers of UO₂ at both 300 and 77°K. When corrected for reflection losses according to the expression¹⁷ $T = [(1-R)^2 e^{-\alpha x}] / [1-R^2 e^{-2\alpha x}]$ these data yielded the absorption coefficients α given in Fig. 5. Bates³ has published absorption measurements which are qualitatively similar in the 3000- to 900-cm⁻¹ region, but show more pronounced discrepancies at lower energies. The absorption coefficient predicted from the bestfitting damped-oscillator model $\lceil \text{Eq.}(1) \rceil$ has been included in the figure as well. It is somewhat surprising that the absorption constants derived from the reflectivity measurements hold so well so far out in the wings $(\alpha_{\text{max}} \approx 3.6 \times 10^4 \text{ cm}^{-1})$. (Although not included in Fig. 5, absorption constants derived from the Kramers-Kronig analysis of the reflectivity agree about equally well over most of this region.)

III. DISCUSSION

Symmetry arguments support the experimental observation of a single, strong infrared resonance in the fluorite structure. The uranium-dioxide lattice consists of three inter-penetrating face-centered-cubic (fcc) lattices, with spacing $2r_0$. The two oxygen sublattices are at $(\frac{1}{4}, \frac{1}{4}, \frac{1}{4})(2r_0)$ and $(-\frac{1}{4}, -\frac{1}{4}, -\frac{1}{4})(2r_0)$ relative to the uranium sublattice. The long-wavelength (k=0)excitations consist of motions of these rigid sublattices against one another. The nine possible such degrees of freedom must form bases for $2F_{1u}+F_{2g}$ irreducible representations of the point group $O_h(m3m)$. The Raman active F_{2g} vibrations can be shown to involve equal and opposite motions of the oxygen sublattices only and do not affect the dielectric dispersion in first order because there is no net dipole moment within the unit cell. One set of F_{1u} normal coordinates represent the acoustic branches. The degeneracy of the remaining F_{1u} set near k=0 is partially lifted because of the macroscopic electric field associated with its longitudinal component.

The frequency of the transverse branch is given by the pole of the *complex* dielectric response, $\epsilon(\omega_{\rm TO}) = \infty$, whereas the frequency of the longitudinal branch is fixed at the (high-frequency) root, $\epsilon(\omega_{LO}) = 0$. Omitting second-order corrections due to damping, ω_{TO} is given by the peak of the imaginary response, $\epsilon''_{\max}(\omega)$ $=\epsilon''(\omega_{\rm TO})$, and $\omega_{\rm LO}$ by the upper root of the real response, $\epsilon'(\omega_{\rm LO}) = 0$. These characteristic frequencies as determined from the Kramers-Kronig analysis of the ThO₂ and UO₂ are given in Table I. Also the total ionic contributions to the static dielectric constant

$$(\epsilon_0 - n^2)_{\rm KK} = \left(\frac{2}{\pi}\right) \int d\omega \frac{\epsilon''(\omega)}{\omega} \tag{3}$$

determined by numerical integration, are given.

The measured values of the high- and low-frequency dielectric constants provide a check on the validity of



¹³ F. Stern, in *Solid State Physics*, edited by F. Seitz and D. Turn-bull (Academic Press Inc., New York, 1963), Vol. 15, p. 333.

 ¹⁴Some additional details of experimental techniques and equipment and data reduction have been published in an earlier paper: J. D. Axe, J. W. Gaglianello, and J. E. Scardefield, Phys. Rev. 139, A1211 (1965).
 ¹⁵See, for example, M. Lax, J. Phys. Chem. Solids 25, 487 (1964).

⁽¹⁹⁶⁴⁾

¹⁶ H. Bilz, L. Genzel, and H. Happ, Z. Physik 160, 535 (1960). ¹⁷ In making these corrections the reflectivity was assumed to be temperature-independent.



FIG. 6. A comparison of the observed absorption of UO2 dispersed in polyethylene (solid curve) with the LO and TO mode frequencies for a large sample and with the mode frequency ω_s calculated for a small spherical particle (see text). The dashed curve, proportional to the imaginary dielectric response, is in-cluded to show the intrinsic linewidth.

the Lyddane-Sachs-Teller (LST) relation (ϵ_0/n^2) $=(\omega_{\rm LO}/\omega_{\rm TO})^2$. As can be seen from Table I the agreement is considerably worse for the UO_2 than for the ThO₂. In fact there is considerable dispersion in the refractive index of UO₂ at visible frequencies¹⁸ and it is likely that an even lower extrapolated value of n^2 should be used although there seems to be no simple unambiguous procedure for performing the extrapolation. This would have the effect of further accentuating the lack of agreement with the Lyddane-Sachs-Teller relation. Deviations of this order can perhaps be realized through anharmonic lattice potentials.¹⁹ An additional factor to be considered in the case of UO2 is the influence of low-lying electronic states of the uranium ion.²⁰ Electronic effects which may violate the adiabatic assumption of the LST relation would be of an indirect nature in UO2, since electric-dipole transitions between these low-lying levels are forbidden by parity. (See below for additional experimental indication of the possible position of these low-lying electronic levels.)

The fluorite lattice can be partially characterized in terms of parameters which appear in expressions exactly analogous to the Szigeti relations^{21,22} for lattices with diatomic unit cells. A shell-model treatment of the fluorite lattice yields⁷

$$Z_{M}'Z_{O}' = -(9v/4\pi e^{2}) [(\epsilon_{0} - n^{2})/(n^{2} + 2)^{2}] \mu_{0} \omega_{\mathrm{TO}}^{2}, \quad (4)$$

$$R_{MO}' = -\left[(\epsilon_0 + 2)/(n^2 + 2) \right] \mu_0 \omega_{\rm TO}^2 \tag{5}$$

 Z_0' and $Z_M' = -2Z_0'$ are the Szigeti effective charges, R_{MO}' is the effective short-range force constant between *M-O* nearest neighbors and μ_0 is the appropriate reduced mass $m_M m_0 [m_M + 2m_0]^{-1}$. If the assumption of central forces is made a simple relationship between the k=0optical-mode frequencies and the lattice compressibility

 β exists, then

$$\mathcal{B}^{-1} = (\frac{1}{3})(c_{11} + 2c_{12}) = (12r_0)^{-1} \\ \times \{2[(\epsilon_0 + 2)/(n^2 + 2)]\mu_0\omega_{\mathrm{TO}}^2 + \mu_R\omega_R^3\}.$$
(6)

Here $2r_0$ is the cubic-unit-cell spacing, $\mu_R = m_0$ and ω_R being, respectively, the reduced mass and frequency characteristic of the Raman active mode. [Small shellmodel polarization corrections have been omitted in Eq. (3)]. The values of these quantities derived for ThO₂ and UO₂ are given in Table I. When compared to the values for the alkaline-earth fluorides the present effective charges are not doubled, as might be expected, but are increased by some 40%; the force constants are roughly doubled. This may be a measure of the increased importance of short-range polarization in the oxide, or increased covalency, or both. The agreement between the measured compressibility of UO2 and that calculated by Eq. (6) is excellent.

There has recently been considerable interest in the interpretation of the fine structure generally seen on the short-wavelength lattice absorption edge in terms of a process whereby two phonons are created at the expense of a single infrared photon.²³⁻²⁷ Wave-vector conservation requires that the two phonons be characterized by equal but opposite wave vectors, and the absorption features then reflect the singularities in the density of $(+\mathbf{k}, -\mathbf{k})$ pair states of the participating phonons. Even for the relatively simple UO₂ structure, the nine phonon branches give rise to 36 two-phonon combination branches for a general $|\mathbf{k}|$,²⁸ and just the evaluation of the number and type of critical points is an undertaking which hardly seems worthwhile in view of the smooth appearance of the data in Fig. 4. To emphasize this point, Fig. 4 also includes the frequencies of what are but a few of the possible critical points at special symmetry points on the zone boundary. These frequencies have been measured by Dolling et $al.^1$ The formal complexity of the situation does not obscure one assignment, however. Inspection of the dispersion curves of Ref. 1 discloses that the $\omega(\mathbf{k})$ surface which at $\mathbf{k}=0$ is associated with the longitudinal, optically active mode intersects the critical points M and L on the zone boundary at nearly the same frequency, $590(\pm 17)$ cm⁻¹. This may be taken as an indication that the surface intersects the whole zone boundary at about this frequency which we denote for simplicity as ω_{LO}' . A similar state of affairs exists for the surface representing the transverse Raman active branch, the characteristic frequency being $\omega_{TR}'=431$ (± 25) cm⁻¹. Phonon branches of this type which show

- ²³ H. Bilz and L. Genzel, Z. Physik **162**, 53 (1962).
 ²⁴ M. Lax and E. Burstein, Phys. Rev. **97**, 39 (1955).

¹⁸ W. P. Ellis, J. Opt. Soc. Am. 54, 265 (1964). The dispersion in ThO₂ is considerably less in the visible region [see W. P. Ellis and R. M. Lindstrom, Opt. Acta 11, 287 (1964)], and can be represented by the simple expression $n^2 - 1 = c[1 - (\lambda_0/\lambda)^2]^{-1}$ with

represented by the simple expression $n^{2} - 1 = c \lfloor 1 - (\lambda_{0}/\lambda)^{2} \rfloor^{2}$ with $\lambda_{0} = 0.117 \mu$ and c = 3.30. ¹⁹ R. A. Cowley, Advan. Phys. **12**, 421 (1963). ²⁰ R. A. Satten, C. L. Schreiber, and E. Y. Wong, J. Chem. Phys. **42**, 162 (1965). ²¹ B. Szigeti, Proc. Roy. Soc. (London) **A204**, 51 (1950). ²² M. Born and K. Huang, *Dynamical Theory of Crystal Lattices* (Oxford University Press, New York, 1954).

 ²⁶ J. L. Birman, Phys. Rev. 131, 1489 (1963).
 ²⁶ W. J. Turner and W. E. Reese, Phys. Rev. 127, 126 (1962). ²⁷ F. A. Johnson and R. Loudon, Proc. Roy. Soc. (London) A281, 274 (1964).

²⁸ When the two phonons in the pair belong to different single-phonon branches, the result is known as a combination branch. Overtone branches (two phonons from the same branch) are not infrared active in crystals with inversion symmetry.

little dispersion along the zone boundary should be associated with sharp peaks in both the single and combined densities of states. The frequency of the observed peak at about 1010 cm⁻¹ agrees quite well with $\omega_{\rm LO}' + \omega_{\rm TR}'$. The intensity of this two-phonon absorption should vary with temperature as $[1+n(\omega_{TR}')]$ $+n(\omega_{\rm LO}')$]. This predicts a ratio $\alpha_{300}/\alpha_{77}=1.4$ at ${\sim}1000~{\rm cm}^{-1}\!,$ which is in reasonable agreement with observation. Finally we note that the highest phonon frequency in UO_2 calculated by Dolling *et al.* is about 700 cm⁻¹. Therefore the rather broad absorption centered at about 1800 cm⁻¹ cannot be due to twophonon absorption and seems too strong to be accounted for by higher order multiple-phonon processes. This suggests that it is electronic in origin, possibly due to magnetic-dipole transitions between states of the $5f^2$ configuration of U4+.

Upon superficial examination, the far-infraredabsorption measurements on UO2 by Tsuboi et al.8 seem inconsistent with the present results. The result obtained by these workers on fine powdered samples dispersed in a polyethylene film is shown in Fig. 6. It is immediately apparent that the absorption is centered about a frequency (412 cm⁻¹) considerably shifted from the value of ω_{TO} as observed by neutron scattering and confirmed by our dielectric-dispersion analysis. Also the absorption is spread over several hundred cm⁻¹, whereas our dielectric-response data indicate a linewidth of about 25 cm⁻¹. We believe that this result can be largely understood as a manifestation of the shape-dependent frequency of long-wavelength modes. That such effects as we are to describe should exist was first pointed out by Frölich,⁹ but in spite of this and more recent work,^{10,11} the effect is not widely appreciated among infrared spectroscopists who are accustomed to thinking of their measurements as pertaining to intrinsic microscopic phenomena, and who routinely use methods similar to those described above in examining dispersed powdered samples.

Qualitatively the explanation lies in the fact that the dipole moment accompanying long-wavelength polar phonons gives rise to macroscopic electric fields, which for large samples (containing many wavelengths of excitation) provide additional restoring forces for longitudinal modes and increase their frequencies relative to transverse modes. The fundamental modes of particles of comparable or smaller dimensions than the natural excitation wavelengths can no longer be classified into longitudinal and transverse, and the mode frequencies become dependent upon the size and shape of the particles, but are constrained to lie roughly within a region $\omega_{TO} \leq \omega \leq \omega_{LO}$. For weak transitions e.g., combination or overtone bands, the difference between ω_{TO} and ω_{LO} is slight. However, for long-wavelength fundamental absorption, the ratio $\omega_{\rm LO}/\omega_{\rm TO}$ can be two or more, and small-particle absorption spectra will consist of contributions of modes distributed over a wide range of frequencies and not directly interpretable in terms of mode frequencies of macroscopic samples.

The easiest case to discuss quantitatively is that of a spherical sample with radius $r \ll \lambda$, the wavelength of the exciting radiation. On the basis of a simple model of a cubic ionic crystal with one infrared active mode, Frölich has shown that the lowest energy mode is a triply degenerate one of uniform polarization (and is therefore infrared active) with a frequency related to the frequency of the macroscopic sample ω_{TO} by the relation $\omega_s^2 = \omega_{\rm TO}^2(\epsilon_0 + 2/n^2 + 2)$. It is a simple matter to rederive this relationship for a small spherical sample embedded in a homogeneous dielectric medium with a dielectric constant ϵ_M . The result is $\omega_s^2 = \omega_{\rm TO}^2(\epsilon_0)$ $+2\epsilon_M/n^2+2\epsilon_M$). As a reference point we have included this small sphere eigenfrequency $\omega_s = 457 \text{ cm}^{-1}$ calculated for UO₂ in polyethylene ($\epsilon_M = 2.0$). This model is oversimplified in several important respects; nevertheless we feel that the above considerations provide a plausible explanation of the apparent incompatibilities in the existing optical data on UO₂ and serve at the same time as warning of the pitfalls possible in interpreting data on finely divides samples.

IV. SUMMARY

Measurements of the infrared reflectivity of UO_2 and ThO_2 , supplemented with static dielectric constant and near-infrared absorption measurements are used to provide the basis for a rather detailed discussion of the long-wavelength lattice dynamics of these materials. The frequencies of both the longitudinal and transverse optically active modes are obtained and for UO_2 they are in excellent agreement with those found by inelastic neutron scattering. The discrepancy between these results and previous absorption measurements on finely divided UO_2 particles is discussed. The principle source of discrepancy is shown to be the correction to the long-wavelength polar-mode frequencies in small particles.

The Lyddane-Sachs-Teller relation seems less well obeyed in UO₂, which has low-lying electronic levels than in ThO₂. The Szigeti effective charges are about 60% of the formal ionic charges. The measured compressibility of UO₂ is in good agreement with that calculated from the optical-mode frequencies using a Szigeti-type relation. It does not seem profitable at the present time to analyze in detail the two-phonon absorption seen above the fundamental resonance in UO₂. However, one band is readily identifiable in terms of a combination of two average phonon frequencies at the Brillouin-zone edge observed by neutron-scattering experiments.

ACKNOWLEDGMENTS

The authors wish to thank H. J. Anderson and Battelle-Northwest for generously providing the UO_2 and ThO_2 single crystals which made this work possible.